

# **Air Quality Criteria for Ozone and Related Photochemical Oxidants (Second External Review Draft)**

## **Volume I of III**

# **Air Quality Criteria for Ozone and Related Photochemical Oxidants**

## **Volume I**

National Center for Environmental Assessment-RTP Office  
Office of Research and Development  
U.S. Environmental Protection Agency  
Research Triangle Park, NC

## **DISCLAIMER**

This document is a second external review draft for review purposes only and does not constitute U.S. Environmental Protection Agency policy. Mention of trade names or commercial products does not constitute endorsement or recommendation for use.

## **PREFACE**

National Ambient Air Quality Standards (NAAQS) are promulgated by the United States Environmental Protection Agency (EPA) to meet requirements set forth in Sections 108 and 109 of the U.S. Clean Air Act (CAA). Sections 108 and 109 require the EPA Administrator (1) to list widespread air pollutants that reasonably may be expected to endanger public health or welfare; (2) to issue air quality criteria for them that assess the latest available scientific information on nature and effects of ambient exposure to them; (3) to set “primary” NAAQS to protect human health with adequate margin of safety and to set “secondary” NAAQS to protect against welfare effects (e.g., effects on vegetation, ecosystems, visibility, climate, manmade materials, etc); and (5) to periodically review and revise, as appropriate, the criteria and NAAQS for a given listed pollutant or class of pollutants.

In 1971, the U.S. Environmental Protection Agency (EPA) promulgated National Ambient Air Quality Standards (NAAQS) to protect the public health and welfare from adverse effects of photochemical oxidants. The EPA promulgates the NAAQS on the basis of scientific information contained in air quality criteria issued under Section 108 of the Clean Air Act. Following the review of criteria as contained in the EPA document, Air Quality Criteria for Ozone and Other Photochemical Oxidants published in 1978, the chemical designation of the standards was changed from photochemical oxidants to ozone (O<sub>3</sub>) in 1979 and a 1-hour O<sub>3</sub> NAAQS was set. The 1978 document focused mainly on the air quality criteria for O<sub>3</sub> and, to a lesser extent, on those for other photochemical oxidants (e.g., hydrogen peroxide and the peroxyacyl nitrates), as have subsequent revised versions of the ozone document.

To meet Clean Air Act requirements noted above for periodic review of criteria and NAAQS, the O<sub>3</sub> criteria document, *Air Quality Criteria for Ozone and Other Photochemical Oxidants*, was next revised and then released in August 1986; and a supplement, *Summary of Selected New Information on Effects of Ozone on Health and Vegetation*, was issued in January 1992. These documents were the basis for a March 1993 decision by EPA that revision of the existing 1-h NAAQS for O<sub>3</sub> was not appropriate at that time. That decision, however, did not take into account some newer scientific data that became available after completion of the 1986 criteria document. Such literature was assessed in the next periodic revision of the O<sub>3</sub> air quality criteria document (completed in 1996) and provided scientific bases supporting the setting by EPA in 1997 of an 8-h O<sub>3</sub> NAAQS that is currently in force together with the 1-h O<sub>3</sub> standard.

The purpose of this revised air quality criteria document for O<sub>3</sub> and related photochemical oxidants is to critically evaluate and assess the latest scientific information published since that assessed in the above 1996 Ozone Air Quality Criteria Document (O<sub>3</sub> AQCD), with the main focus being on pertinent new information useful in evaluating health and environmental effects data associated with ambient air O<sub>3</sub> exposures. However, some other scientific data are also presented and evaluated in order to provide a better understanding of the nature, sources, distribution, measurement, and concentrations of O<sub>3</sub> and related photochemical oxidants and their precursors in the environment. The document mainly assesses pertinent literature published or accepted for publication through 2004.

The present Second Draft O<sub>3</sub> AQCD (dated August 2005) is being released for public comment and review by the Clean Air Scientific Advisory Committee (CASAC) to obtain comments on the organization and structure of the document, the issues addressed, the approaches employed in assessing and interpreting the newly available information on O<sub>3</sub> exposures and effects, and the key findings and conclusions arrived at as a consequence of this assessment. Public comments and recommendations will be taken into account making any appropriate further revisions to this document for incorporation into the final version of the document to be completed and issued by February 28, 2006. Evaluations contained in the present document will be drawn on to provide inputs to associated PM Staff Paper analyses

prepared by EPA's Office of Air Quality Planning and Standards (OAQPS) to pose options for consideration by the EPA Administrator with regard to proposal and, ultimately, promulgation of decisions on potential retention or revision, as appropriate, of the current O<sub>3</sub> NAAQS.

Preparation of this document was coordinated by staff of EPA's National Center for Environmental Assessment in Research Triangle Park (NCEA-RTP). NCEA-RTP scientific staff, together with experts from other EPA/ORD laboratories and academia, contributed to writing of document chapters. Earlier drafts of document materials were reviewed by non-EPA experts in peer consultation workshops held by EPA. The document describes the nature, sources, distribution, measurement, and concentrations of O<sub>3</sub> in outdoor (ambient) and indoor environments. It also evaluates the latest data on human exposures to ambient O<sub>3</sub> and consequent health effects in exposed human populations, to support decision making regarding the primary, health-related O<sub>3</sub> NAAQS. The document also evaluates ambient O<sub>3</sub> environmental effects on vegetation and ecosystems, man-made materials, and surface level solar UV radiation flux and global climate change, to support decision making on secondary O<sub>3</sub> NAAQS.

NCEA acknowledges the valuable contributions provided by authors, contributors, and reviewers and the diligence of its staff and contractors in the preparation of this draft document.

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Ozone Review Panel  
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## Abbreviations and Acronyms

$\alpha$	alpha, probability value
AA	ascorbic acid
ACh	acetylcholine
ADSS	aged and diluted cigarette smoke
AER	air exchange rate
AEROCE	Atmospheric/Ocean Chemistry Experiment
AHR	airway hyperreactivity
AHSMOG	Adventist Health Study on Smog
AIRS	Aerometric Information Retrieval System
AM	alveolar macrophage
ANF	atrial natriuretic factor
AOP2	antioxidant protein 2
APHEA	Air Pollution on Health: European Approach (study)
AQCD	Air Quality Criteria Document
AQS	Air Quality System
ARIC	Atherosclerosis Risk in Communities (study)
ATLAS	atmospheric model by Kurucz
A/V	surface-to-volume ratio
$\beta$	beta-coefficient; slope of an equation
BAL	bronchioalveolar lavage
BALF	bronchioalveolar lavage fluid
BHR	bronchial hyperresponsiveness
BS	black smoke
BSA	body surface area
BMZ	basement membrane zone
BP	blood pressure
C	concentration
$C \times T$	concentration $\times$ time; concentration times duration of exposure
CAA	Clean Air Act

CADS	Cincinnati Activity Diary Study
CAPs	concentrated ambient particles
CAR	centriacinar region
CASAC	Clean Air Scientific Advisory Committee
CASTNet	Clean Air Status and Trends Network
CC16	Clara cell secretory protein
CCSP	Clara cell secretory protein
CD96	Air Quality Criteria Document for Ozone and Related Photochemical Oxidants; O <sub>3</sub> AQCD
C <sub>dyn</sub>	dynamic lung compliance
CE	continuous exercise
CFD	computational fluid dynamics
C <sub>2</sub> H <sub>5</sub> -H	ethane
C <sub>5</sub> H <sub>8</sub>	isoprene
C <sub>6</sub> H <sub>16</sub>	terpene
CHAD	Consolidated Human Activities Database
CH <sub>3</sub> -CHO	acetaldehyde
CH <sub>3</sub> -CCl <sub>3</sub>	methyl chloroform
CH <sub>3</sub> -CO	acetyl
CH <sub>4</sub>	methane
CI	confidence interval
CIE	Commission Internationale de l'Eclairage (International Commission on Illumination)
CINC	cytokine-induced neutrophil chemoattractant
CLM	chemiluminescence method
CMAQ	Community Model for Air Quality
CO	carbon monoxide
CO <sub>2</sub>	carbon dioxide
COD	coefficient of divergence
COP	Conference of Parties
COPD	chronic obstructive pulmonary disease

CRP	C-reactive protein
CTM	chemistry transport model
DHBA	2,3-dehydroxybenzoic acid
DNA	deoxyribonucleic acid
DOAS	differential optical absorption spectroscopy/spectrometry
DPPC	dipalmitoylglycero-3-phosphocholine
DU	Dobson units
$\epsilon$	epsilon, convergence precision
EBC	exhaled breath condensate (fluid)
ECG	electrocardiographic
EDU	ethylenediurea
EEG	electroencephalographic
ELF	epithelial lining fluid
ENA-78	epithelial cell-derived neutrophil-activating peptide 78
ENSO	El Niño-Southern Oscillation
EPA	U.S. Environmental Protection Agency
ETS	environmental tobacco smoke
F	female
FA	filtered air
FACE	free-air carbon dioxide exposure
$f_B$	breathing frequency
FEF	forced expiratory flow
FEF <sub>25-75</sub>	forced expiratory flow between 25 and 75% of vital capacity
FEV <sub>1</sub>	forced expiratory volume in 1 second
FVC	forced vital capacity
GAM	Generalized Additive Model
GCM	general circulation model
GEE	Generalized Estimating Equation
GHGs	greenhouse gases
GLM	Generalized Linear Model

GM-CSF	granulocyte-macrophage colony stimulating factor
G6PD	glucose-6-phosphate dehydrogenase
GR	glutathione reductase
GSH	glutathione; reduced glutathione
GSHPx	glutathione peroxidase
GSTM1	glutathione S-transferase $\mu$ -1 (genotype)
H <sup>+</sup>	hydrogen ion
H <sub>2</sub> CO, HCHO	formaldehyde
HDMA	house dust mite allergen
HFCs	hydrofluorocarbons
HNE	4-hydroxynonenal
HNO <sub>2</sub> , HONO	nitrous acid
HNO <sub>3</sub>	nitric acid
HO	hydroxyl
HO <sub>2</sub>	hydroperoxyl; hydroperoxy
H <sub>2</sub> O <sub>2</sub>	hydrogen peroxide
HR	heart rate
HRV	heart rate variability
H <sub>2</sub> SO <sub>4</sub>	sulfuric acid
IC	inspiratory capacity
ICAM	intracellular adhesion molecule
ICNIRP	International Commission on Non-Ionizing Radiation Protection
IE	intermittent exercise
Ig	immunoglobulin (e.g., IgA, IgE, IgG, IgM)
IL	interleukin (e.g., IL-1, IL-6, IL-8)
iNOS	inducible nitric oxide synthase; NOS-2
ip	intraperitoneal
IPCC	Intergovernmental Panel on Climate Change
IR	infrared
K <sub>a</sub>	intrinsic mass transfer coefficient/parameter



$K_g$	mass transfer coefficient for gas phase
$K_l$	mass transfer coefficient for liquid phase
$K_r$	reaction rate constant
$K_{TB}$	terminal bronchiole region mass transfer coefficient
LDH	lactic acid dehydrogenase
LIDAR	LIght Detection And Ranging
LIS	lateral intercellular space
LLJ	low-level jet
LOESS	locally estimated smoothing splines
LOP	lipid ozonization products
LPS	lipopolysaccharide
LRT	lower respiratory tract; lower airways
LT	leukotriene (e.g., $LTB_4$ , $LTC_4$ , $LTD_4$ , $LTE_4$ )
LT	local time
M	gas molecule
M	male
M	maximum number of iterations
MAP	mean arterial pressure
MCP	monocyte chemotactic protein
MENTOR	Modeling Environment for Total Risk Studies
MI	myocardial infarction
MIP	macrophage inflammatory protein
MMEF	maximal midexpiratory flow
MONICA	Monitoring Trend and Determinants in Cardiovascular Disease (registry)
MPAN	peroxymethacryloyl nitrate; peroxy-methacrylic nitric anhydride
MPO	myeloperoxidase
mRNA	messenger ribonucleic acid
MSA	metropolitan statistical area
MSL	mean sea level
MT	metallothionein

n, N	number
NAAQS	National Ambient Air Quality Standards
NADPH	reduced nicotinamide adenine dinucleotide phosphate
NAS	Normative Aging Study
NCEA-RTP	National Center for Environmental Assessment Division in Research Triangle Park, NC
NCICAS	National Cooperative Inner-City Asthma Study
ND	not detectable; not detected
NEM	National Ambient Air Quality Standards Exposure Model
NF	national forest
NF- $\kappa$ B	nuclear factor kappa B
NH <sub>4</sub> HSO <sub>4</sub>	ammonium bisulfate
NHAPS	National Human Activity Pattern Survey
NIST	National Institute of Standards and Technology
NK	natural killer (cells)
NL	nasal lavage
NM	national monument
NMHCs	nonmethane hydrocarbons
NMMAPS	National Morbidity, Mortality and Air Pollution Study
NO	nitric oxide
NO <sub>2</sub>	nitrogen dioxide
NO <sub>3</sub> <sup>-</sup>	nitrate
NOS	nitric oxide synthase
NOS-1	neuronal nitric oxide synthase
NOS-2	inducible nitric oxide synthase; iNOS
NOS-3	endothelial nitric oxide synthase
NO <sub>x</sub>	nitrogen oxides
NO <sub>y</sub>	reactive nitrogen system components; sum of NO <sub>x</sub> and NO <sub>z</sub> ; odd nitrogen species
NO <sub>z</sub>	difference between NO <sub>y</sub> and NO <sub>x</sub>
NP	national park

NPP	net primary productivity
NQO1wt	NAD(P)H-quinone oxidoreductase wild type (genotype)
NRC	National Research Council
NTP	National Toxicology Program
NTS	nucleus tractus solitarius
NWR	national wildlife refuge
O( <sup>1</sup> D)	electronically excited oxygen atom
O <sub>2</sub>	ground-state oxygen
O <sub>3</sub>	ozone
O <sub>3</sub> *	electronically excited ozone
O( <sup>3</sup> P)	ground-state oxygen atom
OAQPS	Office of Air Quality Planning and Standards
8-OHdG	8-hydroxy-2'-deoxyguanosine
OH	hydroxyl; hydroxy
OTC	open-top chamber
OVA	ovalbumin
O <sub>x</sub>	odd oxygen species
6PGD	6-phosphogluconate dehydrogenase
p	probability value
P <sub>90</sub>	values of the 90th percentile absolute difference in concentrations
PAF	platelet-activating factor
PAN	peroxyacetyl nitrate; peroxyacetic nitric anhydride
PAR	proximal alveolar region
PBL	planetary boundary layer
PBPK	physiologically based pharmacokinetic (approach)
PCI	picryl chloride
PE	postexposure
PEF	peak expiratory flow
PEM	personal exposure monitor
P <sub>enh</sub>	enhanced pause

PG	prostaglandin (e.g., PGD <sub>2</sub> , PGE, PGE <sub>1</sub> , PGE <sub>2</sub> , PGF <sub>1α</sub> , PGF <sub>2α</sub> )
PI	probability interval
PM	particulate matter
PM <sub>2.5</sub>	fine particulate matter (mass median aerodynamic diameter ≤2.5 μm)
PM <sub>10</sub>	combination of coarse and fine particulate matter
PM <sub>10-2.5</sub>	coarse particulate matter (mass median aerodynamic diameter between 10 and 2.5 μm)
PMNs	polymorphonuclear neutrophil leukocytes; neutrophils
pNEM	Probabilistic National Ambient Air Quality Standard Exposure Model
ppb	parts per billion
ppbv	parts per billion by volume
ppm	parts per million
PPN	peroxypropionyl nitrate; peroxypropionic nitric anhydride
PRB	policy relevant background
PSA	picryl sulfonic acid
PUFA	polyunsaturated fatty acid
PWM	pokeweed mitogen
QCE	quasi continuous exercise
r	correlation coefficient
R	intraclass correlation coefficient
R <sup>2</sup>	multiple correlation coefficient
R'CO	acyl
R'C(O)–O <sub>2</sub>	acyl peroxy
RH	relative humidity
R <sub>L</sub>	total pulmonary resistance
RO <sub>2</sub>	organic peroxy; organic peroxy
ROOH	organic peroxides
ROS	reactive oxygen species
RR	ribonucleotide reductase
RRMS	relatively remote monitoring sites
RT	respiratory tract

SAB	Science Advisory Board
SAC	<i>Staphylococcus aureus</i> Cowan 1 strain
SAMD	S-adenosyl methionine decarboxylase
SBUV	solar backscattered ultraviolet radiation
SC	stratum corneum
SD, S-D	Sprague-Dawley (rat)
SD	standard deviation
SES	socioeconomic status
sGAW	specific airways conductance
SHEDS	Simulation of Human Exposure and Dose System
SNPs	single nucleotide polymorphisms
SO <sub>2</sub>	sulfur dioxide
SO <sub>4</sub> <sup>2-</sup>	sulfate
SOD	superoxide dismutase
SOS	Southern Oxidant Study
SP	substance P
SP	surfactant protein (e.g., SP-A, SP-D)
sRAW	specific airways resistance
STE	stratospheric-tropospheric exchange
STRF	Spatio-Temporal Random Field
SUM06	seasonal sum of all hourly average concentrations ≥ 0.06 ppm
SUM08	seasonal sum of all hourly average concentrations ≥ 0.08 ppm
SZA	solar zenith angle
t	<i>t</i> -test statistical value; <i>t</i> statistic
T <sub>3</sub>	triiodothyronine
T <sub>4</sub>	thyroxine
TAR	Third Assessment Report
TB	terminal bronchioles
TBARS	thiobarbituric acid reactive substances
Tc-DTPA	radiolabeled diethylenetriaminepentaacetic acid; <sup>99m</sup> Tc-DTPA

$T_{CO}$	core temperature
$T_{CTL}$	cytotoxic T-lymphocytes
TLC	total lung capacity
TNF	tumor necrosis factor
TNFR	tumor necrosis factor receptor
TOMS	Total Ozone Mapping Satellite; total ozone mapping spectrometer
TRIM	Total Risk Integrated Methodology (model)
TRIM Expo	Total Risk Integrated Methodology Exposure Event (model)
TSP	total suspended particulate
TWA	time-weighted average
UA	uric acid
UNFCCC	United Nations Framework Convention on Climate Change
URT	upper respiratory tract; upper airways
USGCRP	U.S. Global Change Research Program
UV	ultraviolet
UV-A	ultraviolet radiation of wavelengths 320 to 400 nm
UV-B	ultraviolet radiation of wavelengths 280 to 320 nm
UV-C	ultraviolet radiation of wavelengths 200 to 280 nm
VC	vital capacity
$V_D$	anatomic dead space
$\dot{V}_E$	minute ventilation; expired volume per minute
$\dot{V}O_{2max}$	maximal oxygen uptake (maximal aerobic capacity)
VOC	volatile organic compound
$\dot{V}_P$	volumetric penetration
$\dot{V}_{P50\%}$	volume at which 50% of an inhaled bolus is absorbed
$V_T$	tidal volume
$V_{TB}$	terminal bronchiole region volume
W126	cumulative integrated exposure index with a sigmoidal weighting function
WMO/UNEP	World Meteorological Organization/United Nations Environmental Program
WT	wild type

# EXECUTIVE SUMMARY

## E.1 INTRODUCTION

Tropospheric or “surface-level” ozone is one of six major air pollutants regulated by National Ambient Air Quality Standards (NAAQS) under the U.S. Clean Air Act. As mandated by the Clean Air Act, the U.S. Environmental Protection Agency (EPA) must periodically review the scientific bases (or “criteria”) for the various NAAQS by assessing newly available scientific information on a given criteria air pollutant. This draft document, *Air Quality Criteria for Ozone and other Photochemical Oxidants*, is an updated revision of the 1996 Ozone Air Quality Criteria Document (O<sub>3</sub> AQCD) that provided scientific bases for the current O<sub>3</sub> NAAQS set in 1997.

### E.1.1 Clean Air Act Legal Requirements

Clean Air Act (CAA) Sections 108 and 109 govern establishment, review, and revision of U.S. National Ambient Air Quality Standards (NAAQS).

- Section 108 directs the U.S. Environmental Protection Agency (EPA) Administrator to list ubiquitous (widespread) air pollutants that may reasonably be anticipated to endanger public health or welfare and to issue air quality criteria for them. The air quality criteria are to reflect the latest scientific information useful in indicating the kind and extent of all exposure-related effects on public health and welfare expected from the presence of the pollutant in the ambient air.
- Section 109 directs the EPA Administrator to set and periodically revise, as appropriate, two types of NAAQS: (a) *primary NAAQS* to protect against adverse health effects of listed criteria pollutants among sensitive population groups, with an adequate margin of safety, and (b) *secondary NAAQS* to protect against welfare effects (e.g., impacts on vegetation, crops, ecosystems, visibility, climate, man-made materials, etc.). Section 109 also requires peer review of the NAAQS and their underlying scientific bases by the Clean Air Scientific Advisory Committee (CASAC), a committee of independent non-EPA experts.

## **E.1.2 Chronology of Ozone NAAQS Revisions**

In 1971, the U.S. EPA set primary and secondary standards for total photochemical oxidants. However, based on the criteria review completed in 1978, the original primary and secondary NAAQS set in 1971 were revised in 1979 to focus on O<sub>3</sub> as the indicator for new primary and secondary standards that would be attained when the expected number of days per calendar year with maximum 1-h average O<sub>3</sub> concentrations >0.12 ppm did not exceed one. The NAAQS for ambient O<sub>3</sub> were revised in 1997 by replacing the 1-h standards with an 8-h primary standard that is met when the 3-year average of the annual fourth highest daily maximum 8-h average concentration is <0.08 ppm. The 1997 primary NAAQS was based on scientific data from controlled human exposure, laboratory animal, and epidemiological studies and associated analyses presented in the 1996 O<sub>3</sub> AQCD and in the 1996 O<sub>3</sub> Staff Paper (U.S. Environmental Protection Agency, 1996b).

- This revised O<sub>3</sub> AQCD is now being prepared by ORD's National Center for Environmental Assessment (NCEA) to support EPA's ongoing Congressionally-mandated periodic review of O<sub>3</sub> NAAQS under a consent decree (court-ordered) schedule that calls for issuance of the revised AQCD in final form by February 28, 2006. This document assesses the latest available scientific information (published mainly through December 2004) judged to be useful in deriving criteria as scientific bases for decisions on possible revision of the current O<sub>3</sub> NAAQS.
- A separate EPA O<sub>3</sub> Staff Paper will draw upon key findings/conclusions from this document, together with other analyses, to develop and present options for consideration by the EPA Administrator regarding review and possible revision of the O<sub>3</sub> NAAQS.

## **E.1.3 Document Organization and Structure**

Volume I of this document consists of the present Executive Summary and eleven main chapters of this revised O<sub>3</sub> AQCD. Those main chapters focus primarily on interpretative evaluation of key information, whereas more detailed descriptive summarization of pertinent studies and/or supporting analyses are provided in accompanying annexes. Volume II contains the annexes for Chapters 4 through 7, whereas Volume III contains the annex for Chapter 9.



1 The topics covered in the main chapters are as follows:

- 2 ● This Executive Summary summarizes key findings and conclusions from Chapters 1 through  
3 11 of this revised O<sub>3</sub> AQCD.
- 4
- 5 ● Chapter 1 provides a general introduction, including an overview of legal requirements,  
6 the chronology of past revisions of O<sub>3</sub>-related NAAQS, and orientation to the structure of  
7 the document.
- 8
- 9 ● Chapters 2 and 3 provide background information on atmospheric chemistry/physics of O<sub>3</sub>  
10 formation, air quality, and exposure aspects to help to place ensuing discussions of O<sub>3</sub> health  
11 and welfare effects into perspective.
- 12
- 13 ● Chapters 4 through 7 then assess dosimetry aspects, experimental (controlled human exposure  
14 and laboratory animal) studies, and epidemiologic (field/panel; other observational) studies.
- 15
- 16 ● Chapter 8 provides an integrative synthesis of key findings and conclusions derived from the  
17 preceding chapters with regard to ambient O<sub>3</sub> concentrations, human exposures, dosimetry,  
18 and health effects.
- 19
- 20 ● Chapter 9 deals with effects of O<sub>3</sub> on vegetation, crops, and natural ecosystems, whereas  
21 Chapter 10 evaluates tropospheric O<sub>3</sub> relationships to alterations in surface-level UVB flux  
22 and climate change and Chapter 11 assesses materials damage (these all being key types of  
23 welfare effects of relevance to decisions regarding secondary O<sub>3</sub> NAAQS review).
- 24
- 25

## 26 **E.2 ATMOSPHERIC CHEMISTRY AND PHYSICS OF TROPOSPHERIC** 27 **OZONE FORMATION**

28 Key findings/conclusions from Chapter 2 regarding the chemistry and physics of surface-  
29 level O<sub>3</sub> formation include the following:

- 1 ● Ozone ( $O_3$ ) is a secondary pollutant formed by atmospheric reactions involving two classes  
2 of precursor compounds, volatile organic compounds (VOCs) and nitrogen oxides ( $NO_x$ ).  
3 Carbon monoxide also contributes to  $O_3$  formation.  
4
- 5 ● The formation of  $O_3$  and associated compounds is a complex, nonlinear function of many  
6 factors, including the intensity and spectral distribution of sunlight; atmospheric mixing and  
7 other atmospheric processes; and the concentrations of the precursors in ambient air.  
8
- 9 ● The photochemical oxidation of almost all anthropogenic and biogenic VOCs is initiated by  
10 reaction with hydroxyl (OH) radicals. At night, when they are most abundant,  $NO_3$  radicals  
11 oxidize alkenes. In coastal and other select environments, Cl and Br radicals can also initiate  
12 the oxidation of VOCs.  
13
- 14 ● In urban areas, basically all classes of VOCs (alkanes, alkenes, aromatic hydrocarbons,  
15 carbonyl compounds, etc.) and CO are important for ozone formation. Although knowledge  
16 of the oxidative mechanisms of VOCs has improved over the past several years, gaps in  
17 knowledge involving key classes, such as aromatic hydrocarbons, still remain. For example,  
18 only about half of the carbon initially present in aromatic hydrocarbons in smog chamber  
19 studies form compounds that can be identified.  
20
- 21 ● In addition to gas phase reactions, reactions also occur on the surfaces of or within cloud  
22 droplets and airborne particles. Most of the well-established multiphase reactions tend to  
23 reduce the rate of  $O_3$  formation in polluted environments. Direct reactions of  $O_3$  and  
24 atmospheric particles appear to be too slow to reduce  $O_3$  formation significantly at typical  
25 ambient PM levels.  
26
- 27 ● Oxidants other than  $O_3$  are found in the gas phase and in particles. The chemistry occurring  
28 in particle bound-water and, hence, the mechanisms leading to the formation of reactive  
29 oxygen species in particles are largely unknown.  
30

- 1 ● Our basic understanding of meteorological processes associated with summertime O<sub>3</sub>  
2 episodes has not changed over the past several years. However, the realization is growing  
3 that long-range transport processes are important for determining O<sub>3</sub> concentrations at the  
4 surface . In addition to synoptic scale flow fields, nocturnal low-level jets are capable  
5 of transporting pollutants hundreds of km from their sources in either the upper boundary  
6 layer or the lower free troposphere. Turbulence then brings O<sub>3</sub> and other pollutants to  
7 the surface.  
8
- 9 ● Even in the absence of photochemical reactions in the troposphere, some O<sub>3</sub> would be found  
10 near the earth's surface due to its downward transport from the stratosphere. Intrusions of  
11 stratospheric O<sub>3</sub> that reach the surface are rare. Much more common are intrusions that  
12 penetrate to the middle and upper troposphere. However, O<sub>3</sub> transported to the middle and  
13 upper troposphere can still affect surface concentrations through various mechanisms that  
14 mix air between the planetary boundary layer and the free troposphere above.  
15
- 16 ● Chemistry transport models are used to improve understanding of atmospheric chemical and  
17 physical processes, as well as to develop air pollution control strategies. The performance of  
18 these models must be evaluated by comparison with field data as part of an iterative cycle of  
19 model improvement and subsequent evaluation. Discrepancies between model predictions  
20 and observations can be used to point out gaps in current understanding and thus to improve  
21 parameterizations of atmospheric chemical and physical processes.  
22
- 23 ● Model evaluation does not merely involve a straightforward comparison between model  
24 predictions and observed concentration fields of a pollutant of interest (e.g., O<sub>3</sub>). Such  
25 comparisons may not be meaningful because it is difficult to determine if agreement between  
26 measurements and model predictions truly represents an accurate treatment of physical and  
27 chemical processes in the model or the effects of compensating errors in model routines.  
28
- 29 ● The main methods currently in use for routine monitoring of ambient ozone are based on  
30 chemiluminescence or UV absorption. Measurements at most ambient monitoring sites are  
31 based on UV absorption. Both of these methods are subject to interference by other

1 atmospheric components. Studies conducted in Mexico City and in a smog chamber have  
2 found positive interference, but a few studies conducted in urban plumes have not found  
3 evidence for significant positive interference in the UV absorption technique.  
4  
5

### 6 **E.3 ENVIRONMENTAL DISPERSAL, AMBIENT CONCENTRATIONS,** 7 **AND HUMAN EXPOSURE TO OZONE**

8 Key findings/conclusions derived from Chapter 3 are as follows:

- 9 ● Ozone is monitored in populated areas in the United States during “ozone seasons”, which  
10 vary in length depending on location. All monitors should be operational from May to  
11 September. However, in many areas, O<sub>3</sub> is monitored throughout the year.  
12
- 13 ● The median of the mean daily maximum 8-h average O<sub>3</sub> concentration from May to  
14 September 2000 to 2004 across the U.S. was 0.049 ppm on a countywide average basis.  
15 Ninety five per cent of countywide mean daily maximum 8-h average O<sub>3</sub> concentrations were  
16 less than 0.057 ppm for the same period. Because most monitors are located in the East,  
17 these values should not be taken to represent conditions across the country.  
18
- 19 ● The daily maximum 1-h O<sub>3</sub> concentrations tend to be much higher in large urban areas or in  
20 areas downwind of large urban areas. For example, daily maximum 1-h O<sub>3</sub> concentrations in  
21 Houston, TX approached 0.20 ppm during the same period.  
22
- 23 ● Daily maximum 8-h average O<sub>3</sub> concentrations are lower than, but are highly correlated  
24 with, 1-h daily maximum O<sub>3</sub> concentrations. For example, in the Baltimore, MD area, the  
25 correlation coefficient between the two quantities was 0.98 for data obtained from May to  
26 September 1994 to 2004.  
27
- 28 ● Within individual MSAs, O<sub>3</sub> tends to be well correlated across monitoring sites. However,  
29 there can be substantial spatial variations in concentrations. Ozone in city centers tends to be  
30 lower than in regions either upwind or downwind of the center, because of titration by NO  
31 emitted by motor vehicles.

- 1 ● Ozone concentrations tend to peak in early- to mid-afternoon in areas where there is strong  
2 photochemical production and later in the day in areas where transport is more important in  
3 determining O<sub>3</sub> abundance.  
4
- 5 ● Summertime maxima in O<sub>3</sub> concentrations occur in areas in the United States where there is  
6 substantial photochemical activity involving O<sub>3</sub> precursors emitted from human activities.  
7 Maxima can occur anytime from June through August.  
8
- 9 ● Springtime maxima are observed in relatively remote sites in the western United States and  
10 at various other relatively unpolluted sites throughout the Northern Hemisphere. Relatively  
11 high O<sub>3</sub> concentrations can also be found during winter in several cities throughout the  
12 southern United States.  
13
- 14 ● Long-term trends in O<sub>3</sub> concentrations reflect notable decreases over time throughout the  
15 United States, with decreases nationwide of approximately 29% in 2nd highest 1-h O<sub>3</sub>  
16 concentrations from 1980 to 2003 and of about 21% in 4th highest 8-h O<sub>3</sub> concentrations  
17 during the same time period.  
18
- 19 ● These trends include dramatic decreases from peak 1-h O<sub>3</sub> levels of 0.4 to 0.6 ppm seen in  
20 the Los Angeles area at times in the late 1950's to 1970's to current peak levels of 0.17 ppm  
21 and 0.15 ppm (1-h and 8-h avg, respectively) seen in the Los Angeles basin during  
22 2000-2003.  
23
- 24 ● Downward trends in the upper tail of the O<sub>3</sub> concentration distribution do not reflect trends  
25 for O<sub>3</sub> values towards the center of the O<sub>3</sub> concentration distribution nationwide. These latter  
26 concentrations have remained more or less constant, and O<sub>3</sub> values in the lower tail of the  
27 distribution show some evidence of slight increases.  
28
- 29 ● Policy relevant background (PRB) O<sub>3</sub> concentrations are used for assessing risks to human  
30 health associated with O<sub>3</sub> produced from anthropogenic sources in the United States, Canada

1 and Mexico. Because of the nature of the definition of PRB concentrations, they cannot be  
2 derived from observations directly, instead they must be derived from model estimates.

- 3
- 4 ● Current model estimates indicate that PRB O<sub>3</sub> concentrations in the United States surface air  
5 are generally 0.015 ppm to 0.035 ppm. Such concentrations decline from spring to summer  
6 and are generally <0.025 ppm under conditions conducive to high O<sub>3</sub> episodes. PRB Ozone  
7 concentrations may be higher, especially at elevated sites during the spring, due to enhanced  
8 contributions from (a) pollution sources inside and outside North America and  
9 (b) stratospheric O<sub>3</sub> exchange.
  - 10
  - 11 ● Sufficient data for other oxidants (e.g., H<sub>2</sub>O<sub>2</sub>, PAN) and oxidation products (e.g., HNO<sub>3</sub>,  
12 H<sub>2</sub>SO<sub>4</sub>) in the atmosphere are not available for use in epidemiologic time series studies.  
13 Limited data for oxidants besides O<sub>3</sub> in the gas and particle phases suggest that their  
14 combined concentrations are probably <10 % that of O<sub>3</sub>.
  - 15
  - 16 ● Relationships between O<sub>3</sub> and PM<sub>2.5</sub> are complex, in part because PM is not a distinct  
17 chemical species, but is a mix of primary and secondary species. For example, PM<sub>2.5</sub>  
18 concentrations were positively correlated with O<sub>3</sub> during summer, but negatively correlated  
19 with O<sub>3</sub> during the winter at Ft. Meade, MD. Similar relationships were found for PM<sub>10</sub> and  
20 O<sub>3</sub> in data collected in a number of urban areas during the 1980s.
  - 21
  - 22 ● Humans are exposed to O<sub>3</sub> either outdoors or in various microenvironments. Ozone in  
23 indoor environments results mainly from infiltration from outdoors. Once indoors, O<sub>3</sub> is  
24 removed by deposition on and reaction with surfaces and reactions with other pollutants.  
25 Hence, O<sub>3</sub> levels indoors tend to be notably lower than outdoor O<sub>3</sub> concentrations measured  
26 at nearby monitoring sites, although the indoor and ambient O<sub>3</sub> concentrations tend to vary  
27 together (i.e., the higher the ambient, the higher the indoor O<sub>3</sub> levels).
  - 28
  - 29 ● Personal exposure to O<sub>3</sub> tends to be positively associated with time spent outdoors.  
30 Although O<sub>3</sub> concentrations obtained at stationary monitoring sites may not explain the

1 variance in individual personal exposures, they appear to serve reasonably well as surrogate  
2 measures for aggregate personal exposures.

- 3
- 4 ● Atmospheric reactions between O<sub>3</sub> and certain other ambient airborne contaminants, e.g.,  
5 terpenes emitted by vegetation or wood products, contribute to generation of ultrafine  
6 particles, with formation of such particles being observed in both urban and rural areas.  
7 These reactions also occur in indoor environments and involve O<sub>3</sub> infiltrating from outdoors  
8 and terpenes emitted by household products (e.g., air fresheners). Gaseous products  
9 resulting from such reactions may also be toxic.

#### 12 **E.4 DOSIMETRIC STUDIES**

13 Chapter 4 discusses dosimetric issues, including factors that are important to consider in  
14 attempting animal-to-human extrapolations of experimentally-induced O<sub>3</sub> effects.

- 15
- 16 ● Dosimetric studies seek to quantify dose and factors affecting the dose of O<sub>3</sub> and/or its active  
17 metabolites at specific lung regions, target tissues, or cells.
  - 18
  - 19 ● In both humans and animals, the efficiency of O<sub>3</sub> uptake is greater in the nasal passages than  
20 the oral pathway. In the lower respiratory tract, increasing tidal volume increases O<sub>3</sub> uptake,  
21 whereas increasing flow or breathing frequency decreases O<sub>3</sub> uptake. As flow is increased,  
22 O<sub>3</sub> uptake shifts to the smaller peripheral airways.
  - 23
  - 24 ● In adult human females relative to males, the smaller airways and associated larger surface-  
25 to-volume ratio enhance local O<sub>3</sub> uptake and cause somewhat reduced penetration of O<sub>3</sub> into  
26 the distal lung. However, it is not clear from these findings if the actual anatomical location  
27 of O<sub>3</sub> uptake differs between males and females.
  - 28
  - 29 ● Similarly exposed individuals vary in the amount of actual dose received, but O<sub>3</sub> uptake is  
30 not predictive of intersubject variability in FEV<sub>1</sub>.

- 1 ● The efficiency of O<sub>3</sub> uptake is chemical-reaction rate dependent and the reaction products  
2 (hydrogen peroxide, aldehydes, and hydroxyhydroperoxides) created by ozonolysis of lipids  
3 in ELF and cell membranes appear to mediate O<sub>3</sub> toxicity.  
4
- 5 ● Ozone uptake in humans is increased by exposure to NO<sub>2</sub> and SO<sub>2</sub> and decreased during  
6 the O<sub>3</sub> exposure. This suggests that an inflammatory response during exposure to NO<sub>2</sub> and  
7 SO<sub>2</sub> may elicit increased production of O<sub>3</sub>-reactive substrates in the epithelial lining fluid and  
8 that these substrates are depleted by O<sub>3</sub> exposure but not by NO<sub>2</sub> and SO<sub>2</sub> exposures.  
9
- 10 ● New experimental work in rats suggests that the primary site of acute O<sub>3</sub>-induced cell injury  
11 is the conducting airways, whereas prior modeling studies suggested that the proximal  
12 alveolar and centriacinar regions may be principal O<sub>3</sub> target sites.  
13
- 14 ● In most clinical studies, humans are exposed to O<sub>3</sub> during exercise. Under these conditions,  
15 the switch from nasal to oral breathing, coupled with increases in respiratory flow (as occurs  
16 during exercise), causes a shift in the O<sub>3</sub> dose distribution, thusly allowing O<sub>3</sub> to penetrate  
17 deeper into the lung and thereby increasing the potential for damage to bronchiolar and  
18 alveolar tissues.  
19
- 20 ● Comparisons of acute exposures in rats and humans suggest that, though both species have  
21 similar qualitative responses to O<sub>3</sub> exposure, there are interspecies mechanistic disparities  
22 that necessitate careful comparisons of dose-response relationships. Currently available data  
23 suggest that lowest observable effect levels in resting rats are approximately 4- to 5-fold  
24 higher than for exercising humans for toxicological endpoints, including BAL protein and  
25 BAL PMNs.  
26  
27

## 28 **E.5 ANIMAL TOXICOLOGY ASPECTS**

29 Key toxicology findings/conclusions from laboratory animal studies discussed in Chapter 5  
30 include:  
31



## 1 **E.5.1 Respiratory Tract Effects of Short-Term Exposures to Ozone**

2 In general, O<sub>3</sub> concentration and duration of exposure (C and T), respectively, determine  
3 the dose and resultant health effects of O<sub>3</sub>. Concentration usually dominates the response, with  
4 the impact of T being C-dependent (at higher Cs, the impact of T tends to be greater).

### 5 *Effects on Pulmonary Function*

- 6  
7 ● Rapid shallow breathing, which is not protective, but does cause a more evenly distributed  
8 injury pattern, is the most common change in pulmonary function induced by acute (1-8 h)  
9 O<sub>3</sub> exposure of ~0.2 ppm. Decreased lung volumes are observed in rats with acute exposures  
10 at levels of 0.5 ppm. Breathing mechanics (compliance and resistance) are affected at  
11 exposures of ~1.0 ppm.
- 12  
13 ● Attenuation of pulmonary function decrements occurs with 5 days of repeated acute O<sub>3</sub>  
14 exposures, which are not accompanied by concurrent attenuation of lung injury and  
15 morphological changes, indicating that the attenuation does not result in protection against  
16 all the effects of O<sub>3</sub>.
- 17  
18 ● Ozone-induced airway hyperresponsiveness (AHR) occurs in laboratory animals with acute  
19 exposures ( $\leq 1$  h) in the range of 0.5 to 1.0 ppm. Animal studies have shown that O<sub>3</sub>  
20 exposure can augment OVA-induced AHR. A temporal relationship exists between  
21 inflammatory cell influx and O<sub>3</sub>-induced AHR, but inflammation is not a prerequisite of  
22 AHR. Repeated O<sub>3</sub> exposures enhance AHR, possibly by modulating rapidly adapting  
23 airway receptors or by altering the structure of conducting airways. In human asthmatics,  
24 AHR appears to be due, in part, to chronic inflammation and airway remodeling.
- 25  
26 ● Studies using repeated O<sub>3</sub> exposure ( $\leq 0.3$  ppm) of nonsensitized laboratory animals have  
27 shown equivocal results. A few studies in sensitized laboratory animals are consistent with  
28 the O<sub>3</sub>-induced exacerbation of AHR reported in atopic humans with asthma. However,  
29 extrapolation of these data is difficult due to interindividual and interspecies differences in  
30 responsiveness to bronchoprovocation and possible adaptation of airway responsiveness with  
31 long-term, repeated O<sub>3</sub> exposures.

## 1 *Other Respiratory Tract Effects of Ozone*

- 2 ● Due to its high reactivity, O<sub>3</sub> penetrates only about 0.1 to 0.2 μm into the extracellular lining  
3 fluid (ELF) of the respiratory tract. Ozone interacts with a wide range of components in ELF  
4 that include polyunsaturated fatty acids, cholesterol, amino acid residues, reduced  
5 glutathione, uric acid, vitamins C and E, and free amino acids. Ozone's toxicity is dependent  
6 upon a cascade of reaction products, including ozonide, aldehyde, and hydroperoxide.  
7 Saturated phospholipids are thought to reduce the local dose and limit site-specific cell injury  
8 from O<sub>3</sub> exposure.  
9
- 10 ● Antioxidants present in ELF act to protect lung tissue from O<sub>3</sub>-induced injury, but even with  
11 environmentally relevant exposures, the reactivity of O<sub>3</sub> is not quantitatively fully quenched.  
12 Thus, cell injury occurs in both the upper and lower respiratory tract. Short-term exposures  
13 to <1 ppm O<sub>3</sub> increase antioxidant metabolism. Previous O<sub>3</sub> exposure does not appear to be  
14 protective upon re-exposures.  
15
- 16 ● Both short- and long-term exposures to O<sub>3</sub> have been shown to enhance lung xenobiotic  
17 metabolism, possibly due to changes in the number and function of bronchiolar epithelial  
18 Clara cells and alveolar epithelial Type 2 cells. Elevations in enzyme activity appear to  
19 increase as a function of age, suggesting that O<sub>3</sub> exposure can cause greater lung injury in the  
20 older animal. Some studies found an effect on liver xenobiotic enzymes with exposure to O<sub>3</sub>  
21 concentrations as low as 0.1 ppm, whereas others did not detect alterations in metabolic  
22 enzymes even at 1 ppm, the effects appearing to be highly species-specific.  
23
- 24 ● Acute exposures of 0.1 ppm O<sub>3</sub> disrupt the barrier created by airway mucosa in the normal  
25 lung, resulting in an increase in serum proteins, bioactive mediators, and neutrophils in the  
26 interstitium and air spaces of the lung. In rats, a single 3 h exposure to 0.5 ppm O<sub>3</sub> produces  
27 a significant increase in both lung permeability and inflammation. Ozone-induced  
28 permeability changes appear to occur predominantly in the trachea and bronchioalveolar  
29 regions compared to nasal passages. Species differences exist in responses, with guinea pigs  
30 being the most responsive; rabbits the least; and rats, hamsters, and mice intermediate. With  
31 continuing exposure, the increases in BALF protein and PMNs typically peak after a few

1 days and return toward control levels even with continuing exposure. Though inflammation  
2 and increased permeability occur somewhat concurrently, they are distinct events controlled  
3 by independent mechanisms.

- 4
- 5 ● Important mechanisms of O<sub>3</sub>-induced inflammation and injury involve inflammatory  
6 cytokines and chemokines, which are released as a result of stimulation or injury of  
7 macrophages, epithelial cells and PMNs. In vitro exposures to O<sub>3</sub> induce release of the  
8 cytokines IL-6, TNF- $\alpha$ , IL-1 $\beta$ , and IL-8. In vivo exposures of O<sub>3</sub> induce release of MIP-2,  
9 IL-6, MIP-1 $\alpha$ , CINC, eotaxin and fibronectin. Studies utilizing antibodies to selected pro- or  
10 anti-inflammatory cytokines suggest a role of TNF- $\alpha$ , interleukin-10 (IL-10) and IL-1 $\beta$  in  
11 O<sub>3</sub>-induced changes in permeability, inflammation and cytokine release.
- 12
- 13 ● Cell adhesion molecules (e.g., ICAM-1) and extracellular matrix proteins (e.g., fibronectin)  
14 modulate O<sub>3</sub>-induced lung inflammation and injury. Ozone exposure also affects  
15 macrophage functions by increasing their production of nitric oxide, superoxide anion  
16 and PGE<sub>2</sub>.
- 17
- 18 ● Mucociliary clearance is affected in most test species at just under 1 ppm, with lower levels  
19 (~0.1 ppm) increasing clearance and somewhat higher levels decreasing clearance. At O<sub>3</sub>  
20 exposures of 0.1 to 1.2 ppm, alveolar macrophage (AM) function is disrupted and the  
21 number of AM are increased. Ozone exposures are linked to decreased resistance to  
22 microbial pathogens.
- 23
- 24 ● Ozone exposures can enhance or suppress immune responsiveness, depending on the species  
25 studied, the concentration of O<sub>3</sub>, the route of exposure of allergen, and exposure timing.  
26 Continuous exposure to O<sub>3</sub> impairs immune responses for the first several days of exposure,  
27 followed by an adaptation to O<sub>3</sub> that allows a return of normal immune responses. Most  
28 species show little effect of O<sub>3</sub> exposures prior to immunization, but exhibit suppression of  
29 responses to antigen with O<sub>3</sub> exposures post-immunization. Ozone exposures are linked to a  
30 possible interaction between the innate and acquired immune system and a shift in the  
31 immune response towards a Th-2-like pattern. Surfactant proteins A and D, which have an

1 immunomodulatory function in protecting against oxidative stress, are affected by O<sub>3</sub>  
2 exposures. Ozone exposures at levels of 0.1 to 1 ppm O<sub>3</sub> for 1 week have been shown to  
3 cause, in general, increased mortality and morbidity, decreased clearance, increased bacterial  
4 growth, and increased severity of infection at exposure.

- 5  
6 ● Age, gender, nutritional status, genetic variability, exercise and exposure to co-pollutants are  
7 all factors which can impact the effects of O<sub>3</sub>. Control of the ventilatory response to O<sub>3</sub> is  
8 determined, at least in part, by genetic factors. Genetic loci that modulate pulmonary  
9 responses to O<sub>3</sub> differ from each other and from loci controlling inflammatory responses.  
10 The effects of age and gender on lung inflammation are not well characterized, but exercise  
11 during O<sub>3</sub> exposure clearly generally increases susceptibility.
- 12  
13 ● Collagen increases with O<sub>3</sub> exposure and can persist after exposure stops. Rats exposed  
14 acutely or subchronically to 0.4 ppm O<sub>3</sub> showed centriacinar thickening of septa. Collagen  
15 content decreased with postexposure recovery time but not the structural fibrotic changes in  
16 ductular septa and respiratory bronchioles, suggesting that subchronic O<sub>3</sub> exposures in rats  
17 creates a progression of structural lung injury that can evolve to a more chronic form that  
18 likely includes fibrosis.
- 19  
20 ● Ozone-induced alterations in lung structure have been shown across a variety of species  
21 repeatedly exposed to O<sub>3</sub> concentrations as low as 0.15 ppm. Cells in the centriacinar region  
22 (CAR) are the primary targets of O<sub>3</sub>, but ciliated epithelial cells in the nasal cavity and  
23 airways and Type 1 epithelial cells in the gas exchange region are also targeted. Ozone-  
24 induced fibrotic changes in the CAR are maximal at 3 d of exposure and recover 3 d  
25 postexposure with exposures to 0.2 ppm in rodents. Rats with induced allergic rhinitis are  
26 more susceptible to 0.5 ppm than are controls. The proximal respiratory bronchiole receives  
27 the most acute epithelial injury from exposures ≤ 1 ppm, while metabolic effects are greatest  
28 in the distal bronchioles and minor daughter airways.
- 29

## **E.5.2 Respiratory Tract Effects of Chronic (Long-Term) Exposures to Ozone**

A variety of respiratory tract effects have been shown to occur as the result of more chronic, longer-term exposures of laboratory animals to O<sub>3</sub>. Some of the more notable types of effects are as follow.

- □ Chronic O<sub>3</sub> exposures in a range of 0.5 to 1.0 ppm induce a pattern of epithelial hyperplasia which is similar to the pattern of inflammation, with a peak over the first few day, a drop, and then disappearance. In contrast, fibrotic changes in lung tissue increase very slowly over months of exposure, and, after exposure ceases, the changes sometimes persist or increase. Compared to continuous exposure regimens, seasonal episodic exposures demonstrated remodeling in the distal airways, abnormalities in tracheal basement membrane, eosinophil accumulation in conducting airways, and decrements in airway innervation. Also, long-term O<sub>3</sub> exposures have demonstrated that repeated daily exposure of rats to an episodic profile of O<sub>3</sub> caused small, but significant decrements in lung function that were consistent with early indicators of focal fibrogenesis in the proximal alveolar region, without overt fibrosis.

## **E5.3 Other Types of Ozone Exposure Effects Observed in Laboratory Animal Models**

### *Systemic Effects of Ozone*

- Decreased heart rate, core temperature, and blood pressure, all collectively termed the hypothermic response, are other types of effects observed at concentrations of 0.3 to 0.5 ppm. Concentrations of O<sub>3</sub> ≥0.5 ppm cause tissue edema (possibly mediated by atrial natriuretic factor). Additionally, O<sub>3</sub>-induced production of platelet-activating factor and oxysterols suggest mechanisms of cardiovascular injury.
- Neurobehavioral effects attributed to O<sub>3</sub> exposure (0.2 to 1.0 ppm) include decreased motor activity, short- and long-term memory deficits, increased freezing behavior, and decreased exploratory behaviors. Near-ambient exposures to O<sub>3</sub> elicit neuroendocrine effects, including morphological and hormonal changes in the pituitary-thyroid-adrenal axis and alterations of visual and olfactory neural pathways.

- 1 ● No noticeable neurobehavioral or somatic effects have been observed with prenatal  
2 exposures of <1.0 ppm. Effects on neonatal mortality are observed with exposures of 1.0  
3 to 1.5 ppm. Effects on spleen and thymus appear to only occur at high O<sub>3</sub> concentrations  
4 (>1.0 ppm), whereas relevant urban ambient exposures have no effect on systemic immune  
5 function in rats.

### 6 7 ***Genotoxicity Potential of Ozone***

- 8 ● The weight of evidence from new experimental studies, utilizing non-lifetime exposures,  
9 does not appear to support ambient O<sub>3</sub> as a pulmonary carcinogen in laboratory animal  
10 models. New data are in agreement with the 1994 National Toxicology Program  
11 evaluation of O<sub>3</sub> carcinogenicity. However, O<sub>3</sub> could possibly act as a co-carcinogen  
12 functioning to stimulate hyperplasia.

### 13 14 ***Interactions of Ozone with Other Co-occurring Pollutants***

- 15 ● Bases for toxic interactions of O<sub>3</sub> with co-occurring pollutants may include: adsorption  
16 of O<sub>3</sub> onto a co-pollutant with transport to another site; production of toxicologically  
17 active secondary products; biological or chemical alterations at target sites that affect  
18 response to O<sub>3</sub> or the co-pollutant; O<sub>3</sub>- or co-pollutant-induced physiological change, such  
19 as alteration in ventilation pattern, resulting in changes in the penetration or deposition of  
20 one pollutant when another is present.
- 21
- 22 ● Generalizations regarding interactions of O<sub>3</sub> and co-pollutants include: interactions of  
23 O<sub>3</sub>-containing mixtures are generally synergistic; O<sub>3</sub> may produce more significant  
24 biological responses as a component of a mixture than when inhaled alone; and, although  
25 most studies have shown that interaction occurs only at higher than ambient concentrations  
26 with acute exposure, some have demonstrated interactions at more environmentally  
27 relevant levels (e.g., 0.05 to 0.1 ppm O<sub>3</sub> with NO<sub>2</sub>) and with repeated exposures.

### 28 29 ***Effects of Other Photochemical Oxidants***

- 30 ● Ambient concentrations of the most abundant non-O<sub>3</sub> oxidants (peroxyacetyl nitrate,  
31 peroxypropionyl nitrate, and H<sub>2</sub>O<sub>2</sub>) have not been shown as being likely to cause adverse

1 health effects. However, as constituents of ambient air mixes, other ambient oxidants may  
2 contribute to some effects attributed to O<sub>3</sub>.

## 3 4 5 **E.6 CONTROLLED HUMAN EXPOSURE STUDIES**

6 Key findings/conclusions derived from Chapter 6 assessment of experimental human  
7 studies include:

- 8 ● Responses in humans exposed to ambient O<sub>3</sub> concentrations include decreased inspiratory  
9 capacity; mild bronchoconstriction; rapid, shallow breathing pattern during exercise; and  
10 symptoms of cough and pain on deep inspiration. Ozone exposure also results in airway  
11 hyperresponsiveness, inflammation, immune system activation, and epithelial injury.  
12
- 13 ● Young healthy adults exposed to O<sub>3</sub> concentrations of 0.08 ppm develop significant  
14 reversible, transient decrements in pulmonary function if minute ventilation or exposure  
15 duration are increased sufficiently. Healthy children experience similar spirometric  
16 responses but lesser symptoms from O<sub>3</sub> exposure relative to young adults. On average,  
17 spirometric and symptom responses to O<sub>3</sub> exposure appear to decline with increasing age  
18 beyond approximately 18 years of age.  
19
- 20 ● There is a tendency for slightly increased spirometric responses in mild asthmatics and  
21 allergic rhinitics relative to healthy young adults. Spirometric responses in asthmatics appear  
22 to be affected by baseline lung function.  
23
- 24 ● There is a large degree of intersubject variability in physiologic and symptomatic responses  
25 of adults exposed to O<sub>3</sub>. However, responses tend to be reproducible within a given  
26 individual over a period of several months. With increasing O<sub>3</sub> concentration, the  
27 distribution of FEV<sub>1</sub> decrements becomes asymmetrical with a few individuals experiencing  
28 large decrements. An individual's innate susceptibility to ozone may be linked to the genetic  
29 background of an individual. Additional studies, however, are needed to ascertain the link  
30 between susceptibility and polymorphisms.  
31

- Soluble mediators of inflammation such as the cytokines (IL-6, IL-8) and arachidonic acid metabolites (e.g., PGE<sub>2</sub>, PGF<sub>2</sub>", thromboxane, and leukotrienes [LTs] such as LTB<sub>4</sub>) have been measured in the BAL fluid of humans exposed to O<sub>3</sub>. There appears to be no strong correlation between any of the measured cellular and biochemical changes and changes in pulmonary function. A limited number of studies suggest that inflammatory responses may be detected following O<sub>3</sub> exposures that are insufficient to cause decrements in pulmonary function.
- With repeated O<sub>3</sub> exposures over several days, spirometric and symptom responses become attenuated, but this tolerance is lost after about a week without exposure. Some markers of airway inflammation and small airways dysfunction may not be attenuated by repeated O<sub>3</sub> exposures.
- An initial phase of recovery from O<sub>3</sub> exposure in healthy individuals proceeds relatively rapidly, with acute spirometric and symptom responses resolving within about 2 to 4 h. Effects on the small airways, assessed by decrements in FEF<sub>25-75</sub> and altered ventilation distribution at and possibly beyond 24 h, may be partly due to inflammation. Some inflammatory and cellular changes may persist for up to 48 h, but the time course for these parameters in humans has not been explored fully.

## **E.7 EPIDEMIOLOGIC STUDIES**

Many epidemiologic studies, as discussed in Chapter 7, have shown associations of acute exposure to ambient O<sub>3</sub> with a variety of human health endpoints, including pulmonary function, respiratory symptoms, hospital admissions, and mortality. Key findings and conclusions regarding O<sub>3</sub> health effects drawn from the epidemiologic evidence and the issues that may affect the interpretation of the effect estimates can be briefly summarized as follows.

### **E.7.1 Health Effects Associated with Acute Ozone Exposures**

- Field/panel studies of acute O<sub>3</sub> effects. Recent field/panel studies continue to confirm that short-term O<sub>3</sub> exposure is associated with acute decrements in lung function and increased



1 respiratory symptoms, particularly in children and asthmatics. There is also suggestive  
2 evidence that O<sub>3</sub> is related to increased asthma medication use. Taken together with the  
3 evidence from controlled human exposure studies, O<sub>3</sub> is likely causally related to the various  
4 respiratory health outcomes. The current evidence is limited but supportive of a potential  
5 effect of O<sub>3</sub> on heart rate variability, ventricular arrhythmias, and the incidence of  
6 myocardial infarctions.

- 7
- 8 ● Acute O<sub>3</sub> effects on emergency department visits and hospitalizations. Large multicity  
9 studies, as well as many studies from individual cities have reported an association of short-  
10 term O<sub>3</sub> concentrations with respiratory and cardiovascular hospital admissions. Studies  
11 using year-round data noted some inconsistencies in the O<sub>3</sub> effect on daily hospitalizations.  
12 However, studies with data restricted to the summer or warm season, in general, indicated  
13 positive and robust associations between short-term (e.g., 1 h or 8 h) ambient O<sub>3</sub>  
14 concentrations and cardiopulmonary hospital admissions. Results for emergency department  
15 visits are less consistent.
- 16
- 17 ● Acute O<sub>3</sub> effects on mortality. The majority of the studies suggest that an elevated risk of  
18 all-cause mortality is associated with acute exposure to O<sub>3</sub>, especially in the summer or warm  
19 season when O<sub>3</sub> levels are typically high. Slightly greater O<sub>3</sub> effects were observed for  
20 cardiovascular mortality. Results from a recent, large U.S. multicity time-series study  
21 provide the strongest evidence to-date for acute O<sub>3</sub> exposure effects on mortality. Recent  
22 meta-analyses also showed consistent risk estimates that are unlikely to be confounded by  
23 PM; however, future work is needed to better understand the influence of model  
24 specifications on the risk coefficient.
- 25
- 26 ● Age-related differences in O<sub>3</sub> health effects. Supporting evidence exists for heterogeneity in  
27 the effects of O<sub>3</sub> by age. The elderly population (>65 years of age) appear to be at greater  
28 risk of O<sub>3</sub>-related hospitalizations and mortality compared to all age or younger populations.  
29 In addition, potentially adverse respiratory health outcomes were associated with O<sub>3</sub>  
30 exposure in children (<18 years of age).
- 31

- 1 ● Ozone health effects in asthmatics. The effects of O<sub>3</sub> on asthmatics have been examined  
2 widely in both time-series studies and panel studies. Associations of O<sub>3</sub> with various  
3 respiratory health outcomes (including lung function declines, increased respiratory  
4 symptoms, and emergency department visits) were observed. These findings, along with the  
5 pathophysiologic understanding of asthma as a chronic inflammatory disease, indicate that  
6 asthmatics may be a notably susceptible population affected by O<sub>3</sub> exposures.  
7

## 8 **E.7.2 Issues Potentially Affecting Interpretation of Acute Exposure Studies**

- 9 ● Exposure assessment. Exposure misclassification may result from the use of stationary  
10 ambient monitors to determine exposure in population studies. Although central ambient  
11 monitors do not explain the variance of individual personal exposures, significant  
12 correlations are found between aggregate personal O<sub>3</sub> measurements and O<sub>3</sub> concentrations  
13 from ambient monitors. A simulation study indicated that the use of ambient monitor data  
14 will tend to underestimate the O<sub>3</sub> effect. Better understanding of the factors that affect the  
15 relationship between ambient concentrations and personal exposures should help to improve  
16 interpretation of the O<sub>3</sub> effect estimates.  
17
- 18 ● Ozone exposure indices. The three most commonly used daily O<sub>3</sub> exposure indices, 1-h max  
19 O<sub>3</sub>, 8-max O<sub>3</sub>, and 24-h avg O<sub>3</sub>, were found to be highly correlated in studies conducted in  
20 various regions. In addition, effect-size estimates and significance of associations across all  
21 health outcomes were comparable when using standardized distributional increments of  
22 40 ppb, 30 ppb, and 20 ppb for 1-h max O<sub>3</sub>, 8-h max O<sub>3</sub>, and 24-h avg O<sub>3</sub>, respectively.  
23
- 24 ● Lag structures for O<sub>3</sub> exposure and effect. The lag time between O<sub>3</sub> exposure and effect may  
25 differ depending on various factors such as the specific health outcome of interest, the  
26 mechanism of effect, and preexisting health conditions. The majority of the studies found an  
27 immediate O<sub>3</sub> effect, with the strongest associations observed between health outcomes and  
28 O<sub>3</sub> exposure on the same day and/or previous day. Some studies found large cumulative  
29 effects of O<sub>3</sub> over longer lag periods, indicating that multiday lags also may be relevant for  
30 some health outcomes, including mortality.  
31

- 1 ● Sensitivity to model specifications for temporal trends. Ozone effect estimates that were  
2 reported in studies whose main focus was PM often were calculated using the same model  
3 specifications as PM. While the sensitivity of the O<sub>3</sub> risk estimates to alternative model  
4 specifications has not been thoroughly investigated, the limited available evidence indicates  
5 that O<sub>3</sub> effects appear to be robust to various model specifications for temporal trend  
6 adjustment.
- 7
- 8 ● Influence of seasonal factors. An evaluation of the confounding effects of meteorologic  
9 factors and copollutants on O<sub>3</sub> risk estimates is complicated by their changing relationships  
10 with O<sub>3</sub> across seasons. In addition, seasonal or seasonally-modified factors (e.g., air  
11 conditioning use, time spent outdoors) complicate interpretation of all-year effect estimates,  
12 as they affect the relationship between ambient concentrations and personal exposures.  
13 Given the potentially significant influence of season, season-specific analyses are more  
14 informative in assessing O<sub>3</sub> health risks.
- 15
- 16 ● Confounding by copollutants. Multipollutant regression models often are used to adjust for  
17 confounding by copollutants. Although there is some concern regarding the use of  
18 multipollutant models given the varying concavity across pollutants, currently available  
19 results generally suggest that the inclusion of copollutants into the models do not  
20 substantially affect O<sub>3</sub> risk estimates. These findings indicate that effects of O<sub>3</sub> on various  
21 health outcomes are robust and independent of the effects of other copollutants.
- 22
- 23 ● Concentration-response function. In the limited mortality and morbidity studies that have  
24 specifically examined the O<sub>3</sub> concentration-response relationship, the evidence is  
25 inconclusive regarding the detection of any clear effect threshold. Factors such as exposure  
26 measurement error may reduce the ability to detect a threshold in population studies.
- 27
- 28 ● Heterogeneity of O<sub>3</sub> health effects. Consistent O<sub>3</sub> effect estimates have generally been  
29 observed for mortality, hospitalizations, and other respiratory health outcomes in multicity  
30 studies. Some other reported geographic heterogeneity in effect sizes may be attributable to  
31 differences in relative personal exposure to O<sub>3</sub>, which is affected by variations in factors such

1 as air conditioning prevalence and human activity patterns as well as the varying  
2 concentrations and compositions of copollutants present by region.

### 4 **E.7.3 Health Effects Associated with Chronic Ozone Exposure**

5 Many fewer studies have investigated the effects of chronic O<sub>3</sub> exposure on morbidity and  
6 mortality. The strongest evidence is for negative seasonal effects of chronic O<sub>3</sub> exposure on lung  
7 function in adults and children. Less conclusive are longer-term studies investigating the  
8 association of chronic O<sub>3</sub> exposure on yearly lung function, asthma incidence, and respiratory  
9 symptoms. Studies of potential chronic O<sub>3</sub> exposure-mortality relationships have generally  
10 observed inconsistencies across exposure periods, cause-specific mortality outcomes,  
11 and gender.

## 14 **E.8 INTEGRATIVE SYNTHESIS**

15 This section summarizes key conclusions derived from the Chapter 8 integrated synthesis  
16 of information regarding health effects associated with ambient O<sub>3</sub> exposures. The conclusions  
17 were derived based on an integrated analysis of available laboratory animal, human clinical, and  
18 epidemiological studies that have evaluated health effects associated with short-term, repeated,  
19 and long-term exposures to O<sub>3</sub> alone or in combination with other ambient pollutants. The  
20 Chapter 8 synthesis utilized experimental evidence from dosimetric and human and animal  
21 toxicological studies presented in Chapters 4, 5, and 6 both to evaluate the biological plausibility  
22 of health effects observed in epidemiologic studies discussed in Chapter 7 and to inform  
23 delineation of O<sub>3</sub> exposure-dose-response relations and likely underlying mechanisms of action.  
24 These evaluations are also aimed at identifying susceptible populations that are at potentially  
25 greater risk for effects of O<sub>3</sub> exposure.

### 27 ***1. Health effects of acute (short-term) exposures to Ozone***

28 Numerous field panel and time-series epidemiologic studies (using better weather models  
29 and adjustments to confounding copollutants than those assessed in the 1996 O<sub>3</sub> AQCD ) have  
30 evaluated the effects of short-term exposure to O<sub>3</sub> on a wide range of health endpoints, from lung

1 function decrements to mortality. Results from the majority of studies continue to support the  
2 conclusions reported in the 1996 O<sub>3</sub> AQCD.

- 3 ● Panel studies typically have evaluated the effects of short-term O<sub>3</sub> exposure on both healthy  
4 individuals and people with cardiopulmonary diseases. These evaluations included  
5 measurement of lung function changes, respiratory symptoms and use of asthma medication.  
6
- 7 ● Clinical controlled exposure studies in humans indicate changes in lung function and  
8 respiratory symptoms that vary as a function of exposure concentration, duration and level  
9 of exercise.  
10
- 11 ● Newer meta-analyses confirmed the interindividual differences in lung function decrements  
12 reported in the 1996 O<sub>3</sub> AQCD. Age-specific differences in the lung function responses were  
13 also observed. Spirometric responses (due to decrements in lung function) in healthy adults  
14 exposed to near ambient O<sub>3</sub> levels typically resolve to near baseline values within 4-6 h.  
15
- 16 ● Meta-analyses of four controlled human exposure studies (two new and two reported in the  
17 1996 O<sub>3</sub> AQCD) reporting the effects of prolonged (6.6 h) exposures to 0.08 ppm O<sub>3</sub> during  
18 moderate exercise on pulmonary function in young healthy adults (M = 90, F = 30; mean  
19 age, 23 yrs) indicate an absolute FEV<sub>1</sub> decrease of approximately 6%, whereas FEV<sub>1</sub>  
20 increased by about 1% following free air (FA) exposures.  
21
- 22 ● Recent meta-analyses on numerous clinical studies indicate interindividual differences in  
23 lung inflammatory response to short-term O<sub>3</sub> exposures.  
24
- 25 ● Inflammatory and permeability responses also generally resolve (in some instances complete  
26 recovery) within a week or two after cessation of O<sub>3</sub> exposure, but exhibit differential  
27 attenuation profiles between normal healthy subjects and people with preexisting respiratory  
28 diseases. However, some lung inflammation markers may not completely resolve readily,  
29 and mild persistent inflammation has been reported.  
30

- 1 ● Field/panel studies of healthy individuals and asthmatics have found positive associations  
2 between short-term exposure to O<sub>3</sub> and decrements in lung function analogous to those  
3 shown by studies of controlled short-term (1 to 8 h) human exposures to O<sub>3</sub>.  
4
- 5 ● Associations between short-term O<sub>3</sub> exposures and school absenteeism (due to respiratory  
6 illness) have also been suggested.  
7
- 8 ● With regard to cardiac impacts, a limited number of field studies that examined the  
9 relationship between short-term O<sub>3</sub> exposures and cardiovascular effects (heart rate  
10 variability, myocardial infarction) suggest an association.  
11
- 12 ● A large multicity and several single-city studies have indicated a positive association  
13 between increased O<sub>3</sub> levels (especially during the warm season) and increased risk for  
14 hospital admissions. On the other hand epidemiologic data on emergency department visits  
15 do not suggest such an association with increase in ambient O<sub>3</sub> levels.  
16
- 17 ● The results of two large multicity studies from the U.S. and several single-city studies  
18 suggest a positive association between increases in O<sub>3</sub> levels and all-cause (non-accidental)  
19 daily mortality. Meta-analyses on the influence of season suggest a causal association.  
20 Additional meta-analyses on cause-specific mortality are suggestive of a likely positive  
21 association between increases in ambient O<sub>3</sub> levels and cardiovascular mortality.  
22
- 23 ● Short-term O<sub>3</sub>-induced lung function decrements, respiratory symptoms, inflammation and  
24 permeability changes observed in animal toxicology studies are consistent with human  
25 studies.  
26

## 27 ***2. Health effects of repeated short-term exposures to Ozone***

28 The results of new controlled human exposure studies of repeated short-term O<sub>3</sub> exposures  
29 continue to support the health effects findings/conclusions reported in the 1996 O<sub>3</sub> AQCD.

- 30 ● Repeated exposure studies at higher concentrations typically show that FEV<sub>1</sub> response to O<sub>3</sub>  
31 is enhanced on the second of several days of exposure. Such an enhanced response was not

1 observed at lower O<sub>3</sub> concentrations. With repeated O<sub>3</sub> exposures over several days,  
2 spirometric and symptom responses become attenuated, but this tolerance is lost after about a  
3 week without exposure.

- 4  
5 ● In humans repeatedly exposed to 0.4 ppm O<sub>3</sub> for 5 consecutive days, several indicators of  
6 inflammation (e.g., PMN influx, IL-6, PGE<sub>2</sub>, BAL protein, fibronectin) were attenuated after  
7 5 days of exposure. However, lung injury and permeability markers (LDH, IL-8, total  
8 protein, epithelial cells) did not show attenuation, indicating that tissue damage probably  
9 continues to occur during repeated exposure. The recovery of the inflammatory response  
10 occurred for some markers after 10 days, but some responses were not normalized even after  
11 20 days. The continued presence of cellular injury markers indicates a persistent effect that  
12 may not necessarily be recognized due to the attenuation of spirometric and symptom  
13 responses.
- 14  
15 ● Repeated daily exposure to lower concentrations of O<sub>3</sub> (0.125 ppm for 4 days) causes an  
16 increased response to bronchial allergen challenge in subjects with preexisting allergic  
17 airway disease, with or without asthma. In these subjects, changes in airway responsiveness  
18 after O<sub>3</sub> exposure appear to be resolved more slowly than changes in FEV<sub>1</sub> or respiratory  
19 symptoms.

### 21 ***3. Health effects of long-term exposures to Ozone***

22 Assessment of human health effects associated with long-term O<sub>3</sub> exposures is hampered  
23 by the lack of pertinent data from human clinical and epidemiologic studies. Chronic animal  
24 toxicology studies continue to support structural alterations in several regions of the respiratory  
25 tract and identify the centriacinar region of the lung as the most affected region.

- 26 ● □ Animal toxicology studies that utilized exposure regimens to simulate seasonal exposure  
27 pattern also report increased lung injury compared to conventional chronic stable exposures.  
28 One long-term study of infant rhesus monkeys exposed to simulated seasonal O<sub>3</sub> patterns  
29 (0.5 ppm 8h/day for 5 days, every 14 days for 11 episodes) demonstrated: (1) remodeling in  
30 the distal airways; (2) abnormalities in tracheal basement membrane; (3) eosinophil  
31 accumulation in conducting airways; (4) decrements in airway innervation. These findings

1 advance earlier information regarding possible injury-repair processes occurring with  
2 seasonal O<sub>3</sub> exposures.

- 3
- 4 ● Effects of O<sub>3</sub> on the upper respiratory tract of F344 rats exposed to O<sub>3</sub> (0.12, 0.5, or 1.0 ppm  
5 for 20 months) included marked mucous cell metaplasia in the rats exposed to 0.5 and  
6 1.0 pm O<sub>3</sub>, but not at 0.12 ppm O<sub>3</sub>. The persistent nature of the O<sub>3</sub>-induced mucous cell  
7 metaplasia suggests that O<sub>3</sub> exposure may have the potential to induce similar long-lasting  
8 alterations in the airways of humans. Hyperplasia in the nasal epithelium of rats exposed to  
9 0.25 and 0.5 ppm, 8h/day, 7 days/week, for 13 weeks has been reported.
- 10
- 11 ● Pathophysiological changes associated with chronic O<sub>3</sub> exposures observed in animal studies  
12 suggest possible similar alterations in humans. The pulmonary function changes observed in  
13 children in polluted metropolitan areas and lung structural alterations reported in an autopsy  
14 study in Los Angeles suggest a role for long-term ambient O<sub>3</sub> exposure, but such possible  
15 effects need to be further evaluated with improved study design(s).
- 16

#### 17 **4. *Susceptibility factors associated with exposure to ozone***

18 Various factors such as age, gender, nutrition, socioeconomic, activity patterns, and disease  
19 status have been shown to influence the response to environmental air pollutants. Controlled  
20 human exposure studies clearly established differential biological response to O<sub>3</sub> based on  
21 physical activity (exertion) and age. These studies also demonstrated a large variation in  
22 sensitivity and responsiveness to O<sub>3</sub>. The specific factors that contribute to this intersubject  
23 variability are yet to be identified.

- 24 ● □ Increased hospital admissions for asthma and COPD in summer (with increased levels of  
25 ambient O<sub>3</sub>) suggest that people with these respiratory diseases as potential sub-population  
26 for O<sub>3</sub>-induced health effects.
- 27
- 28 ● Similarly, based on O<sub>3</sub>-induced differential responses in lung inflammation and in airway  
29 hyperresponsiveness, asthmatics (including children) appear to have potentially increased  
30 susceptibility to O<sub>3</sub>. However, there is no supportive data from controlled human studies  
31 suggesting individuals with COPD are more sensitive to O<sub>3</sub>-induced health effects.



- 1 ● Animal toxicology studies provided supportive evidence to the observations of varied  
2 susceptibility. Various strains of mice and rats have demonstrated the importance of genetic  
3 background in O<sub>3</sub> susceptibility. Moreover, genetic and molecular characterization studies in  
4 laboratory animals identified genetic loci responsible for both sensitivity and resistance.  
5
- 6 ● Consistent with the 1996 O<sub>3</sub> AQCD, the scarcity of data prevents determination of the role of  
7 ethnic or racial background and nutrition status on O<sub>3</sub>-induced health effects. However,  
8 as presented in this document, exercising (moderate to high physical exertion) healthy  
9 adolescents and asthmatics appear to demonstrate increased responsiveness to ambient  
10 concentrations of O<sub>3</sub> and may be susceptible for O<sub>3</sub>-induced health effects.  
11

##### 12 ***5. Health effects of binary pollutant mixtures containing ozone***

13 A limited number of controlled human exposure studies and a few animal toxicology  
14 studies of binary mixtures containing O<sub>3</sub> suggest potential interactions, depending on specific  
15 exposure regimens and copollutant constituents.

- 16 ● Continuous exposure to SO<sub>2</sub> and NO<sub>2</sub> increased inhaled bolus O<sub>3</sub> absorption, while  
17 continuous exposure to O<sub>3</sub> decreased O<sub>3</sub> bolus absorption. Asthmatics exhibited enhanced  
18 airway reactivity to house dust mite following exposures to O<sub>3</sub>, NO<sub>2</sub>, and the combination of  
19 the two gases. Spirometric response, however, was impaired only by O<sub>3</sub> and O<sub>3</sub>+NO<sub>2</sub> at  
20 higher concentrations.  
21
- 22 ● Animal toxicology studies with O<sub>3</sub> in mixture with NO<sub>2</sub>, formaldehyde, and PM  
23 demonstrated additive, synergistic or antagonistic effects, depending on the exposure  
24 regimen and the endpoints evaluated.  
25
- 26 ● One controlled exposure study of children, designed to approximate exposure conditions of  
27 an epidemiologic study by matching the population and exposure atmosphere (0.1 ppm O<sub>3</sub>,  
28 0.1 ppm SO<sub>2</sub> and 101 µg<sup>m<sup>3</sup></sup> H<sub>2</sub>SO<sub>4</sub>), failed to support the findings of the epidemiologic study.  
29 This study points out difficulties in trying to link the outcomes of epidemiologic and  
30 controlled exposure studies by use of binary pollutant mixtures.  
31

## 1 **E.9 VEGETATION AND ECOLOGICAL EFFECTS**

### 2 **General**

- 3 ● Data published since 1996 continue to support the conclusions of previous O<sub>3</sub> AQCDs that  
4 there is strong evidence that ambient O<sub>3</sub> concentrations cause foliar injury along with growth  
5 and yield damage to numerous common and economically valuable plant and tree species.  
6
- 7 ● Research to date has continued its focus at the species level, with very few new studies at the  
8 ecosystem level. The lack of quantification of biotic and abiotic factors impinging on the  
9 individual to population organizational levels results in a limited ability to scale O<sub>3</sub> responses  
10 to the ecosystem level. Therefore, a high degree of uncertainty remains in our ability to  
11 assess ozone risk to ecological resources and the services they provide.  
12

### 13 **Methodologies**

- 14 ● Since the 1996 AQCD free-air exposure (FACE) systems have come into more frequent use.  
15 FACE systems eliminate many of the concerns raised about closed or open-top chamber  
16 (OTC) experiments including small plot size, altered microclimate within OTCs, and the  
17 effect of charcoal filtering on overall air quality within OTCs. One of the advantages of the  
18 application of plume systems to O<sub>3</sub> research is the ability to compare response of plants in  
19 open-field systems with results from OTCs. In particular, studies with quaking aspen  
20 (*Populus tremuloides* L.) performed in OTCs, FACE, and also at sites along an ambient O<sub>3</sub>  
21 gradient showed that O<sub>3</sub> symptom expression was generally similar, supporting the  
22 previously observed level of variation among aspen clones in OTC studies.  
23
- 24 ● The lack of rural monitors continues to be a major problem in the characterization of O<sub>3</sub>  
25 exposures in remote areas, as well as in linking effects to exposure in natural ecosystems.  
26 Since the 1996 O<sub>3</sub> AQCD, the use of passive samplers has expanded monitoring efforts to  
27 include remote areas that were previously uncharacterized.  
28
- 29 ● Advancements in biomonitoring have been made since the 1996 O<sub>3</sub> AQCD, primarily in the  
30 area of identification and symptom verification of sensitive species . The U.S. Department  
31 of Agriculture (USDA) Forest Service continues its program to monitor O<sub>3</sub> effects in forested

1 ecosystems throughout the United States. Although results are not useful for developing  
2 exposure-response relationships or for quantifying responses to O<sub>3</sub>, they can provide an  
3 annual assessment and correlative information regarding the extent of O<sub>3</sub> injury occurring  
4 across many regions of the United States.

#### 6 **Mode of Action**

- 7 ● The new information available on the mode of action of O<sub>3</sub> is, in part, a result of improved  
8 molecular tools for following rapid changes that occur within the leaf. Many changes occur  
9 within hours or possibly days following O<sub>3</sub> exposure. Other O<sub>3</sub> effects take longer to occur  
10 and tend to be most obvious only under exposure to low O<sub>3</sub> concentrations for long periods.  
11 These low-exposure chronic effects have been linked to the senescence process or some  
12 physiological response very closely linked to senescence (e.g., translocation, reabsorption,  
13 allocation of nutrients and carbon).

#### 15 **Modification of Growth Response**

- 16 ● It has been known for decades that several factors, both biotic and abiotic, alter plant  
17 response to O<sub>3</sub>. However, only a few studies reported since the 1996 O<sub>3</sub> AQCD have  
18 improved our understanding of the role of these interactions in modifying plant O<sub>3</sub> response.  
19
- 20 ● Recent studies have supported the earlier conclusion that O<sub>3</sub> often increases the likelihood  
21 and success of insect attacks, but only with respect to chewing insects. Although it seems  
22 likely that some insect problems could increase as a result of greater O<sub>3</sub> levels, we are still far  
23 from being able to predict the nature of any particular O<sub>3</sub>-plant-insect interaction, its  
24 likelihood, or its severity.
- 25
- 26 ● O<sub>3</sub> exposure generally increases plant diseases associated with facultative necrotrophic plant  
27 pathogens. Generally, pathogens that benefit from damage to cells are enhanced by O<sub>3</sub> stress  
28 of their hosts, whereas pathogens and pests that require healthy hosts are depressed by O<sub>3</sub>  
29 stress.

1 **Exposure Indices**

- 2 ● Exposure indices are metrics that relate measured plant damage (i.e., reduced growth) to  
3 monitored ambient O<sub>3</sub> concentrations over time to provide a consistent metric for reviewing  
4 and comparing exposure-response effects obtained from various studies. Since the 1996 O<sub>3</sub>  
5 AQCD, there has been no direct experimental testing of the adequacy of exposure indices  
6 proposed in 1996; therefore, there is no new information to alter the basic conclusions put  
7 forth in the 1996 O<sub>3</sub> AQCD.  
8
- 9 ● The proposed indices in the 1996 O<sub>3</sub> AQCD (i.e., SUM06, W126, AOT40) included various  
10 functional and statistical summaries of monitored hourly O<sub>3</sub> concentrations over designated  
11 time periods. The few studies that have been published since the 1996 O<sub>3</sub> AQCD continue to  
12 support the earlier conclusions, including the importance of peak concentrations, and the  
13 duration and occurrence of O<sub>3</sub> exposures in altering plant growth and yield.  
14
- 15 ● A large body of new research, mostly out of Europe, addresses the need for an index related  
16 to the actual flux of O<sub>3</sub> into the plant. Despite additional research linking estimates of flux  
17 with plant response since 1996, information is still insufficient to identify a flux-based model  
18 that incorporates the necessary complexity across space and time to be non-site or non-  
19 species specific. Based on the current state of knowledge, exposure indices that cumulate  
20 and differentially weight the higher hourly average concentrations, but include the mid-level  
21 values (e.g., SUM06, W126, AOT40), still represent the best approach for relating vegetation  
22 effects to O<sub>3</sub> exposure in the United States.  
23

24 **Ozone Exposure-Plant Response Relationships**

- 25 ● Data published since 1996 continue to support the conclusions of previous O<sub>3</sub> AQCDs that  
26 there is strong evidence that ambient O<sub>3</sub> concentrations cause foliar injury and growth and  
27 yield damage to numerous common and economically valuable plant and tree species.  
28
- 29 ● In addition to reductions in crop yield, O<sub>3</sub> may also reduce the quality or nutritive value of  
30 annual species. Many recent studies have found O<sub>3</sub> effects on various measures of plant  
31 organs that affect quality, with most of those studies focusing on characteristics important for

1 food or fodder. These studies indicate that ambient O<sub>3</sub> may have economically important  
2 effects on the quality of crop and forage species.

- 3
- 4 ● Results since 1996 support the conclusion of the 1996 O<sub>3</sub> AQCD that deciduous trees are  
5 generally less O<sub>3</sub> sensitive than are most annual plants, with the exception of a few very  
6 sensitive genera such as *Populus* and sensitive species such as black cherry. Various  
7 evergreen tree species and genotypes have widely varying O<sub>3</sub> sensitivities. Based on OTC  
8 studies with seedlings, major evergreen species in the United States are generally less  
9 sensitive than are most deciduous trees, and slower-growing evergreen species are less  
10 sensitive than are faster-growing species. For all types of perennial vegetation, cumulative  
11 effects over more than one growing season may be important; studies for only a single  
12 season may underestimate effects.

13

#### 14 **Ecosystem Effects**

- 15 ● There is evidence that tropospheric O<sub>3</sub> is an important stressor of ecosystems, with  
16 documented impacts on the biotic condition, ecological processes, and chemical/physical  
17 nature of natural ecosystems. Effects on individual keystone species and their associated  
18 microflora and fauna, which have been shown experimentally, may cascade through the  
19 ecosystem to the landscape level, although this has not yet been demonstrated.
- 20
- 21 ● Systematic injury surveys (e.g., USDA Forest Service's ozone bioindicator plot network and  
22 Europe's ICP Forests) demonstrate that foliar injury occurs on O<sub>3</sub> sensitive species in many  
23 regions of the United States and Europe. Frequent lack of correspondence between foliar  
24 symptoms and growth effects means that other methods must be used to estimate the regional  
25 effects of O<sub>3</sub> on tree growth. Investigations of the radial growth of mature trees, combined  
26 with data from many controlled studies with seedlings and a few studies with mature trees  
27 suggest that ambient O<sub>3</sub> is reducing the growth of mature trees in some U.S. locations.
- 28
- 29 ● The study of genetic aspects of O<sub>3</sub> impacts on natural ecosystems has been largely based on  
30 correlations, and it remains to be shown more conclusively whether O<sub>3</sub> affects biodiversity or  
31 genetic diversity.

1 **Economics**

- 2 ● The physical and economic effects on agriculture are well documented and provide useful  
3 information for the consideration of establishing air quality standards for crops. Effects on  
4 forests and natural ecosystems remain problematic, due to limitations in biological response  
5 data and economic methods. The problem is even more acute for valuing natural ecosystem  
6 goods and services.

7  
8  
9 **E.10 TROPOSPHERIC OZONE EFFECTS ON UV-B FLUX AND ITS**  
10 **ROLE IN CLIMATE CHANGE**

11 The molecular properties specific to O<sub>3</sub> include a capacity for absorbing incoming  
12 ultraviolet (UV) and infrared (IR) radiation, and both incoming solar and outgoing terrestrial IR  
13 radiation. Consequently, O<sub>3</sub> plays an essential role in shielding the earth's surface from harmful  
14 levels of UV-B radiation, by way of the stratospheric O<sub>3</sub> layer. Its effectiveness as a screen for  
15 the residual UV-B flux that penetrates the stratosphere and passes into the troposphere and its  
16 role in reducing UV-induced human health effects are addressed in Chapter 10. The radiation-  
17 absorbing properties of O<sub>3</sub> also make it a greenhouse gas (GHG) having global and, more  
18 importantly, regional consequence for climate, as also addressed in Chapter 10. Important  
19 conclusions from Chapter 10 are summarized below.

- 20  
21 ● The distribution of O<sub>3</sub> within the atmosphere. Ozone is distributed very unevenly within the  
22 atmosphere, with ~90% of the total atmospheric burden present in the stratosphere. The  
23 remaining ~10% is distributed within the troposphere, with higher relative concentrations  
24 near the source of its precursors at the surface. Concentrations of O<sub>3</sub> at the mid- and upper-  
25 troposphere vary, depending upon meteorological conditions.
- 26  
27 ● Multiple factors govern the flux of UV-B radiation at the Earth's surface. Latitude and  
28 altitude are the two most important factors that define the residual UV-B flux at the surface.  
29 Natural variation in the total column density of stratospheric O<sub>3</sub> is also an important factor.  
30 All of these factors are followed in importance by tropospheric clouds, particulate matter

1 (PM) and O<sub>3</sub>. The effect of natural stratospheric variation, clouds, PM and tropospheric O<sub>3</sub>  
2 on UV fluxes within the troposphere and at the surface are each very difficult to predict.

- 3
- 4 ● A UV-B “climatology” is needed to predict human exposure levels. A UV-B climatology,  
5 representing patterns and trends in UV-B flux at the Earth’s surface, must be based on  
6 extended in situ observations in order to adequately capture natural variability and the effects  
7 of human activities on atmospheric UV-B absorbers. At present, the body of UV-B  
8 measurements cannot support the development of a climatology.
- 9
- 10 ● Human exposure to UV-B radiation. Quantitative evaluation of human exposure to UV-B  
11 radiation is necessary to perform health risk assessment of UV-B-related health effects.  
12 Individuals who participate in outdoor sports and activities, work outdoors, live in  
13 geographic areas with higher solar flux, and/or engage in high-risk behavior (e.g., extended  
14 sun bathing) can reasonably be projected to be at increased risk for higher UV radiation  
15 exposures. However, little is known about the impact of variability in these factors on  
16 individual exposure to UV radiation.
- 17
- 18 ● Human health effects of UV-B radiation. Exposure to UV-B radiation is associated with  
19 increased risk of erythema, nonmelanoma and melanoma skin cancers, ocular damage, and  
20 immune system suppression. Some studies have attempted to estimate the potential effects  
21 of changes in surface-level UV flux resulting from stratospheric O<sub>3</sub> depletion on these health  
22 outcomes; however, the numerous simplifying assumptions made in the assessments limit the  
23 usefulness of the risk estimates. The effect of changes in surface-level O<sub>3</sub> concentrations on  
24 UV-induced health outcomes cannot yet be critically assessed within reasonable uncertainty.
- 25
- 26 ● Vitamin D-related health benefits of UV-B radiation. A potential health benefit of increased  
27 UV-B exposure relates to the production of vitamin D in humans. Several studies have  
28 found that UV-B radiation, by increasing vitamin D production, is associated with reduced  
29 risks of various cancers. However, as with other impacts of UV-B on human health, this  
30 beneficial effect of UV-B has not been studied in sufficient detail to allow for a credible  
31 health benefits assessment.

- 1 ● Ozone is a potent GHG. Ozone traps incoming solar radiation at both ends of the spectrum,  
2 as well as shortwave radiation that is scattered from high-albedo portions of the Earth's  
3 surface. Outgoing terrestrial IR is absorbed by O<sub>3</sub> within the range where water vapor does  
4 not absorb, so that natural variability in humidity does not alter its radiative impact. These  
5 effects directly force climate. By participating in the oxidative chemistry of the atmosphere,  
6 O<sub>3</sub> can indirectly and negatively force climate by the removal of other greenhouse gases.
- 7
- 8 ● Multiple factors influence the forcing effect of tropospheric O<sub>3</sub>. Estimates of present-day  
9 forcing by O<sub>3</sub> depend upon the available information on pre-industrial and current  
10 concentrations. Both are limited and, therefore, very uncertain. Other factors, including the  
11 albedo of underlying surface, altitude and co-occurrence of PM can also complicate the  
12 calculation of globally-averaged forcing.
- 13
- 14 ● Globally-averaged direct forcing by O<sub>3</sub>. On the basis of the best available information, a  
15 2001 Intergovernmental Panel on Climate Change (IPCC) report offered an estimated value  
16 of  $0.35 \pm 0.15 \text{ Wm}^{-2}$  for the annual, globally-averaged direct forcing by tropospheric O<sub>3</sub>.  
17 Another recent estimate places this value at  $0.5 \pm 0.2 \text{ Wm}^{-2}$ .
- 18
- 19 ● Projections of forcing by O<sub>3</sub> into the future. A CTM-climate modeling intercomparison  
20 study carried out as part of the third assessment by the IPCC yielded an estimated 0.4 to 0.78  
21  $\text{Wm}^{-2}$  forcing by O<sub>3</sub> by the year 2100. The authors of this study concluded that O<sub>3</sub> can be  
22 expected to be an important contributor to climate forcing into the future.
- 23
- 24 ● Climate forcing by O<sub>3</sub> at the regional scale may be its most important impact on climate.  
25 Satellites have detected high O<sub>3</sub> concentrations localized at the regional scale that are  
26 associated with large urban centers and extensive biomass burning. Climate forcing by these  
27 high, regional-scale O<sub>3</sub> concentrations have been estimated to be on the order of  $1 \text{ Wm}^{-2}$  (a  
28 substantial fraction of the direct, globally-averaged forcing due to well-mixed GHGs,  
29 including CO<sub>2</sub>). The impact of climate forcing at this level depends upon the particular  
30 characteristics of the region in which it occurs. At present, regional-scale modeling studies



1 are not available that provide estimates of these effects. Research efforts to do so are  
2 underway.

### 5 **E.11 MATERIALS DAMAGE**

6 The Chapter 11 discussion of O<sub>3</sub> effects on man-made materials mainly summarizes key  
7 information from the 1996 O<sub>3</sub> AQCD, given that little new pertinent research information on O<sub>3</sub>-  
8 related materials damage has been published since then. Key points include:

- 9  
10 ● Ozone and other photochemical oxidants react with many economically important man-made  
11 materials, decreasing their useful life and aesthetic appearance. Materials damaged by O<sub>3</sub>  
12 include elastomers; textiles and fibers; dyes, pigments, and inks; and paints and other surface  
13 coatings.
- 14  
15 ● Elastomeric compounds (natural rubber and synthetic polymers and copolymers of  
16 butadiene, isoprene, and styrene) are highly susceptible, even to low O<sub>3</sub> concentrations.  
17 Ozone damages these compounds by breaking the molecular chain at the carbon-carbon  
18 double bond and by adding a chain of three oxygen atoms directly across the double bond.  
19 This structure change promotes characteristic cracking of stressed/stretched rubber called  
20 “weathering.” Tensile strain produces cracks on the surface of the rubber that increase in  
21 size and number with increased stress/stretching. The rate of crack growth is dependent on  
22 degree of stress, type of rubber compound, O<sub>3</sub> concentration, duration of exposure, O<sub>3</sub>  
23 velocity, and temperature. After initial cracking, further O<sub>3</sub> penetration results in additional  
24 cracking and, eventually, mechanical weakening.
- 25  
26 ● Ozone can damage textiles and fabrics by mechanisms similar to those associated with  
27 elastomers. Generally, synthetic fibers are less affected by O<sub>3</sub> than natural fibers. Overall,  
28 O<sub>3</sub> contribution to degradation of textiles and fabrics is not considered significant.
- 29

- 1     ● Ozone fading of textile dyes is a diffusion-controlled process, with the rate of fading being  
2       controlled by diffusion of the dye to the fiber surface. Many textile dyes react with O<sub>3</sub>. The  
3       rate and severity of the O<sub>3</sub> attack is influenced by the chemical nature of the textile fiber and  
4       the manner in which the dye has been applied.  
5
- 6     ● Paints applied to exterior surfaces of buildings and other structures (e.g., bridges), as well as  
7       several artists' pigments, are also sensitive to fading and oxidation by O<sub>3</sub> at concentrations  
8       found in urban areas.  
9

# 1. INTRODUCTION

This is an update revision of the document, “*Air Quality Criteria for Ozone and Related Photochemical Oxidants*,” published by the U.S. Environmental Protection Agency (EPA) in 1996 (U.S. Environmental Protection Agency, 1996). That 1996 Ozone Air Quality Criteria Document (O<sub>3</sub> AQCD) provided scientific bases for Congressionally-mandated periodic review by the EPA of the National Ambient Air Quality Standards for Ozone (O<sub>3</sub> NAAQS), which culminated in promulgation of new O<sub>3</sub> NAAQS by EPA in 1997.

The present document critically assesses the latest scientific information relative to determining the health and welfare effects associated with the presence of various concentrations of O<sub>3</sub> and related oxidants in ambient air. It builds upon the previous 1996 EPA O<sub>3</sub> AQCD, by focusing on evaluation and integration of information relevant to O<sub>3</sub> NAAQS criteria development that has become available since that covered by the 1996 criteria review; and it will provide scientific bases for the current periodic review of the O<sub>3</sub> NAAQS.

This introductory chapter of the revised O<sub>3</sub> AQCD presents: (a) background information on legislative requirements, the criteria and NAAQS review process, and the history of O<sub>3</sub> NAAQS reviews (including a chronology of changes in key elements of the O<sub>3</sub> standards); (b) an overview of the current O<sub>3</sub> criteria review process and projected schedule (including approaches and procedures used to prepare this document, as well as projected key milestones); and (c) an orientation to the general organizational structure and content of the document.

## 1.1 LEGAL AND HISTORICAL BACKGROUND

### 1.1.1 Legislative Requirements

Two sections of the Clean Air Act (CAA) govern the establishment, review, and revision of National Ambient Air Quality Standards (NAAQS). Section 108 (42 U.S.C. 7408) directs the Administrator of the U.S. Environmental Protection Agency (EPA) to identify ambient air pollutants that may be reasonably anticipated to endanger public health or welfare and to issue air quality criteria for them. These air quality criteria are to reflect the latest scientific

1 information useful in indicating the kind and extent of all identifiable effects on public health or  
2 welfare that may be expected from the presence of a given pollutant in ambient air.

3 Section 109(a) of the CAA (42 U.S.C. 7409) directs the Administrator of EPA to propose  
4 and promulgate primary and secondary NAAQS for pollutants identified under Section 108.  
5 Section 109(b)(1) defines a primary standard as one that, in the judgment of the Administrator, is  
6 requisite to protect the public health (see inset below) based on the criteria and allowing for an  
7 adequate margin of safety. The secondary standard, as defined in Section 109(b)(2), must  
8 specify a level of air quality that, in the judgment of the Administrator, is requisite to protect the  
9 public welfare (see inset below) from any known or anticipated adverse effects associated with  
10 the presence of the pollutant in ambient air, based on the criteria.

11  
12

<b>PUBLIC HEALTH EFFECTS</b>
■□ Effects on the health of the general population, or identifiable groups within the population, who are exposed to pollutants in ambient air
■□ Effects on mortality
■□ Effects on morbidity
■□ Effects on other health conditions including indicators of: <ul style="list-style-type: none"><li>• pre-morbid processes,</li><li>• risk factors, and</li><li>• disease</li></ul>

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<b>PUBLIC WELFARE EFFECTS</b>
■□ Effects on personal comfort and well-being
■□ Effects on economic values
■□ Deterioration of property
■□ Hazards to transportation
■□ Effects on the environment, including: <ul style="list-style-type: none"><li>• animals</li><li>• climate</li><li>• crops</li><li>• materials</li><li>• soils</li><li>• vegetation</li><li>• visibility</li><li>• water</li><li>• weather</li><li>• wildlife</li></ul>

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22 Section 109(d) of the CAA (42 U.S.C. 7409) requires periodic review and, if appropriate,  
23 revision of existing criteria and standards. If, in the Administrator's judgment, the Agency's  
24 review and revision of criteria make appropriate the proposal of new or revised standards, such  
25 standards are to be revised and promulgated in accordance with Section 109(b). Alternatively,  
26 the Administrator may find that revision of the standards is inappropriate and conclude the  
27 review by leaving the existing standards unchanged. Section 109(d)(2) of the 1977 CAA  
28 Amendments also requires that an independent scientific review committee be established to  
29 advise the EPA Administrator on NAAQS matters, including the scientific soundness of criteria  
30 (scientific bases) supporting NAAQS decisions. This role is fulfilled by the Clean Air Scientific  
31 Advisory Committee (CASAC) of EPA's Science Advisory Board (SAB).

## 1.1.2 Criteria and NAAQS Review Process

Periodic reviews by EPA of criteria and NAAQS for a given criteria air pollutant progress through a number of steps, beginning with preparation by EPA's National Center for Environmental Assessment Division in Research Triangle Park, NC (NCEA-RTP) of an air quality criteria document (AQCD). The AQCD provides a critical assessment of the latest available scientific information upon which the NAAQS are to be based. Drawing upon the AQCD, staff of EPA's Office of Air Quality Planning and Standards (OAQPS) prepare a Staff Paper that evaluates policy implications of the key studies and scientific information contained in the AQCD and presents EPA staff conclusions and recommendations for standard-setting options for the EPA Administrator to consider. The Staff Paper is intended to help "bridge the gap" between the scientific assessment contained in the AQCD and the judgments required of the Administrator in determining whether it is appropriate to retain or to revise the NAAQS. Iterative drafts of both the AQCD and the Staff Paper (as well as other analyses, such as exposure and/or risk assessments, supporting the Staff Paper) are made available for public comment and CASAC review. The final versions of the AQCD and Staff Paper incorporate changes made in response to CASAC and public review. Based on the information in these documents, the Administrator proposes decisions on whether to retain or revise the NAAQS, taking into account public comments and CASAC advice and recommendations. The Administrator's proposed decisions are published in the *Federal Register*, with a preamble that presents the rationale for the decisions and solicits public comment. The Administrator makes a final decision after considering comments received on the proposed decisions. The Administrator's final decisions are promulgated in a *Federal Register* notice that addresses significant comments received on the proposal.

NAAQS decisions involve consideration of the four basic elements of a standard: *indicator, averaging time, form, and level*. The indicator defines the pollutant to be measured in the ambient air for the purpose of determining compliance with the standard. The averaging time defines the time period over which air quality measurements are to be obtained and averaged, considering evidence of effects associated with various time periods of exposure. The form of a standard defines the air quality statistic that is to be compared to the level of the standard (i.e., an ambient concentration of the indicator pollutant) in determining whether an area attains the standard. The form of the standard specifies the air quality measurements that

1 are to be used for compliance purposes (e.g., the 98th percentile of an annual distribution of  
2 daily concentrations; the annual arithmetic average), the monitors from which the measurements  
3 are to be obtained (e.g., one or more population-oriented monitors in an area), and whether the  
4 statistic is to be averaged across multiple years. These basic elements of a standard are the  
5 primary focus of the staff conclusions and recommendations in the Staff Paper and in the  
6 subsequent rulemaking, building upon the policy-relevant scientific information assessed in the  
7 AQCD and on the policy analyses contained in the Staff Paper. These four elements taken  
8 together determine the degree of public health and welfare protection afforded by the NAAQS.  
9

### 10 **1.1.3 Regulatory Chronology<sup>1</sup>**

11 On April 30, 1971, the EPA promulgated primary and secondary NAAQS for  
12 photochemical oxidants under Section 109 of the CAA (36 FR 8186). These were set at an  
13 hourly average of 0.08 ppm total photochemical oxidants, not to be exceeded more than 1 h per  
14 year. On April 20, 1977, the EPA announced (42 FR 20493) the first review and updating of the  
15 1970 Air Quality Criteria Document for Photochemical Oxidants in accordance with Section  
16 109(d) of the CAA. In preparing that AQCD, the EPA made two external review drafts of the  
17 document available for public comment, and these drafts were peer reviewed by the  
18 Subcommittee on Scientific Criteria for Photochemical Oxidants of EPA's Science Advisory  
19 Board (SAB). A final revised AQCD for ozone (O<sub>3</sub>) and other photochemical oxidants was  
20 published on June 22, 1978.

21 Based on the 1978 revised AQCD and taking into account the advice and recommendations  
22 of the SAB Subcommittee and public comments, the EPA announced (44 FR 8202) a final  
23 decision to revise the NAAQS for photochemical oxidants on February 8, 1979. That final  
24 rulemaking revised the primary standard from 0.08 ppm to 0.12 ppm, set the secondary standard  
25 to be the same as the primary standard, changed the chemical designation of the standards from  
26 photochemical oxidants to O<sub>3</sub>, and revised the definition of the point at which the standard is  
27 attained as indicated in Table 1-1.  
28

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<sup>1</sup>This following text is excerpted and adapted from the "Proposed Decision on the National Ambient Air Quality Standards for Ozone," 57 FR 35542, 35542-35557 (August, 10, 1992) and the "National Ambient Air Quality Standards for Ozone; Final Rule," 62 FR 38856, 83356-38896 (July 18, 1997).

**Table 1-1. National Ambient Air Quality Standards (NAAQS) for Ozone**

<b>Date of Promulgation</b>	<b>Primary and Secondary NAAQS</b>	<b>Averaging Time</b>
February 8, 1979	0.12 ppm <sup>a</sup> (235 µg/m <sup>3</sup> )	1 h <sup>b</sup>
July 18, 1997	0.08 ppm <sup>a</sup> (157 µg/m <sup>3</sup> )	8 h <sup>c</sup>

<sup>a</sup>1 ppm = 1962 µg/m<sup>3</sup>, 1 µg/m<sup>3</sup> = 5.097 × 10<sup>-4</sup> ppm @ 25 °C, 760 mm Hg.

<sup>b</sup>The standard is attained when the expected number of days per calendar year with a maximum hourly average concentration above 235 µg/m<sup>3</sup> (0.12 ppm) is equal to or less than one.

<sup>c</sup>Based on the 3-year average of the annual fourth-highest daily maximum 8-h average concentration measured at each monitor within an area.

Source: Federal Register (1979, 1997).

1           On March 17, 1982, in response to requirements of Section 109(d) of the CAA, the EPA  
2 announced (47 FR 11561) that it planned to revise the existing 1978 AQCD for O<sub>3</sub> and Other  
3 Photochemical Oxidants; and, on August 22, 1983, it announced (48 FR 38009) that review of  
4 the primary and secondary NAAQS for O<sub>3</sub> had been initiated. The EPA provided a number of  
5 opportunities for expert review and public comment on revised chapters of the AQCD, including  
6 two public peer-review workshops in December 1982 and November 1983. Comments made at  
7 both workshops were considered by EPA in preparing the First External Review Draft that was  
8 made available (49 FR 29845) on July 24, 1984, for public review. On February 13, 1985  
9 (50 FR 6049) and then on April 2, 1986 (51 FR 11339), the EPA announced two public CASAC  
10 meetings, which were held on March 4-6, 1985 and April 21-22, 1986, respectively. At these  
11 meetings, the CASAC reviewed external review drafts of the revised AQCD for O<sub>3</sub> and Other  
12 Photochemical Oxidants. After these two reviews, the Chair summarized CASAC's consensus  
13 view in an October 1986 letter to the EPA Administrator, which stated that the document  
14 "represents a scientifically balanced and defensible summary of the extensive scientific  
15 literature." Taking into account public and CASAC comments on the two external review drafts,  
16 revisions were made by EPA and the final document was released by EPA in August 1986.

17           The first draft of the Staff Paper "Review of the National Ambient Air Quality Standards  
18 for Ozone: Assessment of Scientific and Technical Information" drew upon key findings and  
19 conclusions from the AQCD and was reviewed by CASAC at the April 21-22, 1986 public  
20 meeting. At that meeting, the CASAC recommended that new information on prolonged O<sub>3</sub>

1 exposure effects be considered in a second draft of the Staff Paper. The CASAC reviewed the  
2 resulting second draft and also heard a presentation of new and emerging information on the  
3 health and welfare effects of O<sub>3</sub> at a December 14-15, 1987 public review meeting. The CASAC  
4 concluded that sufficient new information existed to recommend incorporation of relevant  
5 new data into a supplement to the 1986 AQCD (O<sub>3</sub> Supplement) and in a third draft of the  
6 Staff Paper.

7 A draft O<sub>3</sub> Supplement, "Summary of Selected New Information on Effects of Ozone on  
8 Health and Vegetation: Draft Supplement to Air Quality Criteria for Ozone and Other  
9 Photochemical Oxidants," and the revised Staff Paper were made available to CASAC and to the  
10 public in November 1988. The O<sub>3</sub> Supplement assessed selected literature concerning exposure-  
11 and concentration-response relationships observed for health effects in humans and experimental  
12 animals and for vegetation effects that appeared in papers published or in-press from 1986  
13 through early 1989. On December 14-15, 1988, CASAC held a public meeting to review these  
14 documents and then sent the EPA Administrator a letter (dated May 1, 1989), which stated that  
15 the draft O<sub>3</sub> Supplement, the 1986 AQCD, and the draft Staff Paper "provide an adequate  
16 scientific basis for the EPA to retain or revise the primary and secondary standards of ozone."  
17 The CASAC concluded (a) that it would be some time before sufficient new information on the  
18 health effects of multihour and chronic exposure to O<sub>3</sub> would be published in scientific journals  
19 to receive full peer review and, thus, be suitable for inclusion in a criteria document and (b) that  
20 such information could be considered in the next review of the O<sub>3</sub> NAAQS. A final version of  
21 the O<sub>3</sub> Supplement was published in 1992 (U.S. Environmental Protection Agency, 1992).

22 On October 22, 1991, the American Lung Association and other plaintiffs filed suit to  
23 compel the Agency to complete the review of the criteria and standards for O<sub>3</sub> in accordance  
24 with the CAA. The U.S. District Court for the Eastern District of New York subsequently issued  
25 an order requiring the EPA to announce its proposed decision on whether to revise the standards  
26 for O<sub>3</sub> by August 1, 1992 and to announce its final decision by March 1, 1993.

27 The proposed decision on O<sub>3</sub>, which appeared in the Federal Register on August 10, 1992  
28 (57 FR 35542), indicated that revision of the existing 1-h NAAQS was not appropriate at that  
29 time. A public hearing on this decision was held in Washington, DC on September 1, 1992; and  
30 public comments were received through October 9, 1992. The final decision not to revise the  
31 1-h NAAQS was published in the Federal Register on March 9, 1993 (58 FR 13008). However,



1 that decision did not take into consideration a number of more recent studies on the health  
2 and welfare effects of O<sub>3</sub> that had been published since the last of the literature assessed in  
3 the O<sub>3</sub> Supplement (i.e., studies available through 1985 and into early 1986).

4 The Agency initiated consideration of such studies as part of the next congressionally-  
5 mandated periodic review of criteria and NAAQS for Ozone. The new studies were assessed in  
6 revised draft O<sub>3</sub> AQCD chapters that were peer reviewed in July and September 1993  
7 workshops, followed by public release of the First External Review Draft in February 1994 and  
8 CASAC review on July 20-21, 1994. Further drafts of the O<sub>3</sub> AQCD, revised in response to  
9 public comments and CASAC review, were reviewed by CASAC on March 21-25, 1995, and at  
10 a final CASAC review meeting on September 19-20, 1995. The scientific soundness of the  
11 revised O<sub>3</sub> AQCD was recognized by CASAC in a November 28, 1995 letter to the EPA  
12 Administrator; and the final AQCD for O<sub>3</sub> was published in July 1996.

13 The first draft of the associated Staff Paper, “Review of the National Ambient Air Quality  
14 Standards for Ozone: Assessment of Scientific and Technical Information,” was also reviewed  
15 by CASAC at the March 21-22, 1995 public meeting. CASAC also reviewed subsequent drafts  
16 of the Staff Paper at public meetings on September 19-20, 1995 and March 21, 1996, with  
17 completion of CASAC review of the primary and secondary standard portions of the draft Staff  
18 Paper being communicated in letters to the EPA Administrator dated November 30, 1995 and  
19 April 4, 1996, respectively. The final O<sub>3</sub> Staff Paper was published in June 1996.

20 On December 13, 1996 EPA published its proposed decision to revise the O<sub>3</sub> NAAQS  
21 (61 FR 65716). EPA provided extensive opportunities for public comment on the proposed  
22 decision, including several public hearings and two national satellite telecasts. EPA's final  
23 decision to promulgate a new 8-h O<sub>3</sub> NAAQS (see Table 1-1) was published on July 18, 1997  
24 (62 FR 38856).

25 Following promulgation of the new standards, numerous petitions for review of the  
26 standards were filed in the U.S. Court of Appeals for the District of Columbia Circuit (D.C.  
27 Circuit)<sup>2</sup>. On May 14, 1999, the Court remanded the O<sub>3</sub> NAAQS to EPA, finding that section  
28 109 of the CAA, as interpreted by EPA, effected an unconstitutional delegation of legislative

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<sup>2</sup>*American Trucking Associations v. EPA*, No. 97-1441

1 authority<sup>3</sup>. In addition, the Court directed that, in responding to the remand, EPA should  
2 consider the potential beneficial health effects of O<sub>3</sub> pollution in shielding the public from the  
3 effects of solar ultraviolet (UV) radiation. On January 27, 2000, EPA petitioned the U.S.  
4 Supreme Court for certiorari on the constitutional issue (and two other issues), but did not  
5 request review of the D.C. Circuit ruling regarding the potential beneficial health effects of O<sub>3</sub>.  
6 On February 27, 2001 the U.S. Supreme Court unanimously reversed the judgment of the D.C.  
7 Circuit on the constitutional issue, holding that section 109 of the CAA does not delegate  
8 legislative power to the EPA in contravention of the Constitution, and remanded the case to the  
9 D.C. Circuit to consider challenges to the O<sub>3</sub> NAAQS that had not been addressed by that Court's  
10 earlier decisions<sup>4</sup>. On March 26, 2002, the D.C. Circuit issued its final decision, finding the  
11 1997 O<sub>3</sub> NAAQS to be “neither arbitrary nor capricious,” and denied the remaining petitions  
12 for review<sup>5</sup>.

13 On November 14, 2001 EPA proposed to respond to the Court's remand to consider the  
14 potential beneficial health effects of O<sub>3</sub> pollution in shielding the public from the effects of solar  
15 UV radiation by leaving the 1997 8-h NAAQS unchanged. Following a review of information in  
16 the record and the substantive comments received on the proposed response, EPA issued a final  
17 response to the remand, reaffirming the 8 h O<sub>3</sub> NAAQS (68 FR 614, January 6, 2003).

## 18 19 20 **1.2 CURRENT OZONE CRITERIA AND NAAQS REVIEW**

### 21 **1.2.1 Key Milestones and Procedures for Document Preparation**

22 It is important to note at the outset that development of the present document has and will  
23 continue to include substantial external expert review and opportunities for public input through  
24 (a) public workshops involving the general scientific community, (b) iterative reviews of  
25 successive drafts by CASAC, and (c) comments from the public on successive drafts. Extensive  
26 external inputs received through such reviews will help to ensure that the review of the O<sub>3</sub>

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<sup>3</sup> *American Trucking Associations v. EPA*, 175 F.3d 1027 (D.C. Cir., 1999)

<sup>4</sup> *Whitman v. American Trucking Associations*, 531 U.S. 457 (2001)

<sup>5</sup> *American Trucking Associations v. EPA*, 283 F.3d 355, (D.C. Cir. 2002)

1 standards will be based on critical assessment in this document of the latest available pertinent  
2 science.

3 The procedures for developing this revised O<sub>3</sub> AQCD build on experience derived from the  
4 other recent criteria document preparation efforts, with key milestones for development of  
5 this O<sub>3</sub> AQCD being listed in Table 1-2. Briefly, the respective responsibilities for production  
6 of the document and key milestones are as follows. An NCEA-RTP Ozone Team is responsible  
7 for the creation and implementation of a project plan for developing the O<sub>3</sub> AQCD, taking into  
8 account input from individuals in other EPA program and policy offices identified as part of the  
9 EPA Ozone Work Group. The resulting plan, i.e., the Project Work Plan for Revised Air  
10 Criteria for Ozone and Related Photochemical Oxidants (November 2002), was discussed with  
11 CASAC in January 2003. An ongoing literature search that was underway prior to initiation of  
12 work on this document has continued throughout its preparation to identify pertinent O<sub>3</sub>  
13 literature published since early 1996. Under the processes established in Sections 108 and 109  
14 of the CAA, the EPA officially initiated the current criteria and NAAQS review by announcing  
15 the commencement of the review in the Federal Register (65 FR 57810, September, 2000) with a  
16 call for information. That Federal Register notice included (1) a request asking for recently  
17 available research information on O<sub>3</sub> that may not yet have been published and (2) a request for  
18 individuals with the appropriate type and level of expertise to contribute to the writing of O<sub>3</sub>  
19 AQCD materials to identify themselves. The specific authors of chapters or sections of the  
20 proposed document included both EPA and non-EPA scientific experts, who were selected on  
21 the basis of their expertise on the subject areas and their familiarity with the relevant literature.  
22 The project team defined critical issues and topics to be addressed by the authors and provided  
23 direction in order to focus on evaluation of those studies most clearly identified as important for  
24 standard setting.

25 As with other NAAQS reviews, critical assessment of relevant scientific information is  
26 presented in this updated O<sub>3</sub> AQCD. The main focus of this document is the evaluation and  
27 interpretation of pertinent atmospheric science information, air quality data, human exposure  
28 information, and health and welfare effects information newly published since that assessed in  
29 the 1996 O<sub>3</sub> AQCD. Draft versions of AQCD chapter materials were evaluated via expert peer-  
30 consultation workshop discussions (see Table 1-2) that focused on the selection of pertinent  
31 studies to be included in the chapters, the potential need for additional information to be added to

**Table 1-2. Key Milestones for Development of Revised Ozone Air Quality Criteria Document<sup>a</sup>**

<u>Major Milestones</u>	<u>Target Dates</u>
1. Literature Search	Ongoing
2. Federal Register Call for Information	September 2000
3. Draft Project Plan Available for Public Comment	Dec 2001 - March 2002
4. Revised Draft Project Plan Released for CASAC Review	December 2002
5. CASAC Review of Draft Project Work Plan	January 2003
6. Peer-Consultation Workshop on Draft Ecological Effects Materials	April 2003
7. Peer-Consultation Workshops on Draft Atmospheric Science/Exposure and Dosimetry/Health Chapters	July 2004
8. First External Review Draft of O <sub>3</sub> AQCD	January 2005
9. Public Comment Period (90 days)	Feb - April 2005
10. CASAC Public Review Meeting ( <i>First External Review Draft</i> )	May 2005
11. Second External Review Draft of O <sub>3</sub> AQCD	August 2005
12. Public Comment Period (90 days)	Sept - Nov 2005
13. CASAC Public Review Meeting	December 2005
14. Final O <sub>3</sub> AQCD	February 2006

<sup>a</sup> Proposed schedule will be modified from time to time, as necessary, to reflect actual project requirements and progress.

1 the chapters, and the quality of the characterization and interpretation of the literature. The  
 2 authors of the draft chapters then revised them on the basis of the workshop and/or other expert  
 3 review comments<sup>6</sup>. These and other integrative materials were then incorporated into the First  
 4 External Review Draft of this O<sub>3</sub> AQCD (January 2005), which was made available for public  
 5 comment and CASAC review (see Table 1-2).

6 Following review of the First External Review Draft at a May 4-5, 2005 CASAC meeting,  
 7 EPA incorporated revisions into the draft O<sub>3</sub> AQCD in response to comments from CASAC and  
 8 the public and has made this Second External Review Draft (August, 2005) available for further

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<sup>6</sup>It should be noted that materials contributed by non-EPA authors have, at times, been modified by EPA Ozone Team staff in response to internal and/or external review comments and that EPA is responsible for the ultimate content of this O<sub>3</sub> AQCD.

1 public comment and CASAC review according to the schedule projected in Table 1-2. More  
2 specifically, this Second External Review Draft is available for public comment (90 days) during  
3 September-November, 2005, and will be reviewed by CASAC at a public meeting in December  
4 2005 (the site and specific dates to be announced in the Federal Register). The final O<sub>3</sub> AQCD is  
5 to be completed by February 28, 2006, and it is to be made publicly available electronically via  
6 an EPA website and then subsequently printed. Its availability will be announced in the Federal  
7 Register.

8 The EPA's Office of Air Quality Planning and Standards (OAQPS) staff has also prepared  
9 a first draft O<sub>3</sub> Staff Paper drawing upon key information contained in this Second External  
10 Review Draft O<sub>3</sub> AQCD, which presents recommendations regarding whether to retain or, if  
11 appropriate, to revise the O<sub>3</sub> NAAQS. After review of that draft Staff Paper (dated September,  
12 2005) by the public and by CASAC, EPA will take public and CASAC comments into account  
13 in producing a Second Draft Staff Paper. That Second Draft Staff Paper (based on the final  
14 version of this O<sub>3</sub> AQCD) will also be made available for further public comment and CASAC  
15 review before EPA produces a final ozone Staff Paper by September 30, 2006.

## 18 **1.3 ORGANIZATIONAL STRUCTURE OF THE DOCUMENT**

### 19 **1.3.1 General Document Format**

20 The general format used in preparing this draft document is to open each new section for  
21 the updated document with concise summarization of key findings and conclusions from the  
22 previous 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996). After presentation of  
23 such background information, the remainder of each section typically provides an updated  
24 discussion of newer literature and resulting key conclusions. In some cases where no new  
25 information is available, the summary of key findings and conclusions from the previous criteria  
26 document must suffice as the basis for current key conclusions. Increased emphasis is placed in  
27 the main chapters of this revised O<sub>3</sub> AQCD on interpretative evaluation and integration of  
28 evidence pertaining to a given topic than has been typical of previous EPA air quality criteria  
29 documents, with more detailed descriptions of individual studies being provided in a series of  
30 accompanying annexes.

1 A list of references published since completion of the 1996 criteria document was made  
2 available to the authors. The references were selected from information data base searches  
3 conducted by EPA. Additional references have been added to the list (e.g., missed or recently  
4 published papers or “in press” publications) as work has proceeded in creating the draft  
5 document materials. As an aid in selecting pertinent new literature, the authors were also  
6 provided with a summary of issues that need to be addressed in the revised air quality criteria  
7 document for O<sub>3</sub>. These issues were identified by authors and reviewers of the previous  
8 documents and continue to be expanded, as appropriate, based on public discussions, workshops,  
9 or other comments received by EPA.

### 11 **1.3.2 Organization and Content of the Document**

12 This revised AQCD for O<sub>3</sub> and Related Photochemical Oxidants critically assesses  
13 scientific information on the health and welfare effects associated with exposure to the  
14 concentrations of these pollutants in ambient air. The document does not provide a detailed  
15 literature review; but, rather, discusses cited references that reflect the current state of knowledge  
16 on the most relevant issues pertinent to the NAAQS for O<sub>3</sub>. Although emphasis is placed on  
17 discussion of health and welfare effects information, other scientific data are presented and  
18 evaluated in order to provide a better understanding of the nature, sources, distribution,  
19 measurement, and concentrations of O<sub>3</sub> and related photochemical oxidants in ambient air,  
20 as well as the measurement of population exposure to these pollutants.

21 The main focus of the scientific information discussed in the text comes from literature  
22 published since completion of the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency,  
23 1996). Emphasis is placed on studies conducted at or near O<sub>3</sub> concentrations found in ambient  
24 air. Other studies are included if they contain unique data, such as the documentation of a  
25 previously unreported effect or of a mechanism for an observed effect; or if they were multiple-  
26 concentration studies designed to provide exposure-response relationships. Generally, this is not  
27 an issue for human clinical or epidemiology studies. However, for animal toxicology studies,  
28 consideration is given mainly to those studies conducted at less than 1 ppm O<sub>3</sub>. Key information  
29 from studies assessed in the previous O<sub>3</sub> AQCD and whose data impacted the derivation of the  
30 current NAAQS are briefly summarized in the text, along with specific citations to the previous  
31 document. Prior studies are also discussed if they (1) are open to reinterpretation in light of

1 newer data, or (2) are potentially useful in deriving revised standards for O<sub>3</sub>. Generally, only  
2 information that has undergone scientific peer review and has been published (or accepted for  
3 publication) through December 2004 is included in this draft document. A few particularly  
4 pertinent and important new studies published or accepted for publication beyond the end of  
5 2004 are also considered.

6 This document consists of three volumes. The first volume includes an Executive  
7 Summary and Conclusions, as well as Chapters 1 through 11 of the O<sub>3</sub> AQCD. This introductory  
8 chapter (Chapter 1) presents background information on the purpose of the document, legislative  
9 requirements, and the history of past O<sub>3</sub> NAAQS regulatory actions, as well as an overview of  
10 the organization and content of the document. Chapter 2 provides information on the physics  
11 and chemistry of O<sub>3</sub> and related photochemical oxidants in the atmosphere. Chapter 3 covers  
12 tropospheric O<sub>3</sub> environmental concentrations, patterns, and exposure estimates. The  
13 accompanying annexes to each of these background chapters are found in Volume II.

14 Health information pertinent to derivation of the primary O<sub>3</sub> NAAQS is then mainly  
15 covered in the next several chapters (Chapters 4 through 8). Chapter 4 discusses O<sub>3</sub> dosimetry  
16 aspects, and Chapters 5, 6, and 7 discuss animal toxicological studies, human health effects from  
17 controlled-exposure studies, and epidemiologic studies of ambient air exposure effects on human  
18 populations, respectively. Chapter 8 then provides an integrative and interpretive evaluation of  
19 key information relevant to O<sub>3</sub> exposure and health risks, of most pertinence to the review of  
20 primary O<sub>3</sub> NAAQS. The annexes to these health-related chapters are found in Volume II.

21 The remaining three chapters of the document assess welfare effects information pertinent  
22 to the review of secondary O<sub>3</sub> NAAQS. Chapter 9 deals with ecological and other  
23 environmental effects of O<sub>3</sub> and related photochemical oxidants. Chapter 10 assesses  
24 tropospheric O<sub>3</sub> involvement in climate change processes, including determination of solar UV  
25 flux in Earth's lower atmosphere. Lastly, Chapter 11 discusses O<sub>3</sub> effects on man-made  
26 materials as a third type of welfare effect of potential concern. Annex materials related to  
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48



## 2. PHYSICS AND CHEMISTRY OF OZONE IN THE ATMOSPHERE

### 2.1 INTRODUCTION

Ozone ( $O_3$ ) and other oxidants, such as peroxy nitrates and hydrogen peroxide ( $H_2O_2$ ) form in polluted areas by atmospheric reactions involving two main classes of precursor pollutants, volatile organic compounds (VOCs) and nitrogen oxides ( $NO_x$ ). Carbon monoxide (CO) is also important for ozone formation in polluted areas. Ozone is thus a secondary pollutant. The formation of  $O_3$ , other oxidants and oxidation products from these precursors is a complex, nonlinear function of many factors: the intensity and spectral distribution of sunlight; atmospheric mixing and processing on cloud and aerosol particles; the concentrations of the precursors in ambient air; and the rates of chemical reactions of the precursors. Information contained in this chapter and in greater detail in Annex AX2 describes these processes, numerical models that incorporate these processes to calculate  $O_3$  concentrations, and techniques for measuring concentrations of ambient oxidants.

The atmosphere can be divided into several distinct vertical layers, based primarily on the major mechanisms by which they are heated and cooled. The lowest major layer is the troposphere, which extends from the earth's surface to about 8 km above polar regions and to about 16 km above tropical regions. The planetary boundary layer (PBL) is the lower sublayer of the troposphere, extending from the surface to about 1 or 2 km and is most strongly affected by surface conditions. The stratosphere extends from the tropopause, or the top of the troposphere, to about 50 km in altitude (Annex AX2.2.1). The emphasis in this chapter is placed on chemical and physical processes occurring in the troposphere, in particular in the PBL. The processes responsible for producing summertime  $O_3$  episodes are fairly well understood, as outlined in the previous Air Quality Criteria Document for Ozone and Related Photochemical Oxidants (CD96). This chapter mainly considers topics for which there is substantial new information and on topics that form the basis for discussions in later chapters.

## 2.2 CHEMICAL PROCESSES INVOLVED IN OZONE FORMATION AND DESTRUCTION

Ozone occurs not only in polluted urban atmospheres but throughout the troposphere, even in remote areas of the globe. The same basic processes, involving sunlight-driven reactions of  $\text{NO}_x$  and VOCs contribute to  $\text{O}_3$  formation throughout the troposphere. These processes also lead to the formation of other photochemical products, such as peroxyacetyl nitrate (PAN), nitric acid ( $\text{HNO}_3$ ), and sulfuric acid ( $\text{H}_2\text{SO}_4$ ), and to other compounds, such as formaldehyde ( $\text{HCHO}$ ) and other carbonyl compounds, such as aldehydes and ketones.

The photochemical formation of  $\text{O}_3$  in the troposphere proceeds through the oxidation of nitric oxide ( $\text{NO}$ ) to nitrogen dioxide ( $\text{NO}_2$ ) by organic ( $\text{RO}_2$ ) or hydro-peroxy ( $\text{HO}_2$ ) radicals. The photolysis of  $\text{NO}_2$  yields nitric oxide ( $\text{NO}$ ) and a ground-state oxygen atom,  $\text{O}(^3\text{P})$ , which then reacts with molecular oxygen to form  $\text{O}_3$ . Free radicals oxidizing  $\text{NO}$  to  $\text{NO}_2$  are formed during the oxidation of VOCs (Annex AX2.2.2).

The term VOC refers to all carbon-containing gas-phase compounds in the atmosphere, both biogenic and anthropogenic in origin, excluding carbon monoxide ( $\text{CO}$ ) and carbon dioxide ( $\text{CO}_2$ ). Classes of organic compounds important for the photochemical formation of  $\text{O}_3$  include alkanes, alkenes, aromatic hydrocarbons, carbonyl compounds (e.g., aldehydes and ketones), alcohols, organic peroxides, and halogenated organic compounds (e.g., alkyl halides). This array of compounds encompasses a wide range of chemical properties and lifetimes: isoprene has an atmospheric lifetime of approximately an hour, whereas methane has an atmospheric lifetime of about a decade.

In urban areas, compounds representing all classes of VOCs, and  $\text{CO}$  are important for  $\text{O}_3$  formation. In nonurban vegetated areas, biogenic VOCs emitted from vegetation tend to be the most important. In the remote troposphere,  $\text{CH}_4$  and  $\text{CO}$  are the main carbon-containing precursors to  $\text{O}_3$  formation.  $\text{CO}$  also can play an important role in  $\text{O}_3$  formation in urban areas. The oxidation of VOCs is initiated mainly by reaction with hydroxyl ( $\text{OH}$ ) radicals. The primary source of  $\text{OH}$  radicals in the atmosphere is the reaction of electronically excited  $\text{O}$  atoms,  $\text{O}(^1\text{D})$ , with water vapor.  $\text{O}(^1\text{D})$  is produced by the photolysis of  $\text{O}_3$  in the Hartley bands. In polluted areas, the photolysis of aldehydes (e.g.,  $\text{HCHO}$ ), nitrous acid ( $\text{HONO}$ ) and hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) can also be significant sources of  $\text{OH}$  or  $\text{HO}_2$  radicals that can rapidly be converted to  $\text{OH}$  (Eisele et al., 1997). Ozone can oxidize alkenes, and, at night, when they are

1 most abundant, NO<sub>3</sub> radicals also oxidize alkenes. In coastal environments and other selected  
2 environments, atomic Cl and Br radicals can also initiate the oxidation of VOCs (Annex  
3 AX2.2.3).

4 There are a large number of oxidized nitrogen containing compounds in the atmosphere  
5 including NO, NO<sub>2</sub>, NO<sub>3</sub>, HNO<sub>2</sub>, HNO<sub>3</sub>, N<sub>2</sub>O<sub>5</sub>, HNO<sub>4</sub>, PAN and its homologues, other organic  
6 nitrates and particulate nitrate. Collectively these species are referred to as NO<sub>y</sub>. Oxidized  
7 nitrogen compounds are emitted to the atmosphere mainly as NO which rapidly interconverts  
8 with NO<sub>2</sub> and so NO and NO<sub>2</sub> are often “lumped” together into their own group or family,  
9 or NO<sub>x</sub>. NO<sub>x</sub> can be oxidized to reservoir and termination species (PAN and its homologues,  
10 organic nitrates, HNO<sub>3</sub>, HNO<sub>4</sub> and particulate nitrate). These reservoir and termination species  
11 are referred to as NO<sub>z</sub>. The major reactions involving inter-conversions of oxidized nitrogen  
12 species are discussed in Annex AX2.2.4.

13 The photochemical cycles by which the oxidation of hydrocarbons leads to O<sub>3</sub> production  
14 are best understood by considering the oxidation of methane, structurally the simplest VOC.  
15 The CH<sub>4</sub> oxidation cycle serves as a model for the chemistry of the relatively clean or unpolluted  
16 troposphere (although this is a simplification because vegetation releases large quantities of  
17 complex VOCs, such as isoprene, into the atmosphere). In the polluted atmosphere, the  
18 underlying chemical principles are the same, as discussed in Annex AX2.2.5. The conversion of  
19 NO to NO<sub>2</sub> occurring with the oxidation of VOCs is accompanied by the production of O<sub>3</sub> and  
20 the efficient regeneration of the OH radical, which in turn can react with other VOCs.  
21 A schematic overview showing the major processes involved in O<sub>3</sub> production and loss in the  
22 troposphere and stratosphere is given in Figure 2-1.

23 The oxidation of alkanes and alkenes in the atmosphere has been treated in depth in CD96  
24 and is updated in Annexes AX2.2.6 and AX2.2.7. In contrast to simple hydrocarbons containing  
25 one or two carbon atoms, detailed kinetic information about the gas phase oxidation pathways of  
26 many anthropogenic hydrocarbons (e.g., aromatic compounds, such as benzene and toluene),  
27 biogenic hydrocarbons (e.g., isoprene, the monoterpenes), and their intermediate oxidation  
28 products (e.g., epoxides, nitrates, and carbonyl compounds) is lacking. Reaction with OH  
29 radicals represents the major loss process for alkanes. Reaction with chlorine atoms is an  
30 additional sink for alkanes. Stable products of alkane photooxidation are known to include  
31 carbonyl compounds, alkyl nitrates, and *d*-hydroxycarbonyls. Major uncertainties in the

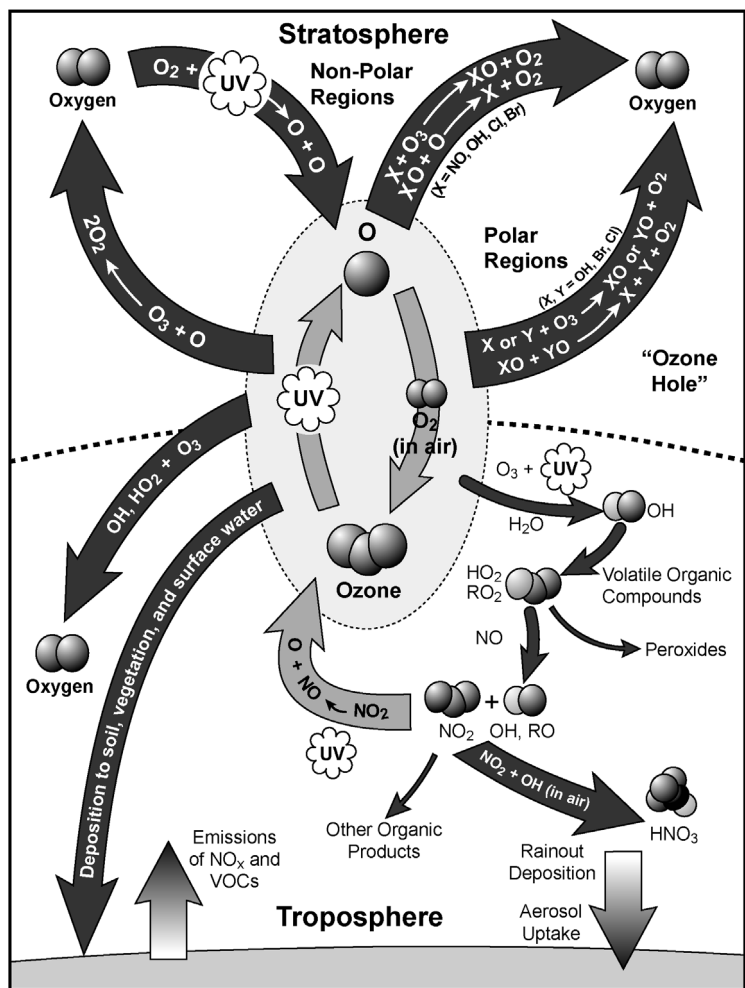


Figure 2-1. Schematic overview of O<sub>3</sub> photochemistry in the stratosphere and troposphere.

1 atmospheric chemistry of the alkanes concern the chemistry of alkyl nitrate formation; these  
 2 uncertainties affect the amount of NO-to-NO<sub>2</sub> conversion occurring and, hence, the amounts  
 3 of O<sub>3</sub> formed during photochemical degradation of the alkanes.

4 The reaction of OH radicals with aldehydes produced during the oxidation of alkanes  
 5 forms acyl (R'CO) radicals, and acyl peroxy radicals (R'C(O)-O<sub>2</sub>) are formed by the further  
 6 addition of O<sub>2</sub>. As an example, the oxidation of ethane (C<sub>2</sub>H<sub>5</sub>-H) yields acetaldehyde  
 7 (CH<sub>3</sub>-CHO). The reaction of CH<sub>3</sub>-CHO with OH radicals yields acetyl radicals (CH<sub>3</sub>-CO).  
 8 The acetyl radicals will then participate with O<sub>2</sub> in a termolecular recombination reaction to form

1 acetyl peroxy radicals, which can then react with NO to form  $\text{CH}_3 + \text{CO}_2$  or they can react  
2 with  $\text{NO}_2$  to form PAN. PAN acts as a temporary reservoir for  $\text{NO}_2$ . Upon the thermal  
3 decomposition of PAN, either locally or elsewhere,  $\text{NO}_2$  is released to participate in the  $\text{O}_3$   
4 formation process again.

5 Alkenes react in ambient air with OH,  $\text{NO}_3$ , and Cl radicals and with  $\text{O}_3$ . All of these  
6 reactions are important atmospheric transformation processes, and all proceed by initial addition  
7 to the  $>\text{C}=\text{C}<$  bonds. Products of alkene photooxidation include carbonyl compounds,  
8 hydroxynitrates and nitratocarbonyls, and decomposition products from the energy-rich  
9 biradicals formed in alkene- $\text{O}_3$  reactions. Major uncertainties in the atmospheric chemistry of  
10 the alkenes concern the products and mechanisms of their reactions with  $\text{O}_3$ , especially the yields  
11 of free radicals that participate in  $\text{O}_3$  formation. Examples of oxidation mechanisms of complex  
12 alkanes and alkenes can be found in comprehensive texts such as Seinfeld and Pandis (1998).

13 The oxidation of aromatic hydrocarbons constitutes an important component of the  
14 chemistry of  $\text{O}_3$  formation in urban atmospheres (Annex AX2.2.8). Virtually all of the important  
15 aromatic hydrocarbon precursors emitted in urban atmospheres are lost through reaction with the  
16 hydroxyl radical. Loss rates for these compounds vary from slow (i.e., benzene) to moderate  
17 (e.g., toluene), to very rapid (e.g., xylene and trimethylbenzene isomers). These loss rates are  
18 very well understood at room temperature and atmospheric pressure and numerous experiments  
19 have been conducted that verify this. However, the mechanism for the oxidation of aromatic  
20 hydrocarbons following reaction with OH is poorly understood, as evident from the poor mass  
21 balance of the reaction products. The mechanism for the oxidation of toluene has been studied  
22 most thoroughly and there is general agreement on the initial steps in the mechanism. However,  
23 at present there is no promising approach for resolving the remaining issues concerning the later  
24 steps. The oxidation of aromatic hydrocarbons also leads to particle formation which could  
25 remove gas-phase constituents that participate in  $\text{O}_3$  formation. The chemistry of secondary  
26 organic aerosol formation from gaseous precursors was summarized in the latest AQCD for  
27 particulate matter.

28 The reactions of oxygenated VOCs are also important components of  $\text{O}_3$  formation (Annex  
29 AX2.2.9). They may be produced either by the oxidation of hydrocarbons or they may be  
30 present in ambient air as the result of direct emissions. For example, motor vehicles and some  
31 industrial processes emit formaldehyde and vegetation emits methanol.

1 As much as 30% of the carbon in hydrocarbons in many urban areas is in the form of  
2 aromatic compounds. Yet, mass balance analyses performed on irradiated smog chamber  
3 mixtures of aromatic hydrocarbons indicate that only about one-half of the carbon is in the form  
4 of compounds that can be identified. The situation is not much better for some smaller  
5 anthropogenic hydrocarbons. For example, only about 60% of the initial carbon can be  
6 accounted for in the OH initiated oxidation of 1,3-butadiene. About two-thirds of the initial  
7 carbon can be identified in product analyses of isoprene oxidation. Adequate analytical  
8 techniques needed to identify and quantify key intermediate species are not available for many  
9 compounds. In addition, methods to synthesize many of the suspected intermediate compounds  
10 are not available so that laboratory studies of their reaction kinetics cannot be performed.  
11 Similar considerations apply to the oxidation of biogenic hydrocarbons besides isoprene.

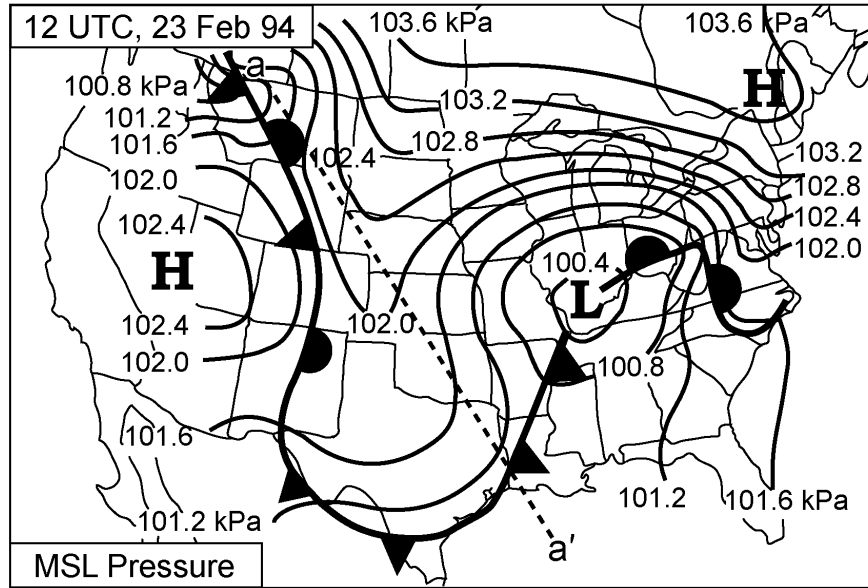
12 In addition to reactions occurring in the gas phase, reactions occurring on the surfaces of or  
13 within cloud droplets and airborne particles also occur. Their collective surface area is huge  
14 implying that collisions with gas phase species occur on very short time scales. In addition to  
15 hydrometeors (e.g., cloud and fog droplets and snow and ice crystals) there are also potential  
16 reactions involving atmospheric particles of varying composition (e.g., wet [deliquesced]  
17 inorganic particles, mineral dust, carbon chain agglomerates and organic carbon particles) to  
18 consider. Most of the well-established multiphase reactions tend to reduce the rate of O<sub>3</sub>  
19 formation in the polluted troposphere. Removal of HO<sub>x</sub> and NO<sub>x</sub> onto hydrated particles will  
20 reduce the production of O<sub>3</sub>. However, the photolysis of HONO formed in reactions such as  
21 these can increase the production of O<sub>3</sub>. The reactions of Br and Cl containing radicals  
22 deplete O<sub>3</sub> in selected environments such as the Arctic during spring, the tropical marine  
23 boundary layer and inland salt lakes. Direct reactions of O<sub>3</sub> and atmospheric particles appear to  
24 be too slow to reduce O<sub>3</sub> formation significantly at typical ambient PM levels. In addition, the  
25 oxidation of hydrocarbons by Cl radicals could lead to the rapid formation of peroxy radicals and  
26 higher rates of O<sub>3</sub> production in selected coastal environments. It should be stressed that  
27 knowledge of multiphase processes is still evolving and there are still many questions that  
28 remain to be answered as outlined in Annex AX2.2.10.

29 The oxidants, other than O<sub>3</sub>, that are formed from the chemistry described above could  
30 exert effects on human health and perhaps also on vegetation. Gas phase oxidants include  
31 PAN, H<sub>2</sub>O<sub>2</sub> and CH<sub>3</sub>OOH and other organic hydroperoxides (Annex AX2.2). In addition to

1 transfer from the gas phase, oxidants can be formed by photochemical reactions occurring in  
2 particles (Annex 2.2.10.6). However, the pathways leading to the formation of oxidants in the  
3 particle phase are not as well understood as they are in the gas phase.  
4  
5

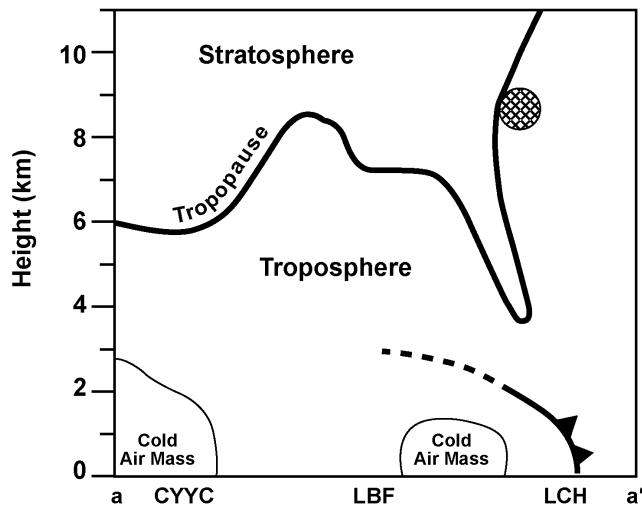
### 6 **2.3 METEOROLOGICAL PROCESSES AFFECTING OZONE**

7 Since CD96, substantial new information about transport processes has become available  
8 from numerical models, field experiments and satellite-based observations. Ozone is produced  
9 naturally by photochemical reactions in the stratosphere as shown in Figure 2-1. Some of this O<sub>3</sub>  
10 is transported downward into the troposphere throughout the year, with maximum contributions  
11 during late winter and early spring mainly in a process known as tropopause folding. Figure  
12 2-2a shows a synoptic situation associated with a tropopause folding event. A vertical cross  
13 section taken through the atmosphere from a to a' is shown in Figure 2-2b. In this figure the  
14 tropopause fold is shown folding downward above and slightly behind the surface cold front,  
15 bringing stratospheric air with it. Although the tropopause is drawn with a solid line, it should  
16 not be taken to mean that it is a material surface, through which there is no exchange. Rather  
17 these folds should be thought of as regions in which mixing of tropospheric and stratospheric air  
18 is occurring (Shapiro, 1980). This imported stratospheric air contributes to the natural  
19 background of O<sub>3</sub> in the troposphere, especially in the free troposphere. It should be noted that  
20 there is considerable uncertainty in the magnitude and distribution of this potentially important  
21 source of tropospheric O<sub>3</sub>. Stratospheric intrusions that reach the surface are rare. Much more  
22 common are intrusions which penetrate only to the middle and upper troposphere. However, O<sub>3</sub>  
23 transported to the upper and middle troposphere can still affect surface concentrations through  
24 various exchange mechanisms that mix air from the free troposphere with air in the planetary  
25 boundary layer. Substantial photochemical production of O<sub>3</sub> in the troposphere also begins in  
26 late winter and early spring; therefore, it cannot be assumed that O<sub>3</sub> present at these times is only  
27 stratospheric in origin. The basic atmospheric dynamics and thermodynamics of stratospheric-  
28 tropospheric exchange are outlined in Annex AX2.3.1.  
29



**Figure 2-2a.** Surface weather chart showing sea level (MSL) pressure (kPa), and surface fronts.

Source: Stull (2000).



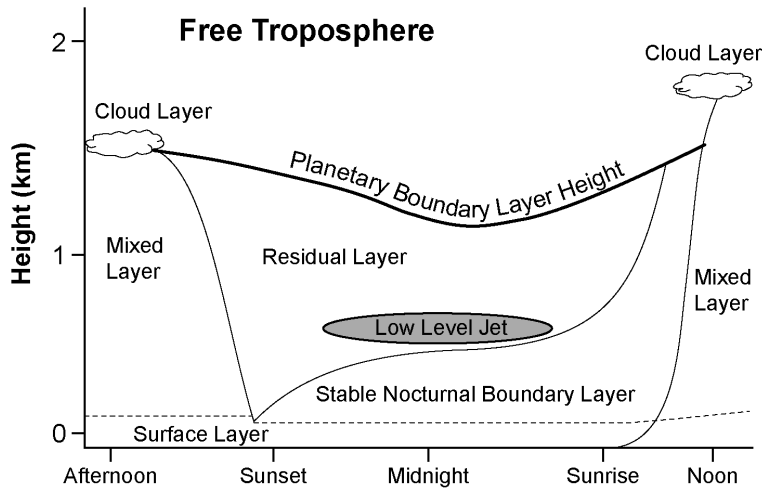
**Figure 2-2b.** Vertical cross section along dashed line (a-a') from northwest to the southeast (CYYC = Calgary, Alberta; LBF = North Platte, NB; LCH = Lake Charles, LA). The approximate location of the jet stream core is indicated by the hatched area. The position of the surface front is indicated by the cold-frontal symbols and the frontal inversion top by the dashed line. Note: This is 12 h later than the situations shown in Figure 2-2a.

Source: Adapted from Stull (2000).



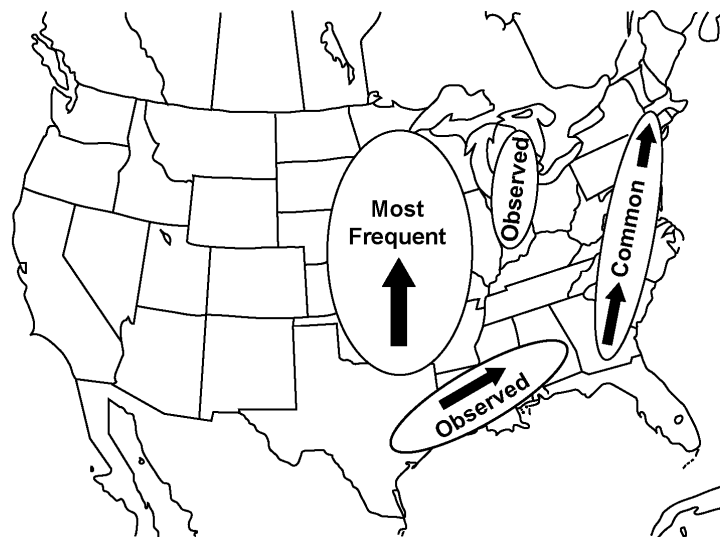
1 Our understanding of the meteorological processes associated with summertime O<sub>3</sub> episodes  
2 remains basically the same as outlined in CD96. Major episodes of high O<sub>3</sub> concentrations in the  
3 eastern United States and in Europe are associated with slow moving, high pressure systems.  
4 High pressure systems during the warmer seasons are associated with the sinking of air, resulting  
5 in warm, generally cloudless skies, with light winds. The sinking of air results in the  
6 development of stable conditions near the surface which inhibit or reduce the vertical mixing  
7 of O<sub>3</sub> precursors. The combination of inhibited vertical mixing and light winds minimizes the  
8 dispersal of pollutants emitted in urban areas, allowing their concentrations to build up.  
9 Photochemical activity involving these precursors is enhanced because of higher temperatures  
10 and the availability of sunlight. In the eastern United States, high O<sub>3</sub> concentrations during a  
11 large scale episode can extend over hundreds of thousands of square kilometers for several days.  
12 These conditions have been described in greater detail in CD96. The transport of pollutants  
13 downwind of major urban centers is characterized by the development of urban plumes.  
14 However, the presence of mountain barriers limits mixing as in Los Angeles and Mexico City  
15 and will result in a higher frequency and duration of days with high O<sub>3</sub> concentrations. Ozone  
16 concentrations in southern urban areas, such as Houston, TX and Atlanta, GA tend to decrease  
17 with increasing wind speed. In northern cities such as Chicago, IL; New York, NY; Boston,  
18 MA; and Portland, ME, the average O<sub>3</sub> concentrations over the metropolitan areas increase with  
19 wind speed indicating that transport of O<sub>3</sub> and its precursors from upwind areas is important  
20 (Husar and Renard, 1998; Schichtel and Husar, 2001).

21 Ozone and other secondary pollutants are determined by meteorological and chemical  
22 processes extending typically over spatial scales of several hundred kilometers (e.g., Civerolo  
23 et al., 2003; Rao et al., 2003). An analysis of the output of regional model studies conducted by  
24 Kasibhatla and Chameides (2000) suggests that O<sub>3</sub> can be transported over a few thousand  
25 kilometers in the upper boundary layer of the eastern half of the United States during specific O<sub>3</sub>  
26 episodes. Convection is capable of transporting O<sub>3</sub> and its precursors vertically through the  
27 troposphere as shown in Annex AX2.3.2. Nocturnal low level jets (LLJs) can also transport  
28 pollutants hundreds of kilometers (Annex AX2.3.3). Schematic diagrams showing the  
29 atmospheric conditions during the formation of low level jets and the regions in which they are  
30 most prevalent are given in Figures 2-3 and 2-4. They have also been observed off the coast of  
31



**Figure 2-3.** The diurnal evolution of the planetary boundary layer while high pressure prevails over land. Three major layers exist (not including the surface layer): a turbulent mixed layer; a less turbulent residual layer which contains former mixed layer air; and a nocturnal, stable boundary layer that is characterized by periods of sporadic turbulence.

Source: Adapted from Stull (1999) Figures 1.7 and 1.12.



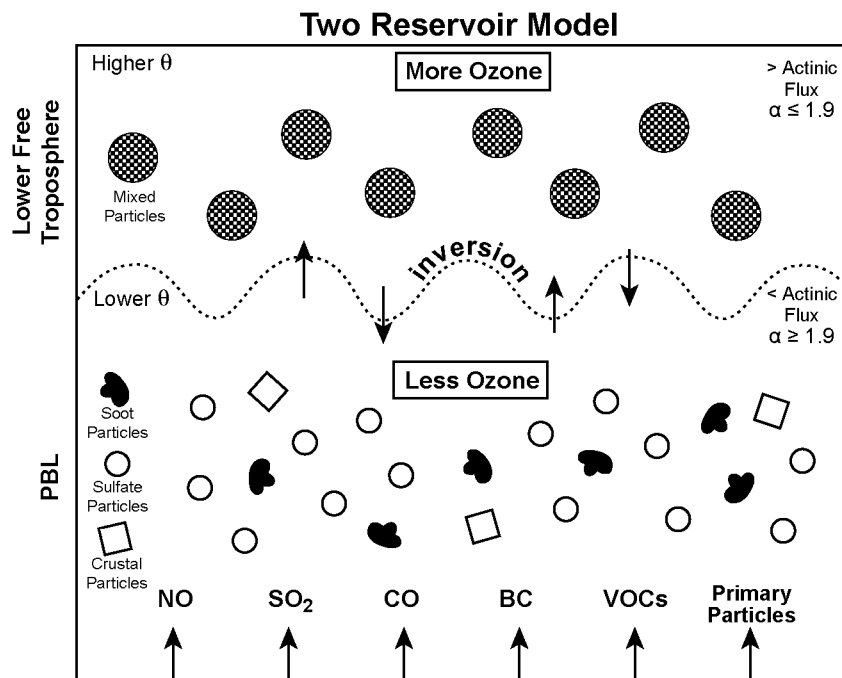
**Figure 2-4.** Locations of low level jet occurrences in decreasing order of prevalence (most frequent, common, observed). These locations are based on 2-years radiosonde data obtained over limited areas. With better data coverage, other low level jets may well be observed elsewhere in the United States.

Source: Bonner (1968).

1 California. Turbulence associated with LLJs can bring these pollutants to the surface and result  
2 in secondary O<sub>3</sub> maxima in the early morning in many locations (Corsmeier et al., 1997).

3 Aircraft observations indicate that there can be substantial differences in mixing ratios of  
4 key species between the surface and the atmosphere above (Fehsenfeld et al., 1996; Berkowitz  
5 and Shaw, 1997). In particular, mixing ratios of O<sub>3</sub> can be higher in the lower free troposphere  
6 (aloft) than in the planetary boundary layer during multiday O<sub>3</sub> episodes (Taubmann et al.,  
7 2004). These conditions are illustrated schematically in Figure 2-5. Convective processes and  
8 small scale turbulence transport O<sub>3</sub> and other pollutants both upward and downward throughout  
9 the planetary boundary layer and the free troposphere. Ozone and its precursors can be  
10 transported vertically by convection into the upper part of the mixed layer on one day, then  
11 transported overnight as a layer of elevated mixing ratios, and then entrained into a growing  
12 convective boundary layer downwind and brought back down to the surface. High  
13 concentrations of O<sub>3</sub> showing large diurnal variations at the surface in southern New England  
14 were associated with the presence of such layers (Berkowitz et al., 1998). Because of wind  
15 shear, winds several hundred meters above the ground can bring pollutants from the west, even  
16 though surface winds are from the southwest during periods of high O<sub>3</sub> in the eastern United  
17 States (Blumenthal et al., 1997). These considerations suggest that in many areas of the United  
18 States, O<sub>3</sub> formation involves processes occurring over hundreds if not thousands of square  
19 kilometers.

20 Although the vast majority of measurements are made near the Earth's surface, there is  
21 substantial photochemistry and transport of O<sub>3</sub> occurring above the boundary layer in the free  
22 troposphere. In the free troposphere, pollutants are chemically more stable and can be  
23 transported over much longer distances and O<sub>3</sub> is produced more efficiently than in the planetary  
24 boundary layer. Results from the Atmosphere/Ocean Chemistry Experiment (AEROCE)  
25 indicated that springtime maxima in surface O<sub>3</sub> over the western North Atlantic Ocean result  
26 from tropopause folding in close proximity to convective clouds (Annex AX2.3.4). The  
27 convection lifts O<sub>3</sub> and its precursors to the free troposphere where they mix with O<sub>3</sub> from the  
28 stratosphere and the mixture is transported eastward. Results from the North Atlantic Regional  
29 Experiment (Annex AX2.3.4) indicated that summertime air is transported along the East Coast  
30 northeastward and upward ahead of cold fronts. New England and the Maritime Provinces of  
31 Canada receive substantial amounts of O<sub>3</sub> and other pollutants through this mechanism.



**Figure 2-5. Conceptual two-reservoir model showing conditions in the PBL and in the lower free troposphere during a multiday O<sub>3</sub> episode. The dotted line represents the top of PBL. Emissions occur in the PBL, where small, unmixed black carbon, sulfate, and crustal particles in the PM<sub>2.5</sub> size range are also shown. Ozone concentrations as well as potential temperature ( $\theta$ ) and actinic flux are lower in the PBL than in the lower free troposphere, while relative humidity and the Angstrom exponent for aerosol scattering ( $\alpha$ ) are higher. Larger, internally mixed sulfate and carbonaceous particles (still in the PM<sub>2.5</sub> size range) and more O<sub>3</sub> exist in the lower free troposphere.**

Source: Taubman et al. (2004).

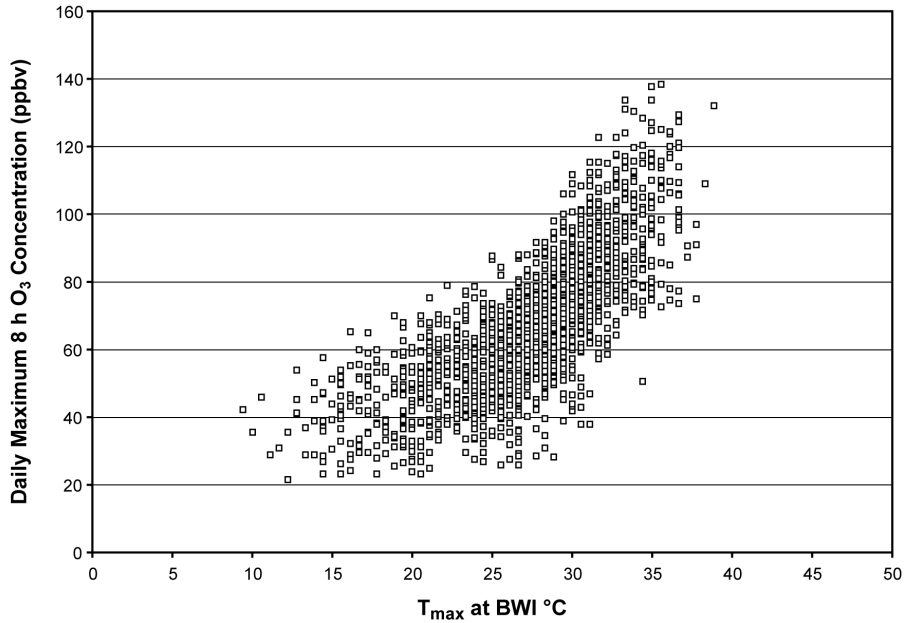
1 Pollutants transported in this way can then be entrained in stronger and more stable westerly  
 2 winds aloft and can travel across the North Atlantic Ocean. The pollutants can then be brought  
 3 to the surface by subsidence in high pressure systems (typically behind the cold front in advance  
 4 of the one mentioned above). Thus, pollutants from North America can be brought down either  
 5 over the North Atlantic Ocean or in Europe. Pollutants can be transported across the North  
 6 Pacific Ocean from Asia to North America in a similar way. Behind an advancing cold front,  
 7 cold and dry stratospheric air is also being transported downward and southward. Stratospheric

1 constituents and tropospheric constituents can then mix by small-scale turbulent exchange  
2 processes. The results of these studies suggest that the mechanisms involved in the long-range  
3 transport of O<sub>3</sub> and its precursors are closely tied to the processes involved in stratospheric-  
4 tropospheric exchange.  
5  
6

## 7 **2.4 RELATIONS OF OZONE TO ITS PRECURSORS**

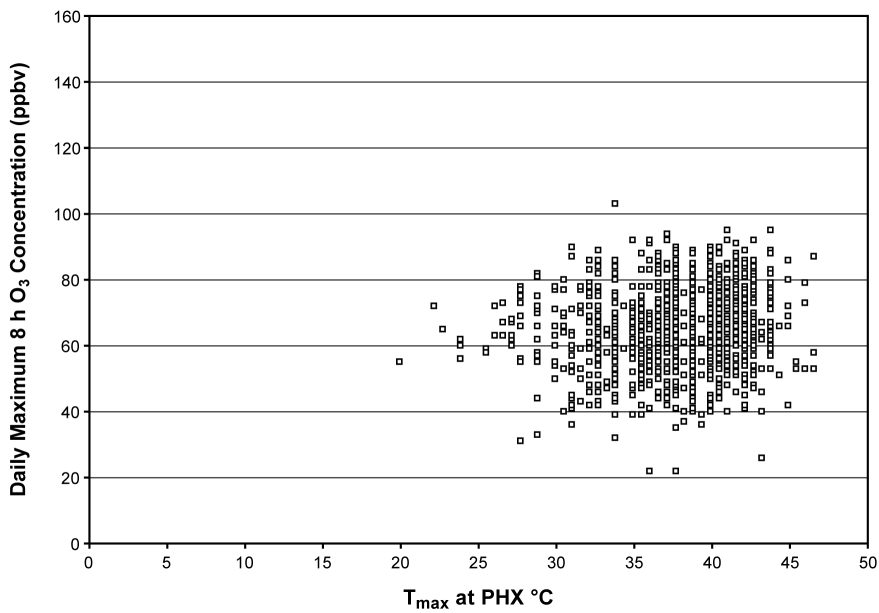
8 The local rate of O<sub>3</sub> formation depends on atmospheric conditions such as the availability  
9 of solar ultraviolet radiation capable of initiating photolysis reactions, air temperatures and the  
10 concentrations of chemical precursors (Annex AX2.3.5). The dependence of daily maximum  
11 8-h O<sub>3</sub> concentrations on daily maximum temperature is illustrated in Figure 2-6 for the  
12 Baltimore, MD area. As can be seen, O<sub>3</sub> concentrations tend to increase with temperature  
13 ( $r = 0.74$ ). However, this trend is absent in data from Phoenix, AZ as can be seen in Figure 2-7  
14 ( $r = 0.14$ ). These figures show that relations of O<sub>3</sub> to precursor variables are location-specific  
15 and relations observed in one area cannot be readily extrapolated to another. Factors that may be  
16 responsible for the differences in O<sub>3</sub> behavior in the two areas are discussed in Section  
17 AX2.3.5.3.

18 Rather than varying directly with emissions of its precursors, O<sub>3</sub> changes in a nonlinear  
19 fashion with the concentrations of its precursors (Annex AX2.4). At the low NO<sub>x</sub> concentrations  
20 found in most environments, ranging from remote continental areas to rural and suburban areas  
21 downwind of urban centers (low - NO<sub>x</sub> regime), the net production of O<sub>3</sub> increases with  
22 increasing NO<sub>x</sub>. At the high NO<sub>x</sub> concentrations found in downtown metropolitan areas,  
23 especially near busy streets and roadways, and in power plant plumes there is scavenging  
24 (titration) of O<sub>3</sub> by reaction with NO (high - NO<sub>x</sub> regime). In between these two regimes there is  
25 a transition stage in which O<sub>3</sub> shows only a weak dependence on NO<sub>x</sub> concentrations. In the  
26 high - NO<sub>x</sub> regime, NO<sub>2</sub> scavenges OH radicals which would otherwise oxidize VOCs to  
27 produce peroxy radicals, which in turn would oxidize NO to NO<sub>2</sub>. In this regime, O<sub>3</sub> production  
28 is limited by the availability of free radicals. The production of free radicals is in turn limited by  
29 the availability of solar UV radiation capable of photolyzing O<sub>3</sub> (in the Hartley bands) or  
30 aldehydes and/or by the abundance of VOCs whose oxidation produce more radicals than they  
31 consume. In the low-NO<sub>x</sub> regime, the overall effect of the oxidation of VOCs is to generate (or



**Figure 2-6. A scatter plot of daily maximum 8-h average O<sub>3</sub> concentrations versus daily maximum temperature for May through September 1994 to 2004 in the Baltimore, MD Air Quality Forecast Area.**

Source: Piety (2005).



**Figure 2-7. A scatter plot of daily maximum 8-h average O<sub>3</sub> concentrations versus daily maximum temperature for May through September 1996 to 2004 at sites downwind of Phoenix, AZ.**

Source: Piety (2005).

1 at least not consume) free radicals, and O<sub>3</sub> production varies directly with NO<sub>x</sub>. There are a  
2 number of ways to refer to the chemistry in these two chemical regimes. Sometimes the terms  
3 VOC-limited and NO<sub>x</sub>-limited are used. However, there are difficulties with this usage because  
4 (1) VOC measurements are not as abundant as they are for nitrogen oxides, (2) rate coefficients  
5 for reaction of individual VOCs with free radicals vary over an extremely wide range, and (3)  
6 consideration is not given to CO nor to reactions that can produce free radicals without involving  
7 VOCs. The terms NO<sub>x</sub>-limited and NO<sub>x</sub>-saturated (e.g., Jaeglé et al., 2001) will be used  
8 wherever possible to more adequately describe these two regimes. However, the terminology  
9 used in original articles will also be used here.

10 The chemistry of OH radicals, which are responsible for initiating the oxidation of  
11 hydrocarbons, shows behavior similar to that for O<sub>3</sub> with respect to NO<sub>x</sub> concentrations (Hameed  
12 et al., 1979; Pinto et al., 1993; Poppe et al., 1993; Zimmerman and Poppe, 1993). These  
13 considerations introduce a high degree of uncertainty into attempts to relate changes in O<sub>3</sub>  
14 concentrations to emissions of precursors. There are no definitive rules governing the levels  
15 of NO<sub>x</sub> at which the transition from NO<sub>x</sub>-limited to NO<sub>x</sub>-saturated conditions occurs. The  
16 transition between these two regimes is highly spatially and temporally dependent and depends  
17 also on the nature and abundance of the hydrocarbons that are present.

18 Trainer et al. (1993) and Olszyna et al. (1994) have shown that O<sub>3</sub> and NO<sub>y</sub> are highly  
19 correlated in rural areas in the eastern United States. Trainer et al. (1993) also showed that O<sub>3</sub>  
20 levels correlate even better with NO<sub>z</sub> than with NO<sub>y</sub>, as may be expected because NO<sub>z</sub> represents  
21 the amount of NO<sub>x</sub> that has been oxidized, forming O<sub>3</sub> in the process. NO<sub>z</sub> is equal to the  
22 difference between measured total reactive nitrogen (NO<sub>y</sub>) and NO<sub>x</sub> and represents the summed  
23 products of the oxidation of NO<sub>x</sub>. NO<sub>z</sub> is composed mainly of HNO<sub>3</sub>, PAN and other organic  
24 nitrates, particulate nitrate, and HNO<sub>4</sub>.

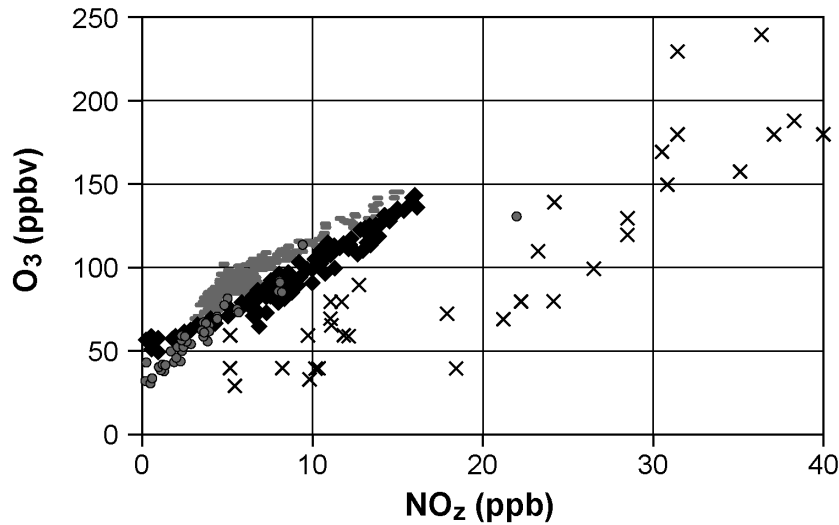
25 Trainer et al. (1993) also suggested that the slope of the regression line between O<sub>3</sub>  
26 and NO<sub>z</sub> can be used to estimate the rate of O<sub>3</sub> production per NO<sub>x</sub> oxidized (also known as  
27 the O<sub>3</sub> production efficiency, or OPE). Ryerson et al. (1998, 2001) used measured correlations  
28 between O<sub>3</sub> and NO<sub>z</sub> to identify different rates of O<sub>3</sub> production in plumes from large point  
29 sources. A number of studies in the planetary boundary layer over the continental United States  
30 have found that the OPE ranges typically from one to nearly ten. However, it may be higher in

1 the upper troposphere and in certain areas, such as the Houston-Galveston area. Observations  
2 indicate that the OPE depends mainly on the abundance of  $\text{NO}_x$ .

3 Various techniques have been proposed to use ambient  $\text{NO}_x$  and VOC measurements to  
4 derive information about the dependence of  $\text{O}_3$  production on their concentrations. For example,  
5 it has been suggested that  $\text{O}_3$  formation in individual urban areas could be understood in terms of  
6 measurements of ambient  $\text{NO}_x$  and VOC concentrations during the early morning (e.g., National  
7 Research Council, 1991). In this approach, the ratio of summed (unweighted) VOC to  $\text{NO}_x$  is  
8 used to determine whether conditions were  $\text{NO}_x$ -limited or VOC limited. This procedure is  
9 inadequate because it omits many factors that are important for  $\text{O}_3$  production such as the impact  
10 of biogenic VOCs (which are typically not present in urban centers during early morning);  
11 important differences in the ability of individual VOCs to generate free radicals (rather than just  
12 total VOC) and other differences in  $\text{O}_3$  forming potential for individual VOCs (Carter et al.,  
13 1995); and changes in the VOC to  $\text{NO}_x$  ratio due to photochemical reactions and deposition as  
14 air moves downwind from urban areas (Milford et al., 1994).

15 Photochemical production of  $\text{O}_3$  generally occurs simultaneously with the production of  
16 various other species such as nitric acid ( $\text{HNO}_3$ ), organic nitrates, and other oxidants such as  
17 hydrogen peroxide. The relative rate of production of  $\text{O}_3$  and other species varies depending on  
18 photochemical conditions, and can be used to provide information about  $\text{O}_3$ -precursor  
19 sensitivity. Sillman (1995) and Sillman and He (2002) identified several secondary reaction  
20 products that show different correlation patterns for  $\text{NO}_x$ -limited and  $\text{NO}_x$ -saturated conditions.  
21 The most important correlations are for  $\text{O}_3$  versus  $\text{NO}_y$ ,  $\text{O}_3$  versus  $\text{NO}_z$ ,  $\text{O}_3$  versus  $\text{HNO}_3$ ,  
22 and  $\text{H}_2\text{O}_2$  versus  $\text{HNO}_3$ . The correlations between  $\text{O}_3$  and  $\text{NO}_y$ , and  $\text{O}_3$  and  $\text{NO}_z$  are especially  
23 important because measurements of  $\text{NO}_y$  and  $\text{NO}_x$  are more widely available than for VOCs.  
24 Measured  $\text{O}_3$  versus  $\text{NO}_z$  (Figure 2-8) shows distinctly different patterns in different locations.  
25 In rural areas and in urban areas such as Nashville, TN,  $\text{O}_3$  is highly correlated with  $\text{NO}_z$ . By  
26 contrast, in Los Angeles, CA,  $\text{O}_3$  is not as highly correlated with  $\text{NO}_z$ , and the rate of increase  
27 of  $\text{O}_3$  with  $\text{NO}_z$  is lower and the  $\text{O}_3$  concentrations for a given  $\text{NO}_z$  value are generally lower.  
28 The different  $\text{O}_3$  versus  $\text{NO}_z$  relations in Nashville, TN and Los Angeles, CA reflects the  
29 difference between  $\text{NO}_x$ -limited conditions in Nashville versus an approach to  $\text{NO}_x$ -saturated  
30 conditions in Los Angeles.





**Figure 2-8. Measured values of O<sub>3</sub> and NO<sub>z</sub> (NO<sub>y</sub> – NO<sub>x</sub>) during the afternoon at rural sites in the eastern United States (grey circles) and in urban areas and urban plumes associated with Nashville, TN (gray dashes); Paris, France (black diamonds); and Los Angeles CA (Xs).**

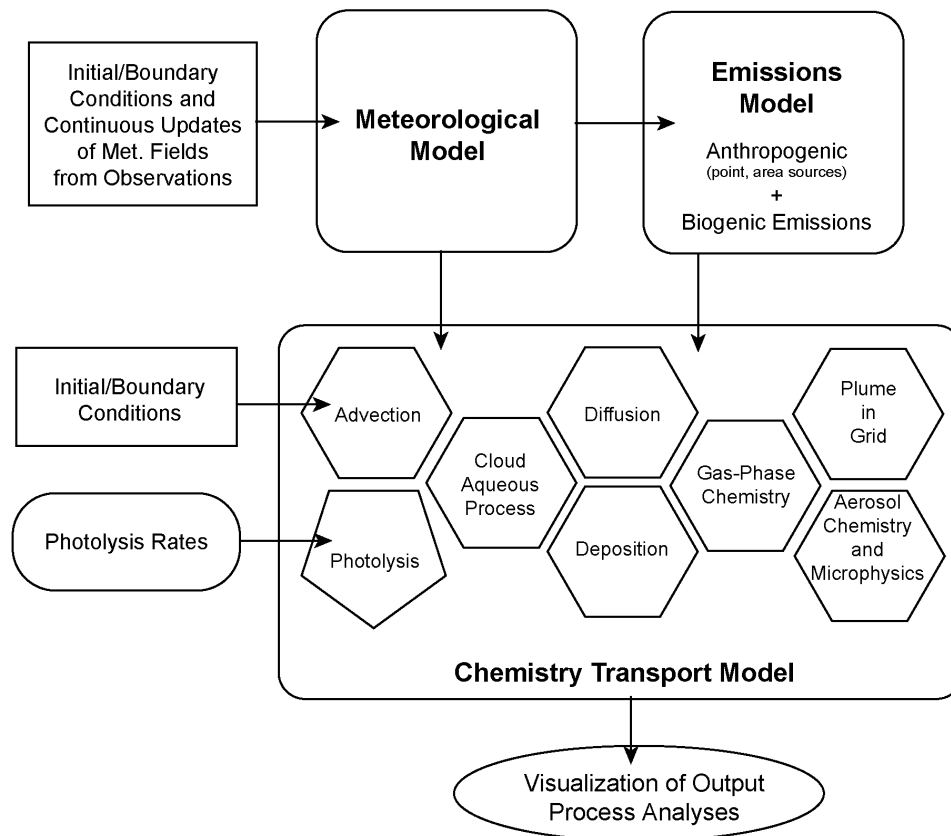
Sources: Trainer et al. (1993), Sillman et al. (1997, 1998), Sillman and He (2002).

1           The difference between NO<sub>x</sub>-limited and NO<sub>x</sub>-saturated regimes is also reflected in  
 2 measurements of hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>). Hydrogen peroxide production is highly sensitive  
 3 to the abundance of free radicals and is thus favored in the NO<sub>x</sub>-limited regime. Measurements  
 4 in the rural eastern United States (Jacob et al., 1995) Nashville, TN (Sillman et al., 1998), and  
 5 Los Angeles, CA (Sakugawa and Kaplan, 1989), show large differences in H<sub>2</sub>O<sub>2</sub> concentrations  
 6 between likely NO<sub>x</sub>-limited and NO<sub>x</sub>-saturated locations.

7  
 8

## 9   **2.5 THE ROLE OF CHEMISTRY-TRANSPORT MODELS IN** 10 **UNDERSTANDING ATMOSPHERIC OZONE**

11           Chemistry-transport models (CTMs) are used to improve understanding of atmospheric  
 12 chemical processes and to develop control strategies (Annex AX2.5). The main components of a  
 13 CTM are summarized in Figure 2-9. Models such as the CMAQ (Community Model for Air  
 14 Quality) system incorporate numerical algorithms describing the processes shown in Figure 2-9.



**Figure 2-9. Main components of a comprehensive atmospheric chemistry modeling system, such as Models-3.**

1 Also shown in Figure 2-9 is the meteorological model used to provide the inputs for calculating  
 2 the transport of species in the CTM. Meteorological models, such as the MM5 model, which  
 3 supply these inputs to the CTMs mentioned above, also provide daily weather forecasts. The  
 4 domains of these models extend typically over areas of millions of square kilometers.

5 Because these models are computationally intensive, it is often impractical to run them  
 6 over larger domains without sacrificing some features. For these reasons, both the  
 7 meteorological model and the CTM rely on boundary conditions that allow processes occurring  
 8 outside the model domain to influence their predictions. The entire system, consisting of  
 9 meteorological model, emissions processor, and output processors shown in Figure 2-9  
 10 constitutes the framework of EPA's Models-3.

1           Because of the large number of chemical species and reactions that are involved in the  
2 oxidation of realistic mixtures of anthropogenic and biogenic hydrocarbons, condensed  
3 mechanisms must be used in atmospheric models. These mechanisms are tested by comparison  
4 with smog chamber data. However, the existing chemical mechanisms often neglect many  
5 important processes such as the formation and subsequent reactions of long-lived carbonyl  
6 compounds, the incorporation of the most recent information about intermediate compounds, and  
7 heterogeneous reactions involving cloud droplets and aerosol particles.

8           Emissions inventories are compiled for O<sub>3</sub> precursors ( NO<sub>x</sub>, VOCs, and CO). Recent  
9 estimates and more detailed discussions of the estimates are given in Annex AX2.5.2.  
10 Anthropogenic NO<sub>x</sub> emissions are associated with combustion processes. Most emissions are in  
11 the form of NO, which is formed at high combustion temperatures from atmospheric nitrogen  
12 and oxygen and from fuel nitrogen. The two largest sources of NO<sub>x</sub> are electric power  
13 generation plants and motor vehicles. Emissions of NO<sub>x</sub> therefore are highest in areas having a  
14 high density of power plants and in urban regions having high traffic density. Natural NO<sub>x</sub>  
15 sources include stratospheric intrusions, lightning, soils, and wildfires. Lightning, fertilized  
16 soils, and wildfires are the major natural sources of NO<sub>x</sub> in the United States. Both nitrifying  
17 and denitrifying organisms in the soil can produce NO<sub>x</sub>, mainly in the form of NO. Emission  
18 rates depend mainly on fertilization levels and soil temperature and moisture. Spatial and  
19 temporal variability in soil NO<sub>x</sub> emissions leads to considerable uncertainty in emissions  
20 estimates. Nationwide, about 60% of lightning generated NO<sub>x</sub> occurs in the southern United  
21 States and about 60% the total NO<sub>x</sub> emitted by soils occurs in the central corn belt of the United  
22 States. The oxidation of NH<sub>3</sub> emitted mainly by livestock and soils, leads to the formation of a  
23 small amount of NO. Uncertainties in natural NO<sub>x</sub> inventories are much larger than for  
24 anthropogenic NO<sub>x</sub> emissions.

25           Hundreds of VOCs, containing mainly two to about twelve carbon atoms, are emitted by  
26 evaporation and combustion processes from a large number of anthropogenic sources. The two  
27 largest source categories in the U.S. EPA's emissions inventories are industrial processes and  
28 transportation. Emissions of VOCs from highway vehicles account for roughly two-thirds of the  
29 transportation-related emissions.

30           The accuracy of VOC emission estimates is difficult to determine, both for stationary and  
31 mobile sources. Evaporative emissions, which depend on temperature and other environmental

1 factors, compound the difficulties of assigning accurate emission factors. In assigning VOC  
2 emission estimates to the mobile source category, models are used that incorporate numerous  
3 input parameters (e.g., type of fuel used, type of emission controls, age of vehicle), each of  
4 which has some degree of uncertainty. Data for the ratio of CO to NO<sub>x</sub> and NMHC to NO<sub>x</sub> in  
5 traffic tunnels (e.g., Pierson et al., 1990) indicated that emissions of NMHCs and CO from motor  
6 vehicles have been underestimated by as much as a factor of two (based on the assumption that  
7 emissions of NO<sub>x</sub> were reasonably well represented in the inventories). However, the results of  
8 more recent studies have been mixed, with many studies showing agreement to within ±50%  
9 (summarized in Air Quality Criteria for Carbon Monoxide [U.S. Environmental Protection  
10 Agency, 2000]). Remote sensing data (Stedman et al., 1991) indicate that about 50% of NMHC  
11 and CO emissions are produced by about 10% of the vehicles. These “super-emitters” are  
12 typically poorly maintained. Vehicles of any age engaged in off-cycle operations (e.g., rapid  
13 accelerations) emit much more than if operated in normal driving modes.

14 Vegetation emits significant quantities of VOCs such as terpenoid compounds (isoprene,  
15 2-methyl-3-buten-2-ol, monoterpenes), compounds in the hexanal family, alkenes, aldehydes,  
16 organic acids, alcohols, ketones, and alkanes. The major chemicals emitted by plants are  
17 isoprene (35%), 19 other terpenoid compounds and 17 non-terpenoid compounds including  
18 oxygenated compounds (40%) (Guenther et al., 2000). Coniferous forests represent the largest  
19 source on a nationwide basis, because of their extensive land coverage. Most biogenic emissions  
20 occur during the summer, because of their dependence on temperature and incident sunlight.  
21 Biogenic emissions are also higher in southern states than in northern states for these reasons and  
22 because of species variations. The uncertainty in natural emissions is about 50% for isoprene  
23 under midday summer conditions and could be as much as a factor of ten higher for some  
24 compounds (Guenther et al., 2000). Uncertainties in both biogenic and anthropogenic VOC  
25 emission inventories prevent determination of the relative contributions of these two categories  
26 at least in many urban areas. On the regional and global scales, emissions of VOCs from  
27 vegetation are much larger than those from anthropogenic sources.

28 The performance of CTMs must be evaluated by comparison with field data as part of a  
29 cycle of model evaluations and subsequent improvements. Discrepancies between model  
30 predictions and observations can be used to point out gaps in current understanding of  
31 atmospheric chemistry and to spur improvements in parameterizations of atmospheric chemical

1 and physical processes. Model evaluation does not merely involve a straightforward comparison  
2 between model predictions and the concentration field of the pollutant of interest. Such  
3 comparisons may not be meaningful because it is difficult to determine if agreement between  
4 model predictions and observations truly represents an accurate treatment of physical and  
5 chemical processes in the CTM or the effects of compensating errors in complex model routines.  
6 Ideally, each of the model components (emissions inventories, chemical mechanism,  
7 meteorological driver) should be evaluated individually, however this is rarely done in practice.  
8 A comparison between free radical concentrations predicted by parameterized chemical  
9 mechanisms and observations suggests that radical concentrations were overestimated by current  
10 chemical mechanisms for  $\text{NO}_x$  concentrations  $< \sim 5$  ppb (Volz-Thomas et al., 2003).

11 In addition to comparisons between concentrations of calculated and measured species,  
12 comparisons of correlations between measured primary VOCs and  $\text{NO}_x$  and modeled VOCs  
13 and  $\text{NO}_x$  are especially useful for evaluating results from chemistry-transport models. Likewise,  
14 comparisons of correlations between measured species and modeled species can be used to  
15 provide information about the chemical state of the atmosphere and to evaluate model  
16 representations (including  $\text{O}_3$  production per  $\text{NO}_x$ ,  $\text{O}_3$ - $\text{NO}_x$ -VOC sensitivity, and the general  
17 accuracy of photochemical representations). A CTM that demonstrates the accuracy of both its  
18 computed VOC and  $\text{NO}_x$  in comparison with ambient measurements and the spatial and temporal  
19 relations among the critical secondary species associated with  $\text{O}_3$  has a higher probability of  
20 representing  $\text{O}_3$ -precursor relations correctly than one that does not.

## 23 **2.6 TECHNIQUES FOR MEASURING OZONE AND ITS PRECURSORS**

24 Several techniques have been developed for sampling and measurement of  $\text{O}_3$  in the  
25 ambient atmosphere at ground level. Although the chemiluminescence method (CLM) using  
26 ethylene is designated as the Federal Reference Method for measuring  $\text{O}_3$ , monitoring in the  
27 NAMS/SLAMS networks is conducted mainly with UV absorption spectrometry using  
28 commercial short path instruments. The primary reference standard instrument is a relatively  
29 long-path UV absorption spectrometer maintained under carefully controlled conditions at NIST  
30 (e.g., Fried and Hodgeson, 1982). Episodic measurements are made with a variety of other

1 techniques based on the principles of chemiluminescence, electrochemistry, differential optical  
2 absorption spectroscopy (DOAS), and LIDAR.

3 In principle, each of these methods is subject to interference. Kleindienst et al. (1993)  
4 found that water vapor could cause a positive interference in the CLM with an average positive  
5 deviation of 3% ozone/% water vapor at 25 °C. The UV absorption spectrometers are subject to  
6 positive interference by atmospheric constituents, such as certain aromatic aldehydes that absorb  
7 at the 253.7 nm Hg resonance line and are at least partially removed by the MnO<sub>2</sub> scrubber.  
8 Parrish and Fehsenfeld (2000) did not find any evidence for significant interference (>1%) in  
9 flights through the Nashville urban plume. The same group tested the air of Houston, El Paso,  
10 Nashville, Los Angeles, San Francisco and the East Coast. They observed only one instance of  
11 substantive positive interference defined as the UV absorption technique showing more than a  
12 few ppb more than the CLM. This occurred in Laporte, TX under heavily polluted conditions  
13 and a low inversion, at night (Jobson et al., 2004). Leston et al. (2005) observed interference of  
14 from 20 to 40 ppb in Mexico City and in a separate smog chamber study. However, the  
15 concentrations of relevant compounds were many times higher than found in U.S. urban areas.  
16 Thus, it is not likely that such interference could be more than a few ppb under typical ambient  
17 conditions. However, Leston et al. (2005) suggested that the use of other materials in the  
18 scrubber could have eliminated the interference seen in their smog chamber study.

19 By far, most measurements of NO are made using the CLM, based on its reaction with O<sub>3</sub>.  
20 Commercial instruments for measuring NO and NO<sub>2</sub> are constructed with an internal converter  
21 for reducing NO<sub>2</sub> to NO and then measuring NO by the CLM. In principle, this technique yields  
22 a measurement of NO<sub>x</sub> with NO<sub>2</sub> found by difference between NO<sub>x</sub> and NO. However, these  
23 converters also reduce NO<sub>z</sub> compounds thereby introducing a positive interference in the  
24 measurement of NO<sub>2</sub>. Other methods for measuring NO<sub>2</sub> are available, such as photolytic  
25 reduction followed by CLM, laser-induced fluorescence and DOAS. However, they require  
26 further development before they can be used for routine monitoring in the NAMS/SLAMS  
27 networks. More detailed descriptions of the issues and techniques discussed above and  
28 techniques for measuring HNO<sub>3</sub> and VOCs can be found in Annex AX2.6.

## 2.7 SUMMARY

Ozone ( $O_3$ ) is formed by atmospheric reactions involving two classes of precursor compounds, volatile organic compounds (VOCs) and nitrogen oxides ( $NO_x$ ). Ozone is thus a secondary pollutant. Ozone is ubiquitous throughout the atmosphere; it is present even in remote areas of the globe. The photochemical oxidation of almost all anthropogenic and biogenic VOCs is initiated by reaction with hydroxyl (OH) radicals. At night, when they are most abundant,  $NO_3$  radicals also oxidize alkenes. In coastal and other select environments, Cl and Br radicals can also initiate the oxidation of VOCs.

In urban areas, basically all classes of VOCs (alkanes, alkenes, aromatic hydrocarbons, carbonyl compounds, etc.) and CO are important for  $O_3$  formation. Although knowledge of the oxidative mechanisms of VOCs has improved over the past several years, gaps in knowledge involving key classes, such as aromatic hydrocarbons, still remain. For example, only about half of the carbon initially present in aromatic hydrocarbons in smog chamber studies form compounds that can be identified.

In addition to gas phase reactions, reactions also occur on the surfaces of, or within cloud droplets and airborne particles. Most of the well-established multiphase reactions tend to reduce the rate of  $O_3$  formation in polluted environments. Reactions of Cl and Br containing radicals deplete  $O_3$  in selected environments such as the Arctic during spring, the tropical marine boundary layer and inland salt lakes. Direct reactions of  $O_3$  with atmospheric particles appear to be too slow to reduce  $O_3$  formation significantly at typical ambient PM levels.

Our basic understanding of the meteorological processes associated with summertime  $O_3$  episodes has not changed over the past several years. However, the realization that long-range transport processes are important for determining  $O_3$  concentrations at the surface is growing. In addition to synoptic scale flow fields, nocturnal low-level jets are capable of transporting pollutants hundreds of km from their sources in either the upper boundary layer or the lower free troposphere. Turbulence then brings  $O_3$  and other pollutants to the surface. On larger scales, important progress has been made in identifying the mechanisms of intercontinental transport of  $O_3$  and other pollutants.

Some  $O_3$  would be found near the earth's surface as the result of its downward transport from the stratosphere, even in the absence of photochemical reactions in the troposphere. Intrusions of stratospheric  $O_3$  that reach the surface are rare. Much more common are intrusions

1 that penetrate to the middle and upper troposphere. However, O<sub>3</sub> transported to the middle and  
2 upper troposphere can still affect surface concentrations through various mechanisms that mix  
3 air between the planetary boundary layer and the free troposphere above.

4 The formation of O<sub>3</sub> and associated compounds is a complex, nonlinear function of many  
5 factors including the intensity and spectral distribution of sunlight; atmospheric mixing and other  
6 atmospheric processes; and the concentrations of the precursors in ambient air. At the  
7 lower NO<sub>x</sub> concentrations found in most environments, ranging from remote continental areas to  
8 rural and suburban areas downwind of urban centers, the net production of O<sub>3</sub> increases with  
9 increasing NO<sub>x</sub>. At the higher concentrations found in downtown metropolitan areas, especially  
10 near busy streets and highways, and in power plant plumes there is net destruction of O<sub>3</sub> by  
11 reaction with NO. In between these two regimes there is a transition stage, in which O<sub>3</sub>  
12 production shows only a weak dependence on NO<sub>x</sub> concentrations. The efficiency of O<sub>3</sub>  
13 production per NO<sub>x</sub> oxidized is generally highest in areas where NO<sub>x</sub> concentrations are lowest  
14 and decrease with increasing NO<sub>x</sub> concentration.

15 Chemistry transport models are used to improve understanding of atmospheric chemical  
16 and physical processes as well as to develop air pollution control strategies. The performance of  
17 these models must be evaluated by comparison with field data as part of a cycle of model  
18 evaluations and subsequent improvements. Discrepancies between model predictions and  
19 observations can be used to point out gaps in current understanding and thus to improve  
20 parameterizations of atmospheric chemical and physical processes. Model evaluation does not  
21 merely involve a straightforward comparison between model predictions and the concentration  
22 fields of a pollutant of interest (e.g., O<sub>3</sub>). Such comparisons may not be meaningful because it is  
23 difficult to determine if agreement between measurements and model predictions truly represents  
24 an accurate treatment of physical and chemical processes in the model or the effects of  
25 compensating errors in model routines.

26 The main methods in use for routine monitoring of ambient O<sub>3</sub> are based on  
27 chemiluminescence or UV absorption. Measurements at most ambient monitoring sites are  
28 based on UV absorption. Both of these methods are subject to interference by other atmospheric  
29 components. One study found large positive interference in Mexico City and in a smog  
30 chamber, but few studies conducted in urban plumes did not find significant positive interference  
31 in the UV absorption technique.



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5

# 3. ENVIRONMENTAL CONCENTRATIONS, PATTERNS, AND EXPOSURE ESTIMATES

## 3.1 INTRODUCTION

### *Identification and Use of Existing Air Quality Data*

Topics discussed in this chapter include the characterization of ambient air quality data for ozone ( $O_3$ ), the uses of these data in assessing the exposure of vegetation to  $O_3$ , concentrations of  $O_3$  in microenvironments, and a discussion of the currently available human exposure data and exposure model development. The information contained in this chapter pertaining to ambient concentrations is taken primarily from the U.S. Environmental Protection Agency (EPA) Air Quality System (AQS; formerly the AIRS database). The AQS contains readily accessible detailed, hourly data that has been subject to EPA quality control and assurance procedures. Data available in AQS were collected from 1979 to 2001. As discussed in previous versions of the  $O_3$  Air Quality Criteria Document or AQCD (U.S. Environmental Protection Agency, 1986, 1996), the data available prior to 1979 may be unreliable due to calibration problems and uncertainties.

As noted in the 1996  $O_3$  AQCD (U.S. Environmental Protection Agency, 1996),  $O_3$  is the only photochemical oxidant other than nitrogen dioxide ( $NO_2$ ) that is routinely monitored and for which a comprehensive database exists. Data for peroxyacetyl nitrate (PAN), hydrogen peroxide ( $H_2O_2$ ), and other oxidants either in the gas phase or particle phase typically have been obtained only as part of special field studies. Consequently, no data on nationwide patterns of occurrence are available for these non- $O_3$  oxidants; nor are extensive data available on the relationships of levels and patterns of these oxidants to those of  $O_3$ . However, available data for gas phase and particle phase oxidants will be discussed.

### *Characterizing Ambient Ozone Concentrations*

The “concentration” of a specific air pollutant is typically defined as the amount (mass) of that material per unit volume of air. However, most of the data presented in this chapter are expressed as “mixing ratios” in terms of a volume-to-volume ratio (parts per million [ppm] or parts per billion [ppb]). Data expressed this way are often referred to as concentrations, both in

1 the literature and in the text, following common usage. Human exposures are expressed in units  
2 of mixing ratio times time.

3 Several different types of indicators are used for evaluating exposures of vegetation to O<sub>3</sub>.  
4 The peak-weighted, cumulative exposure indicators used in this chapter for characterizing  
5 vegetation exposures are SUM06 and SUM08 (the sums of all hourly average concentrations  
6 ≥0.06 and 0.08 ppm, respectively) and W126 (the sum of the hourly average concentrations that  
7 have been weighted according to a sigmoid function that is based on a hypothetical vegetation  
8 response [see Lefohn and Runeckles, 1987]). Further discussion of these exposure indices is  
9 presented in Chapter 9.

10 The EPA has established “ozone seasons” during which measurement of ambient O<sub>3</sub>  
11 concentrations for different locations within the United States and the U.S. territories is required  
12 (CFR, 2000). Table AX3-1 shows the O<sub>3</sub> seasons during which continuous, hourly averaged O<sub>3</sub>  
13 concentrations must be monitored. Monitoring is optional outside of these O<sub>3</sub> seasons and  
14 indeed is conducted during the winter in a number of areas.

15 Data for O<sub>3</sub> in ambient air across the United States are summarized in Section 3.2. The  
16 data are summarized for urban, rural, and relatively remote sites. Relatively remote monitoring  
17 sites (RRMS) are sites that are not strongly influenced by nearby pollution sources and are  
18 located mainly in national parks in the West. However, this does not mean that they are free of  
19 the effects of regional or local pollution, especially during tourist seasons. Data for the spatial  
20 variability of O<sub>3</sub> within urban areas are summarized in Section 3.3. Data for the diurnal and  
21 seasonal variability of O<sub>3</sub> concentrations are given in Section 3.4. The long term temporal  
22 variability of O<sub>3</sub> concentrations is discussed in Section 3.5. Relationships among O<sub>3</sub> and other  
23 species are discussed in Section 3.6. Information about the occurrence of other oxidants and  
24 their relationship to O<sub>3</sub> is given in this section. A discussion of Policy Relevant Background  
25 (PRB) O<sub>3</sub> concentrations is presented in Section 3.7. PRB O<sub>3</sub> concentrations are background O<sub>3</sub>  
26 concentrations used for the purposes of setting the O<sub>3</sub> NAAQS. They are used by the EPA to  
27 assess risks to human health. Indoor sources and emissions of O<sub>3</sub> are discussed in Section 3.8.  
28 Issues related to evaluating human exposure to O<sub>3</sub> are summarized in Section 3.9. Finally, a  
29 summary of key points in Chapter 3 is given in Section 3.10.

## 3.2 AMBIENT AIR QUALITY DATA FOR OZONE

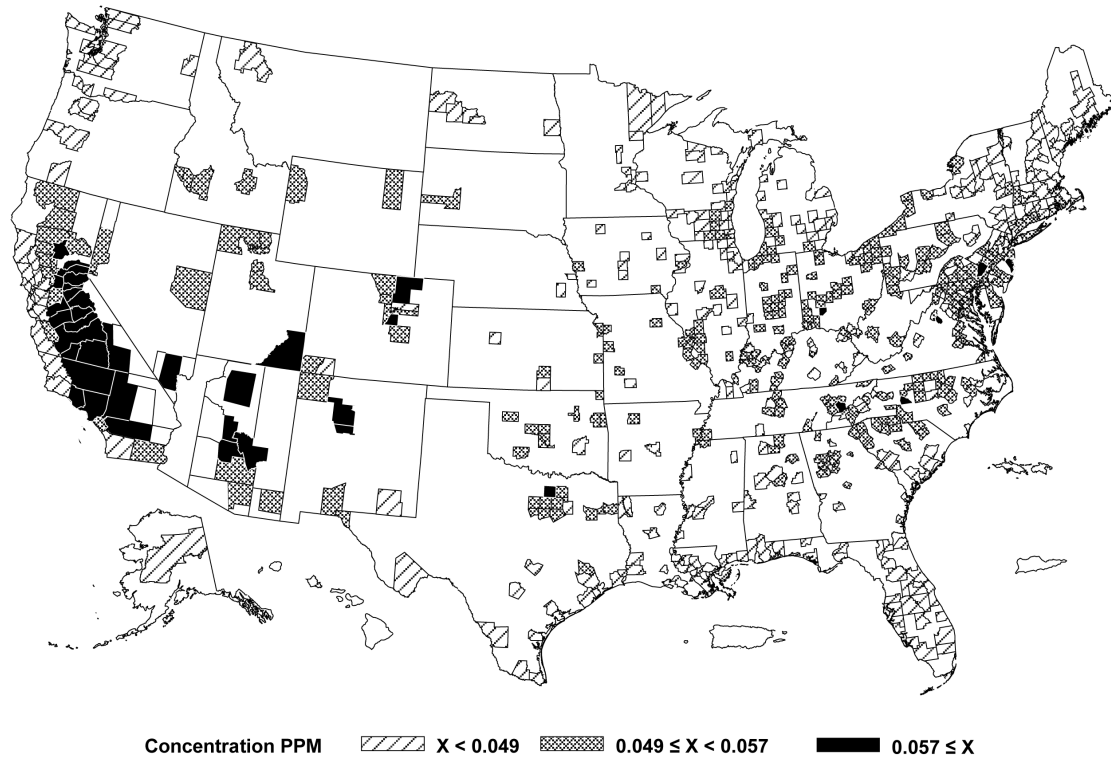
### *Ozone Air Quality at Urban, Suburban, and Nonurban Sites*

Figure 3-1 shows the mean daily maximum 8-h O<sub>3</sub> concentrations and Figure 3-2 shows the 95th percentile values of the daily maximum 8-h O<sub>3</sub> concentrations, based on countywide averages across the United States from May to September 2000 to 2004. The period from May to September was chosen because, although O<sub>3</sub> is monitored for different lengths of time across the country, all O<sub>3</sub> monitors should be operational during these months. Data flagged because of quality control issues were removed with concurrence by the local monitoring agency. Only days with data for 18 of 24 hours were kept, and a minimum of 115 of 153 days were required in each year. Cut points for the tertile distributions on each map were chosen at the median and 95th percentile values. These cut points were chosen as they represent standard metrics for characterizing important aspects of human exposure used by the EPA. Any other percentiles or statistics that are believed to be helpful for characterizing human exposures could also be used. Blank areas on the maps indicate no data coverage. It should be noted that county areas can be much larger in the West than in the East, but monitors are not spread evenly within a county. As a result, the assigned concentration range might not represent conditions throughout a particular county and so large areas in western counties where there are no monitors were blanked out.

As shown in Figure 3-1, the median of the countywide, mean daily maximum 8-h O<sub>3</sub> concentration across the United States is 49 ppb, and the corresponding 95th percentile value is 57 ppb. Though the median and 95th percentile values are fairly close, these results cannot be taken to imply that average O<sub>3</sub> concentrations lie in a relatively narrow range throughout the United States, because data coverage is not as complete in the West as it is in the East. High mean daily maximum 8-h O<sub>3</sub> concentrations are found in California and states in the Southwest as well as in several counties in the East. As shown in Figure 3-2, the nationwide median of the countywide, 95th percentile value of the daily maximum 8-h O<sub>3</sub> concentration is 73 ppb and 5% of these values are above 85 ppb. High values for the 95th percentiles are found in California, Texas, and some counties in the East, but not necessarily in the same counties in the East as shown for the mean daily maximum 8-h concentrations in Figure 3-1.

Although mean O<sub>3</sub> concentrations in Houston, TX were below the nationwide median, its 95th percentile value ranks in the highest 5% nationwide. Conversely, mean O<sub>3</sub> concentrations in southwestern states are among the highest in the United States, but values at the upper end of

**Seasonal (May-September) Mean of Daily Maximum 8-Hour Values, 2002-2004**



**Figure 3-1. Countywide mean daily maximum 8-h O<sub>3</sub> concentrations, May to September 2000 to 2004.**

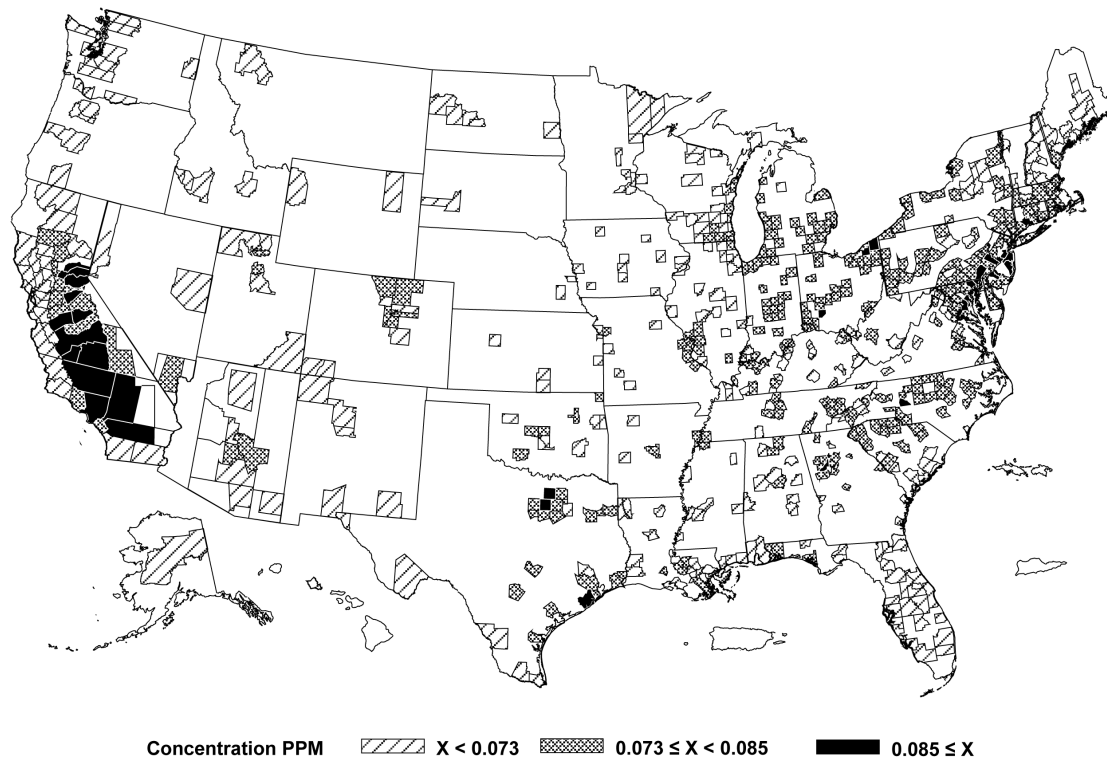
Source: Fitz-Simons et al. (2005).

1 the distribution (e.g., the 95th percentile value) in these states are not among the highest peak  
2 values in the United States. In other areas where the highest mean O<sub>3</sub> concentrations occurred,  
3 such as California; Dallas-Fort Worth, TX; and the Northeast Corridor, the highest peak values  
4 were also observed.

5 Although countywide averages are shown, it should be noted that considerable spatial  
6 variability can exist within a county, especially within urban areas as described in Section 3.3.  
7 In addition, there can also be differences in the diurnal profile of O<sub>3</sub> among monitors within  
8 counties.

9 Box plots showing the percentile distribution of nationwide O<sub>3</sub> concentrations for different  
10 averaging periods (1-h daily maximum, 8-h daily maximum and 24-h daily average) are given in

**Seasonal (May-September) 95th Percentile of Daily Maximum 8-Hour Values, 2002-2004**



**Figure 3-2. Countywide 95th percentile value of daily maximum 8-h O<sub>3</sub> concentrations, May to September 2000 to 2004.**

Source: Fitz-Simons et al. (2005).

1 Figures AX3-4 to AX3-6 and numerical values are given in Table AX3-2. The differences  
2 between the 50th and 95th percentile values can be used to provide indications of differences  
3 in O<sub>3</sub> levels between “typical” O<sub>3</sub> days and “high” O<sub>3</sub> days. These differences are approximately  
4 40, 30, and 25 ppb for the daily 1-h and 8-h daily maxima and 24-h average O<sub>3</sub> concentrations.  
5 As might be expected, the daily maximum 1-h and 8-h O<sub>3</sub> concentrations are highly correlated.

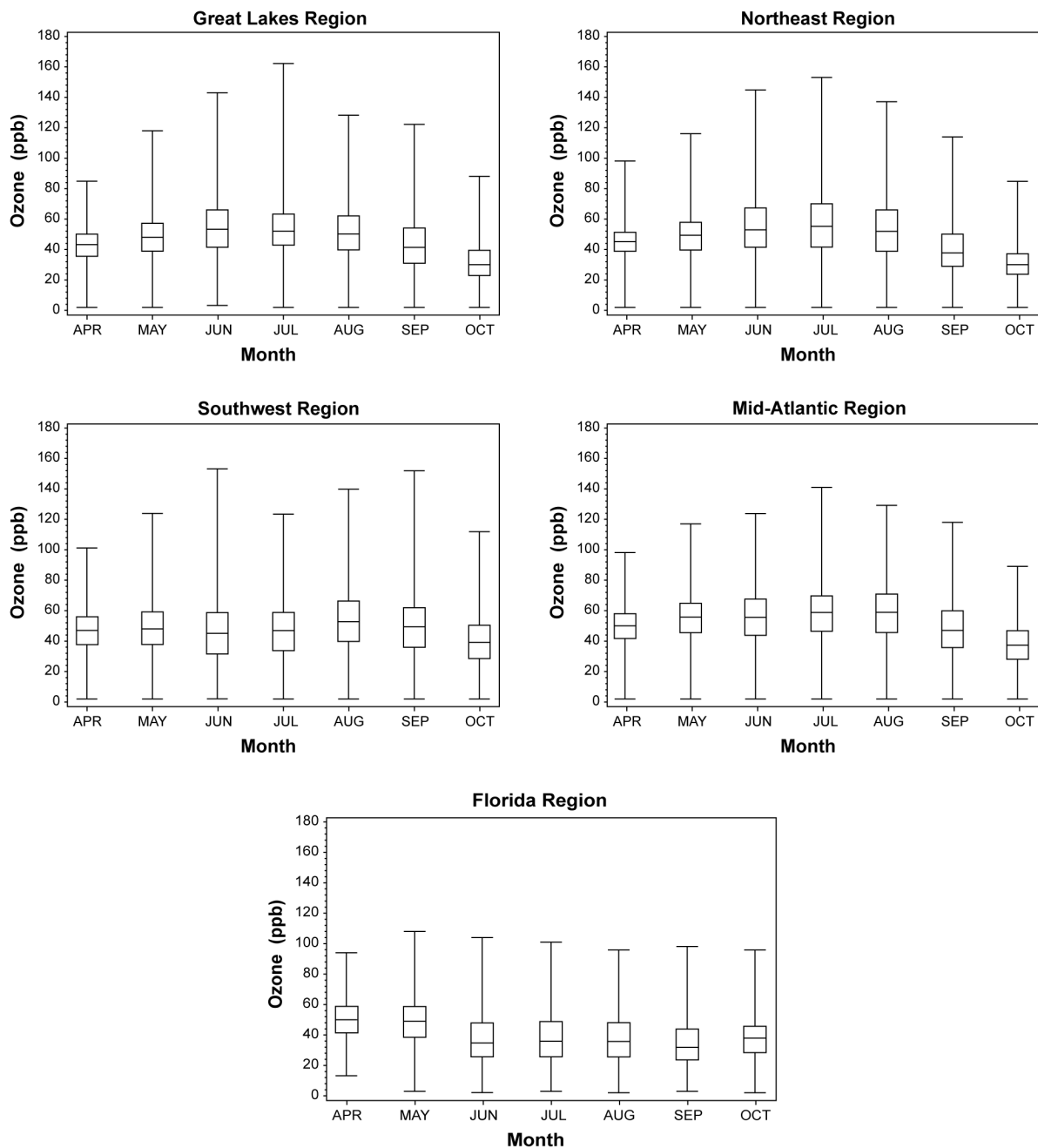
6 Lehman et al. (2004) have shown that the eastern United States can be divided into five  
7 regions, each of which exhibit spatial, relatively coherent patterns of O<sub>3</sub> properties at nonurban  
8 sites. Only sites classified as being rural or suburban and with land usage of forest, agriculture,  
9 or residential were included in the analyses. These criteria were chosen to avoid sites where O<sub>3</sub>  
10 is scavenged by NO that can be found in high concentrations near major sources, such as traffic



1 in urban cores. The five regions, shown in Figure 3-3, are characterized by different patterns  
2 of O<sub>3</sub> properties such as temporal persistence and seasonal variability. Figure 3-3 shows  
3 nonurban, monthly average, daily maximum 8-h O<sub>3</sub> concentrations in the five regions in the  
4 eastern United States from April to October 1993 to 2002.

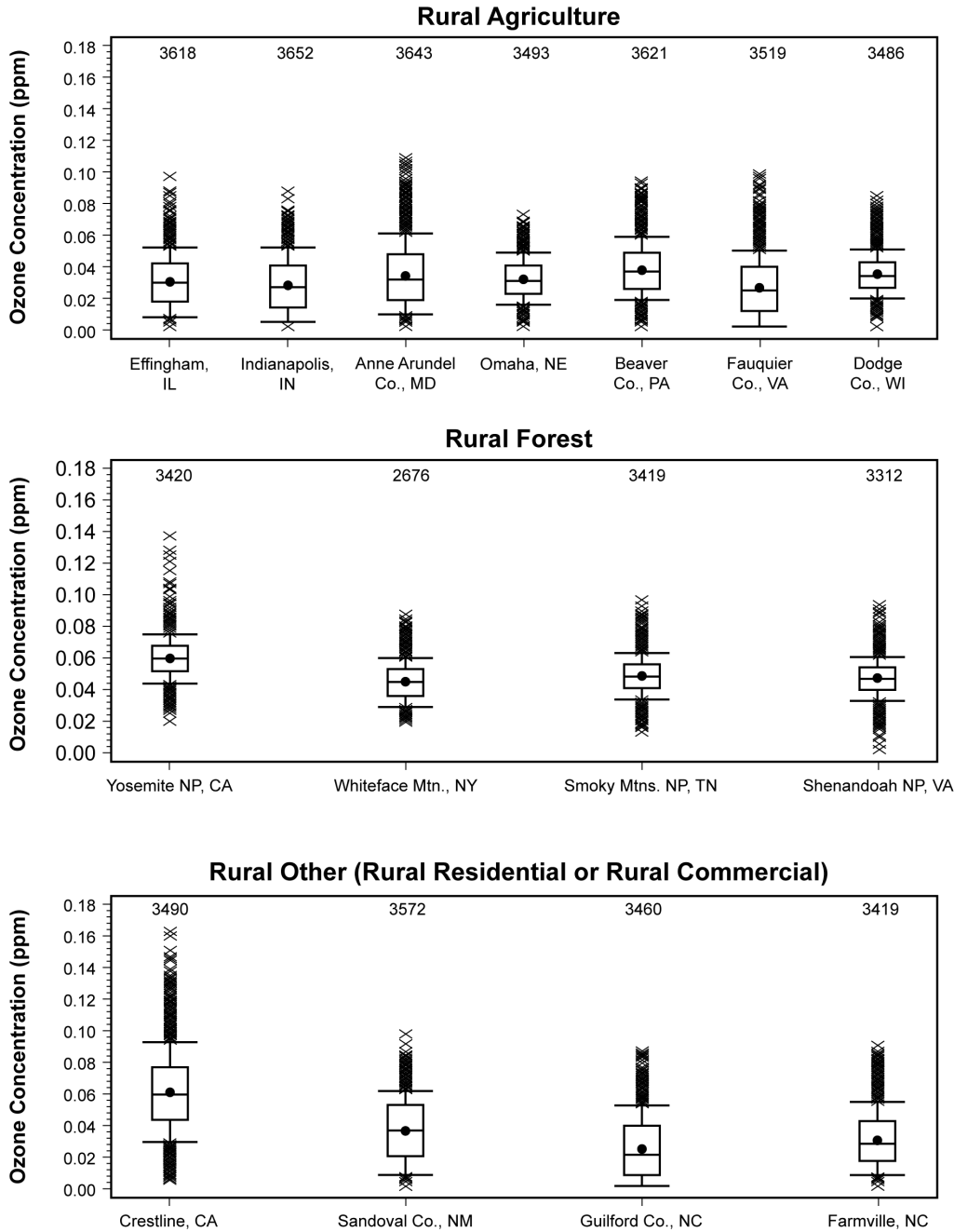
5 Regional differences are immediately apparent. Highest concentrations among all the  
6 regions are generally found in the Mid-Atlantic region (mean of 52 ppb) with highest values  
7 throughout the percentile distribution except for the overall maximum. Lowest mean  
8 concentrations (42 ppb) are found in Florida. In the northern regions (the Northeast, Great  
9 Lakes) and the Mid-Atlantic region, highest median and peak concentrations are found in July,  
10 whereas in the Southwest region, highest median concentrations are found in August, with  
11 highest peaks in June and September, i.e., outside the warmest summer months. In Florida,  
12 highest monthly averaged median and peak concentrations are found during the spring. High O<sub>3</sub>  
13 concentrations tend to be most persistent (3-4 days of persistence) in the southern regions, less  
14 persistent in the Mid-Atlantic region (2-3 days) and least persistent in the northern regions (1 or  
15 2 days). Analyses, such as these, are not available for the western United States, in part because  
16 of the difficulty in defining regions with relatively coherent O<sub>3</sub> properties.

17 Box plots showing the percentile distribution of hourly average O<sub>3</sub> concentrations for  
18 different types of rural sites for 2004 are given in Figures 3-4a (rural-agricultural), 3-4b  
19 (rural-forest) and 3-4c (rural-residential or commercial). Some associated metrics for vegetation  
20 exposures are given in Figures AX3-8 to AX3-10. Note that high O<sub>3</sub> concentrations are found at  
21 sites that are classified as rural, such as Anne Arundel Co., MD; Yosemite NP, CA; and  
22 Crestline, CA. Land use designations do not usually give an accurate picture of exposure  
23 regimes in rural areas, because the land use characterization of “rural” does not imply that a  
24 specific location is isolated from anthropogenic influences. Rather, the characterization refers  
25 only to the current use of the land, not to the presence of sources. Since O<sub>3</sub> produced from  
26 emissions in urban areas is transported to more rural downwind locations, elevated O<sub>3</sub>  
27 concentrations can occur at considerable distances from urban centers. In addition, major  
28 sources of O<sub>3</sub> precursors such as power plants and highways are located in nonurban areas and  
29 also produce O<sub>3</sub> in these areas. Due to lower chemical scavenging in nonurban areas, O<sub>3</sub> tends to  
30 persist longer in nonurban than in urban areas also tending to lead to higher exposures in  
31 nonurban areas influenced by anthropogenic precursor emissions.



**Figure 3-3. Box plots showing daily maximum 8-h  $O_3$  averaged by month over 1993 to 2002 in the five regions in the eastern United States derived by Lehman et al. (2004). The boxes define the interquartile range and the whiskers, the extreme values.**

Source: Lehman et al. (2004).



**Figure 3-4a-c. Hourly average O<sub>3</sub> concentrations observed at selected (a) rural-agricultural (b) rural-forested, and (c) rural-residential or commercial sites for 2004. The whiskers on the box plot represent the 10th and 90th percentile concentrations. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile concentrations. The dots inside the box represent the mean, for the statistic, at all sites. The number of observations is shown above each box plot.**

Source: Fitz-Simons et al. (2005).

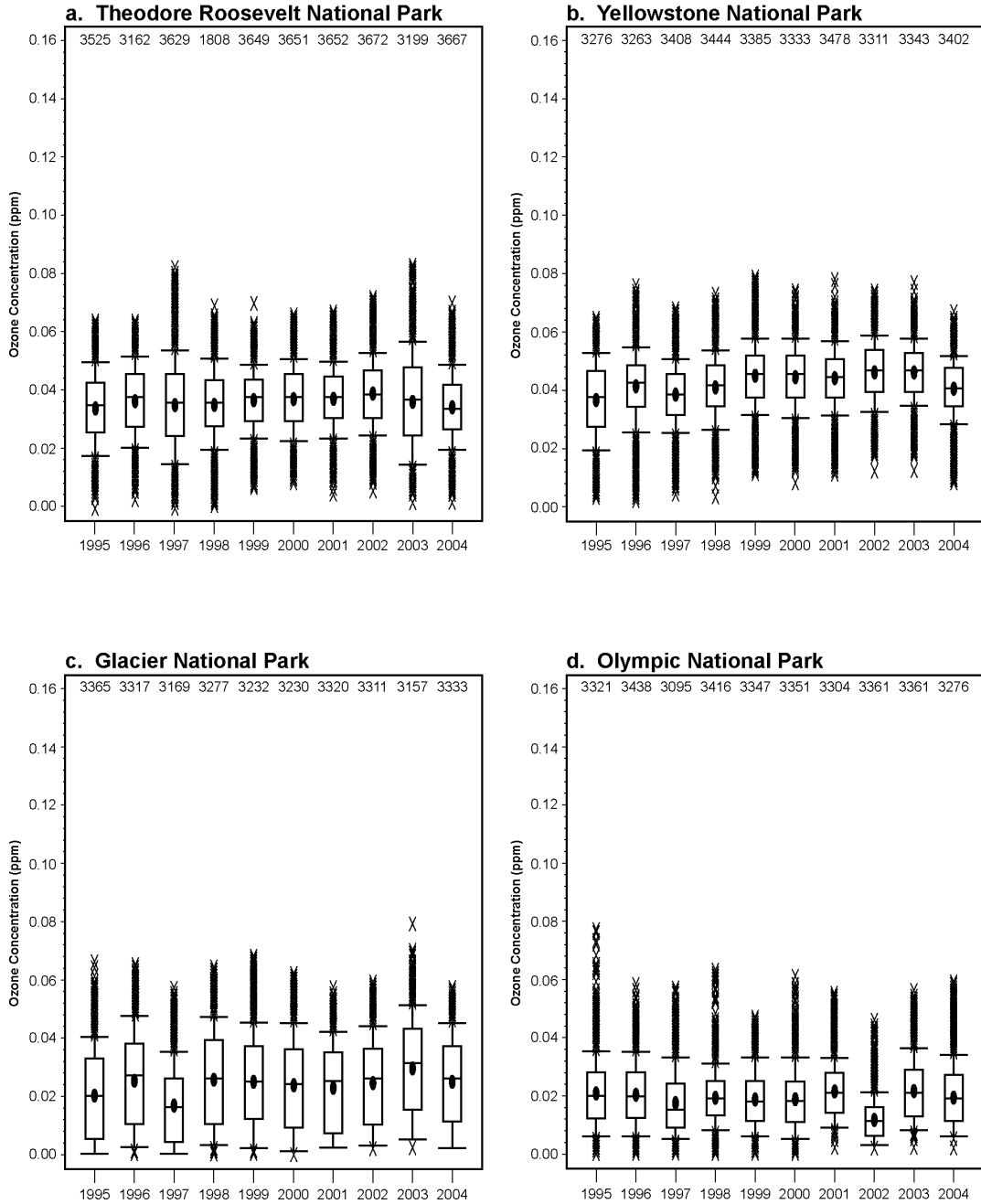
1 **Ozone Air Quality Data at Relatively Remote Monitoring Sites (RRMS)**

2 RRMS are sites that are located in the national parks that tend to be less affected by  
3 obvious pollution sources than other sites. This does not mean that they are completely  
4 unaffected by local pollution, as evidenced by the number of visitors to these national parks.

5 Box plots showing the percentile distribution of annual hourly averaged O<sub>3</sub> concentrations  
6 at four relatively remote monitoring sites (RRMS) are given in Figures 3-5a-d. It is important to  
7 characterize hourly average O<sub>3</sub> concentrations at RRMS so that assessments of the possible  
8 effects of O<sub>3</sub> on human health and vegetation use ranges of concentrations in their experiments  
9 that span the range of O<sub>3</sub> concentrations found in the U.S. In many controlled exposure studies  
10 examining vegetation, O<sub>3</sub> is filtered out of ambient air before it is admitted into the exposure  
11 chambers. As a result, O<sub>3</sub> levels of only a few ppb are used as controls.

12 As can be seen from Figures 3-5a-d, annual mean values of the daily maximum 8-h O<sub>3</sub>  
13 concentration have not changed much over the past 10 years of available data. Mean values  
14 typically range from about 0.020 ppm to about 0.040 ppm. Concentrations only rarely exceed  
15 0.080 ppm, in contrast to observations at other “rural” sites shown in Figures 3-4a-c.

16 The extent to which distributions found at sites with low maximum hourly average  
17 concentrations in the western United States are representative of sites in the eastern and  
18 midwestern United States is debatable because of regional differences in sources of precursors  
19 and transport patterns. Given the high density of sources in the eastern and midwestern  
20 United States, it is unclear whether a site could be found in either of these regions that would not  
21 be influenced by the transport of O<sub>3</sub> from nearby urban areas. Thus, with the exception of the  
22 Voyageurs NP site in Minnesota, observations at RRMS are limited to those obtained in the  
23 western United States. However, not all national park sites in the West can be considered to  
24 be free of strong regional pollution influences, e.g., Yosemite NP (CA) as shown in Figure 3-4b.  
25 Maps showing the nationwide distribution of various metrics for vegetation exposures are given  
26 in Section AX3.2, Figures AX3-13 to AX3-27.



**Figure 3-5a-d. Daily 8-h maximum O<sub>3</sub> concentrations observed at selected national park sites. The whiskers on the box plot represent the 10th and 90th percentile concentrations. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile concentrations. The dots inside the box represent the mean. The number of observations is shown above each box plot.**

Source: Fitz-Simons et al. (2005).

### 3.3 SPATIAL VARIABILITY OF O<sub>3</sub> IN URBAN AREAS

The spatial variability in O<sub>3</sub> concentrations in 24 MSAs across the United States was examined. These MSAs were selected to provide (1) information helpful for risk assessments, (2) a general overview of the spatial variability of O<sub>3</sub> in different regions of the country, and (3) insight into the spatial distribution of O<sub>3</sub> in cities where health outcome studies have been conducted. Statistical analyses of the human health effects of airborne pollutants based on aggregate population time-series data have often relied on ambient concentrations of pollutants measured at one or more central sites in a given metropolitan area. In the particular case of ground-level O<sub>3</sub> pollution, central-site monitoring has been justified as a regional measure of exposure mainly on the grounds that correlations between concentrations at neighboring sites measured over time are usually high. In MSAs with multiple monitoring sites, averages over the monitors have often been used to characterize population exposures. However, substantial differences in concentrations between monitors can exist even though concentrations measured at the monitoring sites are highly correlated, thus leading to the potential for exposure misclassification error.

Metrics for characterizing spatial variability include the use of Pearson correlation coefficients (*r*), values of the 90th percentile absolute difference in O<sub>3</sub> concentrations (*P*<sub>90</sub>), and coefficients of divergence (COD)<sup>1</sup>. These methods of analysis follow those used for characterizing PM<sub>2.5</sub> and PM<sub>10-2.5</sub> concentrations in Pinto et al. (2004) and in the latest edition of the Particulate Matter (PM) AQCD (U.S. Environmental Agency, 2004a). However, the calculations were performed on an hourly basis rather than on a 24-h basis. Data were aggregated over the local O<sub>3</sub> season as indicated in Table AX3-1. The length of the O<sub>3</sub> season varies across the country. In several southwestern states, it lasts all year long. In other areas, such as in New England, the mid-Atlantic states, the Midwest and the Northwest, it can be 6 months long, but typically it lasts from April through October.

---

<sup>1</sup> The COD is defined as follows:

$$COD_{jk} = \sqrt{\frac{1}{p} \sum_{i=1}^p \left( \frac{x_{ij} - x_{ik}}{x_{ij} + x_{ik}} \right)^2} \quad (\text{AX3-1})$$

where  $x_{ij}$  and  $x_{ik}$  represent the 24-h average PM<sub>2.5</sub> concentration for day  $i$  at site  $j$  and site  $k$  and  $p$  is the number of observations.

1 Table 3-1 shows the urban areas chosen, the range of 24-h average O<sub>3</sub> concentrations over  
2 the O<sub>3</sub> season, the range of intersite correlation coefficients, the range of P<sub>90</sub> differences in O<sub>3</sub>  
3 concentrations between site pairs, and the range in COD values. A COD of zero implies that  
4 values in both data sets are identical, and a COD of one indicates that two data sets are  
5 completely different. In general, statistics were calculated for partial MSAs. This was done so  
6 as to obtain reasonable lower estimates of the spatial variability that is present, as opposed to  
7 examining the consolidated MSAs. However, this could not be readily done for Boston, MA  
8 and New York, NY, so statistics were calculated for those consolidated MSAs. More detailed  
9 calculations for a subset of nine MSAs are given in Figures AX3-28 through AX3-36 in  
10 Section AX3.3.

11 As can be seen from Table 3-1, no clearly discernible regional differences were found  
12 in the ranges of parameters analyzed. Additional urban areas would need to be examined to  
13 discern broadscale patterns. The data indicate considerable variability in the concentration  
14 fields. Mean O<sub>3</sub> concentrations vary within individual urban areas by factors of 1.4 to 4.  
15 Intersite correlation coefficients show mixed patterns (i.e., in some urban areas all pairs of sites  
16 are moderately to highly correlated, while other areas show a larger range of correlations).  
17 As may be expected, those areas showing a smaller range of seasonal mean concentrations also  
18 show a smaller range of intersite correlation coefficients. However, there are a number of cases  
19 where sites in an urban area may be moderately to highly correlated, but show substantial  
20 differences in absolute concentrations. In many cases, P<sub>90</sub> values can equal or exceed seasonal  
21 mean O<sub>3</sub> concentrations.

22 It is instructive to compare the metrics for spatial variability shown in Table 3-1 to those  
23 calculated for PM<sub>2.5</sub> and PM<sub>10-2.5</sub> in the PM AQCD (U.S. Environmental Agency, 2004). The  
24 values for concentrations and concentration differences are unique to the individual species, but  
25 the intersite correlation coefficients and the COD values can be directly compared. In general,  
26 the variability in O<sub>3</sub> concentrations is larger than for PM<sub>2.5</sub> concentrations and comparable to that  
27 obtained for PM<sub>10-2.5</sub>. Intersite correlation coefficients in some areas (e.g., Philadelphia, PA;  
28 Atlanta, GA; Portland, OR) can be very similar for both PM<sub>2.5</sub> and for O<sub>3</sub>. However, there is  
29 much greater variability in the concentration fields of O<sub>3</sub> as evidenced by the much higher COD  
30 values. Indeed, COD values are higher for O<sub>3</sub> than for PM<sub>2.5</sub> in each of the urban areas  
31 examined. In all of the urban areas examined for O<sub>3</sub>, some site pairs are always very highly

**Table 3-1. Summary Statistics for the Spatial Variability of O<sub>3</sub> (in ppm) in Selected Urban Areas in the United States**

Urban Area	Number of Sites	Minimum Mean Conc.	Maximum Mean Conc.	Minimum Corr. Coeff.	Maximum Corr. Coeff.	Minimum P <sub>90</sub> <sup>a</sup>	Maximum P <sub>90</sub>	Minimum COD <sup>b</sup>	Maximum COD
Boston, MA	18	0.021	0.033	0.46	0.93	0.012	0.041	0.17	0.45
New York, NY	29	0.015	0.041	0.45	0.96	0.0080	0.044	0.17	0.55
Philadelphia, PA	12	0.020	0.041	0.79	0.95	0.011	0.036	0.23	0.46
Washington, DC	20	0.022	0.041	0.72	0.97	0.010	0.032	0.17	0.45
Charlotte, NC	8	0.031	0.043	0.48	0.95	0.012	0.038	0.17	0.32
Atlanta, GA	12	0.023	0.047	0.63	0.94	0.013	0.045	0.24	0.55
Tampa, FL	9	0.024	0.035	0.74	0.94	0.011	0.025	0.20	0.35
Detroit, MI	7	0.022	0.037	0.74	0.96	0.0090	0.027	0.19	0.36
Chicago, IL	24	0.015	0.039	0.38	0.96	0.0080	0.043	0.16	0.50
Milwaukee, WI	9	0.027	0.038	0.73	0.96	0.0090	0.025	0.18	0.33
St. Louis, MO	17	0.022	0.038	0.78	0.96	0.0090	0.031	0.15	0.41
Baton Rouge, LA	7	0.018	0.031	0.81	0.95	0.0090	0.029	0.23	0.41
Dallas, TX	10	0.028	0.043	0.67	0.95	0.011	0.033	0.16	0.36
Houston, TX	13	0.016	0.036	0.73	0.96	0.0090	0.027	0.20	0.38
Denver, CO	8	0.022	0.044	0.60	0.92	0.013	0.044	0.16	0.46
El Paso, TX	4	0.022	0.032	0.81	0.94	0.012	0.023	0.24	0.31
Salt Lake City, UT	8	0.029	0.048	0.52	0.92	0.012	0.043	0.13	0.51
Phoenix, AZ	15	0.021	0.058	0.29	0.95	0.011	0.057	0.15	0.61
Seattle, WA	5	0.015	0.038	0.63	0.94	0.0080	0.024	0.16	0.46
Portland, OR	5	0.015	0.036	0.73	0.91	0.011	0.025	0.20	0.50
Fresno, CA	6	0.030	0.047	0.90	0.97	0.0090	0.027	0.17	0.40
Bakersfield, CA	8	0.028	0.047	0.23	0.96	0.013	0.052	0.20	0.58
Los Angeles, CA	14	0.010	0.042	0.42	0.95	0.010	0.053	0.22	0.59
Riverside, CA	18	0.018	0.054	0.38	0.95	0.013	0.057	0.15	0.64

<sup>a</sup> P<sub>90</sub> = 90th percentile absolute difference in concentrations.<sup>b</sup> COD = coefficient of divergence for different site pairs.



1 correlated with each other (i.e.,  $r > 0.9$ ) as seen for  $PM_{2.5}$ . These sites also show less variability  
2 in concentration and are probably influenced most strongly by regional production mechanisms.

3 The above considerations indicate that caution should be observed in using data from the  
4 network of ambient  $O_3$  monitors to approximate community-scale human exposures. A similar  
5 conclusion was reached for PM using data from the  $PM_{2.5}$  FRM network, as indicated in  
6 Section 3.4 of the PM AQCD (U.S. Environmental Protection Agency, 2004a).

### 8 **3.3.2 Small-scale Horizontal and Spatial Variability in Ozone Concentrations**

#### 9 *Ozone concentrations near roadways*

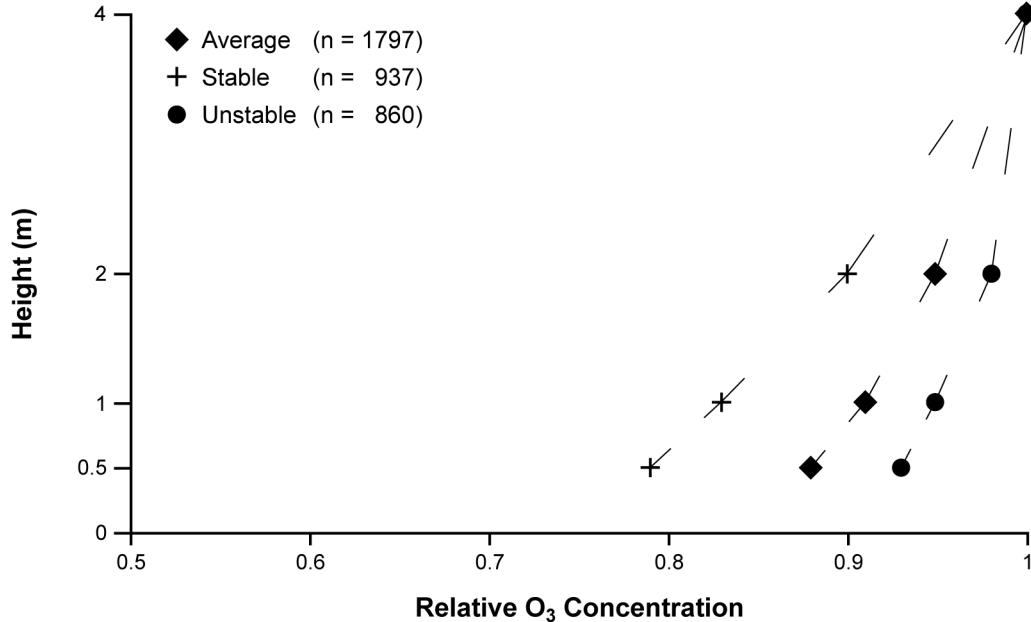
10 Apart from the larger scale variability in surface  $O_3$  concentrations, there is also significant  
11 variability on the micro-scale (< a few hundred meters), especially near roadways and other  
12 sources of emissions that react with  $O_3$ . These sources are not confined to urban areas. Sources  
13 of emissions that react with  $O_3$  such as highways and power plants are also found in rural areas.  
14 Johnson (1995) described the results of studies examining  $O_3$  upwind and downwind of  
15 roadways in Cincinnati, OH. In these studies,  $O_3$  upwind of the roadway was about 50 ppb and  
16 these values were not found again until distances of about 100 m downwind. The  $O_3$  profile  
17 varied inversely with that of  $NO$ , as might be expected. For peak  $NO$  concentrations of 30 ppb  
18 immediately downwind of the road, the  $O_3$  mixing ratio was about 36 ppb, or about 70% of the  
19 upwind value. The magnitude of the downwind depletion of  $O_3$  depends on the emissions of  
20  $NO$ , the rate of mixing of  $NO$  from the roadway and ambient temperature and so depletions  
21 of  $O_3$  downwind of roadways are expected, but with variable magnitude. Guidance for the  
22 placement of  $O_3$  monitors (U.S. Environmental Protection Agency, 1998) states a separation  
23 distance that depends on traffic counts. For example, a minimum separation distance of 100 m  
24 from a road with 70,000 vehicles per day (about 3,000 vehicles per hour) is recommended for  
25 siting an  $O_3$  monitor to avoid interference that would mean a site is no longer representative of  
26 the surrounding area. An average rate of about 3,000 vehicles per hour passing by a monitoring  
27 site implies a road with rather heavy traffic. As noted in Section AX3.3.1 for the Lakewood, CA  
28 monitoring,  $O_3$  levels are lower at sites located near traffic than those located some distance  
29 away and the scavenging of  $O_3$  by emissions of  $NO$  from roadways is a major source of spatial  
30 variability in  $O_3$  concentrations. It should also be noted that scavenging of  $O_3$  by  $NO$  near  
31 roadways was more pronounced before the implementation of stringent  $NO_x$  emissions controls.

1 ***Vertical Variations in Ozone Concentrations***

2 In addition to horizontal variability in O<sub>3</sub> concentrations, there are also variations in the  
3 vertical profile of O<sub>3</sub> in the lowest layers of the atmosphere to consider. The planetary boundary  
4 layer consists of an outer and an inner portion. The inner part of the planetary boundary layer  
5 extends from the surface to about one-tenth the height of the planetary boundary layer. Winds  
6 and transported properties, such as O<sub>3</sub>, are especially susceptible to interactions with obstacles,  
7 such as buildings and trees in the inner boundary layer (atmospheric surface layer) (e.g., Garratt,  
8 1992). Inlets to ambient monitors (typically at heights of 3 to 5 meters) are located in, and  
9 human and vegetation exposures occur in this part of the boundary layer.

10 Photochemical production and destruction of O<sub>3</sub> occur throughout the planetary boundary  
11 layer. However, O<sub>3</sub> is also destroyed on the surfaces of buildings, vegetation, etc. On most  
12 surfaces, O<sub>3</sub> is destroyed with every collision. In addition, O<sub>3</sub> is scavenged by NO emitted by  
13 motor vehicles and soils. These losses imply that the vertical gradient of O<sub>3</sub> should always be  
14 directed downward. The magnitude of the gradient is determined by the intensity of turbulent  
15 mixing in the surface layer.

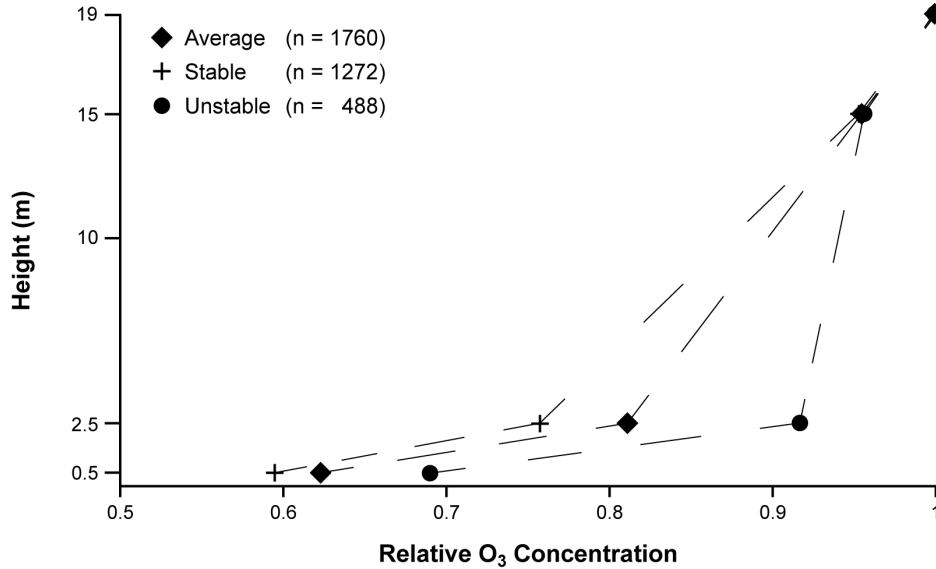
16 Most work characterizing the vertical profile of O<sub>3</sub> near the surface has been performed in  
17 nonurban areas with the aim of calculating fluxes of O<sub>3</sub> and other pollutants through forest  
18 canopies and to crops and short vegetation, etc. Corresponding data are sparse for urban areas.  
19 However, monitoring sites are often set up in open areas such as parks and playgrounds where  
20 surface characteristics may be more similar to those in rural areas than to those in the  
21 surrounding urban area. The vertical profiles of O<sub>3</sub> measured over low vegetation are shown in  
22 Figure 3-6. These measurements were obtained as part of a field campaign to measure the fluxes  
23 of several gas and aerosol phase pollutants using the gradient-flux technique (Horváth et al.,  
24 1995). The labels stable and unstable in the figure refer to atmospheric stability conditions and  
25 average represents the overall average. Ozone concentrations were normalized to their values at  
26 4 m height. As can be seen from the figure, there was a decrease of about 20% in going from a  
27 height of 4 m down to 0.5 m above the surface during stable conditions, but O<sub>3</sub> decreased by  
28 only about 7% during unstable conditions. The average decrease was about 10% for all  
29 measurements. As might be expected, O<sub>3</sub> concentrations at all heights were very highly  
30 correlated with one another. Of course, these values represent averages and there is scatter about  
31 them. Under strongly stable conditions, they fall off toward the surface. However, these



**Figure 3-6. Vertical profile of O<sub>3</sub> obtained over low vegetation. Values shown are relative to concentrations at 4 m above the surface. Ozone concentrations for unstable and unstable conditions were 41.3 and 24.1 ppb, and average O<sub>3</sub> concentration weighted by stability class was 33.1 ppb at 4 m.**

Source. Horváth et al. (1995).

1 conditions tend to occur mainly during night and the stability regime during the day in urban  
 2 areas tends more toward instability because of the urban heat island effect. Figure 3-7 shows the  
 3 vertical profile of O<sub>3</sub> measured in a spruce forest by the same group (Horváth et al., 2003). The  
 4 fall off of O<sub>3</sub> in this case is due to uptake by trees, reaction with ambient NO and with NO  
 5 emitted by the soil in the forest, and reaction with hydrocarbons emitted by the trees in addition  
 6 to deposition on the surface.



**Figure 3-7. Vertical profile of O<sub>3</sub> obtained in a spruce forest. Values shown are relative to concentrations at 19 m above the surface. Mean tree height is 14.5 m. Ozone concentrations for unstable and unstable conditions were 36.7 and 33.8 ppb, and the average O<sub>3</sub> concentration weighted by stability class was 34.6 ppb at 19 m.**

Source: Horváth et al. (2003).

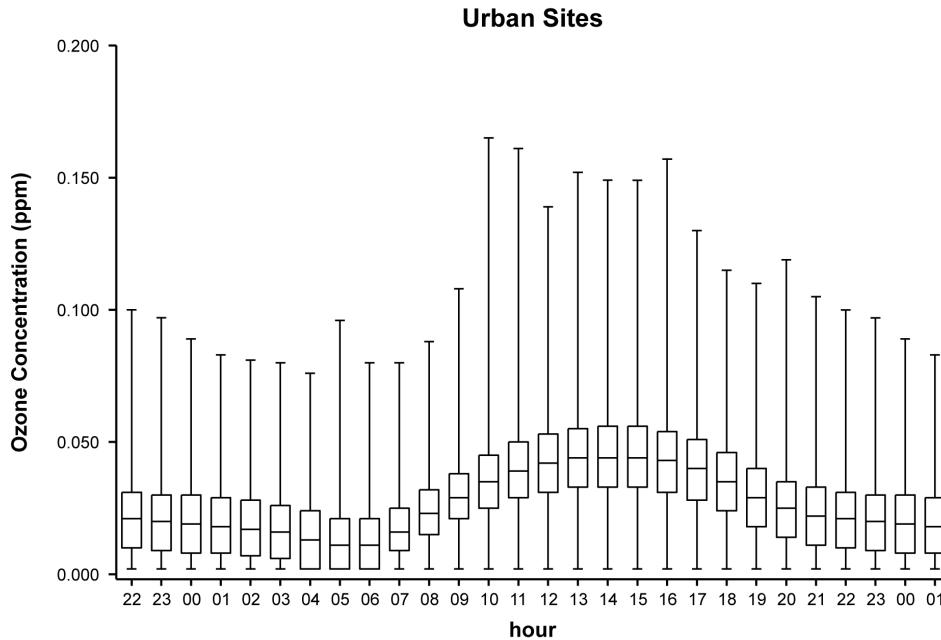
### 3.4 DIURNAL AND SEASONAL VARIABILITY OF OZONE

#### *Diurnal Variability*

Diurnal variations in O<sub>3</sub> at a given location are controlled by a number of factors such as the relative importance of transport versus local photochemical production and loss rates, the timing for entrainment of air from the nocturnal residual boundary layer and the diurnal variability in mixing layer height.

#### *Diurnal Patterns in the Nationwide Data Set*

Composite urban, diurnal variations in hourly averaged O<sub>3</sub> for April through October 2000 to 2004 are shown in Figure 3-8. As can be seen from Figure 3-8, daily 1-h maxima tend to occur in mid-afternoon and daily 1-h minima tend to occur during the early morning. However, there is also considerable spread in these times. Therefore, some caution must be exercised in



**Figure 3-8. Composite, nationwide diurnal variability in hourly averaged O<sub>3</sub> in urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima.**

Source: Fitz-Simons et al. (2005).

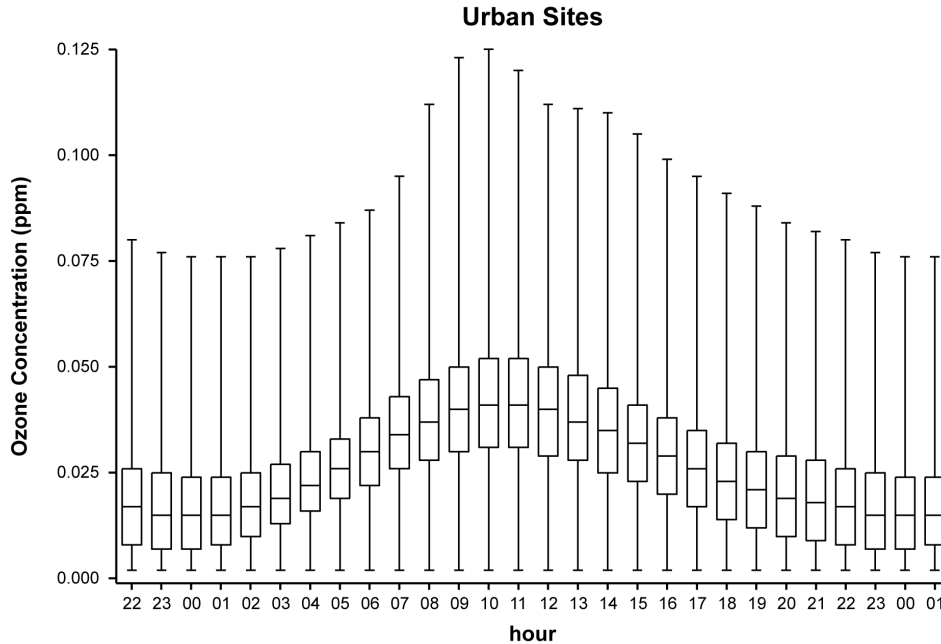
1 extrapolating results from one city to another and when attempting to judge the time of day when  
 2 the daily 1-h maximum occurs.

3 Corresponding data for 8 hour average O<sub>3</sub> data are shown in Figure 3-9. As can be seen  
 4 from Figure 3-9, daily maximum eight hour O<sub>3</sub> concentrations tend to occur from about 10 a.m.  
 5 to about 6 p.m. As can be seen from Figure 3-9, they can also occur at slightly different times  
 6 and the variation in the 8-h averages is smoother than for the 1-h averages. The minima in the  
 7 8 h averages tend to occur starting at about midnight.

8

9 ***Diurnal Patterns in EPA's 12 Cities***

10 The diurnal variability of hourly averaged O<sub>3</sub> in the twelve urban areas considered for  
 11 inclusion in EPA's human health exposure assessment risk assessment for the current review is  
 12 illustrated in Figures 3-10a-l for April to October. Daily maximum 1-h concentrations tend to

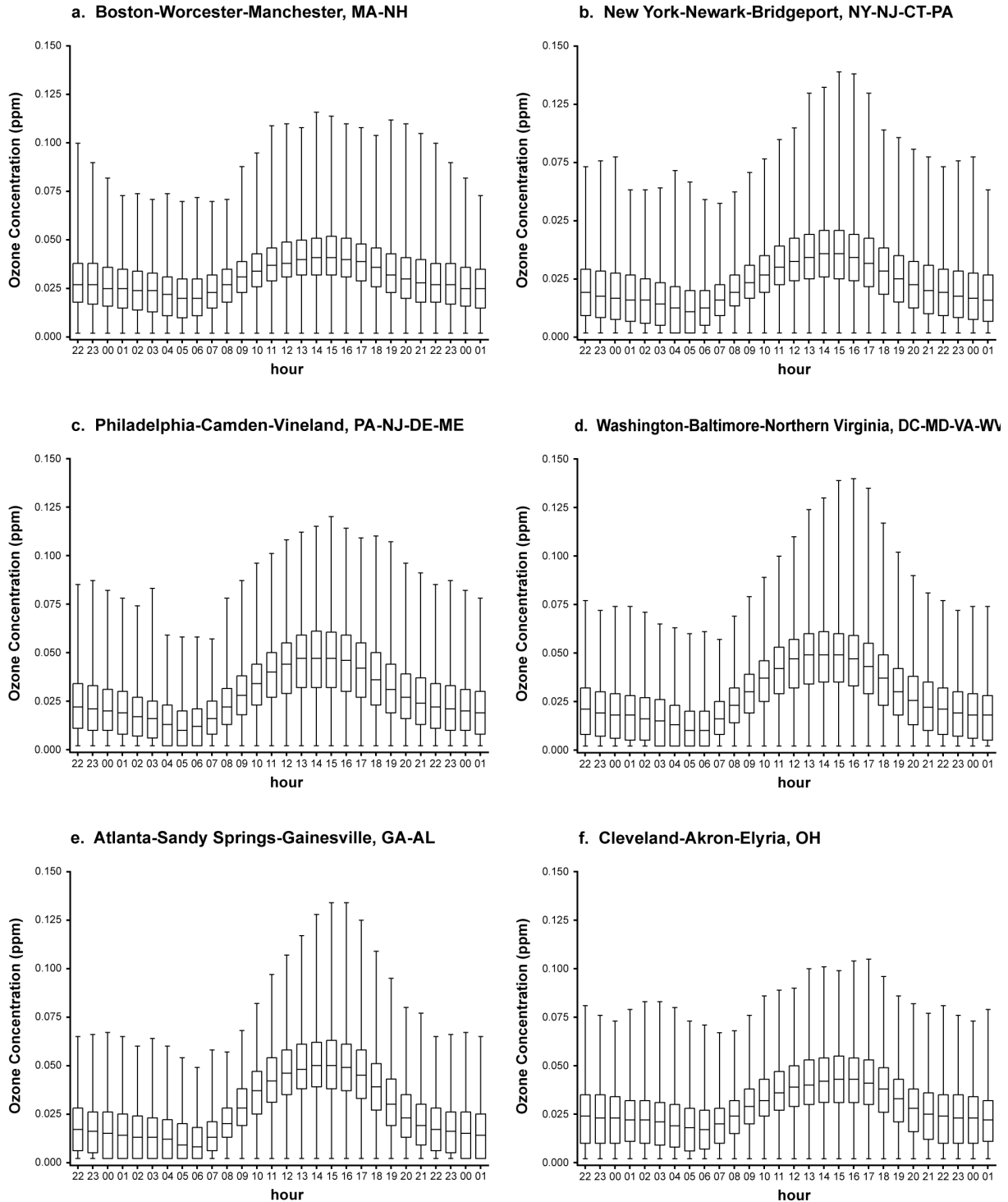


**Figure 3-9. Composite, nationwide diurnal variability in 8 hour average O<sub>3</sub> in urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).

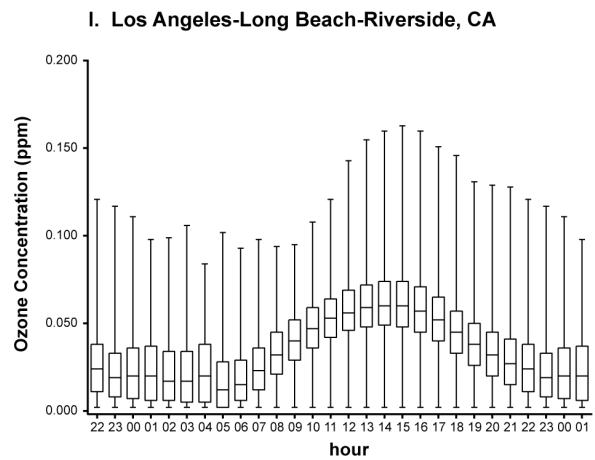
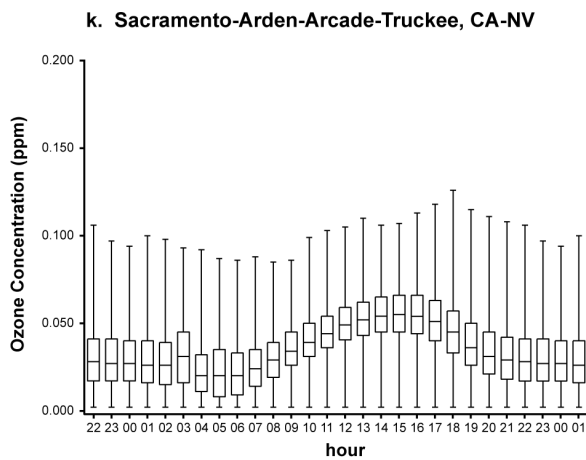
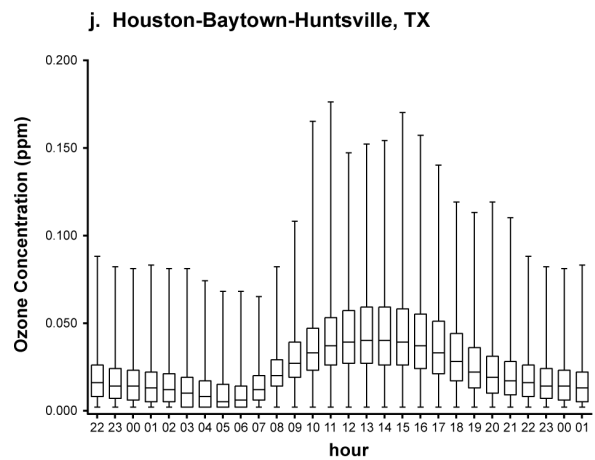
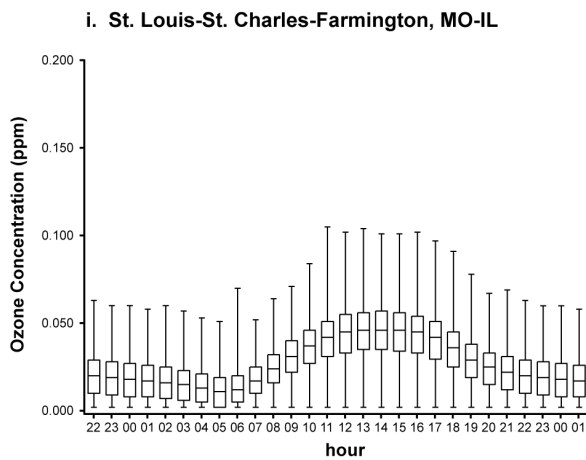
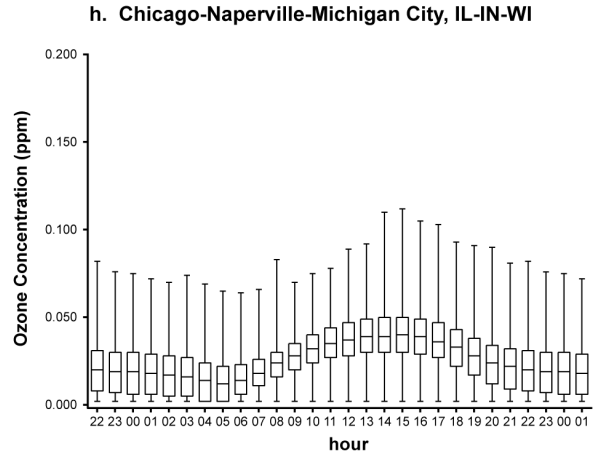
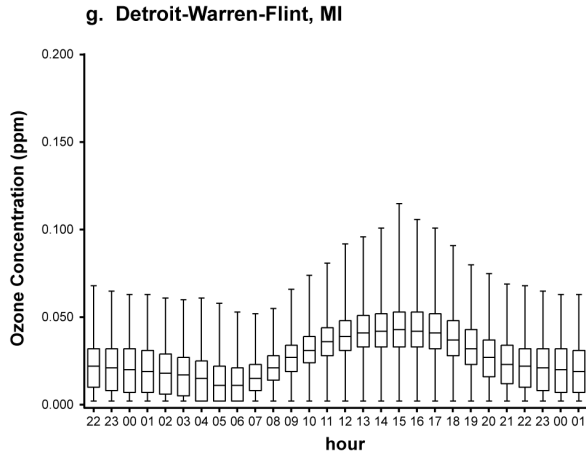
1 occur in mid-afternoon. However, as can be seen from the figures, the diurnal patterns vary  
 2 from city to city, with high values ( $\geq 0.100$  ppm) also occurring either late in the evening as in  
 3 Boston, past midnight as in Los Angeles and Sacramento, or midmorning as in Houston.  
 4 Typically, high values such as these are found during the daylight hours in mid to late afternoon.  
 5 The reasons for the behavior of O<sub>3</sub> during the night at the above-mentioned locations are not  
 6 clear. Measurement issues may be involved or there may be physical causes such as transport  
 7 phenomena, as discussed in Chapter 2. As discussed in Chapter 2, and in greater detail in  
 8 Section AX2.3.3, nocturnal low level jets are capable of producing secondary O<sub>3</sub> maxima at  
 9 night.

10 The diurnal variability of O<sub>3</sub> averaged over 8 hours in the same twelve urban areas is  
 11 shown in Figures 3-11a-l. The diurnal patterns of O<sub>3</sub> are broadly similar between 1-h averages  
 12 and 8-h averages. A typical pattern shows the 8-h daily maximum occurring from about 10 a.m.



**Figure 3-10a-f. Diurnal variability in hourly averaged O<sub>3</sub> in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima.**

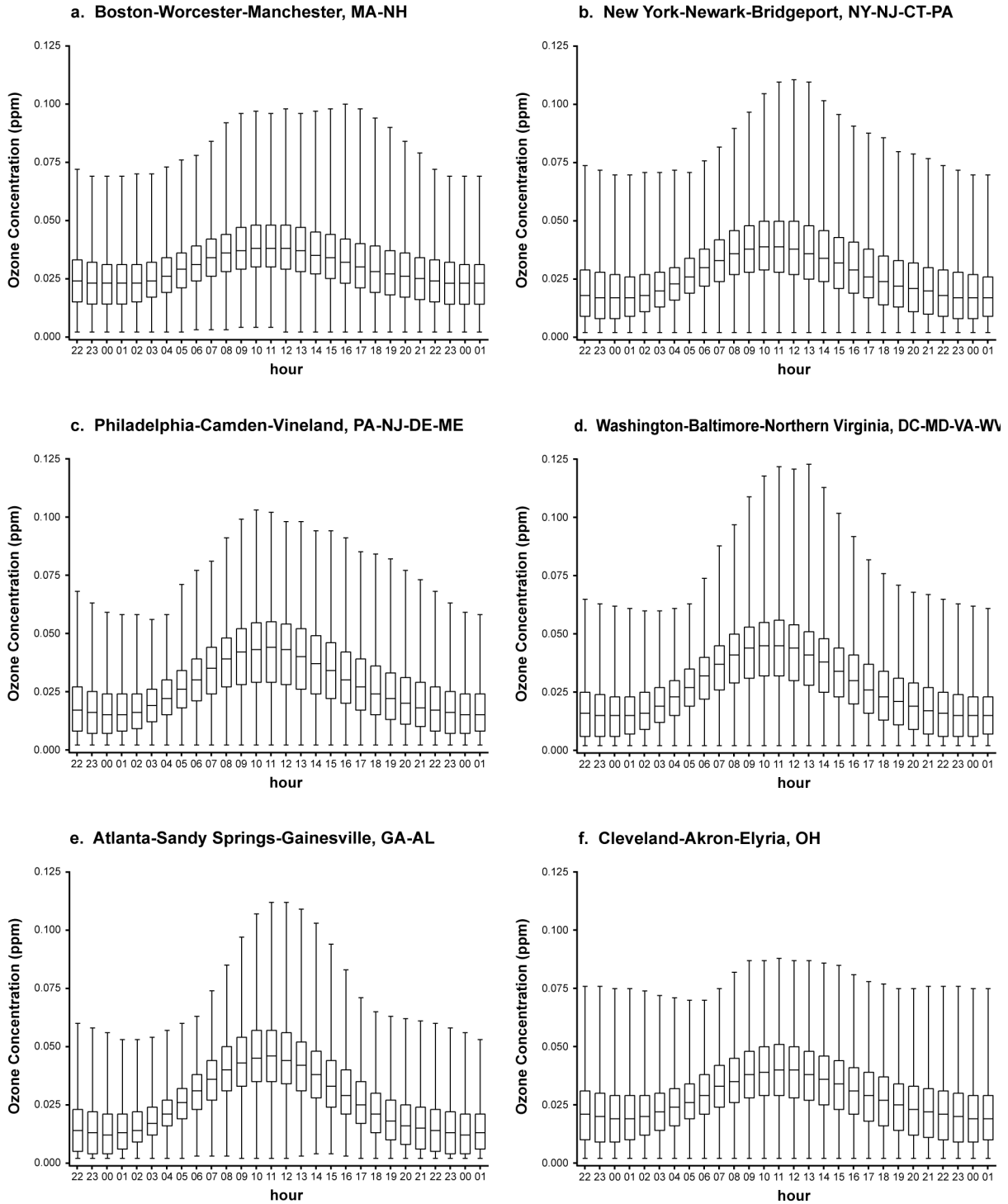
Source: Fitz-Simons et al. (2005).



**Figure 3-10g-l. Diurnal variability in hourly averaged O<sub>3</sub> in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima.**

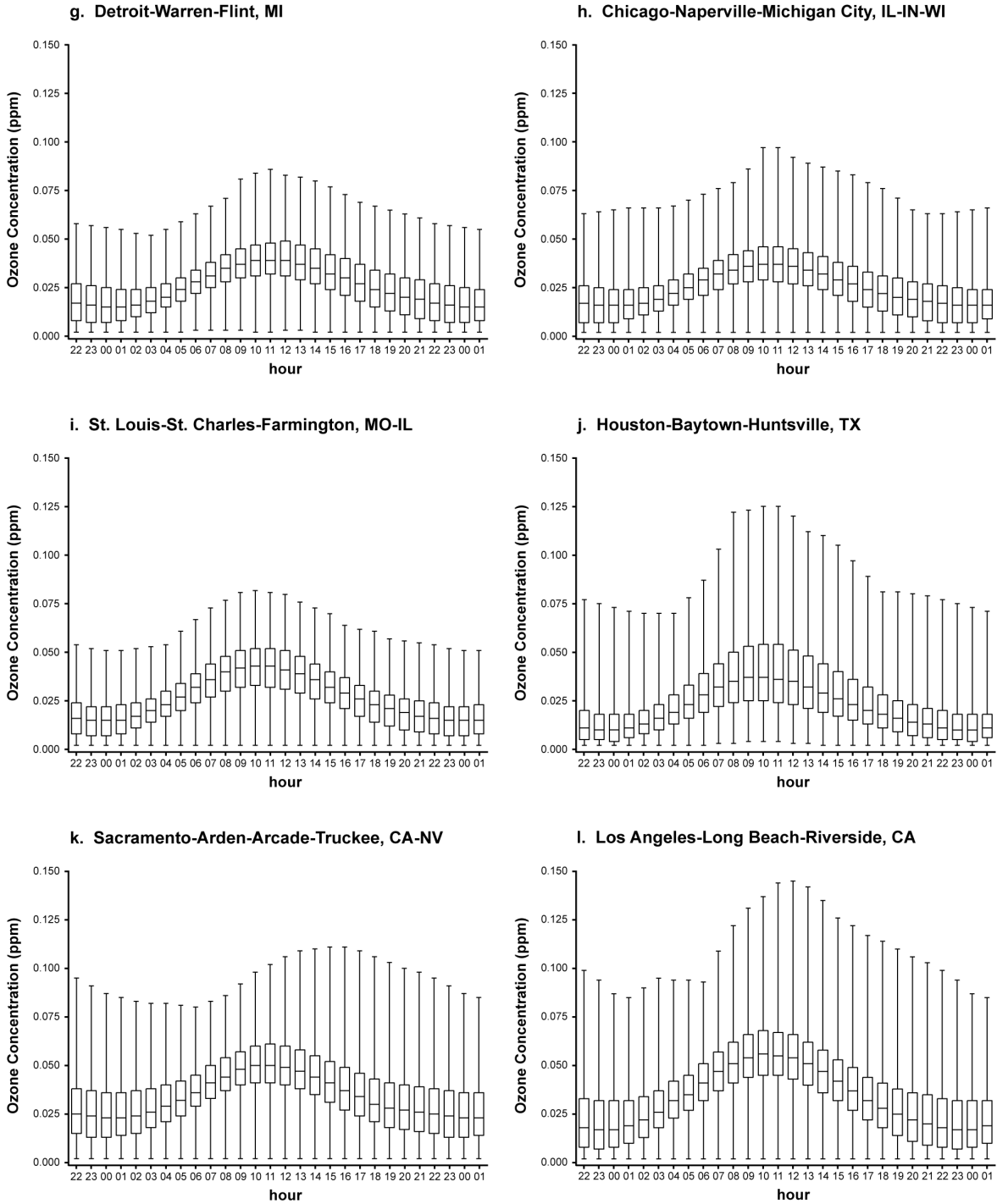
Source: Fitz-Simons et al. (2005).





**Figure 3-11a-f. Diurnal variability in 8 hour averaged O<sub>3</sub> in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).



**Figure 3-11g-l. Diurnal variability in 8 hour averaged O<sub>3</sub> in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).

1 to about 6 p.m., with some deviations from these times. For example, as shown in Figures 3-11a  
2 for Boston and 3-11k for Sacramento, the highest 8-h daily maximum values occur starting in  
3 mid-afternoon and extending into late evening. These results suggest that transport processes are  
4 playing the dominant role in determining the timing of the highest daily maxima in these areas.

5 On days with high 1-h daily maximum concentrations (e.g.,  $\geq 0.12$  ppm) the maxima tend  
6 to occur in a smaller time window centered in the middle of the afternoon, compared to days in  
7 which the maximum is lower. For example, on the high O<sub>3</sub> days the 1-h maximum occurs from  
8 about 11 a.m. to about 6 p.m. However, on days in which the 1-h daily maximum is  
9  $\leq 0.080$  ppm, the daily maximum can occur at any time during the day or night, with only a 50%  
10 probability that it occurs between 1 and 3 p.m., in each of the 12 cities. (The time of day when  
11 the daily maximum 1-h O<sub>3</sub> concentration occurs is illustrated for four of the cities in Figures  
12 AX3-45a-d.). Photochemical reactions in combination with diurnal emissions patterns are  
13 expected to produce mid-afternoon peaks in urban areas. These results suggest that transport  
14 from outside the urban airshed plays the major role for determining the timing of the daily  
15 maxima for low peak O<sub>3</sub> levels. This pattern is typical for the Los Angeles-Long Beach-  
16 Riverside, CA area even on high O<sub>3</sub> days.

17 The same general patterns emerge for the timing of the 1-h daily maximum O<sub>3</sub>  
18 concentration as are found for the daily maximum 8-h average O<sub>3</sub> concentration. As mentioned  
19 above, the daily maximum 8-h O<sub>3</sub> concentrations are generally found between the hours of  
20 10 a.m. and 6 p.m. However, there are significant fractions of the time when this is not the case,  
21 e.g., for high values in Houston, TX and Los Angeles, CA, or in general for lower values at any  
22 of the cities examined. (The time of day when the daily maximum 8-h average O<sub>3</sub>  
23 concentrations occurs is shown for four cities in Figures AX3-46a-d.). Although the 8-h  
24 average O<sub>3</sub> concentration is highly correlated with the daily maximum 1-h average O<sub>3</sub>  
25 concentration, there are situations where the daily maximum 8-h average O<sub>3</sub> concentration might  
26 be driven by very high values in the daily maximum 1-h average O<sub>3</sub> concentration as illustrated  
27 in Figure 3-10j for Houston, TX. In cases such as these, the predicted 8-h average may  
28 overestimate the short-term O<sub>3</sub> concentration later in the day.

29 The patterns of diurnal variability for both 1-h and 8-h averages have remained quite stable  
30 over the 15-year period from 1990 to 2004, with times of occurrence of the daily maxima  
31 varying by no more than an hour from year to year in each of the 12 cities.

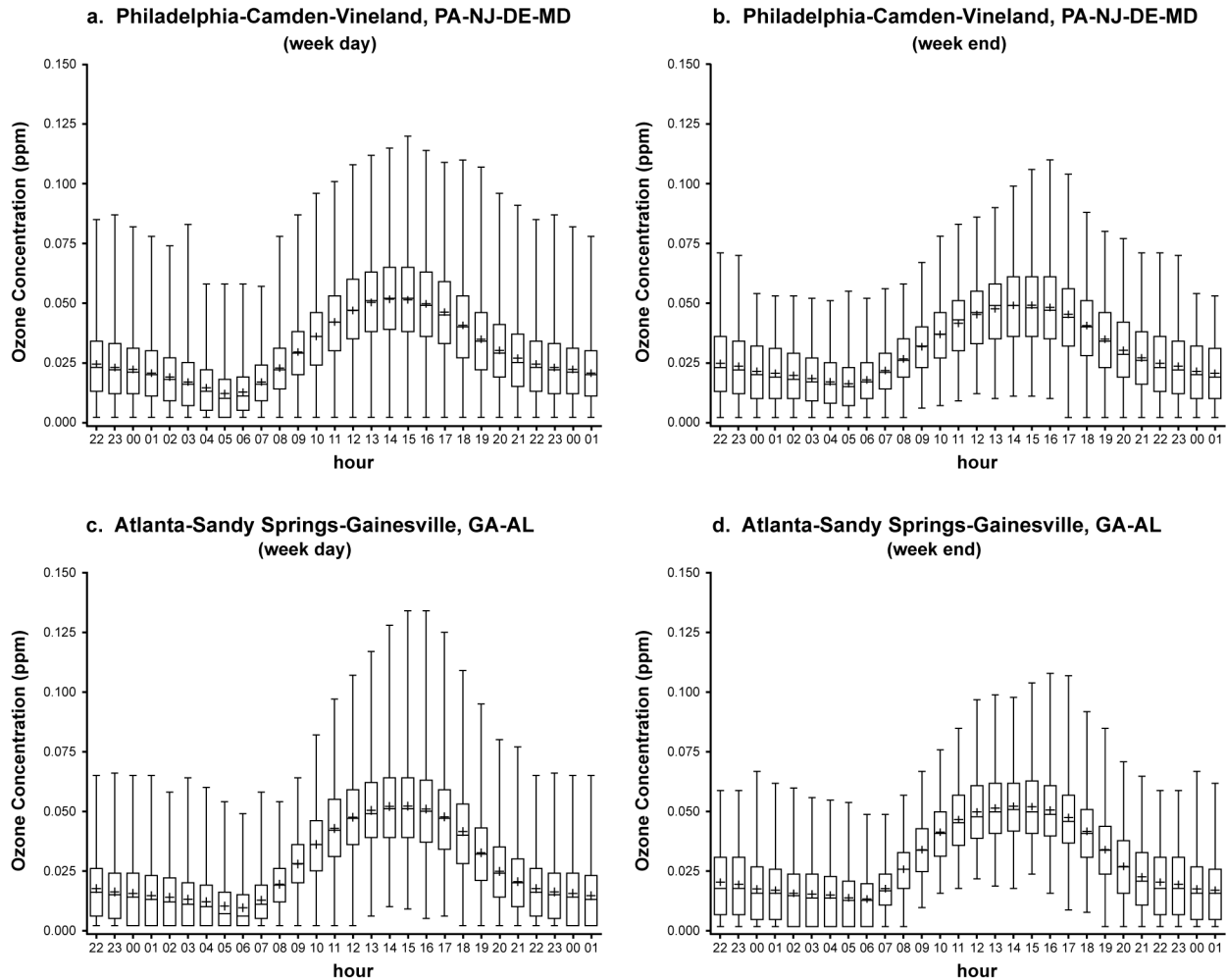
1 ***Weekday/Weekend Differences***

2 Differences in the diurnal behavior of O<sub>3</sub> have been observed in a number of cities (e.g.,  
3 Heuss et al., 2003). Figures 3-12a-h show the contrast in the patterns of hourly averaged O<sub>3</sub> in  
4 the greater Philadelphia, Atlanta, Houston and Los Angeles areas from weekdays to weekends.  
5 Daily maximum concentrations occur basically at the same time on either weekdays or  
6 weekends. Differences are apparent in the hourly concentrations, especially in the extreme  
7 values. Weekday/weekend differences in 8-h average O<sub>3</sub> concentrations are shown in Figures  
8 3-13a-h. As can be seen from a comparison of the weekend versus weekday patterns, there is a  
9 tendency for the lowest values to be higher on weekends than on weekdays. Lower traffic  
10 volumes, in particular diesel truck traffic, lead to less NO emissions and titration of O<sub>3</sub> on  
11 weekends. The spike in values shown for Houston in midmorning shown in Figure 3-12f  
12 resulted from the release of highly reactive hydrocarbons from the petrochemical industry  
13 (which could occur on any day of the week). Otherwise, the maximum O<sub>3</sub> concentrations could  
14 be seen to occur on the weekdays as they do in Philadelphia and Atlanta, in contrast to  
15 Los Angeles. Indeed, the diurnal pattern in Houston is similar to that observed in Atlanta on  
16 weekdays, indicating some overall similarity in the sources of O<sub>3</sub>.

17  
18 ***Spatial Variability in Diurnal Patterns in Urban Areas***

19 Daily maxima in either the 1-h or 8-h averages do not necessarily occur at the same time of  
20 day at each site in the 12 cities, and the diurnal pattern observed at individual sites can vary from  
21 the composites shown in Figures 3-8 and 3-9. Differences between sites are not only related to  
22 distance; they also depend on the presence of nearby sources, such as highways. For example, in  
23 the Los Angeles basin, daily 1-h maxima are reached in the late afternoon in Riverside relatively  
24 close to sites in which the maximum is reached much earlier.

25 The general pattern that emerges from the site-to-site variability within the urban areas  
26 examined is that peaks in 1-h average concentrations are higher and tend to occur later at  
27 downwind sites than in the urban cores. To the extent that monitoring sites are either near to or  
28 remote from sources of precursors in urban/suburban areas, the behavior of O<sub>3</sub> will follow these  
29 basic patterns. Similar relations are found for the 8-h average O<sub>3</sub> concentrations. Differences in  
30 diurnal patterns between sites in urban cores and sites downwind of urban cores are illustrated



**Figure 3-12a-d. Diurnal variations in hourly averaged O<sub>3</sub> on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004.**

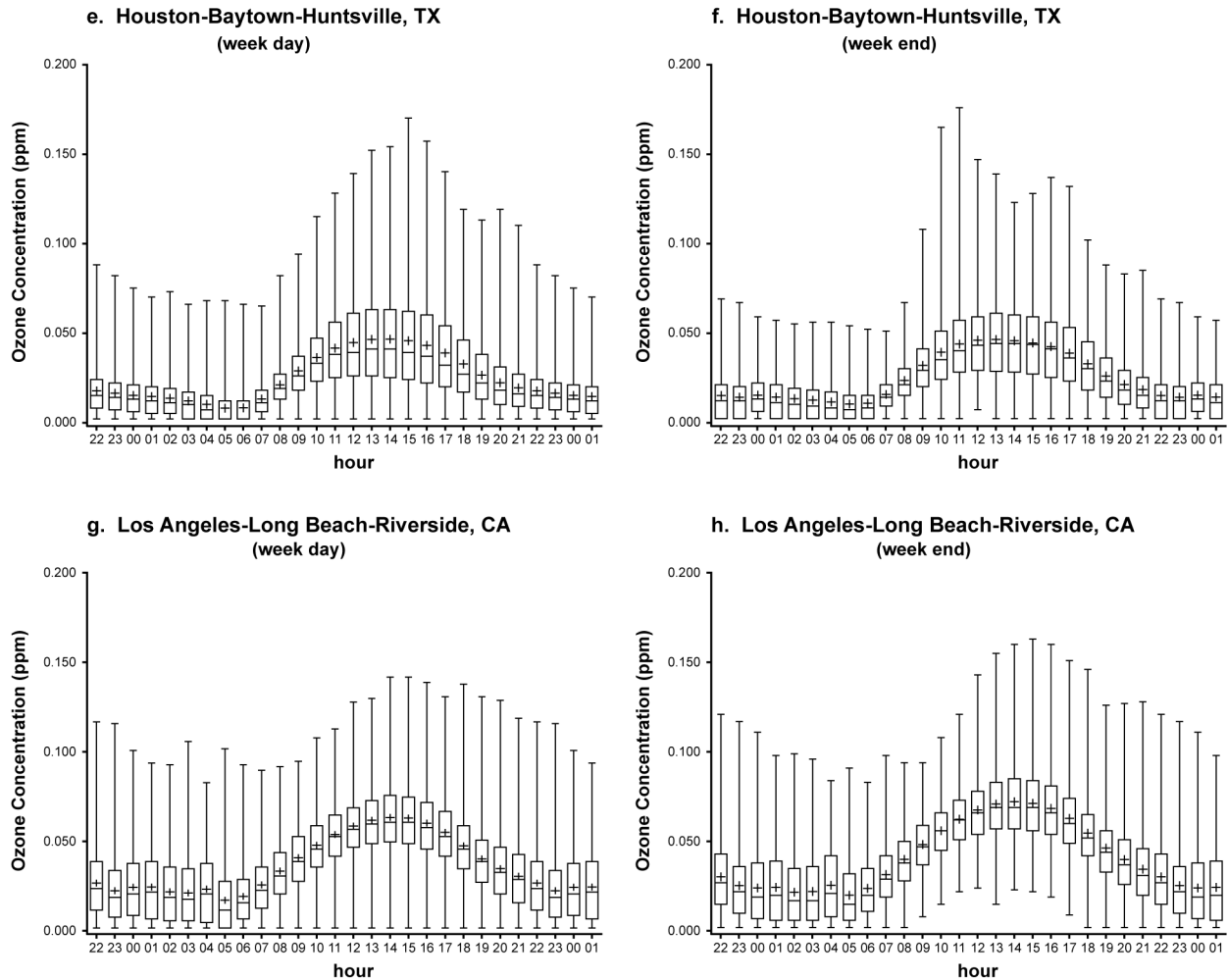
Source: Fitz-Simons et al. (2005).

1 in Figures AX3-49a-b to AX3-51a-b for 1-h average O<sub>3</sub> and in Figures AX3-52a-b to AX3-54a-b  
 2 for Detroit, MI, St. Louis, MO, and Riverside, CA areas.

3

4 ***Seasonal Variability***

5 It should not be assumed that highest O<sub>3</sub> levels are confined to the summer. Highest  
 6 average O<sub>3</sub> concentrations generally occur at background monitoring sites at midlatitudes in the

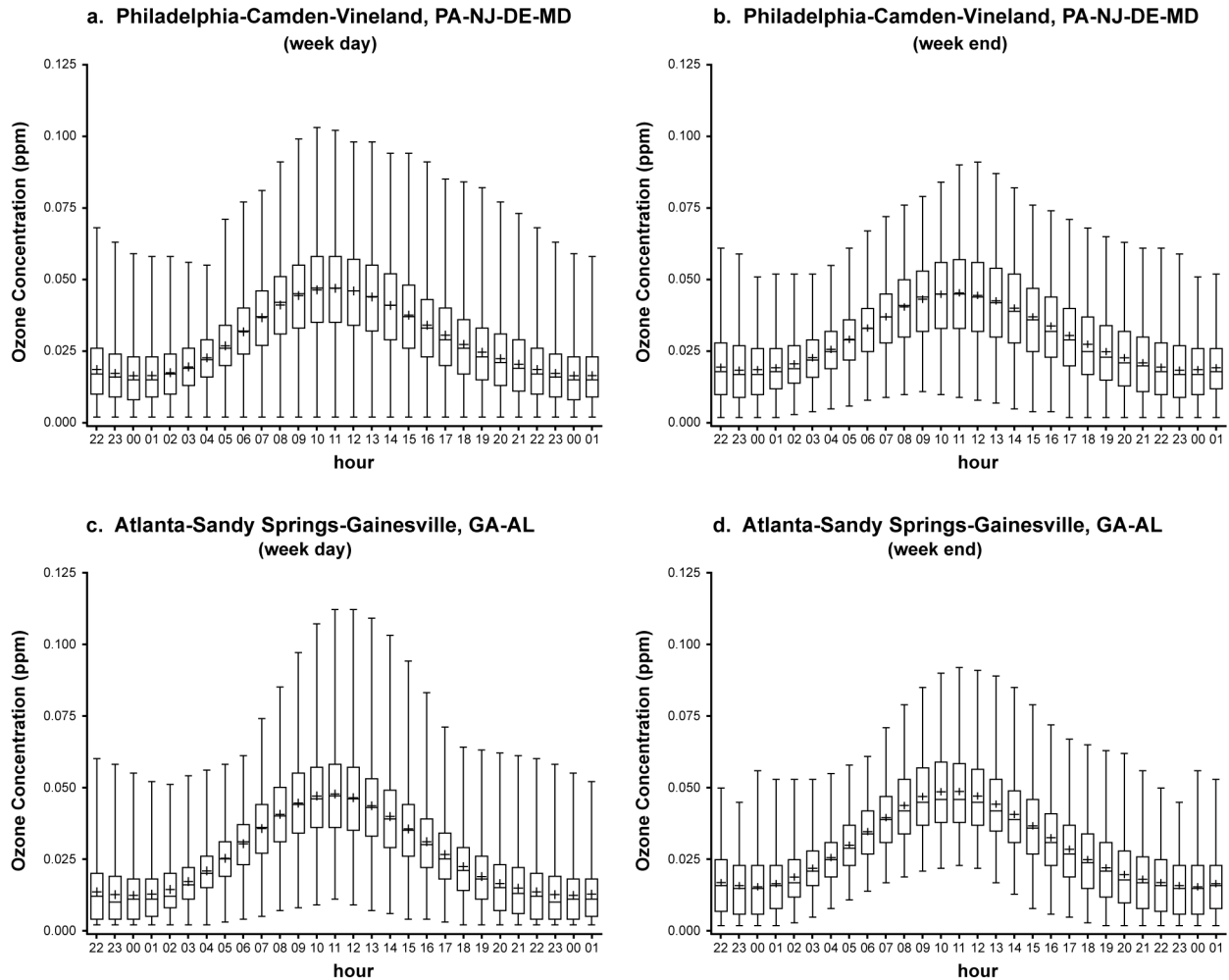


**Figure 3-12e-h. Diurnal variations in hourly averaged O<sub>3</sub> on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004.**

Source: Fitz-Simons et al. (2005).

1 Northern Hemisphere during late winter and spring versus summer as for urban sites or for  
 2 nonurban sites heavily affected by regional pollution sources.

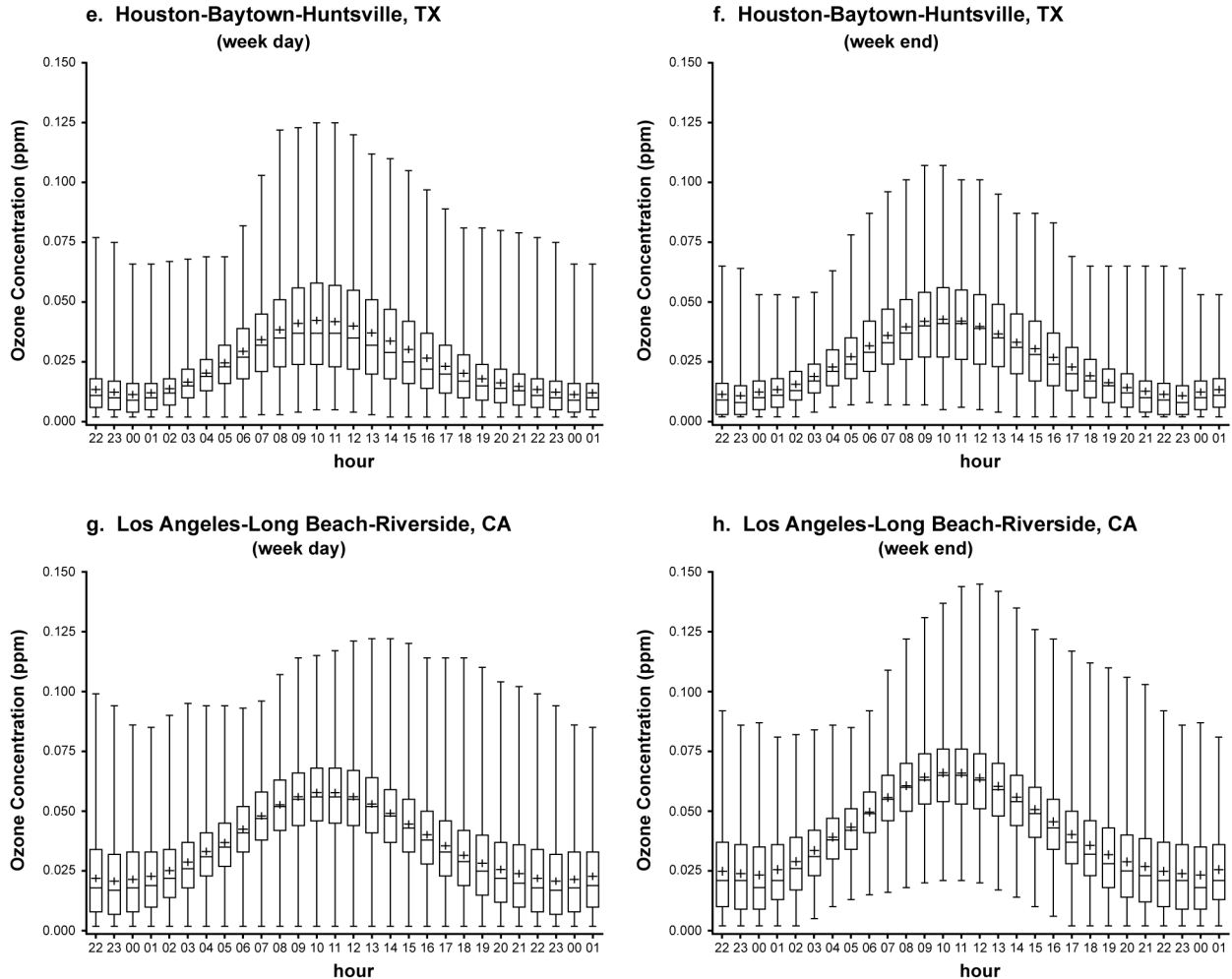
3 High O<sub>3</sub> values are also found at some of the 12 cities outside of summer. The seasonal  
 4 behavior of O<sub>3</sub> varies across the 12 cities. In most northern cities, the extreme values of the daily  
 5 maximum 8-h average O<sub>3</sub> concentration are a little more than half of those during the  
 6 warm season, the ratios of the medians are more similar as can be judged by comparison of



**Figure 3-13a-d. Diurnal variations in 8-h average  $O_3$  on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).

- 1 Figures 3-11a-l with Figures 3-14a-l. Differences are even smaller for the southern cities.
- 2 Indeed, some of the highest  $O_3$  values are found in the Houston CSA outside of summer
- 3 (Figure 3-14j).



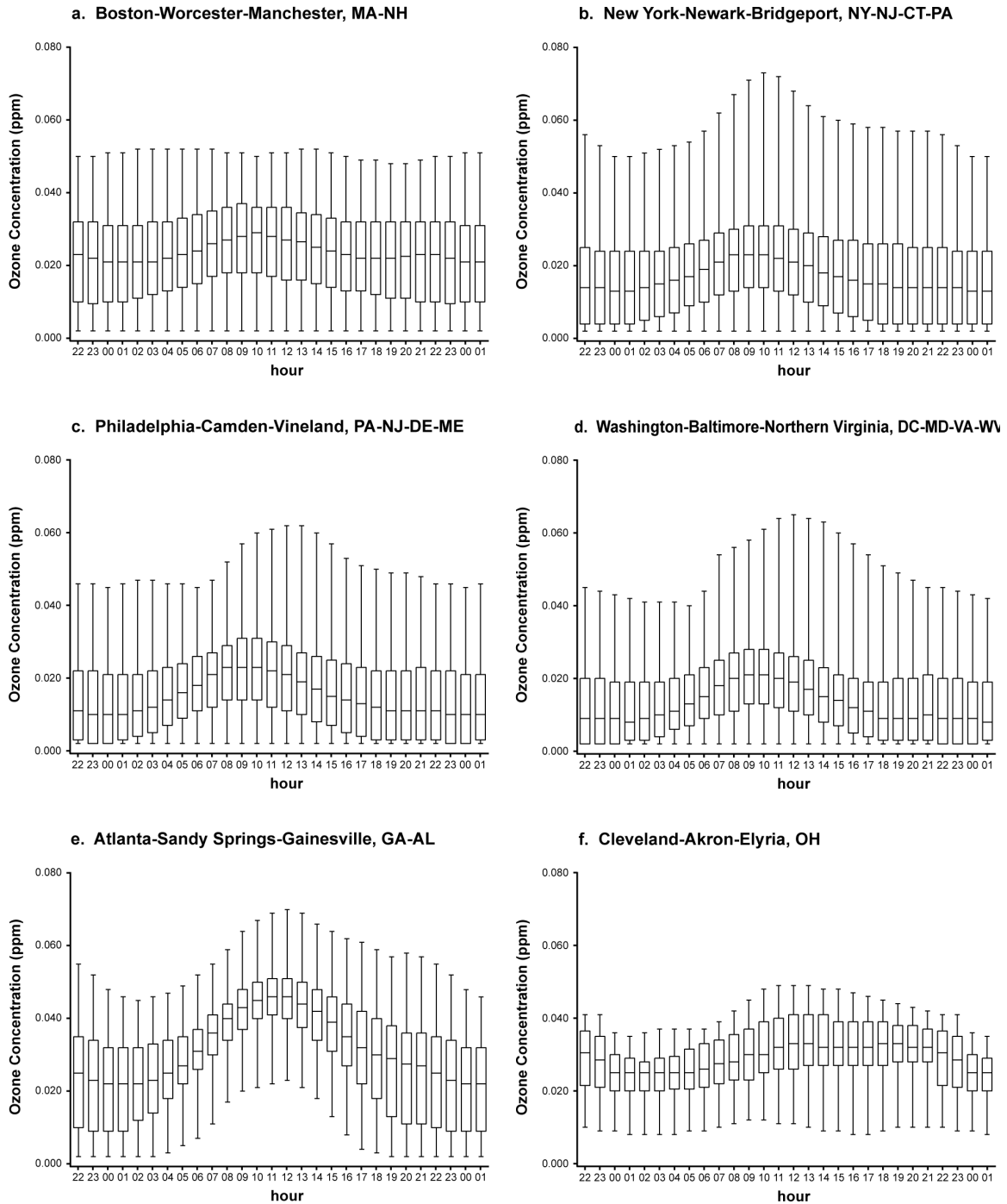
**Figure 3-13e-h. Diurnal variations in 8-h average O<sub>3</sub> on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).

1 ***Diurnal Patterns in Nonurban Areas***

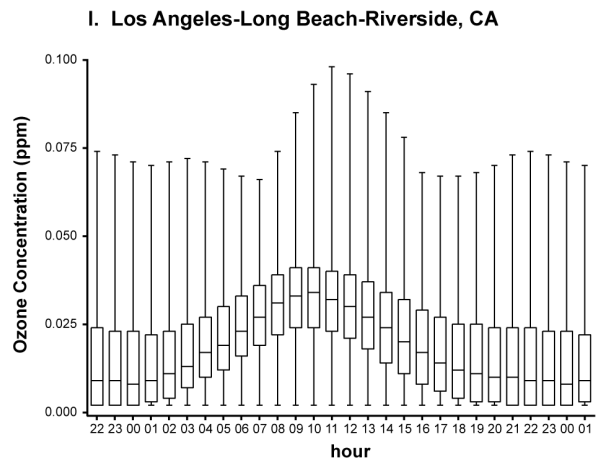
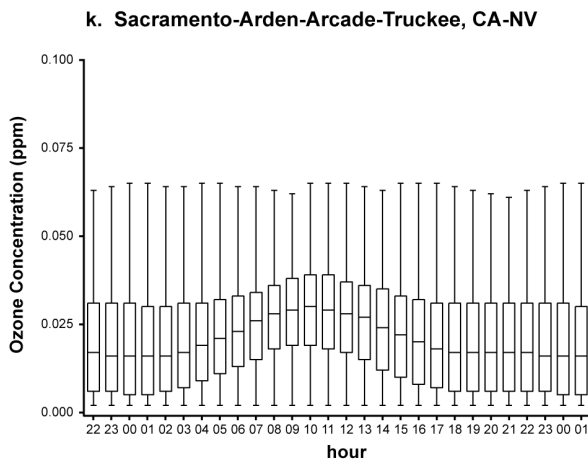
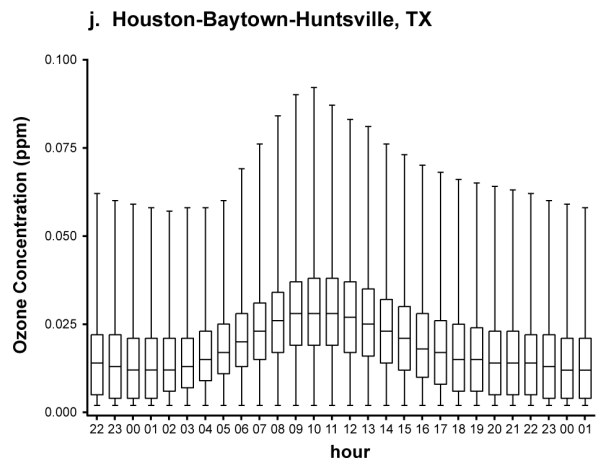
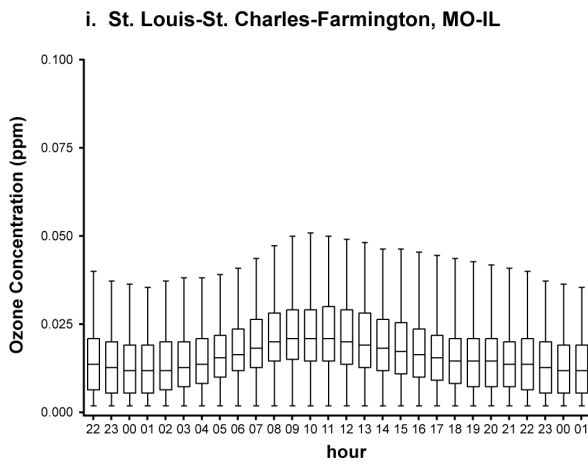
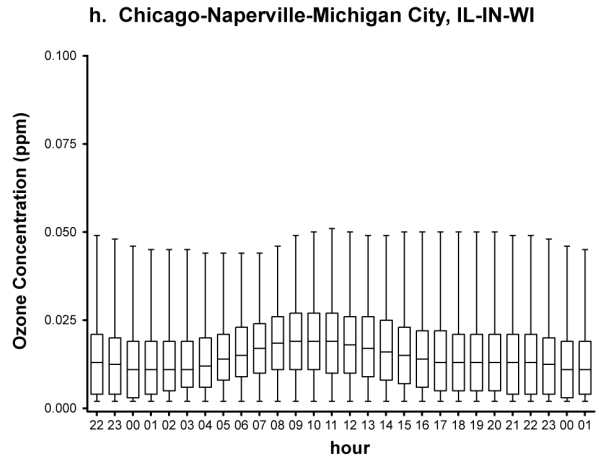
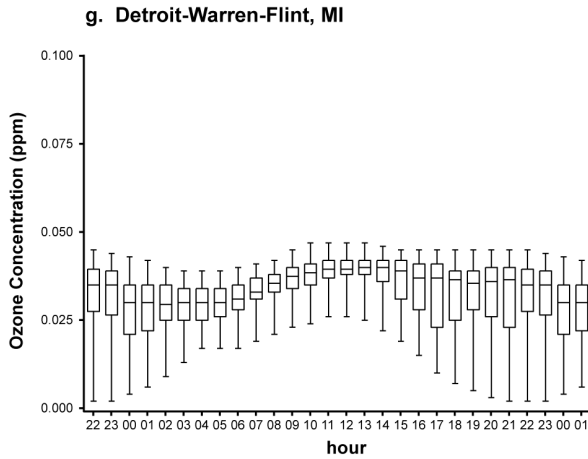
2 Composite diurnal patterns of O<sub>3</sub> are shown in Figure 3-15 for hourly averaged O<sub>3</sub>, and in  
 3 Figure 3-16 for 8 hour average O<sub>3</sub> at rural (CASTNET) sites. As can be seen from a comparison  
 4 of Figures 3-15 and 3-16 with Figures 3-8 and 3-9, diurnal patterns of O<sub>3</sub> are smoother and  
 5 shallower at the rural sites than at the urban sites. Maxima in hourly average O<sub>3</sub> also tend to  
 6 occur in afternoon. However, highest concentrations observed during any particular hour at  
 7 night at the CASTNET sites (~0.130 ppm) are substantially higher than observed in urban areas





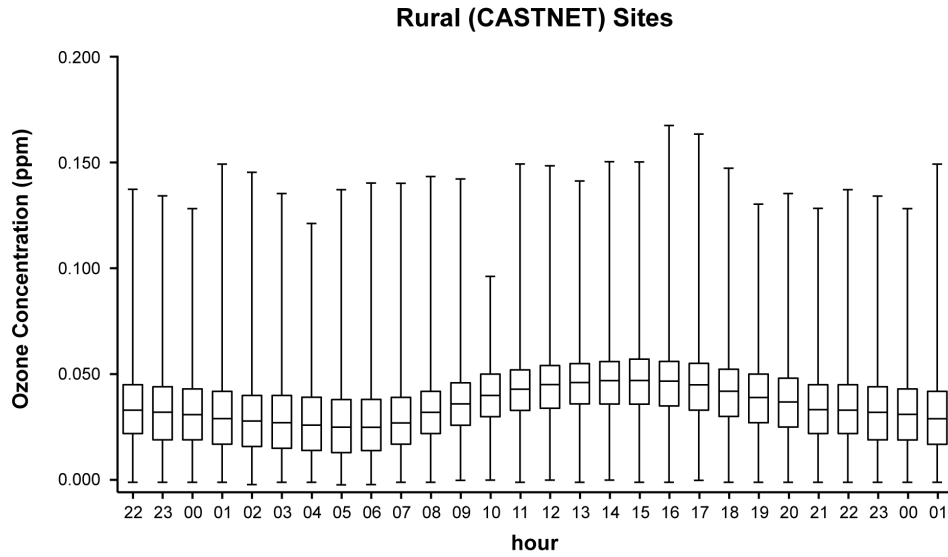
**Figure 3-14a-f. Diurnal variability in 8 hour averaged  $O_3$  in selected urban areas. Values shown are averages from November to March 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).



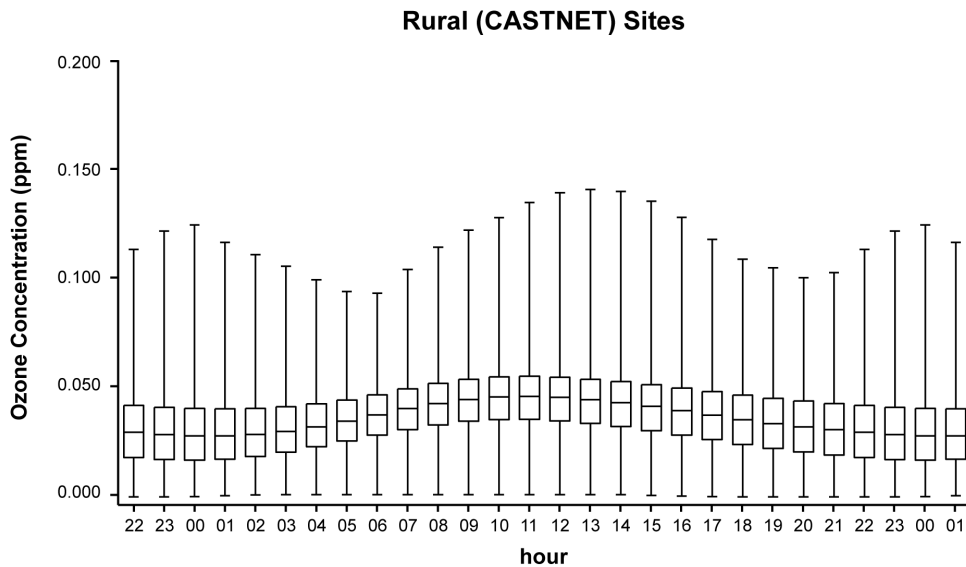
**Figure 3-14g-l. Diurnal variability in 8 hour averaged O<sub>3</sub> in selected urban areas. Values shown are averages from November to March 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima. The hour refers to the start of the 8-h averaging period.**

Source: Fitz-Simons et al. (2005).



**Figure 3-15. Composite diurnal variability in hourly O<sub>3</sub> concentrations observed at CASTNET sites. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima.**

Source: Fitz-Simons et al. (2005).



**Figure 3-16. Composite diurnal variability in 8-h O<sub>3</sub> concentrations observed at CASTNET sites. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers the minima and maxima. The hour refers to the start of the 8-h averaging period.**

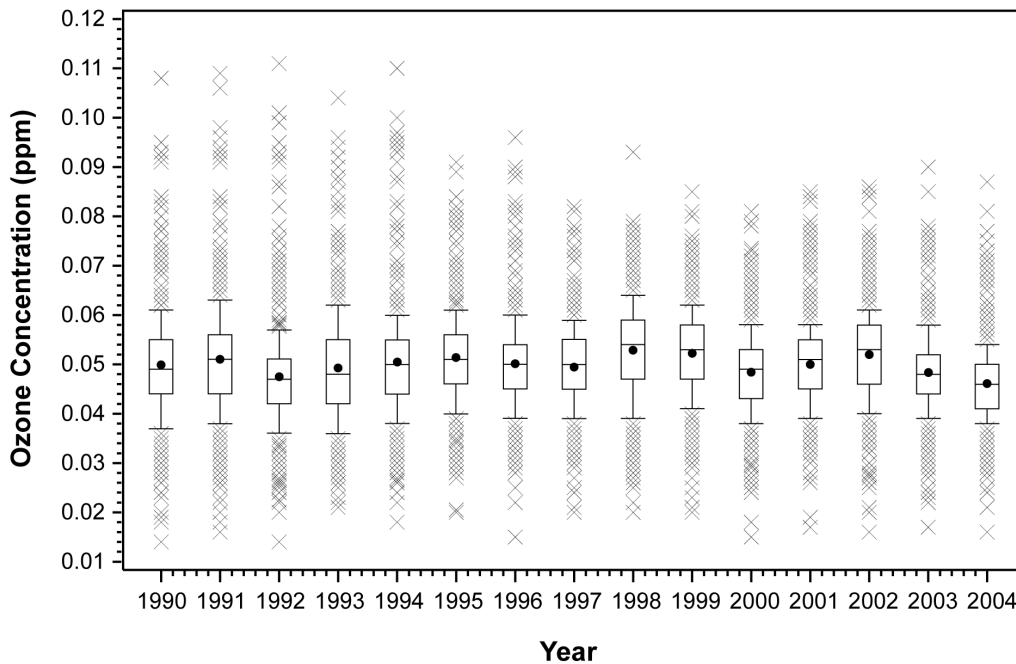
Source: Fitz-Simons et al. (2005).

1 (<0.100 ppm) and daily 1-h maxima at CASTNET sites have exceeded 0.150 ppm. The diurnal  
2 variations in 8-h average O<sub>3</sub> concentrations are also much smaller at the CASTNET sites than at  
3 the urban sites. Note also that the maxima in 8-h average O<sub>3</sub> concentrations are higher at the  
4 CASTNET sites than at the urban sites.

### 7 **3.5 TRENDS IN OZONE CONCENTRATIONS**

8 Year-to-year variability in the nationwide May to September, mean daily maximum 8-h O<sub>3</sub>  
9 concentrations are shown in Figure 3-17. The corresponding year-to-year variability in the 95th  
10 percentile concentrations is shown in Figure 3-18. Data flagged because of quality control issues  
11 were removed with concurrence by the local monitoring agency. Only days with data for 18 of  
12 24 hours were kept, and a minimum of 115 of 153 days were required in each year. Missing  
13 years were filled in using simple linear interpolation, as done in EPA Trends reports. Year-to-  
14 year variability in the 95th percentile values of the daily maximum 8-h O<sub>3</sub> concentrations are  
15 shown in Figure 3-18. Sites considered in this analysis are shown in the map in Figure AX3-3.  
16 As was shown in Figures 3-1 and 3-2, most sites are located in the East. As can be seen from  
17 Figure 3-17, the highest O<sub>3</sub> concentrations have tended to decrease over the past 15 years, while  
18 there has been little change in O<sub>3</sub> concentrations near the center of the distribution. This is  
19 consistent with observations in Europe (Volz-Thomas et al., 2003). Mean O<sub>3</sub> concentrations  
20 were slightly lower in 2003 and 2004 than in earlier years. The summer of 2003 was slightly  
21 cooler than normal in the East (Levinson and Waple, 2004) and the summer of 2004 was much  
22 cooler than normal in the East (Levinson, 2005) accounting in part for the dip in O<sub>3</sub> during these  
23 two years. Observations of O<sub>3</sub> at a number of sites in the Northern Hemisphere likewise do not  
24 show convincing evidence of strong upward trends during the 1990s (Oltmans et al., 1998).  
25 There may even have been a slight increase in O<sub>3</sub> concentrations at the lower end of the  
26 distribution throughout the monitoring period. This would be consistent with data obtained in  
27 Europe, showing that O<sub>3</sub> minima increased during the 1990s. Reduced titration of O<sub>3</sub> by reaction  
28 with NO in response to reductions in NO<sub>x</sub> emissions may be responsible in large measure for this  
29 finding. The concentration of O<sub>x</sub> (NO<sub>2</sub> + O<sub>3</sub>) shows little if any increase at all (Volz-Thomas  
30 et al., 2003).

**Nationwide Trends, May to September  
Mean of Daily Maximum 8-Hour Values, 1990 - 2004**

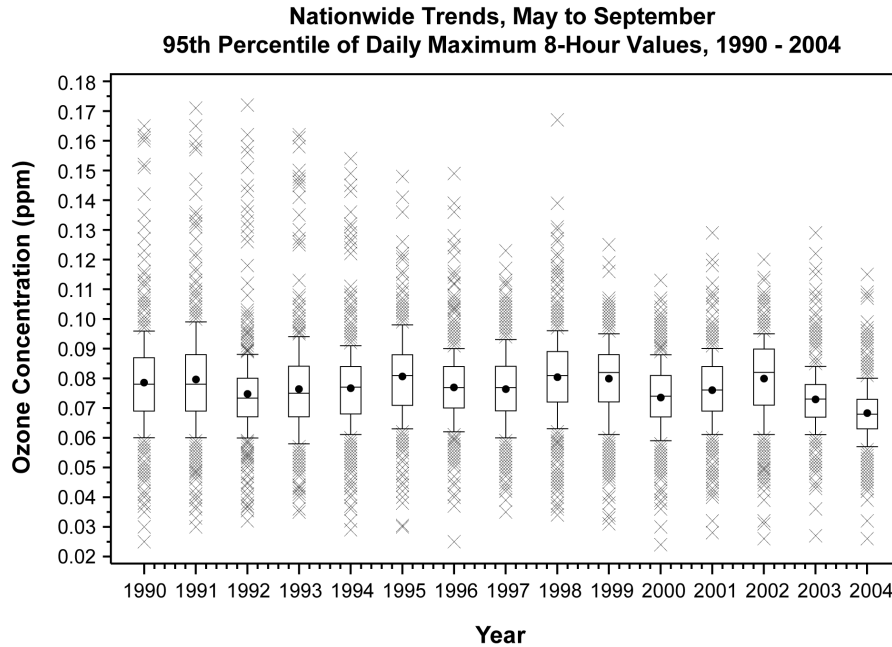


**Figure 3-17. Year-to-year variability in nationwide mean daily maximum 8-h O<sub>3</sub> concentrations. The whiskers on the box plot represent the 10th and 90th percentile concentrations. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile concentrations. The dots inside the box represent the mean, for the statistic, at all sites.**

Source: Fitz-Simons et al. (2005)

1 Trends in compliance metrics such as the fourth highest daily maximum 8-h O<sub>3</sub> concentration  
 2 can be found in the EPA Trends reports.

3 Figures 3-19a-h show year-to-year variability in mean daily 8-h O<sub>3</sub> concentrations  
 4 observed at selected national park sites across the United States. Figures 3-20a-h show year-to-  
 5 year variability in the 95th percentile value of daily maximum 8-h O<sub>3</sub> concentrations at the same  
 6 sites shown in Figures 3-19a-h. The same criteria used for calculating values in Figures 3-17  
 7 and 3-18 were used for calculating the May to September seasonal averages for the national  
 8 parks shown in Figures 3-19a-h and 3-20a-h. Sites at 22 national parks met these criteria, and  
 9 data for all 22 sites are given in Appendix AX3 in Figures AX3-66a-v and AX3-67a-v.



**Figure 3-18. Year-to-year variability in nationwide 95th percentile value of the daily maximum 8-h O<sub>3</sub> concentrations. The whiskers on the box plot represent the 10th and 90th percentile values for the statistic. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile values. The dots inside the box represent the mean, for the statistic, at all sites.**

Source: Fitz-Simons et al. (2005).

1 However, several monitoring sites were moved during the period from 1990 to 2004. Sites were  
 2 moved at Acadia NP in 1996, Joshua Tree NP in 1993, Mammoth Cave NP in 1996, Voyageurs  
 3 NP in 1996, and Yellowstone NP in 1996. These moves often resulted in offsets in O<sub>3</sub> and so  
 4 trends for these locations have not been calculated (cf., Section AX3.6, Table AX3-9). As noted  
 5 in The Ozone Report—Measuring Progress through 2003 (U.S. Environmental Protection  
 6 Agency, 2004b), O<sub>3</sub> trends in national parks in the South and the East are similar to nearby urban  
 7 areas and reflect the regional nature of O<sub>3</sub> pollution. For example, O<sub>3</sub> trends in Charleston, SC  
 8 and Charlotte, NC track those in nearby Cowpens NP and Cape Romaine NP in South Carolina;  
 9 O<sub>3</sub> in Knoxville and Nashville, TN tracks O<sub>3</sub> in Great Smoky NP; O<sub>3</sub> in Philadelphia, PA and  
 10 Baltimore, MD tracks Brigantine NP in New Jersey; and New York, NY and Hartford, CT track  
 11 O<sub>3</sub> in Cape Cod NS. The situation is not as clear in the West, where national parks are affected

1 differently by pollution sources that are located at varying distances away (e.g., Lassen Volcanic  
2 National Park and Yosemite National Park, CA). However, data obtained at these sites still  
3 provide valuable information about the variability in regional background concentrations,  
4 especially since the West has not been broken down into regions as has been done by Lehman  
5 et al. (2004) for the East and shown in Figure 3-3. Comparison of Figures 3-19a-h and 3-20a-h  
6 (in conjunction with Table AX3-9) shows that O<sub>3</sub> concentrations near the center of the  
7 distribution do not necessarily track those at the upper end, as pointed out earlier for nationwide  
8 composite data.

9 Caution should be exercised in using trends calculated at national parks to infer  
10 contributions from distant sources either inside or outside of North America, because of the  
11 influence of regional pollution. For example, using a 15-year record of O<sub>3</sub> from Lassen Volcanic  
12 NP and data from two aircraft campaigns, and observations spanning 18 years from five U.S.  
13 west coast marine boundary layer sites, Jaffe et al. (2003) have estimated that the amount of O<sub>3</sub>  
14 in air arriving from the Eastern Pacific in spring has increased by approximately 10 ppb from the  
15 mid-1980s to the present. This positive trend might be due to increases of emissions of O<sub>3</sub>  
16 precursors in Asia. Positive trends in O<sub>3</sub> were found during all seasons. Although the Lassen  
17 Volcanic NP site is not close to any major emission sources or urban centers, maximum hourly  
18 average O<sub>3</sub> concentrations of >0.080 ppm (during April-May) and >0.100 ppm (during the  
19 summer) occur at Lassen Volcanic NP. Thus, although there is evidence that O<sub>3</sub> levels may be  
20 increasing at some rural locations, there is also evidence that O<sub>3</sub> levels at other locations have  
21 either not increased or have decreased over the same period.

## 22 23 24 **3.6 RELATIONSHIPS BETWEEN OZONE AND OTHER SPECIES**

### 25 *Correlations between Ozone and other Species*

26 In order to understand relationships among atmospheric species, an important distinction  
27 must be made between primary (directly emitted) species and secondary (photochemically  
28 produced) species. In general, it is likely that primary species will be highly correlated with  
29 other primary species, and that secondary species will be highly correlated with other secondary  
30 species. By contrast, primary species are less likely to be correlated with secondary species.  
31 Secondary reaction products tend to correlate with each other, but there is considerable variation.

May to September Mean of Daily Maximum 8-Hour Values, 1990 - 2004

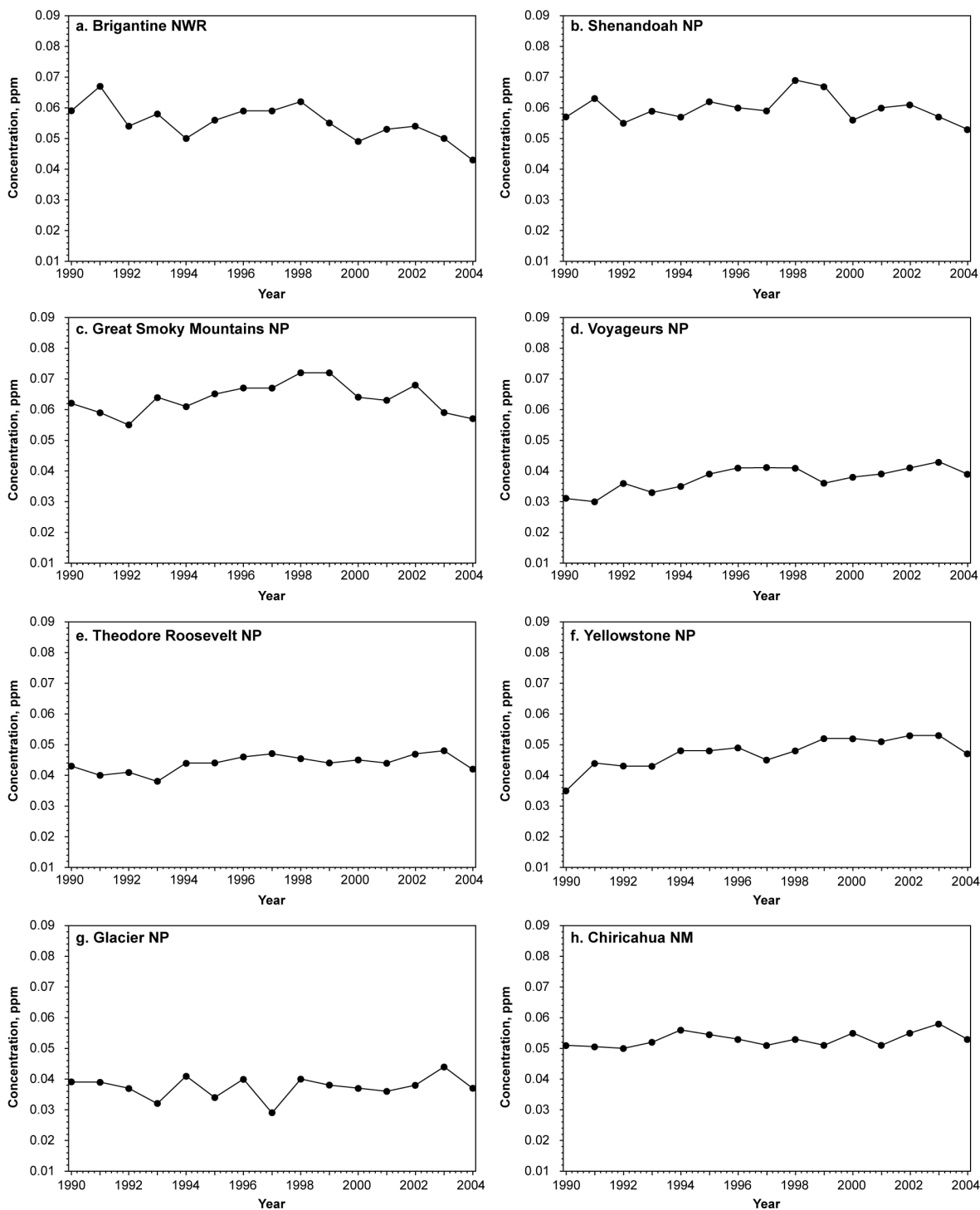


Figure 3-19a-h. Year-to-year variability in mean daily maximum 8-h O<sub>3</sub> concentrations at selected national park (NP), national wildlife refuge (NWR), and national monument (NM) sites.

Source: Fitz-Simons et al. (2005)



May to September 95th Percentile of Daily Maximum 8-Hour Values, 1990 - 2004

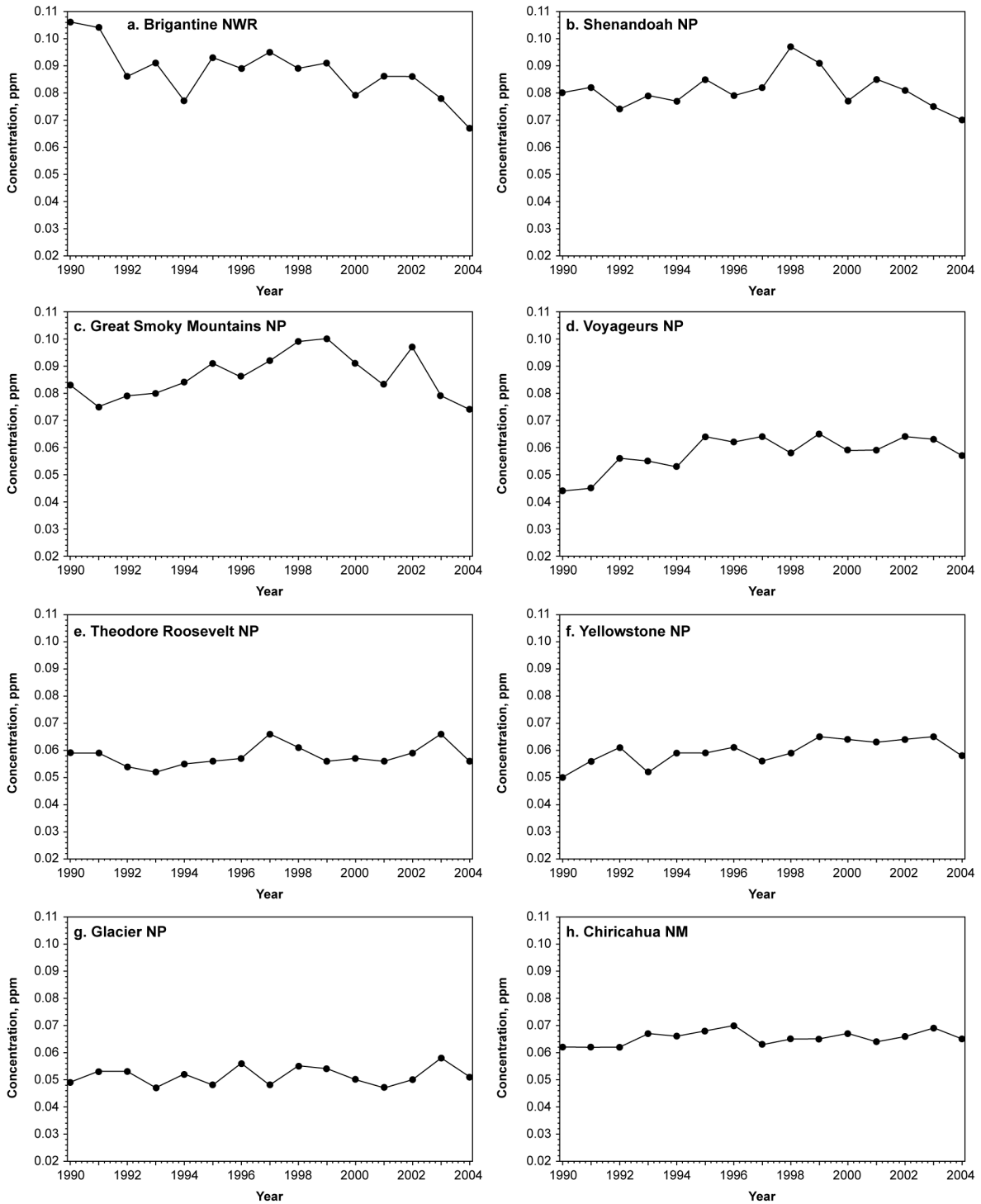


Figure 3-20a-h. Year-to-year variability in 95th percentile of daily maximum 8-h O<sub>3</sub> concentrations at selected national park (NP), national wildlife refuge (NWR), and national monument (NM) sites.

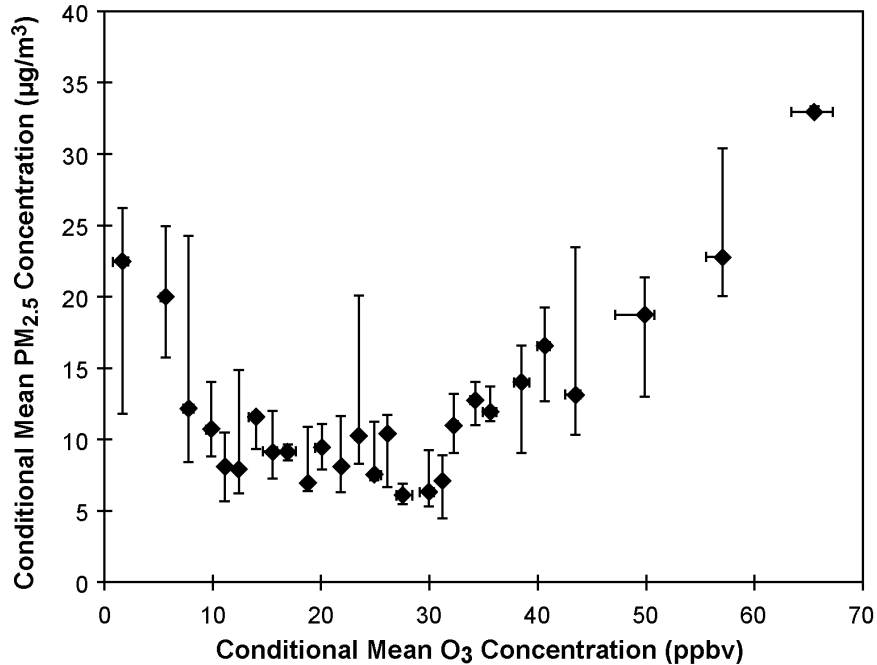
Source: Fitz-Simons et al. (2005).

1 Some species (e.g., O<sub>3</sub> and organic nitrates) are closely related photochemically and are highly  
2 correlated. Others (e.g., O<sub>3</sub> and H<sub>2</sub>O<sub>2</sub>) show a more complex correlation pattern. Further details  
3 are given in Annex AX3 in Section AX3.7.

4 Relationships between primary and secondary components are illustrated by considering  
5 data for O<sub>3</sub> and PM<sub>2.5</sub>. Ozone and PM<sub>2.5</sub> concentrations observed at a monitoring site in Fort  
6 Meade, MD are plotted as binned means for different intervals in Figure 3-21, based on data  
7 collected between July 1999 and July 2001. As can be seen from the figure, PM<sub>2.5</sub> tends to be  
8 negatively correlated with O<sub>3</sub> to the left of the inflection point (at about 30 ppbv O<sub>3</sub>) and tends to  
9 be positively correlated with O<sub>3</sub> to the right of the inflection point. Data to the left of the  
10 minimum in PM<sub>2.5</sub> were collected mainly during the cooler months of the year, while data to the  
11 right of the minimum were collected during the warmer months. This situation arises because  
12 PM<sub>2.5</sub> contains a large secondary component during the summer and has a larger primary  
13 component during winter. During the winter, O<sub>3</sub> comes mainly from the free troposphere, above  
14 the planetary boundary layer and, thus, may be considered a tracer for relatively clean air, and it  
15 is titrated by NO in the polluted boundary layer. Unfortunately, data for PM<sub>2.5</sub> and O<sub>3</sub> are  
16 collected concurrently at relatively few U.S. sites throughout an entire year. So these results,  
17 while highly instructive, are not readily extrapolated to areas where appreciable photochemical  
18 activity occurs throughout the year. Ito et al. (2005) examined the relation between PM<sub>10</sub> and O<sub>3</sub>  
19 on a seasonal basis in several urban areas (cf., Figure 7-24). Although PM<sub>10</sub> contains  
20 proportionately more primary material than does PM<sub>2.5</sub>, relations similar to those shown in  
21 Figure 3-21 are found, reflecting the dominant contribution of PM<sub>2.5</sub> to PM<sub>10</sub>.

### 22 23 ***Other Oxidants***

24 Measurements of gas phase peroxides in the atmosphere were reviewed by Lee et al.  
25 (2000). Ground level measurements of H<sub>2</sub>O<sub>2</sub> taken during the 1970s indicated values of 180 ppb  
26 in Riverside, CA and 10 to 20 ppb during smog episodes in Claremont and Riverside, with  
27 values approaching 100 ppb in forest fire plumes. However, later surface measurements always  
28 found much lower values. For example, in measurements made in Los Angeles and nearby areas  
29 in the 1980s, peak values were always less than about 2 ppb and in a methods intercomparison  
30 study in Research Triangle Park, NC in June 1986, concentrations were <2.5 ppbv. Higher



**Figure 3-21. Binned mean PM<sub>2.5</sub> concentrations versus binned mean O<sub>3</sub> concentrations observed at Fort Meade, MD from July 1999 to July 2001.**

Source: Chen (2002).

1 values ranging up to 5 ppb were found in a few other studies in Kinterbish, Alabama and  
 2 Meadview, Arizona. Several of these studies found strong diurnal variations (typically about a  
 3 factor of three) with maximum values in the mid-afternoon and minimum values in the early  
 4 morning. Mean concentrations of organic hydroperoxides at the surface at Niwot Ridge, CO in  
 5 the summer of 1988 and State Park, GA during the summer of 1991 were all less than a few ppb.

6 Aircraft measurements of hydroperoxide (H<sub>2</sub>O<sub>2</sub>, CH<sub>3</sub>OOH and HOCH<sub>2</sub>OOH)  
 7 concentrations were made as part of the Southern Oxidants Study intensive campaign in  
 8 Nashville, TN in July 1995 (Weinstein-Lloyd et al., 1998). The median concentration of total  
 9 hydroperoxides in the boundary layer between 1100 and 1400 CDT was about 5 ppbv, with more  
 10 than 50% contribution from organic hydroperoxides. Median O<sub>3</sub> was about 70 ppbv at the same  
 11 time. The concentrations of the hydroperoxides depended strongly on wind direction with values  
 12 about 40% lower when winds originated from the N/NW as opposed to the S/SW suggesting that  
 13 local source areas were important.

1 Peroxyacetylnitrate (PAN) is produced during the photochemical oxidation of a wide range  
2 of VOCs in the presence of  $\text{NO}_x$ . It is removed by thermal decomposition and also by uptake to  
3 vegetation (Sparks et al., 2003; Teklemariam and Sparks, 2004). PAN is the dominant member  
4 of the broader family of peroxyacylnitrates (PANs) which includes as other significant  
5 atmospheric components peroxypropionyl nitrate (PPN) of anthropogenic origin, and  
6 peroxyacrylic nitrate (MPAN) produced from oxidation of isoprene. Measurements and  
7 models show that PAN in the United States includes major contributions from both  
8 anthropogenic and biogenic VOC precursors (Horowitz et al., 1998; Roberts et al., 1998).  
9 Measurements in Nashville during the 1999 summertime Southern Oxidants Study (SOS)  
10 showed PPN and MPAN amounting to 14% and 25% of PAN respectively (Roberts et al., 2002).  
11 Measurements during the TexAQS 2000 study in Houston indicated PAN concentrations of up to  
12 6.5 ppbv (Roberts et al., 2003). PAN measurements in southern California during the  
13 SCOS97-NARSTO study indicated peak concentrations of 5-10 ppbv, which can be contrasted to  
14 values of 60 to 70 ppbv measured back in 1960 (Grosjean, 2003). Vertical profiles measured  
15 from aircraft over the U.S. and off the Pacific coasts show PAN concentrations above the  
16 boundary layer of only a few hundred pptv, although there are significant enhancements  
17 associated with long-range transport of pollution plumes including from Asia (Kotchenruther  
18 et al., 2001a; Roberts et al., 2004). Decomposition of this anthropogenic PAN as it subsides  
19 over North America can lead to significant  $\text{O}_3$  production, enhancing the  $\text{O}_3$  background  
20 (Kotchenruther et al., 2001b; Hudman et al., 2004).

21 Oxidants are also present in airborne cloud droplets, rain drops and particulate matter.  
22 Measurements of hydroperoxides, summarized by Reeves (2003), are available mainly for  
23 hydrometeors, but are sparse for ambient particles. Venkatachari et al. (2005a) sampled the  
24 concentrations of total reactive oxygen species (ROS) in particles using a cascade impactor in  
25 Rubidoux, CA during July 2003. Although the species constituting ROS were not identified, the  
26 results were reported in terms of equivalent  $\text{H}_2\text{O}_2$  concentrations. Unlike  $\text{O}_3$  and gas phase  $\text{H}_2\text{O}_2$   
27 which show strong diurnal variability (i.e., about a factor of three variation between afternoon  
28 maximum and early morning minimum), the diurnal variation of particle phase ROS was found  
29 to be much weaker (i.e., less than about 20%) at least for the time between 8 a.m. and midnight.  
30 Because the ROS were measured in the fine aerosol size fraction, which has a lifetime with  
31 respect to deposition of much greater than a day, little loss is expected but their concentrations

1 might also be expected to increase because of nighttime chemistry, perhaps involving NO<sub>3</sub>  
2 radicals. The ROS concentration, about  $7 \times 10^{-9}$  M/m<sup>3</sup> (expressed as equivalent H<sub>2</sub>O<sub>2</sub>), was at  
3 most 1% that of O<sub>3</sub> ( $6.2$  to  $38 \times 10^{-7}$  M/m<sup>3</sup> or 15 to 90 ppb), with highest values at night. In a  
4 companion study conducted in Queens, NY during January and early February 2004,  
5 Venkatachari et al. (2005b) found much lower concentrations of ROS of about  $1.2 \times 10^{-9}$  M/m<sup>3</sup>.  
6 However, O<sub>3</sub> levels were also substantially lower, but ROS concentrations were still less than  
7 1% those of O<sub>3</sub>. It is of interest to note that gas phase OH concentrations measured at the same  
8 time ranged from about  $7.5 \times 10^4$ /cm<sup>3</sup> to about  $1.8 \times 10^6$ /cm<sup>3</sup>, implying the presence of  
9 significant photochemical activity even in New York City during winter.

### 11 *Co-occurrence of Ozone with Other Pollutants*

12 The characterization of co-occurrence patterns under ambient conditions is important for  
13 relating human health and vegetation effects under ambient conditions to controlled research  
14 results as described in Annex AX3.8. Several attempts have been made to characterize gaseous  
15 air pollutant mixtures. The previous 1996 O<sub>3</sub> AQCD discussed various patterns of pollutant  
16 mixtures of SO<sub>2</sub>, NO<sub>2</sub>, and O<sub>3</sub>. Pollutant combinations can occur at or above a threshold  
17 concentration at either the same or different times.

18 The 1996 O<sub>3</sub> AQCD noted that studies of the joint occurrence of gaseous NO<sub>2</sub>/O<sub>3</sub> and  
19 SO<sub>2</sub>/O<sub>3</sub> reached two conclusions: (1) hourly simultaneous and daily simultaneous-only  
20 co-occurrences are fairly rare (when both pollutants were present at an hourly average  
21 concentration  $\geq 0.05$  ppm) and (2) when co-occurrences are present, complex-sequential and  
22 sequential-only co-occurrence patterns predominate. Year-to-year variability was found to be  
23 insignificant.

24 Using 2001 hourly data for O<sub>3</sub> and NO<sub>2</sub> and for O<sub>3</sub> and SO<sub>2</sub>, the co-occurrence patterns for  
25 the data are similar to those of previous studies. As shown in Figure 3-22, fewer than  
26 10 co-occurrences of O<sub>3</sub> and NO<sub>2</sub> were found for most of the collocated monitoring sites.  
27 Likewise, Figure 3-23 shows that fewer than 10 co-occurrences of O<sub>3</sub> and SO<sub>2</sub> were found for  
28 most of the collocated monitoring sites analyzed.

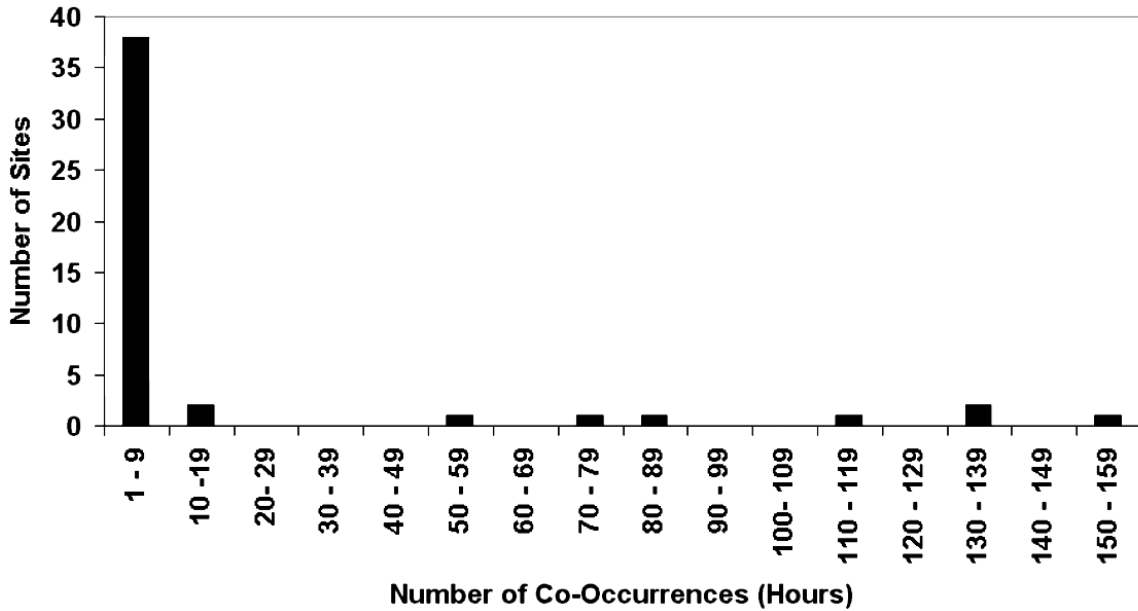


Figure 3-22. The co-occurrence pattern for O<sub>3</sub> and nitrogen dioxide using 2001 data from the AQS. There is co-occurrence when hourly average concentrations of O<sub>3</sub> and another pollutant are both ≥0.05 ppm.

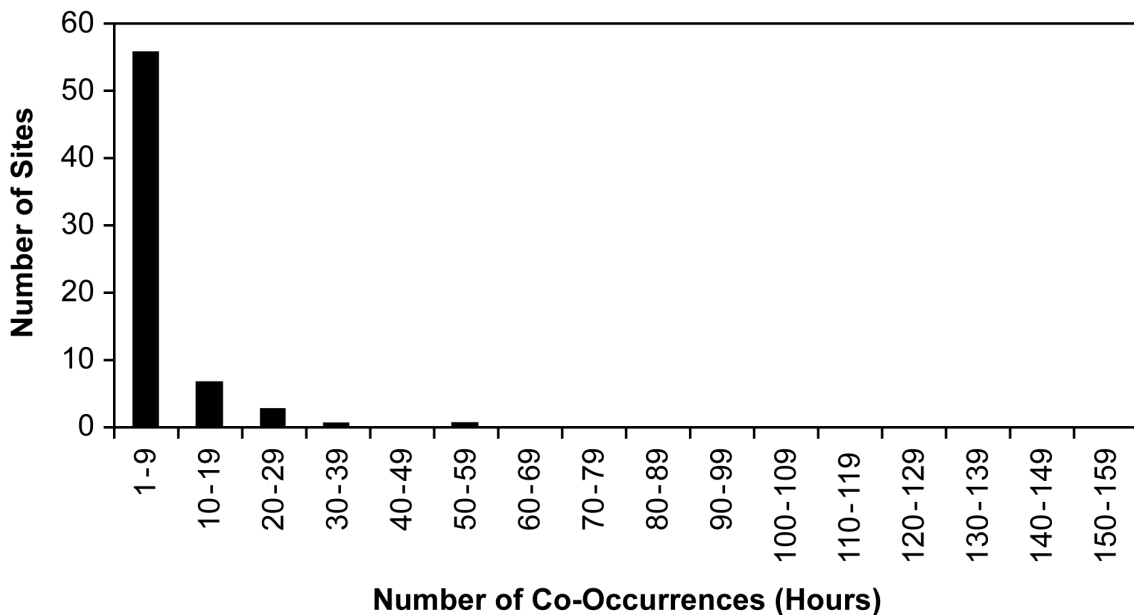
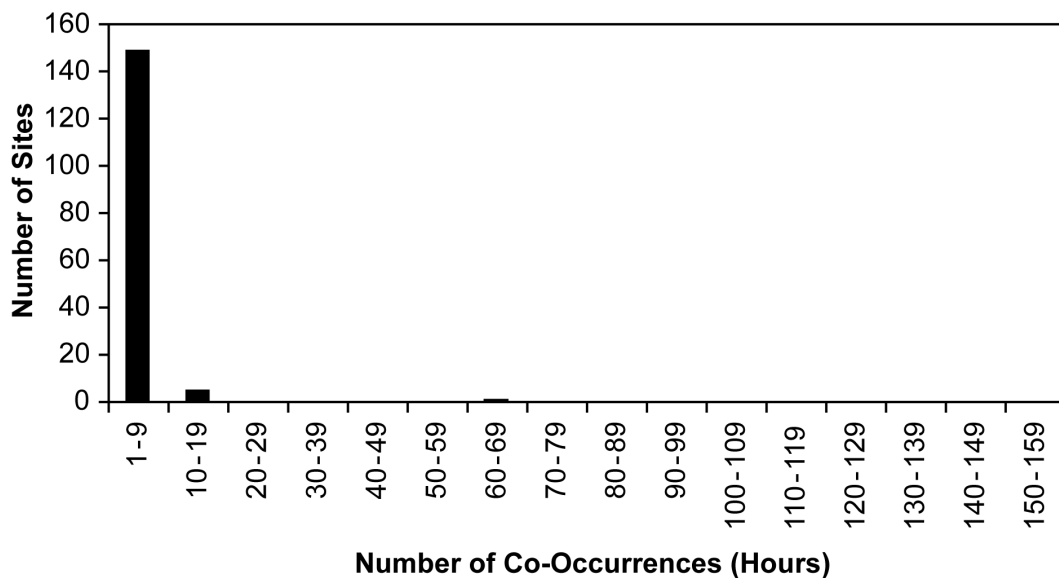


Figure 3-23. The co-occurrence pattern for O<sub>3</sub> and sulfur dioxide using 2001 data from AQS. There is co-occurrence when hourly average concentrations of O<sub>3</sub> and another pollutant are both ≥0.05 ppm.

1 Since 1999, monitoring stations across the United States have routinely measured 24-h  
 2 average concentrations for PM<sub>2.5</sub>. Daily co-occurrence of PM<sub>2.5</sub> and O<sub>3</sub> over a 24-h period was  
 3 also characterized. Because PM<sub>2.5</sub> data are mostly summarized as 24-h average concentrations in  
 4 the AQS database, a daily co-occurrence of O<sub>3</sub> and PM<sub>2.5</sub> was subjectively defined as an hourly  
 5 average O<sub>3</sub> concentration ≥0.05 ppm and a PM<sub>2.5</sub> 24-h concentration ≥40 μg/m<sup>3</sup> (corresponding  
 6 to the EPA Air Quality Index, Level of Concern for PM<sub>2.5</sub>) occurring during the same 24-h  
 7 period. Using 2001 data from the AQS database, the daily co-occurrence of PM<sub>2.5</sub> and O<sub>3</sub> was  
 8 infrequent (Figure 3-24). Only limited data are available on the co-occurrence of O<sub>3</sub> and other  
 9 pollutants (e.g., acid precipitation and acidic cloudwater). In most cases, routine monitoring data  
 10 are not available from which to draw general conclusions.



**Figure 3-24. The co-occurrence pattern for O<sub>3</sub> and PM<sub>2.5</sub> using 2001 data from AQS.**

### 1 **3.7 POLICY RELEVANT BACKGROUND OZONE CONCENTRATIONS**

2 Background O<sub>3</sub> concentrations used for NAAQS-setting purposes are referred to as Policy  
 3 Relevant Background (PRB) O<sub>3</sub> concentrations. Policy Relevant Background concentrations are  
 4 those concentrations that would occur in the United States in the absence of anthropogenic  
 5 emissions in continental North America (defined here as the United States, Canada, and  
 6 Mexico). Policy Relevant Background concentrations include contributions from natural sources

1 everywhere in the world and from anthropogenic sources outside these three countries. For the  
2 purposes of informing decisions about O<sub>3</sub> NAAQS, EPA assesses risks to human health and  
3 environmental effects from O<sub>3</sub> levels in excess of PRB concentrations. Issues concerning the  
4 methodology for estimating PRB O<sub>3</sub> concentrations are described in detail in Annex AX3,  
5 Section AX3.9.

6 Contributions to PRB O<sub>3</sub> include photochemical actions involving natural emissions of  
7 VOCs, NO<sub>x</sub>, and CO as well as the long-range transport of O<sub>3</sub> and its precursors from outside  
8 North America and the stratospheric-tropospheric exchange (STE) of O<sub>3</sub>. Processes involved in  
9 STE are described in detail in Annex AX2.3. Natural sources of O<sub>3</sub> precursors include biogenic  
10 emissions, wildfires, and lightning. Biogenic emissions from agricultural activities are not  
11 considered in the formation of PRB O<sub>3</sub>.

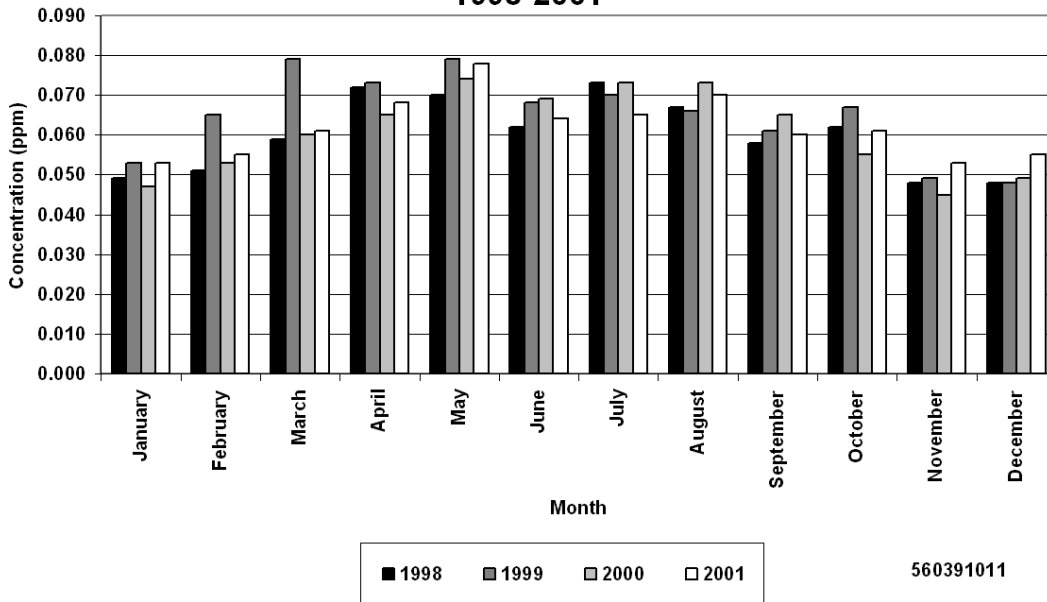
12 Springtime maxima are observed at relatively remote (Annex AX3 and Figures 3-25a,b)  
13 national park sites, located mainly in the western United States and at a number of other  
14 relatively unpolluted monitoring sites throughout the Northern Hemisphere. The major issues  
15 concerning the calculation of PRB O<sub>3</sub> center on the capability of the current generation of global-  
16 scale, three-dimensional, CTMs to simulate the causes of high O<sub>3</sub> concentrations observed at  
17 monitoring sites in relatively unpolluted areas of the United States from late winter through  
18 spring (i.e., February through June). The issues raised do not affect interpretations of the causes  
19 of summertime O<sub>3</sub> episodes as strongly. Summertime O<sub>3</sub> episodes are mainly associated with  
20 slow- moving high-pressure systems characterized by limited mixing between the planetary  
21 boundary layer and the free troposphere, as noted in Annex AX2, Section AX2.3.

22 A large number of case studies document the occurrence of STE mainly during winter and  
23 spring in mid- and high-latitudes in Europe, Asia, and North America. These studies were based  
24 on aircraft, satellite, and ground-based measurements. Considerable uncertainty exists in the  
25 magnitude of the exchange; however, these studies have found that STE occurs throughout the  
26 year, but with a distinct preference for the transport of O<sub>3</sub> directly to the middle and lower  
27 troposphere during late winter and spring. Transport to the upper troposphere occurs throughout  
28 the year.

29 Springtime maxima in tropospheric O<sub>3</sub> observed at high latitudes are also associated with  
30 the winter buildup of O<sub>3</sub> precursors and thermally labile reservoir species, such as PAN and  
31 other reactive nitrogen species. These pollutants originate from all continents in the Northern

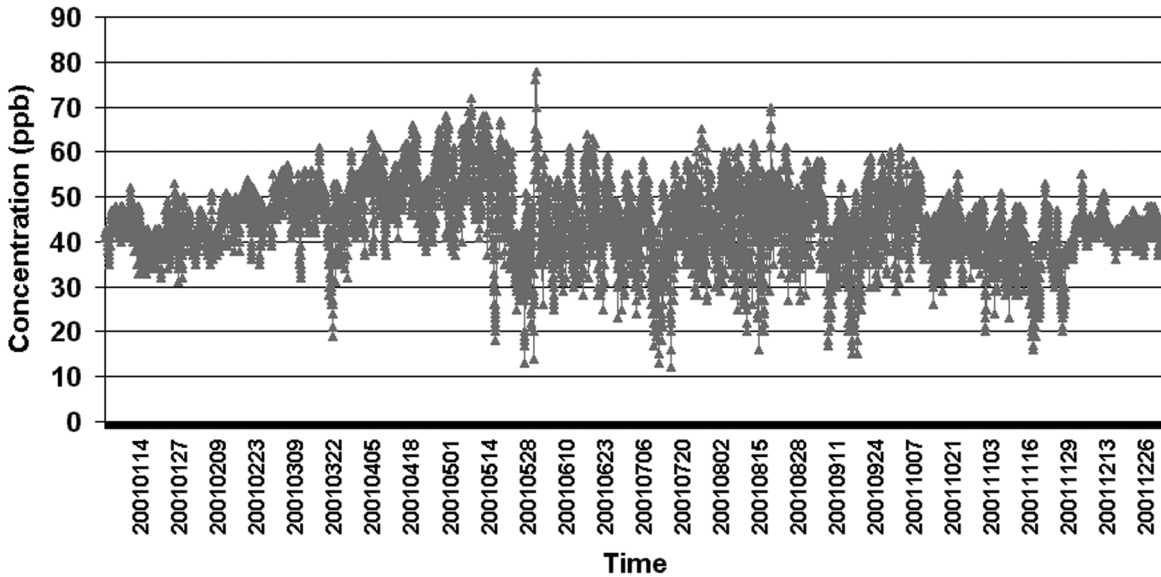


**Yellowstone National Park  
Maximum Hourly Concentration  
1998-2001**



**Figure 3-25a. Monthly maximum hourly average O<sub>3</sub> concentrations at Yellowstone National Park (WY) in 1998, 1999, 2000, and 2001.**

Source: U.S. Environmental Protection Agency (2003a).



**Figure 3-25b. Hourly average O<sub>3</sub> concentrations at Yellowstone National Park (WY) for the period January to December 2001.**

Source: U.S. Environmental Protection Agency (2003a).

1 Hemisphere. Ozone precursor concentrations reach a maximum in late March; and as sunlight  
2 returns to the Arctic, photochemical reactions generate tropospheric O<sub>3</sub> (Section AX3.9.1).  
3 The contribution of Asian sources to the U.S. levels is also largest during spring, reflecting the  
4 efficient lifting of Asian pollution ahead of cold fronts originating in Siberia and transport by  
5 strong westerly winds across the Pacific (e.g., Hudman et al., 2004). The longer lifetime of O<sub>3</sub>  
6 during spring also contributes to springtime maxima (Wang et al., 1998).

7 Estimates of PRB concentrations cannot be obtained solely by examining measurements of  
8 O<sub>3</sub> obtained at RRMS in the United States (Annex AX3, Section AX3.2.3) because of the  
9 long-range transport from anthropogenic source regions within North America. It should also be  
10 noted that it is impossible to determine sources of O<sub>3</sub> without ancillary data that could be used as  
11 tracers of sources or to calculate photochemical production and loss rates. The current definition  
12 of PRB implies that only CTMs can be used to estimate the range of PRB values. On the  
13 synoptic and larger spatial scales at least, all evidence indicates that global CTMs are adequate  
14 tools to investigate the factors controlling tropospheric O<sub>3</sub>; and three-dimensional CTMs, as  
15 typified by Fiore et al. (2003) appear to offer the best methodology for estimating PRB  
16 concentrations that cannot be measured directly (Annex AX3, Section AX3.9.2), at least for  
17 averaging periods of longer than one hour.

18 Previous estimates of background O<sub>3</sub> concentrations, based on different concepts of  
19 background, are given in Table 3-2. Results from global three-dimensional CTMs, where the  
20 background is estimated by zeroing anthropogenic emissions in North America (Table 3-8) are  
21 on the low end of the 25 to 45 ppbv range. Lefohn et al. (2001) have argued that frequent  
22 occurrences of O<sub>3</sub> concentrations above 50 to 60 ppbv at remote northern U.S. sites in spring are  
23 mainly stratospheric in origin. Fiore et al. (2003) used a global CTM to determine the origin of  
24 the high-O<sub>3</sub> events reported by Lefohn et al. (2001), and to conduct a more general quantitative  
25 analysis of background O<sub>3</sub> as a function of season, altitude, and local O<sub>3</sub> concentration.

26 Figure 3-26 shows a comparison between observations obtained at CASTNet sites and  
27 model results of Fiore et al. (2003). They classified the CASTNet monitoring sites into  
28 low-lying sites (generally <1.5 km) and elevated sites (>1.5 km). All elevated sites are in the  
29 West. Results were then aggregated to construct the cumulative probability distributions shown  
30 in Figure 3-26 for the 58 low-altitude sites and the 12 high-altitude sites as well as for the three  
31 seasons. The calculated mean background at the surface sites in spring is 27 ppbv, compared to

**Table 3-2. Previous Estimates of Background O<sub>3</sub> in Surface Air Over the United States**

Study	Method	Time Period	Region	Background Estimate (ppbv)
Trainer et al. (1993)	y-intercept of O <sub>3</sub> vs. NO <sub>y</sub> -NO <sub>x</sub> regression line <sup>a</sup>	Summer 1988	Eastern United States	30-40 <sup>b</sup>
Hirsch et al. (1996)	y-intercept of O <sub>3</sub> vs. NO <sub>y</sub> -NO <sub>x</sub> regression line	May-Sep 1990-1994	Harvard Forest <sup>c</sup>	25 (Sept) – 40 (May) <sup>d</sup>
Altshuller and Lefohn (1996)	y-intercept of O <sub>3</sub> vs. NO <sub>y</sub> regression line, and observations at remote/rural sites	Apr-Oct 1988-1993	Continental United States	25-45 (inland) <sup>e</sup> 25-35 (coastal)
Liang et al. (1998)	Sensitivity simulation in a 3-D model with anthropogenic NO <sub>x</sub> emissions in the continental U.S. set to zero	Full year	Continental United States	20-30 (East) <sup>f</sup> 20-40 (West) (spring maximum)
Lin et al. (2000)	Median O <sub>3</sub> values for the lowest 25th percentiles of CO and NO <sub>y</sub> concentrations	1990-1998	Harvard Forest	35 (fall) – 45 (spring) <sup>g</sup>
Fiore et al. (2002)	O <sub>3</sub> produced outside of the North American boundary layer in a global 3-D model	Summer 1995	Continental United States	15-30 (East) <sup>h</sup> 25-35 (West)

<sup>a</sup>NO<sub>y</sub> is the chemical family including NO<sub>x</sub> and its oxidation products; NO<sub>y</sub>-NO<sub>x</sub> denotes the chemically processed component of NO<sub>y</sub>.

<sup>b</sup> 1300-1700 local time (LT) in flatland and valley sites; all daytime measurements at elevated sites.

<sup>c</sup> rural site in central Massachusetts.

<sup>d</sup> 1100-1700 EST hourly means.

<sup>e</sup> seasonal 7-h (0900-1559) daylight average.

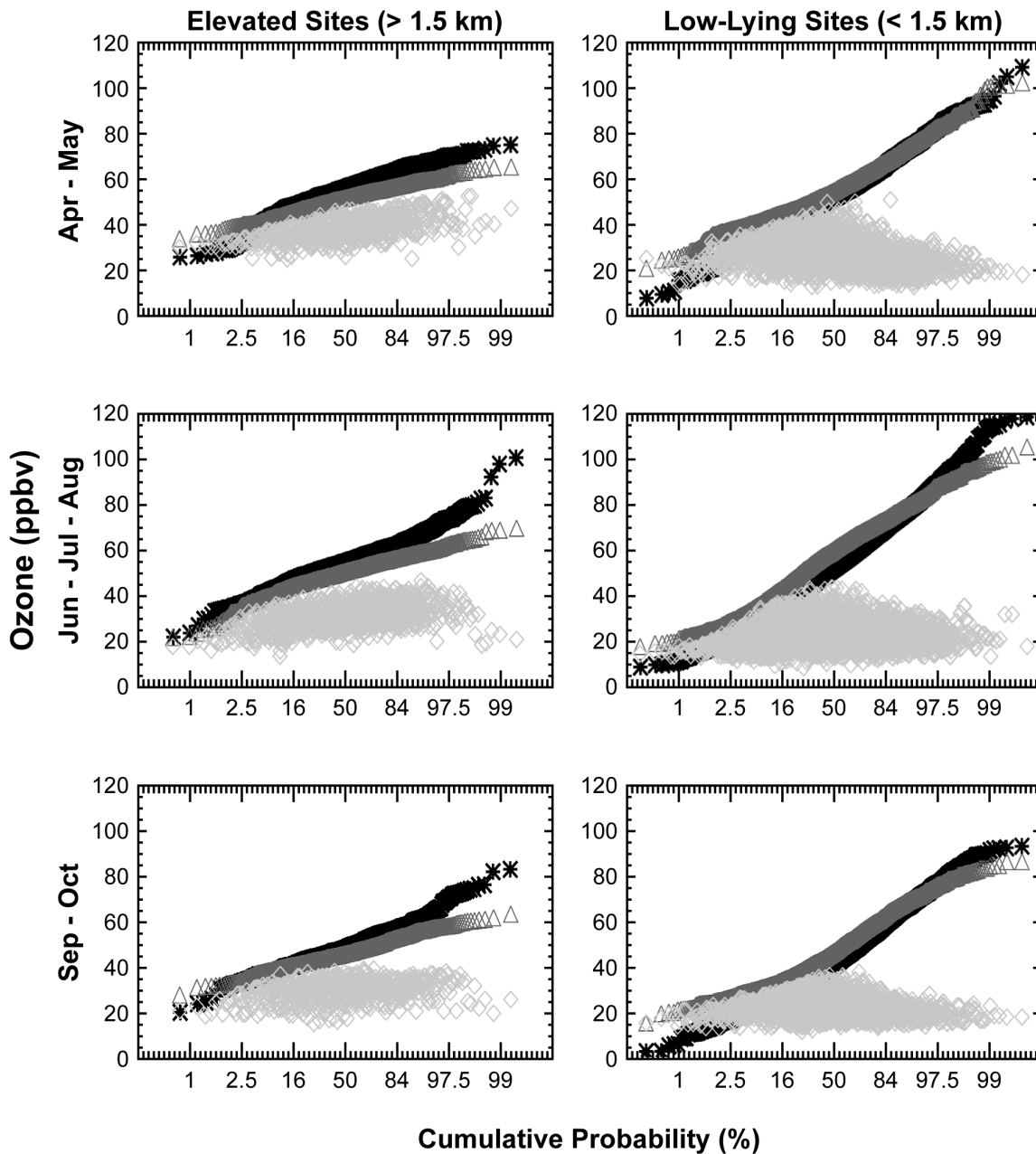
<sup>f</sup> 1300-1600 LT monthly mean.

<sup>g</sup> daily max 8-h averages.

<sup>h</sup> 1300-1700 average.

Source: Fiore et al. (2003).

1 23 ppbv in summer and fall. At these sites, the background is highest for O<sub>3</sub> concentrations near  
2 the center of the distribution, and it declines as total surface O<sub>3</sub> concentrations increase, for  
3 reasons summarized below and discussed by Fiore et al. (2002). The observed O<sub>3</sub> concentration  
4 thus serves a surrogate for meteorological variability (i.e., stagnant versus ventilated conditions),  
5 such that the background O<sub>3</sub> is smaller on days when total O<sub>3</sub> is highest. At the elevated sites,  
6 the calculated mean background is 36 ppbv in spring versus 30 ppbv in the summer and fall.  
7 Background concentrations in the fall resemble those in summer but show less variability and do  
8 not exceed 40 ppbv anywhere in this analysis.



**Figure 3-26.** Estimates of background contribution to surface afternoon (13 to 17 LT) O<sub>3</sub> concentrations in the United States as a function of local O<sub>3</sub> concentration, site altitude, and season. The figure shows cumulative probability distributions of O<sub>3</sub> concentrations for the observations (asterisks) and the model (triangles). The corresponding distribution of background O<sub>3</sub> concentrations is shown as grey diamonds.

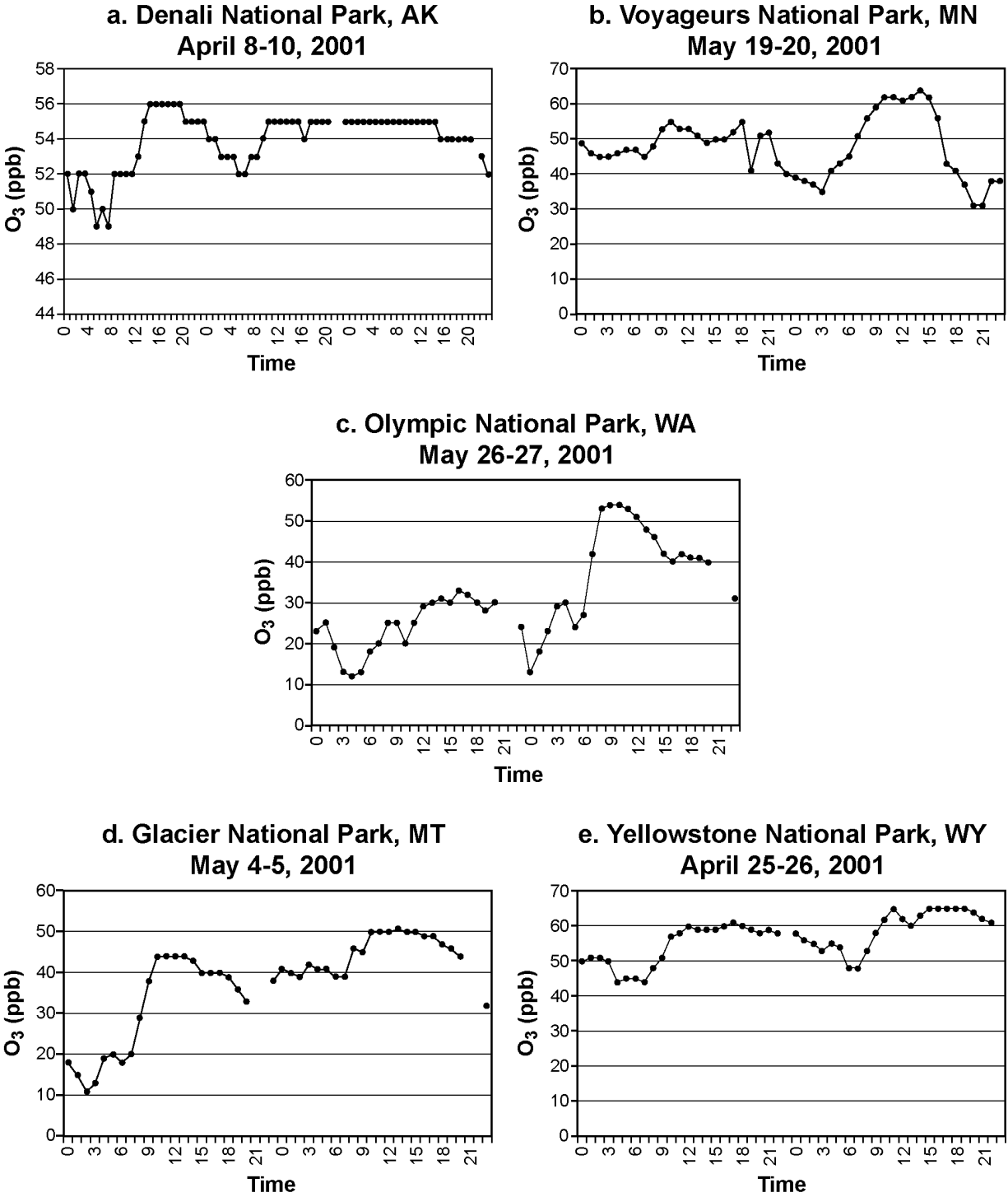
Source: Fiore et al. (2003).

1 Major conclusions from the Fiore et al. (2003) study (discussed in detail in Annex AX3,  
2 Sections AX3.9.3 and AX3.9.4) are:

- 3 • PRB O<sub>3</sub> concentrations in U.S. surface air from 1300 to 1700 local time are generally  
15 to 35 ppbv. They decline from spring to summer and are generally <25 ppbv under  
the conditions conducive to high-O<sub>3</sub> episodes.
- 4 • PRB O<sub>3</sub> concentrations can be represented as a function of season, altitude, and total  
surface O<sub>3</sub> concentration, as illustrated in Figure 3-26.
- 5 • High PRB concentrations (40 to 50 ppbv) occur occasionally at high-elevation sites  
(>1.5 km) in spring due to the free-tropospheric influence, including a 4- to 12-ppbv  
contribution from hemispheric pollution (O<sub>3</sub> produced from anthropogenic emissions  
outside North America). These sites cannot be viewed as representative of low-elevation  
surface sites (Cooper and Moody, 2000), where the background is lower when O<sub>3</sub>  
>60 ppbv.
- 6 • The stratospheric contribution to surface O<sub>3</sub> is of minor importance, typically well  
<20 ppbv. While stratospheric intrusions might occasionally elevate surface O<sub>3</sub>  
at high-altitude sites, these events are rare.

7 Appropriate background concentrations should thus be allowed to vary as a function of  
8 season, altitude, and total O<sub>3</sub> level. The diamonds in Figure 3-26 can be applied for this purpose.  
9 In particular, the depletion of the background during high-O<sub>3</sub> events should be taken into account  
10 (i.e., background O<sub>3</sub> is depleted by reactions in the atmosphere and by deposition to the surface  
11 but is not replenished at a significant rate in the stable, polluted boundary layer). This depletion  
12 is shown in the right-hand panels of Figure 3-26 for the highest O<sub>3</sub> values. Note that the model  
13 is generally able to reproduce the overall frequency distributions in Figure 3-26. Typically,  
14 models produce distributions flatter than are observed. Underpredictions, especially at the upper  
15 end of the frequency distribution during the warmer months, are likely related to sub-grid-scale  
16 processes that the model cannot resolve explicitly. The highest observed O<sub>3</sub> concentrations in all  
17 three seasons and at all altitudes are associated with regional pollution (i.e., North American  
18 anthropogenic emissions), rather than stratospheric influence.

19 Chemistry transport models should be evaluated with observations given earlier in  
20 Chapter 3, in Annex AX3, and to simulate the processes causing the intra-day variability in O<sub>3</sub>  
21 concentrations shown in Figure 3-27 in addition to those summarized in Chapter 2. The diurnal  
22 patterns shown in Figure 3-27 do not fit the smooth pattern shown in Figure 3-15 and indicate



**Figure 3-27. Time-series of hourly average O<sub>3</sub> concentrations observed at five national parks: Denali (AK), Voyageur (MN), Olympic (WA), Glacier (MT), and Yellowstone (WY).**

1 processes capable of producing rapid rises in O<sub>3</sub> at times when substantial photochemical  
2 activity is not present and may indicate stratospheric effects. Higher resolution models capable  
3 of spatially and temporally resolving stratospheric intrusions and capable of resolving O<sub>3</sub>  
4 variability on hourly timescales have not been applied to this problem. Ebel et al. (1991) have  
5 demonstrated that regional-scale CTMs could be used to study individual stratospheric  
6 intrusions. As an example of the utility of different types of models, Zanis et al. (2003) were  
7 able to forecast, observe, and model a stratospheric intrusion (maximum penetration depth was  
8 to slightly >2 km altitude) that occurred from June 20 to 21, 2001, over a large swath of central  
9 Europe. Roelofs et al. (2003) compared results from six global tropospheric CTMs with lidar  
10 observations obtained during that event and concluded that the models qualitatively captured the  
11 features of this intrusion. It was also found that the coarser resolution models overestimated  
12 transport to lower altitudes. The use of higher resolution models, perhaps nested inside the  
13 coarser resolution models, may have helped solve this problem. They would also better address  
14 issues related to temporal (i.e., 1-h versus 8-h averages) and spatial (i.e., populated versus  
15 remote areas) scales needed by policymakers.

16 Although many of the features of the day-to-day variability of O<sub>3</sub> at RRMS in the United  
17 States are simulated reasonably well by Fiore et al. (2003), uncertainties in the calculation of the  
18 temporal variability of O<sub>3</sub> originating from different sources on shorter time scales must be  
19 recognized. The uncertainties stem in part from an underestimate in the seasonal variability in  
20 the STE of O<sub>3</sub> (Fusco and Logan, 2003), the geographical variability of this exchange, and the  
21 variability in the exchange between the free troposphere and the planetary boundary layer in the  
22 model.

23 Ideally, the predictions resulting from an ensemble of models should be compared with  
24 each other and with observations, so that the range of uncertainty inherent in the model  
25 predictions can be evaluated.

### 28 **3.8 OZONE EXPOSURE IN VARIOUS MICROENVIRONMENTS**

29 Humans are exposed to O<sub>3</sub> and related photochemical oxidants through the exchange  
30 boundary, the skin and the openings into the body such as the mouth, the nostrils, and punctures  
31 and lesions in the skin (U.S. Environmental Protection Agency, 1992; Federal Register, 1986).

1 Inhalation exposure to O<sub>3</sub> and related photochemical oxidants is determined by pollutant  
2 concentrations measured in the breathing zone that is not affected by exhaled air as the  
3 individual moves through time and space. A discussion of the basic terminology associated with  
4 exposure appears in AX3.

### 6 ***Quantification of Exposure***

7 Ambient O<sub>3</sub> concentrations vary with time of day (peaking during the latter portion of the  
8 day) and season and among locations. Consequently, exposure to O<sub>3</sub> will change as a function of  
9 time of day and as an individual moves among locations. A hypothetical exposure is  
10 demonstrated in Figure 3-28. The actual dose received also changes during the day and is  
11 dependent on the O<sub>3</sub> concentration in the breathing zone and the individual's breathing rate,  
12 which is, in turn, dependent on the individual's level of exertion.

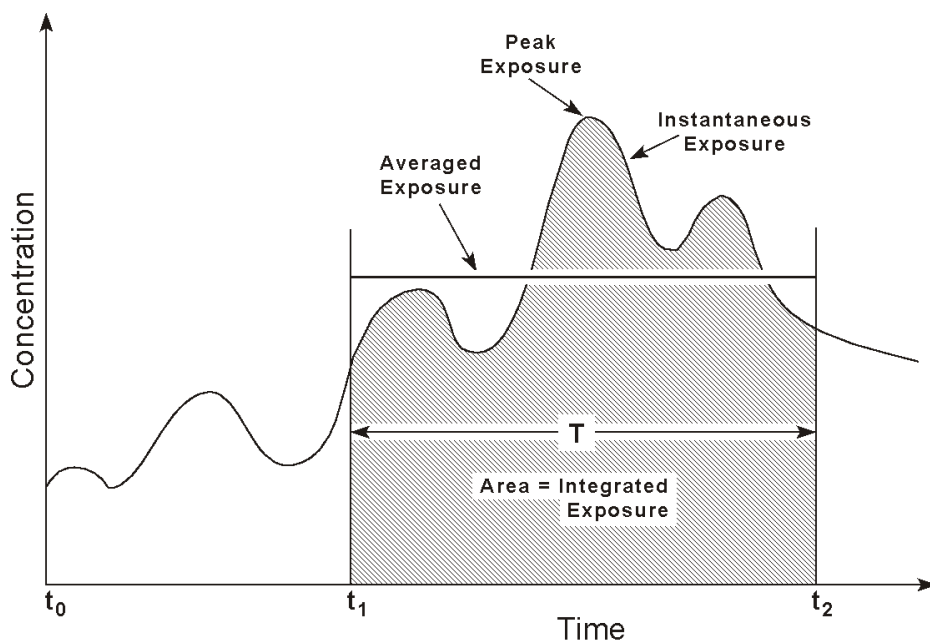
13 When measuring or modeling exposure to O<sub>3</sub> and related photochemical oxidants  
14 consideration should be given to the diurnal weekly (weekday-weekend) and seasonal  
15 variability. Peak concentrations lasting for several hours typically occur toward the latter  
16 portion of the day during the summer months. Regional O<sub>3</sub> episodes often co-occur with high  
17 concentrations of airborne fine particles, making it difficult to assess O<sub>3</sub> dynamics and exposure  
18 patterns. Also, while there are few indoor O<sub>3</sub> sources, O<sub>3</sub> will react with materials and other  
19 pollutants in the indoor environment in an analogous fashion to that occurring in the ambient  
20 atmosphere, potentially exposing subjects to other more toxic pollutants (Nazaroff and Weschler,  
21 2004; Lee and Hogsett, 1999; Wainman et al., 2000; Weschler and Shields, 1997). (See  
22 discussion on O<sub>3</sub> chemistry and indoor sources and concentrations later in this chapter.).

### 24 ***Personal Exposure and Ambient Concentrations***

25 The two approaches for measuring personal exposure are (a) the direct approach, using a  
26 personal exposure monitor (PEM) consisting of a passive sampler worn around the breathing  
27 zone, and (b) the indirect approach, which measures or estimates the O<sub>3</sub> concentrations through  
28 the use of models or biomarkers. Both approaches are associated with measurement error.

29 Although it is difficult to develop passive monitors for personal exposure measurements  
30 because of problems in identifying chemical or trapping reagents that can react with O<sub>3</sub>, several  
31 modified passive samplers have been developed for use in personal O<sub>3</sub> exposure measurements





**Figure 3-28. Hypothetical exposure time profile: pollutant exposure as a function of time showing how the average exposure, integrated exposure, and peak exposure relate to the instantaneous exposure. ( $t_2 - t_1 = T$ )**

Source: U.S. Environmental Protection Agency (2004a).

1 (Bernard et al., 1999; Koutrakis et al., 1993; Avol et al., 1998b; Geyh et al., 1997, 1999). Some  
 2 personal exposure measurements using passive samplers show  $O_3$  exposures below those  $O_3$   
 3 concentrations measured at outdoor stationary sites (Delfino et al., 1996; Avol et al., 1998b;  
 4 Sarnat et al., 2000; Geyh et al., 2000; Brauer and Brook, 1997). However, other studies have  
 5 found strong correlations between  $O_3$  measured at stationary sites and personal monitored  
 6 concentrations (Liard et al., 1999; Bauer and Brook, 1997; Linn et al., 1996; Lee et al., 2004;  
 7 Avol et al., 1998b; O'Neill et al., 2003) when the time spent outdoors, age, gender, and  
 8 occupation of the subjects were considered.

9 The indirect approach determines and measures the concentrations in all of the locations or  
 10 “microenvironments”. The concept of microenvironments is important in the understanding of  
 11 human exposure modeling. Often identified with a perfectly mixed compartment,  
 12 microenvironments are more recently viewed as a controlled volume, indoors or outdoors, that  
 13 can be characterized using a set of either mechanistic or phenomenological governing equations.  
 14 This allows for a nonhomogeneous environment, including sources and sinks within the

1 microenvironment. Microenvironments include indoor residences, other indoor locations,  
2 outdoors near roadways, other outdoor locations, and areas within vehicles.

### 4 ***Microenvironmental Concentration and Ozone Exposure Models***

5 Outdoor concentrations of O<sub>3</sub> are estimated either through emissions-based mechanistic  
6 modeling, or through ambient-data-based modeling. Emissions-based models determine the  
7 spatiotemporal fields of the O<sub>3</sub> concentrations using precursor emissions and meteorological  
8 conditions as inputs. (They are described in Annex AX2.). The ambient-data-based models  
9 determine spatial or spatiotemporal distributions of O<sub>3</sub> through the use of interpolation schemes.  
10 The kriging approach provides standard procedures for generating an interpolated O<sub>3</sub> spatial  
11 distribution for a given period of time (Georgopoulos et al., 1997a,b). The Spatio-Temporal  
12 Random Field (STRF) approach has been used to interpolate monitoring data in both space and  
13 time (Christakos and Vyas, 1998a,b). The STRF approach can analyze information on temporal  
14 trends which cannot be directly incorporated by kriging.

15 Several approaches are available for modeling microenvironmental concentrations:  
16 empirical, mass balance, and detailed computational fluid dynamics (CFD) models. Empirical  
17 relationships provide the basis for future, “prognostic” population exposure models. Mass  
18 balance modeling is the most common approach used to model pollutant concentrations in  
19 enclosed microenvironments. Mass balance modeling ranges from very simple formulations,  
20 assuming ideal (homogeneous) mixing and only linear physicochemical transformations with  
21 sources and sinks, to models that account for complex multiphase chemical and physical  
22 interactions and nonidealities in mixing. Mass balance models take into account the effects of  
23 ventilation, filtration, heterogeneous removal, and direct emission as well as photolytic, thermal,  
24 and chemical reactions. The simplest form of the model is represented by the following  
25 differential equation:

$$\frac{dC_{IN}}{dt} = vC_{OUT} + \frac{S}{V} - vC_{IN}$$

27  
28 where  $dC_{IN}$  is the indoor pollutant concentration (mass/volume),  $dt$  is time in hours,  $v$  is the air  
29 exchange rate,  $C_{OUT}$  is the outdoor pollutant concentration (mass/volume),  $V$  is the volume of the  
30 microenvironment, and  $S$  is the indoor source emission rate. When the model was used to

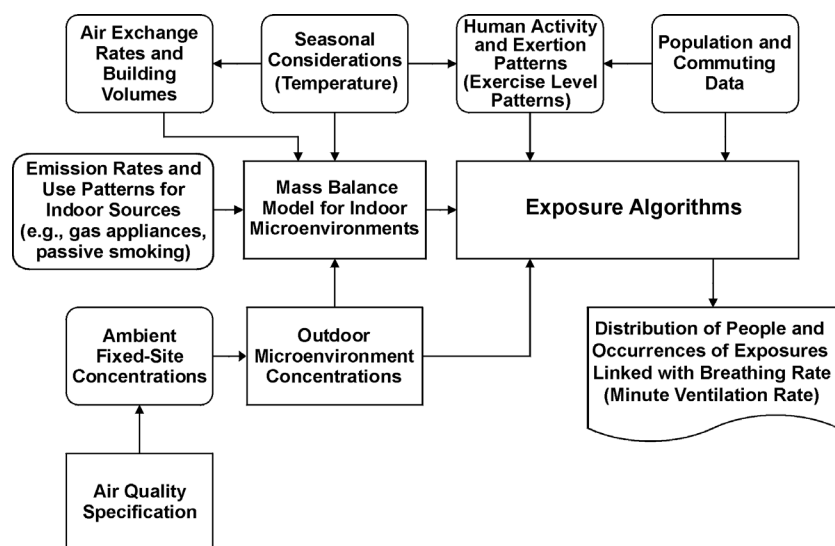
1 estimate indoor O<sub>3</sub> concentrations, indoor concentrations were found to be 33% of outdoor O<sub>3</sub>  
2 concentrations (Freijer and Bloemen, 2000). A more in-depth discussion of the mass balance  
3 model has been reported in Nazaroff and Cass (1986). The pNEM/O<sub>3</sub> model, discussed later in  
4 this chapter, includes a sophisticated mass balance model for indoor and vehicle  
5 microenvironments (Johnson, 2003). CFD models take into account the complex, multiphase  
6 processes that affect indoor concentrations of interacting gas phase pollutants, such as the  
7 interactions of O<sub>3</sub> with indoor sinks and sources (surfaces, gas releases) and with entrained gas  
8 (Sarwar et al., 2001, 2002; Sørensen and Weschler, 2002).

9 Exposure modeling is often used in evaluating exposure to large populations over time.  
10 The use of models is complicated by the fact that O<sub>3</sub> is a secondary pollutant with complex  
11 nonlinear and multiscale dynamics in space and time. Ozone is formed in the atmosphere  
12 through a series of chemical reactions involving precursor VOCs and NO<sub>x</sub>. Therefore, O<sub>3</sub>  
13 exposures may be affected by: (1) emission levels and spatiotemporal patterns of VOCs and  
14 NO<sub>x</sub>; (2) ambient atmospheric as well as indoor microenvironmental transport, removal and  
15 mixing processes; and (3) chemical transformations that take place over a multitude of spatial  
16 scales. The transformations are dependent on the presence of co-occurring pollutants and the  
17 nature of surfaces interacting with the pollutants.

18 Exposure models may be classified as (1) potential exposure models, typically the  
19 maximum outdoor concentrations versus “actual” exposure, including locally modified  
20 microenvironmental outdoor and indoor exposures; (2) population versus “specific individual”-  
21 based exposure models; (3) deterministic versus probabilistic models; and (4) observation versus  
22 mechanistic air quality model-driven estimates of spatially and temporally varying O<sub>3</sub>  
23 concentrations.

24 There are several steps involved in defining exposure models. The steps are based on  
25 frameworks described in the literature over the last 20 years and the structure of various existing  
26 inhalation exposure models (NEM/pNEM, MENTOR/SHEDS, REHEX, TRIM.Expo also known  
27 as APEX, AIRPEX, AIRQUIS). The steps include (1) estimation/ determination of the  
28 background or ambient levels of O<sub>3</sub>; (2) estimation/determination of levels and temporal profiles  
29 of O<sub>3</sub> in various microenvironments; (3) characterization of relevant attributes of individuals or  
30 populations under study (age, gender, weight, occupation, other physiological characteristics);  
31 (4) development of activity event or exposure event sequences; (5) determination of appropriate  
32 inhalation rates during the exposure events; (6) determination of dose; (7) determination of

1 event-specific exposure and intake dose distributions for selected time periods; and  
 2 (8) extrapolation of population sample (or cohort) exposures and doses to the entire populations  
 3 of interest. Figure 3-29 provides a conceptual overview of a current exposure model. A more  
 4 detailed overview of an exposure model can be found in Annex AX3.



**Figure 3-29. Conceptual overview of an exposure model. Model inputs (e.g., activity patterns, ambient monitoring data, air exchange rates) are in round-corner boxes and model calculations are shown in rectangles.**

Source: Johnson et al. (1999).

1 To estimate the actual O<sub>3</sub> dose delivered to the lung, information on the concentration,  
 2 minute ventilation rate, activity level, and the morphology of the respiratory tract are needed.  
 3 Limited data have been compiled for ventilation rates for different age groups, both healthy and  
 4 compromised individuals, at various levels of activity (Klepeis et al., 1996, 2001; Avol et al.,  
 5 1998b; Adams, 1993). Based on the available information, the highest level of outdoor activity  
 6 occurs during the spring and summer months, during the mid- to late afternoon and early  
 7 evening—the times when O<sub>3</sub> concentrations are highest. Children are likely more susceptible to  
 8 the effects of O<sub>3</sub> than other groups. School-age children spend more time outdoors engaged in  
 9 high-level activities than do other groups and breathe more air in than adults relative to body

1 surface area, breathing frequency, and heart rate. Asthmatic children spend the same amount of  
2 time outdoors as other more healthy children but the time spent engaged in high levels of activity  
3 are less.

4 Estimates of activity level have been compiled based on questionnaire data. The National  
5 Human Activity Pattern Survey (NHAPS), a probability-based telephone survey, was conducted  
6 in the early 1990s. The survey concluded that outdoor work-related activities were highest  
7 during the springtime and were more frequent during the morning and early afternoon.  
8 Exercise/sports-related activities were highest from noon to 3 p.m. during the summer months.  
9 During the spring months, exercise/sports-related activities were highest from mid- to late  
10 afternoon (Klepeis et al., 1996, 2001). A pilot study by Gonzales et al. (2003) evaluated the use  
11 of retrospective questionnaires for reconstructing past time-activity and location pattern  
12 information. Ozone concentration estimates using ambient stationary monitors and estimates  
13 derived from diaries and questionnaires differed slightly. However, both estimates were greater  
14 than O<sub>3</sub> personal exposure measurements.

15 Existing comprehensive inhalation exposure models (NEM and pNEM) (Johnson, 2003),  
16 (MENTOR/SHEDS) Burke et al., 2001; McCurdy et al., 2000), and the Air Pollutants Exposure  
17 model (TRIM.Expo) treat human activity patterns as sequences of exposure events in which each  
18 event is defined by a geographic location and microenvironment and then assigned activity diary  
19 records from the CHAD (Consolidated Human Activities Database; [www.epa.gov/chadnet1](http://www.epa.gov/chadnet1))  
20 (Glen et al., 1997; McCurdy, 2000; McCurdy et al., 2000). There are now about 22,600 person-  
21 days of sequential daily activity pattern data in CHAD representing all ages and both genders.  
22 The data for each subject consist of one or more days of sequential activities, in which each  
23 activity is defined by start time, duration, activity type (140 categories), and microenvironment  
24 classification (110 categories). Activities vary from 1 min to 1 h in duration. Activities longer  
25 than 1 h are subdivided into clock-hour durations to facilitate exposure modeling. A distribution  
26 of values for the ratio of oxygen uptake rate to body mass (referred to as metabolic equivalents  
27 or METs) is provided for each activity type listed. A table listing the activity patterns included  
28 in CHAD appears in AX3.

29 pNEM divides the population of interest into representative cohorts based on the  
30 combinations of demographic characteristics (age, gender, and employment), home/work  
31 district, and residential cooking fuel. TRIM.Expo and MENTOR/SHEDS generate a population  
32 demographic file containing a user-defined number of person-records for each census tract of the

1 population based on proportions of characteristic variables (age, gender, employment, and  
2 housing) obtained for the population of interest, and then assigns the matching activity  
3 information from CHAD to each individual record of the population based on the characteristic  
4 variables.

5 The TRIM.Expo model is capable of simulating individual movement through time and  
6 space to provide estimates of exposure to a given pollutant in various microenvironments (e.g.,  
7 indoor, outdoor, and in-vehicle microenvironments). One of the key strengths of the  
8 TRIM.Expo model is its ability to estimate hourly exposures and doses for all simulated  
9 individuals in a sampled population. However, TRIM.Expo is limited in that uncertainties in the  
10 predicted distributions (e.g., age, activity data, commuting patterns, personal activities) have not  
11 been addressed.

12 MENTOR/SHEDS is capable of simulating individuals exposures in various  
13 microenvironments (outdoors, residence, office, school, store, restaurant, bar, and vehicles)  
14 using spatial concentration data for each census tract. The indoor and in-vehicle pollutant  
15 concentrations are calculated using specific equations for the microenvironment and ambient  
16 pollutant concentration relationship. Randomly selected characteristics for a fixed number of  
17 individual are selected to match demographics within the census tract for age, gender,  
18 employment status, and housing type. Smoking prevalence statistics by gender and age is  
19 randomly selected for each individual in the simulation. Diaries for activity patterns are matched  
20 for the simulated individual by demographic characteristics (Burke et al., 2001).

21 An important source of uncertainty in existing exposure modeling involves the creation of  
22 multiday, seasonal, or year-long exposure activity sequences based on 1- to 3-day activity data  
23 for any given individual from CHAD. Currently, appropriate longitudinal data are not available  
24 and the existing models use various rules to derive longer-term activity sequences utilizing 24-h  
25 activity data from CHAD.

26 Of the above models, only NEM/pNEM have been used extensively in O<sub>3</sub> exposure  
27 modeling. The pNEM probabilistic model builds on the earlier NEM deterministic exposure  
28 model. The model takes into consideration the temporal and spatial distribution of people and  
29 O<sub>3</sub> in the area of consideration, variations in O<sub>3</sub> concentrations in the microenvironment, and the  
30 effects of exercise-increased ventilation on O<sub>3</sub> uptake. There are three versions of the pNEM/O<sub>3</sub>  
31 model: (1) general population (Johnson et al., 1996a), (2) outdoor workers (Johnson et al.,  
32 1996b), and (3) outdoor children (Johnson et al., 1996c, 1997). The pNEM models have been

1 applied to nine urban areas and a summer camp. The models used activity data from the  
2 Cincinnati Activity Diary Study (CADS) along with time-activity data from several other  
3 studies. Data from stationary monitoring sites were used to estimate outdoor O<sub>3</sub> exposure.  
4 Indoor O<sub>3</sub> decay was assumed to be proportional to the indoor O<sub>3</sub> concentration. An algorithm  
5 assigned the EVR associated with each exposure event. The EVR for the outdoor children  
6 model was generated using a module based on heart rate data by Spier et al. (1992) and Linn  
7 et al. (1992).

### 9 *Characterization of Exposure*

10 The use of ambient air monitoring stations is the most common surrogate for assigning  
11 exposure in epidemiological studies. Since the primary source of O<sub>3</sub> exposure is the ambient air,  
12 monitoring concentration data would provide the exposure outdoors while exercising, a potential  
13 important exposure to evaluate in epidemiological studies. Monitored concentrations are useful  
14 for a relative assignment of exposure with time if the concentration were uniform across the  
15 region; the time-activity pattern were the same across the population; and the housing  
16 characteristics, such as ventilation rates and the O<sub>3</sub> sinks contributing to its indoor decay rates,  
17 were constant for the study area. Since these factors vary by population and location there will  
18 be errors in the magnitude of the total exposure and in the relative total exposure assignment  
19 based solely on ambient monitoring data.

20 Personal O<sub>3</sub> exposure measurements have been made for potentially susceptible  
21 populations (children, outdoor workers, the elderly, and individuals with chronic obstructive  
22 pulmonary disease). Children and outdoor workers have somewhat higher exposures than other  
23 individuals because they spend more time outdoors engaged in moderate and heavy exertion.  
24 Children are also more active outside and, therefore, have a higher minute ventilation rate than  
25 most adults (Klepeis et al., 1996, 2001). Available exposure studies suggest trends in exposure  
26 magnitude for some populations, however, additional exposure studies are needed to generalize  
27 differences in exposure between the general population and potentially susceptible populations.  
28 Table 3-3 summarizes the findings of available exposure studies.

**Table 3-3. Personal Exposure Concentrations**

Location, Population, Sample Duration	n	Personal Exposure Mean <sup>a</sup> (range) (ppb)	Reference
San Diego, CA, Asthmatics ages 9-18 years, 12 hour	12	12 ± 12 (0-84) 10 weekend 12 weekday	Delfino et al. (1996)
Vancouver, Canada, Adult Workers, Daily High indoor time Moderate indoor time Only outdoor	585	(ND-9) (ND-12) (2-44)	Brauer and Brook (1997)
Southern California, Subjects 10-38 years Spring Fall	24	13.6 ± 2.5 (- to 80) 10.5 ± 2.5 (- to 50)	Liu et al. (1997)
Montpellier, France, Adults, Hourly Winter Summer	16	34.3 ± 17.6 (6.5-88) 15.4 ± 7.7 (6.5-40) 44.1 ± 18.2(11-88)	Bernard et al. (1999)
Souther California, Children 6-12 years, ≥ 6 days Upland - winter - summer Mountain - winter - summer	169	6.2 ± 4.7 (0.5-41) 19 ± 18 (0.5-63) 5.7 ± 4.2 (0.5-31) 25 ± 24 (0.5-72)	Geyh et al. (2000)
Baltimore, MD, Technician, Hourly <sup>b</sup> Winter Summer	1	3.5 ± 7.5 (ND-49) 15 ± 18 (ND-76)	Chang et al. (2000)
Baltimore, MD, Adults 75 ± 7 years, Daily Winter Summer	20	3.5 ± 3.0 (ND-9.9) 0. ± 1.8 (ND-2.8)	Sarnat et al. (2000)

<sup>a</sup>ND = not detected.

<sup>b</sup>Measurements made following scripted activities for 15 days.

1 Ozone concentrations in various microenvironments under a variety of environmental  
2 conditions have been reported in the literature. In the absence of an indoor O<sub>3</sub> source,  
3 concentrations of O<sub>3</sub> indoors are lower than that found in the ambient air. Ozone concentrations  
4 in microenvironments were found to be primarily controlled by ambient O<sub>3</sub> concentrations and  
5 the AER: they increase with increasing AER. To a lesser extent, O<sub>3</sub> concentrations in  
6 microenvironments are influenced by the ambient temperature, time of day, indoor  
7 characteristics (e.g., presence of carpeting), and the presence of other pollutants in the  
8 microenvironment. Table 3-4 describes the findings of the available studies.



## *Factors Affecting Ozone Concentrations*

Ozone and other photochemical oxidants are formed in the ambient air from the reaction of sunlight with vehicle emissions, gasoline fumes, solvent vapors, and power plant and industrial emissions (See Chapter 2 for a discussion of O<sub>3</sub> atmospheric chemistry). Ozone enters the indoor environment primarily through infiltration from outdoors through building components, such as windows, doors, and ventilation systems. There are also a few indoor sources of O<sub>3</sub> (photocopiers, facsimile machines, laser printers, and electrostatic air cleaners and precipitators) (Weschler, 2000). Generally O<sub>3</sub> emissions from office equipment and air cleaners are low except under improper maintenance conditions. Reported O<sub>3</sub> emissions from office equipment range from 1300 to 7900 µg/h (Leovic et al., 1996, 1998). Most air cleaners (particulate ionizers) emitted no or only a small amount (56 to 2757 µg/h) of O<sub>3</sub> during operation (Niu et al., 2001). Emissions from O<sub>3</sub> generators can range from tens to thousands of micrograms per hour (Weschler, 2000; U.S. Environmental Protection Agency, 1996).

Other photochemical oxidants (peroxyacyl nitrates; PAN and PPN) have no known direct emission sources indoors. PAN may be formed in the indoor environment from the reaction of the OH· or NO<sub>3</sub> with acetaldehyde to form the acetyl radical, CH<sub>3</sub>CO (Grosjean et al., 1996). The acetyl radical then reacts with oxygen to form an acetylperoxy radical which reacts with NO<sub>2</sub> to form PAN. Peroxyacyl nitrates primarily occur in the indoor environment from infiltration through the building envelope and through openings in the building envelope.

The concentration of O<sub>3</sub> in indoor environments is dependent on the outdoor O<sub>3</sub> concentration, the AER or outdoor infiltration, indoor circulation rate, and O<sub>3</sub> removal processes through contact with indoor surfaces and reactions with other indoor pollutants. Since O<sub>3</sub> concentrations are generally higher during the warmer months, indoor concentrations will likely be highest during that time period. (See earlier discussion on ambient concentrations of O<sub>3</sub>.)

Air exchange rates vary depending on temperature differences, wind effects, geographical region, type of heating/mechanical ventilation system, and building type (Weschler and Shields, 2000; Colome et al., 1994). The balance of the flow of air in and out of a microenvironment is greatest in a residential building when a window or door is open (Johnson et al., 2004; Howard-Reed et al., 2002). The opening of windows or doors is dependent on the building occupancy, season, housing density, the presence of air conditioning, and wind speed (Johnson and Long, 2004). When windows and doors are closed, the dominant mechanism controlling AERs is

**Table 3-4. Indoor/Outdoor Ozone Concentrations in Various Microenvironments**

Location and Ventilation Conditions	Indoor/Outdoor Concentrations	Comments	Reference
New England States (9) Fall	20 ppb/40 ppb	Schools represented a variety of environmental conditions - varying ambient O <sub>3</sub> concentrations, sources, geographic locations, population density, traffic patterns, building types. Average O <sub>3</sub> concentrations were low in the morning and peaked during the early afternoon. O <sub>3</sub> concentrations averaged for all schools monitored.	NESCAUM (2002)
Mexico City, School Windows/Doors Open (27) Windows/Doors Closed Cleaner Off (41) Windows/Doors Closed Cleaner On (47)	0 to 247 ppb/ 64 to 361 ppb	Study conducted over 4 d period during winter months. Two-minute averaged measurements were taken both inside and outside of the school every 30 min from 10 a.m. to 4 p.m. Estimated air exchange rates were 1.1, 2.1, and 2.5 h <sup>-1</sup> for low, medium, and high flow rates. Ozone concentrations decreased with increasing relative humidity.	Gold et al. (1996)
Mexico City Homes	5 ppb/27ppb (7 d) 7 ppb/37 ppb (14 d)	Ozone monitoring occurred between September and July. Study included 3 schools and 145 homes. Most of the homes were large and did not have air conditioning. Ninety-two percent of the homes had carpeting, 13% used air filters, and 84% used humidifiers. Thirty-five percent opened windows frequently, 43% sometimes, and 22% never between 10 a.m. and 4 p.m. Ozone was monitored at the schools sites from 8 a.m. to 1 p.m. daily for 14 consecutive days. Homes were monitored for continuous 24-h periods for 7 and 14 consecutive days.	Romieu et al. (1998)
Schools	22 ppb/56 to 733 ppb		
Boston, MA, Homes (9) Winter - continuously Summer - continuously	0 to 20.4 ppb/4.4 to 24.5 ppb 0 to 34.2 ppb/8.2 to 51.8 ppb	Study examined the potential for O <sub>3</sub> to react with VOCs to form acid aerosols. Carbonyls were formed. No clear trend of O <sub>3</sub> with AERs. The average AER was 0.9 h <sup>-1</sup> during the winter and 2.6 h <sup>-1</sup> during the summer. Four residences in winter and nine in summer with over 24 h samples collected.	Reiss et al. (1995)
Los Angeles, Homes (239)	13 ppb/37 ppb	Four hundred and eighty-one samples collected inside and immediately outside of home from February to December. Concentrations based on 24-h average O <sub>3</sub> concentrations indoors and outdoors. Low outdoor concentrations resulted in low indoor concentrations. However, high outdoor concentrations resulted in a range of indoor concentrations.	Avol et al. (1998a)

**Table 3-4 (cont'd). Indoor/Outdoor Ozone Concentrations in Various Microenvironments**

Location and Ventilation Conditions	Indoor/Outdoor Concentrations	Comments	Reference
Burbank, CA Telephone Switching Station	0.2/21.1 ppb	Major source of O <sub>3</sub> was transport from outdoors. From early spring to late fall O <sub>3</sub> concentrations peaked during the early afternoon and approach zero at sunset. AER ranged from 1.0 to 1.9 h <sup>-1</sup> .	Weschler et al. (1994)
Munich Germany Office	0.4/0.9 ppb	Indoor concentrations were dependent on the type of ventilation.	Jakobi and Fabian (1997)
Gymnasium	0.49/0.92 ppb		
Classroom	0.54/0.77 ppb		
Residence Bedroom	0.47/1.0 ppb		
Livingroom	0.74/1.0 ppb		
Montpellier, France, Homes (110)	15.5/32.0 ppb	Ozone measurements were made over 5-d periods in and outside of 21 homes during the summer and winter months. The winter I/O ratio was 0.31 compared to 0.46 during the summer months.	Bernard et al. (1999)
Southern CA, Homes Upland	11.8/48.2 ppb	Ozone measurements were taken at 119 homes (57 in Upland and 62 in towns located in the mountains) during April and May. Concentrations were based on average monthly outdoor concentrations and average weekly indoor concentrations. Indoor based on the home location, number of bedrooms, and the presence of an air conditioner.	Geyh et al. (2000) Lee et al. (2002)
Mountains	2.8/35.7 ppb		
Krakow, Poland, Museums Cloth Hall	3.2/25.7-27.4 ppb	Ozone continuously monitored at five museums and cultural centers. Monitoring conducted during the summer months for 21 to 46 h or 28 to 33 days at each of the sites. The indoor concentration was found to be dependent on the ventilation rate, i.e., when the ventilation rate was high the indoor O <sub>3</sub> concentrations approached that of ambient O <sub>3</sub> . Rooms sequestered from the outdoor air or where air was predominantly recycled through charcoal filters the O <sub>3</sub> levels indoors were greatly reduced.	Salmon et al. (2000)
Matejko	8.5/20.0 ppb		
Wawel Castle	2.5/14.7 ppb		
National	1.5/11.0 ppb		
Buildings, Greece Thessalonki	9.39/15.48 ppb	There was no heating/air conditioning system in the building at Thessaloniki. Windows were kept closed during the entire monitoring period. Complete air exchange took place every 3 h. The air conditioning system in continuous use at the Athens site recirculated the air. Complete air exchange was estimated to be 1 h. Monitoring lasted for 30 days at each site but only the 7 most representative days were used.	Drakou et al. (1995)
Athens	8.14/21.66 ppb		

**Table 3-4 (cont'd). Indoor/Outdoor Ozone Concentrations in Various Microenvironments**

<b>Location and Ventilation Conditions</b>	<b>Indoor/Outdoor Concentrations</b>	<b>Comments</b>	<b>Reference</b>
Patrol cars, NC	11.7/28.3 ppb	Patrol cars were monitored Mon. through Thurs. between the hours of 3 p.m. to midnight on 25 occasions during the months of Aug., Sept., and Oct. Outdoor O <sub>3</sub> concentrations were taken from ambient monitoring station. Air inside the patrol car was recirculated cool air.	Riediker et al. (2003)
University of CA Photocopy room	<20 to 40 ppb/—	Room volume was 40 m <sup>3</sup> . Ozone concentrations increased proportionately with increasing use of photocopier.	Black et al. (2000)
Home/office O <sub>3</sub> generators	14 to 200 ppb/—	Room volume was 27 m <sup>3</sup> . Doors and windows were closed. Heating/air conditioning and mechanical ventilation systems were off. Ozone generator was operated for 90 min. High O <sub>3</sub> concentrations noted when O <sub>3</sub> generator used at high setting. AER was 0.3 h <sup>-1</sup> .	Steiber et al. (1995)

1 infiltration through unintentional openings in the building envelope. Williams et al. (2003a,  
2 2003b) reported AERs of 0.001 to 4.87 h<sup>-1</sup> in 37 homes in Research Triangle Park, NC. Chan  
3 et al. (2005) compared air leakage measurements for 70,000 houses. Older and smaller houses  
4 had higher normalized leakage areas than newer and larger houses. Meng et al. (2004) also  
5 attributed higher AERs to the age of the housing stock. AERs for homes in Houston, TX and  
6 Elizabeth, NJ were averaged for all four seasons, the highest AER, 1.22 h<sup>-1</sup>, was noted for homes  
7 in Elizabeth, NJ where the homes were older. Evaluations of AERs for residential structures was  
8 reported by Murray and Burmaster (1995) and includes AERs for 2,844 residential structures in  
9 four different climatic regions by season (winter, spring, summer, and fall). The AER for all  
10 seasons across all regions was 0.76 h<sup>-1</sup> (arithmetic mean) (Region 1: IN, MN, MT, NH, NY1,  
11 VT, WI; Region 2: CO, CT, IL, NJ, NY2, OH, PA, WA; Region 3: CA3, MD, OR, WA;  
12 Region 4: AZ, CA4, FL, TX). The AERs were generally higher during the warm seasons, when  
13 ambient O<sub>3</sub> concentrations are highest. Data for the warmest region during the summer months  
14 may not be representative of all homes because measurements were made in southern California  
15 where windows are open and air conditioning is not used.

16 Average mean (median) AERs of 2.45 (2.24), 1.35 (1.09), and 2.22 (1.79) h<sup>-1</sup> were  
17 reported by Lagus Applied Technology, Inc. (1995) for schools, offices, and retail  
18 establishments in California. Mean AERs for schools, offices, and retail establishments in  
19 Oregon and Washington were 0.32, 0.31, and 1.12 h<sup>-1</sup> (Turk et al., 1989)—considerably less than  
20 that reported by Lagus Applied Technology. Park et al. (1998) reported mean AERs ranging  
21 from 1.0 to 47.5 h<sup>-1</sup> for stationary vehicles under varying ventilating conditions. Where  
22 available, AERs for other studies are included in Table 3-10.

23 The most important removal process for O<sub>3</sub> in the indoor environment is deposition on and  
24 reaction with indoor surfaces. The rate of deposition is material-specific. The removal rate will  
25 depend on the indoor dimensions, surface coverings, and furnishings. Smaller rooms generally  
26 have larger surface-to-volume ratio (A/V) and remove O<sub>3</sub> faster than larger rooms. Fleecy  
27 materials, such as carpets, have larger surface-to-volume ratios and remove O<sub>3</sub> faster than  
28 smooth surfaces (Weschler, 2000). However, the rate of O<sub>3</sub> reaction with carpet diminishes with  
29 cumulative O<sub>3</sub> exposure (Morrison and Nazaroff, 2000, 2002). Weschler (2000) compiled the O<sub>3</sub>  
30 removal rates for a variety of microenvironments. Generally, the removal rates ranged between  
31 3.0 and 4.3 k<sub>d</sub> (A/V)/h<sup>-1</sup>. The highest removal rate, 7.6 k<sub>d</sub> (A/V)/h<sup>-1</sup>, was noted for a clean room  
32 (Weschler et al, 1989).

1 Ozone chemical reactions in the indoor environment are analogous to those reactions  
2 occurring in the ambient air (See discussion on atmospheric chemistry in Chapter 2). Ozone  
3 reacts with unsaturated VOCs in the indoor environment, primarily terpenes or terpene-related  
4 compounds from cleaning products, air fresheners, and wood products. The reactions are  
5 dependent on the O<sub>3</sub> indoor concentration, the indoor temperature and, in most cases, the air  
6 exchange rate/ventilation rate. Some of the reaction products may more negatively impact  
7 human health and artifacts in the indoor environment than their precursors (Wolkoff et al., 1999;  
8 Wilkins et al., 2001; Weschler et al., 1992; Weschler and Shields, 1997; Rohr et al., 2002;  
9 Nøjgaard et al., 2005). Primary reaction products are Criegee biradicals, nitrate radicals, and  
10 peroxyacetyl radicals. Secondary reaction products are hydroxy, alkyl, alkylperoxy,  
11 hydroperoxy, and alkoxy radicals. Reactions with alkenes can produce aldehydes, ketones, and  
12 organic acids (Weschler and Shields, 2000; Weschler et al., 1992).

13 Hydroxyl radicals formed from the reaction of O<sub>3</sub> with VOCs, nitric oxide and  
14 hydroperoxy, and other intermediate products can react with various nitrogen compounds, sulfur  
15 dioxide, carbon monoxide and other compounds to produce significantly more toxic compounds  
16 (Sarwar et al., 2002; Orzechowska and Paulson, 2002; Fick et al., 2003, 2004; Van den Bergh  
17 et al., 2000; Fan et al., 2003; Wilkins et al., 2001; Clausen et al., 2001; Rohr et al., 2002, 2003;  
18 Poupard et al., 2005; Blondeau et al., 2005). The reaction between O<sub>3</sub> and terpenes also has been  
19 shown to increase the concentration of indoor particles (Weschler and Shields, 1999, 2003;  
20 Weschler, 2004; Clausen et al., 2001; Fan et al., 2003; Wainman et al., 2000), possibly from  
21 further reactions of the hydroxy radical with terpenes (Sarwar et al., 2002).

22 Decomposition and formation of PAN in the indoor environment are influenced by NO<sub>2</sub>  
23 and NO. Decomposition of PAN is expected to be a relatively fast process when indoor O<sub>3</sub>  
24 levels are low and when motor vehicle emissions are large or there is an indoor source of NO<sub>x</sub>  
25 (Weschler and Shields, 1997).

### 26 27 ***Factors Affecting the Relationship between Ambient Concentrations and*** 28 ***Personal Exposures to O<sub>3</sub>***

29 Ambient O<sub>3</sub> concentrations vary with the time of day, season of the year, and among  
30 locations. Personal exposure to O<sub>3</sub> is influenced by the microenvironmental concentration and  
31 the amount of time spent in each microenvironment. Because the majority of the population  
32 spends on average nearly 90% of their time in an indoor microenvironment, the majority of the

1 O<sub>3</sub> exposure will occur in the indoor environment. Since there are few indoor sources of O<sub>3</sub>, O<sub>3</sub>  
2 ambient concentration may be the most important factor that affects average population exposure  
3 in the indoor environment.

4 Indoor O<sub>3</sub> concentrations also are affected by several other factors and mechanisms.  
5 Studies have shown that in addition to the ambient O<sub>3</sub> concentrations, indoor O<sub>3</sub> concentrations  
6 are influenced by the air exchange rate or outdoor infiltration, increasing with increasing air  
7 exchange. Once indoors, the O<sub>3</sub> concentration is affected by the indoor circulation rate and O<sub>3</sub>  
8 removal through contact with indoor surfaces and reactions with other indoor pollutants.

9 In some instances, ambient O<sub>3</sub> monitors are located in areas outside the breathing zone.  
10 Studies on the effect of elevation on O<sub>3</sub> concentrations found that concentrations increased with  
11 increasing elevation (Väkevä et al., 1999; Johnson, 1997). Also, since O<sub>3</sub> monitors are  
12 frequently located on rooftops in urban settings, the concentrations measured there may  
13 overestimate the exposure to individuals outdoors in streets and parks, locations where people  
14 exercise and their maximum O<sub>3</sub> exposure is more likely to occur.

15 In epidemiologic studies investigating acute and chronic health outcomes using ambient  
16 monitoring data from stationary monitoring sites, O<sub>3</sub> exposure assessment was affected by the  
17 distance between home and the monitoring site, gender, time-activity patterns (e.g., percentage  
18 of time spent outdoors, type of outdoor activity, time of day during outdoor activity), and indoor  
19 air exchange rates (e.g., ventilation conditions, home characteristics) (Geyh et al., 2000; Lee  
20 et al., 2002, 2004; Liu et al., 1995, 1997; Chang et al., 2000; Chan et al., 2005; O'Neill et al.,  
21 2003; Brauer and Brook, 1997; O'Neill et al., 2003). People that work outdoors tend to be  
22 exposed to higher levels of O<sub>3</sub> (Brauer and Brook, 1997; O'Neill et al., 2003). Geyh et al. (2000)  
23 observed higher indoor and personal O<sub>3</sub> concentrations in a southern California community with  
24 2% air-conditioned homes compared to a community with 93% air-conditioned homes during  
25 the summer (high O<sub>3</sub>) months, but showed no difference in O<sub>3</sub> levels during the winter (low O<sub>3</sub>)  
26 months. Lee et al. (2004) observed that personal O<sub>3</sub> exposure was positively correlated with  
27 outdoor time ( $r = 0.19$ ,  $p < 0.01$ ) and negatively correlated with indoor time ( $r = -0.17$ ,  
28  $p < 0.01$ ). Additional factors that affected indoor O<sub>3</sub> levels were air conditioning, window fans,  
29 and window opening. The O<sub>3</sub> exposure assessment study by Liu et al. (1995) found that after  
30 adjusting for time spent in various indoor and outdoor microenvironments (e.g., car with  
31 windows open, car with windows closed, school, work, home, outdoors near home, outdoors

1 other than near home), mean 12-hour ambient O<sub>3</sub> concentrations explained 32% of the variance  
2 in personal exposure in the summer.

3 In a southern California study by Avol et al. (1998b), boys were found to spend more time  
4 outdoors and be more physically active than girls. Another southern California study found that  
5 boys were outdoors 30 minutes longer than girls, and had higher personal O<sub>3</sub> exposure during  
6 both high and low O<sub>3</sub> months (Geyh et al., 2000).

7 The announcement of smog alerts or air quality indices may influence personal exposures  
8 to O<sub>3</sub> by causing individuals to alter behaviors (avoidance behavior). Neidell (2004), in his  
9 evaluation of the effect of pollution on childhood asthma, examined the relationship between the  
10 issuance of smog alerts or air quality indices for several counties in California and hospital  
11 admissions for asthma in children under age 18 years (not including newborns). Smog alerts are  
12 issued in California on days when O<sub>3</sub> concentrations exceed 200 ppb. There was a significant  
13 reduction in the number of asthma-related hospital admissions in children ages 1 to 12 years on  
14 smog alert days, indicating that avoidance behavior might be present on days of high O<sub>3</sub>  
15 concentrations. Changes in population behavior as a function of concentration complicate the  
16 estimation of health effects from population-based studies; thus, it may be desirable to include  
17 sensitivity analyses that eliminate high O<sub>3</sub> days, particularly in areas where avoidance behavior  
18 is expected.

### 19 20 ***Potential Sources of Error Resulting from the Use of Ambient Ozone Concentrations in*** 21 ***Epidemiological Analyses***

22 There is no clear consensus among exposure analysts as to how well stationary monitor  
23 measurements of ambient O<sub>3</sub> concentrations represent a surrogate for personal O<sub>3</sub> exposure. The  
24 approaches available for assessing exposure in air pollution epidemiology studies, the  
25 microenvironmental (indirect) approach and the personal sampling (direct) approach (Navidi  
26 et al., 1999; Ott, 1982, 1985), are associated with measurement error. To determine personal  
27 exposure using the microenvironmental approach, the concentrations of the various  
28 microenvironments are multiplied by the time spent in each microenvironment. Both the  
29 concentration and time component contribute to the measurement error. There is no time  
30 component to the measurement error in the personal sampling approach, however, the estimation  
31 of exposure using personal monitoring devices contributes to measurement error, especially in  
32 the case of O<sub>3</sub>. Passive badges are commonly used for monitoring O<sub>3</sub> integrated personal



1 exposure. Their sensitivity to wind velocity, badge placement, and interference with other  
2 copollutants may result in measurement error.

3 Results from the error analysis models developed by Navidi et al. (1999) indicated that  
4 neither the microenvironmental nor personal sampling approach gave reliable health effect  
5 estimates when measurement errors were uncorrected. The nondifferential measurement error  
6 biased the effect estimates toward zero under the model assumptions. However, if the  
7 measurement error was correlated with the health response, a bias away from the null could  
8 result. The use of central ambient monitors to estimate exposure also biased the estimates  
9 toward the null. Since most people spend the majority of their time indoors, where O<sub>3</sub> levels  
10 tend to be much lower than outdoor ambient levels, using ambient concentrations to determine  
11 exposure generally overestimates true personal O<sub>3</sub> exposure, resulting in effect estimates biased  
12 toward the null.

13 Several studies have examined the relationship between measured ambient O<sub>3</sub>  
14 concentrations from fixed monitoring sites and personal O<sub>3</sub> exposure (Avol et al., 1998a; Brauer  
15 and Brook, 1995, 1997; Chang et al., 2000; Delfino et al., 1996; Lee et al., 2004; Liard et al.,  
16 1999; Linn et al., 1996; Liu et al., 1995, 1997; O'Neill et al., 2003; Sarnat et al., 2001). In a  
17 Baltimore, MD study of older adults, individuals with COPD, and children, 24-h average  
18 ambient O<sub>3</sub> concentrations from a monitoring site were not found to be significantly associated  
19 with personal O<sub>3</sub> exposure (Sarnat et al., 2001). The mixed regression effect estimates were  
20  $\beta = 0.01$  ( $t = 1.21$ ) and  $\beta = 0.00$  ( $t = 0.03$ ), for summer and winter, respectively. Chang et al.  
21 (2000) compared one-hour personal and ambient O<sub>3</sub> measurements in older adults in various  
22 microenvironments using activity data from the National Human Activity Pattern Survey study  
23 (Klepeis, 1999). There was no correlation between personal and ambient O<sub>3</sub> concentrations in  
24 the indoor residence ( $r = 0.09$  and  $r = 0.05$ , for summer and winter, respectively), although a  
25 moderate correlation was found in other indoor environments such as restaurants, hospitals, and  
26 shopping malls ( $r = 0.34$  in summer,  $r = 0.46$  in winter). In comparison, the correlation in  
27 outdoor environments (near and away from roads) was moderate to high ( $0.68 \leq r \leq 0.91$ ) and  
28 statistically significant. Slopes for the relationship between personal and ambient O<sub>3</sub>  
29 concentrations were not reported in this study.

30 Brauer and Brook (1995, 1997) observed that the daily averaged personal O<sub>3</sub> measurements  
31 and ambient concentrations were well-correlated after stratifying groups by time spent outdoors.  
32 Clinic workers ( $n = 25$ ; 24-hour samples), teenage camp counselors ( $n = 25$ ; 24-hour samples),

1 and farm workers (n = 15; 6-14 h work shift samples) spent 0 to 25%, 7.5 to 45%, and 100% of  
2 their monitored time outdoors, respectively. The personal to ambient O<sub>3</sub> concentration ratios  
3 were significantly different for the clinic workers (0.28) and farm workers (0.96). Ambient O<sub>3</sub>  
4 concentrations and time spent outdoors explained more of the variability in the personal O<sub>3</sub>  
5 measurements for outdoor farm workers compared to the clinical workers. However, the  
6 Spearman correlation coefficients were comparable, 0.60 and 0.64 for the clinic workers and  
7 farm workers, respectively, indicating that the variability of nonambient O<sub>3</sub> exposures was  
8 similar in the two groups. A study by O'Neill et al. (2003) examined 107 pairs of ambient and  
9 personal O<sub>3</sub> measurements from 39 outdoor workers in Mexico City using a longitudinal analysis  
10 method. Two to seven personal measurements were collected on each of the 26 monitoring  
11 days, which were averaged then compared with the ambient concentrations. They estimated that  
12 a 1 ppb increase in ambient O<sub>3</sub> concentration was associated with a 0.56 ppb (95% CI: 0.43,  
13 0.69) increase in personal O<sub>3</sub> concentration. In a Paris, France study by Liard et al. (1999),  
14 adults (n = 55) and children (n = 39) wore passive O<sub>3</sub> monitors for 4 consecutive days during  
15 three periods. For each period, all adults wore the O<sub>3</sub> monitors over the same 4 days. Likewise,  
16 all children wore monitors over the same 4 days for each of the three periods, but on different  
17 days from the adults. The ambient O<sub>3</sub> concentrations from the stationary monitoring sites did not  
18 explain a high percentage of the variance of personal O<sub>3</sub> exposure (nonsignificant [value not  
19 stated] in adults and 21% in children). However, when personal measurements from all subjects  
20 were aggregated for each of the six periods, the 4-day mean personal O<sub>3</sub> exposure was found to  
21 be highly correlated with the corresponding mean ambient concentration (r = 0.83, p < 0.05).  
22 Similarly, a study of Los Angeles school children by Linn et al. (1996) found that daily 24-h  
23 average ambient O<sub>3</sub> concentrations from a central site were well-correlated (r = 0.61) with daily  
24 averaged personal O<sub>3</sub> exposures.

25 The low correlation observed between personal O<sub>3</sub> exposures and ambient O<sub>3</sub>  
26 concentrations in the study by Sarnat et al. (2001) suggests that O<sub>3</sub> concentrations measured at  
27 central ambient monitors do not explain the variance of individual personal exposures.  
28 However, daily averaged personal exposures from the aggregate population have been found to  
29 be correlated with monitored ambient O<sub>3</sub> concentrations, which is of greater relevance in time-  
30 series studies. Although there are correlations between aggregate personal and monitored  
31 ambient O<sub>3</sub> concentrations, the absolute personal concentrations may be considerably lower than  
32 the monitored ambient O<sub>3</sub> concentrations.

1 In summary, results indicate that the relationship between ambient O<sub>3</sub> concentrations and  
2 personal exposure will vary depending on factors such as O<sub>3</sub> concentrations, time spent in the  
3 various microenvironments, and activity levels, creating potential measurement errors. The  
4 expectations based on statistical modeling considerations are that these exposure measurement  
5 errors or uncertainties will reduce the statistical power of the O<sub>3</sub> health effects analysis, making  
6 it difficult to detect a true underlying association between the correct exposure metric and the  
7 health outcome studied. However, until more data on O<sub>3</sub> exposure become available, the use of  
8 monitored ambient O<sub>3</sub> concentrations as a surrogate for exposures is not expected to change the  
9 principal conclusions from O<sub>3</sub> epidemiologic studies using community average health and  
10 pollution data.

### 11 *Exposure to Related Photochemical Oxidants*

12 A variety of related photochemical oxidants produced outdoors, such as PAN and  
13 peroxypropionyl nitrate (PPN), can infiltrate into indoor environments. These compounds are  
14 thermally unstable and decompose to peroxyacetyl radicals and NO<sub>2</sub>. Exposure to related  
15 photochemical oxidants has not been measured, nor are these compounds routinely monitored at  
16 stationary monitoring sites. Available monitored concentrations of related photochemical  
17 oxidants may be found in Annex AX3.

## 18 **3.9 SUMMARY OF KEY POINTS**

19 The median of the daily maximum 8-h O<sub>3</sub> concentration averaged over May to September  
20 is about 0.049 ppm from 2000 to 2004. The daily maximum 1-h O<sub>3</sub> concentrations could have  
21 been much higher in large urban areas or in areas downwind of large urban areas. For example,  
22 in Houston, TX, the daily maximum 1-h O<sub>3</sub> concentrations have approached 0.20 ppm during  
23 this period.

24 Daily maximum 8-h average O<sub>3</sub> concentrations are lower than the maximum 1-h O<sub>3</sub>  
25 concentrations, but they are highly correlated. Within individual MSAs, O<sub>3</sub> concentrations tend  
26 to be well correlated across monitoring sites. However, there can be substantial variations in  
27 O<sub>3</sub> concentrations. Ozone in city centers tends to be lower than in regions either upwind or  
28 downwind because of titration by NO emitted by motor vehicles.

1 Ozone concentrations tend to peak in early- to mid-afternoon in areas where there is strong  
2 photochemical activity and later in the day in areas where transport is more important in  
3 determining the O<sub>3</sub> abundance. Summertime maxima in O<sub>3</sub> concentrations occur in areas in the  
4 United States where there is substantial photochemical activity involving O<sub>3</sub> precursors emitted  
5 from human activities. Monthly maxima can occur anytime from June through August.  
6 However, springtime maxima are observed in national parks, mainly in the western United States  
7 and at a number of other relatively unpolluted monitoring sites throughout the Northern  
8 Hemisphere. For example, the highest O<sub>3</sub> concentrations at Yellowstone NP tend to occur  
9 during April and May. Generally, monthly minima O<sub>3</sub> concentrations tend to occur from  
10 November through February at polluted sites and during the fall at relatively remote sites.

11 Nationwide, daily maximum 8-h O<sub>3</sub> concentrations have decreased at the upper end of the  
12 distribution from 1990 to 2004. However, the daily maximum 8-h O<sub>3</sub> concentrations toward the  
13 center of the distribution have not reflected these changes. Trends have not been consistent at  
14 national park sites; with downward trends observed at some sites and upward or no trends  
15 observed at others. At some sites, trends reversed direction in going from the 98th to the 95th  
16 percentile values.

17 Sufficient data are not available for other atmospheric oxidants (e.g., H<sub>2</sub>O<sub>2</sub>, PAN) and  
18 oxidation products (e.g., HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>) to relate concentrations of O<sub>3</sub> to these species for use in  
19 time series studies. Data for these species are only obtained as part of specialized field studies.  
20 In general, secondary species, such as HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, H<sub>2</sub>O<sub>2</sub>, and PAN, are expected to be at least  
21 moderately correlated with O<sub>3</sub>. On the other hand, primary species are expected to be more  
22 highly correlated with each other than with secondary species, provided that the primary species  
23 originate from common sources. Concentrations of other oxidants are much lower than for O<sub>3</sub>  
24 and range from ≤1% for oxidants in particles to several percent for gas phase species. The  
25 relationship of O<sub>3</sub> to PM<sub>2.5</sub> is complex, because PM is not a distinct chemical species but is a mix  
26 of primary and secondary species. PM<sub>2.5</sub> concentrations were positively correlated with O<sub>3</sub>  
27 during summer, but negatively correlated with O<sub>3</sub> during winter at Ft. Meade, MD. PM<sub>10</sub>  
28 concentrations show similar relations with O<sub>3</sub>.

29 Co-occurrences of O<sub>3</sub> (defined when both pollutants are present at an hourly average  
30 concentration of ≥0.05 ppm) with NO<sub>2</sub> and SO<sub>2</sub> are rare. For example, there were fewer than  
31 10 co-occurrences with either NO<sub>2</sub> or SO<sub>2</sub> in 2001. The number of co-occurrences for O<sub>3</sub> and  
32 PM<sub>2.5</sub> (defined as an hourly average O<sub>3</sub> concentration ≥0.05 ppm and a 24-h average PM<sub>2.5</sub>

1 concentration  $\geq 40 \mu\text{g}/\text{m}^3$  occurring during the same 24-h period) also tended to be infrequent  
2 (<10 times) at most sites, but there were up to 20 such co-occurrences at a few sites.

3 Policy relevant background  $\text{O}_3$  concentrations are used for assessing risks to human health  
4 associated with  $\text{O}_3$  produced from anthropogenic sources in continental North America. Because  
5 of the nature of the definition of PRB concentrations, they cannot be directly derived from  
6 monitored concentrations, instead they must be derived from modeled estimates. Current model  
7 estimates indicate that ambient air PRB concentrations in the United States are generally  
8 0.015 ppm to 0.035 ppm. They decline from spring to summer and are generally  $<0.025$  ppm  
9 under conditions conducive to high  $\text{O}_3$  episodes. However, PRB concentrations can be higher,  
10 especially at elevated sites during spring, due to enhanced contributions from hemispheric  
11 pollution and stratospheric exchange.

12 Ozone exposure changes as a function of time of day, season, and microenvironment.  
13 Ambient  $\text{O}_3$  concentrations are generally higher during warmer seasons and during the weekday,  
14 peaking during the later portion of the day. Ozone concentrations in indoor microenvironments  
15 are generally lower than those concentrations encountered in the ambient air. There are few  
16 indoor sources of  $\text{O}_3$ . Ozone occurs in indoor microenvironments primarily through infiltration  
17 through the building envelop and through windows, doors, and ventilation systems. The indoor  
18  $\text{O}_3$  concentration is dependent on the outdoor concentration, the AER, indoor circulation rate,  
19 and removal processes. Consequently, measured and modeled exposures should take into  
20 consideration  $\text{O}_3$  diurnal weekly and seasonal variability and varying microenvironmental  
21 concentrations.

22 Once indoors,  $\text{O}_3$  reacts with indoor surfaces, including surface coverings and furnishings.  
23 Ozone also will react with VOCs in indoor environments, primarily terpenes or terpene-related  
24 compounds. Ozone reactions with pollutants indoors are analogous to those reactions occurring  
25 in the ambient air, potentially exposing subjects to compounds significantly more toxic than  $\text{O}_3$ .  
26 The reaction products include Criegee biradicals, nitrate radicals, peroxyacetyl radicals, and  
27 hydroxy, alkyl, alkyperoxy, hydroperoxy, and alkoxy radicals. The hydroxy radical will react  
28 with various nitrogen compounds, sulfur dioxide, carbon monoxide, and other compounds. The  
29 formation of submicron particles has been attributed to the reaction of  $\text{O}_3$  and the hydroxy  
30 radical with terpene and terpene-related compounds.

31 The available approaches for measuring personal  $\text{O}_3$  exposure include the direct approach,  
32 using a PEM, and the indirect approach, which measures or models exposure in the

1 microenvironments the individual encounters. Both approaches are associated with  
2 measurement errors.

3 There are difficulties in identifying chemical trapping agents for PEMs that can react with  
4 O<sub>3</sub>, and PEMs are sensitive to wind velocity, badge placement, and interference with other  
5 copollutants. Some studies using PEMs have shown personal O<sub>3</sub> exposures below those  
6 concentrations measured at stationary monitoring sites, while other studies have found strong  
7 correlations between O<sub>3</sub> measured at stationary monitoring sites and personal monitored  
8 concentrations.

9 The use of measured O<sub>3</sub> concentrations from stationary ambient monitoring sites as  
10 surrogates for personal exposure may be affected by the O<sub>3</sub> ambient concentration, percentage of  
11 time spent outdoors, and type of outdoor activity. Epidemiologic studies investigating health  
12 outcomes using data from stationary monitoring sites found O<sub>3</sub> exposure to be affected by the  
13 distance between the subjects' location and the stationary monitor, individual activity patterns,  
14 and the O<sub>3</sub> concentration in the microenvironment.

15 The use of exposure models to evaluate O<sub>3</sub> exposure to large populations over time is  
16 complicated by the fact that O<sub>3</sub> is a secondary pollutant with complex nonlinear and multiscale  
17 dynamics in space and time. The existing comprehensive inhalation exposure models (NEM,  
18 pNEM, MENTOR/SHEDS, TRIM.Expo) treat human activity patterns as sequences of exposure  
19 events. Estimates of activity levels are assigned from CHAD, the Consolidated Human  
20 Activities Database.

21 Ambient O<sub>3</sub> concentrations are estimated using emissions-based mechanistic models or  
22 ambient-data-based models. Models for estimating microenvironmental concentrations include  
23 the empirical, mass balance, and detailed CFD models. Mass balance modeling is the most  
24 common modeling approach to estimating concentrations in enclosed microenvironments. The  
25 pNEM/O<sub>3</sub> population exposure model, the model used more extensively in O<sub>3</sub> exposure  
26 modeling, includes a sophisticated mass balance model for indoor and vehicle  
27 microenvironments. There are three versions of the pNEM/O<sub>3</sub> model: the general population,  
28 outdoor workers, and outdoor children.

29 Results from O<sub>3</sub> exposure studies indicate that the relationship between ambient O<sub>3</sub>  
30 concentrations and personal exposure/dose will vary depending on O<sub>3</sub> concentrations and time  
31 spent in the various microenvironments, particularly the time spent outdoors where O<sub>3</sub>  
32 concentrations tend to be higher, and the personal activity level. Consequently, the O<sub>3</sub>

1 exposure/dose may differ from the concentrations measured at stationary monitoring sites.  
2 However, until more data on O<sub>3</sub> exposure become available, the use of monitored ambient O<sub>3</sub>  
3 concentrations as a surrogate for exposures is not expected to change the principal conclusions  
4 from O<sub>3</sub> epidemiologic studies using community average health and pollution data.

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## 4. DOSIMETRY, SPECIES HOMOLOGY, SENSITIVITY, AND ANIMAL-TO-HUMAN EXTRAPOLATION

### 4.1 INTRODUCTION

The dosimetry of ozone (O<sub>3</sub>) in humans has been examined in a series of studies published in the past decade. These studies further characterize the dose of O<sub>3</sub> delivered to various sites in the respiratory tract (RT). Ozone, classified as a reactive gas, interacts with surfactant, antioxidants, and other compounds in the epithelial lining fluid (ELF). Researchers have attempted to obtain a greater understanding of how these complex interactions affect O<sub>3</sub> uptake and O<sub>3</sub>-induced injury. New work has also been completed evaluating species differences in responses to O<sub>3</sub> exposures, which allow more accurate quantitative extrapolation from animals to humans.

This chapter is not intended to be a complete overview of O<sub>3</sub> dosimetry and animal-to-human comparisons, but rather, it is an update of the dosimetry/extrapolation chapter from the last O<sub>3</sub> criteria document (U.S. Environmental Protection Agency, 1996), or 1996 O<sub>3</sub> AQCD, and other reviews of the earlier published literature. The framework for presenting this chapter is first a discussion in Section 4.2 of general concepts of the dosimetry of O<sub>3</sub> in the RT. Bolus-response studies are then presented in Section 4.2.1 followed by general uptake studies in Section 4.2.2. Dosimetry modeling is presented in Section 4.2.3 followed by the summary and conclusions for the dosimetry material in Section 4.2.4. The chapter continues in Section 4.3 with a discussion of species comparisons and ends with a discussion of animal-to-human extrapolation. More detailed discussions of the studies are presented in the supporting material to this chapter (Annex AX4). The toxicological effects of O<sub>3</sub> in laboratory animals and in vitro test systems are discussed in Chapter 5 and direct effects of O<sub>3</sub> in humans are discussed in Chapter 6. The historical O<sub>3</sub> literature is very briefly summarized in this chapter, providing a very concise overview of previous work. The reader is referred to the 1996 O<sub>3</sub> AQCD for more detailed discussion of the literature prior to the early 1990s.

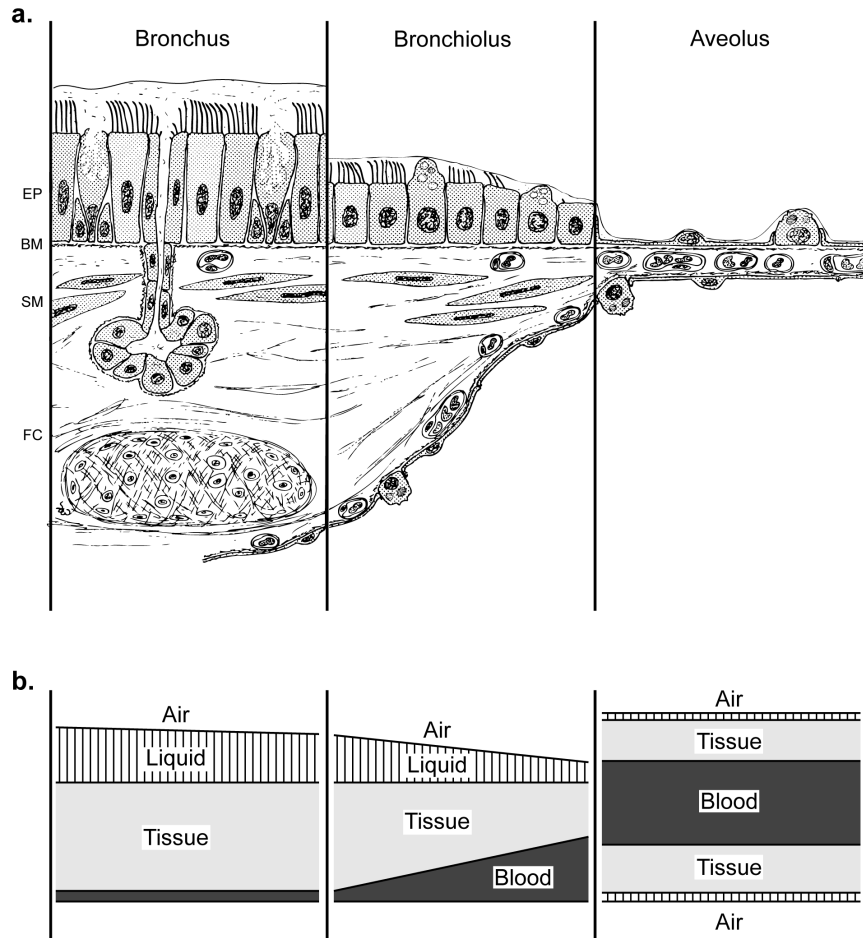
## 4.2 DOSIMETRY OF OZONE IN THE RESPIRATORY TRACT

Ozone dosimetry refers to the measurement or estimation of the amount of O<sub>3</sub> or its reaction products reaching and persisting at specific sites in the RT following an exposure. The compound most directly responsible for toxic effects may be the inhaled gas O<sub>3</sub> or one of its chemical reaction products. Complete identification of the actual toxic agents and their integration into dosimetry is a complex issue that has not been resolved. Dosimetric studies attempt to quantify the amount of O<sub>3</sub> retained in the lung (i.e., not exhaled) or the dose of O<sub>3</sub> or its active metabolites delivered to target cells or tissues (i.e., dose per cell or tissue surface area). For comparison, epidemiologic studies may simply consider exposure concentration while clinical studies may consider the total amount of O<sub>3</sub> inhaled (product of exposure concentration, duration, and minute ventilation). Hence, dosimetric studies seek to accurately quantify dose to target lung regions or tissues, whereas epidemiologic and clinical studies typically consider exposures.

Understanding dosimetry as it relates to O<sub>3</sub>-induced injury is complex due to the fact that O<sub>3</sub> interacts primarily with the ELF which contains surfactant and antioxidants. In the upper airways ELF is thick and highly protective against oxidant injury. Figure 4-1 illustrates the structure of the lower airways with progression from the large airways to the alveolus. In lower airways ELF is thinner, has lower levels of antioxidants, and thus, allows more cellular injury. Adding to the complexity is the fact that O<sub>3</sub> can react with molecules in the ELF to create even more reactive metabolites, which can then diffuse within the lung or be transported out of the lung to generate systemic effects.

A considerable number of dosimetric studies were summarized in the 1996 O<sub>3</sub> AQCD. These studies provided estimates of absorbed O<sub>3</sub> in the RT as a whole or in regions such as the upper airways (URT) or lower airways (LRT), defined as being proximal or distal to the tracheal entrance, respectively. Estimates were obtained for both humans and animals via direct measurement and mathematical modeling. The mathematical models also estimated O<sub>3</sub> doses to specific target sites such as the proximal alveolar region (PAR; first generation distal to the terminal bronchioles) and the centriacinar region (CAR; junction of conducting airways and gas exchange region).





**Figure 4-1. Structure of lower airways with progression from the large airways to the alveolus. Panel (a) illustrates basic airway anatomy. Structures are epithelial cells, EP; basement membrane, BM; smooth muscle cells, SM; and fibrocartilaginous coat, FC. Panel (b) illustrates the relative amounts of liquid, tissue, and blood with distal progression. In the bronchi there is a thick liquid lining over a relatively thick layer of tissues. Even highly soluble materials moving from the air into the liquid layer have minimal systemic access via the blood. With distal progress, the protective liquid lining diminishes allowing increased access of compounds crossing the air-liquid interface to the tissues and the blood.**

Source: Panel (a) reproduced with permission (Weibel, E. R. [1980] Design and structure of the human lung. In: Fishman, A. P., ed. Pulmonary Diseases and Disorders. New York, NY: McGraw-Hill; p. 231).

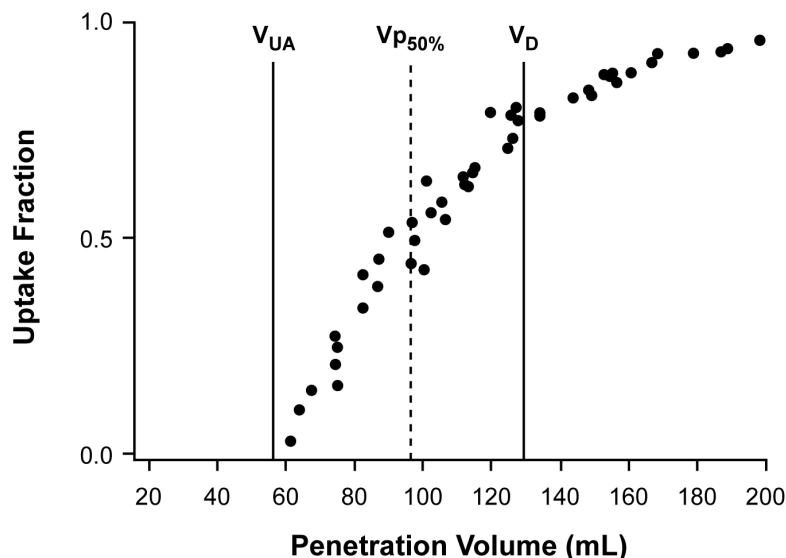
1 In general, the consensus of experimental and modeling studies summarized in the  
2 1996 O<sub>3</sub> AQCD supported the following conclusions: (1) for the URT, animal and human  
3 studies suggested that O<sub>3</sub> uptake is greater in the nose than the mouth but the effect of flow on  
4 uptake was equivocal; (2) for the LRT, predicted tissue doses (O<sub>3</sub> flux to liquid-tissue interface)  
5 were very low in the trachea, increased to a maximum in the terminal bronchioles or first airway  
6 generation in the pulmonary region, and rapidly decreased with distal progression; (3) increasing  
7 tidal volume (V<sub>T</sub>) increases O<sub>3</sub> uptake, whereas, increasing flow or breathing frequency (f<sub>B</sub>)  
8 decreases O<sub>3</sub> uptake; (3) increasing flow shifts O<sub>3</sub> uptake to the smaller peripheral airways, i.e.,  
9 toward the CAR; and (4) similarly, the effect of exercise is to significantly increase the  
10 pulmonary region total dose (mass of O<sub>3</sub>) and the CAR dose (mass per unit surface area).

11 Some cross-species *in vivo* comparisons were described in the 1996 O<sub>3</sub> AQCD.  
12 For instance, comparing bronchoalveolar lavage (BAL) cells from rats and humans, it was  
13 estimated that a 0.4 ppm O<sub>3</sub> exposure in exercising humans gave 4 to 5 times the O<sub>3</sub> dose  
14 (retained) relative to rats exposed at rest to the same concentration. *In vitro* dosimetry studies in  
15 the 1996 O<sub>3</sub> AQCD using isolated lung preparations showed that uptake efficiency is chemical-  
16 reaction dependent, indicating the importance of reaction product formation. These reaction  
17 products, created mainly by the ozonolysis of polyunsaturated fatty acids, included hydrogen  
18 peroxide, aldehydes, and hydroxyhydroperoxides, which are mediators of O<sub>3</sub> toxicity. Other  
19 products are created by the reaction of O<sub>3</sub> with other ELF constituents, all of which must be  
20 considered in understanding the dosimetry of O<sub>3</sub>.

21 The next two sections (4.2.1 and 4.2.2) review the available new experimental studies  
22 on O<sub>3</sub> dosimetry, all of which were conducted by Ultman and colleagues. Table AX4-1 in  
23 Annex AX4 summarizes these studies.

#### 24 25 **4.2.1 Bolus-Response Studies**

26 The bolus-response method has been used by the Ultman group as an approach to explore  
27 the distribution of O<sub>3</sub> absorption in the airways of humans. This non-invasive method consists  
28 of an injection of a known volume and concentration of O<sub>3</sub> during inspiration. Ozone uptake is  
29 the amount of O<sub>3</sub> absorbed during a breath relative to the amount contained in the inhaled  
30 bolus. Figure 4-2 illustrates the uptake of a series of O<sub>3</sub> boli as a function of volumetric  
31 penetration (V<sub>p</sub>), i.e., the volume between the center of mass of an inhaled bolus and the end of



**Figure 4-2. Ozone uptake fraction as a function of volumetric penetration ( $V_p$ ) in a representative subject. Each point represents the  $O_3$  uptake of a bolus inspired by the subject. The volumes,  $V_{UA}$  and  $V_D$ , are the volume of the upper airways and anatomical dead space, respectively, and  $V_{P_{50\%}}$  is the  $V_p$  at which 50% of the inspired bolus was absorbed. In 47 healthy subjects (24 M, 23 F), Ultman et al. (2004) found that  $V_{P_{50\%}}$  was well correlated with  $V_D$  ( $r = 0.57$ ,  $p < 0.001$ ) and better correlated with the volume of the conducting airways, i.e.,  $V_D$  minus  $V_{UA}$ , ( $r = 0.65$ ,  $p = 0.001$ ).**

Source: Adapted from Ultman et al. (2004).

1 inspiration. The inspired  $O_3$  boli (for which the uptake fractions are illustrated in Figure 4-2)  
 2 were 20 ml of 2 ppm  $O_3$ . Kabel et al. (1994) have previously shown that varying the  $O_3$   
 3 concentration of inspired boli between 0.4 and 4 ppm does not affect the distribution of uptake as  
 4 a function of  $V_p$ .

5 The  $O_3$  bolus-technique was used by Bush et al. (1996a) to ascertain differences in lung  
 6 anatomy and gender that can alter the exposure-dose cascade. Forced vital capacity (FVC), total  
 7 lung capacity (TLC) and anatomic dead space ( $V_D$ ) were determined for ten male and ten female  
 8 subjects, who then inhaled to a 20 ml bolus of 3 ppm  $O_3$  injected into the airstream. In all  
 9 subjects, dosimetry differences could be explained by differences in  $V_D$ . In a subsequent study,  
 10 Ultman et al. (2004) showed that the volume at which 50% of an inspired  $O_3$  bolus is absorbed

1 was better associated with the volume of the conducting airways than  $V_D$  (see Figure 4-2).  
2 Bush et al. (1996a) pointed out that the applicability of their results may be limited because of  
3 their assumptions that the intrinsic mass transfer parameter ( $K_a$ ) was independent of location in  
4 the RT and that there was no mucous resistance. They further suggested that the dependence  
5 of  $K_a$  on flowrate and  $V_D$  be restricted to flowrates  $\leq 1000$  mL/s until studies at higher rates have  
6 been performed.

7 Nodelman and Ultman (1999) demonstrated that the uptake distributions of  $O_3$  boli were  
8 sensitive to the mode of breathing and to the airflow rate. As flowrates increased from 150 to  
9 1000 mL/s,  $O_3$  penetrated deeper into the lung and penetration was further increased by oral  
10 relative to nasal breathing. The authors suggest that the switch from nasal to oral breathing  
11 coupled with increases in respiratory flow as occurs during exercise causes a shift in the  $O_3$  dose  
12 distribution, allowing  $O_3$  to penetrate deeper into the lung, increasing the potential for damage to  
13 bronchiolar and alveolar tissues.

14 More recently, Ultman et al. (2004) measured  $O_3$  uptake using the bolus technique in  
15 60 young healthy nonsmoking adults (32 M, 28 F). Bolus were inspired at a rate of 1 mL/s,  
16 equivalent to a moderate exercise rate with a minute ventilation of 30 L/min. Figure 4-2  
17 illustrates the  $O_3$  uptake fraction as a function of  $V_p$  in a representative subject. Anatomic dead  
18 space was measured in 47 of the subjects (24 M, 23 F). In these subjects, the volume at which  
19 50% of an inhaled bolus was absorbed ( $V_{P_{50\%}}$ ) was correlated with  $V_D$  ( $r = 0.57$ ,  $p < 0.001$ ) and  
20 the volume of the conducting airways, i.e.,  $V_D$  minus the volume of the upper airways, ( $r = 0.65$ ,  
21  $p = 0.001$ ). Both  $V_{P_{50\%}}$  and  $V_D$  were significantly greater in males than females, although the  
22 volume of the upper airways was not. These findings suggest that in females the smaller  
23 airways, and associated larger surface-to-volume ratio, enhance local  $O_3$  uptake and cause  
24 reduced penetration of  $O_3$  into the distal lung. It is not clear from these findings, however, if the  
25 actual anatomical location of  $V_{P_{50\%}}$  differed between males and females.

26 A few studies have measured the effect of a continuous pollutant exposure on  $O_3$  bolus  
27 uptake. Asplund et al. (1996) randomly exposed young healthy adults (8 M, 3 F) for 2 h  
28 [presumably at rest] to 0.0 (air), 0.12, or 0.36 ppm  $O_3$  on 3 separate occasions separated by at  
29 least 1-wk. Ozone bolus uptake was measured preexposure and subsequently at 30 minute  
30 intervals during the exposure. Ozone uptake over the  $V_p$  range of 70 to 120 ml increased  
31 after the air exposure, decreased slightly after the 0.12 ppm  $O_3$  exposure, and decreased

1 more substantially following the 0.36 ppm O<sub>3</sub> exposure. Relative to uptake during the air  
2 exposure, O<sub>3</sub> bolus uptake was significantly decreased by 30 minutes of the 0.12 and  
3 0.36 ppm O<sub>3</sub> exposures and remained significantly decreased for the duration of these exposures.

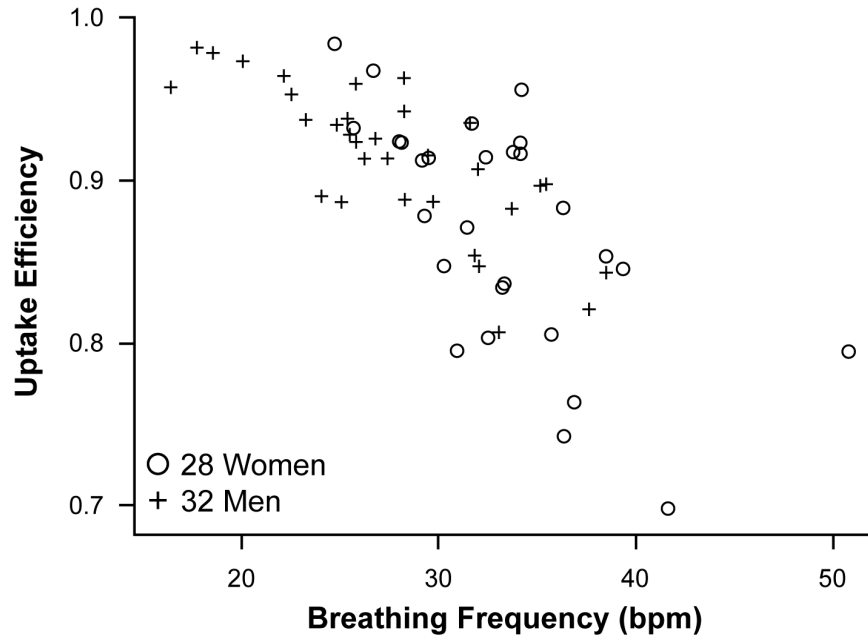
4 Using a similar protocol, Rigas et al. (1997) randomly exposed young healthy adults (6 M,  
5 6 F) for 2 h at rest to filtered air, 0.36 ppm NO<sub>2</sub>, 0.75 ppm NO<sub>2</sub>, 0.36 ppm SO<sub>2</sub>, or 0.36 ppm O<sub>3</sub>.  
6 Ozone bolus uptake (V<sub>p</sub> range of 70 to 120 ml ) was measured preexposure and every 30 minute  
7 during the exposures. The results of an *F* test indicated that exposure duration (30-, 60-, 90-,  
8 120-min) was not a significant factor, but treatment (NO<sub>2</sub>, SO<sub>2</sub>, etc.) was (*p* < 0.01). Ozone  
9 bolus uptake was increased by 30 minutes during the NO<sub>2</sub> and SO<sub>2</sub> exposures and decreased  
10 during the O<sub>3</sub> exposure. The authors suggested that there may be increased production of  
11 an O<sub>3</sub>-reactive substrate in the ELF due to airway inflammation. During NO<sub>2</sub> and SO<sub>2</sub> exposures  
12 the substrate was not depleted by these gases and so could react with the O<sub>3</sub> bolus. During O<sub>3</sub>  
13 exposure the substrate was depleted, causing the fractional absorption of the O<sub>3</sub> bolus to  
14 decrease.

#### 16 **4.2.2 General Uptake Studies**

17 Ultman and colleagues have recently completed some general uptake studies to determine  
18 the ratio of O<sub>3</sub> uptake to the quantity of O<sub>3</sub> inhaled. Uptake efficiency was determined at  
19 exposures of 0.2 or 0.4 ppm O<sub>3</sub> while exercising at a minute volume of approximately 20 L/min  
20 for 60 minutes or 40 L/min for 30 minutes in both men and women (Rigas et al., 2000). Uptake  
21 efficiency ranged from 0.56 to 0.98 and had a statistically significant but weak dependence on  
22 concentration, minute volume, and exposure time. Intersubject differences had the largest  
23 influence on uptake efficiency, resulting in a variation of approximately 10%. As the quantity  
24 of O<sub>3</sub> retained by the RT is equal to uptake efficiency times the quantity of O<sub>3</sub> inhaled, relatively  
25 large changes in concentration, minute volume, or exposure time may result in relatively large  
26 changes in the amount of O<sub>3</sub> retained by the RT or absorbed locally. The authors concluded that  
27 for exposure times <2 h, inhaled dose (product of O<sub>3</sub> concentration, exposure duration, and  
28 minute ventilation) is a reasonable predictor of actual uptake as long as there are fixed  
29 concentrations of O<sub>3</sub> and fixed levels of exercise. More importantly, similarly exposed  
30 individuals vary in the amount of actual dose received.

1           Santiago et al. (2001) studied the effects of airflow rate (3 to 15 L/min) and O<sub>3</sub>  
2 concentration (0.1, 0.2, or 0.4 ppm) on O<sub>3</sub> uptake in nasal cavities of males and females.  
3 As would be expected, uptake efficiency in the nose was inversely related to the flowrate and the  
4 concentration of O<sub>3</sub> in the inlet air. They computed a gas-phase diffusion resistance of <24% of  
5 overall diffusion resistance which suggested to them that simultaneously occurring diffusion and  
6 chemical reactions in the mucous layer were the limiting factors in O<sub>3</sub> uptake. Difference in O<sub>3</sub>  
7 uptake ranged from 0.63 to 0.97 at flowrates of 3 L/min and 0.25 to 0.50 at 15 L/min. The small  
8 effects of flowrate and concentration on uptake efficiency were statistically significant, but  
9 intersubject differences accounted for approximately half of the total variation in uptake  
10 efficiency. Both these general uptake studies, done at environmentally relevant O<sub>3</sub>  
11 concentrations, indicate that inter-individual differences in fractional uptake are extremely  
12 important in O<sub>3</sub> dose-response relationships.

13           In the research mentioned above, Ultman et al. (2004) also completed continuous exposure  
14 studies. The same 60 subjects were exposed continuously for 1 h to either clean air or 0.25 ppm  
15 ozone while exercising at a target minute ventilation of 30 L/min. This is the first study to assess  
16 ventilatory and dosimetric parameters for an entire hour of exposure. Additionally they  
17 measured bronchial cross-sectional area available for gas diffusion in addition to other  
18 ventilatory parameters. At a fixed minute ventilation of 30 L/min, the uptake fraction of O<sub>3</sub>  
19 decreased with increasing  $f_B$  (see Figure 4-3) and increased with increasing  $V_T$ . The uptake  
20 fraction was significantly greater in males (91.4%) than females (87.1%), which is consistent  
21 with the larger  $f_B$  and smaller  $V_T$  of the females than males. There was a small but significant  
22 reduction in the breath-by-breath uptake of O<sub>3</sub> from 90.6% on average for the first 15 minutes to  
23 87.3% on average for the last 15 minutes of exposure. Ozone uptake rate correlated with percent  
24 changes in individual bronchial cross-sectional area but did not correlate with individual FEV<sub>1</sub>  
25 responses. Neither of these parameters correlated with the penetration volume determined in the  
26 bolus studies mentioned above. The authors concluded that the intersubject differences in forced  
27 respiratory responses were not due to differences in O<sub>3</sub> uptake. However, these data did partially  
28 support the hypothesis that changes in cross-sectional area available for gas diffusion are related  
29 to overall O<sub>3</sub> retention.  
30



**Figure 4-3. Ozone uptake efficiency as a function of breathing frequency at a minute ventilation of 30 L/min. The uptake efficiency was well correlated with breathing frequency ( $r = -0.723$ ,  $p < 0.001$ ) and tidal volume (*not illustrated*;  $r = 0.490$ ,  $p < 0.001$ ).**

Source: From Ultman et al. (2004).

### 4.2.3 Dosimetry Modeling

When all of the animal and human in vivo O<sub>3</sub> uptake efficiency data are compared, there is a good degree of consistency across data sets, which raises the level of confidence with which these data sets can be used to support dosimetric model formulations. Models predict that the net O<sub>3</sub> dose (O<sub>3</sub> flux to air-liquid interface) gradually decreases distally from the trachea toward the end of the TB and then rapidly decreases in the pulmonary region. However, the tissue dose (O<sub>3</sub> flux to liquid-tissue interface) is low in the trachea, increases to a maximum in the terminal bronchioles and the first generation of the pulmonary region, and then decreases rapidly distally into the pulmonary region. The increased V<sub>T</sub> and flow, associated with exercise in humans or CO<sub>2</sub>-stimulated ventilation increases in rats, shifts O<sub>3</sub> dose further into the periphery of the lung, causing a disproportionate increase in distal lung dose.

1           Localized damage to lung tissue has been modeled showing variation of O<sub>3</sub> dose among  
2 anatomically equivalent ventilatory units as a function of path length from the trachea with  
3 shorter paths showing greater dose (Overton and Graham, 1995). More recent data indicate that  
4 the primary site of acute cell injury occurs in the conducting airways (Postlethwait et al., 2000).  
5 These data must be considered when developing models that attempt to predict site-specific  
6 locations of O<sub>3</sub>-induced injury. The early models computed relationships between delivered  
7 regional dose and response with the assumption that O<sub>3</sub> was the active agent responsible for  
8 injury. It is now known that reactive intermediates such as hydrohydroperoxides and  
9 aldehydes are important agents mediating the response to O<sub>3</sub> (further discussed in Section 5.3.1).  
10 Thus, models must consider O<sub>3</sub> reaction/diffusion in the ELF and ELF-derived reactions  
11 products.

12           Table AX4-2 in the annex presents a summary of new theoretical studies of the uptake  
13 of O<sub>3</sub> by the RTs (or regions) of humans and laboratory animals that have become available  
14 since the 1996 review. They are discussed below.

15           Overton and Graham (1995) created a rat model combining multiple path anatomic models  
16 and one-dimensional convection-dispersion equations which simulates transport and uptake  
17 of O<sub>3</sub> in airways and airspaces of the modeled TB region. Predictions from this model  
18 realistically detail O<sub>3</sub> transport and uptake of different but morphologically equivalent sites.  
19 Using computational fluid dynamics (CFD), Cohen-Hubal et al. (1996) modeled the effect of the  
20 mucus layer thickness in the nasal passage of a rat. Predictions of overall uptake were within the  
21 range of measured uptake. Predicted regional O<sub>3</sub> flux was correlated with measured cell  
22 proliferation for the CFD simulation that incorporated two regions, each with a different mucus  
23 thickness. But using bolus-response data described above, Hu et al. (1994) and Bush et al.  
24 (2001) estimate a reaction rate constant that is more than 1000 times as large as that used by  
25 Cohen-Hubal et al. (1996).

26           With a RT dosimetry model, Overton et al. (1996) investigated the sensitivity of uptake  
27 efficiency, proximal alveolar region (PAR) dose, and PAR dose ratio to TB region volume ( $V_{TB}$ )  
28 and TB region expansion in humans and rats. The PAR was defined as the first generation distal  
29 to terminal bronchioles and the PAR dose ratio was defined as the ratio of a rat's predicted PAR  
30 dose to a human's predicted PAR dose. This ratio relates human and rat exposure concentrations  
31 so that both species receive the same PAR dose. In rats, the PAR is a region of major damage



1 from  $O_3$ . For each species, three values of  $V_{TB}$  were used: a mean value from the literature and  
2 the mean  $\pm$  twice the SD. For both the rat and human simulations, there were several general  
3 findings: (1) uptake efficiency and PAR dose both increased with decreasing  $V_{TB}$ , e.g., using the  
4 highest TB region mass transfer coefficient ( $k_{TB}$ ), the PAR dose for  $V_{TB} - 2SD$  was five times  
5 greater than the PAR dose for  $V_{TB} + 2SD$ , (2) uptake efficiency and PAR dose both decreased  
6 with TB expansion relative to no expansion, 3) PAR dose increased with tidal volume,  
7 4) PAR dose increased with decreasing  $k_{TB}$ , and 5) uptake efficiency increased with  $k_{TB}$ .

8 Bush et al. (2001) modified their single-path model (Bush et al., 1996b) so that simulations  
9 would coincide with experimental uptake efficiency data for  $O_3$  and  $Cl_2$  during oral and nasal  
10 breathing. Relative to their original model, the Bush et al. (2001) model added lung expansion  
11 and modified the mass transfer coefficients for both the gas-phase ( $k_g$ ) and the liquid-phase ( $k_l$ ).  
12 Consistent with Overton et al. (1996), considering expansion of the TB airways reduced uptake  
13 efficiency versus no expansion. As very little inhaled  $O_3$  reaches the peripheral lung, it was not  
14 surprising that alveolar expansion had minimal affect on uptake efficiency. Ignoring the  $O_3$   
15 reaction rate constant ( $k_r$ ), the simulations for  $O_3$  and  $Cl_2$  were nearly the same since the gas-  
16 phase diffusion coefficients of  $O_3$  and  $Cl_2$  are similar. But for a given  $V_p$  the TB airways of the  
17 lung, experimental bolus uptake are always less for  $O_3$  than for  $Cl_2$ . The authors surmised that  
18 the difference between the uptake for these gases could be explained adequately based solely on  
19 the diffusive resistance of  $O_3$  in airways surface fluid (modeled by  $k_r$ ). Qualitatively, model  
20 simulations also agreed well with the experimental data of Gerrity et al. (1995).

21 Age- and gender-specific differences in both regional and systemic uptake in humans was  
22 modeled using a physiologically-based pharmacokinetic (PBPK) approach (Sarangapani et al.,  
23 2003). The model estimated that regional (URT, TB, pulmonary) extraction efficiency of  $O_3$  is  
24 relatively insensitive to age and gender.

25 A recent attempt was made (Mudway and Kelly, 2004) to model  $O_3$  dose-inflammatory  
26 response using a meta-analysis of 23 exposures in published human chamber studies. The  $O_3$   
27 concentrations ranged from 0.08 to 0.6 ppm and the exposure durations ranged from 60 to  
28 396 minutes. The analysis showed linear relationships between  $O_3$  dose and neutrophilia in  
29 bronchoalveolar lavage fluid (BALF). Linear relationships were also observed between  $O_3$  dose  
30 and protein leakage into BALF, which suggested to the authors that a large-scale study could  
31 determine a possible  $O_3$  threshold level for these inflammatory responses. These recent findings

1 seem consistent with the linear relationship between O<sub>3</sub> dose to pulmonary tissues normalized  
2 for body weight and lavage fluid protein in rats, guinea pigs, and rabbits (Miller et al., 1988).

#### 4 **4.2.4 Summary and Conclusions - Dosimetry**

5 Ozone is a highly reactive gas and powerful oxidant with a short half-life. Uptake occurs  
6 in mucous membranes of the RT where O<sub>3</sub> reacts with components of the ELF. Uptake  
7 efficiency is chemical-reaction dependent and the reaction products (hydrogen peroxide,  
8 aldehydes, and hydroxyhydroperoxides) created by ozonolysis of polyunsaturated fatty acids  
9 mediate O<sub>3</sub> toxicity. The 1996 O<sub>3</sub> AQCD reported that uptake of O<sub>3</sub> in rat is about 0.50 and in  
10 humans at rest is about 0.8 to 0.95. In humans, about 0.07 of the O<sub>3</sub> is removed in the  
11 larynx/trachea, about 0.50 in the head, and about 0.43 in the lungs, where the primary site of  
12 damage was believed to be the CAR. Increasing flow shifted O<sub>3</sub> uptake distally toward smaller  
13 airways of the lung. Studies in humans showed that increasing minute ventilation with exercise  
14 (by increasing both breathing frequency and tidal volume) causes only a small decrease in  
15 uptake efficiency by the total RT. The nasal passages appeared to absorb more O<sub>3</sub> than the oral  
16 passages. Comparing BAL cells, a 0.4 ppm exposure in exercising humans showed 4 to 5 times  
17 the retained dose of O<sub>3</sub> relative to rats exposed at rest to the same concentration.

18 New research on O<sub>3</sub> uptake has been performed in humans but not in laboratory animals.  
19 Bolus-response studies demonstrated that a previous continuous exposure to O<sub>3</sub> decreases the  
20 absorption of a bolus of O<sub>3</sub>, probably due to depletion of compounds able to absorb O<sub>3</sub>.  
21 Continuous exposure to NO<sub>2</sub> and SO<sub>2</sub> increased absorption of a bolus of O<sub>3</sub>. These data are of  
22 some relevance to environmental exposures where humans may receive differing concentrations  
23 of O<sub>3</sub> depending on time of day. Verifying prior work, the bolus-response method was used to  
24 demonstrate that O<sub>3</sub> bolus uptake is sensitive to the mode of breathing and to the airflow rate.  
25 As flow is increased from 150 to 1000 mL/s, O<sub>3</sub> boli penetrated deeper into the lung and  
26 penetration was further increased by oral versus nasal breathing. This suggests that the switch  
27 from nasal to oral breathing coupled with increases in respiratory flow as occurs during exercise  
28 causes a shift in regional O<sub>3</sub> dose deeper into the lung, increasing the potential of damage to  
29 bronchiolar and alveolar tissues. The finding that O<sub>3</sub> uptake is inversely related to airflow also  
30 agrees with earlier animal studies.

1 New general uptake study data demonstrate that exercising men and women receiving  
2 0.2 or 0.4 ppm O<sub>3</sub> at 20 L/min for 60 minutes or 40 L/min for 30 minutes absorb 0.56 to 0.98.  
3 The absorbed fraction or FA is affected only by large changes in concentration, minute volume,  
4 and exposure time. This suggests that for exposure times <2 h, inhaled dose is a reasonable  
5 predictor of actual uptake as long as there are fixed concentrations of O<sub>3</sub> and fixed levels of  
6 exercise. Individuals exposed to similar concentrations vary considerably in the amount of  
7 actual dose received. This intersubject variability was also demonstrated in a study of O<sub>3</sub> uptake  
8 in nasal cavities of men and women. The FA in the nose was inversely related to the flowrate  
9 and the concentration of O<sub>3</sub>, suggesting that simultaneously occurring diffusion and chemical  
10 reactions in the mucous layer were the limiting factors in O<sub>3</sub> uptake. Both these general uptake  
11 studies, done at environmentally relevant O<sub>3</sub> concentrations, indicate that inter-individual  
12 differences in fractional uptake, which can range from 0.25 to 0.97, are extremely important  
13 in O<sub>3</sub> dose-response relationships.

14 The consistency of uptake data generated in animal and human studies allow a high level  
15 of confidence in their use in dosimetry modeling. Early models predicted that net O<sub>3</sub> dose to  
16 ELF and tissue gradually decreases distally from the trachea toward the end of the TB and then  
17 rapidly decreases in the pulmonary region. Exercise-induced or CO<sub>2</sub>-stimulated increases in V<sub>T</sub>  
18 and flow, shift O<sub>3</sub> dose further into the periphery of the lung, causing a disproportionate increase  
19 in distal lung dose. Localized damage to lung tissue has been modeled showing variation of O<sub>3</sub>  
20 dose among anatomically equivalent ventilatory units as a function of path length from the  
21 trachea with shorter paths showing greater damage.

22 New models have produced some refinements of earlier models such as: (1) the use of  
23 mucus resistance and thickness in describing O<sub>3</sub> dosimetry and determining the patterns  
24 of O<sub>3</sub>-induced lesions; (2) the shape of the dose versus generation plot along any path from the  
25 trachea to alveoli is independent of path, with the tissue dose decreasing with increasing  
26 generation index; (3) simulations sensitive to conducting airway volume but relatively  
27 insensitive to characteristics of the respiratory airspace; (4) the importance of TB region  
28 expansion; (5) the importance of dose received in the PAR both inter-individual differences and  
29 extrapolations based on dose; and (6) reevaluation of mass transfer coefficients for conducting  
30 airways. Additionally, more recent data indicate that the primary site of acute cell injury occurs

1 in the conducting airways and that reactive intermediates in the ELF, rather than O<sub>3</sub> itself, are  
2 responsible for pulmonary injury. These data must be considered when developing new models.

### 3 4 5 **4.3 SPECIES HOMOLOGY, SENSITIVITY, AND ANIMAL-TO-HUMAN** 6 **EXTRAPOLATION**

7 Basic similarities exist across human and other animals species with regard to basic  
8 anatomy, physiology, biochemistry, cell biology, and disease processes. However, there are  
9 obviously some species differences that have the potential to affect both the patterns of O<sub>3</sub>  
10 uptake in the respiratory tract as well as responses. For instance, primates are oronasal breathers  
11 with a dichotomous branching lung structure, whereas, rodents are obligate nasal breathers with  
12 a monopodial branching lung structure (Miller et al., 1993). Even when comparing nasal  
13 breathing, differences in the nasal structure between primates and rodents can affect both the site  
14 and amount of gaseous uptake in this region (DeSesso, 1993; Morgan et al., 1989). Cellular  
15 profiles also differ between species as a function of location in the respiratory tract (Miller et al.,  
16 1993; Plopper et al., 1989; Stone et al., 1992).

17 The homology as it exists creates similarities in acute O<sub>3</sub>-induced effects, especially in the  
18 respiratory tract and in lung defense mechanisms. Rodents appear to have a slightly higher  
19 tachypneic response to O<sub>3</sub>, which is clearly concentration-dependent in most species and shows  
20 parallel exacerbation when hyperventilation (e.g., exercise or CO<sub>2</sub>) is superimposed. What is not  
21 known is whether this is evidence of pulmonary irritant sensitivity, perhaps as a prelude to  
22 toxicity, or whether tachypnea is a defensive action taken by the respiratory system to minimize  
23 distal lung O<sub>3</sub> deposition. Airway or lung resistance in humans is not affected appreciably by  
24 acute exposure to O<sub>3</sub>, except under conditions of heavy exercise; animals appear to need high-  
25 level exposures or special preparations that bypass nasal scrubbing. Dynamic lung compliance  
26 (C<sub>dyn</sub>) has been shown to have small magnitude decreases in response to O<sub>3</sub> in some studies  
27 across species, but it is thought that these changes are of little biological significance for ambient  
28 exposures. Spirometric changes, the hallmark of O<sub>3</sub> response in humans, occur in rats, but to a  
29 lesser degree. It is unclear, however, the degree to which anesthesia (rat) and the comparability  
30 of hyperventilation induced by CO<sub>2</sub> (rat) or exercise (human) may influence this difference in

1 responsiveness. Collectively, the acute functional response of laboratory animals to O<sub>3</sub> appears  
2 quite homologous to that of the human.

3 Examination of BAL constituents show that the influx of inflammatory cells and protein  
4 from the serum is influenced by species, but perhaps to less extent than by ventilation and  
5 antioxidant status. Adjustment for these factors can modulate responses to approximate animal  
6 responses to those of humans. Unfortunately, these influential factors are rarely measured and,  
7 even less often, controlled. Increases in protein levels in BALF with O<sub>3</sub> exposures in guinea pigs  
8 are also a factor in the species' susceptibility to the effects of O<sub>3</sub>. Species comparisons of  
9 acute O<sub>3</sub> exposures to mice, guinea pigs, rats, hamsters, and rabbits found that guinea pigs were  
10 the most responsive (to  $\geq 0.2$  ppm); rabbits were the least responsive (2.0 ppm only); and rats,  
11 hamsters, and mice were intermediate (effects at  $\geq 1.0$  ppm). Rats and humans have subtle  
12 species-specific differences in inflammatory responses to O<sub>3</sub> in terms of the timing of PMN  
13 influx in the nasal and bronchoalveolar regions.

14 When humans are exposed to O<sub>3</sub> repeatedly for several consecutive days, lung function  
15 decrements subside, and normal spirometric parameters are regained (see Section 6.6). This  
16 phenomenon of functional attenuation also has been demonstrated in rats, not only in terms of  
17 spirometry, but also in terms of the classic tachypneic ventilatory response. Full or partial  
18 attenuation of some BAL parameters also appears to occur in both rats and humans, but exposure  
19 scenario appears to play a role; other cellular changes do not attenuate (see Section 6.9.4).  
20 Existing epidemiologic studies provide only suggestive evidence that persistent or progressive  
21 deterioration in lung function is associated with long-term oxidant-pollutant exposure (See  
22 Chapter 7). With chronic, repeated exposures to  $\geq 0.12$  ppm O<sub>3</sub>, however, laboratory animals  
23 demonstrate changes in lung structure, function, and biochemistry that are indicative of airway  
24 irritation and inflammation with the possible development of chronic lung disease (U.S.  
25 Environmental Protection Agency, 1996). Based on the apparent homology of these responses  
26 between humans and laboratory animals, animal studies appear to provide a means for assessing  
27 such chronic health concerns.

28 A species' susceptibility to the effects of O<sub>3</sub> exposure may be due, in part, to biochemical  
29 differences among species. Evidence for this is provided by differences in activity of SD rat and  
30 rhesus monkey CYP monooxygenases elicited by O<sub>3</sub> exposure (Lee et al., 1998). Additional  
31 characterization of species- and region-specific CYP enzymes will create a better understanding

1 of the differences in response to O<sub>3</sub>. This will allow more accurate extrapolation from animal  
2 exposures to human exposures and toxic effects.

3 Antioxidant metabolism varies widely among species, which can greatly influence the  
4 effects of O<sub>3</sub> (discussed in greater detail in 5.2.1.3). The guinea pig appears to be the species  
5 most susceptible to O<sub>3</sub>. Early studies ranked mice > rats > guinea pigs in order of antioxidant  
6 responsiveness to O<sub>3</sub> challenge. Guinea pigs have been shown to have lower basal levels of  
7 GSH transferase activity, lower activity of GSH peroxidases, and lower levels of vitamin E  
8 compared to rats. These lower levels of antioxidants combined with increases in protein levels  
9 in BALF (discussed above) with O<sub>3</sub> exposures likely explain, at least in part, the species'  
10 susceptibility to the effects of O<sub>3</sub>.

11 Because cytokine and chemokine responses are so important in an animal's defense  
12 against O<sub>3</sub> exposure, comparisons of differences in species expression and activity of these  
13 inflammatory mediators is necessary. Arsalane et al. (1995) compared guinea pig and human  
14 AM recovered in BALF and subsequently exposed in vitro to 0.1 to 1 ppm for 60 minutes.  
15 Measurement of inflammatory cytokines showed a peak at 0.4 ppm in both species. Guinea pig  
16 AM had an increase in IL-6 and TNF $\alpha$  while human AM had increases in TNF $\alpha$ , IL-1b, IL-6 and  
17 IL-8. This exposure also caused an increase in mRNA expression for TNF $\alpha$ , IL-1b, IL-6 and  
18 IL-8 in human cells. At 0.1 ppm exposures, only TNF $\alpha$  secretion was increased. These data  
19 suggest similar cytokine responses in guinea pigs and humans, both qualitatively and  
20 quantitatively.

21 Species differences in morphological responses to O<sub>3</sub> exposure have been characterized by  
22 Dormans et al. (1999), as discussed in previous sections. Dormans et al. (1999) continuously  
23 exposed rats, mice, and male guinea pigs to filtered air, 0.2, or 0.4 ppm O<sub>3</sub> for 3, 7, 28, and  
24 56 days. The animals exposed for 28 days were examined at 3, 7, or 28 days PE. Depending  
25 on the endpoint studied, the species varied in sensitivity. Greater sensitivity was shown in the  
26 mouse as determined by biochemical endpoints, persistence of bronchiolar epithelial  
27 hypertrophy, and recovery time. Guinea pigs were more sensitive in terms of the inflammatory  
28 response though all three species had increases in the inflammatory response after three days that  
29 did not decrease with exposure. These data on inflammation are in general agreement with  
30 Hatch et al., (1986), discussed above. In all species, the longest exposure to the highest O<sub>3</sub>  
31 concentration caused increased collagen in ductal septa and large lamellar bodies in Type II

1 cells, but that response also occurred in rats and guinea pigs at 0.2 ppm. No fibrosis was seen at  
2 the shorter exposure times and the authors question whether fibrosis occurs in healthy humans  
3 after continuous exposure. The authors do not rule out the possibility that some of these  
4 differences may be attributable to differences in total inhaled dose or dose actually reaching a  
5 target site. Overall, the authors rated mice as most susceptible, followed by guinea pigs and rats.

6 Comparisons of airway effects in rats, monkeys and ferrets resulting from exposures of  
7 1.0 ppm O<sub>3</sub> for 8 h (Sterner-Kock et al. 2000) demonstrated that monkeys and ferrets had similar  
8 inflammatory responses and epithelial necrosis. The response of these two species was more  
9 severe than that seen in rats. These data suggest that ferrets are a good animal model for O<sub>3</sub>-  
10 induced airway effects due to the similarities in pulmonary structure between primates and  
11 ferrets. However, the mechanisms of O<sub>3</sub> effects at these high concentrations may differ from  
12 those at more realistic levels.

13 A number of species, including nonhuman primates, dogs, cats, rabbits, and rodents, have  
14 been used to study the effects of O<sub>3</sub> exposure on airway bronchoconstriction. A commonly used  
15 model of bronchospasm utilizes guinea pigs acutely exposed to high O<sub>3</sub> concentrations (2 to  
16 3 ppm) to induce airway hyperreactivity (AHR). As mentioned earlier, the model is helpful for  
17 determining mechanistic aspects of AHR, but is not really relevant for extrapolation to potential  
18 airway responses in humans exposed to ambient levels of O<sub>3</sub>. Additionally, guinea pigs have  
19 been shown to have AHR in other studies that is very similar to asthmatic humans, but the utility  
20 of guinea pig data is somewhat limited by their disparity from other animal models.

21 The rat is a key species used in O<sub>3</sub> toxicological studies, but the rat has both behavioral  
22 and physiological mechanisms that can lower core temperature in response to acute exposures,  
23 thus limiting extrapolation of rat data to humans. Iwasaki et al. (1998) evaluated cardiovascular  
24 and thermoregulatory responses to O<sub>3</sub> at exposure of 0.1, 0.3, and 0.5 ppm O<sub>3</sub> 8 hrs/day for  
25 4 consecutive days. A dose-dependent disruption of HR and T<sub>co</sub> was seen on the first and second  
26 days of exposure, which then recovered to control values. Watkinson et al. (2003) exposed  
27 rats to 0.5 ppm O<sub>3</sub> and observed this hypothermic response, which included lowered HR,  
28 lowered T<sub>co</sub>, and increased inflammatory components in BALF. The authors suggested that the  
29 response is an inherent reflexive pattern that can possibly attenuate O<sub>3</sub> toxicity in rodents. They  
30 discuss the cascade of effects created by decreases in T<sub>co</sub>, which include: (1) lowered metabolic  
31 rate, (2) altered enzyme kinetics, (3) altered membrane function, (4) decreased oxygen

1 consumption and demand, (5) reductions in minute ventilation, which would act to limit the dose  
2 of O<sub>3</sub> delivered to the lungs. These effects are concurrent with changes in HR which lead to:  
3 (1) decreased CO, (2) lowered BP, and (3) decreased tissue perfusion, all of which may lead to  
4 functional deficits. The hypothermic response has not been observed in humans except at very  
5 high exposures, which complicates extrapolation of effects in rats to humans.

6 The importance of animal studies derives from their utilization in determining cause-effect  
7 relationships between exposure and health outcome, but the animal data must be integrated with  
8 epidemiological studies and controlled human clinical studies. Animal studies can corroborate  
9 both clinical and epidemiology studies and further provide important data that is impossible to  
10 collect in human studies. Toxic pulmonary and extrapulmonary effects following O<sub>3</sub> exposure  
11 have been well-studied in rodents, nonhuman primates, and a few other species; so,  
12 extrapolation, both qualitative and quantitative, to human exposures and consequent health  
13 effects is possible. Quantitative extrapolation, required to determine what specific exposure is  
14 likely to cause an effect in humans, is theoretically founded on the equivalency of mechanisms  
15 across species. At the molecular level, O<sub>3</sub> acts on the carbon-carbon double bond in  
16 polyunsaturated fatty acids and on sulfhydryl groups in proteins, both of which are found within  
17 cell membranes in animals and humans. At higher levels of cellular organization, cells affected  
18 in animals (e.g., AMs, Type 1 cells) have similar functions in humans, and organ systems (e.g.,  
19 respiratory system) have major interspecies similarities. However, interspecies differences do  
20 occur and complicate extrapolation.

21 Quantitative extrapolation, which involves a combination of dosimetry and species  
22 sensitivity, still requires more research before it can be fully realized. Knowledge of dosimetric  
23 animal-to-human extrapolation is more advanced than that of species-sensitivity, but  
24 extrapolation models have not been completely validated, and therefore, significant uncertainties  
25 remain. Mathematical modeling of O<sub>3</sub> deposition in the lower respiratory tract (i.e., from the  
26 trachea to alveoli) of several animal species and humans shows that the pattern of regional dose  
27 is similar, but that absolute values differ. In spite of structural and ventilatory differences  
28 between species, the greatest predicted tissue dose is to the CAR. Even though the CAR of rats  
29 has very rudimentary respiratory bronchioles, compared to well-developed ones in primates, the  
30 CAR of both rats and nonhuman primates respond similarly to O<sub>3</sub>.



1 Experimental measurement of delivered O<sub>3</sub> doses estimate that total respiratory uptake is  
2 ~47% in laboratory animals and ~87% in exercising humans, while nasopharyngeal removal is  
3 ~17% in rats and ~40% in humans. The previous O<sub>3</sub> AQCD (U.S. Environmental Protection  
4 Agency, 1996) provided the first quantitative animal-to-human extrapolation of morphological  
5 changes in the proximal alveolar region using rat and monkey studies. The extrapolation  
6 predicted that a 9-year-old child would have a 20% or 75% increase in PAR tissue thickness if  
7 their sensitivity to O<sub>3</sub> was equal to that of a rat or monkey, respectively. Adults would have  
8 15 or 70% increase, suggesting the potential for chronic effects in humans. In spite of the  
9 significant uncertainties, this extrapolation raises concern about the potential for chronic effects  
10 in humans

11 Experiments using 2 h exposures to 0.4 ppm <sup>18</sup>O<sub>3</sub> suggested that exercising (15 min  
12 intervals, rest and exercise at 60 L/min) humans received a 4- to 5-fold higher <sup>18</sup>O<sub>3</sub>  
13 concentrations in BAL than resting rats (Hatch et al., 1994). That level of exposure increased  
14 BAL protein and PMNs in humans, while a concentration of 2.0 ppm in rats was necessary for  
15 similar effects. Caveats in the interpretation of <sup>18</sup>O<sub>3</sub> studies include: (1) only a very small  
16 portion of the labeled compound is recoverable to assess incorporation; and (2) if species being  
17 compared differ in physiocochemical factors controlling mass transfer and downstream O<sub>3</sub>  
18 metabolism, it could cause significant differences in the amount of inhaled <sup>18</sup>O<sub>3</sub> that is detected  
19 during subsequent tissue analysis. Further, species differences in pulmonary anatomy,  
20 ventilation, antioxidants, and susceptibility all influence dose, repair processes, and tolerance to  
21 subsequent O<sub>3</sub> exposure. Important differences between exercising humans and resting rats that  
22 can affect tissue O<sub>3</sub> dose include: (1) increased ventilation and O<sub>3</sub> delivery with exercise;  
23 (2) decreased pulmonary ventilation and body temperature during O<sub>3</sub> exposure in rats;  
24 (3) diminished dose received in rats due to their burying their noses in their fur during exposure;  
25 and (4) increased concentration of antioxidants in ELF in rats compared to humans. These  
26 antioxidants are important for converting O<sub>3</sub> to inactive products before toxicity occurs (Kari  
27 et al., 1997; Gunnison and Hatch, 1999; Plopper et al., 1998), though this quenching is not  
28 quantitative. These and possibly other differences between rats and humans suggest that a  
29 2 ppm exposure in nonexercising rats approximates a 0.4 ppm exposure in exercising humans.  
30 Further comparisons of exercising human exposure to 0.1 ppm for 6 hours (Devlin et al., 1991)

1 and resting rat exposure to 0.3 ppm show inflammatory and permeability changes in humans but  
2 not rats.

### 3 4 **4.3.1 Summary and Conclusions: Species Homology, Sensitivity, and** 5 **Animal-to-Human Extrapolation**

6 Comparisons of acute exposures in rats and humans suggest that, though both species have  
7 similar qualitative responses to O<sub>3</sub> exposure, there are interspecies mechanistic disparities that  
8 necessitate careful comparisons of dose-response relationships. There is no perfect nonhuman  
9 species with which to model O<sub>3</sub> toxicity. All have limitations that must be considered when  
10 attempting to extrapolate to human exposures. Awareness of these limitations, even at the level  
11 of subtle strain differences within a test species, is extremely important. The currently available  
12 data suggest that LOELs in resting rats are approximately 4- to 5-fold higher than for exercising  
13 humans for toxicological endpoints including BAL protein and BAL PMNs. Studies comparing  
14 species-specific differences in O<sub>3</sub>-induced effects showed that guinea pigs were the most  
15 susceptible, rabbits the least susceptible, and rodents intermediate in susceptibility. The recent  
16 work being done utilizing various mouse strains with differing sensitivities to O<sub>3</sub> will help us to  
17 understand the extremely complex inter-individual differences in human sensitivity to O<sub>3</sub>.

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1                   **5. TOXICOLOGICAL EFFECTS OF OZONE AND**  
2                   **RELATED PHOTOCHEMICAL OXIDANTS IN**  
3                   **LABORATORY ANIMALS AND IN VITRO**  
4                   **TEST SYSTEMS**

5  
6  
7                   **5.1 INTRODUCTION**

8                   A wide range of effects of ozone (O<sub>3</sub>) has been demonstrated in laboratory animals. The  
9                   major research findings are that environmentally relevant levels of O<sub>3</sub> cause lung inflammation;  
10                  decreases in host defenses against infectious lung disease; acute changes in lung function,  
11                  structure, and metabolism; chronic lung disease, some elements of which are irreversible; and  
12                  systemic effects on target organs (e.g., brain, heart, liver, immune system) distant from the lung.  
13                  The research also has served to expand the understanding of mechanisms of O<sub>3</sub> toxicity and the  
14                  relationships between concentration and duration of exposure.

15                 The framework for presenting the health effects of O<sub>3</sub> in animals begins with a presentation  
16                 of respiratory tract effects, followed by systemic effects, and then interactions of O<sub>3</sub> with other  
17                 common co-occurring pollutants. The information discussed in this chapter is founded on a very  
18                 wide body of literature on studies in laboratory animals and on in vitro test systems of animal  
19                 cell lines and organ systems that may mimic responses in intact animals. The direct effects of O<sub>3</sub>  
20                 in humans are discussed in the following chapter (Chapter 6).

21                 This chapter is not intended to be a compendium of all that is known about O<sub>3</sub>; rather, it is  
22                 an update of the toxicology chapter from the last O<sub>3</sub> criteria document (U.S. Environmental  
23                 Protection Agency, 1996), or 1996 O<sub>3</sub> CD, and other reviews of the earlier published literature.  
24                 The historical O<sub>3</sub> literature is very briefly summarized in an opening paragraph of each section  
25                 or subsection. This paragraph is intended as a very concise overview of previous work, and the  
26                 reader is referred to the 1996 O<sub>3</sub> CD for more detailed discussion of the literature prior to the  
27                 early 1990's. Each section then continues with brief discussions of the key new studies (or  
28                 somewhat older studies that were not included in the previous CD). Longer discussions of new  
29                 studies are included where warranted. Sections are ended with comparisons of data from the  
30                 previous CD with new data and basic conclusions are drawn. Summaries of new studies and  
31                 results are provided in tables in Annex AX5.

1 Except for nitrogen dioxide (NO<sub>2</sub>), the subject of another criteria document (U.S.  
2 Environmental Protection Agency, 1993), there is very little relevant information on other  
3 photochemical oxidants in the published literature. What is known about the effects of these  
4 other oxidants is also summarized briefly in this chapter.  
5  
6

## 7 **5.2 RESPIRATORY TRACT EFFECTS OF OZONE**

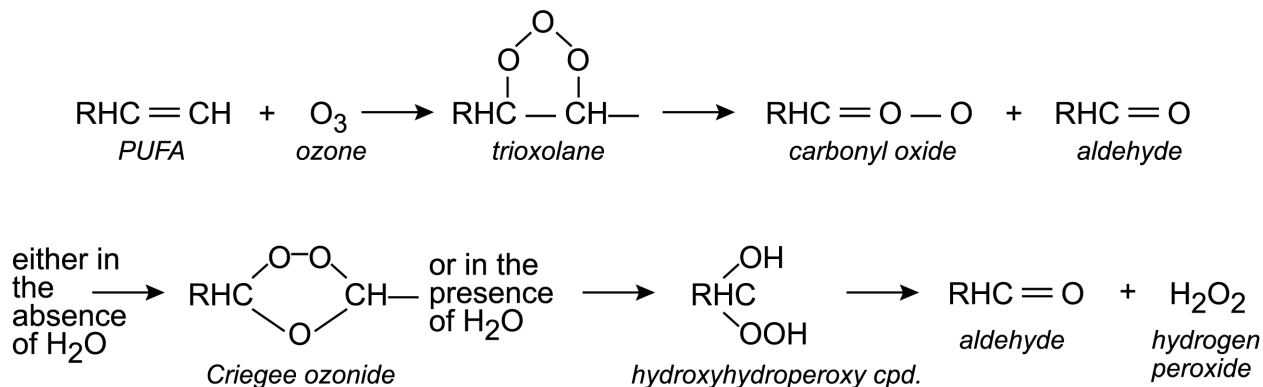
### 8 **5.2.1 Biochemical Effects**

9 Biochemically detected effects of O<sub>3</sub> are integrally involved in effects on both structure  
10 and function (respiratory and nonrespiratory) of the respiratory tract. Changes in xenobiotic  
11 metabolism, antioxidant metabolism and oxygen consumption, lipids and arachidonic acid  
12 metabolism, and collagen metabolism are all observed with O<sub>3</sub> exposure, though the mechanisms  
13 and associations are not fully understood.  
14

#### 15 **5.2.1.1 Cellular Targets of O<sub>3</sub> Interaction**

16 Ozone has the potential to interact with a wide range of different cellular components that  
17 include polyunsaturated fatty acids (PUFAs); some protein amino acid residues; and some  
18 low-molecular-weight compounds that include glutathione (GSH), urate, vitamins C and E, and  
19 free amino acids. Early work demonstrated that O<sub>3</sub>, being a highly reactive compound, does not  
20 penetrate much beyond the epithelial lining fluid (ELF). Reaction/diffusion analyses suggest  
21 that O<sub>3</sub>, at environmentally-relevant concentrations, diffuses no more than 0.1 to 0.2 μm into the  
22 ELF. Ozone-induced cell damage most likely results from its reactions with PUFAs to form  
23 stable but less reactive ozonide, aldehyde, and hydroperoxide reaction products. These reaction  
24 products (Crige ozonides and hydroxyhydroperoxides) may act as signal transduction  
25 molecules involved in signaling of cellular responses such as inflammation, and thus mediate O<sub>3</sub>  
26 toxicity. These reactions are summarized in Figure 5-1 and studies published since the 1996  
27 AQCD are listed in Table AX5-1.

28 Frampton et al. (1999) demonstrated the ozonation of PUFA to form nonanal and hexanal  
29 in rat BAL after exposures to 0.22 ppm O<sub>3</sub> for 4 h with exercise. Increases in nonanal were not  
30 accompanied by significant changes in lung function, in epithelial permeability, or in airway  
31 inflammation. Hexanal levels did not increase significantly and levels of both aldehydes



**Figure 5-1. Major secondary products of ozone interaction with epithelial lining fluid and lung cells.**

1 returned to baseline by 18 h PE. Pryor et al. (1996) exposed rats to 0.5 to 10 ppm O<sub>3</sub> both with  
 2 and without 5% CO<sub>2</sub> to measure the amount of aldehyde generated in BAL, and also the rate of  
 3 disappearance of aldehydes from the ELF following the O<sub>3</sub> exposure. Ozone exposure with CO<sub>2</sub>  
 4 increased the tidal volume and the yield of aldehydes with a maximal aldehyde yield at 2.5 ppm  
 5 for 1 h. Absolute yields were impossible to ascertain in this system because deposition of O<sub>3</sub> is  
 6 unknown and aldehyde recovery is not complete due to loss of aldehyde by volatilization and by  
 7 diffusion into underlying tissue. The data showed that at 0.5 ppm O<sub>3</sub> with 5% CO<sub>2</sub>, levels of  
 8 hexanal and nonanal increased at 30 minutes, decreased slightly from that level at 60 minutes,  
 9 was maximal at 90 minutes and then dropped to 60 minutes levels at 120 minutes. Levels of  
 10 heptanal did not change appreciably during this time course. Levels of these aldehydes were  
 11 dependent on a dynamic relationship between their production and the disappearance from the  
 12 ELF. The authors stated that O<sub>3</sub> is the limiting reagent in this process because the amount of  
 13 PUFA far exceeds the amount of O<sub>3</sub> on a molar basis. Because of the limitations of measuring  
 14 aldehydes in this study paradigm, it is not useful for quantitative dosimetry; however, the authors  
 15 suggest the study does serve to demonstrate the use of aldehydes as biomarkers of O<sub>3</sub> exposure  
 16 since nonanol is produced in an O<sub>3</sub>-specific pathway.

17 Postlethwait et al. (1998) utilized three biologically relevant models (isolated epithelial  
 18 lining fluid, intact lung, and liposome suspensions) to determine the O<sub>3</sub>-induced production of  
 19 heptanal, nonanal and hexanal in an attempt to estimate formation of lipid-derived bioactive

1 compounds. Exposures used were 0.25 to 1.0 ppm for 30 to 60 minutes. Data suggest that  
2 PUFAs directly react with O<sub>3</sub> and the amount of bioactive lipids produced is inversely related to  
3 ascorbic acid availability. The authors caution that there are limitations to the use of  
4 measurements of these reactions products in determining O<sub>3</sub> dose-response relationships due to  
5 the heterogenous nature of O<sub>3</sub> reactions in the epithelial lining fluid. Connor et al. (2004) have  
6 recently examined the reactive absorption of O<sub>3</sub> (0.3 to 1.1 ppm for 1 to 2 h) within ELF using  
7 interfacial films composed of dipalmitoylglycero-3-phosphocholine (DPPC) and rat lung lavage  
8 fluid. The films reduced O<sub>3</sub> reactive absorption by antioxidants. Further experiments using a  
9 human lung fibroblast cell line exposed to O<sub>3</sub> demonstrated that ascorbic acid (AA) produced  
10 cell injury that high levels of O<sub>3</sub> and AA were needed to induce cell injury, and the DPPC films  
11 reduced the amount of cell injury. From these data the authors suggest that O<sub>3</sub> reactions with  
12 ELF substrates cause cell injury that films of active, saturated phospholipids reduce the local  
13 dose of O<sub>3</sub>-derived reaction products, and that these interfacial phospholipids modulate the  
14 distribution of inhaled O<sub>3</sub> and the extent of site-specific cell injury.

15 Recent studies have examined the formation of ozonation products such as  
16 4-hydroxynonenal (HNE), a toxic aldehyde that reacts with cysteine, histamine, and lysine  
17 amino acid residues and creates protein adducts. Hamilton et al. (1998) demonstrated (see  
18 Chapter 6) using human AM exposed to 0.4 ppm O<sub>3</sub> for 1 h that exposure caused apoptosis, an  
19 increase in a 32-kDa protein adduct, and an increase in ferritin and a 72-kDa heat shock protein.  
20 By exposing AM to HNE in vitro, all of these effects are replicated, which the authors interpret  
21 to mean that creation of protein adducts and apoptotic cell death are cellular toxic effect of acute  
22 O<sub>3</sub> exposure and that it is mediated, at least in part by HNE.

23 These recent reports combined with observations reported in the previous O<sub>3</sub> CD (US  
24 Environmental Protection Agency, 1996) suggest that interactions of O<sub>3</sub> with cellular  
25 components and ELF generate toxic ozonation products and mediate toxic effects through these  
26 products.

### 27 28 **5.2.1.2 Monooxygenases**

29 Both short- and long-term exposures to O<sub>3</sub> have been shown to enhance lung xenobiotic  
30 metabolism, possibly as a result of changes in the number and function of bronchiolar epithelial  
31 Clara cells and alveolar epithelial Type 2 cells. Studies of the effects of O<sub>3</sub> on lung

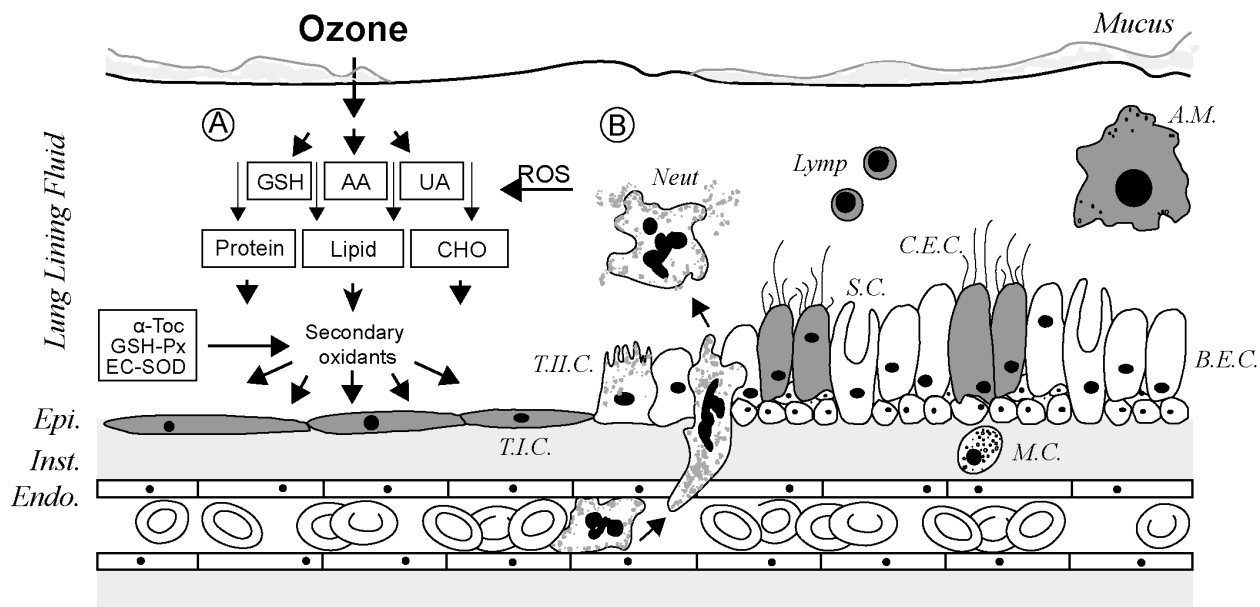


1 monooxygenases are listed in Table AX5-2. Early studies showed that exposure to O<sub>3</sub> increased  
2 CYP 2B1 (the major CYP isoform in rat lung) content and activity in rat lung. Ozone exposures  
3 also caused hypertrophy and hyperplasia of CYP 2B1-immunoreactive Clara cells. Comparisons  
4 of rat and rhesus monkey CYP isoforms demonstrated species-specific and region-specific (e.g.,  
5 trachea, parenchyma) differences in the activities of P450 isoforms (Lee et al., 1998)

6 Watt et al. (1998) found that 1 ppm O<sub>3</sub> in both acute (8h, 1 ppm) and chronic (90 days,  
7 1 ppm) exposures in rat increased CYP 2E1 in a region-specific manner. Paige et al. (2000a)  
8 showed that a long term exposure (0.8 ppm, 8h/day for 90 days) increased the activity of CYP  
9 2B in distal lung but not trachea or intrapulmonary airways. Studies have focused on P450 gene  
10 expression to examine possible genetic mechanisms that may explain differential O<sub>3</sub>-sensitivity  
11 (Mango et al., 1998). Mice (129 strain) deficient in Clara cell secretory protein (CCSP<sup>-/-</sup>),  
12 which are oxidant-sensitive, were exposed to 1 ppm O<sub>3</sub> for 2 hours. The CCSP null mice  
13 demonstrated increases in IL-6 and metallothionein (Mt) mRNA that preceded decreases in  
14 Clara cell CYP 2F2 mRNA (normally expressed at high levels in mouse lung) levels. In 129  
15 strain wildtype (WT) mice, RNA levels changed similarly, to a lesser degree. These data  
16 suggest a protective role against oxidant damage for CCSP, and further, that genetic  
17 susceptibility to oxidant stress may be mediated, in part, by the gene coding for CCSP.

### 18 19 **5.2.1.3 Antioxidants, Antioxidant Metabolism, and Mitochondrial Oxygen Consumption**

20 Ozone also undergoes reactions with ascorbic acid (AA), reduced glutathione (GSH), and  
21 uric acid (UA), all antioxidants present in ELF (see Figure 5-2, A). In vivo experiments have  
22 shown that reactions with O<sub>3</sub> occur preferentially with antioxidants compared to proteins and  
23 lipids also present in ELF. This is a protective interaction, but even with environmentally  
24 relevant exposures to O<sub>3</sub>, the reactivity of O<sub>3</sub> is not quantitatively quenched. Antioxidants offer  
25 some protection from O<sub>3</sub> exposure but often are not maintained at concentrations sufficient to  
26 fully protect the lung. Thus, O<sub>3</sub>-induced cell injury occurs in both the lower and upper  
27 respiratory tract. Early work has shown that acute (1 week) exposures to <1 ppm O<sub>3</sub> increase  
28 antioxidant metabolism, including levels of cytosolic enzymes glucose-6-phosphate  
29 dehydrogenase (G6PD), 6-phosphogluconate dehydrogenase (6PGD), glutathione reductase  
30 (GR), and glutathione peroxidase (GSHPx). Re-exposure after a recovery period causes  
31 increases equivalent to first-time exposures, thus previous exposure appears to not be protective.



**Figure 5-2. (Reprinted from Molecular Aspects of Medicine, I.S. Mudway and F.J. Kelly, Ozone and the Lung: a sensitive issue, page 36, (2000), with permission from Elsevier).**

1           Increases in enzyme activity appear to increase as a function of age, suggesting that O<sub>3</sub>  
2 exposure can cause greater lung injury in the older animal. This has been attributed to  
3 differences in dose reaching lung target sites, differing base levels of antioxidants and  
4 antioxidant enzymes, and differences in cellular sensitivity. Species differences exist in  
5 antioxidant metabolism, with guinea pigs being very sensitive to O<sub>3</sub> due to their diminished  
6 increases in antioxidants and antioxidant enzymes. Chronic exposures of rats to urban patterns  
7 of O<sub>3</sub> (daily peaks of 0.25 ppm) caused increases in GSHPx and GR, but not superoxide  
8 dismutase (SOD). The enzyme changes could be accounted for by differences in the steady-state  
9 cell population or in cellular antioxidant capacity. More recent studies examining antioxidants  
10 and O<sub>3</sub> exposure are listed in Table AX5-3.

11           Ozone induced both site- and cell-specific changes in copper-zinc (Cu-Zn) and manganese  
12 (Mn) SOD in rats exposed to 1.0 ppm O<sub>3</sub> for up to 3 months (Weller et al., 1997). Cu-Zn SOD  
13 labeling was decreased in epithelial cells in airways and parenchyma. Mn SOD labeling was  
14 increased in both AM and epithelial type II cells of the centriacinar region (CAR), which the

1 authors suggest may allow these cells to tolerate further O<sub>3</sub> exposure. This work is in agreement  
2 with earlier work suggesting a role of SOD in protection of cells against oxidative stress.

3 Freed et al. (1999) evaluated the role of antioxidants in O<sub>3</sub>-induced oxidant stress in dogs  
4 (exposed to 0.2 ppm in a 6 h exposure) by inhibiting the antioxidant transport using probenecid  
5 (an anion-transport inhibitor). Blocking antioxidant transport caused heterogeneously  
6 distributed increases in peripheral airway resistance and reactivity, supporting the hypothesis  
7 that in the lung periphery, endogenous antioxidants moderate the effects of O<sub>3</sub> and that this  
8 exposure is a subthreshold stimulus for producing effects on peripheral airway resistance and  
9 reactivity in dogs. The authors further found that treatment with probenecid also inhibited O<sub>3</sub>-  
10 induced neutrophilic inflammation, providing evidence for a dissociation between airway  
11 function and inflammation. This suggests that O<sub>3</sub>-induced inflammation and airway  
12 hyperreactivity (AHR) are independent phenomena operating through multiple mechanistic  
13 pathways.

14 Mudway and Kelly (1998) modeled the interactions of O<sub>3</sub> with ELF antioxidants using a  
15 continually mixed, interfacial exposure set up with O<sub>3</sub> concentrations of ranging from 0.1 to  
16 1.5 ppm for durations ranging from 30 to 720 min. Uric acid was ranked the most O<sub>3</sub>-reactive,  
17 AA the second most reactive, and GSH the least reactive. Thus, they concluded that GSH is not  
18 an important substrate for O<sub>3</sub>, while UA appeared to be the most important reactive substrate,  
19 which confers protection from O<sub>3</sub> by removing it from inhaled air and limiting the amount that  
20 reaches the distal lung. By providing a substrate for O<sub>3</sub> reactions in the ELF, UA effectively  
21 reduces the diffusive resistance of O<sub>3</sub> (see Bush et al., 2001) in the TB airways and thus may  
22 serve to limit the amount of O<sub>3</sub> reaching the distal lung. The authors acknowledge limitations in  
23 extrapolating these data to in vivo O<sub>3</sub> exposures due to the absence of surfactant lipids and  
24 airway mucus in the model.

#### 25 26 **5.2.1.4 Lipid Metabolism and Content of the Lung**

27 One of the major postulated molecular mechanisms of action of O<sub>3</sub> is peroxidation of  
28 mono- and polyunsaturated fatty acids and unsaturated neutral lipids in the lung. Because all of  
29 these lipids appear both in cell membranes and as secretions in the ELF, it is difficult to ascertain  
30 which lipid pool contributes to the formation of lipid ozonation products. As mentioned, O<sub>3</sub> can  
31 penetrate only about 0.1 to 0.2 μm into the ELF, so it is unlikely that O<sub>3</sub> reacts directly with

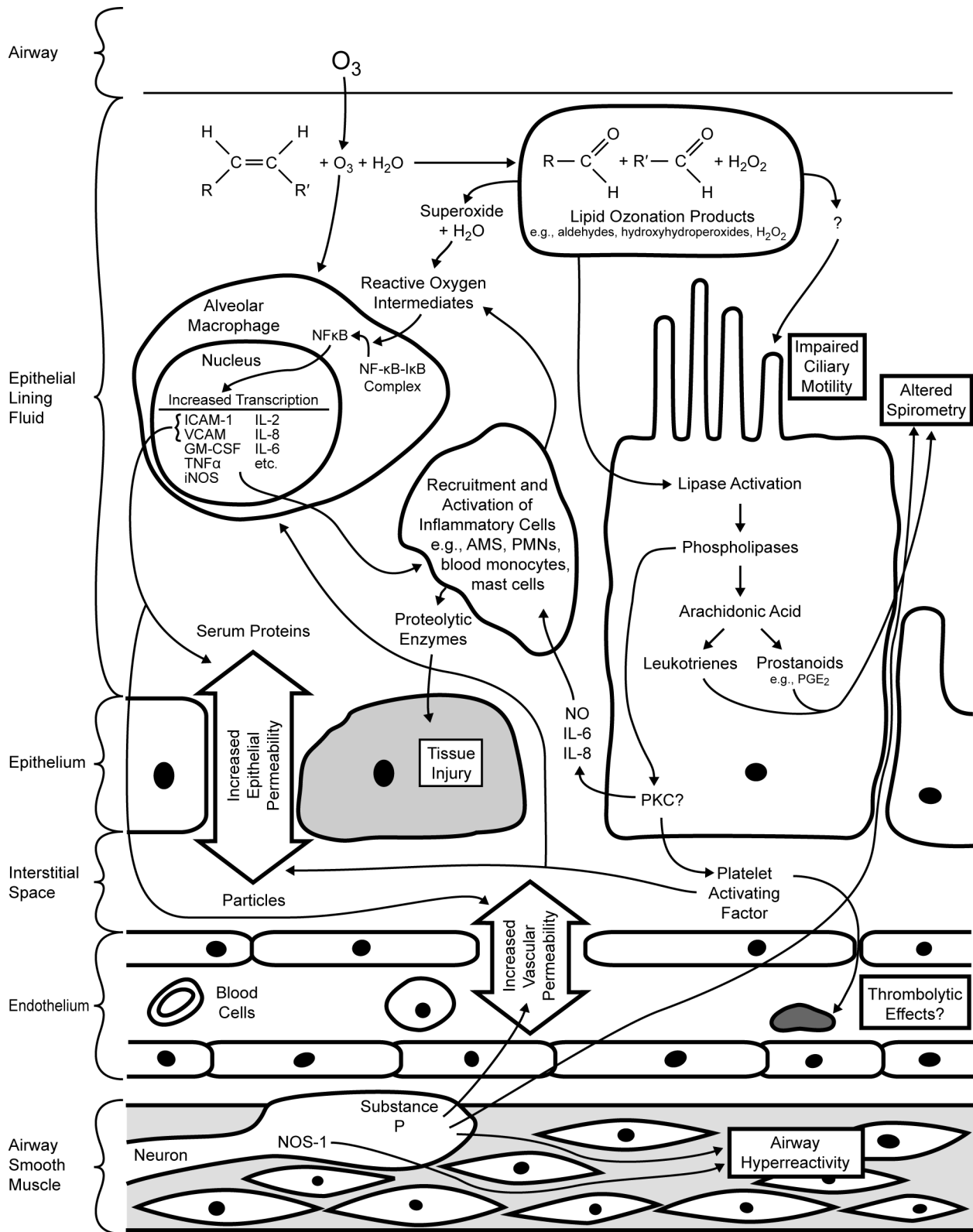
1 epithelial cell membranes, except in regions of distal lung where ELF is very thin or absent. The  
2 inflammatory cascade (shown in Figure 5-3) initiated by O<sub>3</sub> generates a mix of secondary  
3 reactants (e.g., aldehydes) which then are likely to oxidize lipids and proteins in cell membranes.

4 In both acute and short-term studies, a variety of lung lipid changes occur, including an  
5 increase in arachidonic acid. Metabolism of arachidonic acid produces a variety of biologically  
6 active mediators that can, in turn, affect host defenses, lung function, the immune system, and  
7 other functions. The protein A component of surfactant is also a primary target of O<sub>3</sub> interaction.  
8 During the first few days of O<sub>3</sub> exposure, the changes in lung lipid biosynthesis can be accounted  
9 for by the alveolar epithelial proliferative repair. With longer exposures (e.g., 0.12 ppm for  
10 90 days) an increase in PUFAs and a decrease in cholesterol-esters are seen, indicative of  
11 long-term alterations of surfactant lipid composition.

12 Several new studies listed in Table AX5-4 examined the effects of O<sub>3</sub> exposure on  
13 phospholipids in lung tissue. Ozonation of PUFAs has been shown to generate other aldehydes  
14 such as nonanal and hexanal in rat (Pryor et al., 1996; Frampton et al., 1999). These aldehydes  
15 are short-lived and found to not affect lung function (Frampton et al., 1999). These observations  
16 suggest that levels of these aldehydes are dependent on a dynamic relationship between their  
17 production and their disappearance from the ELF.

18 Pryor et al. (1995) proposed a cascade mechanism whereby ozonation products cause  
19 activation of specific lipases, which then trigger the activation of second messenger pathways  
20 (e.g., phospholipase A<sub>2</sub> or phospholipase C). This group (Kafoury et al., 1999) showed that  
21 exposure of cultured human bronchial epithelial cells to the lipid ozonation product 1-palmitoyl-  
22 2-(9-oxononanoyl)-sn-glycero-3-phosphocholine elicited release of platelet-activating factor  
23 (PAF) and prostaglandin E<sub>2</sub>, but not IL-6. The lipid ozonation product 1-hydroxy-1-  
24 hydroperoxynonane caused release of PAF and IL-6 in these cells, but not prostaglandin E<sub>2</sub>.  
25 These results suggest to the authors that O<sub>3</sub>-induced production of lipid ozonation products  
26 causes release of proinflammatory mediators that then generate an early inflammatory response.

27 Very new work (Ballinger et al., 2005) has shown that ozone-induced membrane oxidation  
28 is augmented by antioxidants present in ELF. They utilized a red cell membrane model exposed  
29 to 0.8 ppm O<sub>3</sub> for 30 min. The monolayer of cells was intermittently covered by an aqueous film  
30 consisting of rat BALF or BALF plus added antioxidants. AA and GSH induced dose-dependent



Adapted from: Pryor et al. (1995); Krishna et al. (1998); Bhalla et al. (1999)

**Figure 5-3. Mechanisms of ozone toxicity.**

1 oxidative damage to the cell membrane proteins and lipids via secondary oxidant formation.  
2 The authors concluded that early in O<sub>3</sub> exposure, ELF antioxidants are high enough to drive  
3 reactive absorption of O<sub>3</sub> into the ELF and to concurrently quench secondary reaction products,  
4 thus limiting cell injury. With continued exposure, antioxidants levels decrease such that  
5 unreacted O<sub>3</sub> and cytotoxic products can diffuse to the cell membranes, causing injury.  
6 Limitations of this in vitro study are the possible differences in chemical species and  
7 mechanisms compared to in vivo systems.

8 Uhlson et al. (2002) reacted O<sub>3</sub> with calf lung surfactant which resulted in the production  
9 of 1-palmitoyl-2-(9'-oxo-nonanoyl)-glycerophosphocholine (16:0a/9-al-GPCho). The biological  
10 activity of this oxidized phospholipid included: (1) decreased macrophage viability,  
11 (2) induction of apoptosis in pulmonary epithelial-like A549 cells, (3) and release of IL-8 from  
12 A549 cells. Exposure levels of 0.125 ppm O<sub>3</sub> for 2–4 h in this system were capable of  
13 generating biologically active phospholipids that were capable of mediating toxic effects of O<sub>3</sub>.

14 In addition to PUFA, cholesterol, the most abundant neutral lipid present in ELF, is also a  
15 target of O<sub>3</sub>. Pulfer and Murphy (2004) demonstrated the ozonolysis of cholesterol in an in vitro  
16 system using BALF isolated from rats that had been exposed to 2.0 ppm O<sub>3</sub> for 4 h. Production  
17 of 5-hydroperoxy-*B*-homo-6-oxa-cholestan-3,7a-diol, 5β,6β-epoxycholesterol, and 3β-hydroxy-  
18 5oxo-5,6-seco-cholestan-6-al was shown. Additionally, both 5β,6β-epoxycholesterol and its  
19 most abundant metabolite, cholestan-6-oxo-3β,5α-diol, were demonstrated to be cytotoxic to  
20 16-HBE cells and to inhibit cholesterol synthesis. Studies (Pulfer et al., 2005) in C57BL/6J mice  
21 exposed to 0.5, 1.0, 2.0 or 3.0 ppm O<sub>3</sub> for 3 h demonstrated that these oxysterols were produced  
22 in vivo also. The authors suggest that this may be an additional mechanism of O<sub>3</sub> toxicity.  
23 Though these oxysterol reaction products have not been fully characterized, they may be  
24 involved in O<sub>3</sub>-induced inflammation by disrupting cellular membranes or altering signaling  
25 between cells. Similar oxysterols have been implicated in the inflammatory cascade associated  
26 with atherosclerosis.

27 Thus, new work has attempted to elucidate the mechanisms by which reactions of O<sub>3</sub> with  
28 lipids create phospholipids that then mediate downstream toxic effects. It is uncertain whether  
29 these described changes in lipid content and/or metabolism lead to significant changes in surface  
30 tension or compliance properties of the lung.

31

### 5.2.1.5 Ozone Interactions with Proteins and Effects on Protein Synthesis

Epithelial lining fluid contains proteins arising from airway secretions and from blood. Ozone can react with four amino acid residues (cysteine, histidine, methionine, and tryptophan) and can cause oxidation of functional groups on proteins, including aldehydes, alcohol, amines and sulfhydryls. A number of enzymes have been shown to be inhibited by O<sub>3</sub> including cholinesterase,  $\alpha$ 1-antiproteinase, and prostaglandin synthetase. Additionally, O<sub>3</sub> decreases the inhibitory activity of  $\alpha$ 1-proteinase inhibitor, which is implicated in development of emphysema. Surfactant protein A (SP-A) is a target for O<sub>3</sub> toxicity by modulation of SP-A self association, vesicle aggregation, phospholipid secretion, and stimulation of AM superoxide anion generation (see Section 5.2.2.3). Further, O<sub>3</sub> is thought to interfere in SP-A's homeostatic role in surfactant release from alveolar Type 2 cell lamellar bodies and its subsequent uptake by Type 2 cells and AMs.

Lung collagen, collagen synthesis, and prolyl hydroxylase activity associated with fibrogenesis have been shown to increase in rodents with O<sub>3</sub> exposure of  $\geq 0.45$  ppm. Some studies have shown that this increase persists after exposure stops and that there is an influence of exposure pattern on the response. The increased collagen has been correlated with structural changes in the lung. Rats exposed to an urban pattern of O<sub>3</sub> with daily peaks of 0.25 ppm for 38 weeks displayed extracellular matrix thickening. Increased levels of collagen in CAR were demonstrated in female rats exposed to 0.5 to 1.0 ppm O<sub>3</sub> for 6 h/day for 20 months and in monkeys exposed to 0.61 ppm for 1 year. Both increased age and health status (e.g., emphysemic) were implicated in the increased collagen formation in response to O<sub>3</sub> exposure.

A time-course study (van Bree et al., 2001 Table AX5-5) evaluating the lung injury and changes in collagen content in rats exposed acutely or subchronically to 0.4 ppm O<sub>3</sub> demonstrated CAR thickening of septa which progressed from 7 through 56 days of exposure. Though collagen content decreased with PE recovery, the structural fibrotic changes in ductular septa and respiratory bronchioles persisted, suggesting that subchronic O<sub>3</sub> exposures in rats creates a progression of structural lung injury that can evolve to a more chronic form, which included fibrosis. The biological relevance and adverse health effects of altered protein synthesis and collagen accumulation are uncertain.

### 1 **5.2.1.6 Differential Gene Expression**

2 Gohil et al. (2003) examined differential gene expression in C57BL/6 mice exposed to  
3 1 ppm O<sub>3</sub> for three consecutive nights for 8 hours (see Table AX5-6). Ozone exposure induced  
4 changes in expression of 260 genes ( 80% repressed and 20% induced). Differentially expressed  
5 genes included those involved in progression of the cell cycle such as *S*-adenosyl methionine  
6 decarboxylase 3 (SAMDC3), ribonucleotide reductase (RR), and clusterin. Increased transcription  
7 of these genes suggests O<sub>3</sub>-induced activation of the cell cycle with subsequent cellular  
8 proliferation. This is in accord with the finding of increased epithelial proliferation with acute  
9 O<sub>3</sub> exposure as discussed in studies in Sections 5.2.4.1 and 5.2.5.1. Several NF-κB-induced  
10 genes were upregulated, included serum amyloid protein, topoisomerase IIα, monocyte  
11 chemoattractant protein, platelet-derived growth factor, and inhibitor of apoptosis. Upregulation  
12 of these genes suggests to the authors that they may account for O<sub>3</sub>-induced proliferation of  
13 nonciliated cells and Clara cells. Downregulation of transcripts for isoforms of myosins and  
14 actins were also observed, which may explain, in part, a mechanism of O<sub>3</sub>-induced vascular  
15 permeability. Several members of the CYP family were downregulated, including 2a4, and 2e1,  
16 and 2f2, as were aryl-hydrocarbon receptor and several glutathione transferases. Metallothionein  
17 1 and 2 and lactotransferrin were upregulated, indicative of their function as antioxidants and  
18 anti-inflammatory agents. Ozone-induced suppression of immune function is suggested by  
19 downregulation of transcripts encoding major histocompatibility complex genes,  
20 lymphocyte-specific proteins, and immunoglobulins. Section 5.2.2.3 discusses the effects of O<sub>3</sub>  
21 exposure on the immune system.

22 Quinlan et al. (1994) have reviewed the regulation of antioxidant enzymes in lung after  
23 oxidant injury. A comparison of alterations in gene expression in rat following O<sub>3</sub> or hyperoxia  
24 exposure, both of which induce reactive oxygen species and injury to vascular endothelial cells  
25 and cells of the alveoli, show that both ~1 ppm O<sub>3</sub> and 85-95% O<sub>2</sub> increase expression of  
26 CuZnSOD, glutathione peroxidase, and catalase. Studies in mice (Johnston et al, 1998) also  
27 demonstrate that changes in gene expression indicative of inflammation and epithelial injury that  
28 occur with hyperoxia in mice (95% O<sub>2</sub>) compare to similar injury that occurs following O<sub>3</sub>  
29 exposure.



### 5.2.1.7 Summary and Conclusions - Biochemical Effects

Ozone has been shown to interact with a wide range of different cellular components including PUFAs, amino acid residues, and some low-molecular-weight compounds (GSH, urate, vitamins C and E). As O<sub>3</sub> does not penetrate much beyond the ELF, damage likely results from its PUFA ozonation products (mostly hydroxyhydroperoxides) involvement in signaling of cellular responses such as inflammation. New work has shown that ozonation of PUFA also forms the aldehydes nonanal, heptanal, and hexanal, the production of which is dependent on AA availability. Saturated phospholipids are thought to reduce the local dose and limit site-specific cell injury from O<sub>3</sub> exposure. Another ozonation product HNE creates protein adducts that have been linked to apoptosis and heat shock proteins in vitro.

Both short- and long-term exposures to O<sub>3</sub> have been shown to enhance lung xenobiotic metabolism, possibly as a result of changes in the number and function of bronchiolar epithelial Clara cells and alveolar epithelial Type 2 cells. This modulation is both species- and region-specific and includes the isoforms CYP 2B1, CYP 2E1. CCSP is also involved in inflammatory responses to O<sub>3</sub> exposure. Mice strains with differing sensitivities to O<sub>3</sub> show that responses in protein, LDH and inflammatory cell influx are due to CCSP levels and changes in lung epithelial permeability.

Reactions of O<sub>3</sub> with AA, GSH, and UA (all antioxidants present in ELF) are a protective mechanism. But even with environmentally relevant exposures, the reactivity of O<sub>3</sub> is not quantitatively quenched and cell injury occurs in both the lower and upper respiratory tract. Early work has shown that short-term exposures to <1 ppm O<sub>3</sub> increase antioxidant metabolism. Re-exposure after a recovery period causes increases equivalent to first-time exposures, suggesting that previous exposure is not protective. Increases in enzyme activity appear to increase as a function of age, suggesting that O<sub>3</sub> exposure can cause greater lung injury in the older animal. Long-term urban patterns of exposure to O<sub>3</sub> (daily peaks of 0.25 ppm) caused increases in GSHPx and GR, but not SOD. Recent work has suggested that endogenous antioxidants moderate the effects of O<sub>3</sub> and that this exposure is a subthreshold stimulus for producing effects on peripheral airway resistance and reactivity, thus indicating a dissociation between airway function and inflammation.

In both acute and short-term studies, a variety of lung lipid changes occur with O<sub>3</sub> exposure, including an increase in AA. With longer exposures (e.g., 0.12 ppm for 90 days),

1 an increase in PUFAs and a decrease in cholesterol-esters are seen, indicative of long-term  
2 alterations of surfactant lipid composition. Whether these changes in lipid content and/or  
3 metabolism lead to significant changes in surface tension or compliance properties of the lung  
4 remains unknown. New studies evaluating O<sub>3</sub>-induced alterations in lipid metabolism have not  
5 been completed.

6 Collagen, a structural protein involved in fibrosis, increases with O<sub>3</sub> exposure, and some  
7 studies have shown that this increase persists after exposure stops. Urban patterns of exposure  
8 (daily peaks of 0.25 ppm for 38 weeks) created extracellular matrix thickening. Increases in  
9 centriacinar collagen were demonstrated in female rats exposed to 0.5 to 1.0 ppm O<sub>3</sub> for 6 h/day  
10 for 20 months and in monkeys exposed to 0.61 ppm for 1 year. New work examining the time  
11 course of lung injury and changes in collagen content in rats exposed acutely or subchronically  
12 to 0.4 ppm O<sub>3</sub> showed centriacinar thickening of septa. Collagen content decreased with PE  
13 recovery but not the structural fibrotic changes in ductular septa and respiratory bronchioles,  
14 which suggests that subchronic O<sub>3</sub> exposures in rats creates a progression of structural lung  
15 injury that can evolve to a more chronic form, which includes fibrosis.

## 16 17 **5.2.2 Lung Host Defenses**

18 Defense mechanisms, including the mucociliary clearance system, AMs, and humoral- and  
19 cell-mediated immune system, exist in the lung to protect it from infectious and neoplastic  
20 disease and inhaled particles. Summaries of key new animal studies examining the effects of O<sub>3</sub>  
21 on lung host defenses are presented in Table AX5-7 of Annex AX5. Acute human exposures  
22 to O<sub>3</sub> result in similar effects on AMs (see Chapter 6).

### 23 24 **5.2.2.1 Clearance**

25 Early studies of the effect of O<sub>3</sub> on the mucociliary escalator showed morphological  
26 damage to ciliated epithelial cells of the tracheobronchial tree at doses of <1 ppm.  
27 Functionally, O<sub>3</sub> slowed particle clearance in rats at doses of 0.8 ppm for 4 h and in rabbits at  
28 0.6 ppm for 2 h exposures. Acute exposures at 0.5 ppm O<sub>3</sub> in sheep caused increased basal  
29 secretion of glycoproteins, while longer exposures reduced tracheal glycoprotein secretions,  
30 both of which can alter the effectiveness of the mucociliary escalator. Early postnatal exposures  
31 of sheep to 1 ppm O<sub>3</sub> caused retardation of normal morphologic development of the tracheal

1 epithelium, decreased epithelial mucosa density, decreased tracheal mucous velocity, and  
2 delayed development of carbohydrate composition. Conversely, alveolar clearance in rabbits  
3 after acute exposure (0.1 ppm, 2 h/day, for 1 to 4 days) is increased. Longer exposures showed  
4 no effect and increased O<sub>3</sub> (1.2 ppm) slowed clearance. This pattern of clearance occurs in rats  
5 also. A study using rat tracheal explants exposed to O<sub>3</sub> for 10 min (Churg et al., 1996) showed  
6 that uptake of TiO<sub>2</sub> and asbestos was enhanced at 0.01 and 0.1 ppm, respectively. The authors  
7 attribute the increased uptake as a direct effect of O<sub>3</sub>, suggesting mediation by H<sub>2</sub>O<sub>2</sub> or hydroxyl  
8 radical. Studies of the clearance of the radiolabeled chelate <sup>99m</sup>Tc diethylenetriamine pentaacetic  
9 acid (Tc-DTPA) have shown that clearance is significantly increased following a 3 h exposure  
10 to 0.8 ppm O<sub>3</sub> in SD rats (Pearson and Bhalla, 1997). Examination of regional clearance  
11 of <sup>99m</sup>Tc-DTPA in dogs following a 6 h isolated sublobar exposure to 0.4 ppm O<sub>3</sub> or air showed  
12 that O<sub>3</sub> decreased the clearance halftime by 50% at 1 day following exposure (Foster and Freed,  
13 1999). Clearance was still elevated at 7 d PE but had recovered by 14 d. So, a single local  
14 exposure to O<sub>3</sub> increases transepithelial clearance but without any influence on contralateral  
15 segments, i.e., only for epithelia directly exposed to O<sub>3</sub>.

16 Alveolar clearance is slower than tracheobronchial clearance and involves particle  
17 movement through interstitial pathways to the lymphatic system or movement of particle-laden  
18 AMs to the bottom of the mucociliary escalator. Exposures of rabbits to 0.1 ppm accelerated  
19 clearance while 1.2 ppm slowed clearance. A chronic exposure has been shown to slow  
20 clearance. New evaluations of the effects of O<sub>3</sub> on alveolar clearance have not been performed.

#### 22 **5.2.2.2 Alveolar Macrophages**

23 A primary function of AMs is to clear the lung of infectious and noninfectious particles by  
24 phagocytosis, detoxification, and removal. Further, AMs secrete cellular mediators that recruit  
25 and activate inflammatory cells in the lungs (see Figure 5-3). Ozone has been shown to inhibit  
26 phagocytosis at 0.1 ppm for 2 h in rabbits. This inhibition returns to control levels if exposures  
27 are repeated for several days. The production of superoxide anion radicals and the activity AM  
28 lysosomal enzymes (both involved in bactericidal activity) are inhibited by 3 h exposures to  
29 0.4 and 0.25 ppm O<sub>3</sub> in rodents and rabbits, respectively. Production of IFN $\gamma$  was decreased in  
30 rabbit AM by 1 ppm O<sub>3</sub> for 3 h.

1 New studies have shown that O<sub>3</sub> affects AM chemotaxis, cell adhesion, and surface  
2 expression of cell adhesion molecules (Bhalla, 1996). AM from SD rats exposed to 0.8 ppm O<sub>3</sub>  
3 for 3 h showed greater mobility and greater adhesion than air exposed controls. This increased  
4 mobility and adhesion were attenuated by CD16b and ICAM-1 antibodies, suggesting these  
5 adhesion molecules modulate O<sub>3</sub>-induced inflammation. Antibodies to TNF $\alpha$  and IL1 $\alpha$  also  
6 mitigated AM adherence, suggesting further that the inflammatory response to O<sub>3</sub> is mediated by  
7 these cytokines (Pearson and Bhalla, 1997). Cohen et al. (1996) showed that 1 ppm O<sub>3</sub> for 4 h  
8 reduces binding of INF  $\gamma$  to AM in WEHI-3 cells, and additionally reduces phagocytic activity,  
9 production of reactive oxygen intermediates, and elevation of intracellular Ca<sup>++</sup>.

10 Cohen et al. (2001, 2002) exposed male F-344 rats to either 0.1 or 0.3 ppm O<sub>3</sub> for 4 h/day,  
11 5 days/week or either 1 or 3 weeks. In this study, superoxide anion production was increased at  
12 1 week. Hydrogen peroxide production was reduced at both exposure concentrations and  
13 durations and was further reduced with INF $\gamma$  stimulation, suggesting that one effect of O<sub>3</sub> is  
14 compromised killing of bacteria by AM due to the reduction in hydrogen peroxide production.

15 Ozone treatment (2 ppm O<sub>3</sub>, 3 h in female SD rats) caused a time-dependent increase in  
16 NO levels in both AM and type II epithelial cells that was correlated with increased expression  
17 of iNOS mRNA and protein (Laskin et al., 1998). Inhibition of NF- $\kappa$ B, caused a dose-dependent  
18 inhibition of NO and iNOS production. Additionally, O<sub>3</sub> caused a time-dependent increase  
19 in NF- $\kappa$ B binding activity in the nucleus of both cell types. The authors hypothesize that O<sub>3</sub>  
20 exposure causes the cytokines TNF $\alpha$  and IL-1 $\beta$  to bind to surface receptors and initiate  
21 intracellular signaling pathways in AM leading to activation of NF- $\kappa$ B, its entry into the nucleus,  
22 and its binding to the regulatory sequences of genes such as iNOS to allow their transcription.  
23 Additional studies (Laskin et al., 2002) using AM isolated from C57Bl6x129 mice with a  
24 targeted disruption of the gene for iNOS showed no toxicity to 0.8 ppm O<sub>3</sub> for 3 h, as measured  
25 by BALF protein levels and nitrotyrosine staining of the lung. Additionally, mice  
26 overexpressing human Cu, Zn superoxide dismutase (SOD) and mice with a targeted disruption  
27 of p50 NF- $\kappa$ B were also resistant to O<sub>3</sub> toxicity. WT mice exposed to O<sub>3</sub> showed an increase in  
28 expression of STAT-1, a protein that binds to the regulatory region of iNOS. Taken together,  
29 these results suggest to the authors that a number of proteins including NF- $\kappa$ B, phosphoinoside  
30 3-kinase, and STAT-1 that bind to and regulate expression of iNOS are modulated by O<sub>3</sub>  
31 exposure. The same iNOS knockout mice strain exposed to 0.8 ppm O<sub>3</sub> for 3 h (Fakhrzadeh

1 et al., 2002) showed no increase in AM superoxide anion and prostaglandin. These data provide  
2 further evidence the NO and its reactive oxidative product peroxynitrite are important in O<sub>3</sub>-  
3 induced lung injury. Further discussions of the role of nitric oxide synthase/reactive nitrogen  
4 and cytokines/chemokines in O<sub>3</sub>-induced inflammation are provided in Section 5.2.3.

### 5 6 **5.2.2.3 Immune System**

7 Other than by natural protection (e.g., opsonizing antibody, nonspecific phagocytosis by  
8 AM), the immune system defends the lung by mounting three major waves of response: natural  
9 killer (NK) cells (nonspecific lymphocytes that kill viruses, bacteria, and tumor cells), followed  
10 by cytotoxic T-lymphocytes (T<sub>CTL</sub>- lymphocytes that lyse specifically recognized microbial and  
11 tumor-cell targets), followed by antigen-specific antibodies. These T-cell types are involved  
12 with other immunologically active cells (e.g., B-cells and AM), which in a complex manner,  
13 interact in immunological defense. To date, only a few of these mechanisms have been  
14 investigated in the context of their role in O<sub>3</sub> susceptibility. The effects of O<sub>3</sub> on the immune  
15 system are complex and depend on the exposure parameters and observation periods. T-cell-  
16 dependent functions appear to be more affected than B-cell-dependent functions. Generally,  
17 there is an early immunosuppressive effect that can, with continued exposure, either return to  
18 normal or actually enhance immunity. Changes in immune cell population occur with O<sub>3</sub>  
19 exposure including T:B-cell ratios in the MLN. Natural killer (NK) cell activity increases with  
20 1 week exposures of 0.2 to 0.4 ppm O<sub>3</sub> but decreases with exposures to 0.82 ppm. Ozone  
21 exposure has also shown to be responsible for enhancement of allergic sensitization at levels of  
22 0.5 to 0.8 ppm for 3 days. Studies of the effects of O<sub>3</sub> on the immune system are summarized in  
23 Table AX5-7.

24 Garssen et al. (1997) have studied the effects of O<sub>3</sub> on non-IgE-mediated pulmonary  
25 hyper-immune reactions induced by picryl chloride (PCI). BALB/c mice sensitized with PCI,  
26 both actively and passively (by adoptive transfer of lymphoid cells from pre-sensitized mice),  
27 were then challenged with picryl sulfonic acid (PSA). The mice were exposed to 12 h of 0.4, 0.8,  
28 or 1.6 mg/m<sup>3</sup> O<sub>3</sub> during one night, at 4 days or 7 days after skin sensitization (which was either  
29 just before or just after PSA challenge, i.e., during the induction or effector phase).  
30 Nonsensitized mice showed no changes in tracheal reactivity to carbacol with O<sub>3</sub> exposure.  
31 Sensitized mice were hyperreactive to carbachol 48 h after PSA challenge, whereas sensitized

1 mice exposed to all concentrations of O<sub>3</sub> showed no significant tracheal hyperreactivity to  
2 carbachol. The sensitized mice also demonstrated a suppressed inflammatory reaction (PMN)  
3 with 1.6 mg O<sub>3</sub> exposure. Ozone exposure following PSA challenge also caused a suppression  
4 of tracheal hyperresponsiveness. In a separate experiment wherein mice were exposed to O<sub>3</sub>  
5 before sensitization and then lymphoid cells from these mice were injected into nonexposed  
6 mice, the recipients also demonstrated an inhibition of the induction of hyperreactivity. These  
7 results are opposite to the effect on type I (IgE-mediated) allergic reactions, which the authors  
8 suggest is due to activation of Th-2 cell-dependent reactions that are possibly potentiated by O<sub>3</sub>  
9 or to a direct effect by O<sub>3</sub> on Th-1 cells or other cells that are crucial for the tracheal  
10 hyperreactivity and inflammation seen in this mouse model.

11 Kleeberger et al. (2000, 2001a) have demonstrated a potential interaction between the  
12 innate and acquired immune system with O<sub>3</sub> exposure. Using O<sub>3</sub>-susceptible (C57BL/6J)  
13 and O<sub>3</sub>-resistant (C3H/HeJ) mice, they identified a candidate gene on chromosome 4, Toll-like  
14 receptor 4 (*Tlr4*). Ozone exposure (0.3 ppm for 24 to 72 h) of C3H/HeJ and C3H/HeOuJ mice,  
15 the latter differing from the O<sub>3</sub>-resistant strain by a polymorphism in the coding region of *Tlr4*,  
16 demonstrated greater protein concentrations in the OuJ strain. The two strains exhibited  
17 differential expression of *Tlr4* mRNA with O<sub>3</sub> exposure. Thus, a quantitative trait locus on  
18 chromosome 4 appears to be responsible for a significant portion of the genetic variance in  
19 O<sub>3</sub>-induced lung hyperpermeability. In these mouse strains lavageable protein concentration was  
20 lowered by inhibition of inducible nitric oxide synthase (iNOS) and by targeted disruption of  
21 *Nos2*. Comparisons of C3H/HeJ and C3H/HeOuJ O<sub>3</sub> exposures demonstrated reduced *Nos2* and  
22 *Tlr4* mRNA levels in the O<sub>3</sub>-resistant C3H/HeJ mice. These data are consistent with the  
23 hypothesis that O<sub>3</sub>-induced lung hyperpermeability is mediated by iNOS. These studies suggest  
24 a role for TLR4 in the host response to O<sub>3</sub> similar to the role it has demonstrated in  
25 lipopolysaccharide (LPS) sensitivity (Schwartz 2002; Wells et al. 2003). TLR4 signaling is  
26 thought to be critical to linking the innate and acquired immune system through antigen  
27 presenting cells and Th1/Th2 differentiation.

28 Ozone exposure has been shown to affect Ig responses both in vitro and in mice. Becker  
29 et al. (1991) demonstrated changes in IgG production in cultured human lymphocytes with O<sub>3</sub>  
30 exposures of 1.0, 0.5, and 0.1 ppm for 2 h. Subsequent to O<sub>3</sub> exposure, cells were stimulated  
31 with pokeweed mitogen (PWM, a T-cell-dependent stimulus) or *Staphylococcus aureus* Cowan 1

1 strain (SAC, a T-cell-independent stimulus). Both B and T cells were affected by O<sub>3</sub>.  
2 T cells also demonstrated an increase in IL-6 and a decrease in IL-2, which suggested to the  
3 authors that O<sub>3</sub> may have direct effects on IgG producing cells and concurrently an effect that  
4 is mediated by altered production of T cell immunoregulatory molecules. Responses to  
5 repeated O<sub>3</sub> (0.08 - 0.25 ppm) and OVA (1%) exposures were compared in “IgE-high responder”  
6 (BALB/c) and “IgE-low responder” (C57BL/6) mice (Neuhaus-Steinmetz et al., 2000). Ozone  
7 appeared to shift the immune response toward a Th2-like pattern in the two mouse strains with  
8 differing potentials for developing allergic reactions.

9 Another study (Depuydt et al., 2002) demonstrated that O<sub>3</sub> (0.1 ppm for 2 h) increases  
10 allergen-induced airway inflammation in previously sensitized mice but has no effect on the  
11 sensitization process itself. This study uses OVA-pulsed dendritic cells instead of systemic  
12 adjuvant, which the authors consider a more relevant model of sensitization as it clearly  
13 separates the immune response from the challenge and does not obscure regulatory processes as  
14 does i.p. injections of OVA. They further suggest that dendritic cells, the principal antigen-  
15 presenting cells in the airway, are an important component of O<sub>3</sub>-induced eosinophilic airway  
16 inflammation.

17 Surfactant protein A and D (SP-A and SP-D) were shown to create an inflammatory  
18 feedback loop with perturbations in lung immune defenses (reviewed in Hawgood and Poulain,  
19 2001). Earlier studies suggested that SP-A is a target for O<sub>3</sub> toxicity by causing inhibition of  
20 SP-A self-association and SP-A-mediated lipid vesicle aggregation. Further, O<sub>3</sub> reduced the  
21 ability of SP-A to inhibit phospholipid secretion by alveolar type II cells and reduced the  
22 capacity of SP-A to induce superoxide anion production and enhance phagocytosis of herpes  
23 simplex virus. Bridges et al. (2000) reported that both SP-A and SP-D directly protect surfactant  
24 phospholipids and macrophages from oxidative damage by blocking accumulation of TBARS  
25 and conjugated dienes.

26 Eight human variants of SP-A in CHO cells exposed to O<sub>3</sub> (1ppm for 4 h) showed  
27 decreased ability to stimulate cytokine (TNF- $\alpha$  and IL-8) production in THP-1 cells, a  
28 macrophage-like cell line (Wang et al., 2002). Each variant had a unique time- and  
29 dose-dependent pattern of stimulation of cytokine production with O<sub>3</sub> exposure which the  
30 authors attribute to possible differences in susceptibility to O<sub>3</sub> oxidation. Targeted disruption of  
31 mouse SP-A and SP-D (Hawgood et al, 2002) caused increases in BAL phospholipid,

1 macrophage, and protein through 24 weeks of age. Further, the deficient mice developed patchy  
2 lung inflammation and air space enlargement consistent with emphysema. Future experiments  
3 using these null mice will help to establish the role of SP-A and SP-D in pulmonary host defense  
4 to O<sub>3</sub> exposure.  
5

#### 6 **5.2.2.4 Interactions with Infectious Microorganisms**

7 Ozone-induced dysfunction of host defense systems results in enhanced susceptibility to  
8 bacterial lung infections. Acute exposures of 0.08 ppm (3 h) O<sub>3</sub> can overcome the ability of  
9 mice to resist infection (by decreasing lung bactericidal activity) with Streptococcal bacteria,  
10 resulting in mortality. Changes in antibacterial defenses are dependent on exposure regimens,  
11 species and strain of test animal, species of bacteria, and age of animal, with young mice more  
12 susceptible to the effects of O<sub>3</sub>. The effect of O<sub>3</sub> exposure on antibacterial host defenses appears  
13 to be concentration- and time-dependent. Early studies using the mouse “infectivity model,”  
14 consisting of exposure to clean air or O<sub>3</sub> followed by exposure to an aerosolized microorganism,  
15 showed that the difference in mortality between O<sub>3</sub>-exposed groups and controls is  
16 concentration-related. Chronic exposures (weeks, months) of 0.1 ppm do not cause greater  
17 effects on infectivity than short exposures, due to defense parameters becoming reestablished  
18 with prolonged exposures.

19 More recent studies of O<sub>3</sub>-induced modulation of cell-mediated immune responses showed  
20 effects on the onset and persistence of infection. Cohen et al. (2001, 2002) exposed male F-344  
21 rats subchronically to either 0.1 or 0.3 ppm O<sub>3</sub> for 4 h/day 15 days/week, for 1 or 3 weeks.  
22 Subsequent exposure with viable *Listeria monocytogenes* demonstrated no observed effect on  
23 cumulative mortality but did show a concentration-related effect on morbidity onset and  
24 persistence. These data suggest that O<sub>3</sub> may cause a possible imbalance between Th-1 and Th-2  
25 cells, which can subsequently lead to suppression of the resistance to intracellular pathogens.

26 Effects of O<sub>3</sub> on viral infections are dependent on the temporal relationship between O<sub>3</sub>  
27 exposure and viral infection. Only high concentrations (1.0 ppm O<sub>3</sub>, 3 h/day, 5 days, mice)  
28 increased viral-induced mortality. No detrimental effects were seen with a 120-day exposure to  
29 0.5 ppm O<sub>3</sub> on acute lung injury from influenza virus administered immediately before O<sub>3</sub>  
30 exposure started. But there were O<sub>3</sub>-enhanced postinfluenzal alveolitis and lung parenchymal  
31 changes. As O<sub>3</sub> does not affect lung influenza viral titers, it apparently does not impact antiviral



1 clearance mechanisms. In general, the evidence suggests that O<sub>3</sub> can enhance both bacterial and  
2 viral lung infections, but the key mechanisms have not yet been identified. New studies on the  
3 interactions of O<sub>3</sub> and viral infections have not been published.

#### 4 5 **5.2.2.5 Summary and Conclusions - Lung Host Defenses**

6 New data on lung host defenses support earlier work which suggests that mucociliary  
7 clearance is affected in most test species at just under 1 ppm, with lower levels (~0.1 ppm)  
8 increasing clearance and somewhat higher levels decreasing clearance. These data also propose  
9 mechanisms whereby O<sub>3</sub> affects clearance, which include uptake being a direct effect of O<sub>3</sub>, but  
10 modulated by ROS and hydroxyl radicals.

11 Alveolar macrophage function is disrupted by O<sub>3</sub> as shown by a number of studies  
12 demonstrating inhibition of phagocytosis at concentrations ranging from 0.1 to 1.2 ppm. This  
13 inhibition returns to control levels if exposures are repeated for several days. Two new studies  
14 corroborate earlier findings of increases in AM number in that same exposure range. In this  
15 environmentally relevant exposure range, new studies support older findings of decreased  
16 resistance to microbial pathogens as shown by the endpoints examining superoxide radical  
17 formation, altered chemotaxis/motility, decreased INF $\gamma$  levels, decreased lysosomal activity,  
18 increased PGE levels, and increased NO mRNA and protein.

19 New research evaluating the effects of O<sub>3</sub> on immune function advances previous work that  
20 has shown that exposures can enhance or suppress immune responsiveness depending on the  
21 species studied, concentration of O<sub>3</sub>, route of exposure of allergen, and timing of exposure.  
22 Continuous exposure to O<sub>3</sub> impairs immune responses for the first several days of exposure,  
23 followed by an adaptation to O<sub>3</sub> that allows a return of normal immune responses. Most species  
24 show little effect of O<sub>3</sub> exposures prior to immunization, but a suppression of responses to  
25 antigen in O<sub>3</sub> exposures post-immunization. The use of mouse strains with genetically  
26 determined sensitivity or resistance to O<sub>3</sub> indicated a possible interaction between the innate and  
27 acquired immune system, and further, that O<sub>3</sub> may shift the immune response towards a Th-2-  
28 like pattern. Work has also focused the deleterious effects of O<sub>3</sub> exposure on SP-A and SP-D  
29 and their immunomodulatory function in protecting against oxidative stress.

30 Several new studies evaluating the effects of O<sub>3</sub> exposures on infectious microorganisms  
31 are in concurrence with previous studies which showed, in general, increased mortality and

1 morbidity, decreased clearance, increased bacterial growth, and increased severity of infection at  
2 exposure levels of 0.1 to 1 ppm O<sub>3</sub> for 1 week.

### 3 4 **5.2.3 Inflammation and Lung Permeability Changes**

5 The normal lung has an effective barrier function that controls bidirectional flow of fluids  
6 and cells between the air and blood compartments. Ozone disrupts this function, resulting in two  
7 well-characterized effects of O<sub>3</sub> exposure, lung inflammation and increased permeability, which  
8 are distinct events controlled by independent mechanisms. Ozone initiates inflammation of lung  
9 tissue by reactions with antioxidants and lipids in ELF (discussed in 5.2.1, see Figure 5-2).

10 Secondary reaction products generated in this process then cause changes in cell membranes,  
11 disruption of the lung barrier leading to leakage of serum proteins, influx of polymorphonuclear  
12 leukocytes (PMNs), release of bioactive mediators, and movement of compounds from the  
13 airspaces into the blood. This increased permeability allows accumulation of co-occurring  
14 pollutants into the lung tissue. The framework for presenting this stereotypical response to O<sub>3</sub>  
15 consists of discussions covering: 1) the time course of these changes; 2) concentration × time  
16 (C × T) relationships; 3) susceptibility factors; 4) mediators of inflammation; and 5) nitric oxide  
17 and reactive nitrogen.

18 Rats appear to be more resistant to O<sub>3</sub>-induced inflammation than humans (see Chapter 4).  
19 With comparable exposure protocols, both species have similar observed inflammatory and  
20 permeability changes, i.e., controlled human exposure studies discussed in Chapter 6 indicate  
21 that the majority of acute responses in humans are similar to those observed in animals.

22 Ozone also increases the permeability from the air to the blood compartment. Ozone  
23 (0.8 ppm; 2 h) caused a 2-fold increase of the transport of labeled DTPA from the rat tracheal  
24 lumen to the blood. This coincided with a 2-fold increase in the number of endocytic vesicles in  
25 epithelial cells that contained intraluminally instilled HRP as a tracer. These studies also suggest  
26 an uneven disruption of tight junctions and alternate transport through endocytotic mechanisms.  
27 In studies aimed at detecting the effects of O<sub>3</sub> exposure on regional permeability, O<sub>3</sub> increased  
28 the transmucosal transport of DTPA and BSA more in the trachea and bronchoalveolar zone than  
29 in the nose. These changes in barrier integrity may allow increased entry of antigens and other  
30 bioactive compounds (e.g., bronchoconstrictors) into lung tissue. Data from analyses at regular

1 intervals PE indicate that maximal increases in BALF protein, albumin and number of PMNs  
2 occur 8 to 18 h (depending on the study) after an acute exposure ceases.

3 Increases in permeability and inflammation have been observed at levels as low as  
4 0.1 ppm O<sub>3</sub> for 2 h/day for 6 days in rabbit and 0.12 ppm in mice (24 h exposure) and rats  
5 (6 h exposure). After acute exposures, the influence of the time of exposure increases as the  
6 concentration of O<sub>3</sub> increases. The exact role of inflammation in causation of lung disease is not  
7 known, nor is the relationship between inflammation and changes in lung function. Table  
8 AX5-8 in Annex AX5 summarizes new key studies describing the potential for O<sub>3</sub> exposure  
9 effects on lung permeability and inflammation.

### 11 **5.2.3.1 Time Course of Inflammation and Lung Permeability Changes**

12 The maximal increase in BALF protein, albumin, and PMN occurs in most species 8 to  
13 18 h after the cessation of acute exposures of 0.5 to 1.0 ppm. A study of OVA-sensitized male  
14 Dunkin-Hartley guinea pigs exposed to 1.0 ppm O<sub>3</sub> for 3 h showed that levels of PMN  
15 significantly increased at 3 h PE, but BAL protein levels did not, suggesting a lack of correlation  
16 between the two endpoints (Sun et al., 1997). Increased PMN without a concordant increase in  
17 BAL protein levels were found when the guinea pigs were exposed to 1.0 ppm O<sub>3</sub> for 1 h and  
18 evaluated 24-h PE. The first group also had an increase in AHR, but not the second group,  
19 which suggests a dissociation between PMN levels and AHR.

20 Earlier work demonstrated that O<sub>3</sub> exposures of 0.8 to 1 ppm in rat and guinea pig  
21 transiently increase the permeability from the air to the blood compartment. This permeability is  
22 greatest in trachea and bronchoalveolar zone, and may allow increased entry of antigens and  
23 other bioactive compounds (e.g., bronchoconstrictors) into lung tissues. The time course of the  
24 influx of PMNs into the lung and the BALF fluid levels of macrophage inflammatory protein-2  
25 (MIP-2) were found to be roughly similar to that for proteins (Bhalla and Gupta, 2000).  
26 Adherence of neutrophils to pulmonary vascular endothelium is maximal within 2 h after  
27 exposure and returns to control levels by 12 h PE (Lavnikova et al., 1998). Cheek et al. (1995)  
28 cultured monolayers of rat alveolar type II cells and exposed them to 0.1 or 0.5 ppm O<sub>3</sub> for 0.5 h  
29 to evaluate the effects of O<sub>3</sub> on permeability. Permeability increased dose-dependently and the  
30 higher exposures elicited greater numbers of injured epithelial cells. Exposure to 0.1 ppm O<sub>3</sub>  
31 was thought to expedite the restoration of epithelial barrier functions, while in higher exposures,

1 neutrophils exacerbated the O<sub>3</sub>-induced injury. Vesely et al. (1999a) have demonstrated that  
2 neutrophils contribute to the repair process in O<sub>3</sub>-injured airway epithelium and they may play a  
3 role in removal of O<sub>3</sub>-injured cells.

4 Exposures of 3 to 7 days have been found to cause increases in BALF protein and PMNs  
5 that typically peak after a few days (depending upon species tested and exposures) and return  
6 towards control even with continuing exposure. Van Bree et al. (2002) observed lower  
7 BALF levels of protein, fibronectin, IL-6 and inflammatory cells in rats exposed for 5 days to  
8 0.4 ppm O<sub>3</sub> than in rats exposed for 1 day, suggesting adaptation to O<sub>3</sub> exposure. Postexposure  
9 challenge with single O<sub>3</sub> exposures at different time points showed recovery of susceptibility  
10 to O<sub>3</sub>. McKinney et al. (1998) observed differences in IL-6 levels due to repetitive exposures  
11 and demonstrated a role of IL-6 in the adaptive response induced by repeated O<sub>3</sub> exposures of  
12 0.5 ppm for 4 h.

#### 14 **5.2.3.2 Concentration and Time of Exposure**

15 The relative influence of concentration and duration of exposure (i.e., C × T) has been  
16 investigated extensively in rats, using BALF protein as an endpoint. Earlier work utilizing  
17 concentrations of 0.1 to 2 ppm O<sub>3</sub> and durations of 1 to 8 h has shown that the interaction  
18 between C and T is complex. At these levels of exposure, concentration generally dominates the  
19 response. C × T studies using the endpoints of changes in lung protein or cell type showed that  
20 acute damage is a function of cumulative dose. The impact of T is C-dependent (at higher Cs,  
21 the impact of T is greater); at the lowest C and T values, this dependence appears to be lost. The  
22 controlled human exposure data described in Chapter 6 concur with most animal data, showing  
23 that concentration of O<sub>3</sub> is the most important factor determining O<sub>3</sub> responses, and that duration  
24 of exposure and ventilation rate are secondary factors.

25 New studies evaluating C × T relationships in animal models have not been completed.  
26 However, a full understanding of C × T relationships in ambient exposures must include the  
27 recognition that ‘real world’ exposures are cyclic in nature, due to the daily and seasonal  
28 variations in O<sub>3</sub> levels. The concentration of O<sub>3</sub>, the duration of the exposure, and duration  
29 between exposures are all relevant to the type and level of O<sub>3</sub>-induced injury.

### 5.2.3.3 Susceptibility Factors

Factors that have been studied for potential impact on the effects of O<sub>3</sub> exposure include age, gender, nutritional status, exposure to co-pollutants, exercise, and genetic variability. A full characterization of the effects of age on O<sub>3</sub> responses has not been completed. Data available indicate that effects of age on O<sub>3</sub> responses are endpoint-dependent, with young mice, rats, and rabbits having greater prostaglandin levels with exposure and senescent rats having greater IL-6 and N-acetyly-β-D-glucosaminidase levels with exposure.

A study (Johnston et al., 2000a) compared gene expression of chemokines and cytokine in newborn and 8-week-old C57Bl/6J mice exposed to 1.0 or 2.5 ppm for 4, 20, or 24 h. The newborn mice displayed increased levels of Mt mRNA only, while the 8-week-old mice had increases in MIP-1α, MIP-2, IL-6, and Mt mRNA. Comparisons were made with mice of the same age groups with exposures to endotoxin (10 ng/mouse for 10 min). Both age groups displayed similar cytokine/chemokine profiles with endotoxin exposure. This suggested to the authors that the responses to endotoxin, which does not cause epithelial injury, and the responses to O<sub>3</sub>, which does, demonstrate that differences in inflammatory control between newborn and adult mice is secondary to epithelial injury.

Pregnancy and lactation increased the susceptibility of rats to acute O<sub>3</sub>, but no clear effects of gender have been identified. The effects of vitamin C deficiency on O<sub>3</sub> responses are unclear. Ascorbate-deficient guinea pigs exposed to O<sub>3</sub> demonstrated only minimal effects on injury and inflammation (Kodavanti et al., 1995). Utilizing a diet-restricted (20% of the freely-fed diet) rat model, Elsayed (2001) demonstrated higher survivability on exposure to higher O<sub>3</sub> (0.8 ppm continuously for 3 days) compared to freely-fed rats. Pre-exposure to sidestream cigarette smoke had been found to cause increased lung injury (Yu et al., 2002). In vitro studies on the macrophages from smoke + O<sub>3</sub>-exposed animals responded by a greater release of TNF-α following LPS stimulation when compared to macrophages exposed to air, smoke or O<sub>3</sub> (0.5 ppm, 24 h) alone.

Lines of evidence illustrate that genetic background is an extremely important determinant of susceptibility to O<sub>3</sub>. Earlier studies using inflammation-prone (susceptible) C57BL/6J (B6) and inflammation-resistant C3H/HeJ (C3) mouse strains and high doses of O<sub>3</sub> (2 ppm for 3 h) identified *Inf-2* as a locus controlling susceptibility. Further studies in these two strains of mice using more relevant exposures (0.3 ppm for 72 h) identified that the acute and subacute

1 exposures are controlled by two distinct genes, referred to as *Inf-1* and *Inf-2*, respectively  
2 (Tankersley and Kleeberger, 1994). Exposures to 0.3 ppm O<sub>3</sub> for 48 or 72 h, when repeated  
3 fourteen days after the initial exposures, caused a smaller increase in BALF protein and number  
4 of macrophages, lymphocytes and epithelial cells in both strains, but PMN number was greater  
5 in both strains compared to initial exposure (Paquette et al., 1994). Kleeberger et al. (1997) also  
6 identified another potential susceptibility gene, tumor necrosis factor (*Tnf*, which codes for the  
7 pro-inflammatory cytokine TNF- $\alpha$ ) on a qualitative trait locus on mouse chromosome 17.  
8 By neutralizing the function of TNF- $\alpha$  with a specific antibody, they were able to confer  
9 protection against O<sub>3</sub> (0.3 ppm, 48 h) injury in susceptible mice. The group then demonstrated a  
10 role for TNF receptor 1 and 2 (TNFR1 and TNFR2, respectively) signaling in subacute (0.3 ppm  
11 for 48 h) O<sub>3</sub>-induced pulmonary epithelial injury and inflammation (Cho et al., 2001). TNFR1  
12 and TNFR2 knockouts were less sensitive to subacute O<sub>3</sub> exposure than WT C57BL/6J mice.

13 An integrated and more comprehensive effort to identify the genetic basis for the  
14 susceptibility to O<sub>3</sub>-induced lung injury was reported by Savov et al. (2004). In this report, acute  
15 lung injury to high dose of O<sub>3</sub> (2 ppm for 3 h) was assessed and integrated with physiological,  
16 biochemical, and genetic observations using 9 inbred mouse strains. This work indicated the  
17 presence of genetic loci on chromosomes 1, 7, and 15 associated with phenotypic  
18 characteristics for resistance to acute O<sub>3</sub>-induced lung injury. They identified C3H/HeJ and  
19 A/J as consistently O<sub>3</sub>-resistant, C57BL/6J and 129/SvIm as consistently O<sub>3</sub>-vulnerable, and  
20 CAST/Ei, BTBR, DBA/2J, FVB/NJ, and BALB/cJ as intermediate in response to O<sub>3</sub>.

21 Ozone-induced changes in CCSP (called CC16 by this group) expression were evaluated in  
22 five inbred mouse strains: C57BL/6J and CBA both considered sensitive to acute O<sub>3</sub>-induced  
23 inflammation, C3H/HeJ and AKR/J both considered resistant, and SJL/J considered intermediate  
24 (Broeckeaert et al., 2003). Two exposures paradigms were used, 1.8 ppm O<sub>3</sub> for 3 h or  
25 0.11 ppm O<sub>3</sub>, 24/h day for up to 3 days, and BALF and serum was assayed immediately after  
26 exposure or at 6 h PE. Both exposure levels caused a transient increase in CC16 in serum that  
27 correlated with BALF changes in protein, LDH, and inflammatory cells. There was an inverse  
28 relationship between preexposure levels of CC16 in BALF and epithelial damage based on  
29 serum CC16 levels and BALF markers of inflammation. There was also an inverse relationship  
30 between preexposure levels of albumin in BALF and lung epithelium damage. Based on these  
31 results, the authors conclude that a major determinant of susceptibility to O<sub>3</sub> is basal lung

1 epithelial permeability. As all of the mouse strains had similar levels of preexposure CC16  
2 mRNA, they explored the possible role of CC16 isozymes in differences among strains. The  
3 CC16 monomer a 7kD protein exists in two isoforms with differing pI values, CC16a (4.9) and  
4 CC16b (5.2). To evaluate the role of CC16 isoform profiles in permeability differences between  
5 C57BL/6J and C3H/HeJ, this group evaluated the CC16 protein profiles in BALF of the strains  
6 before and after O<sub>3</sub> exposure following two-dimensional protein electrophoresis analysis.  
7 C57BL/6J mice had lower levels of CC16a (the more acidic form) than C3H/HeJ. But both the  
8 strains had similar levels of CC16b. Based on these observations, Broeckaert et al (2003)  
9 conclude that greater epithelial permeability observed in C57BL/6J may be due to difference in  
10 the expression of CC16a and possibly other antioxidant/inflammatory proteins.

11 Wattiez et al. (2003) examined BALF protein from C57BL/6J (O<sub>3</sub>-sensitive) and  
12 C3H/HeJ (O<sub>3</sub>-resistant) mice exposed to filtered air using a two-dimensional polyacrylamide  
13 gel approach to analyze the protein profiles. C3H/HeJ mice expressed 1.3 times more Clara cell  
14 protein16 (CC16) than C57BL/6J mice and, further, expressed more of the acidic isoform of  
15 CC16. Strain-specific differential expression of isoforms of the antioxidant protein 2 (AOP2),  
16 the isoelectric point 5.7 isoform in C3H/HeJ and isoelectric point 6.0 isoform in C57BL/6J were  
17 observed. These data suggest a protective role for CCSP against oxidative damage, and further,  
18 that genetic susceptibility to oxidant stress may be moderated, in part, by the gene coding for  
19 CCSP. Taken together, these mouse studies of genetic susceptibility are useful for  
20 understanding underlying mechanisms leading to O<sub>3</sub>-induced effects. However, at this point,  
21 corresponding human polymorphisms have not yet been identified which associate with differing  
22 human sensitivities to O<sub>3</sub>.

#### 23 24 **5.2.3.4 Mediators of Inflammatory Response and Injury**

25 Ozone reacts with lipids in the ELF or epithelial cell membranes, creating ozonation  
26 products which then stimulate airway epithelial cells, AMs, and PMNs to release a host of  
27 pro-inflammatory mediators including cytokines, chemokines, reactive oxygen species,  
28 eicosanoids, and platelet activating factor (see Figure 5-3). While neutrophils in the lung  
29 characterize an inflammatory response to O<sub>3</sub>, the release of chemotactic mediators by  
30 inflammatory cells indicates their state of activation and their role in continued inflammation  
31 and injury. At O<sub>3</sub> exposures of ≥ 1 ppm, these mediators recruit PMN, and increase expression of

1 MIP-2 mRNA or BALF levels of MIP-2 (Driscoll et al., 1993; Haddad et al., 1995; Bhalla and  
2 Gupta, 2000). The increased mRNA expression was associated with an increased neutrophilia in  
3 the lung. Zhao et al. (1998) showed that 0.6 ppm O<sub>3</sub> exposure for 2 h in mice and rats causes an  
4 increase in monocyte chemotactic protein-1 (MCP-1).

5 Fibronectin, an extracellular matrix glycoprotein, is thought to have a role in lung  
6 inflammation and inflammatory disorders, and has shown to be increased with exposure to  
7 1 ppm O<sub>3</sub> for 14 days. Gupta et al. (1998) observed an increase in both fibronectin protein and  
8 mRNA expression in the lung of rats exposed to 0.8 ppm O<sub>3</sub> for 3 h. A mechanistic role of  
9 fibronectin in O<sub>3</sub>-induced inflammation and injury was suggested on the basis of comparability  
10 of temporal changes in BALF protein, fibronectin and alkaline phosphatase activity with  
11 exposures of 1 ppm for 3 h (Bhalla et al., 1999). Studies have reported an effect of O<sub>3</sub> on other  
12 cytokines and inflammatory mediators. An increase occurred for cytokine-induced neutrophil  
13 chemoattractant (CINC) and NF-κB expression in vivo (Koto et al., 1997), for IL-8 in vivo and  
14 in vitro (Chang et al., 1998), TNFα, fibronectin, IL-1 and CINC release by macrophages ex vivo  
15 (Pendino et al., 1994; Ishii et al., 1997), and NF-κB and TNFα (Nichols et al., 2001; see 6.9.2).  
16 An increase in lung CINC mRNA occurred within 2 h after the end of a 3 h exposure of rats to  
17 1 ppm O<sub>3</sub>. The CINC mRNA expression was associated with neutrophilia at 24 h PE. Exposure  
18 of guinea pig AMs recovered in BALF and exposed in vitro to 0.4 ppm O<sub>3</sub> for 1 h produced a  
19 significant increase in IL-6 and TNFα (Arsalane et al., 1995). An exposure of human AMs to an  
20 identical O<sub>3</sub> concentration increased TNFα, IL-1β, IL-6 and IL-8 and their mRNAs. Ozone  
21 exposure (0.3 to 2.5 ppm, 1-48 h) of mice caused an increase in IL-6, MIP-1α, MIP-2,  
22 eotaxin and Mt abundance (Johnston et al., 1999a). The IL-6 and Mt increase was enhanced in  
23 mice deficient in CCSP, suggesting a protective role of Clara cells and their secretions (Mango  
24 et al., 1998). CCSP deficiency, also increased sensitivity of mice to O<sub>3</sub>, as determined by an  
25 increase in abundance of MIP-1α and MIP-2 following a 4 h exposure to 1.0 ppm O<sub>3</sub>  
26 (Johnston et al., 1999b).

27 Mast cells, which are located below the epithelium, release proinflammatory mediators and  
28 have been shown to contribute to O<sub>3</sub>-induced epithelial damage. Greater increases in lavageable  
29 macrophages, epithelial cells and PMNs were observed in mast cell-sufficient mice than in mast  
30 cell-deficient mice exposed to 0.26 ppm O<sub>3</sub> for 8 h per day, 5 days per week (Kleeberger et al.,  
31 2001b). Increases in inflammatory cells were also observed in mast cell-deficient mice repleted



1 of mast cells, however O<sub>3</sub>-induced permeability changes were similar between genotypic groups  
2 exposed to 0.26 ppm. When a RBL-2H3 mast cell line was exposed to 0.1 to 1.0 ppm O<sub>3</sub> for 1 h,  
3 spontaneous release of serotonin and modest generation of prostaglandin D<sub>2</sub> occurred only under  
4 conditions that caused cytotoxicity (Peden and Dailey, 1995). Additionally, O<sub>3</sub> inhibited IgE-  
5 and A23187-induced degranulation. Mast cells recovered from O<sub>3</sub>-exposed peripheral airways  
6 of ascaris sensitive dogs released significantly less histamine and PGD<sub>2</sub> following *in vitro*  
7 challenge with ascaris antigen or calcium ionophore (Spannhake, 1996). Ozone (0.4 ppm,  
8 5 weeks) exposure also promoted eosinophil recruitment in the nose and airways in response to  
9 instillation of OVA or OVA-pulsed dendritic cells and aggravated allergy like symptoms in  
10 guinea pigs (Iijima et al., 2001).

11 The role of PMNs and cellular mediators in lung injury and epithelial permeability has  
12 been investigated using antibodies and inhibitors of known specificity to block inflammatory cell  
13 functions and cytokine activity. Treatment of rats with cyclophosphamide prior to O<sub>3</sub> exposure  
14 (0.8 ppm, 48 h) resulted in a decreased recovery of PMNs in the BALF and attenuated  
15 permeability induced by O<sub>3</sub> (Bassett et al., 2001).

16 Pretreatment of animals with antiserum against rat neutrophils abrogated PMN  
17 accumulation in the lung, but did not alter permeability increase produced by O<sub>3</sub>. Studies  
18 utilizing antibodies to selected pro- or anti-inflammatory cytokines suggest a role of TNF $\alpha$ ,  
19 IL-10, and IL-1 $\beta$  in O<sub>3</sub>-induced changes in permeability, inflammation and cytokine release  
20 (Ishii et al., 1997; Reinhart et al., 1999; Bhalla et al., 2002) in exposures of ~1 ppm for 3-6 h.  
21 An attenuation of O<sub>3</sub>-induced increases in permeability and inflammation was also observed in  
22 mice treated, either before or after exposure, with UK-74505, a platelet-activating factor (PAF)  
23 receptor antagonist (Longphre et al., 1999). These results were interpreted to indicate that  
24 O<sub>3</sub>-induced epithelial and inflammatory changes are mediated in part by activation of PAF  
25 receptors.

26 Ozone exposure stimulates macrophage motility towards a chemotactic gradient, and  
27 macrophages isolated from rats exposed to 0.8 ppm O<sub>3</sub> for 3 h adhered to epithelial cells  
28 (ARL-14) in culture to a greater extent than macrophages from air-exposed controls (Bhalla,  
29 1996). Both macrophage motility and chemotaxis were attenuated by antibodies to cell adhesion  
30 molecules CD-11b and ICAM-1, suggesting a role for cell adhesion molecules in O<sub>3</sub>-induced  
31 cellular interactions. This may also explain the increased tissue localization and reduced

1 recovery of macrophages in BALF (Pearson and Bhalla, 1997) following O<sub>3</sub> exposure (0.8 ppm,  
2 3 h). Studies investigating the mechanisms of PMN recruitment in the lung have explored the  
3 role of cell adhesion molecules that mediate PMN-endothelial interactions. An exposure of  
4 female rats to O<sub>3</sub> (1 ppm, 2 h) had an attenuating effect on CD-18 expression on AMs and  
5 vascular PMNs, but the expression of CD62L, a member of selection family, on vascular PMNs  
6 was not affected (Hoffer et al., 1999). In monkeys, O<sub>3</sub>-induced (0.8 ppm, 8 h) inflammation was  
7 blocked by treatment with a monoclonal antibody to CD18, suggesting dependence of PMN  
8 recruitment on this adhesion molecule (Hyde et al., 1999). Treatment of monkeys with CD18  
9 antibody also reduced tracheal expression of the β6 integrin (Miller et al., 2001) suggesting that  
10 lung epithelial cell expression of this adhesion molecule is associated with sites of neutrophil  
11 recruitment. A single 3 h exposure of rats to 1 ppm O<sub>3</sub> caused an elevation in concentration of  
12 ICAM-1, but not CD-18, in the BALF (Bhalla and Gupta, 2000). Takahashi et al. (1995a) found  
13 an increase in tissue expression of ICAM-1 in mice exposed to 2 ppm O<sub>3</sub> for 3 h, noting a  
14 temporal correlation of inflammatory activity and ICAM-1 expression which varied in different  
15 regions of the lung. A comparable pattern of time-related changes in total protein, fibronectin  
16 and alkaline phosphatase activity in the BALF of rats exposed to 0.8 ppm O<sub>3</sub> for 3 h was also  
17 noted by Bhalla et al. (1999). Together, these studies support the role of extracellular matrix  
18 protein and cell adhesion molecules in the induction of lung inflammation and injury.

#### 19 20 **5.2.3.5 The Role of Nitric Oxide Synthase and Reactive Nitrogen in Inflammation**

21 Nitric oxide (NO) is a messenger molecule involved in many biological processes,  
22 including inflammation (see Figure 5-3). Cells in the respiratory tract (including mast cells,  
23 neutrophils, epithelial cells, neurons, and macrophages) produce three differing forms of nitric  
24 oxide synthase (NOS), the enzyme that catalyzes the formation of NO. NOS-1 (neuronal) and  
25 NOS-3 (endothelial) are constitutively expressed, whereas NOS-2 (also referred to as iNOS) is  
26 inducible, commonly by pro-inflammatory cytokines. Macrophages isolated from O<sub>3</sub>-exposed  
27 (0.8 ppm for 3 h) mice produced increased amounts of NO, superoxide anion, and PGE<sub>2</sub>,  
28 but production of these mediators by macrophages from NOS knockout mice was not  
29 elevated (Fakhrzadeh et al., 2002). Additionally, mice deficient in NOS or mice treated  
30 with N<sup>G</sup>-monomethyl-L-arginine, an inhibitor of total NOS, were protected from O<sub>3</sub>-induced  
31 permeability, inflammation, and injury, suggesting a role of NO in the production of O<sub>3</sub> effects

1 (Kleeberger et al., 2001a; Fakhrzadeh et al., 2002). These results contrast with a study showing  
2 that O<sub>3</sub> exposure (of 1 ppm for 8 h/night for 3 nights) produced greater injury, as determined by  
3 measurement of MIP-2, matrix metalloproteinases, total protein, cell content and tyrosine  
4 nitration of whole lung protein, in iNOS knockout mice than in wild type mice (Kenyon et al.,  
5 2002). This group suggests that protein nitration is related to inflammation and is not dependent  
6 on iNOS-derived NO. They point out the possible experimental differences, such as O<sub>3</sub>  
7 concentration, for inconsistency between their results and those of Kleeberger et al. (2001a).

8 Rats pretreated with ebselen, a potent anti-inflammatory, immunomodulator, and  
9 NO/peroxynitrite scavenger, then exposed to 2 ppm O<sub>3</sub> for 4 h had decreased numbers of  
10 neutrophils, lowered albumin levels, and inhibited nitration of tyrosine residues in BALF 18 h  
11 PE, though macrophage iNOS expression was not changed (Ishii et al., 2000a). These results  
12 suggest an iNOS-independent mechanism for O<sub>3</sub>-induced inflammation. Jang et al. (2002)  
13 showed dose-dependent increases in nitrate (indicative of in vivo NO generation) with O<sub>3</sub>  
14 exposure (0.12, 0.5, 1, or 2 ppm for 3 h). Functional studies of enhanced pause (P<sub>enh</sub>)  
15 demonstrated increases with O<sub>3</sub> exposure which were also dose-dependent. Western blot  
16 analysis of lung tissue showed increases in NOS-1, but not in NOS -3 or iNOS isoforms. These  
17 results suggest that in mice NOS-1 may induce airway responsiveness by a neutrophilic airway  
18 inflammation. The literature regarding the effects of O<sub>3</sub> exposure on NOS activity is complex  
19 and conflicting. Similarly, the issue of protein nitration as it relates to cell injury due to O<sub>3</sub>  
20 exposure is somewhat controversial.

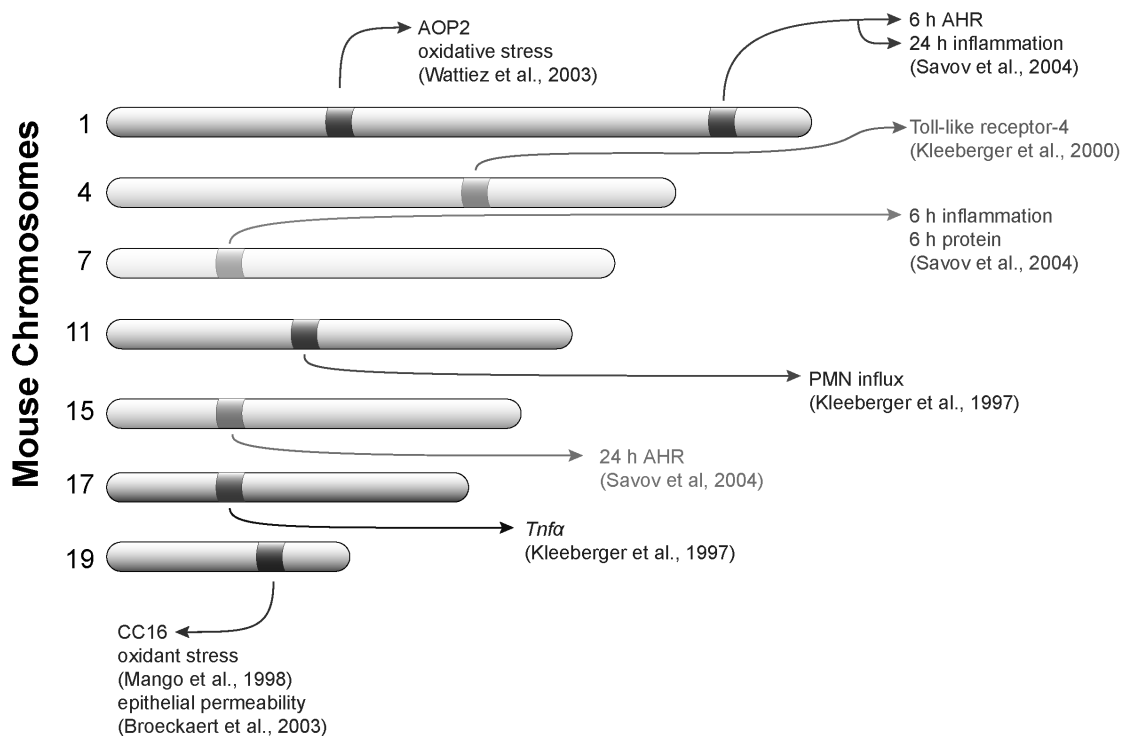
### 21 22 **5.2.3.6 Summary and Conclusions - Inflammation and Permeability Changes**

23 Figure 5-3 depicts many of the inflammatory and permeability changes that occur with O<sub>3</sub>  
24 exposure. Additionally, the figure demonstrates links between inflammatory/permeability  
25 responses and altered spirometric responses (discussed in Section 5.2.5), ciliary motility  
26 (discussed in Section 5.2.2.1), airway hyperreactivity (discussed in Section 5.2.5.3), and possible  
27 thrombolytic effects (Section 5.3.3). Airway mucosa in the normal lung serves as an effective  
28 barrier that controls bidirectional flow of fluids and cells between the air and blood  
29 compartments. Ozone disrupts this function, resulting in a cascade of effects which includes an  
30 increase in serum proteins, bioactive mediators, and PMNs in the interstitium and air spaces of  
31 the lung. Damaged epithelial cells release cytokines, which function to recruit and activate AMs

1 and PMNs. PMN recruitment into the lung is maximal at several hours PE. PMN recruitment is  
2 followed by blood monocytes which enter the lung and enlarge to become AMs. The AMs  
3 persist for days to weeks, phagocytizing injured cells. Activated PMNs and AMs continue the  
4 cascade of effects by further releasing inflammatory mediators, which serve to amplify the initial  
5 effects of O<sub>3</sub>. Generally, the initiation of inflammation is an important component of the defense  
6 process; however, its persistence and/or repeated occurrence can result in adverse health effects.  
7 Activation of this inflammatory cascade takes several hours. Chemical mediators released early  
8 in the cascade contribute to effects on pulmonary function. Events later in the cascade, by  
9 which time O<sub>3</sub>-induced alterations pulmonary function have attenuated, are related to sustained  
10 inflammation. Further, mechanistic separation of inflammation, permeability, and airway  
11 hyperreactivity (AHR) is suggested by the temporal disparities of their increases.  
12 The O<sub>3</sub>-induced disruption of the tight junctions between epithelial cells also increases  
13 the permeability between the air and blood compartments. This disruption, occurring with  
14 exposures of 0.8 ppm for 2 h, is greater in the trachea and bronchoalveolar zone than in the  
15 nose and allows entry of particles, including bioactive compounds, into the lung tissue.

16 For environmentally-relevant exposures to O<sub>3</sub>, concentration of exposure dominates the  
17 response. Studies evaluating C × T relationships have not been published recently. Other  
18 factors that have been studied for potential impact on the effects of O<sub>3</sub> include age, gender,  
19 nutritional status, genetic variability, exercise and exposure to co-pollutants. The effects of age  
20 on lung inflammation are not well known. After an acute exposure to 0.8 or 1 ppm, young mice,  
21 rats, and rabbits had greater changes in prostaglandins in BALF, but there were no age-  
22 dependent effects on BALF protein or cell number. Comparisons of male and female animals,  
23 and vitamin C or ascorbate deficiency did not reveal significant differences in the effects of O<sub>3</sub>,  
24 but exercise during exposure increased susceptibility.

25 Important new work has revealed that susceptibility to O<sub>3</sub> is, in part, genetically  
26 determined. Mouse strains with differing sensitivities to O<sub>3</sub> have identified genes on separate  
27 loci controlling various aspects of inflammation, providing additional evidence for the  
28 mechanistic separation of responses to O<sub>3</sub>. The research is summarized in Figure 5-4.  
29 Kleeberger's group has identified Inf-1, which modulates acute inflammatory responses; Inf-2  
30 which modulates responses to subacute exposures; and TNF-α and TNF receptors, which are  
31 involved in inflammatory responses. Other research groups have identified loci linked to other



**Figure 5-4. Mouse chromosomes on which genes or gene loci have been identified that modulate responses to O<sub>3</sub>.**

1 endpoints. This line of research provides a ground work for understanding the underlying  
 2 mechanisms of O<sub>3</sub>-induced injury, and can shed light on genes responsible for human  
 3 susceptibility to O<sub>3</sub>.

4 Recent studies have placed a major focus on mediators released from inflammatory cells to  
 5 understand the mechanisms of O<sub>3</sub>-induced inflammation and injury. Cytokines and chemokines  
 6 have been shown to be released as a result of stimulation or injury of macrophages, epithelial  
 7 cells and PMNs. Exposure of guinea pig AMs recovered in BALF and exposed *in vitro* to  
 8 0.4 ppm O<sub>3</sub> produced a significant increase in IL-6 and TNF $\alpha$ . An exposure of human AMs  
 9 to an identical O<sub>3</sub> concentration increased TNF $\alpha$ , IL-1 $\beta$ , IL-6 and IL-8. The expression  
 10 of MIP-2 mRNA or BALF levels of MIP-2 increased in mice and rats exposed to O<sub>3</sub>  
 11 concentrations  $\geq 1$  ppm. An increase after O<sub>3</sub> exposure has also been reported for other cytokines  
 12 and inflammatory mediators, including CINC and fibronectin. The CINC mRNA expression

1 was associated with neutrophilia at 24 hrs PE. Ozone exposure of mice also caused an increase  
2 in IL-6, MIP-1 $\alpha$  and eotaxin in mice. Further understanding of the role of mediators has  
3 come from studies utilizing antibodies and inhibitors of known specificity. In these studies  
4 treatment of rats with an anti IL-6 receptor antibody prior to a nighttime exposure to O<sub>3</sub>  
5 abolished O<sub>3</sub>-induced cellular adaptive response following a subsequent exposure. Studies  
6 utilizing antibodies to selected pro- or anti-inflammatory cytokines suggest a role of TNF $\alpha$ ,  
7 interleukin-10 (IL-10) and IL-1 $\beta$  in O<sub>3</sub>-induced changes in permeability, inflammation and  
8 cytokine release.

9 Studies investigating the mechanisms of PMN recruitment in the lung have explored the  
10 role of cell adhesion molecules that mediate PMN-endothelial cell interactions. An increase in  
11 tissue expression of ICAM-1 occurred in mice exposed to 0.8 ppm O<sub>3</sub>. A comparable pattern of  
12 time-related changes in total protein, fibronectin and alkaline phosphatase activity in the BALF  
13 was observed in rats exposed to 1 ppm O<sub>3</sub>. In monkeys, the O<sub>3</sub>-induced inflammation was  
14 blocked by treatment with a monoclonal antibody to CD18, suggesting dependence of PMN  
15 recruitment on this adhesion molecule. Together, these studies support the role of extracellular  
16 matrix protein and cell adhesion molecules in lung inflammation and injury.

17 Ozone exposure also affects macrophage functions, and consequently their role in lung  
18 inflammation. Macrophages isolated from O<sub>3</sub>-exposed mice produced increased amounts of  
19 NO, superoxide anion and PGE<sub>2</sub>, but production of these mediators by macrophages from  
20 NOS knockout mice was not elevated. Additionally, mice deficient in NOS or mice treated  
21 with N<sup>G</sup>-monomethyl-L-arginine, an inhibitor of total NOS, were protected from O<sub>3</sub>-induced  
22 permeability, inflammation and injury. These findings suggest a role of NO in the production  
23 of O<sub>3</sub> effects.

#### 24 25 **5.2.4 Morphological Effects**

26 Most mammalian species show generally similar morphological responses to <1 ppm O<sub>3</sub>,  
27 which differ only by region, cell type, exposure parameters, and length of time between exposure  
28 and examination. Constant low exposures to O<sub>3</sub> create an early bronchoalveolar exudation,  
29 which declines with continued exposure and drops in the PE period. Epithelial hyperplasia also  
30 starts early, increases in magnitude for several weeks, plateaus with continuing exposure, and  
31 declines slowly during PE. Interstitial fibrosis has a later onset, continues to increase throughout

1 the exposure, and can continue to increase after the exposure ends. Nonhuman primates respond  
2 more than rats at this concentration, due to differences in antioxidants, the CAR (predicted to  
3 receive the highest dose of O<sub>3</sub>), the presence of respiratory bronchioles, acinar volume, and  
4 differences in the nasal cavity's ability to "scrub" the O<sub>3</sub>. Ciliated epithelial cells of the airway,  
5 Type 1 epithelial cells of the gas-exchange region, and ciliated cells in the nasal cavity are the  
6 cells most affected by O<sub>3</sub>. Ozone-damaged ciliated cells are replaced by nonciliated cells (which  
7 are unable to provide clearance function) and Type 1 cells are replaced by Type 2 cells, which  
8 are thicker and produce more lipids. Inflammation also occurs, especially in the CAR, wherein  
9 the tissue is thickened as collagen accumulates. At exposures of 0.25 ppm O<sub>3</sub> (8 h/day, 18 mo)  
10 in monkeys, the distal airway is remodeled as bronchiolar epithelium replaces the cells present in  
11 alveolar ducts. In both rodents and monkeys, it appears that the natural seasonal patterns of O<sub>3</sub>  
12 exposure alters morphology more than continuous exposures, thus long-term animal studies with  
13 uninterrupted exposures may underestimate morphological effects.

#### 14 15 **5.2.4.1 Acute and Subchronic Exposure Effects**

16 Morphological effects of key acute and subchronic exposure studies are summarized in  
17 Table AX5-9. Harkema et al. (1997a) reviewed toxicological studies of the nasal epithelial  
18 response to short-term O<sub>3</sub>. New information regarding the effects of O<sub>3</sub> in this region include  
19 demonstrations that the topical anti-inflammatory corticosteroid fluticasone propionate prevents  
20 inflammation and mucous cell metaplasia in rats after cumulative O<sub>3</sub> exposure (0.5 ppm O<sub>3</sub>,  
21 8 h/day, for 3 or 5 days) (Hotchkiss et al., 1998). Exposure to bacterial endotoxin, a common  
22 ambient air toxicant, can potentiate mucous cell metaplasia in the nasal transitional epithelium of  
23 rats caused by a previous 3 day 0.5 ppm O<sub>3</sub> exposure (Fanucchi et al., 1998). Male F344/N Hsd  
24 rats were intranasally instilled with endotoxin after exposure to filtered air (FA) or 0.5 ppm O<sub>3</sub>,  
25 (8 h/d for 3 d). Mucous cell metaplasia was not found in the air/endotoxin group, but was found  
26 in the O<sub>3</sub>/saline group and was most severe in the O<sub>3</sub>/endotoxin group. A similar synergistic  
27 effect was demonstrated by Wagner et al. (2001a,b) with exposure of Fischer rats to O<sub>3</sub> for 8 h  
28 per day for 3 days and endotoxin. Ozone alone created epithelial lesions in the nasal transitional  
29 epithelium, while endotoxin alone caused lesions in the epithelium of the nose and conducting  
30 airways. The enhanced O<sub>3</sub>-induced mucous cell metaplasia was related to neutrophilic  
31 inflammation.

1 Pre-metaplastic responses, such as mucin mRNA upregulation, neutrophilic inflammation,  
2 and epithelial proliferation, were shown to be responsible for O<sub>3</sub>-induced mucous cell metaplasia  
3 in the transitional epithelium of rats (Cho et al., 1999a, 2000). Male F344/N rats exposed to O<sub>3</sub>  
4 (0.5 ppm, 8 h/d for 1, 2, or 3 d) demonstrated a rapid increase in an airway-specific mucin gene  
5 mRNA after exposure to O<sub>3</sub>, both before and during the onset of mucous cell metaplasia.  
6 Neutrophilic inflammation coincided with epithelial DNA synthesis and upregulation of  
7 rMuc-5AC, but was resolved before the development of epithelial metaplasia. The mucous cell  
8 metaplasia was neutrophil-dependent, whereas O<sub>3</sub>-induced epithelial cell proliferation and mucin  
9 gene upregulation were neutrophil-independent.

10 Dormans et al. (1999) compared the extent and time course of fibrotic changes in mice,  
11 rats, and guinea pigs exposed to 0.2 and 0.4 ppm O<sub>3</sub> for 3, 7, 28, and 56 days. They found a  
12 concentration-related centriacinar inflammation in all three species, with a maximum after  
13 3 days of exposure and total recovery within 3 days after exposure. Repair of O<sub>3</sub>-induced  
14 damage by removal of injured epithelial cells is enhanced by the influx of neutrophils (Hyde  
15 et al., 1999; Veseley et al., 1999b; Miller et al., 2001; see Section 5.2.3). A study examining the  
16 kinetics of early cellular responses to O<sub>3</sub> utilized bromodeoxyuridine to label s-phase cells  
17 (Hotchkiss et al., 1997). Labeling indices for rat nasal transitional epithelial cell DNA were  
18 greatest 20 to 24 h after O<sub>3</sub> (0.5 ppm for 8 h) exposure, and remained greater than control at  
19 36 h PE.

20 Very few published studies have explicitly explored susceptibility factors such as species,  
21 gender, age, antioxidant defense, acute and chronic airway disease, and exercise. Most typical  
22 laboratory species studied have qualitatively similar effects associated with O<sub>3</sub> exposure.  
23 Dormans et al. (1999) compared morphological, histological, and biochemical effects in the rat,  
24 mouse, and guinea pig following O<sub>3</sub> exposure and recovery in clean air. Wistar RIV:Tox male  
25 rats, NIH male mice, and Hartley Crl:(HA)BR male guinea pigs were continuously exposed to  
26 FA, 0.2, or 0.4 ppm for 3, 7, 28, and 56 days. Recovery from 28 days of exposure was studied at  
27 intervals of 3, 7, and 28 days PE. The mouse was the most sensitive as shown by a concentration  
28 and exposure-time dependent persistence of bronchiolar epithelial hypertrophy, elevated lung  
29 enzymes, and slow recovery from exposure. Exposure to the high dose for 56 d in both rats and  
30 guinea pigs caused increased amounts of collagen in ductal septa and large lamellar bodies in



1 Type II cells. The inflammatory response was greater in the guinea pig. Overall, the authors  
2 rated mice as most susceptible, followed by guinea pigs and rats.

3 Ferrets, monkeys, and rats were exposed to O<sub>3</sub> (1.0 ppm, 8 h) to compare airway effects  
4 Sterner-Kock et al. (2000). The ferrets and monkeys had similar epithelial necrosis and  
5 inflammation that was more severe than that found in rats. Because ferrets have a similar  
6 pulmonary structure as humans (e.g., well-developed respiratory bronchioles and submucosal  
7 glands), the authors concluded that the ferret would be a better model than rodents  
8 for O<sub>3</sub>-induced airway effects. Age susceptibility is dependent on the endpoint examined  
9 (see Chapter 4 for discussions of age-related differences in O<sub>3</sub> dosimetry). One study (Dormans  
10 et al., 1996) demonstrated that O<sub>3</sub>-induced centriacinar lesions are larger in younger rats than in  
11 older rats with exposures to 0.4 ppm for 1 to 7 days.

12 New studies have examined O<sub>3</sub>-induced morphological effects in compromised laboratory  
13 animals. Rats with endotoxin-induced rhinitis were more susceptible to mucous cell metaplasia  
14 in the nasal transitional epithelium caused by a 3 day exposure to 0.5 ppm O<sub>3</sub> (Cho et al., 1999b).  
15 Wagner et al. (2002) reported a similar O<sub>3</sub>-induced enhancement of inflammatory and epithelial  
16 responses associated with allergic rhinitis. Brown Norway rats were exposed to 0.5 ppm O<sub>3</sub>,  
17 8 h/day for 1 day or 3 consecutive days and then immediately challenged intranasally with either  
18 saline or ovalbumin (OVA). Multiple exposures to O<sub>3</sub> caused greater increases in  
19 mucosubstances produced in the nose by allergen challenge.

20 Recent research has focused on the concept of O<sub>3</sub> susceptible and nonsusceptible sites  
21 within the respiratory tract, including in situ antioxidant status and metabolic activity. Plopper  
22 et al. (1998) examined whether the variability of acute epithelial injury to short-term O<sub>3</sub>  
23 exposure within the tracheobronchial tree is related to local tissue doses of O<sub>3</sub> or to local  
24 concentrations of reduced glutathione (GSH). Adult male rhesus monkeys exposed to O<sub>3</sub> (0.4 or  
25 1.0 ppm for 2 h) demonstrated significant cellular injury at all sites, but the most damage, along  
26 with increased inflammatory cells, occurred in the proximal respiratory bronchiole.  
27 A significant reduction in GSH was found in the proximal bronchus at 0.4 ppm O<sub>3</sub> and in the  
28 respiratory bronchiole at 1.0 ppm O<sub>3</sub>. A significant decrease in the percent of macrophages,  
29 along with significant increases in the percent of neutrophils and eosinophils, and a doubling of  
30 total lavage protein, were found after exposure to 1.0 ppm O<sub>3</sub> only. The authors concluded that

1 the variability of local O<sub>3</sub> dose in the respiratory tract was related to inhaled O<sub>3</sub> concentration  
2 and was closely associated with local GSH depletion and with the degree of epithelial injury.

3 Plopper and colleagues (e.g., Watt et al., 1998; Paige et al., 2000a) explored the site-  
4 specific relationship between epithelial effects of O<sub>3</sub> exposure and the metabolism of  
5 bioactivated compounds within the respiratory tract of rats. The distribution of CYP2E1-  
6 dependent activity, measured with a selective substrate (p-nitrocatechol), was found to be  
7 highest in the distal bronchioles and minor daughter airways, and lower in the lobar bronchi and  
8 major daughter airways. Short-term O<sub>3</sub> exposure (1 ppm for 8 h) increased CYP2E1 activity in  
9 the lobar bronchi/major daughter airways only; however, long-term O<sub>3</sub> exposure (1 ppm for  
10 90 days) decreased CYP2E1 activity in the major and minor airways, further complicating the  
11 interpretation of O<sub>3</sub> effects based on concentration and duration of exposure and recovery. Rats  
12 treated i.p. with 1-nitronaphthalene, a pulmonary toxicant requiring metabolic activation, and  
13 exposed to 0.8 ppm O<sub>3</sub>, 8h/day for 90 days showed greater histopathologic and morphometric  
14 effects in the CAR of the lung (Paige et al., 2000b). Despite reported tolerance to oxidant stress  
15 after long-term O<sub>3</sub> exposure, there was increased severity of ciliated cell toxicity.

#### 16 17 **5.2.4.2 Summary of Acute and Subchronic Morphological Effects**

18 Short-term exposures to O<sub>3</sub> cause similar alterations in lung structure in a variety of  
19 laboratory animal species at concentrations of 0.15 ppm in rats and lower concentrations in  
20 primates. Cells in the CAR are the primary targets of O<sub>3</sub>, but ciliated epithelial cells in the nasal  
21 cavity and airways and Type 1 epithelial cells in the gas exchange region are also targets. New  
22 work has shown that a topical anti-inflammatory corticosteroid can prevent these effects in nasal  
23 epithelia, while exposure to bacterial endotoxin can potentiate the effects. Ozone-induced  
24 fibrotic changes in the CAR are maximal at 3 d of exposure and recover 3 d PE with exposures  
25 of 0.2 ppm in rodents. New studies of susceptibility factors demonstrated that ferrets and  
26 monkeys have similar inflammatory and necrotic responses to 1 ppm O<sub>3</sub>, which differs from  
27 lesser injury seen in rats. Rats with induced allergic rhinitis are more susceptible to 0.5 ppm  
28 than are controls. Important new work has demonstrated variability of local O<sub>3</sub> dose and  
29 subsequent injury in the RT due to depletion of GSH. The proximal respiratory bronchiole  
30 receives the most acute epithelial injury from exposures ≤ 1 ppm, while metabolic effects were  
31 greatest in the distal bronchioles and minor daughter airways.

### 5.2.4.3 Subchronic and Chronic Exposure Effects

Summaries of new studies of morphological effects of subchronic and chronic exposures are listed in Table AX5-10 in Annex AX5. In general, as the duration of exposure lengthens, there is not a concomitant linear increase in the intensity of effect of a given endpoint. Rather, as exposure proceeds past 1 week to 1 year, Type 1 cell necrosis and inflammatory responses generally decrease to near control values, and hyperplastic and fibrotic changes remain elevated. After long-term exposure ended, some indicies of fibrosis persisted and in some cases became more severe during PE periods in clean air.

Effects of O<sub>3</sub> on the upper respiratory tract of F344 rats exposed to O<sub>3</sub> (0.12, 0.5, or 1.0 ppm for 20 months) included marked mucous cell metaplasia in the rats exposed to 0.5 and 1.0 pm O<sub>3</sub>, but not at 0.12 ppm O<sub>3</sub> (Harkema et al., 1997a). In a follow-up study, hyperplasia was found in the nasal epithelium of rats exposed to 0.25 and 0.5 ppm, 8h/day, 7 days/week, for 13 weeks (Harkema et al., 1999). The mucous cell metaplasia, and associated intraepithelial mucosubstances, induced by 0.5 ppm O<sub>3</sub> persisted for 13 weeks after exposure. An acute (8 h) exposure to 0.5 ppm O<sub>3</sub> 13 weeks after the chronic exposure induced an additional increase of mucosubstances in the nasal epithelium of rats but not in rats chronically exposed to 0 or 0.25 ppm O<sub>3</sub>. The persistent nature of the O<sub>3</sub>-induced mucous cell metaplasia in rats reported in this study suggests that O<sub>3</sub> exposure may have the potential to induce similar long-lasting alterations in the airways of humans.

No significant changes in nasal tissue were seen in rats continuously exposed for 49 days to the ambient air of Mexico City, Mexico (Moss et al., 2001). A rat study using 6-month exposures to ambient air of Sao Paulo, with a disparate pollutant composition than that of Mexico City, demonstrated development of secretory hyperplasia in rats (Lemos et al., 1994). However, without information on differences in ambient pollution composition in the two cities, the studies cannot be compared. Because of the persistent nature of these changes in the controlled studies with rats, and the fact that the upper airways of humans are probably more sensitive, like the monkey, the authors suggested that long-term exposure to ambient levels of O<sub>3</sub> could induce significant nasal epithelial lesions that may compromise the upper respiratory tract defense mechanisms of exposed human populations.

Rats exposed to 0.5 ppm O<sub>3</sub> for 1 month exhibited Bcl-2 in protein extracts of nasal epithelium (Tesfaigzi et al., 1998). Further, after 3 and 6 months of exposure, the number of

1 metaplastic mucous cells in the transitional epithelium was indirectly related to the percentage of  
2 cells that were Bcl-2 positive . Cells from rats exposed to FA did not express any Bcl-2. This  
3 study suggests that apoptosis regulators like Bcl-2 may play a role in the development and  
4 resolution of mucous cell metaplasia in the nasal airway.

5 A spectrum of lesions was reported (Herbert et al., 1996) in the nasal cavity and  
6 centriacinar lung of male and female mice exposed to 0.5 or 1.0 ppm of O<sub>3</sub> for 2 years, which  
7 persisted with continued exposure for 30 months. These lesions included bone loss in the  
8 maxilloturbinates, mucosal inflammation, mucous cell metaplasia in the nasal transitional  
9 epithelium and increased interstitial and epithelial thickening in the proximal alveolar region.  
10 In the CAR, there were increased numbers of nonciliated cells. However, changes in other  
11 endpoints including lung function and lung biochemistry were not evident. The investigators'  
12 interpretation of the entire study is that rodents exposed to the two higher O<sub>3</sub> concentrations had  
13 some structural hallmarks of chronic airway disease in humans.

14 A chronic study using a simulated, seasonal O<sub>3</sub>-exposure pattern was reported by Plopper  
15 and colleagues (Evans et al., 2003; Schelegle et al., 2003a; Chen et al., 2003; Plopper and  
16 Fanucchi, 2000). Infant rhesus monkeys (30 days old) were exposed to FA, house dust mite  
17 allergen aerosol (HDMA), or O<sub>3</sub> + HDMA. The 0.5 ppm O<sub>3</sub> exposures were 8 h/day for 5 days,  
18 every 14 days for a total of 11 O<sub>3</sub> episodes. Half of the monkeys were sensitized to house dust  
19 mite allergen (*Dermatophagoides farinae*) at 14 and 28 days of age. The sensitized monkeys  
20 were exposed to HDMA for 2 h/day on Days 3-5 of the FA or O<sub>3</sub> exposures. The lungs were  
21 removed during the last FA exposure and the right and left cranial and right middle lobes were  
22 separately inflation fixed. Microdissection and morphometric analyses were performed on the  
23 conducting airways to the level of the most proximal respiratory bronchiole. Repeated  
24 exposures to O<sub>3</sub> or O<sub>3</sub> + HDMA over a 6-month period resulted in an atypical development of  
25 the basement membrane zone of airways in nonsensitized developing monkeys. Remodeling in  
26 the distal conducting airways was found in the sensitized monkeys as a result of the damage and  
27 repair processes occurring with repeated exposure (Evans et al., 2003; Schelegle et al., 2003a).  
28 Lung function changes in these monkeys (Schelegle et al., 2003b), and associated adaptation of  
29 the respiratory motor responses (Chen et al., 2003), are described in Section 5.2.5.2.  
30 Collectively, these findings provide a pathophysiologic basis for changes in airway function

1 described in children growing up in polluted metropolitan areas (e.g., Tager, 1999)  
2 (see Chapter 7).

3 Necropsy of the left caudal lobe of these infant monkeys showed accumulation of  
4 eosinophils and mucous cells within the combined epithelial and interstitial compartments in the  
5 conducting airways and in the terminal/respiratory bronchioles (Schelegle et al., 2003a) . House  
6 dust mite sensitization and HDMA challenge alone, or combined with O<sub>3</sub> exposure, resulted in  
7 significantly greater eosinophil accumulation in the conducting airways when compared to FA  
8 and O<sub>3</sub> only exposures. A significant accumulation of eosinophils was found in the  
9 terminal/respiratory bronchioles of the sensitized monkeys challenged with HDMA when  
10 compared to monkeys exposed to FA, O<sub>3</sub>, and HDMA + O<sub>3</sub>. The mean mass of mucous cells  
11 increased in the fifth generation conducting airways of sensitized animals challenged with  
12 HDMA alone and when combined with O<sub>3</sub> exposure, and in the terminal bronchioles of  
13 sensitized animals exposed to HDMA + O<sub>3</sub>. The tracheal basement membrane of HDMA-  
14 sensitized monkeys exposed to HDMA or to HDMA + O<sub>3</sub> was significantly increased over  
15 controls; however, there were no significant changes in the airway diameter of proximal and  
16 mid-level airways. Exposures of sensitized young monkeys to HDMA alone, or to O<sub>3</sub> alone,  
17 resulted in eosinophilia of the mid-level conducting airways and the terminal/respiratory  
18 bronchioles, but without alterations in airway structure or function. The authors interpreted  
19 these findings to indicate that the combination of cyclic O<sub>3</sub> exposure and HDMA challenge in  
20 HDMA-sensitized infant monkeys act synergistically to produce an allergic-reactive airway  
21 phenotype characterized by significant eosinophilia of midlevel conducting airways,  
22 transmigration of eosinophils into the lumen, and an altered structural development of  
23 conducting airways that is associated with increased airway resistance and nonspecific airway  
24 reactivity (see Section 5.2.5).

25 Examination of development of the tracheal basement membrane zone (BMZ) in these  
26 monkeys (Evans et al., 2003) showed that with exposures to either O<sub>3</sub> or HDMA + O<sub>3</sub>, BMZ  
27 development was affected. Abnormalities in the BMZ included: (1) irregular and thin collagen  
28 throughout the BMZ; (2) perlecan depleted or severely reduced; (3) FGFR-1 immunoreactivity  
29 reduced; (4) FGF-2 immunoreactivity absent in perlecan-deficient BMZ, but present in the  
30 lateral intercellular space (LIS), in basal cells, and in attenuated fibroblasts; (5) syndecan-4  
31 immunoreactivity increased in basal cells. The authors interpret these data as suggesting that O<sub>3</sub>

1 targets cells are associated with synthesis of epithelial BMZ perlecan. The absence of FGF-2,  
2 normally stored in the BMZ, could affect downstream signaling in airway epithelium and could  
3 be responsible for the abnormal development of the airway seen in this study, and thus be an  
4 important mechanism modulating O<sub>3</sub>-induced injury. Midlevel bronchi and bronchioles from  
5 these monkeys (Larson et al., 2004) demonstrated decrements in the density of epithelial nerves  
6 in the axial path between the sixth and seventh airway generations in exposures to O<sub>3</sub>.  
7 Combined O<sub>3</sub> + HDMA exposures exacerbated this reduction. They attribute this loss of nerve  
8 plexuses to neural regression or stunted nerve development, the latter corroborated by the Evans  
9 et al. (2003) finding of decreased growth factors following O<sub>3</sub> exposure. Additionally, they  
10 found streaks or clusters of cells immunoreactive for protein gene product 9.5 (PGP 9.5, a pan-  
11 neuronal marker) and negative for calcitonin gene-related peptide. The functional significance  
12 of this is unknown but suggests to the authors a possible injury-repair process induced by O<sub>3</sub>.

13 Remodeling of the distal airways and CAR is one of the most disturbing aspects of the  
14 morphological changes occurring after subchronic and chronic exposure to O<sub>3</sub>. Recently,  
15 bronchiolization was reported in rats exposed to 0.4 ppm O<sub>3</sub> for only 56 days (van Bree et al.,  
16 2001). They also found collagen formation progressively increased with increasing O<sub>3</sub> exposure  
17 and persisted into PE recovery. In addition to centriacinar remodeling, Pinkerton et al. (1998)  
18 reported thickening of tracheal, bronchial, and bronchiolar epithelium after 3 or 20 months  
19 exposure to 1 ppm O<sub>3</sub>, but not to 0.12 ppm. Although some older literature had reported that  
20 chronic exposures to ≤1.0 ppm O<sub>3</sub> cause emphysema, none of the more recent literature supports  
21 this hypothesis.

#### 22 23 **5.2.4.4 Summary and Conclusions - Subchronic and Chronic Morphological Effects**

24 The progression of effects during and after a chronic exposure at a range of 0.5 to 1.0 ppm  
25 is complex, with inflammation peaking over the first few days of exposure, then dropping, then  
26 plateauing, and finally, largely disappearing. Epithelial hyperplasia follows a somewhat similar  
27 pattern. Effects of 0.5 ppm O<sub>3</sub> for 20 months on the nasal mucosa include atrophy of nasal  
28 turbinates and mucous cell metaplasia, which persisted long after the exposure ceased. Fibrotic  
29 changes in the tissue increase very slowly over months of exposure, and, after exposure ceases,  
30 the changes sometimes persist or increase. The pattern of exposure in this same concentration  
31 range determines effects, with 18 mo of daily exposure causing less morphologic damage than

1 exposures on alternating months. This is important, given that environmental O<sub>3</sub> exposure is  
2 typically seasonal. Plopper and colleagues' long term study of infant rhesus monkeys exposed to  
3 simulated, seasonal O<sub>3</sub> (0.5 ppm 8 h/day for 5 days, every 14 days for 11 episodes)  
4 demonstrated: (1) remodeling in the distal airways; (2) abnormalities in tracheal basement  
5 membrane; (3) eosinophil accumulation in conducting airways; and (4) decrements in airway  
6 innervation. These findings advance earlier information regarding possible injury-repair  
7 processes occurring with seasonal O<sub>3</sub> exposures.  
8

## 9 **5.2.5 Effects on Pulmonary Function**

### 10 **5.2.5.1 Acute and Subchronic Exposure Effects on Pulmonary Function**

11 Numerous pulmonary function studies of the effects of acute O<sub>3</sub> exposure (defined here  
12 as ≤1 week of exposure) in several animal species have been conducted and generally show  
13 responses similar to those of humans (e.g., increased breathing frequency, decreased tidal  
14 volume, increased resistance, decreased forced vital capacity (FVC) and changes in the  
15 expiratory flow-volume curve). These effects are seen at 0.25 to 0.4 ppm O<sub>3</sub> for several h in a  
16 number of species. At concentrations of ≥1 ppm, breathing mechanics (compliance and  
17 resistance) are affected. The breathing pattern returns to normal after O<sub>3</sub> exposure. In rats  
18 exposed to 0.35 to 1 ppm O<sub>3</sub> for 2 h/day for 5 days, there was a pattern of attenuation of  
19 pulmonary function responses similar to that observed in humans. Concurrently, there was no  
20 attenuation of biochemical indicators of lung injury or of morphological changes.

21 Work demonstrating attenuation of pulmonary functions (see Table AX5-11) was  
22 completed by Wiester et al. (1996) who exposed male Fischer 344 rats to 0.5 ppm O<sub>3</sub> for either  
23 6 or 23 h/day over 5 days. Ozone-induced changes in lung volume were attenuated during the  
24 5 exposure days and returned to control levels after 7 days recovery. The responses to repeated  
25 O<sub>3</sub> exposure in rats were exacerbated by reduced ambient temperature, presumably as a result of  
26 increased metabolic activity.

27 Researchers have utilized inbred mouse strains with varying ventilatory responses to O<sub>3</sub> to  
28 attempt to model susceptible populations. As differences were seen in inflammatory responses  
29 to acute O<sub>3</sub> exposures in C57BL/6J and C3H/HeJ mice, comparisons were made of their  
30 ventilatory responses also (Tankersley et al., 1993). Following an exposure of 2 ppm O<sub>3</sub> for 3 h,

1 breathing frequency ( $f$ ), tidal volume ( $V_T$ ), and minute ventilation were measured 1 and 24 h in  
2 both normocapnia (or air at  $\sim 0\%$   $\text{CO}_2$ ) and hypercapnia (5 or 8%  $\text{CO}_2$ ). They demonstrated that  
3 acute  $\text{O}_3$  exposures caused altered hypercapnic ventilatory control, which varied between strains.  
4 This suggested to the authors that  $\text{O}_3$ -induced alterations in ventilation are determined, at least in  
5 part, by genetic factors. A caveat regarding studies such as this using high exposure  
6 concentrations is that events observed at high concentrations may differ from those observed at  
7 near-ambient  $\text{O}_3$  levels.

8 Paquette et al. (1994) measured ventilatory responses in C57BL/6J and C3H/HeJ mice  
9 given repeated acute exposures of 0.3 ppm for 48 and 72 h. The two strains had differing  
10 responses to both normocapnia and hypercapnia. Normocapnic  $V_E$  was greater following  
11 subacute  $\text{O}_3$  exposure in C57BL/6J mice than in C3H/HeJ mice, due to increased  $f$  and  
12 reduced  $V_T$ , respectively. This suggests that the increased  $V_T$  in C57BL/6J mice may contribute  
13 to the increased susceptibility to lung injury due to a greater dose of  $\text{O}_3$  reaching the lower lung.  
14 Hypercapnic ventilatory responses following subacute  $\text{O}_3$  exposures demonstrated reduced  $V_E$   
15 (due to decreased  $V_T$ ) in C57BL/6J only. Evaluations of  $\text{O}_3$  dosimetry were performed in these  
16 two strains using  $^{18}\text{O}_3$ -labeled ozone (2 ppm for 2-3 h) (Slade et al., 1997). Immediately after  
17 exposures of 2 ppm  $^{18}\text{O}_3$  for 2-3 h, C3H/HeJ mice had 46% less  $^{18}\text{O}$  in lungs and 61% less in  
18 trachea, than C57BL/6J. Additionally, C3H/HeJ mice had a greater body temperature decrease  
19 following  $\text{O}_3$  exposure than C57BL/6J mice, suggesting that the differences in susceptibility  
20 to  $\text{O}_3$  are due to differences the ability to decrease body temperature and, consequently decrease  
21 the dose of  $\text{O}_3$  to the lung.

22 Tracheal transepithelial potential has also been shown to differ in eight mouse strains 6 h  
23 after exposure to 2 ppm  $\text{O}_3$  for 3 h (Takahashi et al., 1995b). AKR/J, C3H/HeJ, and CBA/J were  
24 identified as resistant strains and 129/J, A/J, C57BL/6J, C3HeB/FeJ and SJL/J were identified as  
25 susceptible strains. The authors noted that strains' responses to this parameter did not show  
26 concordance with inflammatory responses, suggesting to the authors that the two phenotypes are  
27 not controlled by the same genetic factors.

28 Savov et al. (2004) characterized ventilatory responses in nine mouse strains exposed to  $\text{O}_3$   
29 (2.0 ppm  $\text{O}_3$  for 3 h). C57BL/6J was hyporeactive to MCh prior to  $\text{O}_3$ , but was very responsive  
30 to MCh following  $\text{O}_3$ . Conversely, C3H/HeJ had an intermediate baseline  $P_{\text{enh}}$  and a small



1 response to MCh following O<sub>3</sub> exposure. This study corroborates the evidence of no consistent  
2 relationship between respiratory P<sub>enh</sub> and inflammation.

### 3 4 **5.2.5.2 Summary and Conclusions - Acute and Subchronic Effects on** 5 **Pulmonary Function**

6 Early work has demonstrated that during acute exposure of ~0.2 ppm O<sub>3</sub> in rats, the most  
7 commonly observed alterations are increased frequency of breathing and decreased tidal volume  
8 (i.e., rapid, shallow breathing). Exposures of ~1.0 ppm O<sub>3</sub> affect breathing mechanics  
9 (compliance and resistance). Additionally, decreased lung volumes are observed in rats with  
10 acute exposures at levels of 0.5 ppm. New work utilizing inbred mouse strains with varying  
11 ventilatory responses to O<sub>3</sub> has suggested that: (1) control of the ventilatory response is  
12 determined, at least in part, by genetic factors; (2) increased V<sub>T</sub> in some strains may contribute  
13 to lung injury due to a greater dose of O<sub>3</sub> reaching the lower lung; (3) some strains' ability to  
14 reduce body temperature may account for their decreased O<sub>3</sub>-induce lung injury; and (4) tracheal  
15 transepithelial potential is determined, in part, by genetic factors. Importantly, the genetic loci  
16 that appear to be modulating various aspects of pulmonary responses to O<sub>3</sub> differ from each  
17 other and from loci controlling inflammatory responses.

18 Exposures of 2 h/day for 5 days create a pattern of attenuation of pulmonary function in  
19 both rats and humans without concurrent attenuation of lung injury and morphological changes,  
20 indicating that the attenuation did not result in protection against all the effects of O<sub>3</sub>.  
21 Chronic O<sub>3</sub> exposure studies evaluating pulmonary function are not available. Earlier work  
22 has demonstrated that repeated daily exposure of rats to an episodic profile of O<sub>3</sub> caused small,  
23 but significant decrements in lung function that were consistent with early indicators of focal  
24 fibrogenesis in the proximal alveolar region, without overt fibrosis.

### 25 26 **5.2.5.3 Ozone Effects on Airway Responsiveness**

27 Effects of O<sub>3</sub> on airway reactivity have been observed in a variety of species at an exposure  
28 range of 0.5 to 1 ppm. Many of the new studies on pulmonary function in laboratory animals  
29 allow a better prediction of the effects of O<sub>3</sub> exposure on the exacerbation of asthma symptoms  
30 and the risk of developing asthma in humans. However, it is necessary to understand the factors  
31 that determine airway responsiveness across different mammalian species as discussed in  
32 Chapter 4.

1 Traditional studies of airway responsiveness require sedation in both infants and laboratory  
2 animals. Laboratory animal studies employ intravenous agonist challenges as well as inhalation  
3 challenges, though inhaled agonist challenges are preferred in humans. Exercise testing is  
4 not possible with sedation unless exercise is “simulated” by increasing ventilation using  
5 elevated  $F_iCO_2$ ; and the need for artificial ventilation in laboratory animal studies may cause  
6 breathing patterns that affect  $O_3$  deposition. Joad et al. (2000) reported that when 1 ppm  $O_3$  for  
7 90 min is administered to isolated rat lung at either 2.4 mL/40 bpm or 1.2 mL/80 bpm, the more  
8 rapid breathing pattern elicits less epithelial cell injury than the slower breathing pattern.  
9 Though this study design does not really model rapid shallow breathing elicited in the intact  
10 animal, it shows greater reduction in injury in the proximal axial airway compared to its adjacent  
11 airway branch and terminal bronchiole. The rapid, shallow breathing pattern protects the large  
12 conducting airways of rats, but causes a more even distribution of epithelial cell injury to the  
13 terminal bronchioles (Schelegle et al., 2001). Postlethwait et al. (2000) demonstrated that the  
14 conducting airways are the primary site of acute cytotoxicity from  $O_3$  exposure. Three-  
15 dimensional mapping of the airway tree in SD rat isolated lung exposed to 0, 0.25, 0.5, or  
16 1.0 ppm  $O_3$  showed a concentration-dependent increase in injured cells. Injury was evident in  
17 proximal and distal conduction airways, lowest in terminal bronchioles, and highest in the small  
18 side branches downstream of bifurcations. These exposure levels did not concurrently elicit  
19 changes in LDH activity or total protein in BALF, suggesting that the mapping technique is a  
20 more sensitive measure of injury and is useful in dosimetry studies.

21 Whole-body plethysmography of unanesthetized, unrestrained rodents has been used to  
22 indirectly measure pulmonary resistance (Shore et al., 2002; Goldsmith et al., 2002; Jang et al.,  
23 2002). However, these indices of inspiratory/expiratory pressure differences, including  
24 enhanced pause ( $P_{enh}$ ) may be less sensitive than direct measurements of lung airflow resistance  
25 (Murphy, 2002). Changes in airway structure caused by viral infections also must be considered  
26 when evaluating laboratory animal studies. Animals with acute viral illness have morphological  
27 evidence of inflammatory cell infiltration, bronchiolar wall edema, epithelial hyperplasia, and  
28 increased airway mucous plugs that can cause airway narrowing, air trapping, and serious  
29 functional changes in the lung (Folkerts et al., 1998).

30 Exercise-induced bronchoconstriction in humans appears to be mediated by changes in the  
31 tonicity of the airway lining fluid (Anderson and Daviskas, 2000). Brannan et al. (1998) suggest

1 that a test in laboratory animals based on the inhalation of mannitol aerosol (hyperosmolar)  
2 might be feasible and provide information similar to that from exercise challenges in cooperative  
3 children and adults. Unfortunately, there have been few reports of mannitol or adenosine  
4 monophosphate challenges in laboratory animals; most studies have utilized histamine,  
5 methacholine, acetylcholine, or carbachol to determine outcome. In active humans with asthma,  
6 adenosine monophosphate challenges appear to better reflect ongoing airway inflammation than  
7 histamine or methacholine challenges (Polosa and Holgate, 1997; Avital et al, 1995a,b), and  
8 might be useful in identifying mechanisms of asthma in laboratory animals and their  
9 responsiveness to environmental pollutants.

10 The increased responsiveness to bronchoconstrictor challenge in asthma is thought to result  
11 from a combination of structural and physiological factors that include increased inner-wall  
12 thickness, increased smooth-muscle responsiveness, and mucus secretion. These factors also  
13 are likely to determine a level of innate airway responsiveness that is genetically influenced.  
14 Chapter 6 (Section 6.8) discusses cellular and biochemical changes that have been identified  
15 in human asthmatics. These studies suggest that the mechanisms involved in AHR are  
16 multifactorial, with general agreement that there is an inconsistent relationship between AHR  
17 and markers of inflammation.

18 A large data base of laboratory animal research has been collected on the role of O<sub>3</sub> in  
19 producing an increase in AHR (see Table AX5-12). Exposure levels ( $\geq 1$  ppm for  $\geq 30$  min)  
20 in many of these studies are not environmentally relevant, but information may be obtained  
21 regarding the mechanisms of action of O<sub>3</sub> concerning: O<sub>3</sub> concentration and peak response time,  
22 inhaled versus intravenous challenge with nonspecific bronchoconstrictors, neurogenic  
23 mediation, neutrophilic inflammation, and interactions with specific biological agents (e.g.,  
24 antigens and viruses). However, as with other toxicants, high-dose and low-dose mechanisms  
25 may differ, so interpretation of results must take this into consideration.

26 Many species of laboratory animals have been used to study the effects of O<sub>3</sub> on airway  
27 bronchoconstriction. Ozone-induced AHR in guinea pigs has been used to model human  
28 bronchospasm (van Hoof et al., 1996; 1997a,b; Matsubara et al., 1997a,b; Sun and Chung, 1997;  
29 Aizawa et al., 1999a,b; Tsai et al., 1998; Nakano et al., 2000). Because these studies were done  
30 at 2 to 3 ppm O<sub>3</sub>, these results are not directly relevant for extrapolation to potential airway  
31 responses in humans exposed to ambient levels of O<sub>3</sub>. Humans with reactive airway disease

1 (e.g., asthma) appear to be sensitive to ambient levels of O<sub>3</sub> (see Chapters 6 and 7) and the  
2 current understanding is that O<sub>3</sub> exacerbates airway responsiveness to specific allergens,  
3 presumably by nonspecifically increasing AHR.

4 Shore et al. (2000, 2002) have shown that O<sub>3</sub>-induced AHR is reduced in immature rats and  
5 mice. SD rats exposed to 2 ppm O<sub>3</sub> at ages 2, 4, 6, 8, or 12 weeks and A/J mice exposed to 0.3 to  
6 3 ppm for 3 h at age 2, 4, 8, or 12 weeks had similar concentration-related decreases in V<sub>E</sub> except  
7 at the youngest ages. This smaller decrement in V<sub>E</sub> suggested a delivered dose that was much  
8 greater in the younger animals. This group (Shore et al., 2003) has also recently shown that  
9 obese mice have greater ventilatory responses to O<sub>3</sub>. Exposures of 2.0 ppm O<sub>3</sub> for 3 h to lean,  
10 WT C57BL/6J and *ob/ob* mice (mice with a genetic defect in the coding for leptin, the satiety  
11 hormone) showed that the *ob/ob* mice had enhanced AHR and inflammation compared to the  
12 WT mice. These data correlate with epidemiological data showing increased incidence of  
13 asthma in overweight children.

14 Increased AHR to various nonspecific bronchoconstrictive agents (e.g., ACh,  
15 methacholine, histamine, carbachol) given by inhalation or intravenous routes has been  
16 previously shown in laboratory animals exposed to O<sub>3</sub> concentrations ≤1.0 ppm. Dye et al.  
17 (1999) showed hyperresponsiveness to methacholine in rats 2 h after exposure to 2 ppm O<sub>3</sub> for  
18 2 h. AHR can be induced by specific antigens as well as O<sub>3</sub>. The most commonly used  
19 laboratory animal model is the OVA sensitized guinea pig. Animals sensitized with OVA have  
20 been shown to have similar responses to nonspecific bronchoconstrictors as control animals.

21 OVA-sensitized guinea pigs (Sun et al., 1997) and mice (Yamauchi et al., 2002) were used  
22 to determine the enhancement of antigen-induced bronchoconstriction by acute, high-level O<sub>3</sub>  
23 (1.0 ppm O<sub>3</sub> for 1 h). Male Dunkin-Hartley guinea pigs were sensitized by i.p. injection of OVA  
24 and exposed to O<sub>3</sub> alone, OVA aerosol, or O<sub>3</sub> + OVA. Ozone exposure alone increased  
25 bronchial responsiveness to ACh at 3 h, but not 24 h, while OVA alone had no effect. Combined  
26 exposure to O<sub>3</sub> and OVA (1 ppm for 1 h, then 3 min OVA) increased bronchial responsiveness to  
27 ACh 3 h after O<sub>3</sub> exposure. At 24 h following O<sub>3</sub> exposure, AHR increased when OVA  
28 challenge was performed at 21 h, suggesting that O<sub>3</sub> pre-exposure can potentiate OVA-induced  
29 AHR. Neutrophil counts in the BALF increased at 3 and 24 h after O<sub>3</sub> exposure alone but were  
30 not further increased when O<sub>3</sub> exposure was combined with OVA airway challenge; however

1 protein content of the BALF did increase at 3 and 24 h in the O<sub>3</sub> and OVA groups. Thus, this  
2 study also indicates that high-ambient O<sub>3</sub> exposure can augment antigen (OVA)-induced AHR in  
3 guinea pigs.

4 Yamauchi et al. (2002) sensitized male C57BL/6 mice by i.p. injection of OVA and then  
5 exposed them to O<sub>3</sub>. The sensitized mice had AHR to methacholine. Ozone exposure caused  
6 significant decreases in dynamic lung compliance, minute ventilation, and P<sub>a</sub>O<sub>2</sub> in OVA-  
7 sensitized mice, but not in controls. A marker of inflammation (soluble intercellular adhesion  
8 molecule-1 [sICAM-1]) was elevated in the BAL fluid of OVA-sensitized mice, but sICAM-1  
9 levels were not significantly changed by O<sub>3</sub> exposure, indicating that the O<sub>3</sub>-induced AHR to  
10 methacholine was not caused by O<sub>3</sub>-induced inflammation.

11 Ozone-induced AHR may be temporally associated with inflammatory cells stimulated by  
12 cytokines (Koto et al., 1997), mast cells (Igarashi et al., 1998; Noviski et al., 1999), or by oxygen  
13 radicals (Takahashi et al., 1993). One study, however, has shown that inflammation is not a  
14 prerequisite of AHR (Koto et al., 1997), and it has been suggested that O<sub>3</sub>-induced AHR may be  
15 epithelium dependent (McGraw et al., 2000). For example, neonatal rats pretreated with  
16 capsaicin, which will permanently destroy C-fibers and prevent O<sub>3</sub>-induced (1 ppm, 8 h) release  
17 of neuropeptides (Vesely et al., 1999a), and then exposed to O<sub>3</sub> when adults, showed a marked  
18 increase in airway responsiveness to inhaled aerosolized methacholine (Jimba et al., 1995).  
19 Takebayashi et al. (1998) has shown that depletion of tachykinins by capsaicin treatment, or by a  
20 specific tachykinin receptor antagonist, can block the induction of AHR by O<sub>3</sub>. The seemingly  
21 disparate responses in laboratory animals may be due to species- or strain-specific differences in  
22 inherent reactivity to bronchoconstrictors, or to inherent differences in susceptibility to O<sub>3</sub>-  
23 induced inflammation (Zhang et al., 1995; Depuydt et al., 1999; Dye et al., 1999).

24 Studies that may be potentially relevant to ambient levels of O<sub>3</sub> were conducted in vivo, in  
25 an isolated perfused lung model, and in ex vivo lung segments using multihour and repeated  
26 multihour exposures with ambient levels of O<sub>3</sub>. A study on the relationship between O<sub>3</sub>-induced  
27 AHR and tracheal epithelial function was conducted in New Zealand white rabbits by Freed  
28 et al. (1996). Rabbits exposed to O<sub>3</sub> (0.2 ppm for 7 h) demonstrated significantly decreased  
29 tracheal transepithelial potential difference but no changes in lung resistance. Changes in the  
30 compartmentalized lung resistance, measured in response to ACh challenge before and after  
31 bilateral vagotomy, were not significantly different in air-exposed rabbits; however, bilateral

1 vagotomy did enhance peripheral lung reactivity in O<sub>3</sub>-exposed rabbits. The ACh-induced  
2 a 140% increase in lung resistance with O<sub>3</sub> exposure was two times higher than with air  
3 exposure, indicating that ambient-level O<sub>3</sub> exposure affects tracheal epithelial function in rabbits  
4 and increases central airway reactivity, possibly through vagally-mediated mechanisms.

5 Pulmonary mechanics and hemodynamics were studied in the New Zealand white rabbit  
6 isolated perfused lung model that allowed partitioning of the total pressure gradient into arterial,  
7 pre- and post-capillary, and venous components (Delaunois et al., 1998). Exposures to O<sub>3</sub>  
8 (0.4 ppm for 4 h) were followed by evaluation of airway responsiveness to ACh, substance P  
9 (SP), or histamine immediately or 48 h later. Ozone inhibited pulmonary mechanical reactivity  
10 to all three bronchoconstrictors that persisted for 48 h and modified vasoreactivity of the  
11 vascular bed, but only at 48 h PE. Arterial segmental pressure, normally insensitive to ACh and  
12 SP, was significantly elevated by O<sub>3</sub>; precapillary segmental pressure decreased in response to  
13 Ach, suggesting that O<sub>3</sub> can induce direct vascular constriction, but the vascular responses are  
14 variable and depend on the agonist used and on the species studied.

15 Airway responsiveness to the same three compounds was evaluated by Segura et al. (1997)  
16 in guinea pigs exposed to O<sub>3</sub> (0.15, 0.3, 0.6, or 1.2 ppm for 4 h). Ozone did not cause AHR to  
17 ACh or histamine, except at the highest concentration (1.2 ppm O<sub>3</sub>) for histamine. However, O<sub>3</sub>  
18 did cause AHR to SP at ≥0.3 ppm, suggesting that O<sub>3</sub> destroys neutral endopeptidases  
19 (responsible for SP inactivation) in airway epithelial cells. Vargas et al. (1998), in a follow-up  
20 study, demonstrated that guinea pigs chronically exposed to 0.3 ppm O<sub>3</sub> for 4 h/day became  
21 adapted to SP-induced AHR. Ozone caused increased sensitivity to SP after 1, 3, 6, 12, and  
22 24 days of exposure that was associated with airway inflammation; however, after 48 days of  
23 exposure, the increased sensitivity to SP was lost.

24 This study is in accordance with Szarek et al. (1995) who demonstrated that AHR  
25 associated with acute O<sub>3</sub> exposures does not persist during long-term exposure to near-ambient-  
26 levels of O<sub>3</sub> (≤1 ppm). Fischer 344 rats, exposed to 0.0, 0.12, 0.5, or 1.0 ppm O<sub>3</sub>, 6 h/day,  
27 5 days/week for 20 months, demonstrated significantly reduced responses to bethanechol, ACh,  
28 and electrical field stimulation in eighth generation airway segments. This suggests that some  
29 adaptation had taken place during long-term exposure, possibly increased inner wall thickness.

30 It is well known that the changes in breathing pattern and lung function caused by O<sub>3</sub> are  
31 attenuated with repeated daily exposures for at least 3 to 5 days. But guinea pigs exposed to

1 0.5 ppm O<sub>3</sub>, 8 h/day for 7 days showed enhancement of responsiveness of rapidly adapting  
2 airway receptors (Joad et al., 1998). Repeated exposure increased receptor activity to SP,  
3 methacholine, and hyperinflation; there were no significant effects on baseline or SP- and  
4 methacholine-induced changes in lung compliance and resistance, suggesting that the  
5 responsiveness of rapidly adapting receptors was enhanced.

6 Male and female Hartley guinea pigs exposed to O<sub>3</sub> (0.1 and 0.3 ppm, 4 h/day, 4 days/week  
7 for 24 weeks) were evaluated for airway responsiveness following ACh or OVA inhalation  
8 challenges (Schlesinger et al., 2002a,b). Ozone exposure did not cause AHR in nonsensitized  
9 animals but did exacerbate AHR to both ACh and OVA in sensitized animals that persisted for  
10 4 weeks after exposure. The effects of O<sub>3</sub> on airway responsiveness were gender independent  
11 and were concentration-related for the ACh challenges.

12 Schelegle et al. (2003a) evaluated airway responsiveness in infant rhesus monkeys exposed  
13 to a 5 day O<sub>3</sub> episode repeated every 14 days over a 6-month period. Half of the monkeys were  
14 sensitized to house dust mite allergen (HDMA; *Dermatophagoides farinae*) at 14 and 28 days of  
15 age before exposure to a total of 11 episodes of O<sub>3</sub> (0.5 ppm, 8 h/day for 5 days followed  
16 by 9 days of FA), HDMA, or O<sub>3</sub> + HDMA. Baseline R<sub>aw</sub> was significantly elevated after  
17 10 exposure episodes in the HDMA + O<sub>3</sub> group compared to the FA, HDMA, and O<sub>3</sub> exposure  
18 groups. Aerosol challenge with HDMA at the end of the 10th episode did not significantly  
19 affect R<sub>aw</sub>, V<sub>T</sub>, f<sub>B</sub>, or S<sub>a</sub>O<sub>2</sub>. Aerosol challenge with histamine was not significantly different after  
20 6 episodes; however, the EC150 R<sub>aw</sub> for the HDMA + O<sub>3</sub> group was significantly reduced after  
21 10 episodes when compared to the FA, HDMA, and O<sub>3</sub> exposure groups, indicating the  
22 development of AHR in this group sometime between episodes 6 and 10. The results are  
23 consistent with altered structural development of the conducting airways.

24 During repeated episodic exposures to O<sub>3</sub>, respiratory responses are first altered to a rapid,  
25 shallow breathing pattern, which has long been considered protective, especially to the deep  
26 lung. This dogma has been discounted recently as discussed above (Schelegle et al., 2001).  
27 Alfaro et al. (2004) examined the site-specific deposition of <sup>18</sup>O (1 ppm 2 h) at breathing  
28 frequencies of 80, 120, 160, or 200 breaths/minute (bpm). At all frequencies, parenchymal areas  
29 had a lower content of <sup>18</sup>O than trachea and bronchi. As breathing frequency increased from 80  
30 to 160 bpm, the deposition showed a reduction in midlevel trachea and an increase in both  
31 mainstream bronchi. At this frequency there was also an increase in deposition in parenchyma

1 supplied by short (cranial) airway paths, consistent with results seen by Schelegle et al., (2001).  
2 At 200 bpm <sup>18</sup>O deposition in trachea increased, concurrent with increases in right cranial and  
3 caudal bronchi regions. Right cranial parenchymal content decreased at 200 bpm, whereas right  
4 caudal parenchymal levels did not change at any breathing frequency. The authors list some  
5 limitations of this study, such as the possible effect on regional distribution of ventilation by use  
6 of the negative-pressure ventilator, the effect of paralysis on airway geometry, and possible  
7 translocation of <sup>18</sup>O during the 2 h exposure period. These two studies provide evidence  
8 that O<sub>3</sub>-induced rapid, shallow breathing creates a more evenly distributed injury pattern,  
9 with possibly greater protection from focal injury to the large conducting airways including the  
10 trachea and the left mainstem bronchus.

11 Another study of the adaptive phenomena in SD rats used an exposure paradigm consisting  
12 of 5 days of daily 8 h 1 ppm O<sub>3</sub> exposures followed by 9 days of recovery in FA (Schelegle  
13 et al., 2003b). This O<sub>3</sub>/FA pattern was repeated for 4 cycles and demonstrated that the O<sub>3</sub>-  
14 induced rapid shallow breathing pattern was followed by adaptation that occurred with each  
15 cycle. However, the release of SP from the trachea, the neutrophil content, and cell  
16 proliferation became attenuated after the first cycle, suggesting a disconnect from the rapid  
17 shallow breathing response. Hypercellularity of the CAR epithelium and thickening of the CAR  
18 interstitium, not linked to changes in cell proliferation, were also found. The authors suggest  
19 mechanism(s) of injury from repeated O<sub>3</sub> exposures consisting of diminished neutrophilic  
20 inflammation/and or release of mitogenic neuropeptides, depressed cell proliferative response,  
21 and cumulative distal airway lesion.

22 Following the initial response of a rapid, shallow breathing pattern, animals eventually  
23 adapt with continued episodic exposure despite the continued presence of epithelial damage,  
24 altered structural development, and inflammation of the airways. Chen et al. (2003) used a  
25 subset of the monkeys from the Schlegle et al. (2003a) study to demonstrate that attenuation  
26 of O<sub>3</sub>-induced rapid shallow breathing and lung function changes typically seen with repeated O<sub>3</sub>  
27 exposure may be caused by the adaptation of the respiratory motor responses. This episodic O<sub>3</sub>  
28 exposure appeared to create neuroplasticity of the nucleus tractus solitarius (NTS; a region of the  
29 brainstem which controls respiration), including increased nonspecific excitability of the NTS  
30 neurons, an increased input resistance, and an increased spiking response to intracellular  
31 injections of depolarizing current.



#### 1 **5.2.5.4 Summary and Conclusions - Effects on Airway Responsiveness**

2 Ozone-induced AHR has been reported in a number of laboratory species at an exposure  
3 range of 0.5 to 1.0 ppm and in human asthmatics at ambient levels. In asthmatics, O<sub>3</sub> is thought  
4 to exacerbate AHR to specific allergens by nonspecifically increasing AHR. New studies have  
5 demonstrated that AHR in asthmatics is due in part to chronic inflammation and airway  
6 remodeling. Animal studies have shown that O<sub>3</sub> exposure can augment OVA-induced AHR.  
7 Importantly, there is a temporal relationship between inflammatory cell influx and O<sub>3</sub>-induced  
8 AHR, but inflammation is not a prerequisite of AHR. Repeated O<sub>3</sub> exposures enhance AHR,  
9 possibly by modulating rapidly adapting airway receptors or by altering the structure of  
10 conducting airways.

11 Currently reported investigations on AHR with repeated O<sub>3</sub> exposure to nonsensitized  
12 laboratory animals have shown equivocal results, especially at the most relevant ambient O<sub>3</sub>  
13 concentrations of  $\leq 0.3$  ppm. The few available studies in sensitized laboratory animals are  
14 consistent with the O<sub>3</sub>-induced exacerbation of AHR reported in atopic humans with asthma (see  
15 Chapter 6) but the results are difficult to extrapolate because of interindividual and interspecies  
16 differences in responsiveness to bronchoprovocation and possible adaptation of airway  
17 responsiveness with long-term, repeated O<sub>3</sub> exposures. Therefore, further studies in laboratory  
18 animals are needed to investigate responses to the different challenges in relation to  
19 measurements of airway inflammation and the other physiological and structural factors known  
20 to contribute to airway responsiveness in human subjects.

21 Important new information indicates that rapid shallow breathing in response to O<sub>3</sub> causes  
22 a more evenly distributed injury pattern rather than protects from injury. New insights into the  
23 mechanisms of O<sub>3</sub>-induced AHR suggest that: (1) exercise-induced bronchoconstriction may be  
24 mediated by changes in tonicity of the bronchial smooth muscles; (2) vagally-mediated  
25 mechanisms may affect tracheal epithelial function and increase central airway reactivity;  
26 (3) O<sub>3</sub> may induce direct vascular constriction; (4) O<sub>3</sub> may destroy neural endopeptidases in  
27 airway epithelial cells, thus preventing the inactivation of SP; and (5) repeated O<sub>3</sub> exposures may  
28 diminish neutrophilic inflammation, depress cell proliferation, and cause cumulative distal  
29 airway lesions.

## 5.2.6 Genotoxicity Potential of Ozone

There has been an historical interest in the ability of ground-level pollution to cause cancer, especially lung cancer. This interest has been amplified in recent years by results of an epidemiologic study that suggest association of increased risks of incident lung cancer with elevated long-term ambient concentrations of O<sub>3</sub>, PM<sub>10</sub>, and SO<sub>2</sub> in nonsmoking California males (Beeson et al., 1998; Abbey et al., 1999). However, another larger, nationwide American Cancer Society study (Pope et al., 2002) showed no significant effect of O<sub>3</sub> on mortality risk, but positive associations between warm season (July-September) O<sub>3</sub> concentrations and cardiopulmonary mortality. Studies of children and young adults of southwest metropolitan Mexico City, repeatedly exposed to high levels of O<sub>3</sub>, PM, NO<sub>x</sub>, aldehydes, metals, and other components in a complex ambient mixture, also report DNA damage in blood leukocytes and nasal epithelial cells (Valverde et al., 1997; Calderón-Garcidueñas et al., 1999) and abnormal nasal biopsies (Calderón-Garcidueñas et al., 2001a). (See Chapter 6 for a discussion of the human studies.)

A number of experimental studies have been done to explore the mutagenic/carcinogenic potential of O<sub>3</sub>. In vitro studies are difficult to interpret due to very high exposure levels and culture systems that allowed the potential formation of artifacts. Some recently published in vivo exposure studies (see Table AX5-13) found increased DNA strand breaks in respiratory cells from guinea pigs (Ferng et al., 1997) and mice (Bornholdt et al., 2002) but, again, only on exposure to high doses of O<sub>3</sub> (1 ppm for 72 h and 1 or 2 ppm for 90 min, respectively).

Exposing the A/J mouse strain (known to have a high incidence of spontaneous pulmonary adenomas) to 0.12, 0.50, and 1.0 ppm O<sub>3</sub> for 6 h/day, 5 days/week for up to 9 months, Witschi et al. (1999) did not find O<sub>3</sub> exposure-related differences in lung tumor multiplicity or incidence. Similarly, in a subchronic exposure study (B6C3F<sub>1</sub> mice to 0.5 ppm O<sub>3</sub> for 6 h/day, 5 days/week for 12 weeks) Kim et al. (2001) did not find statistically significant increases in the incidence of lung tumors. Significant differences in mean body weight as well as mean absolute and relative weights of several organs (e.g., liver, spleen, kidney, testes, and ovary) were observed between O<sub>3</sub>-exposed and air-exposed mice. Histopathologic examination of major organs revealed oviductal carcinomas in 3/10 O<sub>3</sub>-exposed female mice.

### 1 **5.2.6.1 Summary and Conclusions - Genotoxicity Potential of Ozone**

2 The weight of evidence from new experimental studies does not appear to support  
3 ambient O<sub>3</sub> as a pulmonary carcinogen in laboratory animal models. These new data are in  
4 agreement with a study of carcinogenicity of O<sub>3</sub> from the NTP study (National Toxicology  
5 Program, 1994; Boorman et al., 1994), which was negative in male and female rats, ambiguous  
6 in male mice, and positive only in female mice at high concentrations of O<sub>3</sub> (i.e., 1.0 ppm).  
7 As none of the new experimental studies of genotoxicity provided lifetime exposure durations  
8 such as those used in NTP cancer studies, the observation of no effects must be tempered by  
9 consideration of the limited duration of the exposure. Overall, then, the new animal studies are  
10 inconclusive as are the epidemiologic studies discussed in Chapter 7, which may be due to  
11 significant species differences in this health endpoint. Also, O<sub>3</sub> could act as a co-carcinogen  
12 functioning to stimulate hyperplasia. In epidemiology studies, exposures typically consist of  
13 mixtures of co-pollutants, some of which are known carcinogens (see Section 5.4.3).

## 14 15 16 **5.3 SYSTEMIC EFFECTS OF OZONE EXPOSURE**

17 Ozone indirectly affects organs beyond the respiratory system due to O<sub>3</sub> reaction products  
18 entering the bloodstream and being transported to target sites. Extra-pulmonary effects could  
19 also be due to the exposure-related production of mediators, metabolic products and cell  
20 trafficking. Although systemic effects are of interest and indicate a very broad array  
21 of O<sub>3</sub> effects, they are of limited influence and difficult to interpret. By protecting from  
22 respiratory tract effects, these systemic effects will likely be protected against also. Systemic  
23 effects are only summarized briefly here and in Table AX5-14.

### 24 25 **5.3.1 Neurobehavioral Effects**

26 Animal behavior, both motor activity and operant behavior, has been shown to be  
27 suppressed by acute O<sub>3</sub> exposures (3 to 6 h) of 0.12 ppm. There is a dose dependent decrease in  
28 activity with increasing exposure levels. Additionally, these lowered activity levels tend to  
29 attenuate with longer exposure periods. New studies in adult laboratory animals confirm that  
30 environmentally relevant O<sub>3</sub> concentrations from 0.2 to 1.0 ppm can decrease motor activity and  
31 affect short- and long-term memory, as tested by passive avoidance conditioning in 4 h

1 exposures in rats (Rivas-Arancibia et al., 1998; Avila-Costa et al., 1999; Dorado-Martinez et al.,  
2 2001), or water-maze learning tasks in mice following a 30-day exposure (Sorace et al., 2001).  
3 The effects have been attributed to reactive oxygen/nitrogen species and/or ozonation products.  
4 The memory deficits could be blocked by administration of vitamin E (Guerrero et al, 1999) or  
5 taurine (Rivas-Arancibia et al., 2000). Increased freezing and decreased exploratory behaviors  
6 were accompanied by decreased serotonin levels and increased levels of NO, glutamate,  
7 dopamine and striatal lipoperoxidation in rats exposed to 1 ppm of O<sub>3</sub> for 4 h (Rivas-Arancibia  
8 et al., 2003). The O<sub>3</sub>-exposed animals also demonstrated neuronal cytoplasm and dendrite  
9 vacuolation and dilation of rough endoplasmic reticulum cisterns, which the authors interpret as  
10 a neurodegenerative process resulting from the oxidative stress of acute O<sub>3</sub> exposure. Niño-  
11 Cabrera et al. (2002) demonstrated that a 0.7 ppm O<sub>3</sub> exposure for 4 h can induce ultrastructural  
12 alterations in the hippocampus and prefrontal cortex in aged rats. These are areas of the brain  
13 where degenerative age-related changes in learning and memory functions have been reported  
14 (Bimonte et al., 2003).

15 Paz (1997) reviewed a series of studies that demonstrated significant alterations of  
16 electroencephalographic (EEG) patterns during sleep in animals acutely exposed to O<sub>3</sub> (0.35 to  
17 1.0 ppm). Rats and cats both showed loss of paradoxical sleep time after 2 to 8 h of O<sub>3</sub> exposure  
18 (Paz and Bazan-Perkins, 1992; Paz and Huitrón-Reséndiz, 1996). Increased total wakefulness,  
19 alterations in circadian rhythm, and a permanent 50% loss of paradoxical sleep time were shown  
20 in rat pups born to dams exposed to 1.0 ppm O<sub>3</sub> during gestation (Haro and Paz, 1993). Effects  
21 on sleep patterns were associated with alterations in brain neurotransmitter levels (Huitrón-  
22 Reséndiz et al., 1994; González-Piña and Paz, 1997) thought to be caused by O<sub>3</sub> reaction  
23 products or prostaglandins (Koyama and Hayaishi, 1994). The permanent effects in pups caused  
24 by high O<sub>3</sub> exposure during gestation were attributed to the diminished antioxidant capability of  
25 fetal tissue (Günther et al., 1993).

26 High, nonambient levels of O<sub>3</sub> (e.g., >1.0 ppm) affect visual and olfactory neural pathways  
27 in the rat. For example, Custodio-Ramirez and Paz (1997) reported a significant delay in visual  
28 evoked potentials recorded in the visual cortex and the lateral geniculate nucleus of male Wistar  
29 rats acutely exposed to high levels of O<sub>3</sub> (1.5, and 3.0 ppm for 4 h). Colin-Barenque et al.  
30 (1999), using the same strain, reported cytological and ultrastructural changes in the granule  
31 layer of the olfactory bulb after a 4-h exposure to 1 to 1.5 ppm O<sub>3</sub>. Although these neural effects

1 are thought to be caused by O<sub>3</sub> reaction products, especially free radicals, the studies do not add  
2 much to an understanding of the underlying mechanisms.

### 3 4 **5.3.2 Neuroendocrine Effects**

5 Early studies suggested an interaction of O<sub>3</sub> with the pituitary-thyroid-adrenal axis because  
6 thyroidectomy, hypophysectomy, and adrenalectomy protected against the lethal effects of high  
7 concentrations of O<sub>3</sub>. Concentrations of 0.7 to 1.0 ppm O<sub>3</sub> for a 1 day exposure in male rats  
8 caused changes in the parathyroid; thymic atrophy; decreased serum levels of thyroid stimulating  
9 hormone, triiodothyronine (T<sub>3</sub>), thyroxine (T<sub>4</sub>), free T<sub>4</sub>, and protein binding; and increased  
10 prolactin. In more recent studies, increased toxicity to O<sub>3</sub> was reported in hyperthyroid rats by  
11 Huffman et al. (2001) and T<sub>3</sub> supplementation was shown to increase metabolic rate and  
12 pulmonary injury in the lungs of O<sub>3</sub>-treated animals (Sen et al., 1993).

13 The mechanisms by which O<sub>3</sub> affects neuroendocrine function are not well understood.  
14 Cottet-Emard et al. (1997) examined catecholamine activity in rat sympathetic efferents and  
15 brain areas of prime importance to adaptation to environmental stressors. Exposures of  
16 0.5 ppm O<sub>3</sub> for 5 days caused inhibition of norepinephrine turnover in heart (-48% of the  
17 control level) but not in lungs and failed to modify the tyrosine hydroxylase activity in superior  
18 cervical ganglia and the catecholamine content in the adrenal glands. In the CNS, O<sub>3</sub> inhibited  
19 tyrosine hydroxylase activity in noradrenergic brainstem cell groups and decreased  
20 catecholamine turnover in the cortex (-49%) and striatum (-18%) but not in the hypothalamus.  
21 This suggests that high ambient levels of O<sub>3</sub> can produce marked neural disturbances in  
22 structures involved in the integration of chemosensory inputs, arousal, and motor control, effects  
23 that may be responsible for some of the behavioral effects seen with O<sub>3</sub> exposure.

### 24 25 **5.3.3 Cardiovascular Effects**

26 Studies of the effects on hematological parameters and blood chemistry in rats have shown  
27 that erythrocytes are a target of O<sub>3</sub>. Exposures to 1.0 ppm O<sub>3</sub> for 3 h have been found to  
28 decrease heart rate (HR), mean arterial pressure (MAP), and core temperature (T<sub>co</sub>) and to induce  
29 arrhythmias with some exposures in rats. These effects are more pronounced in adult and awake  
30 rats than in younger or sleeping animals. Exposures of 0.2 ppm for 48 h have been shown to

1 cause bradycardia, while exposures of 0.1 ppm for 3 days have been shown to cause  
2 bradyarrhythmia in these animals.

3 A more recent study of rats exposed to FA for 6 h, followed 2 days later by a 5 h exposure  
4 to 0.1 ppm O<sub>3</sub>, 5 days later by a 5 h exposure to 0.3 ppm O<sub>3</sub>, and 10 days later by a 5 h exposure  
5 to 0.5 ppm O<sub>3</sub> used the head-out plethysmograph for continuous measurements (Arito et al.,  
6 1997). Each of the O<sub>3</sub> exposures was preceded by a 1 h exposure to FA. Transient rapid shallow  
7 breathing with slightly increased HR appeared 1-2 min after the start of O<sub>3</sub> exposures and was  
8 attributed to an olfactory response. Persistent rapid shallow breathing with a progressive  
9 decrease in HR occurred with a latent period of 12 h. During the last 90-min of exposure,  
10 averaged values for relative minute ventilation tended to decrease with the increase in O<sub>3</sub>  
11 concentration for young (4-6 mo) but not old (20-22 mo) rats.

12 Studies utilizing radiotelemetry transmitters in unanesthetized and unrestrained rats,  
13 Watkinson et al. (1995; 2001) and Highfill and Watkinson (1996) demonstrated that when HR  
14 was reduced during a 5 day 0.5 ppm O<sub>3</sub> exposure, the T<sub>co</sub> and activity levels also decreased. The  
15 decreases in T<sub>co</sub> and blood pressure reported by in these studies and by Arito et al. (1997)  
16 suggest that the changes in ventilation and HR are mediated through physiological and  
17 behavioral defense mechanisms in an attempt to minimize the irritant effects of O<sub>3</sub> inhalation.  
18 Decreased activity was previously reported in laboratory animals during exposure to O<sub>3</sub>  
19 (see above).

20 Similar cardiovascular and thermoregulatory responses in rats to O<sub>3</sub> were reported by  
21 Iwasaki et al. (1998). Repeated exposure to 0.1, 0.3, and 0.5 ppm O<sub>3</sub> 8 h/day for 4 consecutive  
22 days caused disruption of circadian rhythms of HR and T<sub>co</sub> on the first and second exposure days  
23 that was concentration-dependent. The decreased HR and T<sub>co</sub> recovered to control values on the  
24 third and fourth days of O<sub>3</sub> exposure.

25 The thermoregulatory response to O<sub>3</sub> was further characterized by Watkinson et al. (2003).  
26 Male Fischer-344 rats were exposed to 0.0 ppm for 24 h/day (air), 0.5 ppm for 6 h/day  
27 (intermittent) or 0.5 ppm for 23 h/day (continuous) at 3 temperatures, 10 °C (cold), 22 °C  
28 (room), or 34 °C (warm). Another protocol examined the effects of O<sub>3</sub> exposure (0.5 ppm) and  
29 exercise described as rest, moderate, heavy or CO<sub>2</sub>-stimulated ventilation. Both intermittent and  
30 continuous O<sub>3</sub> exposure caused decreases in HR and T<sub>co</sub> and increases in BALF inflammatory  
31 markers. Exercise in FA caused increases in HR and T<sub>co</sub> while exercise in O<sub>3</sub> caused decreases

1 in those parameters. Carbon dioxide and O<sub>3</sub> induced the greatest deficits in HR and T<sub>co</sub>. Several  
2 factors were suggested that may modulate the hypothermic response, including dose, animal  
3 mass, and environmental stress.

4 Laboratory animals exposed to relatively high O<sub>3</sub> concentrations (≥0.5 ppm) demonstrate  
5 tissue edema in the heart and lungs. This may be due to increased circulating levels of atrial  
6 natriuretic factor (ANF), which is known to mediate capillary permeability, vasodilation, and  
7 blood pressure (Daly et al., 2002). Increased levels of ANF were reported in the heart, lungs,  
8 and circulation of rats exposed to 0.5 ppm O<sub>3</sub> for 8 h ( Vesely et al., 1994a,b,c).

9 Earlier work demonstrated O<sub>3</sub>-induced release of functionally active PAF from rodent  
10 epithelial cells and the presence of PAF receptors on AMs. New work examining lipid  
11 metabolism (Section 5.2.1.4) and mediators of inflammatory response and injury  
12 (Section 5.2.3.4) confirm these earlier studies that PAF (Kafoury et al., 1999) and PAF  
13 receptors (Longphre et al., 1999) are involved in responses to O<sub>3</sub>. In addition to the role of PAF  
14 in pulmonary inflammation and hyperpermeability, this potent inflammatory mediator may have  
15 clotting and thrombolytic effects, though this has not been demonstrated experimentally (see  
16 Figure 5-2). This cardiovascular effect may explain, in part, epidemiologic findings of heart  
17 attack and stroke (see Chapter 7). The findings of Pulfer and Murphy (2004); Pulfer et al.,  
18 (2005); Section 5.2.1.4), describing the in vitro and in vivo production of two biologically active  
19 oxysterols, are also suggestive of a mechanism whereby O<sub>3</sub> exposure may be implicated in the  
20 increased risk of cardiopulmonary disease.

#### 21 22 **5.3.4 Reproductive and Developmental Effects**

23 Early studies of pre- and postnatal exposure to O<sub>3</sub> were performed at relatively high  
24 concentrations. Teratogenic effects were not observed with intermittent exposures of 0.44 to  
25 1.97 ppm O<sub>3</sub> during any part of gestation. Continuous exposure during mid-gestation increased  
26 the resorption of embryos while exposures during late gestation delayed some behavioral  
27 developments (e.g., righting, eye opening). There were no effects on neonatal mortality up to  
28 1.5 ppm O<sub>3</sub>, whereas some transient effects on weight gain were observed at exposures of  
29 0.6 ppm O<sub>3</sub>.

30 More recent studies tend to confirm previous conclusions that prenatal exposures to O<sub>3</sub>  
31 concentrations <1.0 ppm do not cause major or widespread somatic or neurobehavioral effects in

1 the offspring of laboratory animals. These studies generally add some weight toward a negative  
2 interpretation of the importance of contributions of low, ambient O<sub>3</sub> to lower birth weights and  
3 gross development defects reported in neonates born to women exposed to typical ambient  
4 pollution (e.g., Renner, 2002; Chen et al., 2002; Ritz and Yu, 1999). Some postnatal O<sub>3</sub>  
5 exposure studies continue to find a few, subtle or borderline somatic and behavioral deficits that  
6 will require further research to better assess potential risk to developing humans.

7 Studies of somatic and neurobehavioral development in female CD-1 mice exposed during  
8 pregnancy (days 7 to 17) to O<sub>3</sub> (0, 0.4, 0.8, or 1.2 ppm) failed to show any O<sub>3</sub> effects on  
9 reproductive or behavioral performance (Bignami et al., 1994). The study did find significant  
10 decreases in body weight gain and delayed eye opening in pups in the 1.2 ppm exposure group.  
11 The lack of effect on behavioral performance contrasts with earlier findings, which may be due  
12 to the use of different species, differing exposure durations, cross-fostering used in the latter  
13 study, different species, and exposure durations during pregnancy. A second study using CD-1  
14 mice exposed in utero from conception through day 17 of pregnancy to 0, 0.2, 0.4, and  
15 0.6 ppm O<sub>3</sub> found no significant deficits in reproductive performance, postnatal somatic and  
16 neurobehavioral development, or adult motor activity (Petruzzi et al., 1995). A third study by  
17 the same group (Petruzzi et al., 1999), using O<sub>3</sub> exposures (0.3, 0.6, or 0.9 ppm) which continued  
18 postnatally until weaning, showed subtle changes in handedness and morphine reactivity.  
19 Exposures to 0.6 ppm O<sub>3</sub> caused a reduced preference for the right paw in adulthood. Exposures  
20 to 0.9 ppm O<sub>3</sub> altered hot plate avoidance after i.p. treatment with morphine in adulthood.

21 CD-1 mice exposed to 0.6 ppm O<sub>3</sub> from birth through weaning demonstrated no  
22 impairment of navigational performance during acquisition and only subtle changes during  
23 reversal (Dell'Omo et al., 1995a). Additionally, there were no O<sub>3</sub>-induced effects on  
24 reproductive performance, but offspring showed a significant reduction in body weight. Effects  
25 on neurobehavioral development with this exposure were minor, with some attenuation of  
26 activity responses and impairment of passive avoidance acquisition (Dell'Omo et al. (1995b)).  
27 The offspring of CD-1 mice continuously exposed from 30 days prior to the formation of  
28 breeding pairs until PND 17 to 0.0, 0.3, or 0.6 ppm O<sub>3</sub> showed only small and selective effects  
29 on somatic and sensorimotor development (Sorace et al., 2001).

30 Morphological changes were found in the anterior cerebellar lobe of rat pups born to dams  
31 exposed during the entire gestation period to very high (1.0 ppm) O<sub>3</sub> concentrations for 12 h/day.



1 (Rivas-Manzano and Paz, 1999). Additionally, the dams displayed significantly fewer  
2 implantations, increased rate of reabsorptions, a high incidence of spontaneous abortion, and  
3 offspring with low birth weight, as noted by previous investigators.  
4

### 5 **5.3.5 Effects on the Liver, Spleen, and Thymus**

6 Early investigations of the effects of O<sub>3</sub> on liver centered on xenobiotic metabolism, and  
7 the prolongation of sleeping time, which was observed at 0.1 ppm O<sub>3</sub>. In some species, only  
8 adults and especially females were affected. In rats, high (1.0 to 2.0 ppm for 3 h) acute O<sub>3</sub>  
9 exposures caused increased production of NO by hepatocytes and enhanced protein synthesis  
10 (Laskin et al., 1994; 1996). The O<sub>3</sub>-associated effects shown in the liver are thought to be  
11 mediated by inflammatory cytokines or other cytotoxic mediators released by activated  
12 macrophages in the lungs (Vincent et al., 1996; Laskin et al., 1998; Laskin and Laskin, 2001).  
13 Except for the earlier work on xenobiotic metabolism, the responses occurred only after very  
14 high acute O<sub>3</sub> exposures.

15 Examinations of the effects of O<sub>3</sub> on spleen and thymus have shown that O<sub>3</sub> primarily  
16 affects T-cell mediated systemic immunity. As with the O<sub>3</sub>-associated effects shown in the liver,  
17 most of the statistically significant changes occurred after acute exposures to very high O<sub>3</sub>  
18 concentrations and relate to systemic oxidative stress. Using more relevant ambient urban O<sub>3</sub>  
19 exposure patterns, effects were not found on systemic immune function of rats.  
20

### 21 **5.3.6 Effects on Cutaneous and Ocular Tissues**

22 Ozone exposure not only affects various organ systems, when inhaled, but also has direct  
23 effects on the exposed skin and eyes. The outermost layer of the skin (stratum corneum; SC)  
24 may be oxidized, which can lead to compromise of the skin barrier and an epidermal  
25 proinflammatory response (Weber et al., 2001; Thiele, 2001). These effects are found only at  
26 very high concentrations (>1-5 ppm) and have not been shown at more relevant ambient levels  
27 of exposure. The skin possesses a well-developed defense system against oxidative stress,  
28 utilizing nonenzymatic (e.g., vitamin C and E, glutathione, uric acid,  $\alpha$ -tocopherol) and  
29 enzymatic (e.g., superoxide dismutase, catalase, glutathione reductase and peroxidase)  
30 antioxidants (Cross et al., 1998). Ocular tissues have similar antioxidant protective function as  
31 the skin but are not as well studied (Mucke, 1996; Rose et al., 1998). Effects of ground-level

1 smog on the eyes have been reported but generally are attributed to related photochemical  
2 oxidants like peroxyacetyl nitrate (Vyskocil et al., 1998) or possibly to atmospheric O<sub>3</sub>  
3 precursors or reaction products like aldehydes. As in other tissues, O<sub>3</sub> may have disparate  
4 high-dose and low-dose mechanisms of effect on skin and eyes, so results must be interpreted  
5 in this light.

6 Hairless mice (SKH-1) exposed to O<sub>3</sub> (0.8 to 10 ppm for 2 h) were used to demonstrate that  
7 O<sub>3</sub> depletes the low molecular weight antioxidants (e.g.,  $\alpha$ -tocopherol, vitamin C, glutathione,  
8 uric acid) in the SC at  $\geq 1.0$  ppm and causes increased MDA at  $\geq 5$  ppm (Weber et al, 1999, 2000,  
9 2001). Valacchi et al. (2000) demonstrated that preexposure to 0.5 O<sub>3</sub> for 2 h followed by low-  
10 dose ultraviolet (UV) radiation (0.33 MED) caused depletion of  $\alpha$ -tocopherol. This suggests that  
11 combined low doses of UV radiation and near-ambient levels of O<sub>3</sub> may cause oxidative stress  
12 on the SC. Prolonged exposure to 0.8 ppm O<sub>2</sub> for 6 h also induces cellular stress responses that  
13 included the formation of HNE protein adducts, HSP27, and heme-oxygenase-1 in the deeper  
14 cellular layers of the skin that continued for up to 18 h after O<sub>3</sub> exposure, followed by repair  
15 processes (Valacchi et al., 2003).

16 The importance of O<sub>3</sub> and UV-induced cellular protein oxidation found in murine skin  
17 models to possibly similar environmentally-induced changes in human SC keratins was  
18 identified by Thiele et al. (1998, 1999) and Thiele (2001). Using the presence of carbonyl  
19 groups in proteins as a marker of reactive oxygen mediated protein oxidation, they reported  
20 higher carbonyl levels in the upper SC from the tanned skin of humans and in the skin of healthy  
21 human volunteers exposed to model chemical oxidants (e.g., hypochlorite, benzoyl peroxide)  
22 that were inversely correlated with vitamin E levels. The environmentally-induced oxidative  
23 damage identified in human SC represents an early pathophysiological stage in the development  
24 of barrier disruption and inflammation, and possibly has implications for the process of  
25 desquamation. The relevance of potentiation of environmental oxidative stress by O<sub>3</sub> exposure  
26 of human skin needs further study.

### 27 28 **5.3.7 Summary and Conclusions - Systemic Effects of Ozone**

29 Neurobehavioral effects of O<sub>3</sub> at concentrations of 0.2 to 1.0 ppm include decreased motor  
30 activity, short- and long-term memory deficits, increased freezing behavior, and decreased  
31 exploratory behaviors. These effects have been associated with reactive oxygen/nitrogen

1 species, ozonation products, altered neurotransmitter levels, morphological changes in several  
2 brain regions, and altered EEG patterns during sleep. Neuroendocrine effects of O<sub>3</sub> include  
3 morphological and hormonal changes in the pituitary-thyroid-adrenal axis at concentrations  
4 of ~0.75 ppm and alterations of visual and olfactory neural pathways at concentrations >1 ppm.  
5 Mechanisms underlying these effects are not understood at this time. Cardiovascular effects  
6 of O<sub>3</sub> at concentrations of 0.3 to 0.5 ppm include decreased HR, T<sub>CO</sub>, and BP, which have been  
7 termed a hypothermic response. Concentrations of O<sub>3</sub> ≥0.5 ppm cause tissue edema (possibly  
8 mediated by ANF).

9 Prenatal exposures to O<sub>3</sub> concentrations <1.0 ppm did not cause noticeable somatic or  
10 neurobehavioral effects in offspring, while concentrations of 1.0 to 1.5 ppm caused varying  
11 effects on neonatal mortality. Some studies have shown an effect of O<sub>3</sub> on liver xenobiotic  
12 enzymes at concentrations as low as 0.1 ppm, while other studies have shown no alterations in  
13 metabolic enzymes at even 1 ppm, with the effects appearing to be highly-species specific.  
14 Effects on spleen and thymus appear to only occur at high O<sub>3</sub> concentrations (>1.0 ppm), while  
15 relevant ambient, urban exposures have no effect on systemic immune function in rats. Effects  
16 of O<sub>3</sub> on cutaneous and ocular tissue are only seen at high, nonrelevant concentrations.  
17  
18

#### 19 **5.4 INTERACTIONS OF OZONE WITH OTHER CO-OCCURRING** 20 **POLLUTANTS**

21 Ozone is part of a complex mixture of air pollutants with a composition and pattern that  
22 varies geographically and temporally (by hour of the day, day of the week, and season). Health  
23 effects caused by the complex mixture are undoubtedly different (either subtly or significantly)  
24 from the additive effects of a few of the hundreds of compounds present. The only disciplinary  
25 approach that can evaluate a “real-world” complex mixture is epidemiology (Chapter 7).  
26 However, because of the difficulty in evaluation of causative factors and quantitative  
27 relationships in epidemiology studies, it is useful to consider animal toxicological studies of  
28 mixtures. Such studies can be divided into three categories: (1) ambient air mixtures,  
29 (2) laboratory-generated complex mixtures (e.g., gasoline combustion mixtures having  
30 ultraviolet-irradiation, other reaction mixtures with O<sub>3</sub> and several other components), and  
31 (3) binary mixtures. In most cases, experimental designs in the first two classes did not have

1 an O<sub>3</sub>-only group, making it difficult to impossible to discern the influence of O<sub>3</sub>. The more  
2 recent mixture studies that are discussed here typically have been with NO<sub>2</sub>, sulfuric acid  
3 (H<sub>2</sub>SO<sub>4</sub>), or ammonium sulfate ([NH<sub>4</sub>]<sub>2</sub>SO<sub>4</sub>).

4 Interpreting the mixture studies in terms of real-world risk is difficult because laboratory  
5 exposure patterns do not always represent real-world exposure patterns. For example, in the real  
6 world, nitrogen dioxide (NO<sub>2</sub>) often peaks before O<sub>3</sub> peaks, with a mixture occurring between  
7 the peaks, but most laboratory exposures used mixtures only. Also, most studies of O<sub>3</sub> and NO<sub>2</sub>  
8 mixtures used ambient levels of O<sub>3</sub> and levels of NO<sub>2</sub> high above ambient. As shall be seen, all  
9 interaction possibilities have occurred, depending upon the composition of the mixture, the  
10 endpoint examined, and the exposure regimen. In some cases, no interaction was found. Most  
11 often, additivity (the effects of the mixture are equal to the sum of the effects of the individual  
12 components) or synergism (the effects of the mixture are greater than the sum of the effects of  
13 the individual components) was observed. Antagonism (the effects of the mixture are less than  
14 the sum of the individual components) was rarely found.

#### 16 **5.4.1 Ozone and Nitrogen Oxides**

17 The most commonly studied copollutant in binary mixtures with O<sub>3</sub> is NO<sub>2</sub>. Both early  
18 work and more recent studies indicate that, although interaction may occur between these two  
19 pollutants, in general, O<sub>3</sub> often masked the effects of the NO<sub>2</sub> or accounted for most of the  
20 response, due to the greater toxicity of O<sub>3</sub>. Very generally, additivity occurred after acute  
21 exposure and synergism occurred with prolonged exposure. Interpreting the mixture studies is  
22 challenging because laboratory exposure patterns rarely simulate real-world exposure patterns.  
23 In the case of NO<sub>2</sub> and O<sub>3</sub>, NO<sub>2</sub> typically peaks before O<sub>3</sub>, with a mixture occurring between the  
24 peaks, but most laboratory exposures used mixtures only. Also, most studies of O<sub>3</sub> and NO<sub>2</sub>  
25 mixtures used ambient levels of O<sub>3</sub> and levels of NO<sub>2</sub> high above ambient. Table AX5-15 lists  
26 more recent studies evaluating coexposures to NO<sub>2</sub> and O<sub>3</sub>.

27 Chronic exposures of rats to O<sub>3</sub> (0.8 ppm) and NO<sub>2</sub> (14.4 ppm) for 6 h/day caused  
28 development of respiratory insufficiency and severe weight loss. Half of these animals died after  
29 55 to 78 days of exposure due to severe fibrosis (Farman et al., 1997). Increased total lung  
30 collagen and elastin were observed, with loss of mature collagen, suggesting breakdown and  
31 remodeling of the lung parenchyma. Morphological examination following these coexposures

1 demonstrates a sequence of events starting with increasing inflammatory and mild fibrotic  
2 changes for the first 3 weeks of exposure, stabilized or even reduced changes after 4 to 6 weeks,  
3 and severe increases over 7 to 9 weeks of exposure (Farman et al., 1999). This suggests that  
4 repair processes occurring during the middle 4 to 6 weeks of exposure become overwhelmed,  
5 leading to progressive fibrosis after 7 to 8 weeks of exposure. When the coexposure was  
6 extended for 90 days, lesions were noted far into the acinus, but the extent of tissue involvement  
7 was the same after 7, 78, and 90 days of exposure. At the end of exposure, high levels of  
8 procollagen types I and III mRNA were observed within central acini in the lungs from the  
9 combined exposure group but not in lungs from the rats exposed to O<sub>3</sub> or NO<sub>2</sub> alone.

10 Sprague-Dawley rats exposed to 0.3 ppm O<sub>3</sub> and the combined exposure of O<sub>3</sub> and  
11 1.2 ppm NO<sub>2</sub> for 3 d demonstrated significant DNA single-strand breaks in AMs (Bermúdez  
12 et al., 1999). No changes were caused by NO<sub>2</sub>-only exposure. The same exposures stimulated  
13 the activity of polyADPR synthetase, suggesting a response to lung cellular DNA repair caused  
14 by oxidant-induced lung injury (Bermúdez, 2001). The laboratory animal model of progressive  
15 pulmonary fibrosis, utilizing long-term, combined O<sub>3</sub> (0.4 to 0.8 ppm) and high-level NO<sub>2</sub> (7 to  
16 14 ppm) exposure, causes an initial acute pulmonary inflammation, followed by adaptation and  
17 repair, and eventually causing pulmonary fibrosis after 6 to 13 weeks of exposure (Ishii et al.,  
18 2000a; Weller et al., 2000). Unfortunately, this model is not very useful for understanding  
19 potential interactive effects of ambient concentrations of O<sub>3</sub> and NO<sub>2</sub>.

## 21 **5.4.2 Ozone and Other Copollutants**

### 22 *Ozone and Formaldehyde*

23 Early studies with combined exposures to O<sub>3</sub> and formaldehyde (HCHO) found evidence  
24 of both synergistic and non-interactive effects. Newer work listed in Table AX5-16 includes  
25 studies of biochemical and histopathological endpoints in rats exposed to 0.4 ppm O<sub>3</sub> and  
26 3.6 ppm HCHO, alone and combined, for 8 h/day for 3 days (Casseo and Feron, 1994). They  
27 demonstrated no interactive effects in the nasal respiratory epithelium, despite the high levels of  
28 HCHO when compared to typical ambient levels of 1 to 10 ppb (e.g., Rehle et al., 2001). Mautz  
29 (2003) studied changes in breathing pattern and epithelial cell proliferation using exposures of  
30 0.6 ppm O<sub>3</sub> and 10 ppm HCHO alone and in combination for 3 h with exercise at two times  
31 resting ventilation. Even with exercise, HCHO does not substantially penetrate to the lower

1 respiratory tract to interact with O<sub>3</sub> and does not alter breathing patterns to modify local O<sub>3</sub> dose.  
2 Parenchymal injury was, therefore, due to O<sub>3</sub> alone. In the nasal transitional epithelium and in  
3 the trachea, however, combined exposure produced additive effects due to the increased volume  
4 of toxicants during exercise. No other combined pollutant studies have been published in the  
5 peer-reviewed literature, although two studies compared the respiratory effects of O<sub>3</sub> to HCHO.  
6 Nielsen et al., (1999) compared upper airway sensory irritation caused by HCHO concentrations  
7 up to 4 ppm to the lower airway irritation caused by O<sub>3</sub>. Using BALB/c mice, they continuously  
8 measured  $f_B$ ,  $V_T$ , expiratory flow,  $T_i$ ,  $T_e$ , and respiratory patterns during acute, 30-min exposures.  
9 They reported a no effect level of 0.3 ppm for HCHO and 1.0 ppm for O<sub>3</sub>.

10 Thus, O<sub>3</sub> and HCHO do not appear to have additive effects, except during exercise, and  
11 that is due to increased volume of gas reaching the tissue. Any possible synergism occurs in the  
12 nasal epithelium. HCHO exerts its effects primarily in the upper respiratory tract, whereas the  
13 primary site of acute cell injury from O<sub>3</sub> occurs in the conducting airways. EPA is currently  
14 completing a toxicological and epidemiological review and risk characterization for  
15 formaldehyde.

### 16 *Ozone and Tobacco Smoke*

17 Early studies of combined exposures of O<sub>3</sub> (1 ppm) and tobacco smoke demonstrated  
18 altered airway responsiveness to inhaled bronchoconstrictor challenge and tracheal vascular  
19 permeability in guinea pigs. Table AX5-17 lists studies completed since the 1996 AQCD  
20 evaluating coexposures of tobacco smoke and O<sub>3</sub>.

21 Wu et al. (1997) reported that inhalation of cigarette smoke evokes a transient  
22 bronchoconstrictive effect in anesthetized guinea pigs. Total pulmonary resistance ( $R_L$ ) and  
23 dynamic lung compliance ( $C_{dyn}$ ) were compared before and after acute exposure to 1.5 ppm O<sub>3</sub>  
24 for 1 h. Cigarette smoke alone (7 ml) at a low concentration (33%) induced a mild and  
25 reproducible bronchoconstriction that slowly developed and reached its peak after a delay  
26 of >1 min. After O<sub>3</sub> exposure, the same cigarette smoke inhalation challenge evoked an intense  
27 bronchoconstriction that occurred more rapidly, reaching its peak within 20 s, and was sustained  
28 for >2 min. Pretreatment with selective antagonists of neurokinin type 1 and 2 receptors  
29 completely blocked the enhanced airway responsiveness suggesting that O<sub>3</sub> exposure induced  
30

1 AHR to inhaled cigarette smoke, which resulted primarily from the bronchoconstrictive effect of  
2 endogenous tachykinins.

3 The above studies were conducted with undiluted tobacco smoke and high O<sub>3</sub>  
4 concentrations. To determine the effects of aged and diluted sidestream cigarette smoke (ADSS)  
5 as a surrogate of environmental tobacco smoke (ETS) on O<sub>3</sub>-induced lung injury, Yu et al.  
6 (2002) exposed male B6C3F1 mice to (1) FA, (2) ADSS, (3) O<sub>3</sub>, or (4) ADSS followed by O<sub>3</sub>  
7 (ADSS/O<sub>3</sub>). Exposure to 30 mg/m<sup>3</sup> ADSS, 6 h/day for 3 days, followed by exposure to  
8 0.5 ppm O<sub>3</sub> for 24 h was associated with a significant increase in the number of cells recovered  
9 by BAL compared with exposure to ADSS alone or O<sub>3</sub> alone. Neutrophils, lymphocytes, and  
10 total protein levels in BAL were increased following the combined exposure when compared  
11 with all other groups. Within the CAR, the percentage of proliferating cells was unchanged from  
12 control following exposure to ADSS alone but was significantly elevated following exposure  
13 to O<sub>3</sub> and further augmented in a statistically significant manner in mice exposed to ADSS/O<sub>3</sub>.  
14 Following exposure to O<sub>3</sub> alone or ADSS/O<sub>3</sub>, the ability of AMs to release IL-6 under LPS  
15 stimulation was significantly decreased, while exposure to ADSS alone or ADSS/O<sub>3</sub> caused a  
16 significantly increased release of TNF $\alpha$  from AMs under LPS stimulation. These data suggest  
17 that ADSS exposure enhances the sensitivity of animals to O<sub>3</sub>-induced lung injury.

18 Acute exposure to ETS also may make a healthy person more susceptible to sequential O<sub>3</sub>  
19 exposure by affecting lung barrier function or the underlying epithelium. Toxicological studies  
20 with components of ETS (e.g., nicotine receptor agonists, acrolein, and oxidants) have shown  
21 that the vagal bronchopulmonary C-fibers are stimulated by acute exposures that initiate both  
22 central and local responses (Bonham et al., 2001; Mutoh et al., 2000). The central responses  
23 (e.g., tachypnea, cough, bronchoconstriction, increased mucous secretion) are more protective of  
24 the lungs; however, local responses may include increased sensitization of the C-fibers to other  
25 irritants, including O<sub>3</sub>. Active tobacco smokers should not be similarly affected because they  
26 already have significant chronic airway inflammation and increased mucus production. In fact,  
27 chronic smokers appear to have diminished lung function responses to O<sub>3</sub> (see Chapter 6).

### 29 **5.4.3 Complex (Multicomponent) Mixtures Containing Ozone**

30 Ambient pollution in most areas is a complex mix of more than two chemicals. A number  
31 of new studies have examined the effects of exposure to multicomponent atmospheres

1 containing O<sub>3</sub>. Some of these studies attempted to simulate photochemical reaction products  
2 occurring under actual atmospheric conditions. However, the results of these studies are often  
3 difficult to interpret because of chemical interactions between the components, as well as the  
4 resultant production of variable amounts of numerous secondary reaction products, and a lack of  
5 precise control over the ultimate composition of the exposure environment. In addition, the role  
6 of O<sub>3</sub> in the observed biological responses is often obscure. Prior studies using irradiated  
7 automobile exhaust mixtures containing total oxidant concentrations (expressed as O<sub>3</sub>) in the  
8 range of 0.2 to 1.0 ppm have demonstrated pulmonary function changes in several species.

9 A more recent attempt has been made to examine multicomponent mixtures resulting from  
10 the reaction of O<sub>3</sub> with unsaturated hydrocarbons [e.g., isoprene (C<sub>5</sub>H<sub>8</sub>) and terpene (C<sub>10</sub>H<sub>16</sub>)],  
11 producing HCHO, formic acid, acetone, acrolein, acetic acid, and other oxidation products, many  
12 of which are strong airway irritants. Wilkins et al. (2001) evaluated sensory irritation by  
13 measuring mean f<sub>B</sub> in the mouse bioassay and found a 50% reduction after 30 min of exposure to  
14 reaction products of O<sub>3</sub> and isoprene. The mixture at this time period contained <0.2 ppm O<sub>3</sub>, so  
15 the authors attributed the observed effects to the oxidation products. Clausen et al. (2001), using  
16 the same mouse model, evaluated the reaction products of O<sub>3</sub> and limonene. A 33% reduction in  
17 mean f<sub>B</sub> was produced after 30 min of exposure to the mixture containing <0.3 ppm O<sub>3</sub>, again  
18 implicating the effects of strong irritant products. Further work needs to be done with these  
19 complex reaction mixtures because of their potential impact on the respiratory tract. The results  
20 would be particularly important, however, to the reaction of O<sub>3</sub> indoors (see Chapter 3).

21 Pollutant mixtures containing acid aerosols comprise another type of commonly examined  
22 exposure atmosphere (studies summarized in Table AX5-18). Earlier studies that employed  
23 simultaneous single, repeated, or continuous exposures of various animal species to mixtures of  
24 acid sulfates and O<sub>3</sub> found responses for several endpoints, including tracheobronchial  
25 mucociliary clearance, alveolar clearance, pulmonary mechanics, and lung morphology, to be  
26 due solely to O<sub>3</sub>. Some synergism was noted for bacterial infectivity, response to antigen, and  
27 effects on lung protein content and the rate of collagen synthesis.

28 More recent studies found some differences in airway responses to inhaled acid particle-O<sub>3</sub>  
29 mixtures that may have been partly due to airway dosimetry. Various physical and chemical  
30 mechanisms may be responsible (see Schlesinger, 1995). For example, physical adsorption or  
31 absorption of O<sub>3</sub> or its reaction products on a particle could result in transport to more sensitive



1 sites, or to sites where O<sub>3</sub>, by itself, would not normally be reactive (Madden et al., 2000).  
2 Chemical reactions on the surface of particles can form secondary products that are more  
3 toxicologically active, or chemical characteristics of the particle may change the residence time  
4 or reactivity of oxidation products at the site of deposition. The hypothesis that synergism  
5 between O<sub>3</sub> and sulfates is due to decreased pH changing the residence time or reactivity of  
6 reactants, such as free radicals, was tested by Chen et al. (1995) and El-Fawal et al. (1995).  
7 Male New Zealand white rabbits were exposed for 3 h to 125 µg/m<sup>3</sup> H<sub>2</sub>SO<sub>4</sub>, 0.1, 0.3, or  
8 0.6 ppm O<sub>3</sub>, and to combinations. Chen et al. (1995) demonstrated that decreased pH following  
9 exposure to acid aerosol was correlated with phagocytic activity and capacity of harvested  
10 macrophages and that exposure to O<sub>3</sub>/ H<sub>2</sub>SO<sub>4</sub> removed this relationship. El-Fawal et al. (1995)  
11 showed that responsiveness of rabbit harvested bronchial rings to ACh was increased following a  
12 3 h O<sub>3</sub> exposure, but that 0.1 to 0.6 ppm O<sub>3</sub>/0.5 to 0.125 mg/m<sup>3</sup> H<sub>2</sub>SO<sub>4</sub> combinations resulted in  
13 antagonism.

14 As discussed in Section 5.2.2.1, Churg et al. (1996) demonstrated increased uptake of  
15 asbestos or TiO<sub>2</sub> in response to 10 min O<sub>3</sub> (up to 1.0 ppm) pre-exposure suggesting that low  
16 concentrations of O<sub>3</sub> may increase the penetration of some types of PM into epithelial cells.  
17 Using human epithelial cell cultures, Madden et al. (2000) demonstrated a greater potency for  
18 ozonized diesel PM to induce prostaglandin E<sub>2</sub> production. This suggests that 0.1 ppm O<sub>3</sub> for  
19 24 h can modify the biological activity of PM derived from diesel exhaust.

20 Effects of combined exposures of O<sub>3</sub> and resuspended urban particles on cell proliferation  
21 in epithelial cells of the terminal bronchioles and the alveolar ducts were examined by Vincent  
22 et al. (1997) and Adamson et al. (1999). Rats exposed to 0.8 ppm O<sub>3</sub> in combination with 5 or  
23 50 mg/m<sup>3</sup> particles for 4 h demonstrated greatly potentiated proliferative effects compared to O<sub>3</sub>  
24 exposure alone. These findings using resuspended dusts, although at high concentrations, are  
25 consistent with the studies demonstrating interaction between H<sub>2</sub>SO<sub>4</sub> aerosols and O<sub>3</sub>. Effects of  
26 acute coexposure to 0.6 ppm O<sub>3</sub> and fine or ultrafine H<sub>2</sub>SO<sub>4</sub> (0.5 to 0.3 mg/m<sup>3</sup>) aerosols on lung  
27 morphology were examined by Kimmel et al. (1997). They demonstrated that alveolar septal  
28 volume was increased in animals co-exposed to O<sub>3</sub> and ultrafine, but not fine, H<sub>2</sub>SO<sub>4</sub>.  
29 Interestingly, cell proliferation was increased only in animals co-exposed to fine H<sub>2</sub>SO<sub>4</sub> and O<sub>3</sub>,  
30 as compared to animals exposed to O<sub>3</sub> alone. Subchronic exposure to acid aerosols (20 to  
31 150 µg/m<sup>3</sup> H<sub>2</sub>SO<sub>4</sub>) had no interactive effect on the biochemical and morphometric changes

1 produced by either intermittent or continuous exposure to 0.12 to 0.2 ppm O<sub>3</sub> for up to 90 days,  
2 which suggests that the interactive effects of O<sub>3</sub> and acid aerosol coexposure in the lung  
3 disappeared during the long-term exposure (Last and Pinkerton, 1997). Sindhu et al. (1998)  
4 observed an increase in rat lung putrescine levels after repeated, combined exposures to O<sub>3</sub> and a  
5 nitric acid vapor for 40 weeks.

6 Other studies have examined interactions between carbon particles and O<sub>3</sub>. The  
7 interactions of intratracheally instilled carbon particles followed by either a 7-day or 60-day  
8 exposure to 0.5 ppm O<sub>3</sub> in rats was evaluated by Creutzenberg et al. (1995). The carbon  
9 particles caused diminished phagocytotic capacity and chemotactic migration capability of AMs  
10 that was stimulated by the subsequent O<sub>3</sub> exposure. Inflammatory responses following  
11 exposures to low- and high-concentration mixtures of O<sub>3</sub> and acidic aerosols (0.2 ppm O<sub>3</sub> +  
12 50 µg/m<sup>3</sup> carbon + 100 µg/m<sup>3</sup> H<sub>2</sub>SO<sub>4</sub>; 0.4 ppm O<sub>3</sub> + 250 µg/m<sup>3</sup> carbon + 500 µg/m<sup>3</sup> H<sub>2</sub>SO<sub>4</sub>,  
13 respectively) for 1 or 5 days was examined by Kleinman et al. (1999). The response with  
14 the O<sub>3</sub>-particle mixture was greater after 5 days (4 h/day) than after day 1. This contrasted  
15 with O<sub>3</sub> exposure alone (0.4 ppm), which caused marked inflammation on acute exposure,  
16 but no inflammation after 5 consecutive days of exposure.

17 The effects of a mixture of elemental carbon particles, 0.2 ppm O<sub>3</sub>, and 0.5 mg/m<sup>3</sup>  
18 ammonium bisulfate on rat lung collagen content and macrophage activity was examined by  
19 Kleinman et al. (2000). Decreases in lung collagen, and increases in macrophage respiratory  
20 burst and phagocytosis were observed relative to other pollutant combinations. Mautz et al.  
21 (2001) used a similar mixture (i.e., elemental carbon particles, 0.16 to 0.59 ppm O<sub>3</sub>, ammonium  
22 bisulfate 0.5 to 0.22 mg/m<sup>3</sup>, but with 0.11 to 0.39 ppm NO<sub>2</sub> also) and exposure regimen as  
23 Kleinman et al. (2000). Also observed were decreases in pulmonary macrophage Fc-receptor  
24 binding and phagocytosis and increases in acid phosphatase staining. Bronchoalveolar epithelial  
25 permeability and cell proliferation were increased. Altered breathing-patterns also were  
26 observed, with some adaptations occurring.

27 Bolarin et al. (1997) exposed rats to 50 or 100 µg/m<sup>3</sup> carbon particles in combination with  
28 ammonium bisulfate and 0.2 ppm O<sub>3</sub>. Despite 4 weeks of exposure, they observed no changes in  
29 protein concentration in lavage fluid or blood prolyl 4-hydroxylase, an enzyme involved in  
30 collagen metabolism. Slight decreases in plasma fibronectin were present in animals exposed to  
31 the combined pollutants versus O<sub>3</sub> alone. Thus, the potential for adverse effects in the lungs of

1 animals challenged with a combined exposure to particles and gaseous pollutants is dependent  
2 on numerous factors, including the gaseous co-pollutant, concentration, and time.

3 In a complex series of studies, Oberdörster and colleagues examined the interaction of  
4 several pulmonary oxidative stress pollutants. Elder et al. (2000a,b) reported the results of  
5 combined exposure to ultrafine carbon particles (100  $\mu\text{g}/\text{m}^3$ ) and  $\text{O}_3$  (1 ppm for 6 h) in young  
6 and old Fischer 344 rats that were pretreated with aerosolized endotoxin. In old rats, exposure to  
7 carbon and  $\text{O}_3$  produced an interaction that resulted in a greater influx in neutrophils than that  
8 produced by either agent alone. This interaction was not seen in young rats. Oxidant release  
9 from lavage fluid cells also was assessed and the combination of endotoxin, carbon particles,  
10 and  $\text{O}_3$  produced an increase in oxidant release in old rats. This mixture produced the opposite  
11 response in the cells recovered from the lungs of the young rats, indicating that the lungs of the  
12 aged animals underwent greater oxidative stress in response to a complex pollutant mix of  
13 particles,  $\text{O}_3$ , and a biogenic agent. Johnston et al. (2000a; 2002) reported the results of  
14 combined exposure to  $\text{O}_3$  (1.0 and 2.5 ppm for 4, 20, or 24 h) and low-dose endotoxin, or to  $\text{O}_3$   
15 and endotoxin separately, in newborn and adult C57BL/6J mice. In the first study, adult (8 wk  
16 old) mice showed greater sensitivity to  $\text{O}_3$  than newborn (36 h old) mice on the basis of mRNAs  
17 encoding for various chemokines and cytokines. In contrast, adult and newborn mice responded  
18 similarly 2 h after endotoxin exposure (10 ng for 10 min), suggesting that age differences  
19 in  $\text{O}_3$ -generated inflammation is secondary to epithelial cell injury. In the second study, 8 wk  
20 old mice exposed to  $\text{O}_3$  (1 ppm for 24 h) followed by endotoxin (37.5 EU for 10 min) showed  
21 increased responsiveness over either exposure alone, on the basis of increased expression of  
22 chemokine and cytokine messages and increased BAL fluid levels of protein and PMNs.

23 Fanucchi et al. (1998) and Wagner et al. (2001a,b) examined the synergistic effect of  
24 coexposure to  $\text{O}_3$  and endotoxin on the nasal transitional epithelium of rats that also was  
25 mediated, in part, by neutrophils. Fisher 344 rats intranasally instilled with endotoxin and  
26 exposed to 0.5 ppm  $\text{O}_3$ , 8 h per day, for 3 days developed mucous cell metaplasia in the nasal  
27 transitional epithelium, an area normally devoid of mucous cells; whereas, intratracheal  
28 instillation of endotoxin (20  $\mu\text{g}$ ) caused mucous cell metaplasia rapidly in the respiratory  
29 epithelium of the conducting airways. A synergistic increase of intraepithelial mucosubstances  
30 and morphological evidence of mucous cell metaplasia were found in rat maxilloturbinates upon  
31 exposure to both  $\text{O}_3$  and endotoxin, compared to each pollutant alone. A similar response was

1 reported in OVA-sensitized Brown Norway rats exposed to 0.5 ppm O<sub>3</sub>, 8 h/day for 3 days  
2 (Wagner et al., 2002), indicating that coexposure to O<sub>3</sub> and inflammatory biogenic substances  
3 like allergens (e.g., OVA) or bacterial endotoxin can augment epithelial and inflammatory  
4 responses in rat nasal passages.

5 In follow-up studies, Wagner et al. (2003) reported that coexposure of rats to O<sub>3</sub> and  
6 endotoxin also enhanced epithelial and neutrophilic inflammatory responses in the pulmonary  
7 airways. Fisher 344 rats were intranasally instilled with endotoxin and exposed to 1.0 ppm O<sub>3</sub>  
8 for 8 h, which was repeated 24 h later. Three days after the last exposure, BALF was analyzed  
9 for inflammatory cells and secreted mucosubstances (mucin 5AC), and lung tissue was  
10 processed for morphometric analysis. Endotoxin instillation alone caused a dose-dependent  
11 increase in BALF neutrophils that was further increased 2-fold in O<sub>3</sub>-exposed rats given 20 µg  
12 endotoxin, the highest dose. Mucin glycoprotein 5AC also was increased in the BALF at this  
13 dose but not at lower endotoxin doses. Ozone exposure alone did not cause mucus  
14 hypersecretion, but it did potentiate mucus secretion in rats given both 2 and 20 µg endotoxin  
15 and increased intraepithelial mucosubstances 2-fold, which was further substantiated by  
16 significant increases in mucin gene (rMuc5AC) mRNA levels in the conducting airways.

17 The effect of O<sub>3</sub> modifying the biological potency of PM (diesel PM and carbon black) was  
18 examined by Madden et al. (2000) in rats. Reaction of NIST Standard Reference Material  
19 # 2975 diesel PM with 0.1 ppm O<sub>3</sub> for 48 hr increased the potency (compared to unexposed or  
20 air-exposed diesel PM) to induce neutrophil influx, total protein, and LDH in lung lavage fluid in  
21 response to intratracheal instillation. Exposure of the diesel PM to high, nonambient O<sub>3</sub>  
22 concentration (1.0 ppm) attenuated the increased potency, suggesting destruction of the bioactive  
23 reaction products. Unlike the diesel particles, carbon black particles exposed to 0.1 ppm O<sub>3</sub> did  
24 not exhibit an increase in biological potency, which suggested that the reaction of organic  
25 components of the diesel PM with O<sub>3</sub> were responsible for the increased potency.

26 Ulrich et al. (2002) investigated the effect of ambient PM from Ottawa Canada (EHC-93)  
27 on O<sub>3</sub>-induced inflammation. Male Wistar rats were exposed to 0.8 ppm O<sub>3</sub> for 8 h and allowed  
28 to recover before intratracheal instillation of 0.5, 1.5, and 5 mg of EHC-93 in 0.3 ml of saline.  
29 The high concentrations of PM used were sufficient to induce pulmonary inflammation, which  
30 was not exacerbated by pre-exposure to O<sub>3</sub>. Rats from the combined exposure group did have

1 higher and more persistent lung lavage protein and albumin levels, as well as increased plasma  
2 fibrinogen levels when compared to PM exposure alone.

3 The interaction of PM and O<sub>3</sub> was further examined in a murine model of OVA-induced  
4 asthma. Kobzik et al. (2001) investigated whether coexposure to inhaled, concentrated ambient  
5 particles (CAPs) from Boston, MA and to O<sub>3</sub> could exacerbate asthma-like symptoms. On days  
6 7 and 14 of life, half of the BALB/c mice used in this study were sensitized by intraperitoneal  
7 (ip) injection of OVA and then exposed to OVA aerosol on three successive days to create the  
8 asthma phenotype. The other half received the ip OVA but were exposed to a phosphate-  
9 buffered saline aerosol (controls). The mice were further subdivided (n ≥ 61/group) and exposed  
10 for 5 h to CAPs, ranging from 63 to 1,569 µg/m<sup>3</sup>, 0.3 ppm O<sub>3</sub>, CAPs + O<sub>3</sub>, or to FA. Pulmonary  
11 resistance and airway responsiveness to an aerosolized MCh challenge were measured after  
12 exposures. A small, statistically significant increase in pulmonary resistance and airway  
13 responsiveness, respectively, was found in both normal and asthmatic mice immediately after  
14 exposure to CAPs alone and to CAPs + O<sub>3</sub> but not to O<sub>3</sub> alone or to FA. By 24 h after exposure,  
15 the responses returned to baseline levels. There were no significant increases in airway  
16 inflammation after any of the pollutant exposures. In this well-designed study of a small-animal  
17 model of asthma, O<sub>3</sub> and CAPs did not appear to be synergistic. In further analysis of the data  
18 using specific elemental groupings of the CAPs, the acutely increased pulmonary resistance  
19 was found to be associated with the AlSi fraction of PM. Thus, some components of  
20 concentrated PM<sub>2.5</sub> may affect airway caliber in sensitized animals, but the results are difficult  
21 to extrapolate to people with asthma.

22 Animal studies have examined the adverse cardiopulmonary effects of complex mixtures in  
23 urban and rural environments of Italy (Gulisano et al., 1997), Spain (Lorz and López, 1997), and  
24 Mexico (Vanda et al., 1998; Moss et al., 2001). Some of these studies have taken advantage of  
25 the differences in pollutant mixtures of urban and rural environments to report primarily  
26 morphological changes in the nasopharynx and lower respiratory tract (Gulisano et al., 1997;  
27 Lorz and López, 1997) of lambs and pigeons, respectively, after natural, continuous exposures to  
28 ambient pollution. Each study has provided evidence that animals living in urban air pollutants  
29 have greater pulmonary changes than those that would occur in a rural and presumably cleaner,  
30 environment. However, these studies either did not report ambient O<sub>3</sub> levels, or reported only  
31 annual means.

1 The study by Moss et al. (2001) examined the nasal and lung tissue of rats exposed  
2 (23 h/day) to Mexico City air for up to 7 weeks and compared them to controls similarly exposed  
3 to FA. No inflammatory or epithelial lesions were found using quantitative morphological  
4 techniques; however, the concentrations of pollutants were low. Extrapolation of these results to  
5 humans is restricted, however, by uncontrolled exposure conditions, small sample sizes, and  
6 other unknown exposure and nutritional factors in the studies in mammals and birds, and the  
7 negative studies in rodents. They also bring up the issue of which species of “sentinel” animals  
8 is more useful for predicting urban pollutant effects in humans. Thus, in these field studies, it is  
9 difficult to assign a specific role to any specific component of the mixture for the significant  
10 cardiopulmonary effects reported.

11 Similar morphological changes (Calderón-Garcidueñas et al., 2000a; 2001) and chest X-ray  
12 evidence of mild lung hyperinflation (Calderón-Garcidueñas et al., 2000b) have been reported in  
13 children residing in urban and rural areas of Mexico City. (See Chapter 7 for details of these  
14 studies.) The ambient air in urban areas, particularly in southwestern Mexico City, is a complex  
15 mixture of particles and gases, including high concentrations of O<sub>3</sub> and aldehydes that previously  
16 have been shown to cause airway inflammation and epithelial lesions in humans (e.g., Calderón-  
17 Garcidueñas et al., 1992, 1994, 1996) and laboratory animals (Morgan et al., 1986; Heck et al.,  
18 1990; Harkema et al., 1994, 1997a,b). The described effects demonstrate a persistent, ongoing  
19 upper and lower airway inflammatory process and chest X-ray abnormalities in children residing  
20 predominantly in highly polluted areas. Again, extrapolation of these results to urban  
21 populations of the United States is difficult because of the unique complex mixture of urban  
22 air in Mexico City, uncontrolled exposure conditions, and other unknown exposure and  
23 nutritional factors.

#### 24 **5.4.4 Summary and Conclusions - Interactions of Ozone with other** 25 **Co-occurring Pollutants**

26 It is difficult to summarize the role that O<sub>3</sub> plays in exposure responses to binary mixtures,  
27 and even harder to determine its role in responses to multicomponent, complex atmospheres.  
28 Though the specific mechanisms of action of the individual pollutants within a mixture may be  
29 known, the exact bases for toxic interactions have not been elucidated clearly. Certain generic  
30 mechanisms that may underlie pollutant interactions: (1) physical, involving adsorption of one  
31

1 pollutant onto another and subsequent transport to more or less sensitive sites or to sites where  
2 one of the components of the mixture normally would not deposit in concentrated amounts  
3 (probably not involved in O<sub>3</sub>-related interactions); (2) production of secondary products that may  
4 be more toxicologically active than the primary materials, demonstrated or suggested in a  
5 number of studies as a basis for interaction between O<sub>3</sub> and NO<sub>2</sub> and between O<sub>3</sub> and PM;  
6 (3) biological or chemical alterations at target sites that affect response to O<sub>3</sub> or the copollutant,  
7 which which has been suggested to underlie interactions with mixtures of O<sub>3</sub> and acid sulfates;  
8 4) O<sub>3</sub>- or copollutant-induced physiological change, such as alteration in ventilation pattern,  
9 resulting in changes in the penetration or deposition of one pollutant when another is present.  
10 This has been implicated in enhanced responses to various O<sub>3</sub>-containing mixtures with exercise.

11 Evaluation of interactions between O<sub>3</sub> and copollutants is a complex procedure. Responses  
12 are dependent on a number of host and environmental factors, such that different studies using  
13 the same copollutants may show different types or magnitudes of interactions. The occurrence  
14 and nature of any interaction is dependent on the endpoint being examined and is also highly  
15 related to the specific conditions of each study, such as animal species, health status, exposure  
16 method, dose, exposure sequence, and the physicochemical characteristics of the copollutants.  
17 Because of this, it is difficult to compare studies, even those examining similar endpoints, that  
18 were performed under different exposure conditions. Thus, any description of interactions is  
19 really valid only for the specific conditions of the study in question and cannot be generalized to  
20 all conditions of exposure to a particular chemical mixture. Furthermore, it is generally not  
21 possible to extrapolate the effect of pollutant mixtures from studies on the effects of each  
22 component when given separately. In any case, what can be concluded from the database is that  
23 interactions of O<sub>3</sub>-containing mixtures are generally synergistic (antagonism has been noted in a  
24 few studies), depending on the various factors noted above, and that O<sub>3</sub> may produce more  
25 significant biological responses as a component of a mixture than when inhaled alone.  
26 Furthermore, although most studies have shown that interaction occurs only at higher than  
27 ambient concentrations with acute exposure, some have demonstrated interaction at more  
28 environmentally relevant levels (e.g., 0.05 to 0.1 ppm O<sub>3</sub> with NO<sub>2</sub>) and with repeated exposures.

## 5.5 EFFECTS OF OTHER PHOTOCHEMICAL OXIDANTS

Peroxyacetyl nitrate (PAN) and peroxypropionyl nitrate (PPN) are the most abundant non-O<sub>3</sub> oxidants in ambient air of industrialized areas, other than the inorganic nitrogenous oxidants such as NO<sub>2</sub>, and possibly HNO<sub>3</sub>. Ambient levels of PAN and PPN were reported to be decreasing over the 1990's and available air quality data (Grosjean et al., 2001; Grosjean, 2003; Jakobi and Fabian, 1997) indicate that present peak concentrations of PAN and PPN in ambient air from urban areas are in the low ppb range (e.g., <1 to 10 ppb). The levels found in nonurban areas are considerably lower (Gaffney et al., 1993).

Reactions occur in the troposphere between O<sub>3</sub> and hydrocarbons (e.g., d-limonene) to produce epoxides, hydroperoxides, and peroxides. The majority of the measured ambient hydroperoxides produced is hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), although a small amount of organic hydroperoxides (ROOH) also may be formed. Friedlander and Yeh (1998) have estimated that atmospheric aerosols can carry as high as 1 mM of H<sub>2</sub>O<sub>2</sub> and organic hydroperoxides (e.g., hydroxymethylhydroperoxide) in the associated water. In vitro cell and tissue damage are induced by high concentrations of liquid phase H<sub>2</sub>O<sub>2</sub> (50 μM to 1 mM). Morio et al. (2001) (see Table AX5-19) demonstrated that a 2 h exposure of 10 and 20 ppb of inhaled H<sub>2</sub>O<sub>2</sub> vapor can penetrate the lower lung where it causes inflammation. It is likely that hygroscopic components of PM transport ambient H<sub>2</sub>O<sub>2</sub> into the lower lung and induce tissue injury as well. Exposure of rats to a H<sub>2</sub>O<sub>2</sub>-fine particle mixture (215 or 429 μg/m<sup>3</sup> ammonium sulfate) resulted in increased neutrophil influx, and production of inflammatory mediators by AMs (Morio et al., 2001). Hygroscopic secondary organic aerosols generated by the O<sub>3</sub>/hydrocarbon reactions and their co-occurrence with H<sub>2</sub>O<sub>2</sub> also provides another possible mechanism, yet to be validated, whereby H<sub>2</sub>O<sub>2</sub> can be transported into the lower respiratory tract (e.g., Friedlander and Yeh, 1998). Interaction of inhaled O<sub>3</sub> with unsaturated fatty acids on cell membranes and mucus in the airways generates epoxides, hydroperoxides, and secondary ozonation products such as 4-hydroxynonenal (see Section 5.2.1)

Inhalation toxicological information on the effects of the non-O<sub>3</sub> oxidants has been limited to a few studies on PAN, but at concentrations much higher (approximately 100- to 1,000 fold) than levels typically found in ambient air. Such high acute levels cause changes in lung morphology, behavioral modifications, weight loss, and susceptibility to pulmonary infections. Therefore, acute toxicity of PAN is much lower than O<sub>3</sub>, and it is unlikely that present ambient



1 PAN levels would affect pulmonary function responses to O<sub>3</sub> (reviewed in Vyskocil et al., 1998).  
2 Cytogenetic studies indicate that PAN is not a potent mutagen, clastogen, or DNA damaging  
3 agent in mammalian cells in vivo or in vitro at concentrations several orders of magnitude higher  
4 than the generally encountered ambient air levels in most cities (Vyskocil et al., 1998;  
5 Kligerman et al., 1995; Heddle et al., 1993). Some studies suggest that PAN may be a weak  
6 bacterial mutagen at concentrations much higher than exist in present urban atmospheres  
7 (DeMarini et al., 2000; Kleindienst et al., 1990).

8 An additional level of complexity exists due to the possibility that other ambient oxidants  
9 may contribute to effects attributed to O<sub>3</sub>. As discussed in Chapter 2, both short-lived radicals  
10 and secondary particles containing highly polar compounds are generated in the troposphere by  
11 the same photochemical mechanisms that produce O<sub>3</sub>. It is plausible that, in addition to the  
12 direct effects of O<sub>3</sub>, health effects are produced by ambient exposures to these gaseous and  
13 particulate secondary compounds. Little is known regarding the composition of these reaction  
14 products, and little research has been undertaken evaluating their toxicologic effects. Due to the  
15 many oxidizing species present in the atmosphere, interpretation of toxicology data based on O<sub>3</sub>  
16 exposures alone have the potential for underestimating health effects of ambient oxidant  
17 mixtures.

### 19 **5.5.1 Summary and Conclusions - Effects of Other Photochemical Oxidants**

20 Concentrations of PAN and PPN (<1 to 10 ppb) in ambient air are unlikely to affect  
21 pulmonary function or cause DNA damage. Levels of 10-20 ppm H<sub>2</sub>O<sub>2</sub> can penetrate to the  
22 lower lung directly or be transported there by PM, where inflammation can result; however,  
23 ambient H<sub>2</sub>O<sub>2</sub> levels of are typically < ~5 ppb. As toxicology studies of other photochemical  
24 oxidants are rare, quantitative scientific evaluations of possible health effects of environmental  
25 exposures cannot be completed at this time.

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# 6. CONTROLLED HUMAN EXPOSURE STUDIES OF OZONE AND RELATED PHOTOCHEMICAL OXIDANTS

## 6.1 INTRODUCTION

In the previous chapter, results of ozone (O<sub>3</sub>) studies in laboratory animals and in vitro test systems were presented. The extrapolation of results from animal studies is one mechanism by which information on potential adverse human health effects from exposure to O<sub>3</sub> is obtained. More direct evidence of human health effects due to O<sub>3</sub> exposure can be obtained through controlled human exposure studies of volunteers or through field and epidemiologic studies of populations exposed to ambient O<sub>3</sub> (*see Chapter 7*). Controlled human exposure studies typically use fixed concentrations of O<sub>3</sub> under carefully regulated environmental conditions and subject activity levels. This chapter discusses studies in which volunteers were exposed for up to 8 h to between 0.08 to 0.75 ppm O<sub>3</sub> while at rest or during varying intensities of exercise.

The majority of controlled human studies have investigated the effects of exposure to O<sub>3</sub> in young nonsmoking healthy adults (18 to 35 years of age) performing continuous exercise (CE) or intermittent exercise (IE). Varied combinations of O<sub>3</sub> concentration, exercise routine, and exposure duration have been used in these studies. The responses to ambient O<sub>3</sub> concentrations include decreased inspiratory capacity; mild bronchoconstriction; rapid, shallow breathing patterns during exercise; and symptoms of cough and pain on deep inspiration. Reflex inhibition of inspiration results in a decrease in forced vital capacity (FVC) and total lung capacity (TLC) and, in combination with mild bronchoconstriction, contributes to a decrease in the forced expiratory volume in 1 s (FEV<sub>1</sub>). In addition to physiological pulmonary responses and respiratory symptoms, O<sub>3</sub> has been shown to result in airway hyperresponsiveness, epithelial permeability, and inflammation.

The most salient observations from studies reviewed in the 1996 EPA Ozone Air Quality Criteria Document or O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996) were that: (1) young healthy adults exposed to O<sub>3</sub> concentrations  $\geq 0.08$  ppm develop significant reversible, transient decrements in pulmonary function if minute ventilation ( $\dot{V}_E$ ) or duration of exposure is



1 increased sufficiently, (2) children experience similar spirometric responses but lesser symptoms  
2 from O<sub>3</sub> exposure relative to young adults, (3) O<sub>3</sub>-induced spirometric responses are decreased in  
3 the elderly relative to young adults, (4) there is a large degree of intersubject variability in  
4 physiologic and symptomatic responses to O<sub>3</sub> but responses tend to be reproducible within a  
5 given individual over a period of several months, and (5) subjects exposed repeatedly to O<sub>3</sub> for  
6 several days develop a tolerance to successive exposures, as demonstrated by an attenuation of  
7 responses, which is lost after about a week without exposure.

8 There are several important limitations associated with these clinical studies: (1) the  
9 ability to study only short-term, acute effects; (2) difficulties in trying to link short-term effects  
10 with long-term consequences; (3) the use of a small number of volunteers that may not be  
11 representative of the general population; and (4) the statistical limitations associated with the  
12 small sample size. Sample size affects the power of a study, and having a small number of  
13 samples causes a risk of Type II error, i.e., the incorrect conclusion that no difference exists  
14 between treatments or groups when comparisons are not significantly different. This affects the  
15 confidence in estimates of a minimum O<sub>3</sub> concentration at which some degree of pulmonary  
16 impairment will occur in both the general population and susceptible subpopulations. As a  
17 result, the conclusions drawn from many of the studies cited in this chapter may underestimate  
18 the presence of responses at low O<sub>3</sub> concentrations and low activity levels.

19 Most of the scientific information summarized in this chapter comes from the literature  
20 published since the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996). In addition  
21 to further study of physiological pulmonary responses and symptoms of breathing discomfort,  
22 much of this literature has focused on mechanisms of inflammation and cellular responses to  
23 injury induced by O<sub>3</sub> inhalation. A more thorough discussion and review of this literature  
24 appears in Annex AX6 of this document. In summarizing the literature, effects are described if  
25 they are statistically significant at a probability (p-value) of less than 0.05; otherwise, trends are  
26 noted as such.

27 As spirometry typically *improves* in healthy young adults with exercise exposures to  
28 filtered air (FA), the term “O<sub>3</sub>-induced” is used herein and in the annex to designate effects that  
29 have been corrected for responses during FA exposures. For healthy adults, an O<sub>3</sub>-induced  
30 change in lung function is the difference between the *decrement* experienced with O<sub>3</sub> exposure  
31 and the *improvement* observed with FA exposure. However, the distinction between an O<sub>3</sub>-

1 induced change and a post- versus preexposure change is particularly important in individuals  
2 with respiratory disease who may experience exercise-induced *decrements* in pulmonary  
3 function during both FA and O<sub>3</sub> exposures. Hence, in subjects with respiratory disease, exercise-  
4 induced responses could be mistaken for O<sub>3</sub>-induced responses in the absence of a correction for  
5 FA responses.

## 6 7 8 **6.2 PULMONARY FUNCTION EFFECTS OF OZONE EXPOSURE** 9 **IN HEALTHY SUBJECTS**

### 10 **6.2.1 Introduction**

11 As reviewed in the 1986 and 1996 O<sub>3</sub> AQCD's (U.S. Environmental Protection Agency,  
12 1986, 1996), 0.5 ppm is the lowest O<sub>3</sub> concentration at which statistically significant reductions  
13 in FVC and FEV<sub>1</sub> have been reported in sedentary subjects. On average, young adults (n = 23;  
14 mean age, 22 yrs) exposed at rest for 2 h to 0.5 ppm O<sub>3</sub> had O<sub>3</sub>-induced decrements of ~4% in  
15 FVC and ~7% in FEV<sub>1</sub> (Folinsbee et al., 1978; Horvath et al., 1979). During exercise,  
16 spirometric and symptoms responses are observed at lower O<sub>3</sub> concentrations. For acute  
17 exposures of 2 h or less to  $\geq 0.12$  ppm O<sub>3</sub>, if  $\dot{V}_E$  is sufficiently increased by exercise, healthy  
18 human subjects generally experience decreases in TLC, inspiratory capacity (IC), FVC, FEV<sub>1</sub>,  
19 mean forced expiratory flow from 25% to 75% of FVC (FEF<sub>25-75</sub>), and tidal volume (V<sub>T</sub>)  
20 and increases in specific airways resistance (sRaw), breathing frequency (f<sub>B</sub>), and airway  
21 responsiveness. These exposures also cause symptoms of cough, pain on deep inspiration,  
22 shortness of breath, throat irritation, and wheezing. With exposures of 4- to 8-h in duration,  
23 statistically significant pulmonary function and symptoms responses are observed at  
24 lower O<sub>3</sub> concentrations and lower  $\dot{V}_E$  than in shorter duration studies.

### 25 26 **6.2.2 Acute Exposure for Up to 2 h**

27 With heavy CE ( $\dot{V}_E = 89$  L/min), an O<sub>3</sub>-induced decrement of 9.7% in FEV<sub>1</sub> has  
28 been reported for healthy young adults (n = 17; age, 24 ± 3 yrs) exposed for only 1 h to  
29 0.12 ppm O<sub>3</sub> (Gong et al., 1986). With moderate-to-heavy IE (15 min intervals of rest and  
30 exercise [ $\dot{V}_E = 68$  L/min]), McDonnell et al. (1983) reported a physiologically small, but

1 significant, O<sub>3</sub>-induced decrement of 3.4% in FEV<sub>1</sub> for young healthy adults (n = 22, age,  
2 22 ± 3 yrs) exposed for 2 h to 0.12 ppm O<sub>3</sub>. Using the same 2 h IE exposure protocol, Linn et al.  
3 (1986) found no statistically significant spirometric responses at O<sub>3</sub> concentrations of 0.16 ppm  
4 and lower. However, the subjects in the Linn et al. (1986) study were potentially exposed  
5 concurrently in Los Angeles to ambient O<sub>3</sub> levels of between 0.12 and 0.16 ppm and were on  
6 average 3 yrs older than the subjects in the McDonnell et al. (1983) study. (*The attenuating*  
7 *effects of increasing age and repeated O<sub>3</sub> exposures are discussed in Sections 6.5.1 and 6.6,*  
8 *respectively.*) The disparities between the Linn et al. (1986) and McDonnell et al. (1983) studies  
9 demonstrate the difficulty in determining a no-effect-level for O<sub>3</sub> based on relatively small study  
10 populations.

11 Studies analyzing large data sets (≥300 subjects) provide better predictive ability of acute  
12 changes in FEV<sub>1</sub> at low levels of O<sub>3</sub> and  $\dot{V}_E$  than possible via comparisons between smaller  
13 studies. Such an analysis was performed by McDonnell et al. (1997), who examined FEV<sub>1</sub>  
14 responses in 485 healthy white males (18 to 36 years of age; subjects recruited from the area  
15 around Chapel Hill, NC) exposed once for 2 h to O<sub>3</sub> concentrations of up to 0.40 ppm at rest or  
16 with IE. Decrements in FEV<sub>1</sub> were modeled by sigmoid-shaped curve as a function of subject  
17 age, O<sub>3</sub> concentration,  $\dot{V}_E$ , and duration of exposure. Regarding applicability to the general  
18 population, the McDonnell et al. (1997) model has an apparent limitation of considering only  
19 data for white males. However, two other large studies (n = 372; 18 to 35 yrs of age; subjects  
20 recruited from the area around Chapel Hill, NC) found no significant gender or race effects on  
21 spirometric responses to O<sub>3</sub> exposure (Seal et al., 1993, 1996).

22 Ultman et al. (2004) recently reported pulmonary responses in 60 young healthy  
23 nonsmoking adults (32 M, 28 F) exposed to 0.25 ppm O<sub>3</sub> for 1 h with CE at a target  $\dot{V}_E$  of  
24 30 L/min. Consistent with findings reported in the 1996 O<sub>3</sub> criteria document, considerable  
25 intersubject variability in FEV<sub>1</sub> decrements was reported by Ultman et al. (2004) with responses  
26 ranging from a 4% improvement to a 56% decrement. One-third of the subjects had FEV<sub>1</sub>  
27 decrements of >15% and 7% of the subjects had decrements of >40%. It should be pointed out  
28 that the McDonnell et al. (1997) model predicts only *average* responses. In a more recent study,  
29 McDonnell et al. (1999) also reported a model predicting average symptom responses from O<sub>3</sub>  
30 exposure. Unfortunately, neither of these papers (McDonnell et al., 1997, 1999) provide

1 predictions of intersubject variability in response. (*Section 6.4 of this Chapter discusses*  
2 *intersubject variability in response to O<sub>3</sub> exposure*).

3 In addition to the overt effects of O<sub>3</sub> exposure on the large airways as indicated by  
4 spirometric responses, O<sub>3</sub> exposure also affects the function of the small airways and  
5 parenchymal lung. Foster et al. (1993, 1997) examined the effect of O<sub>3</sub> on ventilation  
6 distribution in healthy adult males. In healthy nonsmoking males (26.7 ± 7 years old) exposed to  
7 FA or 0.33 ppm O<sub>3</sub> for 2 h with IE, there was a significant reduction in the ventilation to the  
8 lower-lung (31% of lung volume) and significant increases in ventilation to the upper- and  
9 middle-lung regions relative to the FA values in 7 of the 9 subjects (Foster et al., 1993).  
10 In another study, 15 healthy nonsmoking males (25.4 ± 2 years old) were exposed to FA or  
11 0.35 ppm O<sub>3</sub> for 2.2 h with IE (Foster et al., 1997). Following O<sub>3</sub> exposure, an inert gas washout  
12 was delayed and resembled a two-compartment washout, whereas pre-O<sub>3</sub> exposure a log-linear  
13 gas clearance as a function of expired volume resembled a single-compartment washout. The  
14 pronounced slow phase of gas washout occurring post-O<sub>3</sub>, represented a 24% decrease in the  
15 washout rate relative to pre-O<sub>3</sub>. At 24-h post-O<sub>3</sub>, 6 of the 12 subjects still had [or developed]  
16 a delayed washout relative to the pre-O<sub>3</sub> maneuver. This suggests a prolonged O<sub>3</sub> effect on the  
17 small airways and ventilation distribution in some individuals.

### 19 **6.2.3 Prolonged Ozone Exposures**

20 In the exposure range of 0.08 to 0.16 ppm O<sub>3</sub>, a number of studies using moderate  
21 quasi continuous exercise (QCE; 50 min exercise and 10 min rest per h) for 4 to 8 h have  
22 shown significant responses under the following conditions: 0.16 ppm for 4 h with QCE  
23 at  $\dot{V}_E \approx 40$  L/min (Folinsbee et al., 1994), 0.08 to 0.12 ppm for 6.6 h with QCE at  $\dot{V}_E \approx 35$  to  
24 40 L/min (Adams, 2002; Adams, 2003a; Folinsbee et al., 1988; Horstman et al., 1990), and  
25 0.12 ppm for 8 h of IE (30 min per h) at  $\dot{V}_E \approx 40$  L/min (Hazucha et al., 1992). Symptoms and  
26 spirometric responses increased with duration of exposure, O<sub>3</sub> concentration, and total  $\dot{V}_E$ .  
27 Airway resistance is only modestly affected with moderate or even heavy exercise combined  
28 with O<sub>3</sub> exposure (Folinsbee et al., 1978; McDonnell et al., 1983; Seal et al., 1993).

### 6.2.3.1 Effect of Exercise Ventilation Rate on FEV<sub>1</sub> Response to 6.6 h Ozone Exposure

It is well established that response to O<sub>3</sub> exposure is a function of  $\dot{V}_E$  in studies of 2 h or less in duration (*See Section AX6.2.2*). It is reasonable to expect that response to a prolonged 6.6-h O<sub>3</sub> exposure is also a function of  $\dot{V}_E$ , although quantitative analyses are lacking. Data from five similar prolonged exposure studies are available for evaluation of FEV<sub>1</sub> responses as a function of exercise  $\dot{V}_E$  (Adams, 2000; Adams and Ollison, 1997; Folinsbee et al., 1988, 1994; Horstman et al., 1990). Each of these studies exposed similarly aged subjects (mean ages 22 to 25 yrs) to 0.12 ppm O<sub>3</sub> for 6.6 h. In total, ten sets of mean FEV<sub>1</sub> decrements were available for exercise  $\dot{V}_E$  ranging from 20 to 43 L/min, although no data were available for  $\dot{V}_E$  between 20 and 30 L/min (*data illustrated in Figure AX6-2*). As in 2 h exposure studies, FEV<sub>1</sub> decrements are a function of  $\dot{V}_E$  in prolonged 6.6-h exposure studies as demonstrated by a significant correlation between these variables (Pearson,  $r = 0.95$ ,  $p < 0.001$ ; Spearman,  $r = 0.84$ ,  $p < 0.01$ ).

### 6.2.3.2 Exercise Ventilation Rate as a Function of Body/Lung Size on FEV<sub>1</sub> Response to 6.6 h Ozone Exposure

Based on the assumption that the total inhaled O<sub>3</sub> dose (product of O<sub>3</sub> concentration, exposure duration, and  $\dot{V}_E$ ) is proportional to the lung size, exercise  $\dot{V}_E$  is typically selected to be a multiple of body surface area (BSA) or FVC. Data from several recent studies do not support the contention that  $\dot{V}_E$  should be normalized. In an analysis of data from 485 young adults, McDonnell et al. (1997) found that any effect of BSA, height, or baseline FVC on percent decrement in FEV<sub>1</sub> was small to nonexistent. This is consistent with Messineo and Adams (1990), who compared pulmonary function responses in young adult women having small ( $n = 14$ ) or large ( $n = 14$ ) lung sizes (mean FVC of 3.74 and 5.11 L, respectively) and found no significant group difference in FEV<sub>1</sub> decrements. For 30 subjects (15M, 15F) exposed to 0.12 ppm O<sub>3</sub> for 6.6 h, Adams (2000) also reported that FEV<sub>1</sub> responses were more closely related to  $\dot{V}_E$  than to  $\dot{V}_E$  normalized to BSA. The O<sub>3</sub> dosimetry study of Bush et al. (1996) suggested that normalization of the O<sub>3</sub> dose might more appropriately be a function of anatomic dead space. Ozone penetrates deeper into the lungs of individuals with larger conducting airway

1 volumes, however, FEV<sub>1</sub> responses in subjects exposed for 2 h to 0.25 ppm O<sub>3</sub> did not appear to  
2 be associated with O<sub>3</sub> uptake (Ultman et al., 2004).

### 3 4 **6.2.3.3 Comparison of 2 h IE to 6.6 h O<sub>3</sub> Exposure Effects on Pulmonary Function**

5 Adams (2003b) examined whether prolonged 6.6-h QCE exposure to a relatively low O<sub>3</sub>  
6 concentration (0.08 ppm) and the 2-h IE exposure at a relatively high O<sub>3</sub> concentration (0.30  
7 ppm) elicited consistent individual subject FEV<sub>1</sub> responses. Individual subject O<sub>3</sub> exposure  
8 reproducibility was first examined via a regression plot of the postexposure FEV<sub>1</sub> response to the  
9 6.6-h chamber exposure as a function of postexposure FEV<sub>1</sub> response to the 2-h IE chamber  
10 exposure. The R<sup>2</sup> of 0.40, although statistically significant, was substantially less than that  
11 observed in a comparison of individual FEV<sub>1</sub> response to the two 2-h IE exposures by chamber  
12 and face mask, respectively (R<sup>2</sup> = 0.83). The Spearman rank order correlation for the chamber  
13 6.6-h and chamber 2-h exposure comparison was also substantially less (0.49) than that obtained  
14 for the two 2-h IE exposures (0.85). The primary reason for the greater variability in the  
15 chamber 6.6-h exposure FEV<sub>1</sub> response as a function of that observed for the two 2-h IE  
16 exposures is very likely related to the increased variability in response upon repeated exposure  
17 to O<sub>3</sub> concentrations lower than 0.18 ppm (R = 0.57, compared to a mean R of 0.82 at higher  
18 concentrations) reported by McDonnell et al. (1985a). This rationale is supported by the lower r<sup>2</sup>  
19 (0.40) observed by Adams (2003b) for the FEV<sub>1</sub> responses found in 6.6 h chamber and face  
20 mask exposures to 0.08 ppm O<sub>3</sub>, compared to an r<sup>2</sup> of 0.83 observed for responses found at 0.30  
21 ppm O<sub>3</sub>.

### 22 23 **6.2.4 Triangular Ozone Exposures**

24 To further explore the factors that determine responsiveness to O<sub>3</sub>, Hazucha et al. (1992)  
25 designed a protocol to examine the effect of varying, rather than constant, O<sub>3</sub> concentrations.  
26 Subjects were exposed to an O<sub>3</sub> level that increased linearly from 0 to 0.24 ppm for the first 4 h  
27 and then decreased linearly from 0.24 to 0 ppm over the second 4 h of the 8 h exposure  
28 (triangular concentration profile) and to a constant level exposure of 0.12 ppm O<sub>3</sub> for 8 h. While  
29 total inhaled O<sub>3</sub> doses for the constant and the triangular concentration profile were almost  
30 identical, the FEV<sub>1</sub> response was dissimilar. For the constant 0.12 ppm O<sub>3</sub> exposure, FEV<sub>1</sub>  
31 declined ~5% by the fifth hour and then remained at that level. With the triangular O<sub>3</sub>

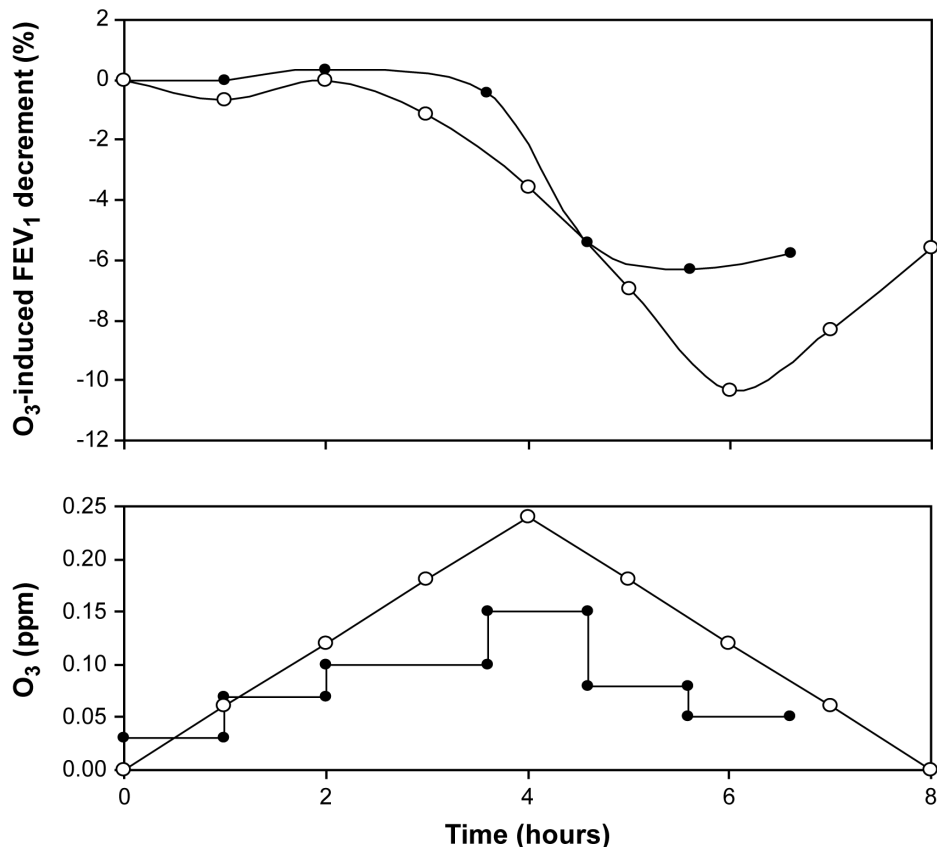
1 concentration profile, there was minimal FEV<sub>1</sub> response over the first 3 h followed by a rapid  
2 decrease in FEV<sub>1</sub> (-10.3%) over the next 3 h. During the seventh and eighth hours, mean FEV<sub>1</sub>  
3 decrements improved to -6.3% as the O<sub>3</sub> concentration decreased from 0.12 to 0.00 ppm  
4 (mean = 0.06 ppm).

5 More recently, Adams (2003a) used a less abrupt triangular O<sub>3</sub> exposure profile at  
6 concentrations assumed to be typical of outdoor ambient conditions (beginning at 0.03 ppm,  
7 increasing steadily to 0.15 ppm in the fourth hour and decreasing steadily to 0.05 ppm at 6.6 h  
8 (mean = 0.08 ppm). Postexposure values for FEV<sub>1</sub> and symptoms were not significantly  
9 different between the 6.6 h triangular and a square-wave 0.08 ppm O<sub>3</sub> exposure. During the  
10 triangular exposure, however, FEV<sub>1</sub> responses became statistically significant after 4.6 h,  
11 whereas, they were not significant until 6.6 h during the square wave exposure (Adams, 2003a).  
12 Perhaps due to the lower O<sub>3</sub> concentrations, evidence of FEV<sub>1</sub> response recovery with the  
13 triangular exposure was less pronounced than as observed by Hazucha et al. (1992). Figure 6-1  
14 illustrates the average O<sub>3</sub>-induced FEV<sub>1</sub> responses and the O<sub>3</sub> exposure schemes for the Adams  
15 (2003a) and Hazucha et al. (1992) studies. For completeness, other studies have also used a  
16 triangular exposure profile but for a shorter duration of only 130 minutes (Foster and  
17 Stetkiewicz, 1996; Foster et al., 1996).

18 With square-wave O<sub>3</sub> exposures between 0.08 to 0.12 ppm, FEV<sub>1</sub> decrements may increase  
19 with time of exposure (and O<sub>3</sub> dose) or reach a plateau (Horstman et al., 1990; McDonnell et al.,  
20 1991). For the triangular exposures used by Hazucha et al. (1992) and Adams (2003a),  
21 maximal FEV<sub>1</sub> responses occurred 1 h to 2 h after peak O<sub>3</sub> concentration and 1 h to 2 h before  
22 the maximal O<sub>3</sub> dose occurred (at the end of the O<sub>3</sub> exposure). These two studies suggest that  
23 depending upon the profile of the exposure, the triangular exposure can potentially lead to  
24 higher FEV<sub>1</sub> responses than square wave exposures at overall equivalent ozone doses.  
25

## 26 **6.2.5 Mechanisms of Pulmonary Function Responses**

27 Inhalation of O<sub>3</sub> for several hours while physically active elicits both subjective respiratory  
28 tract symptoms and acute pathophysiologic changes. The typical symptomatic response  
29 consistently reported in studies is that of tracheobronchial airway irritation. Depending on the  
30 individual's responsiveness to O<sub>3</sub>, this is accompanied by several pathophysiologic changes such  
31 as decrements in lung capacities and volumes, bronchoconstriction, airway hyperresponsiveness,



**Figure 6-1. Triangular exposure profile – O<sub>3</sub>-induced FEV<sub>1</sub> decrements (top panel) and O<sub>3</sub> concentrations (bottom panel) as a function of exposure duration. Open (○) and closed (●) circles illustrate average data from Hazucha et al. (1992) and Adams (2003a), respectively. For clarification, the “O<sub>3</sub>-induced FEV<sub>1</sub> decrement” is the FEV<sub>1</sub> response following O<sub>3</sub> exposure minus the FEV<sub>1</sub> response following FA.**

1 airway inflammation, immune system activation, and epithelial injury. The severity of  
 2 symptoms and the magnitude of response depend on inhaled dose, O<sub>3</sub> sensitivity of an  
 3 individual, and the extent of tolerance resulting from the individual’s previous exposures.  
 4 The development of effects is time-dependent during both exposure and recovery periods with  
 5 considerable overlap of evolving and receding effects. The time sequence, magnitude and the  
 6 type of responses of this complex series of events, both in terms of development and recovery,  
 7 indicate that several mechanisms, activated at different times of exposure, must contribute to the  
 8 overall lung function response (U.S. Environmental Protection Agency, 1996).

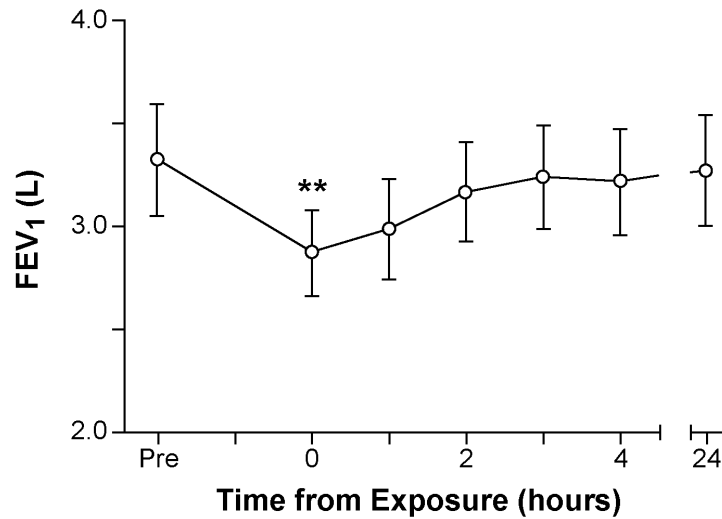


1 Available information on recovery from O<sub>3</sub> exposure indicates that an initial phase of  
2 recovery proceeds relatively rapidly, and some 40 to 65% of the acute spirometric and symptom  
3 response appears to occur within about 2 h (Folinsbee and Hazucha, 1989). Following a 2 h  
4 exposure to 0.4 ppm O<sub>3</sub> with IE, Nightingale et al. (2000) observed a 13.5% decrement in FEV<sub>1</sub>.  
5 By 3 h postexposure, however, only a 2.7% FEV<sub>1</sub> decrement persisted as illustrated in  
6 Figure 6-2. A similar postexposure recovery in FVC was also observed. Gerrity et al. (1993)  
7 suggested that for healthy young adults transient increases in mucus clearance (mediated by  
8 cholinergic receptors) due to O<sub>3</sub> exposure may be coincident to pulmonary function responses,  
9 i.e., the transient increases in clearance and decrements in lung function return to baseline values  
10 within 2 to 3 h postexposure. However, there is some indication that the spirometric responses,  
11 especially at higher O<sub>3</sub> concentrations, are not fully recovered within 24 h (Folinsbee and  
12 Horvath, 1986; Folinsbee et al., 1998). In hyperresponsive individuals, the recovery takes  
13 longer, as much as 48 hours, to return to baseline values. Collectively, these observations  
14 suggest that there is a rapid recovery of O<sub>3</sub>-induced spirometric responses and symptoms, which  
15 may occur during resting exposure to O<sub>3</sub> (Folinsbee et al., 1977) or as O<sub>3</sub> concentration is  
16 reduced during exposure (Hazucha et al., 1992), and a slower phase, which may take at least  
17 24 h to complete (Folinsbee and Hazucha, 2000). Repeated exposure studies at higher  
18 concentrations typically show that FEV<sub>1</sub> response to O<sub>3</sub> is enhanced on the second of several  
19 days of exposure (Table AX6-8). This enhanced response suggests a residual effect of the  
20 previous exposure, about 22 h earlier, even though the preexposure spirometry may be the  
21 same as on the previous day. The absence of the enhanced response with repeated exposure at  
22 lower O<sub>3</sub> concentrations may be the result of a more complete recovery or less damage to  
23 pulmonary tissues (Folinsbee et al., 1994).

#### 24 **6.2.5.1 Pathophysiologic Mechanisms**

##### 25 *Breathing pattern changes*

26 Human studies consistently report that inhalation of O<sub>3</sub> alters the breathing pattern without  
27 significantly affecting minute ventilation. A progressive decrease in tidal volume and a  
28 “compensatory” increase in frequency of breathing to maintain steady minute ventilation during  
29 exposure suggests a direct modulation of ventilatory control. These changes parallel a response  
30 of many animal species exposed to O<sub>3</sub> and other lower airway irritants (Tepper et al., 1990).  
31



**Figure 6-2. Recovery of FEV<sub>1</sub> responses following a 2 h exposure to 0.4 ppm O<sub>3</sub> with IE. Immediately postexposure, FEV<sub>1</sub> was significantly (\*\*p < 0.001) decreased. At 3 h postexposure, FEV<sub>1</sub> was at 97% of the preexposure value.**

Adapted from Nightingale et al. (2000).

1 Bronchial C-fibers and rapidly adapting receptors appear to be the primary vagal afferents  
 2 responsible for O<sub>3</sub>-induced changes in ventilatory rate and depth in both humans (Folinsbee and  
 3 Hazucha, 2000) and animals (Coleridge et al., 1993; Hazucha and Sant'Ambrogio, 1993;  
 4 Schelegle et al., 1993).

5 The potential modulation of breathing pattern by activation of sensory afferents located in  
 6 extrathoracic airways by O<sub>3</sub> has not yet been studied in humans. Nasal only O<sub>3</sub> exposure of rats  
 7 produces changes in breathing pattern that are similar to changes observed in humans (Kleinman  
 8 et al., 1999).

9

10 *Symptoms and lung function changes*

11 As discussed, in addition to changes in ventilatory control, O<sub>3</sub> inhalation by humans will  
 12 also induce a variety of symptoms, reduce vital capacity (VC) and related functional measures,  
 13 and increase airway resistance.

14 Schelegle et al. (2001) demonstrated that the reduction in VC due to O<sub>3</sub> exposure is a reflex  
 15 action and not a voluntary termination of inspiration as result of discomfort. They reported

1 that O<sub>3</sub>-induced symptom responses (mediated in part by bronchial C-fibers) are substantially  
2 reduced by inhaled topical anesthetic. However, the anesthetic had a minor and irregular effect  
3 on pulmonary function decrements and tachypnea. Since respiratory symptom responses were  
4 largely abolished, these findings support reflex inhibition of VC due to stimulation of both  
5 bronchial and pulmonary C-fibers.

6 The involvement of nociceptive bronchial C-fibers modulated by opioid receptors  
7 in limiting maximal inspiration and eliciting subjective symptoms in humans was studied  
8 by Passannante et al. (1998). Sufentanil (an opioid agonist and analgesic) rapidly  
9 reversed O<sub>3</sub>-induced symptom responses and reduced spirometric decrements in “strong”  
10 responders. The incomplete recovery in FEV<sub>1</sub> following sufentanil administration, however,  
11 suggests involvement of non-opioid receptor modulated mechanisms as well. Interestingly,  
12 naloxone (opioid receptor antagonist) had no significant effect on FEV<sub>1</sub> decrements in “weak”  
13 responders. Plasma levels of β-endorphin (a potent pain suppressor) were not related with O<sub>3</sub>  
14 responses.

#### 15 16 *Airway hyperreactivity*

17 In addition to limitation of maximal inspiration and its effects on other spirometric  
18 endpoints, activation of airway sensory afferents also plays a role in receptor-mediated  
19 bronchoconstriction and an increase in airway resistance. Despite this common mechanism,  
20 post-O<sub>3</sub> pulmonary function changes and either early or late bronchial hyperresponsiveness  
21 (BHR) to inhaled aerosolized methacholine or histamine are poorly correlated either in time or  
22 magnitude. Fentanyl and indomethacin, the drugs that have been shown to attenuate O<sub>3</sub>-induced  
23 lung function decrements in humans, did not prevent induction of BHR when administered to  
24 guinea pigs prior to O<sub>3</sub> exposure (Yeadon et al., 1992). Neither does post-O<sub>3</sub> BHR seem to be  
25 related to airway baseline reactivity. These findings imply that the mechanisms are either not  
26 related or are activated independently in time. Animal studies (with limited support from human  
27 studies) have suggested that an early post-O<sub>3</sub> BHR is, at least in part, vagally mediated (Freed,  
28 1996) and that stimulation of C-fibers can lead to increased responsiveness of bronchial smooth  
29 muscle independently of systemic and inflammatory changes which may be even absent (Joad  
30 et al., 1996). In vitro study of isolated human bronchi have reported that O<sub>3</sub>-induced airway  
31 sensitization involves changes in smooth muscle excitation-contraction coupling (Marthan,

1 1996). Characteristic O<sub>3</sub>-induced inflammatory airway neutrophilia which at one time was  
2 considered a leading BHR mechanism, has been found in a murine model, to be only  
3 coincidentally associated with BHR, i.e., there was no cause and effect relationship (Zhang et al.,  
4 1995). However, this observation does not rule out involvement of other cells such as  
5 eosinophils or T-helper cells in BHR modulation. There is some evidence that release of  
6 inflammatory mediators by these cells can sustain BHR and bronchoconstriction. In vitro and  
7 animal studies have also suggested that airway neutral endopeptidase activity can be a strong  
8 modulator of BHR (Marthan et al., 1996; Yeadon et al., 1992). Late BHR observed in some  
9 studies is plausibly due to a sustained damage of the airway epithelium and continual release of  
10 inflammatory mediators (Foster et al., 2000). Thus, O<sub>3</sub>-induced BHR appears to be a product of  
11 many mechanisms acting at different time periods and levels of the bronchial smooth muscle  
12 signaling pathways (*The effects of O<sub>3</sub> on BHR are described in Section 6.8*).  
13

#### 14 **6.2.5.2 Mechanisms at a Cellular and Molecular Level**

15 Stimulation of vagal afferents by O<sub>3</sub> and reactive products, the primary mechanism of lung  
16 function impairment, is enhanced and sustained by what can be considered in this context to be  
17 secondary mechanisms activated at a cellular and molecular level. The complexity of these  
18 mechanisms is beyond the scope of this section and the reader is directed to Section 6.9 of this  
19 chapter for greater detail. A comprehensive review by Mudway and Kelly (2000) discusses the  
20 cellular and molecular mechanisms of O<sub>3</sub>-induced pulmonary response in great detail.

21 Stimulation of bronchial C-fibers by O<sub>3</sub> not only inhibits maximal inspiration but, through  
22 local axon reflexes, induces neurogenic inflammation. This pathophysiologic process is  
23 characterized by release of tachykinins and other proinflammatory neuropeptides. Ozone  
24 exposure has been shown to elevate the C-fiber-associated tachykinin, substance P, in human  
25 bronchial lavage fluid (Hazbun et al. 1993) and to deplete neuropeptides synthesized and  
26 released from C-fibers in human airway epithelium rich in substance P-immunoreactive axons.  
27 Substance P and other transmitters are known to induce granulocyte adhesion and subsequent  
28 transposition into the airways, increase vascular permeability and plasma protein extravasation,  
29 cause bronchoconstriction, and promote mucus secretion (Solway and Leff, 1991). Although the  
30 initial pathways of neurogenic, antigen-induced, and innate immune-mediated inflammation are  
31 not the same, they eventually converge leading to further amplification of airway inflammatory

1 processes by subsequent release of cytokines, eicosanoids, and other mediators. Significantly  
2 negative correlations between O<sub>3</sub>-induced leukotriene (LTC<sub>4</sub>/D<sub>4</sub>/E<sub>4</sub>) production and spirometric  
3 decrements (Hazucha et al., 1996), and an increased level of postexposure PGE<sub>2</sub>, a mediator  
4 known to stimulate bronchial C-fibers, show that these mediators play an important role in  
5 attenuation of lung function due to O<sub>3</sub> exposure (Mohammed et al., 1993; Hazucha et al., 1996).  
6 Moreover, because the density of bronchial C-fibers is much lower in the small than large  
7 airways, the reported post O<sub>3</sub> dysfunction of small airways assessed by decrement in FEF<sub>25-75</sub>  
8 (Weinman et al., 1995; Frank et al., 2001) may be due in part to inflammation. Also, because of  
9 the relative slowness of inflammatory responses as compared to reflex effects, O<sub>3</sub>-triggered  
10 inflammatory mechanisms are unlikely to initially contribute to progressive lung function  
11 reduction. It is plausible, however, that when fully activated, they sustain and possibly further  
12 aggravate already impaired lung function. Indeed, a prolonged recovery of residual spirometric  
13 decrements following the initial rapid improvement after exposure termination could be due to  
14 slowly resolving airway inflammation. Bronchial biopsies performed 6 h postexposure have  
15 shown that O<sub>3</sub> caused a significant decrease in immunoreactivity to substance P in the  
16 submucosa (Krishna et al., 1997). A strong negative correlation with FEV<sub>1</sub> also suggests that the  
17 release of substance P may be a contributing mechanism to persistent post-O<sub>3</sub>  
18 bronchoconstriction (Krishna et al., 1997). Persistent spirometry changes observed for up to  
19 48 h postexposure could plausibly be sustained by the inflammatory mediators, many of which  
20 have bronchoconstrictive properties (Blomberg et al., 1999).

## 23 **6.3 SUBJECTS WITH PREEXISTING DISEASE**

24 Individuals with respiratory disease are of primary concern in evaluating the health effects  
25 of O<sub>3</sub> because even a small change in function is likely to have more impact on a person with  
26 reduced reserve, i.e., O<sub>3</sub>-induced effects are superimposed on preexisting pulmonary impairment.

### 28 **6.3.1 Subjects with Chronic Obstructive Pulmonary Disease**

29 For patients with COPD performing light to moderate IE, no decrements in pulmonary  
30 function were observed after 1- and 2-h exposures to ≤0.30 ppm O<sub>3</sub> (Kehrl et al., 1985; Linn  
31 et al., 1982a, 1983a; Solic et al., 1982), and only small decreases in forced expiratory volume

1 were observed for 3-h exposures of chronic bronchitics to 0.41 ppm O<sub>3</sub> (Kulle et al., 1984).  
2 More recently, Gong et al. (1997a) found no significant difference in response between age-  
3 matched controls and COPD patients to a 4 h exposure to 0.24 ppm O<sub>3</sub> with IE. Although the  
4 clinical significance is uncertain, small transient decreases in arterial blood oxygen saturation  
5 have also been observed in some of these studies.

### 6 7 **6.3.2 Subjects with Asthma**

8 Based on studies reviewed in the 1996 criteria document (U.S. Environmental Protection  
9 Agency, 1996), asthmatic subjects appear to be at least as sensitive to acute effects of O<sub>3</sub> as  
10 healthy nonasthmatic subjects.

11 Several recent studies support a tendency for slightly increased spirometric responses in  
12 mild asthmatic versus healthy subjects. Alexis et al. (2000) reported reductions in FVC (12%,  
13 10%) and FEV<sub>1</sub> (13%, 11%) for 13 mild asthmatic and 9 healthy subjects, respectively, exposed  
14 to 0.4 ppm O<sub>3</sub> for 2 h with IE ( $\dot{V}_E = 30$  L/min). The FVC and FEV<sub>1</sub> responses were attenuated  
15 by indomethacin in the healthy subjects but not the asthmatics. As assessed by the magnitude of  
16 reductions in mid-flows (viz. FEF<sub>25</sub>, FEF<sub>50</sub>, FEF<sub>60p</sub>, FEF<sub>75</sub>) following O<sub>3</sub> exposure, the small  
17 airways tended to be more affected in asthmatics than healthy subjects. In a larger study, Jörres  
18 et al. (1996) exposed 24 asthmatics, 12 allergic rhinitics, and 10 healthy subjects to 0.25 ppm O<sub>3</sub>  
19 for 3 h with IE. The O<sub>3</sub>-induced FEV<sub>1</sub> decrements tended to be greater in the diseased  
20 populations (allergic rhinitis, 14.1%; asthmatic, 12.5%; healthy controls, 10.2%). Scannell et al.  
21 (1996) exposed 18 asthmatics to 0.2 ppm O<sub>3</sub> for 4 h with IE ( $\dot{V}_E \approx 25$  L/min/m<sup>2</sup> BSA).

22 An O<sub>3</sub>-induced increase in sRaw tended to be greater in the asthmatics compared to 81 healthy  
23 subjects who underwent similar experimental protocols (Aris et al., 1995; Balmes et al., 1996).

24 Increased sensitivity of asthmatics to O<sub>3</sub> was also demonstrated in the epidemiological  
25 study by Höpfe et al. (2003). Relevant pulmonary function responses (>10% drop in FEV<sub>1</sub>,  
26 FVC, or PEF, and/or >20% increase in sRaw) subsequent to O<sub>3</sub> exposure were experienced  
27 by 22 of 43 young asthmatics (mean age, 15 yrs) versus only 6 of 43 young athletes (mean  
28 age, 18 yrs). Participants were asked to engage in their normal activities for 2 h in the  
29 afternoon (61-62 ppb O<sub>3</sub>, on average) prior to pulmonary function testing. The estimated  
30 activity level during O<sub>3</sub> exposures was lower in the asthmatics ( $\dot{V}_E \approx 25$  L/min) than the  
31 athletes ( $\dot{V}_E \approx 80$  L/min). As discussed in Sections 6.2.2 and 6.2.3.1, responses to O<sub>3</sub> increase

1 with  $\dot{V}_E$ . Hence, in the absence of some underlying susceptibility to adverse O<sub>3</sub> effects, the  
2 asthmatics would actually be expected to respond far less than the athletes who had a 3.2-fold  
3 greater  $\dot{V}_E$ .

4 Similar O<sub>3</sub>-induced spirometric responses are suggested by some studies. The Scannell  
5 et al. (1996) study of 18 asthmatics reported FEV<sub>1</sub> and FVC decrements that were similar to 81  
6 healthy subjects (Aris et al., 1995; Balmes et al., 1996). Similar group decrements in FEV<sub>1</sub> and  
7 FVC were reported by Hiltermann et al. (1995), who exposed 6 asthmatics and 6 healthy  
8 subjects to 0.4 ppm O<sub>3</sub> for 2 h with light IE. Basha et al. (1994) also reported similar spirometric  
9 responses between 5 asthmatic and 5 healthy subjects exposed to 0.2 ppm O<sub>3</sub> for 6 h with IE.  
10 The lack of significant differences in the Hiltermann et al. (1995) and Basha et al. (1994) studies  
11 is not compelling given the extremely small sample sizes and corresponding lack of statistical  
12 power. The Basha et al. (1994) study was also confounded by the asthmatics having an average  
13 preexposure FEV<sub>1</sub> that was about 430 mL lower (a 12% difference) on the O<sub>3</sub>-day relative to the  
14 air-day. Hence, only the Scannell et al. (1996) study supports similar O<sub>3</sub>-induced spirometric  
15 responses in asthmatics versus healthy subjects.

16 One study has reported that asthmatics tend to have smaller O<sub>3</sub>-induced FEV<sub>1</sub> decrements  
17 relative healthy subjects (3% versus 8%, respectively) when exposed to 0.2 ppm O<sub>3</sub> for 2 h with  
18 IE (Mudway et al., 2001). However, the asthmatics in the Mudway et al. (2001) study also  
19 tended to be older than the healthy subjects, which could partially explain their lesser response.

20 In a longer exposure duration (7.6 h) study, Horstman et al. (1995) exposed 17 mild-to-  
21 moderate asthmatics and 13 healthy controls to 0.16 ppm O<sub>3</sub> or FA with quasi continuous  
22 exercise ( $\dot{V}_E \approx 30$  L/min). The FEV<sub>1</sub> decrement observed in the asthmatics was significantly  
23 greater than in the healthy subjects (19% versus 10%, respectively). There was also tendency for  
24 a greater O<sub>3</sub>-induced decrease in FEF<sub>25-75</sub> in asthmatics relative to the healthy subjects (24%  
25 versus 15%, respectively). A significant positive correlation in asthmatics was also reported  
26 between O<sub>3</sub>-induced spirometric responses and baseline lung function, i.e., responses increased  
27 with severity of disease.

28 With repeated O<sub>3</sub> exposures asthmatics, like healthy subjects (*see Section 6.6*),  
29 develop tolerance. Gong et al. (1997b) exposed 10 asthmatics to 0.4 ppm O<sub>3</sub>, 3 h per day with  
30 IE ( $\dot{V}_E \approx 32$  L/min), for 5 consecutive days. Symptom and spirometric responses were greatest  
31 on the first (-35 % FEV<sub>1</sub>) and second (-34 % FEV<sub>1</sub>) exposure days, and progressively

1 diminished toward baseline levels ( $-6\%$  FEV<sub>1</sub>) by the fifth exposure day. Similar to healthy  
2 subjects, asthmatics lost their tolerance 4 and 7 days later.

3 Some, but not all, studies have reported that asthmatics have a somewhat exaggerated  
4 inflammatory response to acute O<sub>3</sub> exposure relative to healthy controls (e.g., McBride et al.,  
5 1994; Basha et al., 1994; Peden et al., 1995, 1997; Peden, 2001a; Scannell et al., 1996;  
6 Hiltermann et al., 1997, 1999; Michelson et al., 1999; Vagaggini et al., 1999; Newson et al.,  
7 2000; Holz et al., 2002) also (*see Section 6.9 and Tables AX6-3 and -12*). For example, at 18-h  
8 post-O<sub>3</sub> exposure (0.2 ppm, 4-h with IE) and corrected for FA responses, Scannell et al. (1996)  
9 found significantly increased neutrophils in 18 asthmatics (12%) compared to 20 healthy  
10 subjects (4.5%). This inflammatory response difference was observed despite no group  
11 differences in spirometric responses to O<sub>3</sub>. Inflammatory responses do not appear to be  
12 correlated with lung function responses in either asthmatic or healthy subjects (Balmes et al.,  
13 1996, 1997; Holz et al., 1999). This lack of correlations between inflammatory and spirometric  
14 responses may be due to differences in the time kinetics of these responses (Stenfors et al.,  
15 2002). In addition, airway responsiveness to inhaled allergens is increased by O<sub>3</sub> exposure in  
16 subjects with allergic asthma for up to 24 h (*see Section 6.8*).

### 18 **6.3.3 Subjects with Allergic Rhinitis**

19 Allergic rhinitis is a condition defined by inflammation of the nasal membranes. Nayak  
20 (2003) recently reviewed the commonalities between asthma and allergic rhinitis. Clinically,  
21 greater than 60% of asthmatics have allergic rhinitis and slightly less than 40% of allergic  
22 rhinitics have asthma. Leukotrienes and histamine are well-recognized mediators of responses  
23 (viz., inflammation, hyperresponsiveness, and bronchoconstriction) in both asthma and allergic  
24 rhinitis. Although, rhinitis and asthma are distinguished as affecting the upper and lower  
25 airways, respectively, it has been suggested that these diseases are manifestations of the same  
26 disease entity.

27 Given the prevalence of concomitant asthma and rhinitis and their common response  
28 mediators, it should be expected that allergic rhinitics might respond more similarly to  
29 asthmatics than healthy individuals. Regarding spirometric responses, Jörres et al. (1996)  
30 provide the only data demonstrating a trend in support of this supposition.



1 Studies demonstrating the interaction between air pollutants and allergic processes in the  
2 human nasal airways and rhinoconjunctival tissue have been reviewed by Peden (2001b) and  
3 Riediker et al. (2001), respectively. Ozone exposure of subjects with allergic rhinitis has been  
4 shown to induce nasal inflammation and increase airway responsiveness to nonspecific  
5 bronchoconstrictors.

6 Peden et al. (1995), who studied allergic asthmatics exposed to O<sub>3</sub>, found that O<sub>3</sub> causes an  
7 increased response to nasal allergen challenge in addition to nasal inflammatory responses.  
8 Their data suggested that allergic subjects have an increased immediate response to allergen  
9 after O<sub>3</sub> exposure. In a follow-up study, Michelson et al. (1999) reported that 0.4 ppm O<sub>3</sub> did not  
10 promote early-phase-response mediator release or enhance the response to allergen challenge in  
11 the nasal airways of mild, asymptomatic dust mite-sensitive asthmatic subjects. Ozone did,  
12 however, promote an inflammatory cell influx, which helps induce a more significant late-phase  
13 response in this population.

14 Jörres et al. (1996) found that O<sub>3</sub> causes an increased response to bronchial allergen  
15 challenge in subjects with allergic rhinitis. This study also measured responses in healthy  
16 subjects and mildly allergic asthmatics (*see Sections AX6.3.2 and AX6.8*). All subjects were  
17 exposed to 0.25 ppm O<sub>3</sub> for 3 h with IE. Statistically significant O<sub>3</sub>-induced decrements in FEV<sub>1</sub>  
18 occurred in rhinitics (14.1%), asthmatics (12.5%), and the healthy controls (10.2%), but these  
19 responses did not differ statistically between groups. Methacholine responsiveness was  
20 significantly increased in asthmatics, but not in subjects with allergic rhinitis. Airway  
21 responsiveness to an individual's historical allergen (either grass and birch pollen, house dust  
22 mite, or animal dander) was significantly increased after O<sub>3</sub> exposure when compared to FA  
23 exposure. The authors concluded that subjects with allergic rhinitis, but without asthma, could  
24 be at risk if a high O<sub>3</sub> exposure is followed by a high dose of allergen.

25 Holz et al. (2002) extended the results of Jörres et al. (1996) by demonstrating that  
26 repeated daily exposure to lower concentrations of O<sub>3</sub> (0.125 ppm for 4 days) causes an  
27 increased response to bronchial allergen challenge in subjects with preexisting allergic airway  
28 disease, with or without asthma. These investigators observed no major difference in the pattern  
29 of bronchial allergen response between asthmatics or rhinitics, except for a 10-fold increase in  
30 the dose of allergen required to elicit a similar response ( $\geq 20\%$  decrease in FEV<sub>1</sub>) in the  
31 asthmatic subjects. Early phase responses were more consistent in subjects with rhinitis and

1 late-phase responses were more pronounced in subjects with asthma. There also was a tendency  
2 towards a greater effect of O<sub>3</sub> in subjects with greater baseline response to specific allergens  
3 (chosen on the basis of skin prick test and history, viz., grass, rye, birch, or alder pollen, house  
4 dust mite, or animal dander). These data suggest that the presence of allergic bronchial  
5 sensitization, but not a history of asthma, may be a key determinant of increased airway allergen  
6 responsiveness following exposure to O<sub>3</sub> (*for a more complete discussion of airway*  
7 *responsiveness*) see Section AX6.8.

### 9 **6.3.4 Subjects with Cardiovascular Disease**

10 Possibly due to the age of subjects studied, O<sub>3</sub> exposure does not appear to result in  
11 significant pulmonary function impairment or evidence of cardiovascular strain in patients with  
12 cardiovascular disease relative to healthy controls. Gong et al. (1998) exposed 10 hypertensive  
13 and 6 healthy adult males, 41 to 78 years of age, to 0.3 ppm O<sub>3</sub> for 3 h with IE at 30 L/min. For  
14 all subjects combined (no significant group differences), there was an O<sub>3</sub>-induced decrement of  
15 7% in FEV<sub>1</sub> and an 70% increase in the alveolar-arterial oxygen tension gradient. The overall  
16 results did not indicate any major acute cardiovascular effects of O<sub>3</sub> in either the hypertensive or  
17 normal subjects. Gong et al. (1998) suggested that by impairing alveolar-arterial oxygen  
18 transfer, the O<sub>3</sub> exposure could potentially lead to adverse cardiac events by decreasing oxygen  
19 supply to the myocardium. However, the subjects in their study had sufficient functional reserve  
20 so as to not experience significant ECG changes or myocardial ischemia and/or injury (*see*  
21 *Section 6.10 for additional discussion*).

## 24 **6.4 INTERSUBJECT VARIABILITY AND REPRODUCIBILITY** 25 **OF RESPONSE**

26 Analysis of factors that contribute to intersubject variability is important for the  
27 understanding of individual responses, mechanisms of response, and health risks associated with  
28 acute O<sub>3</sub> exposures. A large intersubject variability in response to O<sub>3</sub> has been reported by  
29 numerous investigators (Adams et al., 1981; Aris et al., 1995; Folinsbee et al., 1978; Kulle et al.,  
30 1985; McDonnell et al., 1983). The magnitude of individual variability in FEV<sub>1</sub> response in 2 h  
31 IE exposures increases at higher O<sub>3</sub> concentrations (Kulle et al., 1985; McDonnell et al., 1983).

1 McDonnell (1996) examined the FEV<sub>1</sub> response data from three 6.6-h exposure studies  
2 conducted at the EPA Health Effects Research Laboratory and showed that the FEV<sub>1</sub> responses  
3 in FA were small with most tightly grouped around zero. With increasing O<sub>3</sub> concentrations  
4 between 0.08 and 0.12 ppm, the mean response became asymmetrical with a few individuals  
5 experiencing quite large decrements in FEV<sub>1</sub> (*Intersubject variability observed in O<sub>3</sub> dosimetry*  
6 *studies is discussed in Chapter 4.2).*

7 As an example of the variation in spirometric responses to O<sub>3</sub> exposure, Hazucha et al.  
8 (2003) analyzed the distribution of O<sub>3</sub> responsiveness in 240 subjects (18 to 60 years of age)  
9 exposed to 0.42 ppm O<sub>3</sub> (on 3 occasions) for 1.5 h with IE at  $\dot{V}_E = 20$  L/min/m<sup>2</sup> BSA. Across  
10 all ages, 18% of subjects were weak responders ( $\leq 5\%$  FEV<sub>1</sub> decrement), 39% were moderate  
11 responders, and 43% were strong responders ( $\geq 15\%$  FEV<sub>1</sub> decrement). Younger subjects  
12 ( $\leq 35$  years of age) were predominately strong responders, whereas, older subjects ( $> 35$  years of  
13 age) were mainly weak responders. The influence of age on intersubject variability was also  
14 noted by Passannante et al. (1998) who found that subjects under 35 years of age were more like  
15 to be strong responders than older individuals. In contrast to these clinical studies, Höppe et al.  
16 (2003) observed relevant pulmonary function responses ( $> 10\%$  drop in FEV<sub>1</sub>, FVC, or PEF,  
17 and/or  $> 20\%$  increase in sRaw) subsequent to O<sub>3</sub> exposure in 27% of elderly adults (n = 41;  
18 mean age, 81 yrs;  $\dot{V}_E \approx 10$  L/min) versus only 14% of young athletes (n = 43; mean age,  
19 18 yrs;  $\dot{V}_E \approx 80$  L/min). In the absence of some underlying susceptibility to adverse O<sub>3</sub> effects,  
20 the elderly adults would be expected to respond far less than the athletes who had an estimated  
21 8-fold greater  $\dot{V}_E$ .

22 For repeated exposures, Hazucha et al. (2003) reported that the reproducibility of FEV<sub>1</sub>  
23 responses was related to the length of time between exposures. The Spearman correlation  
24 coefficient of 0.54 was found between responses for exposures separated by 105 days (median),  
25 whereas, a correlation coefficient of 0.85 was found between responses for exposures separated  
26 by only 7 days (median). The more reproducible the subject's response, the more precisely it  
27 indicates his/her intrinsic responsiveness. In 2 h IE O<sub>3</sub> exposures, McDonnell et al. (1985b)  
28 found a relatively poor FEV<sub>1</sub> reproducibility (R = 0.58) at the lowest concentration, 0.12 ppm,  
29 due, in part, to a lack of specific O<sub>3</sub> response or a uniformly small response in the majority of  
30 subjects. It was concluded that for 2 h IE O<sub>3</sub> exposures equal to or greater than 0.18 ppm, the  
31 intersubject differences in magnitude of change in FVC and FEV<sub>1</sub> are quite reproducible over

1 time (21 to 385 days; mean = 33 days) and are due primarily to differences in intrinsic  
2 responsiveness of individual subjects to O<sub>3</sub> exposure.

3 Intersubject variability, mechanisms of response, and health risks associated with acute O<sub>3</sub>  
4 exposures are complicated by a poor association between various O<sub>3</sub>-induced responses. In a  
5 retrospective study of 485 male subjects (ages 18 to 36 yrs) exposed to one of six O<sub>3</sub>  
6 concentrations at one of three activity levels for 2 h, McDonnell et al. (1999) observed  
7 significant, but low, Spearman rank order correlations between FEV<sub>1</sub> response and symptoms of  
8 cough (R = 0.39), shortness of breath (R = 0.41), and pain on deep inspiration (R = 0.30). These  
9 authors concluded that these responses are related mechanistically to some degree, but indicated  
10 that there does not appear to be a single factor which is responsible for the observed individual  
11 differences in O<sub>3</sub> responsiveness across the spectrum of symptom and lung function responses.

12 The effect of large intersubject variability on the ability to predict individual  
13 responsiveness to O<sub>3</sub> was demonstrated by McDonnell et al. (1993). These investigators  
14 analyzed the data of 290 male subjects (18 to 32 years of age) who underwent repeat 2 h IE  
15 exposures to one or more O<sub>3</sub> concentrations ranging from 0.12 to 0.40 ppm. They attempted to  
16 identify personal characteristics (i.e., age, height, baseline pulmonary function, presence of  
17 allergies, and past smoking history) that might predict individual differences in FEV<sub>1</sub> response.  
18 Only age contributed significantly to intersubject responsiveness (younger subjects were more  
19 responsive), accounting for just 4% of the observed variance. Interestingly, O<sub>3</sub> concentration  
20 accounted for only 31% of the variance, strongly suggesting the importance of as yet undefined  
21 individual characteristics that determine FEV<sub>1</sub> responsiveness to O<sub>3</sub>. The authors concluded that  
22 much individual variability in FEV<sub>1</sub> response to O<sub>3</sub> remains unexplained.

## 23 24 25 **6.5 FACTORS MODIFYING RESPONSIVENESS TO OZONE**

### 26 **6.5.1 Influence of Age**

27 Children, adolescents, and young adults (<18 yrs of age) appear, on average, to have nearly  
28 equivalent spirometric responses to O<sub>3</sub>, but have greater responses than middle-aged and older  
29 adults when exposed to a comparable O<sub>3</sub> doses (U.S. Environmental Protection Agency, 1996).  
30 Symptomatic responses to O<sub>3</sub> exposure, however, appear to increase with age until early  
31 adulthood and then gradually decrease with increasing age (U.S. Environmental Protection

1 Agency, 1996). In contrast to young adults, the diminished symptomatic responses in children  
2 and the elderly may put them at an increased risk for continued exposure. Although no new  
3 laboratory studies investigating O<sub>3</sub> responses in children have been published since the last O<sub>3</sub>  
4 AQCD, the epidemiological studies published during the last decade (*see section 7.2.3.1 for*  
5 *details*) are generally in agreement with the earlier laboratory studies.

6 The ensuing discussion in this section will provide information on average FEV<sub>1</sub> responses  
7 to O<sub>3</sub> exposure as a function of age in healthy adults ranging from 18 to 50 years of age. As was  
8 specifically addressed in Section 6.4, however, there is considerable intersubject variability in  
9 responses and this between subject variability increases with increasing O<sub>3</sub> dose (*see Figure*  
10 *AX6-6*). Epidemiological studies also report such intersubject variability. For instant, Höpfe  
11 et al. (2003) observed relevant pulmonary function responses (>10% drop in FEV<sub>1</sub>, FVC,  
12 or PEF, and/or >20% increase in sRaw) subsequent to O<sub>3</sub> exposure in 50% of healthy  
13 children (n = 44; mean age, 7 yrs;  $\dot{V}_E \approx 30$  L/min), 14% of young athletes (n = 43; mean age,  
14 18 yrs;  $\dot{V}_E \approx 80$  L/min), and 27% of elderly adults (n = 41; mean age, 81 yrs;  $\dot{V}_E \approx 10$  L/min).

15 Beyond approximately 18 years of age, spirometric and symptom responses to O<sub>3</sub> exposure  
16 begin to decline with increasing age. In healthy individuals, the rate of decline in O<sub>3</sub>  
17 responsiveness appears to be greater in younger (18 to 35 yrs) versus middle aged (35 to 55 yrs)  
18 individuals (Passannante et al., 1998; Hazucha et al., 2003). Beyond this age (>55 yrs), acute O<sub>3</sub>  
19 exposure elicits minimal spirometric changes. An average FEV<sub>1</sub> decrement of ~3% has been  
20 reported by Gong et al. (1997a) for this older population under a “worst case” exposure scenario  
21 (0.24 ppm O<sub>3</sub> with 4 h IE). Although Gong et al. (1997a) and others have examined responses  
22 to O<sub>3</sub> exposure in subjects of various ages, the exposure conditions differ between most studies  
23 so that age effects remain uncertain.

24 Three recent studies, which analyzed large data sets ( $\geq 240$  subjects) of similarly exposed  
25 subjects, show clearly discernable changes in FEV<sub>1</sub> responses to O<sub>3</sub> as a function of age.  
26 Seal et al. (1996) analyzed O<sub>3</sub>-induced spirometric responses in 371 young nonsmokers  
27 (18 to 35 years of age) exposed for 2.3 h during IE at a  $\dot{V}_E$  of 25 L/min/m<sup>2</sup> BSA. On average,  
28 for the same O<sub>3</sub> concentration (C), the response of 25, 30, and 35 year old individuals are  
29 predicted to be 83, 65, and 48%, respectively, of the response in a 20 year old. For example,  
30 a 5.4% decrement in FEV<sub>1</sub> is predicted for 20 year old exposed to 0.12 ppm O<sub>3</sub> for 2.3 h

1 IE ( $\dot{V}_E = 25 \text{ L/min/m}^2 \text{ BSA}$ ), whereas, a similarly exposed 35 yr old is predicted to have only a  
2 2.6% decrement.

3 McDonnell et al. (1997) examined FEV<sub>1</sub> responses in 485 healthy white males (18 to  
4 36 years of age) exposed once for 2 h to an O<sub>3</sub> concentration of 0.0, 0.12, 0.18, 0.24, 0.30, or  
5 0.40 ppm at rest or one of two levels of IE ( $\dot{V}_E$  of 25 and 35 L/min/m<sup>2</sup> BSA). For the same  
6 exposure conditions (C,  $\dot{V}_E$ , and duration), the average responses of 25, 30, and 35 year old  
7 individuals are predicted to be 69, 48, and 33%, respectively, of the response in 20 year olds.  
8 Hazucha et al. (2003) analyzed the distribution of O<sub>3</sub> responsiveness in 240 subjects (18 to  
9 60 years of age) exposed to 0.42 ppm O<sub>3</sub> for 1.5 h with IE at  $\dot{V}_E = 20 \text{ L/min/m}^2 \text{ BSA}$ . In males,  
10 the FEV<sub>1</sub> responses of 25, 35, and 50 year olds are predicted to be 94, 83, and 50% ,  
11 respectively, of the average response in 20 year old males. In females, the FEV<sub>1</sub> responses of 25,  
12 35, and 50 year olds are predicted to be 82, 46, and 18%, respectively, of the average response in  
13 20 year old females.

14 For subjects aged 18 to 36 yrs, McDonnell et al. (1999) recently reported that symptom  
15 responses from O<sub>3</sub> exposure also decrease with increasing age. Whether the same age-dependent  
16 pattern of O<sub>3</sub> sensitivity decline also holds for airway reactivity or inflammatory endpoints has  
17 not been determined.

## 19 **6.5.2 Gender and Hormonal Influences**

20 Several studies have suggested that physiological differences between the genders may  
21 predispose females to a greater susceptibility to O<sub>3</sub>. Lower plasma and nasal lavage fluid levels  
22 of uric acid (the most prevalent antioxidant) in females relative to males may be a contributing  
23 factor (Housley et al., 1996). Consequently, reduced absorption of O<sub>3</sub> in the upper airways may  
24 promote its deeper penetration. Dosimetric measurements have shown that the absorption  
25 distribution of O<sub>3</sub> is independent of gender when absorption is normalized to anatomical dead  
26 space (Bush et al., 1996). More recently, Ultman et al. (2004) reported that the whole lung  
27 uptake fraction of O<sub>3</sub> was significantly greater in males (91.4%) than females (87.1%). But, this  
28 increase in O<sub>3</sub> uptake in the males was consistent with their larger tidal volume and slower  
29 breathing frequency relative to the females. Furthermore, O<sub>3</sub> uptake was not correlated with  
30 spirometric responses. Thus, a differential removal of O<sub>3</sub> by uric acid seems to have minimal

1 effect. In general, the spirometric responses of young healthy females to O<sub>3</sub> exposure appears  
2 comparable to the responses of young males (Hazucha et al., 2003). Although, during the  
3 follicular phase of the menstrual cycle, lung function response to O<sub>3</sub> is enhanced (Fox et al.,  
4 1993).

### 6 6.5.3 Racial, Ethnic, and Socioeconomic Status Factors

7 A few epidemiologic studies have implied that minorities are more responsive to O<sub>3</sub> than  
8 caucasians. However, this may be more of a consequence of the overall quality of health care  
9 and socioeconomic status (SES) than an innate sensitivity to oxidants (Gwynn and Thurston,  
10 2001; Seal et al, 1996). The paucity of data has prevented making any definitive conclusions on  
11 the influence of race, ethnic or other related factors on the responsiveness to O<sub>3</sub>.

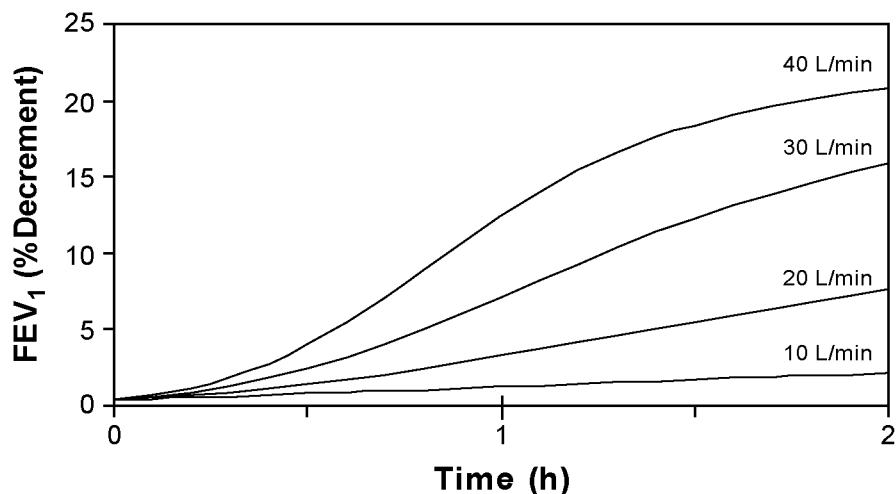
### 13 6.5.4 Influence of Physical Activity

14 Any physical activity will increase minute ventilation and therefore the dose of inhaled O<sub>3</sub>.  
15 Consequently, the intensity of physiological response following an acute exposure will be  
16 strongly associated with minute ventilation (*see Figures 6-3 and AX6-2*).

### 18 6.5.5 Environmental Factors

19 Since the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996) few human  
20 laboratory studies have examined the potential influence of environmental factors such as rural  
21 versus urban environment, passive cigarette smoke exposure, and bioactive agents such as  
22 endotoxin on healthy individual's pulmonary function changes due to O<sub>3</sub>.

23 New controlled human exposure studies have confirmed that smokers are less responsive  
24 to O<sub>3</sub> than nonsmokers. Spirometric and plethysmographic pulmonary function decline,  
25 nonspecific airway hyperreactivity, and inflammatory response of smokers to O<sub>3</sub> were all weaker  
26 than those reported for nonsmokers. Although all these responses are intrinsically related, the  
27 functional association between them, as in nonsmokers, has been weak. Similarly, the time  
28 course of development and recovery of these effects, as well their reproducibility, was not  
29 different from nonsmokers. Chronic airway inflammation with desensitization of bronchial  
30 nerve endings and an increased production of mucus may plausibly explain the pseudo-  
31 protective effect of smoking (Frampton et al., 1997; Torres et al., 1997).



**Figure 6-3. Predicted  $O_3$ -induced decrements in  $FEV_1$  as a function of exposure duration and level of IE (line labels are  $\dot{V}_E$  levels) in young healthy adults (20 yrs of age) exposed to 0.3 ppm  $O_3$ . The illustrated activity levels range from rest ( $\dot{V}_E = 10$  L/min) to moderate exercise ( $\dot{V}_E = 40$  L/min). Predictions are for Model 1 coefficients in Table 3 of McDonnell et al. (1997).**

Source: Based on McDonnell et al. (1997).

1 The effect of environmental tobacco smoke (ETS) on  $O_3$  responses has received very little  
 2 attention. In one study, preexposure of mice to sidestream cigarette smoke (ETS surrogate)  
 3 elicited no immediate effects, but potentiated subsequent  $O_3$ -induced inflammatory responses  
 4 (Yu et al., 2002) (*See Chapter 5.4.2 for additional ETS details*). Endotoxin is a biologically  
 5 active component of both mainstream and sidestream tobacco smoke (Hasday et al., 1999) which  
 6 might contribute to the potentiation of  $O_3$  effects.

7 The influence of ambient temperature on pulmonary effects induced by  $O_3$  exposure in  
 8 humans has been studied infrequently under controlled laboratory conditions. Several  
 9 experimental human studies have reported additive effects of heat and  $O_3$  exposure (see U.S.  
 10 Environmental Protection Agency, 1986, 1996). Foster et al. (2000) exposed 9 young healthy  
 11 subjects for 130 min (IE 10 min at 36 to 39 l/min) to filtered air and to ramp profile  $O_3$  at 22 °C  
 12 and 30 °C, 45-55% RH. The  $O_3$  exposure started at 0.12 ppm, reached the peak of 0.24 ppm  
 13 midway through and subsequently declined to 0.12 ppm at the end of exposure. At the end of  
 14 exposure  $FEV_1$  decreased significantly ( $p < 0.5$ ) by ~8% at 22 °C and ~6.5% at 30 °C. One day



1 (19 h) later, the decline of 2.3% from baseline was still significant ( $p < 0.05$ ) at both  
2 temperatures. FVC decrements were smaller and significant only for the 22 °C condition  
3 immediately postexposure. There was a decline in specific airway conductance (sGaw;  $p < 0.05$ )  
4 at 30 °C but not at 22 °C. The nonspecific bronchial responsiveness to methacholine assessed  
5 as PC<sub>50</sub> sGaw was significantly ( $p < 0.05$ ) higher one day following O<sub>3</sub> exposure at both  
6 temperatures but more so at 30 °C. Thus, these findings suggest that elevated temperature may  
7 partially attenuate spirometric responses but enhance airway reactivity.

### 8 9 **6.5.6 Oxidant-Antioxidant Balance**

10 The first line of defense against oxidative stress is antioxidant present in epithelial lining  
11 fluid (ELF) which scavenge free radicals and limit lipid peroxidation. Exposure to O<sub>3</sub> depletes  
12 the antioxidant level in nasal ELF probably due to scrubbing of O<sub>3</sub> (Mudway et al., 1999),  
13 however, the concentration and the activity of antioxidant enzymes either in ELF or plasma do  
14 not appear to be related to O<sub>3</sub> responsiveness (Avisar et al., 2000; Blomberg et al., 1999; Samet  
15 et al., 2001). Carefully controlled studies of dietary antioxidant supplementation have  
16 demonstrated some protective effects of  $\alpha$ -tocopherol and ascorbate on spirometric lung function  
17 from O<sub>3</sub> but not on the intensity of subjective symptoms and inflammatory response including  
18 cell recruitment, activation and release of mediators (Samet et al., 2001; Trenga et al., 2001).  
19 Dietary antioxidants have also afforded partial protection to asthmatics by attenuating post-  
20 exposure bronchial hyperresponsiveness (Trenga et al., 2001). The field studies performed in  
21 Mexico City (*described in Section 7.2.3.1*) and animal studies (*described in Section 5.2.1.3*) have  
22 also demonstrated the protective effects of ELF antioxidants during O<sub>3</sub> exposures.

### 23 24 **6.5.7 Genetic Factors**

25 Several recent studies (Yang et al., 2005; David et al., 2003; Romieu et al., 2004) have  
26 reported that genetic polymorphism of antioxidant enzymes and inflammatory genes may  
27 modulate pulmonary function and inflammatory response to O<sub>3</sub> challenge. It appears that  
28 healthy carriers of NAD(P)H:quinone oxidoreductase wild type (NQO1wt) in combination with  
29 glutathione S-transferase  $\mu$ -1 (GSTM1null) genotype are more responsive to O<sub>3</sub>. The authors  
30 have implied that the interindividual variability in O<sub>3</sub> responsiveness (FEV<sub>1</sub> changes) is related

1 to the polymorphism of these enzymes. Adults with GSTM1null only genotype did not show O<sub>3</sub>  
2 hyperresponsiveness (Bergamaschi et al., 2001). A subsequent study from the same laboratory  
3 reported a positive association between O<sub>3</sub> responsiveness, as characterized by the level of  
4 oxidative stress and inflammatory mediators (8-isoprostane, LTB<sub>4</sub> and TBARS) in EBC fluid,  
5 and the antioxidant enzyme polymorphism. However, none of the spirometric lung function  
6 endpoints were affected by ozone exposure (Corradi et al., 2002). It is of interest to note, that  
7 human nasal mucosa biopsies of GSTM1 deficient subjects showed higher antioxidant enzymes  
8 activity than biopsies of GSTM1 positive individuals when exposed to ozone (Otto-Knapp et al.,  
9 2003).

10 Asthmatic children with a genetic deficiency of GSTM1 were reported to be more  
11 responsive to ambient O<sub>3</sub> exposure, as assessed by decrements in FEF<sub>25-75</sub>, in this field study.  
12 Antioxidant supplementation (vit. C and E) attenuated post-ozone lung function response in  
13 these children (Romieu et al., 2004). More specific genotyping has shown that ozone  
14 responsiveness of asthmatic children may be related to the presence of variant Ser allele for  
15 NQO1. The presence of at least one NQO1 Ser allele in combination with GSTM1 null  
16 genotype lowered the risk of asthma in ozone exposed asthmatic children relative to Pro/Pro  
17 genotype (David et al., 2003).

18 The influence of functional polymorphism in TNF- $\alpha$ , lymphotoxin-  $\alpha$  (LTA), TLR4, SOD2  
19 and GPX1 genes on ozone-induced lung function changes in healthy individuals, mild asthmatics  
20 and subjects with rhinitis was varied. Of the inflammatory genes studied only TNF- $\alpha$  has  
21 appeared to show some promise as one of the genetic factors of susceptibility. However, as the  
22 authors stated “the functional significance of individual TNF- $\alpha$  polymorphisms remains  
23 controversial” (Yang et al., 2005).

24 These recent studies have shown that individual’s innate susceptibility to ozone may be  
25 linked to genetic background of an individual. Although a number of potential ozone  
26 susceptibility genes have been identified, additional better designed and controlled studies are  
27 needed to ascertain the link between susceptibility and polymorphism.  
28  
29  
30

## 6.6 REPEATED O<sub>3</sub> EXPOSURE EFFECTS

Based on studies reviewed here and in the previous O<sub>3</sub> criteria documents (U.S. Environmental Protection Agency, 1986, 1996), several conclusions can be drawn about repeated 1 to 2-h O<sub>3</sub> exposures. Repeated exposures to O<sub>3</sub> can cause an enhanced (i.e., greater) pulmonary function response on the second day of exposure (*see Tables AX6-8 and AX6-9 for added detail*). This enhancement appears to be dependent on the interval between the exposures (24 h is associated with the greatest increase) and is absent with intervals >3 days (Bedi et al., 1985; Folinsbee and Horvath, 1986; Schonfeld et al., 1989). An enhanced response also appears to depend to some extent on the magnitude of the initial response (Horvath et al., 1981). Small responses to the first O<sub>3</sub> exposure are less likely to result in an enhanced response on the second day of O<sub>3</sub> exposure (Folinsbee et al., 1994). With continued daily exposures (i.e., beyond the second day) there is an attenuation of pulmonary function responses, typically after 3 to 5 days of repeated exposure. This attenuated response persists for less than 1 week (Kulle et al., 1982; Linn et al., 1982b) or as long as 2 weeks (Horvath et al., 1981). In temporal conjunction with pulmonary function changes, symptoms induced by O<sub>3</sub>, such as cough, pain on deep inspiration, and chest discomfort, are increased on the second exposure day and attenuated with repeated exposure thereafter (Folinsbee et al., 1980, 1998; Foxcroft and Adams, 1986; Linn et al., 1982b). O<sub>3</sub>-induced changes in airway responsiveness persist longer and attenuate more slowly than pulmonary function and symptoms responses (Dimeo et al., 1981; Kulle et al., 1982), although this has been studied only on a limited basis (Folinsbee et al., 1994). In longer-duration (4 h to 6.6 h), lower-concentration studies that do not cause an enhanced second-day response, the attenuation of response to O<sub>3</sub> appears to proceed more rapidly (Folinsbee et al., 1994) [*Effects of repeated exposures on inflammatory responses are discussed in Section 6.9.4*].

## 6.7 EFFECTS ON EXERCISE PERFORMANCE

The effects of acute O<sub>3</sub> inhalation on endurance exercise performance have been examined in numerous controlled laboratory studies. These studies were discussed in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996) and can be divided into two categories: (1) those that examined the effects of acute O<sub>3</sub> inhalation on maximal oxygen uptake ( $\dot{V}O_{2\max}$ ) and

1 (2) those that examined the effects of acute O<sub>3</sub> inhalation on the ability to complete strenuous  
2 continuous exercise protocols of up to 1 h in duration.

3 In brief, endurance exercise performance and  $\dot{V}O_{2\max}$  may be limited by acute exposure  
4 to O<sub>3</sub> (Adams and Schelegle, 1983; Schelegle and Adams, 1986; Gong et al., 1986; Foxcroft and  
5 Adams, 1986; Folinsbee et al., 1977; Linder et al., 1988). Gong et al. (1986) and Schelegle and  
6 Adams (1986) found that significant reductions in maximal endurance exercise performance may  
7 occur in well-conditioned athletes while they perform CE ( $\dot{V}_E > 80$  L/min) for 1 h at O<sub>3</sub>  
8 concentrations  $\geq 0.18$  ppm. Reports from studies of exposure to O<sub>3</sub> during high-intensity  
9 exercise indicate that breathing discomfort associated with maximal ventilation may be an  
10 important factor in limiting exercise performance in some, but not all, subjects.

## 13 **6.8 EFFECTS ON AIRWAY RESPONSIVENESS**

14 Airway or bronchial hyperresponsiveness (BHR) refers to a condition in which the  
15 propensity for the airways to bronchoconstrict due to a variety of stimuli becomes augmented.  
16 Airway responsiveness is typically quantified by measuring the decrement in pulmonary  
17 function (i.e., spirometry or plethysmography) following the inhalation of small amounts of an  
18 aerosolized bronchoconstrictor agent (specific [antigen, allergen] or nonspecific [methacholine,  
19 histamine]) or a measured stimulus (e.g., exercise, cold air).

20 Ozone exposure causes an increase in nonspecific airway responsiveness as indicated by a  
21 reduction in the concentration of methacholine or histamine required to produce a given  
22 reduction in FEV<sub>1</sub> or increase in SRaw. Increased airway responsiveness is an important  
23 consequence of exposure to O<sub>3</sub> because its presence means that the airways are predisposed to  
24 narrowing on inhalation of a variety of stimuli (e.g., specific allergens, SO<sub>2</sub>, cold air).

25 Ozone exposure of asthmatic subjects, who characteristically have increased airway  
26 responsiveness at baseline, can cause further increases in responsiveness (Kreit et al., 1989).  
27 Similar relative changes in airway responsiveness are seen in asthmatics exposed to O<sub>3</sub> despite  
28 their markedly different baseline airway responsiveness. Several studies (Jörres et al., 1996;  
29 Kehrl et al., 1999; Molfino et al., 1991) have been published suggesting an increase in specific  
30 (i.e., allergen-induced) airway reactivity. An important aspect of increased airway

1 responsiveness after O<sub>3</sub> exposure is that this represents a plausible link between ambient O<sub>3</sub>  
2 exposure and increased hospital admissions for asthma.

3 Changes in airway responsiveness after O<sub>3</sub> exposure appear to be resolved more slowly  
4 than changes in FEV<sub>1</sub> or respiratory symptoms (Folinsbee and Hazucha, 2000). Furthermore, in  
5 studies of repeated exposure to O<sub>3</sub>, changes in airway responsiveness tend to be somewhat less  
6 susceptible to attenuation with consecutive exposures than changes in FEV<sub>1</sub> (Dimeo et al., 1981;  
7 Folinsbee et al., 1994; Gong et al., 1997b; Kulle et al., 1982). Increases in airway  
8 responsiveness do not appear to be strongly associated with decrements in lung function or  
9 increases in symptoms.

10 The mechanism of O<sub>3</sub>-induced increases in airway responsiveness is only partially  
11 understood, but it appears to be associated with a number of cellular and biochemical changes  
12 in airway tissue. Although inflammation could play a role in the increase in airway  
13 responsiveness, cyclooxygenase inhibitors have not been effective at blocking the O<sub>3</sub>-induced  
14 influx of PMNs into bronchoalveolar lavage (BAL) fluid (Hazucha et al., 1996; Ying et al.,  
15 1990). Therefore, O<sub>3</sub>-induced airway responsiveness may not be due to the presence of PMNs  
16 in the airway or to the release of arachidonic acid metabolites. Rather, it seems likely that the  
17 mechanism for this response is multifactorial, possibly involving the presence of cytokines,  
18 prostanoids, or neuropeptides; activation of macrophages, eosinophils, or mast cells; and  
19 epithelial damage that increases direct access of mediators to the smooth muscle or receptors in  
20 the airways that are responsible for reflex bronchoconstriction.

## 23 **6.9 EFFECTS ON INFLAMMATION AND HOST DEFENSE**

### 24 **6.9.1 Introduction**

25 Short-term exposure of humans to O<sub>3</sub> can cause acute inflammation and long-term  
26 exposure of laboratory animals results in a chronic inflammatory state (*see Chapter 5*). The  
27 relationship between repetitive bouts of acute inflammation in humans caused by O<sub>3</sub> and the  
28 development of chronic respiratory disease is unknown.

29 The presence of neutrophils (PMNs) in the lung has long been accepted as a hallmark of  
30 inflammation and is an important indicator that O<sub>3</sub> causes inflammation in the lungs. It is  
31 apparent, however, that inflammation within airway tissues may persist beyond the point that

1 inflammatory cells are found in BAL fluid (BALF). Soluble mediators of inflammation such as  
2 the cytokines (IL-6, IL-8) and arachidonic acid metabolites (e.g., PGE<sub>2</sub>, PGF<sub>2α</sub>, thromboxane,  
3 and leukotrienes [LTs] such as LTB<sub>4</sub>) have been measured in the BAL fluid of humans exposed  
4 to O<sub>3</sub>. In addition to their role in inflammation, many of these compounds have  
5 bronchoconstrictive properties and may be involved in increased airway responsiveness  
6 following O<sub>3</sub> exposure.

7 Some recent evidence suggests that changes in small airways function may provide a  
8 sensitive indicator of O<sub>3</sub> exposure and effect, despite the fact that inherent variability in their  
9 measurement by standard spirometric approaches make their assessment difficult (Frank et al.,  
10 2001). Observations of increased functional responsiveness of these areas relative to the more  
11 central airways, and of persistent effects following repeated exposure, may indicate that further  
12 investigation of inflammatory processes in these regions is warranted.

## 14 **6.9.2 Inflammatory Responses in the Upper Respiratory Tract**

15 The nasal passages constitute the primary portal for inspired air at rest and, therefore,  
16 the first region of the respiratory tract to come in contact with airborne pollutants. Nikasinovic  
17 et al. (2003) recently reviewed the literature of laboratory-based nasal inflammatory studies  
18 published since 1985. Nasal lavage (NL) has provided a useful tool for assessing O<sub>3</sub>-induced  
19 inflammation in the nasopharynx. Increased levels of PMNs in the NL fluid of humans exposed  
20 to 0.5 ppm O<sub>3</sub> at rest for 4 h has been reported (Graham et al., 1988; Bascom et al., 1990).

21 Graham and Koren (1990) compared inflammatory mediators present in both the NL and  
22 BAL fluids of humans exposed to 0.4 ppm O<sub>3</sub> for 2 h. Similar increases in PMN were observed  
23 in NL and BAL, suggesting a qualitative correlation between inflammatory changes in the lower  
24 airways (BAL) and the upper respiratory tract (NL). Torres et al. (1997) compared NL and BAL  
25 in smokers and nonsmokers exposed to 0.22 ppm O<sub>3</sub> for 4 h. In contrast to Graham and Koren  
26 (1990), they did not find a relationship between numbers or percentages of PMNs in the nose  
27 and the lung, perhaps in part due to the variability observed in their NL recoveries. Albumin, a  
28 marker of epithelial cell permeability, was increased 18 h later, but not immediately after  
29 exposure, as seen by Bascom et al. (1990).

30 McBride et al. (1994) reported that asthmatic subjects were more sensitive than  
31 nonasthmatics to upper airway inflammation at an O<sub>3</sub> concentration (0.24 ppm for 1.5 h with

1 light IE) that did not affect pulmonary function. In the asthmatics, there was a significant  
2 increase in the number of PMNs in NL fluid both immediately and 24 h after exposure. Peden  
3 et al. (1995) also found that exposure to 0.4 ppm O<sub>3</sub> had a direct nasal inflammatory effect and a  
4 priming effect on response to nasal allergen challenge. A subsequent study in dust  
5 mite-sensitive asthmatic subjects indicated that O<sub>3</sub> at this concentration enhanced eosinophil  
6 influx in response to allergen but did not promote early mediator release or enhance the nasal  
7 response to allergen (Michelson et al., 1999). Similar to observations made in the lower airways,  
8 the presence of O<sub>3</sub> molecular “targets” in nasal lining fluid is likely to provide some level of  
9 local protection against exposure. In a study of healthy subjects exposed to 0.2 ppm O<sub>3</sub> for 2 h,  
10 Mudway and colleagues (1999) observed a significant depletion of uric acid in NL fluid at 1.5 h  
11 following exposure.

### 13 **6.9.3 Inflammatory Response in the Lower Respiratory Tract**

14 As reviewed in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996), acute  
15 exposure to O<sub>3</sub> results in an inflammatory reaction, increased epithelial cell permeability, and  
16 may stimulate fibrogenic processes. Inflammatory markers are observed in BALF of healthy  
17 subjects by 1 h post-O<sub>3</sub> exposure and may persist for at least 18 to 24 h. Not all inflammatory  
18 markers, however, follow the same time course. Studies published since the 1996 O<sub>3</sub> AQCD  
19 support these earlier findings.

20 Inflammatory effects have been assessed *in vivo* by lavage (proximal airway and  
21 bronchoalveolar), bronchial biopsy, and more recently, induced sputum. *In vitro* studies of  
22 human alveolar macrophages (AM) and airway epithelial cells exposed to O<sub>3</sub> suggest that most  
23 mediators found in the BALF of O<sub>3</sub>-exposed humans are produced by epithelial cells (U.S.  
24 Environmental Protection Agency, 1996). Recent evidence suggests that the release of  
25 mediators from AMs may be modulated by the products of O<sub>3</sub>-induced oxidation of airway  
26 lining fluid components, such as human surfactant protein A (Wang et al., 2002).

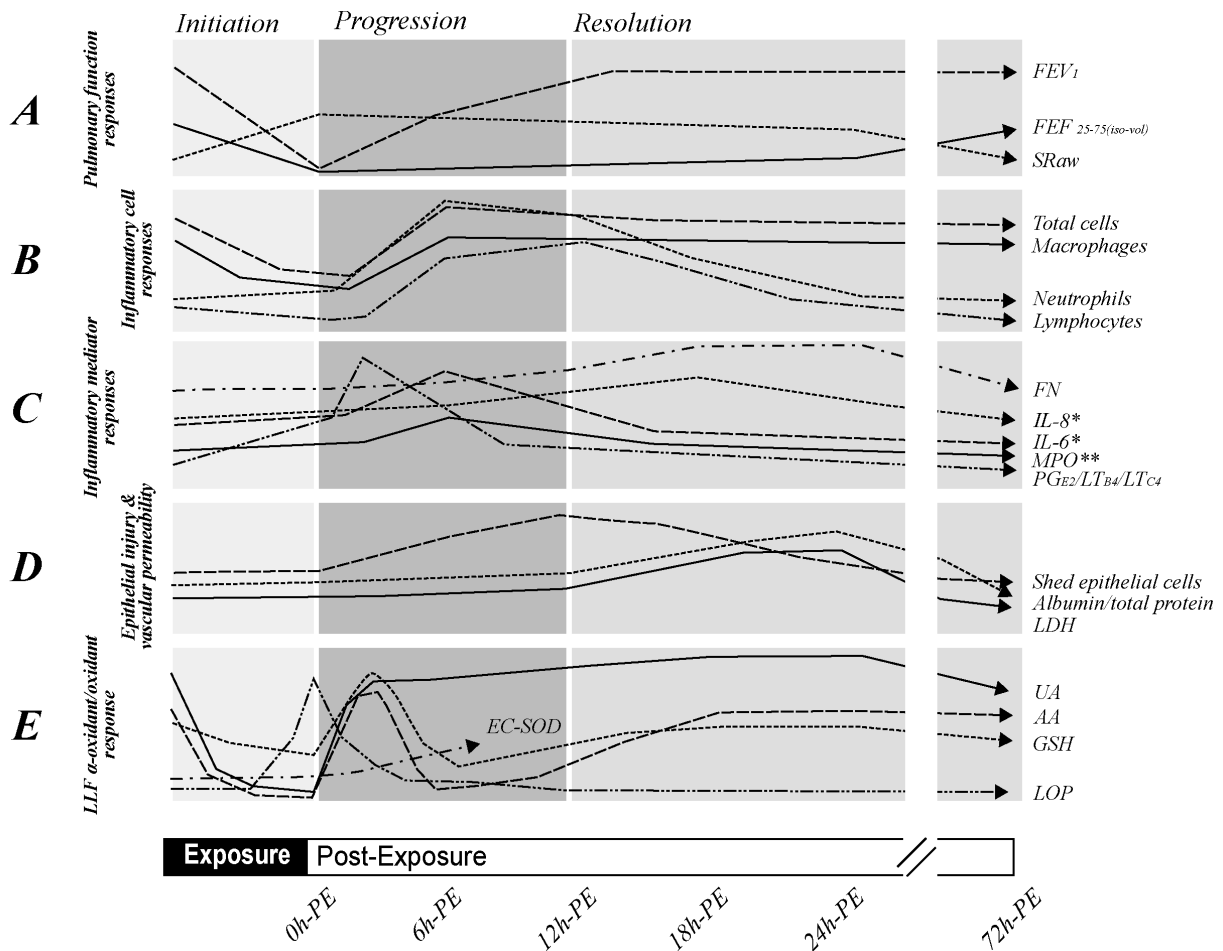
27 Spirometric responses to O<sub>3</sub> are independent from inflammatory responses and markers of  
28 epithelial injury (Balme et al., 1996; Blomberg et al., 1999; Hazucha et al., 1996; Torres et al.,  
29 1997). Significant inflammatory responses to O<sub>3</sub> exposures that did not elicit significant  
30 spirometric responses have been reported (Holz et al., 2005; McBride et al., 1994). A meta-

1 analysis of 21 studies (Mudway and Kelly, 2004), showed that PMN influx in health subjects is  
2 associated with total O<sub>3</sub> dose (product of O<sub>3</sub> concentration, exposure duration, and  $\dot{V}_E$ ).

3 The time course of the inflammatory response to O<sub>3</sub> in humans has not been fully  
4 characterized. From a review of the literature by Mudway and Kelly (2000), Figure 6-4  
5 illustrates a plausible time course of acute O<sub>3</sub> responses. As the figure shows, different markers  
6 have peak responses at different times. Studies in which lavages were performed 1 h after O<sub>3</sub>  
7 exposure (1 h at 0.4 ppm or 4 h at 0.2 ppm) have demonstrated that the inflammatory responses  
8 are quickly initiated (Devlin et al., 1996; Schelegle et al., 1991; Torres et al., 1997).  
9 Inflammatory mediators and cytokines such as IL-8, IL-6, and PGE<sub>2</sub> are greater at 1 h than at  
10 18 h post-O<sub>3</sub> exposure (Devlin et al., 1996; Torres et al., 1997). However, IL-8 still remains  
11 elevated at 18 h post-O<sub>3</sub> (4 h at 0.2 ppm O<sub>3</sub> versus FA) in healthy subjects and more so in  
12 asthmatics (Balmes et al., 1996; Scannell et al., 1996). Schelegle et al. (1991) found increased  
13 PMNs in the “proximal airway” lavage at 1, 6, and 24 h after O<sub>3</sub> exposure (4 h at 0.2 ppm O<sub>3</sub>),  
14 with a peak response at 6 h. Although, at 18 to 24 h after O<sub>3</sub> exposure, PMNs remain elevated  
15 relative to 1 h postexposure (Schelegle et al., 1991; Torres et al., 1997). In addition to the influx  
16 of PMNs and (in allergic asthmatics) eosinophils, lymphocyte numbers in BALF are elevated  
17 significantly at 6 h following exposure (2 h at 0.2 ppm O<sub>3</sub>) of healthy subjects (Blomberg et al.,  
18 1997). Flow cytometry also indicated the increased presence of CD3+, CD4+ and CD8+ T cell  
19 subsets. This same laboratory later demonstrated that within 1.5 h following exposure of healthy  
20 subjects to the same O<sub>3</sub> regimen, expression of human leukocyte antigen (HLA)-DR on lavaged  
21 macrophages underwent a significant, 2.5-fold increase (Blomberg et al., 1999).

22 The inflammatory responses to O<sub>3</sub> exposure have also been studied in asthmatic subjects  
23 (Basha et al., 1994; Scannell et al., 1996; Peden et al., 1997). In these studies, asthmatics  
24 showed significantly more neutrophils in the BALF (18 h post-exposure) than similarly exposed  
25 healthy individuals. In one of these studies (Peden et al., 1997), which included only allergic  
26 asthmatics who tested positive for *Dematophagoides farinae* antigen, there was an eosinophilic  
27 inflammation (2-fold increase), as well as neutrophilic inflammation (3-fold increase). In a  
28 study of subjects with intermittent asthma exposed to 0.4 ppm O<sub>3</sub> for 2 h, increases in eosinophil  
29 cationic protein, neutrophil elastase and IL-8 were found to be significantly increased 16 h post-  
30 exposure and comparable in induced sputum and BALF (Hiltermann et al, 1999). Scannell et al.  
31 (1996) also reported that IL-8 tends to be higher in the BALF of asthmatics compared to





**Figure 6-4.** Time course of acute responses seen in humans exposed to O<sub>3</sub>. Responses are divided into three phases: initiation, during O<sub>3</sub> exposure; progression, where responses develop post-O<sub>3</sub> exposure; and resolution, during which responses return baseline levels. \*The IL-8 response is shown with a progressive increase peaking at 18 h postexposure (18h-PE). The IL-8 literature is inconclusive and neutrophils have been demonstrated in the absence of an IL-8 increase. \*\*Few studies have measured MPO, however, a trend for an increase at 6h-PE has been reported.

**Abbreviations:** LOP (lipid ozonation products), GSH (reduced glutathione), AA (ascorbic acid), UA (uric acid), LDH (lactate dehydrogenase), PGE<sub>2</sub> (prostaglandin E2), LT<sub>B4</sub> (leukotriene B4), LT<sub>C4</sub> (leukotriene C4), MPO (myeloperoxidase).

Source: Reprinted from Mudway and Kelly (2000) with permission from Elsevier.

1 nonasthmatics following O<sub>3</sub> exposure, suggesting a possible mediator for the significantly  
2 increased neutrophilic inflammation in those subjects. Bosson et al. (2003) found significantly  
3 greater the epithelial expression of IL-5, IL-8, granulocyte-macrophage colony-stimulating  
4 factor (GM-CSF) and epithelial cell-derived neutrophil-activating peptide 78 (ENA-78) in  
5 asthmatics compared to healthy subjects following exposure to 0.2 ppm O<sub>3</sub> for 2 h. Stenfors and  
6 colleagues (2002) were unable to detect a difference in the increased neutrophil numbers  
7 between 15 mild asthmatic and 15 healthy subjects by bronchial wash at the 6 h post-exposure  
8 time point. However, the asthmatics were on average 5 years older than the healthy subjects in  
9 this study and it is not yet known how age affects inflammatory responses. It is also possible  
10 that the time course of neutrophil influx differs between healthy and asthmatic individuals.

11 Vagaggini et al. (2002) investigated the effect of prior allergen challenge on responses in  
12 mild asthmatics exposed for 2 h to 0.27 ppm O<sub>3</sub> or filtered air. At 6 h post-exposure, eosinophil  
13 numbers in induced sputum were found to be significantly greater after O<sub>3</sub> than after air. Studies  
14 such as these suggest that the time course of eosinophil and neutrophil influx following O<sub>3</sub>  
15 exposure can occur to levels detectable within the airway lumen by as early as 6 h. They also  
16 suggest that the previous or concurrent activation of proinflammatory pathways within the  
17 airway epithelium may enhance the inflammatory effects of O<sub>3</sub>. For example, in an *in vitro*  
18 study of epithelial cells from the upper and lower respiratory tract, cytokine production induced  
19 by rhinovirus infection was enhanced synergistically by concurrent exposure to O<sub>3</sub> at 0.2 ppm for  
20 3 h (Spannhake et al, 2002).

21 Although the release of mediators has been demonstrated to occur at exposure  
22 concentrations and times that are minimally cytotoxic to airway cells, potentially detrimental  
23 latent effects have been demonstrated in the absence of cytotoxicity. These include the  
24 generation of DNA single strand breaks (Kozumbo et al., 1996), the loss of cellular replicative  
25 activity (Gabrielson et al., 1994) in bronchial epithelial cells exposed *in vitro*, and the formation  
26 of protein and DNA adducts. A highly toxic aldehyde formed during O<sub>3</sub>-induced lipid  
27 peroxidation is 4-hydroxynonenal (HNE). Healthy human subjects exposed to 0.4 ppm O<sub>3</sub> for  
28 1 h underwent BAL 6 h later. Analysis of lavaged AMs by Western Blot indicated increased  
29 levels of a 32-kDa HNE-protein adduct, as well as 72-kDa heat shock protein and ferritin  
30 in O<sub>3</sub>-versus air-exposed subjects (Hamilton et al., 1998). In a recent study of healthy subjects  
31 exposed to 0.1 ppm O<sub>3</sub> for 2 h (Corradi et al., 2002), formation of 8-hydroxy-2'-deoxyguanosine

1 (8-OHdG), a biomarker of reactive oxidant species (ROS)-DNA interaction, was measured in  
2 peripheral blood lymphocytes. At 18 h post exposure, 8-OHdG was significantly increased in  
3 cells compared to pre-exposure levels, presumably linked to concurrent increases in chemical  
4 markers of ROS. Of interest, the increase in 8-OHdG was only significant in a subgroup of  
5 subjects with the wild genotype for NQ01 and the null genotype for GSTM1, suggesting that  
6 polymorphisms in redox enzymes may confer “susceptibility” to O<sub>3</sub> in some individuals.

7 The generation of ROS following exposure to O<sub>3</sub> has been shown to be associated with a  
8 wide range of responses. In a recent study, ROS production by alveolar macrophages lavaged  
9 from subjects exposed to 0.22 ppm for 4 h was assessed by flow cytometry (Voter et al., 2001).  
10 Levels were found to be significantly elevated 18 h post exposure and associated with several  
11 markers of increased permeability. An *in vitro* study of human tracheal epithelial cells exposed  
12 to O<sub>3</sub> indicated that generation of ROS resulted in decrease in synthesis of the bronchodilatory  
13 prostaglandin, PGE<sub>2</sub>, as a result of inactivation of prostaglandin endoperoxide G/H synthase 2  
14 (Alpert et al., 1997).

#### 16 **6.9.4 Adaptation of Inflammatory Responses**

17 Physiologic and symptomatic responses in humans following repeated exposure to O<sub>3</sub> were  
18 discussed in Section 6.6. Inflammatory responses upon repeated O<sub>3</sub> exposures are discussed in  
19 this section. Animal studies suggest that while inflammation may be diminished with repeated  
20 exposure, underlying damage to lung epithelial cells continues (Tepper et al., 1989). Markers  
21 from BALF following both 2-h (Devlin et al., 1997) and 4-h (Christian et al., 1998; Jörres et al.,  
22 2000) repeated O<sub>3</sub> exposures (up to 5 days) indicate that there is ongoing cellular damage  
23 irrespective of the attenuation of some cellular inflammatory responses of the airways,  
24 pulmonary function, and symptom responses.

25 Devlin et al. (1997) examined the inflammatory responses of humans repeatedly exposed  
26 to 0.4 ppm O<sub>3</sub> for 5 consecutive days. Several indicators of inflammation (e.g., PMN influx,  
27 IL-6, PGE<sub>2</sub>, BAL protein, fibronectin) were attenuated after 5 days of exposure (i.e., values were  
28 not different from FA). Several markers (LDH, IL-8, total protein, epithelial cells) did not show  
29 attenuation, indicating that tissue damage probably continues to occur during repeated exposure.  
30 The recovery of the inflammatory response occurred for some markers after 10 days, but some  
31 responses were not normalized even after 20 days. The continued presence of cellular injury

1 markers indicates a persistent effect that may not necessarily be recognized due to the  
2 attenuation of spirometric and symptom responses.

3 Christian et al. (1998) randomly subjected healthy subjects to a single exposure and to 4  
4 consecutive days of exposure to 0.2 ppm O<sub>3</sub> for 4 h. Both “bronchial” and “alveolar” fractions  
5 of the BAL showed decreased numbers of PMNs and fibronectin concentration at day 4 versus  
6 the single exposure, and a decrease in IL-6 levels in the alveolar fraction. Following a similar  
7 study design and exposure parameters, Jörres et al. (2000) found both functional and BAL  
8 cellular responses to O<sub>3</sub> were abolished at 24 h postexposure following the fourth exposure day.  
9 However, levels of total protein, IL-6, IL-8, reduced glutathione and ortho-tyrosine were still  
10 increased significantly. In addition, visual scores for bronchitis, erythema and the numbers of  
11 neutrophils in the mucosal biopsies were increased. Their results indicate that, despite reduction  
12 of some markers of inflammation in BAL and measures of large airway function, inflammation  
13 within the airways persists following repeated exposure to O<sub>3</sub>.

14 Holz et al. (2002) made a comparison of early and late responses to allergen challenge  
15 following O<sub>3</sub> in subjects with allergic rhinitis or allergic asthma. With some variation, both early  
16 and late FEV<sub>1</sub> and cellular responses in the two subject groups were significantly enhanced by 4  
17 consecutive days of exposure to 0.125 ppm O<sub>3</sub> for 3 h.

18 In another study, Frank and colleagues (2001) exposed healthy subjects to FA and to O<sub>3</sub>  
19 (0.25 ppm, 2 h) on 4 consecutive days each, with pulmonary function measurements being made  
20 prior to and following each exposure. BAL was performed on day 5, 24 h following the last  
21 exposure. On day 5, PMN numbers remained significantly higher following O<sub>3</sub> compared to FA.  
22 Of particular note in this study was the observation that small airway function, assessed by  
23 grouping values for isovolumetric FEF<sub>25-75</sub>, Vmax50 and Vmax75 into a single value, showed  
24 persistent reduction from day 2 through day 5. These data suggest that techniques monitoring  
25 the function in the small peripheral airway regions, the primary sites of O<sub>3</sub> uptake in the lung,  
26 may provide important information regarding both acute and cumulative effects of O<sub>3</sub> exposure.

### 27 28 **6.9.5 Effect of Anti-Inflammatory and Other Mitigating Agents**

29 Pretreatment of healthy subjects with non-steroidal anti-inflammatory drugs (ibuprofen,  
30 etc.) has been found to partially suppress development of airway inflammation and pulmonary  
31 function changes (U.S. Environmental Protection Agency, 1996). Although atropine blocked the

1 increase in Raw in response to O<sub>3</sub> exposure, it did not alter the spirometric or symptom  
2 responses (Beckett et al., 1985). Similarly, albuterol and salbutamol, which had no effect  
3 on O<sub>3</sub>-induced changes in spirometry, also had no effect of symptom responses (McKenzie et al.,  
4 1987; Gong et al., 1988). The anti-inflammatory medications indomethacin and ibuprofen,  
5 which partially inhibit the spirometric responses to O<sub>3</sub> exposure, also cause a reduction in  
6 respiratory symptoms (Schelegle et al., 1987; Hazucha et al., 1994). Indomethacin attenuates  
7 decrements in FEV<sub>1</sub> and FVC in healthy subjects, but not asthmatics (Alexis et al.,  
8 2000). In contrast, inhalation of the corticosteroid budesonide does not prevent or even  
9 attenuate O<sub>3</sub>-induced responses in healthy subjects as assessed by measurements of lung  
10 function, bronchial reactivity and airway inflammation (Nightingale et al., 2000). In asthmatic  
11 subjects, budesonide decreases airway neutrophil influx following O<sub>3</sub> exposure (Vagaggini  
12 et al., 2001). This suggests that corticosteroids may be effective only when the inflammation  
13 is already present, such as in asthmatics.

14 Holz et al. (2005) studied inflammatory responses in healthy ozone-responders (>10%  
15 increase in sputum neutrophils from O<sub>3</sub>) pretreated with single doses (the highest shown to be  
16 safe and well tolerated) of inhaled fluticasone and oral prednisolone. The O<sub>3</sub> exposure caused  
17 small changes in FEV<sub>1</sub> (-3.6% ± 6.8%) that were not significantly different from baseline or  
18 between treatment groups (i.e., prescreening, placebo, fluticasone, and prednisolone). Relative  
19 to placebo, the inhaled or oral corticosteroids significantly reduced O<sub>3</sub>-induced neutrophil levels.  
20 These authors note that their study design was intended to test the anti-inflammatory effects of  
21 the steroids and that such high-dose regimens should not be considered for potential long-term  
22 patient treatment.

### 23 **6.9.6 Changes in Host Defense Capability Following Ozone Exposures**

24 A number of studies clearly show that a single acute exposure (1 to 4 h) of humans to  
25 moderate concentrations of O<sub>3</sub> (0.2 to 0.6 ppm) while exercising at moderate to heavy levels  
26 results in a number of cellular and biochemical changes in the lung including an inflammatory  
27 response characterized by increased numbers of PMNs, increased permeability of the epithelial  
28 cells lining the respiratory tract, cell damage, and production of proinflammatory cytokines and  
29 prostaglandins. This response can be detected as early as 1 h after exposure (Koren et al., 1991;  
30 Schelegle et al., 1991) and persists for at least 18 h (Aris et al., 1993; Koren et al., 1989). The  
31

1 response profile of these mediators is not defined adequately, although it is clear that the time  
2 course of response varies for different mediators and cells (Devlin et al., 1997; Schelegle et al.,  
3 1991). These changes also occur in humans exposed to 0.08 and 0.10 ppm O<sub>3</sub> for 6 to 8 h  
4 (Devlin et al., 1991; Peden et al., 1997). Ozone also causes inflammatory changes in the nose, as  
5 indicated by increased levels of PMNs and albumin, a marker for increased epithelial cell  
6 permeability. Nasal lavage analyses, however, are not necessarily parallel to BAL analyses.

7       There appears to be no strong correlation between any of the measured cellular and  
8 biochemical changes and changes in lung function measurements, suggesting that different  
9 mechanisms may be responsible for these processes (Balmes et al., 1996; Devlin et al., 1991).  
10 The idea of different mechanisms is supported by a study in which ibuprofen, a cyclooxygenase  
11 inhibitor, blunted the O<sub>3</sub>-induced decrements in lung function without altering the O<sub>3</sub>-induced  
12 increase in PMNs or epithelial cell permeability (Hazucha et al., 1996). In vitro studies suggest  
13 that epithelial cells are the primary target of O<sub>3</sub> in the lung and that O<sub>3</sub> induces them to produce  
14 many of the mediators found in the BAL fluid of humans exposed to O<sub>3</sub>. Although O<sub>3</sub> does not  
15 induce AMs to produce these compounds in large quantities, it does directly impair the ability of  
16 AMs to phagocytize and kill microorganisms.

17       A number of studies have found that O<sub>3</sub> exposures increases epithelial cell permeability  
18 through direct (technetium-99m labeled diethylene triamine pentaacetic acid, <sup>99m</sup>Tc-DTPA,  
19 clearance) and indirect (e.g. increased BAL albumin, protein) techniques. Kehrl et al. (1987)  
20 showed increased <sup>99m</sup>Tc-DTPA clearance in healthy young adults at 75 minutes postexposure  
21 to 0.4 ppm O<sub>3</sub> for 2 h. More recently, Foster and Stetkiewicz (1996) have shown that  
22 increased <sup>99m</sup>Tc-DTPA clearance persists for at least 18-20 h post-O<sub>3</sub> exposure (130 min to  
23 average O<sub>3</sub> concentration of 0.24 ppm) and the effect is greater at the lung apices than at the  
24 base. Increased BAL protein, suggesting O<sub>3</sub>-induced changes in epithelial permeability, have  
25 also been reported at 1 h and 18 h postexposure (Balmes et al., 1996; Devlin et al., 1997). A  
26 recent meta-analysis of results from 21 publications (Mudway and Kelly, 2004), showed that  
27 increased BAL protein is associated with total ozone dose (product of O<sub>3</sub> concentration,  
28 exposure duration, and  $\dot{V}_E$ ). Changes in permeability associated with acute inflammation may  
29 provide increased access of inhaled antigens, particles, and other substances to the smooth  
30 muscle, interstitial cells, and the blood.

1 In addition to affecting epithelial permeability and AM-mediated clearance in the  
2 respiratory region of the lung, mucociliary clearance of the tracheobronchial airways is also  
3 affected by O<sub>3</sub> exposure. Only two studies (Foster et al., 1987; Gerrity et al., 1993) have  
4 investigated the effect of O<sub>3</sub> exposure on mucociliary particle clearance in humans. Foster et al.  
5 (1987) measured clearance during and after a 2 h exposure to 0.4 ppm O<sub>3</sub>. Gerrity et al. (1993)  
6 measured clearance at 2 h postexposure (0.4 ppm O<sub>3</sub>), by which time, sRaw had returned to  
7 baseline and FVC was within 5% of baseline (versus an 11% decrement immediately  
8 postexposure). Foster et al. (1987) found a stimulatory effect of acute O<sub>3</sub> exposure on  
9 mucociliary clearance. Gerrity et al. (1993), who observed no effect on clearance, suggested that  
10 transient clearance increases are coincident to pulmonary function responses. Investigators in  
11 both studies suggested that O<sub>3</sub>-induced increases in mucociliary clearance could be mediated by  
12 cholinergic receptors.

## 15 **6.10 EXTRAPULMONARY EFFECTS OF OZONE**

16 Ozone reacts rapidly on contact with respiratory system tissue and is not absorbed or  
17 transported to extrapulmonary sites to any significant degree as such. Human exposure studies  
18 discussed in the previous criteria documents (U.S. Environmental Protection Agency, 1986,  
19 1996) failed to demonstrate any consistent extrapulmonary effects. More recently, some human  
20 exposure studies have attempted to identify specific markers of exposure to O<sub>3</sub> in blood. Foster  
21 et al. (1996) found a reduction in the serum levels of the free radical scavenger  $\alpha$ -tocopherol  
22 after O<sub>3</sub> exposure. Liu et al. (1997, 1999) used a salicylate metabolite, 2,3, dehydroxybenzoic  
23 acid (DHBA), to indicate increased levels of hydroxyl radical which hydroxylates salicylate to  
24 DHBA. Increased DHBA levels after exposure to 0.12 and 0.40 ppm suggest that O<sub>3</sub> increases  
25 production of hydroxyl radical. The levels of DHBA were correlated with changes in  
26 spirometry.

27 Gong et al. (1998) observed a statistically significant O<sub>3</sub>-induced increase the alveolar-to-  
28 arterial PO<sub>2</sub> gradient in both healthy (n = 6) and hypertensive (n = 10) adult males (41-78 years  
29 old) exposed for 3 h with IE ( $\dot{V}_E \approx 30$  L/min) to 0.3 ppm O<sub>3</sub>. The mechanism for the decrease in  
30 arterial oxygen tension in the Gong et al. (1998) study could be due to an O<sub>3</sub>-induced ventilation-  
31 perfusion mismatch. Foster et al. (1993) has demonstrated that even in relatively young healthy

1 adults ( $26.7 \pm 7$  years old), O<sub>3</sub> exposure can cause ventilation to shift away from the well  
2 perfused basal lung. This effect of O<sub>3</sub> on ventilation distribution [and by association the small  
3 airways] may persist beyond 24-h post-exposure (Foster et al., 1997). Gong et al. (1998)  
4 suggested that by impairing alveolar-arterial oxygen transfer, the O<sub>3</sub> exposure could potentially  
5 lead to adverse cardiac events by decreasing oxygen supply to the myocardium. The subjects in  
6 the Gong et al. (1998) study had sufficient functional reserve so as to not experience significant  
7 ECG changes or myocardial ischemia and/or injury.

8 Effects of O<sub>3</sub> exposure on alveolar-arterial oxygen gradients may be more pronounced in  
9 patients with preexisting obstructive lung diseases. Relative to healthy elderly subjects, COPD  
10 patients have reduced gas exchange and low SaO<sub>2</sub>. Any inflammatory or edematous responses  
11 due to O<sub>3</sub> delivered to the well ventilated regions of the COPD lung could further inhibit gas  
12 exchange and reduce oxygen saturation. In addition, O<sub>3</sub>-induced vasoconstriction could also  
13 acutely induce pulmonary hypertension. Inducing pulmonary vasoconstriction and hypertension  
14 in these patients would perhaps worsen their condition, especially if their right ventricular  
15 function was already compromised.

## 18 **6.11 EFFECTS OF OZONE MIXED WITH OTHER POLLUTANTS**

19 Over the past 10 years only a handful of human controlled studies have examined the  
20 effects of pollutant mixtures containing O<sub>3</sub>. The studies summarized in this section complement  
21 the studies reviewed in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996).  
22 *(The complexities of O<sub>3</sub> and co-pollutant exposures in animal studies are discussed in*  
23 *Section 5.4.4).*

24 The results of a controlled study on children (Linn et al., 1997), designed to approximate  
25 exposure conditions of an epidemiologic study (Neas et al., 1995) by matching the population  
26 and exposure atmosphere (0.1 ppm O<sub>3</sub>, 0.1 ppm SO<sub>2</sub> and 101 µg/m<sup>2</sup> H<sub>2</sub>SO<sub>4</sub>), did not support the  
27 findings of this epidemiologic study. The study points out the difficulties in attempting to link  
28 the outcomes of epidemiologic and controlled studies. Another vulnerable population,  
29 asthmatics, demonstrated enhanced airway reactivity to house dust mite following exposures  
30 to O<sub>3</sub>, NO<sub>2</sub>, and the combination of the two gases. Spirometric response, however, was impaired  
31 only by O<sub>3</sub>, and O<sub>3</sub> + NO<sub>2</sub> at higher concentrations (Jenkins et al., 1999). Continuous exposure



1 to SO<sub>2</sub> and NO<sub>2</sub> increases inhaled bolus O<sub>3</sub> absorption, while continuous exposure to O<sub>3</sub>  
2 decreases O<sub>3</sub> bolus absorption (Rigas et al., 1997). Inhalation of a mixture of PM<sub>2.5</sub> and O<sub>3</sub> by  
3 healthy subjects increased brachial artery tone and reactivity (Brook et al., 2002). Since no other  
4 cardiovascular endpoints were affected by the exposure, the pathophysiological importance of  
5 this observation remains uncertain. However, acute pulmonary hypertension due to O<sub>3</sub>-induced  
6 vasoconstriction could pose a risk to individuals with cardiovascular disease (*see Section 6-10*).

7 All in all, the contention that air pollutant mixtures elicit stronger pathophysiologic effects  
8 than individual pollutants of the mix is only weakly supported by human studies of either healthy  
9 or at-risk population.

## 12 **6.12 CONTROLLED STUDIES OF AMBIENT AIR EXPOSURES**

13 A large amount of informative O<sub>3</sub> exposure-effects data has been obtained in controlled  
14 laboratory exposure studies under a variety of different experimental conditions. However,  
15 laboratory simulation of the variable pollutant mixtures present in ambient air is not practical.  
16 Thus, the exposure effects of one or several artificially generated pollutants (i.e., a simple  
17 mixture) on pulmonary function and symptoms may not explain responses to ambient air where  
18 complex pollutant mixtures exist.

### 20 **6.12.1 Mobile Laboratory Studies**

21 Quantitatively useful information on the effects of acute exposure to photochemical  
22 oxidants on pulmonary function and symptoms responses from field studies using a mobile  
23 laboratory were presented in prior criteria documents (U.S. Environmental Protection Agency,  
24 1986, 1996). Relative to controlled exposure studies, mobile laboratory ambient air studies  
25 suffer the additional limitation of a dependence on ambient outdoor conditions. Consistent with  
26 controlled exposure studies, mobile studies in California demonstrated that pulmonary effects  
27 from exposure to ambient air in Los Angeles are related to O<sub>3</sub> concentration and level of  
28 exercise. Healthy subjects with a history of allergy also appeared to be more responsive to O<sub>3</sub>  
29 than “nonallergic” subjects (Linn et al., 1980, 1983b), although a standardized evaluation of  
30 atopic status was not performed.

## 6.12.2 Aircraft Cabin Studies

Respiratory symptoms and pulmonary function effects resulting from exposure to O<sub>3</sub> in commercial aircraft flying at high altitudes, and in altitude-simulation studies, have been assessed previously (U.S. Environmental Protection Agency, 1986, 1996). Commercial aircraft cabin O<sub>3</sub> levels were reported to be very low (average concentration 0.01 to 0.02 ppm) during 92 randomly selected smoking and nonsmoking flights in 1989 (Nagda et al., 1989). None of these flights recorded O<sub>3</sub> concentrations exceeding the 3-h time-weighted average (TWA) standard of 0.10 ppm promulgated by the U.S. Federal Aviation Administration (FAA, 1980), probably due to the use of O<sub>3</sub>-scrubbing catalytic filters (Melton, 1990).

Ozone contamination aboard high-altitude aircraft also has been an interest to the U.S. Air Force because of complaints by crew members of frequent symptoms of dryness and irritation of the eyes, nose, and throat and an occasional cough (Hetrick et al., 2000). Despite the lack of ventilation system modifications as used in commercial aircraft, the O<sub>3</sub> concentrations never exceeded the FAA ceiling limit of 0.25 ppm and exceeded the 3-h TWA of 0.10 ppm only 7% of the total monitored flight time (43 h). The authors concluded that extremely low average relative humidity (12%) during flight operations was most likely responsible for the reported symptoms.

## 6.13 SUMMARY

Responses in humans exposed to ambient O<sub>3</sub> concentrations include decreased inspiratory capacity; mild bronchoconstriction; rapid, shallow breathing pattern during exercise; and symptoms of cough and pain on deep inspiration. Reflex inhibition of inspiration results in a decrease in forced vital capacity (FVC) and, in combination with mild bronchoconstriction, contributes to a decrease in the forced expiratory volume in 1 s (FEV<sub>1</sub>). In addition to physiological pulmonary responses and symptoms of breathing discomfort, O<sub>3</sub> exposure also results in airway hyperresponsiveness, inflammation, immune system activation, and epithelial injury. With repeated O<sub>3</sub> exposures over several days, spirometric and symptom responses become attenuated, but this tolerance is lost after about a week without exposure. Airway responsiveness also appears to be attenuated with repeated O<sub>3</sub> exposures, but less than FEV<sub>1</sub>.

1 Unlike spirometric and symptom responses, airway inflammation and small airways dysfunction  
2 may not become attenuated by repeated O<sub>3</sub> exposures.

3 Young healthy adults exposed to O<sub>3</sub> concentrations  $\geq 0.08$  ppm develop significant  
4 reversible, transient decrements in pulmonary function if minute ventilation ( $\dot{V}_E$ ) or duration of  
5 exposure are increased sufficiently. The pattern of FEV<sub>1</sub> response appears to depend on the O<sub>3</sub>  
6 exposure profile. Triangular exposure profiles can potentially lead to greater FEV<sub>1</sub> responses  
7 than square wave exposures at equivalent average O<sub>3</sub> doses. O<sub>3</sub>-induced decrements in FEV<sub>1</sub> do  
8 not appear to depend on gender, race, body surface area, height, lung size, or baseline FVC in  
9 young healthy adults. Healthy children experience similar spirometric responses but lesser  
10 symptoms from O<sub>3</sub> exposure relative to young adults. On average, spirometric and symptom  
11 responses to O<sub>3</sub> exposure appear to decline with increasing age beyond approximately 18 years  
12 of age. There is a large degree of intersubject variability in physiologic and symptomatic  
13 responses of healthy adults exposed to O<sub>3</sub>. However, responses tend to be reproducible within a  
14 given individual over a period of several months. With increasing O<sub>3</sub> concentration, the  
15 distribution of FEV<sub>1</sub> decrements becomes asymmetrical with a few individuals experiencing  
16 large decrements.

17 There is a tendency for slightly increased spirometric responses in mild asthmatics and  
18 allergic rhinitics relative to healthy young adults. Spirometric responses in asthmatics appear to  
19 be affected by baseline lung function, i.e., responses increase with disease severity. With  
20 repeated daily O<sub>3</sub> exposures, spirometric responses of asthmatics become attenuated; however,  
21 airway responsiveness becomes increased in subjects with preexisting allergic airway disease  
22 (with or without asthma). Possibly due to patient age, O<sub>3</sub> exposure does not appear to cause  
23 significant pulmonary function impairment or evidence of cardiovascular strain in patients with  
24 cardiovascular disease or chronic obstructive pulmonary disease relative to healthy subjects.

25 Available information on recovery from O<sub>3</sub> exposure indicates that an initial phase of  
26 recovery in healthy individuals proceeds relatively rapidly, with acute spirometric and symptom  
27 responses resolving within about 2 to 4 h. Small residual lung function effects are almost  
28 completely resolved within 24 hours. Effects of O<sub>3</sub> on the small airways, assessed by persistent  
29 decrement in FEF<sub>25-75</sub> and altered ventilation distribution, may be due in part to inflammation.  
30 Indeed, a prolonged recovery of residual spirometric decrements following the initial rapid  
31 recovery could be due to slowly resolving airway inflammation. In hyperresponsive individuals,

1 this recovery takes longer, as much as 48 hours, to return to baseline values. Persistent  
2 spirometry changes observed for up to 48 h postexposure could plausibly be sustained by the  
3 inflammatory mediators. Cellular responses (e.g., release of immunomodulatory cytokines)  
4 appear to still be active as late as 20 h postexposure. More slowly developing inflammatory and  
5 cellular changes may persist for up to 48 h, but the time course for these parameters in humans  
6 has not been explored fully.

7 Soluble mediators of inflammation such as the cytokines (IL-6, IL-8) and arachidonic acid  
8 metabolites (e.g., PGE<sub>2</sub>, PGF<sub>2α</sub>, thromboxane, and leukotrienes [LTs] such as LTB<sub>4</sub>) have been  
9 measured in the BAL fluid of humans exposed to O<sub>3</sub>. Many of these compounds have  
10 bronchoconstrictive properties and may be involved in increased airway responsiveness  
11 following O<sub>3</sub> exposure. Some indicators of inflammation (e.g., PMN influx, IL-6, PGE<sub>2</sub>,  
12 fibronectin) are attenuated with repeated O<sub>3</sub> exposures. However, indicating that tissue damage  
13 probably continues to occur during repeated O<sub>3</sub> exposure, other markers (LDH, IL-8, total  
14 protein, epithelial cells) do not show attenuation. There appears to be no strong correlation  
15 between any of the measured cellular and biochemical changes and changes in lung function  
16 measurements. A limited number of studies suggest that inflammatory responses may be  
17 detected following O<sub>3</sub> exposures that are insufficient to cause decrements in pulmonary function.  
18 Whether airway reactivity or inflammatory responses to O<sub>3</sub> are dependent on the age of the  
19 exposed individual, such as spirometric responses, has not been determined.

20 Dietary antioxidant supplementation attenuates O<sub>3</sub>-induced spirometric responses but not  
21 the intensity of subjective symptoms nor inflammatory responses. Dietary antioxidants also  
22 afforded partial protection to asthmatics by attenuating postexposure bronchial  
23 hyperresponsiveness.

24

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# 7. EPIDEMIOLOGIC STUDIES OF HUMAN HEALTH EFFECTS ASSOCIATED WITH AMBIENT OZONE EXPOSURE

## 7.1 INTRODUCTION

This chapter evaluates current epidemiologic literature on health and physiological effects of ambient O<sub>3</sub> exposure. Epidemiologic studies linking community ambient O<sub>3</sub> concentrations to health effects were reported in the 1996 Ozone Air Quality Criteria Document (O<sub>3</sub> AQCD; U.S. Environmental Protection Agency, 1996a). Many of those studies reported that pulmonary function decrements, respiratory symptoms, and hospital and emergency department admissions in human populations were associated with ambient levels of O<sub>3</sub>. Numerous more recent epidemiologic studies discussed in this chapter evaluate the relationship of ambient O<sub>3</sub> to morbidity and mortality, and thereby provide an expanded basis for assessment of health effects associated with exposures to O<sub>3</sub> at concentrations currently encountered in the United States.

As discussed elsewhere in this document (Chapters 5 and 6), a substantial amount of experimental evidence links O<sub>3</sub> exposure unequivocally with respiratory effects in laboratory animals and humans. These include structural changes in the bronchiolar-alveolar transition (centriacinar) region of the lung, biochemical evidence of acute cellular/tissue injury, inflammation, increased frequency and severity of experimental bacterial infection, and temporary reductions in mechanical lung function. These effects have been observed with exposure to O<sub>3</sub> at ambient or near-ambient concentrations. Thus, many of the reported epidemiologic associations of ambient O<sub>3</sub> with respiratory health effects have considerable biological credibility. Accordingly, the new epidemiologic studies of ambient O<sub>3</sub> assessed here are best considered in combination with information from the other chapters on ambient O<sub>3</sub> concentration and exposure (Chapter 3), and toxicological effects of O<sub>3</sub> in animals and humans (Chapters 5 and 6, respectively). The epidemiologic studies constitute important information on associations between health effects and exposures of human populations to “real-world” O<sub>3</sub> and also help to identify susceptible subgroups and associated risk factors. A wide variety of oxidants in both the gaseous and particulate phases have not been examined in relation to health

1 outcomes in the literature. Therefore, discussion in this chapter is limited to studies of human  
2 health effects associated with ambient O<sub>3</sub> exposure.

### 3 4 **7.1.1 Approach to Identifying Ozone Epidemiologic Studies**

5 Numerous O<sub>3</sub> epidemiologic papers have been published since completion of the 1996 O<sub>3</sub>  
6 AQCD. The U.S. Environmental Protection Agency (NCEA-RTP) has implemented a  
7 systematic approach to identify relevant epidemiologic studies for consideration in this chapter.  
8 In general, an ongoing search has been employed in conjunction with other strategies to identify  
9 O<sub>3</sub> epidemiologic literature pertinent to developing criteria for the O<sub>3</sub> National Ambient Air  
10 Quality Standards (NAAQS). A publication base was established using Medline, Pascal,  
11 BIOSIS, NTIS, and Embase, and a set of search terms proven by prior use to identify pertinent  
12 literature. The search strategy was reexamined and modified to enhance identification of  
13 published papers. PubMed was added to the search regime.

14 While the above search regime provided good coverage of the relevant literature,  
15 additional approaches augmented the traditional search methods. First, a Federal Register  
16 Notice was issued requesting information and published papers from the public at large. Next,  
17 non-EPA chapter authors, expert in this field, identified literature on their own. NCEA-RTP  
18 staff also identified publications as an element of their assessment and interpretation of the  
19 literature. Finally, additional potentially relevant publications were included following external  
20 review as a result of comments from both the public and CASAC. The combination of these  
21 approaches is believed to produce a comprehensive collection of studies appropriate for review  
22 and assessment here. The principal objective criteria used for selecting literature for the present  
23 assessment is to include all identified studies that evaluated the relationship between measured  
24 ambient O<sub>3</sub> levels and a human health outcome. New studies accepted for publication through  
25 December 2004, as identified using the approaches above, have been included in this AQCD and  
26 additional efforts have been made to assess more recent studies.

### 27 28 **7.1.2 Approach to Assessing Epidemiologic Evidence**

29 Definitions of the various types of epidemiologic studies assessed have been provided in an  
30 earlier PM AQCD (U.S. Environmental Protection Agency, 1996b). Briefly, epidemiologic  
31 studies are generally divided into two groups, *morbidity* studies and *mortality* studies. *Morbidity*

1 studies evaluate O<sub>3</sub> effects on a wide range of health endpoints, including the following:  
2 changes in pulmonary function, respiratory symptoms, self-medication in asthmatics, and airway  
3 inflammation; changes in cardiovascular physiology/functions; and cardiopulmonary emergency  
4 department visits and hospital admissions. *Mortality* studies investigate O<sub>3</sub> effects on total  
5 (nonaccidental) mortality and cause-specific mortality, providing evidence related to a clearly  
6 adverse endpoint. The epidemiologic strategies most commonly used in O<sub>3</sub> health studies are  
7 prospective cohort studies, ecologic studies, time-series semi-ecologic studies, and case-  
8 crossover studies. All of these are observational studies rather than experimental studies.

9 The approach to assessing epidemiologic evidence has been stated most recently in the  
10 2004 PM AQCD (U.S. Environmental Protection Agency, 2004) and is summarized here. The  
11 critical assessment of epidemiologic evidence presented in this chapter is conceptually based  
12 upon consideration of salient aspects of the evidence of associations so as to reach fundamental  
13 judgments as to the likely causal significance of the observed associations (see Hill, 1965). The  
14 general evaluation of the strength of the epidemiologic evidence reflects consideration not only  
15 of the magnitude and precision of reported O<sub>3</sub> effect estimates and their statistical significance,  
16 but also of the robustness of the effects associations. Statistical significance corresponds to the  
17 allowable rate of error (Type I error) in the decision problem constructed from assuming that a  
18 simple null hypothesis of no association is true. It is a conditional probability; for statistical  
19 significance, typically there is a less than 0.05 chance of rejecting the null hypothesis given that  
20 it is true. Robustness of the associations is defined as stability in the effect estimates after  
21 considering a number of factors, including alternative models and model specifications, potential  
22 confounding by copollutants, as well as issues related to the consequences of measurement error.

23 Consideration of the consistency of the effects associations, as discussed in the following  
24 sections, involves looking across the results of multiple- and single-city studies conducted by  
25 different investigators in different places and times. Relevant factors are known to exhibit much  
26 variation across studies, including, for example, the presence and levels of copollutants, the  
27 relationships between central measures of O<sub>3</sub> and exposure-related factors, relevant demographic  
28 factors related to sensitive subpopulations, and climatic and meteorological conditions. Thus, in  
29 this case, consideration of consistency and the related heterogeneity of effects are appropriately  
30 understood as an evaluation of the similarity or general concordance of results, rather than an  
31 expectation of finding quantitative results within a very narrow range.

1 Looking beyond the epidemiologic evidence, evaluation of the biological plausibility of the  
2 O<sub>3</sub>-health effects associations observed in epidemiologic studies reflects consideration of both  
3 exposure-related factors and dosimetric/toxicologic evidence relevant to identification of  
4 potential biological mechanisms. Similarly, coherence of health effects associations reported in  
5 the epidemiologic literature reflects consideration of information pertaining to the nature of the  
6 various respiratory- and cardiac-related mortality and morbidity effects and biological markers  
7 evaluated in toxicologic and human clinical studies. These broader aspects of the assessment are  
8 only touched upon in this chapter but are more fully integrated in the discussion presented in  
9 Chapter 8.

10 In assessing the relative scientific quality of epidemiologic studies reviewed here and to  
11 assist in interpreting their findings, the following considerations were taken into account:

- 12 (1) To what extent are the aerometric data/exposure metrics used of adequate quality and  
sufficiently representative to serve as credible exposure indicators, well-reflecting  
geographic or temporal differences in study population pollutant exposures in the  
range(s) of pollutant concentrations evaluated?
- 13 (2) Were the study populations well defined and adequately selected so as to allow for  
meaningful comparisons between study groups or meaningful temporal analyses of  
health effects results?
- 14 (3) Were the health endpoint measurements meaningful and reliable, including clear  
definition of diagnostic criteria utilized and consistency in obtaining dependent  
variable measurements?
- 15 (4) Were the statistical analyses used appropriate, and properly performed and  
interpreted?
- 16 (5) Were likely important covariates (e.g., potential confounders or effect modifiers)  
adequately controlled for or taken into account in the study design and statistical  
analyses?
- 17 (6) Were the reported findings internally consistent, biologically plausible, and coherent  
in terms of consistency with other known facts?

18 These guidelines provide benchmarks for judging the relative quality of various studies and  
19 in assessing the body of epidemiologic evidence. Detailed critical analysis of all epidemiologic  
20 studies on O<sub>3</sub> health effects, especially in relation to all of the above questions, is beyond the  
21 scope of this document. Of most importance for present purposes are those studies which



1 provide useful qualitative or quantitative information on concentration-response relationships for  
2 health effects associated with ambient air levels of O<sub>3</sub> likely to be encountered in the U.S. among  
3 healthy and susceptible populations.  
4

### 5 **7.1.3 Considerations in the Interpretation of Epidemiologic Studies of** 6 **Ozone Health Effects**

7 Prior to discussing results from recent O<sub>3</sub> epidemiologic studies, issues and questions  
8 arising from the study designs and analysis methods used in the assessment of O<sub>3</sub> effect  
9 estimates will be briefly presented. Study design can restrict the health effect parameters that  
10 can be estimated. Separate considerations need to be made for acute versus chronic effect  
11 studies as well as individual versus aggregate-level analyses. Time-series studies and panel  
12 studies are most frequently conducted in air pollution epidemiologic research. Aggregate-level  
13 exposure and/or outcome data are often used in these types of studies. Analyses using  
14 administrative health outcome data (e.g., numbers of deaths and emergency hospital admissions)  
15 have inherent limitations as well as strengths (Virnig and McBean, 2001). The impact of study  
16 design or the loss of information due to aggregation largely depends upon exposure variation  
17 (Sheppard et al., 2005).

18 This section mainly focuses on the topics of exposure assessment and model specification  
19 in air pollution epidemiologic studies. Potential biases that may result from O<sub>3</sub> exposure  
20 measurement error, and choice of exposure index and lag period are first presented.  
21 A discussion of model specification issues and potential confounding by temporal factors,  
22 meteorological effects, seasonal trends, and copollutants follow.  
23

#### 24 **7.1.3.1 Exposure Assessment and Measurement Error in Epidemiologic Studies**

25 In many air pollution epidemiologic studies, especially time-series studies with  
26 administrative data on mortality and hospitalization outcomes, data from central ambient  
27 monitoring sites generally are used as the estimate of exposure. Personal exposures of individual  
28 study participants generally are not directly observed in epidemiologic studies. The use of O<sub>3</sub>  
29 concentrations from ambient monitors as surrogate measures for personal O<sub>3</sub> exposures was  
30 discussed previously in Section 3.9. Routinely collected ambient monitor data, though readily

1 available and convenient, may not represent true personal exposure, which includes both  
2 ambient and non-ambient source exposures.

3 In several studies focused on evaluating exposure to O<sub>3</sub>, measurements were made in a  
4 variety of indoor environments, including homes (Lee et al., 2004), schools (Linn et al., 1996),  
5 and the workplace (Liu et al., 1995). Indoor O<sub>3</sub> concentrations were, in general, approximately  
6 one-tenth of the outdoor concentrations in these studies. Few indoor sources of O<sub>3</sub> exist,  
7 possible sources being office equipment (e.g., photocopiers, laser printers) and air cleaners.  
8 As described in Section 3.8 of this document, O<sub>3</sub> in the indoor environment is largely dependent  
9 on the outdoor ambient O<sub>3</sub> concentration. Other factors that influence the O<sub>3</sub> concentration  
10 indoors include the air exchange rate, outdoor infiltration, indoor circulation rate, and O<sub>3</sub>  
11 removal process.

12 Sheppard (2005) states that non-ambient exposures typically vary across individuals but  
13 are not likely to have strong temporal correlations. In contrast, ambient concentrations for  
14 individuals should be highly correlated as they vary over time similarly for everyone because of  
15 changes in source generation, weather, and season. The independence of ambient and  
16 non-ambient exposure sources has important implications for selection of study designs that are  
17 most effective for estimating health effects (Sheppard, 2005). In an ideal situation, studies of air  
18 pollution health effects would be conducted at the individual level, with information on personal  
19 exposure to the various pollutants. However, determining accurate personal exposure  
20 information is difficult and generally impractical. A simulation study by Sheppard et al. (2005)  
21 examining non-reactive pollutants observed that there was no noticeable difference between  
22 effect estimates using either total personal exposure or ambient concentration data when  
23 non-ambient source exposures were independent of ambient source exposures in time-series  
24 studies. Sheppard (2005) concludes that for estimating acute effects, ambient concentration  
25 measurements are adequate in time-series studies. In the case of O<sub>3</sub>, there are limited  
26 non-ambient sources; thus, ambient concentrations of O<sub>3</sub> are also likely to be adequate in the  
27 analysis of O<sub>3</sub> health effects in time-series studies. Even with the lower exposure variation when  
28 using only ambient concentration data, the large sample sizes and longer study duration make  
29 time-series studies quite powerful.

30 As discussed thoroughly in the 2004 PM AQCD (Section 8.4.5), the resulting exposure  
31 measurement error and its effect on the estimates of relative risk must be considered. In theory,

1 there are three components to exposure measurement error in time-series studies as described by  
2 Zeger et al. (2000): (1) the use of average population rather than individual exposure data;  
3 (2) the difference between average personal ambient exposure and ambient concentrations at  
4 central monitoring sites; and (3) the difference between true and measured ambient  
5 concentrations. Zeger et al. indicated that the first and third error components were largely  
6 Berksonian errors; although they would increase the standard errors, they would not bias the  
7 risk estimate. However, the second error component resulting from the difference between  
8 average personal ambient exposure and outdoor ambient concentration levels might attenuate  
9 the risk estimate.

10 The impact of exposure measurement error on O<sub>3</sub> effect estimates was demonstrated in a  
11 study by Navidi et al. (1999). In this study, a simulation was conducted using data from the  
12 University of Southern California Children's Health Study of the long-term effects of air  
13 pollutants on children. The effect estimate from computed "true" O<sub>3</sub> exposure was compared to  
14 effect estimates from exposure determined using several methods: (1) ambient stationary  
15 monitors; (2) the microenvironmental approach (multiply concentrations in various  
16 microenvironments by time present in each microenvironment); and (3) personal sampling.  
17 Effect estimates based on all three exposure measures were biased towards the null. The bias  
18 that results when using the microenvironmental and personal sampling approach is due to  
19 nondifferential measurement error. Use of ambient monitors to determine exposure will  
20 generally overestimate true personal O<sub>3</sub> exposure (assumes that subjects are outdoors 100% of  
21 their time and not in close proximity to sources that reduce O<sub>3</sub> levels such as NO emissions from  
22 mobile sources), thus generally their use will result in effect estimates that are biased towards the  
23 null.

24 Zidek (1997) notes that a statistical analysis must balance bias and imprecision (error  
25 variance). Ignoring measurement error in air pollution epidemiologic studies often will result in  
26 underestimated risk estimates. In a reanalysis of the study by Burnett et al. (1994) on the acute  
27 respiratory effects of ambient air pollution, Zidek et al. (1998) observed that accounting for  
28 measurement error, as well as a few additional changes to the analysis, resulted in qualitatively  
29 similar conclusions. However, while the original analysis by Burnett et al. found that 5% of  
30 daily respiratory admissions in the summer months was attributable to O<sub>3</sub>, Zidek et al. calculated  
31 that O<sub>3</sub> was associated with a 14% increase in respiratory admissions. Available data and

1 analysis limit our ability to weigh the importance of uncertainty due to measurement error  
2 relative to other sources in the studies reviewed.

3 As discussed in Section 3.9, there is suggestive evidence that ambient O<sub>3</sub> concentrations  
4 from central monitors may serve as valid surrogate measures for aggregate personal O<sub>3</sub>  
5 exposures in time-series studies. However, using ambient concentrations to determine exposure  
6 generally overestimates true personal O<sub>3</sub> exposures, resulting in biased descriptions of  
7 underlying concentration-response relationships. These effect estimates, though conservative  
8 from a testing perspective, must be evaluated and used with caution as they may lead to an  
9 underestimation of the overall impact of air pollution on health effects. A better understanding  
10 of the relationship between ambient concentrations and personal exposures, and the factors that  
11 affect the relationship will improve the interpretation of ambient concentration-population health  
12 response associations observed in epidemiologic studies.

### 13 14 **7.1.3.2 Ozone Exposure Indices Used**

15 The O<sub>3</sub>-related effect estimates for mortality and morbidity health outcomes are usually  
16 presented in this document as a relative risk, or risk rate relative to a baseline mortality or  
17 morbidity rate. These relative risks are based on an incremental change in exposure.  
18 To enhance comparability between studies, presenting these relative risks by a uniform exposure  
19 increment is needed. However, determining a standard increment is complicated by the use of  
20 different O<sub>3</sub> exposure indices in the existing health studies. The three daily O<sub>3</sub> exposure indices  
21 that most often appear in the literature are 1-h average maximum (1-h max), 8-h average  
22 maximum (8-h max), and 24-h average (24-h avg) concentrations. As levels are lower and less  
23 variable for the longer averaging times, relative risks of adverse health outcomes for a specific  
24 numeric concentration range are not directly comparable across metrics. Using the nationwide  
25 distributional data for O<sub>3</sub> monitors in U.S. Metropolitan Statistical Areas, increments  
26 representative of a low-to-high change in O<sub>3</sub> concentrations were approximated based on annual  
27 mean to 95th percentile differences (Langstaff, 2003), as follows:

28  
29

Daily Exposure Index	Exposure Increment (ppb)
1-h max O <sub>3</sub>	40
8-h max O <sub>3</sub>	30
24-h avg O <sub>3</sub>	20

30  
31  
32

1 In the following discussion sections, efforts were made to standardize the O<sub>3</sub> excess risks using  
2 these increments, except as noted, so that risk estimates could be compared across studies. Note  
3 that in the Annex Tables, effect estimates are not standardized; results are presented in the tables  
4 as they are reported in the papers.

### 6 **7.1.3.3 Lag Time: Period between Ozone Exposure and Observed Health Effect**

7 Lags of exposure may reflect the distribution of effects across time in a population and the  
8 potential mechanisms of effects. The choice of lag days for the relationship between exposure  
9 and health effects depends on the hypothesis being tested and the mechanism involved in the  
10 expression of the outcome. Effects can occur acutely with exposure on the same or previous  
11 day, cumulatively over several days, or after a delayed period of a few days. With knowledge  
12 of the mechanism of effect, the choice of lag days can be determined prior to analysis.  
13 For example, one can expect cough to occur acutely after exposure with a lag of 0 or 1 day, as  
14 O<sub>3</sub> can act as a short-term irritant. However, an O<sub>3</sub>-related inflammatory response may not lead  
15 to asthma exacerbation until several days later. An asthmatic may be impacted by O<sub>3</sub> on the first  
16 day of exposure, have effects triggered further on the second day, then report to the emergency  
17 room for an asthmatic attack three days after exposure. Further, within a population of  
18 asthmatics, exacerbation of asthma symptoms may be observed over a period of several days,  
19 since each asthmatic has individual aspects of the disease and may be affected by the exposure  
20 differently depending on his/her sensitivity and disease severity. The results from controlled  
21 human studies may be useful in assessing the adequacy of lags for some respiratory health  
22 outcomes.

23 Some studies attempted to examine the overall impact of O<sub>3</sub> through distributed lag  
24 models. Schildcrout and Heagerty (2005) compared regression analyses using single-day versus  
25 distributed lag models. The single-day lag model calculates a risk estimate that assumes  
26 dependence only on exposure from the specified day. In contrast, the distributed lag model  
27 provides an estimate that is a summary measure of the cumulative distributed lag effect from all  
28 included lag days. The standard error of the cumulative sum of the individual distributed lag  
29 coefficients takes into consideration the variance-covariance of the multiple lags, and is therefore  
30 larger than the standard error of the single-day lag coefficient. Thus, if the underlying O<sub>3</sub>-health  
31 outcome relationship was a single-day effect, then modeling the relationship with a distributed

1 lag model would make the estimate less significant. On the other hand, if the effect of O<sub>3</sub> on  
2 health outcomes persisted over several days, then applying a single-day lag model would result  
3 in an underestimation of the multiday effects. The correct choice requires balancing variance  
4 and bias.

5 As the parameters estimated from single-day lag versus multiday lag models are not the  
6 same, interpretation and comparison of these results may be difficult. When comparing the  
7 impacts of these different models, the nuance of increments used in calculating the estimates is  
8 different depending on the model. For example, an excess percent mortality risk “per 20 ppb  
9 increase in 24-h avg O<sub>3</sub>” in a distributed lag model including lag 0- through 6-days tacitly means  
10 a 20 ppb increase in each of the seven days. The difference in the exposure scenarios in the  
11 single-day versus multiday lag model (i.e., 20 ppb increase in one day versus several consecutive  
12 days) complicates a simple comparison of risk estimates from two different models using “the  
13 same increment.”

14 Only a limited number of studies have hypothesized a priori the lag structure to be  
15 examined. Most of the O<sub>3</sub> time-series studies examined relatively small numbers of single-day  
16 lag models, typically lags of 0 through 3 days. Sheppard et al. (1999) notes that when  
17 considering single-day lag estimates it is important to consider the effect estimate in the context  
18 of the pattern of adjacent lags as these estimates contain information from the adjacent days  
19 owing to serial correlation of the pollutant series. In many cases, a pattern of positive  
20 associations across several lag days were reported. For the respiratory and cardiovascular  
21 outcomes investigated, the “most significant” lags were generally 0- or 1-day lags, suggesting  
22 that the majority of the single-day associations are immediate, not a random pattern in which  
23 associations can be observed on any of the lags examined with equal probabilities. For example,  
24 two recent meta-analyses of O<sub>3</sub>-mortality effects observed that the combined estimate from  
25 0-day lag models was larger than the estimate from longer lag days (Bell et al., 2005; Levy  
26 et al., 2005).

27 Bias resulting from the selection of lags has not been examined specifically for O<sub>3</sub> effects.  
28 However, the issue of lags has been investigated for PM and the results of this analysis are most  
29 likely of relevance for O<sub>3</sub>. Lumley and Sheppard (2000) performed a simulation study to  
30 examine model selection bias in air pollution epidemiology using PM<sub>2.5</sub> as an example.  
31 Sheppard et al. (1999; reanalysis Sheppard, 2003) had investigated the association between

1 asthma hospital admissions and ambient PM<sub>2.5</sub> concentrations over an eight-year period in  
2 Seattle, WA. Note that the results from Lumley and Sheppard (2000) and Sheppard et al. (1999)  
3 were based on GAM using default convergence criteria (see Section 7.1.3.7). A negative control  
4 analysis, using simulated data with no association between PM exposure and the health outcome,  
5 and a positive control analysis, in which a specified non-zero excess risk is added to the  
6 simulation, were performed for comparison. The bias from selection of best of seven lags  
7 (0 to 6 days) and residual seasonal confounding in the negative control analysis (median log  
8 relative risk of 0.0013) was approximately half the log relative risk estimated from the observed  
9 data (0.0027), after adjusting for season and temperature. In the positive control model (true log  
10 relative risk of 0.0083), the bias was small (median log relative risk of 0.0080). Results from  
11 these simulations indicate that bias from selection of lags may be small, but of the same  
12 magnitude as the estimated health impacts.

13 Selection of lag periods should depend on the hypothesis of the study and the potential  
14 mechanism of the effect. When the mechanism of the health effect is unknown, investigating the  
15 association between outcome and exposure using cumulative distributed lag models may be  
16 informative. Analyzing a large number of lags and simply choosing the largest and most  
17 significant results may bias the air pollution risk estimates away from the null. Most studies  
18 have shown that O<sub>3</sub> has a fairly consistent, immediate effect on health outcomes, including  
19 respiratory hospitalizations and mortality. Several studies also observed significant O<sub>3</sub> effects  
20 over longer cumulative lag periods, suggesting that in addition to single-day lags, multiday lags  
21 should be investigated to fully capture a delayed O<sub>3</sub> effect on health outcomes. In this document,  
22 discussion largely focuses on effect estimates from 0- and 1-day lags, with some consideration of  
23 cumulative, multiday lag effects. It is not straightforward to compare and contrast results from  
24 single-day versus multiday lag models because the parameters estimated from these models are  
25 not the same. These complications need to be taken into consideration when interpreting results  
26 from various lag models.

#### 27 28 **7.1.3.4 Model Specification to Adjust for Temporal Trends and Meteorologic Effects**

29 Several challenges present themselves with respect to designing and interpreting  
30 time-series studies. The principal challenge facing the analyst in the daily time-series context is  
31 avoiding bias due to confounding by short-term temporal factors operating over time scales from

1 days to seasons. In the current regression models used to estimate short-term effects of air  
2 pollution, two major potential confounders need to be considered: (1) seasonal trend and other  
3 “long-wave” temporal trends; and (2) weather effects. Both of these variables tend to predict a  
4 significant fraction of fluctuations in time-series. Unfortunately, as O<sub>3</sub> has strong seasonal  
5 cycles and is formed more at higher temperatures, both terms are also highly correlated with O<sub>3</sub>.  
6 The correlation of O<sub>3</sub> with these confounding terms tends to be higher than that for PM or other  
7 gaseous pollutants. In the U.S., the mass concentration of PM<sub>2.5</sub> generally does not have strong  
8 seasonal cycles like O<sub>3</sub> because PM<sub>2.5</sub> tends to reflect both primary emissions (throughout the  
9 year, but often higher in winter in most U.S. cities) and secondary aerosols (higher in summer).  
10 Therefore, PM<sub>2.5</sub> and O<sub>3</sub> effect estimates from studies primarily designed to examine PM<sub>2.5</sub> health  
11 effects may not be comparable as model specifications that may be appropriate for PM<sub>2.5</sub> may  
12 not necessarily be adequate for O<sub>3</sub>.

13 An examination of recent time-series studies indicates that several types of fitting  
14 approaches have been used to adjust for temporal trends and weather effects. The use of  
15 parametric and nonparametric smoothers with varying degrees of freedom per year has emerged  
16 as the prevailing approach. The use of larger degrees of freedom to adjust for potential  
17 confounding by time-varying factors may inadvertently result in ascribing more effects to these  
18 unmeasured potential confounders and mask the air pollution effect. Often smaller pollution  
19 effect estimates are observed when more degrees of freedom are used. Currently, the degrees of  
20 freedom used to adjust for temporal trends in time-series studies generally range from 4 to  
21 12 degrees of freedom per year using either nonparametric or parametric smoothers. Statistical  
22 diagnostics such as Akaike’s Information Criteria, residual autocorrelation, or dispersion of the  
23 regression model often are used to choose or evaluate the adequacy of the degrees of freedom for  
24 temporal trend. However, these diagnostics do not guarantee “adequate” control for temporal  
25 confounding, as choosing the appropriate extent of smoothing requires prior knowledge of the  
26 nature of the confounding (e.g., shape and duration of influenza epidemics).

27 The issue of model specifications to adjust for temporal trends and weather variables in  
28 time-series studies was a consideration of several researchers that conducted sensitivity analyses  
29 of PM data (Health Effects Institute, 2003). The sensitivity of O<sub>3</sub> coefficients to model  
30 specifications for temporal trend adjustment has not been as well-studied. Recent multicity  
31 studies examined the sensitivity of O<sub>3</sub> coefficients to the extent of smoothing for adjustment of



1 temporal trends and meteorologic factors (Bell et al., 2004; Huang et al., 2005; Ito et al., 2005).  
2 Most, if not all, O<sub>3</sub> studies used the same model specifications to estimate the excess risks for  
3 PM and other gaseous pollutants. The model specification designed to control confounding by  
4 meteorological and temporal factors for PM may not be necessarily adequate for O<sub>3</sub>. As noted  
5 above, O<sub>3</sub> is expected to have the strongest correlation with both temporal (seasonal) trend and  
6 weather effects. The strong annual cycle in O<sub>3</sub> concentrations presents a unique problem in  
7 time-series analyses where time trends are fitted simultaneously with pollution and other model  
8 terms (i.e., co-adjustment). In this setting, the annual O<sub>3</sub> cycle itself may compete with the  
9 smooth function of time to explain some of the annual, cyclic behavior in the health outcome,  
10 which can result in biased effect estimates for O<sub>3</sub> when data for all seasons are analyzed  
11 together.

12 Current weather models used in time-series analyses can be classified into: (1) quantile  
13 (e.g., quartile, quintile) indicators; (2) parametric functional forms such as V- or U-shape  
14 functions; and (3) parametric (e.g., natural splines) or nonparametric (e.g., locally estimated  
15 smoothing splines [LOESS]) smoothing functions. More recent studies tend to use smoothing  
16 functions. While these methods provide flexible ways to fit health outcomes as a function of  
17 temperature and other weather variables, there are two major issues that need further  
18 examination to enable more meaningful interpretation of O<sub>3</sub> morbidity and mortality effects.

19 The first issue is the interpretation of weather or temperature effects. Most researchers  
20 agree about the morbidity and mortality effects of extreme temperatures (i.e., heat waves or cold  
21 spells). However, as extreme hot or cold temperatures, by definition, happen rarely, much of the  
22 health effects occur in the mild or moderate temperature range. Given the significant correlation  
23 between O<sub>3</sub> and temperature, ascribing the association between temperature and health outcomes  
24 solely to temperature effects may underestimate the effect of O<sub>3</sub>. The second issue is that  
25 weather model specifications are fitted for year-round data in most studies. Such models will  
26 ignore the correlation structure that can change across seasons, resulting in inefficiency  
27 and model mis-specification. This is particularly important for O<sub>3</sub>, which appears to change  
28 its relationship with temperature as well as with other pollutants across seasons.

29 This changing relationship between O<sub>3</sub> and temperature, as well as O<sub>3</sub> and other pollutants  
30 across seasons, and its potential implications to health effects modeling have not been examined  
31 thoroughly in the time-series literature. Even the flexible smoother-based adjustments for

1 seasonal and other time-varying variables cannot fully take into account these complex  
2 relationships. One obvious way to alleviate or avoid this complication is to analyze data by  
3 season. While this practice reduces sample size, its extent would not be as serious as PM (which  
4 is collected only every sixth day in most locations) because O<sub>3</sub> is collected daily, though only in  
5 warm seasons in some states. An alternative approach is to include separate O<sub>3</sub> concentration  
6 variables for each season (by multiplying O<sub>3</sub> concentrations by a season indicator variable).

7 In locations where seasonal variability may be a factor, O<sub>3</sub> effect estimates calculated using  
8 year-round data can be misleading, as the changing relationship between O<sub>3</sub>, temperature, and  
9 other pollutants across seasons may have a significant influence on the estimates. Analyses have  
10 indicated that confounding from seasonal variability may be controlled effectively by stratifying  
11 the data by season.

#### 13 **7.1.3.5 Confounding Effects of Copollutants**

14 Extensive discussions on the issues related to confounding effects among air pollutants in  
15 time-series studies are provided in Section 8.4.3 of the 2004 PM AQCD. Since the general  
16 issues discussed in that document are applicable to all pollutants, such discussions are not  
17 repeated here. What was not discussed in the 2004 PM AQCD was the issue of changing  
18 relationships among air pollutants across seasons. Compared to other pollutants, O<sub>3</sub> has strong  
19 seasonal cycles. Ambient O<sub>3</sub> levels are typically higher in the summer or warm season, often  
20 referred to as the O<sub>3</sub> season. In the winter or colder months, O<sub>3</sub> levels tend to be much lower  
21 compared to the summer months. During the winter in some urban locations, O<sub>3</sub> mainly comes  
22 from the free troposphere and can be considered a tracer for relatively clean air (i.e., cold, clear  
23 air coming down from the upper atmosphere), as discussed in Chapter 3 of this AQCD. The  
24 clean air is associated with the passage of cold fronts and the onset of high-pressure conditions,  
25 which occur with colder temperatures. Thus, sunny clear winter days following a high-pressure  
26 system are the days when air pollution levels from primary emissions (e.g., NO<sub>2</sub>, SO<sub>2</sub>, and PM  
27 from local sources) tend to be lower and O<sub>3</sub> is relatively higher. This can lead to negative  
28 correlations between O<sub>3</sub> and the primary pollutants in the winter. As shown in Figure 3-24 in the  
29 Chapter 3 Annex, the relationship between O<sub>3</sub> and PM<sub>2.5</sub> was U-shaped for the year-round data in  
30 Fort Meade, MD. The negative PM<sub>2.5</sub>/O<sub>3</sub> slope was in the range of O<sub>3</sub> concentrations less than  
31 30 ppb, providing supporting evidence of the aforementioned winter phenomenon. Thus, the

1 correlation between O<sub>3</sub> and PM for year-round data may be misleading. The high reactivity of  
2 O<sub>3</sub> with certain copollutants further complicates the analysis. For example, the reaction between  
3 NO (emitted from motor vehicles) and O<sub>3</sub> results in reduced O<sub>3</sub> levels but increased NO<sub>2</sub> levels  
4 during high traffic periods.

5 Multipollutant regression models often are used to assess potential confounding by  
6 copollutants; however, there are limitations to these models. Zidek et al. (1996) examined,  
7 through simulation, the joint effects of multicollinearity and measurement error in a Poisson  
8 regression model. The results illustrated the transfer of effects from the “causal” variable to the  
9 confounder. However, in order for the confounder to have a larger effect size than the true  
10 predictor, the correlation between the two covariates had to be very high ( $r \geq 0.9$ ), with moderate  
11 error ( $\alpha > 0.5$ ) for the true predictor and no error for the confounder in their scenarios. The  
12 transfer-of-causality effect was lessened when the confounder also became subject to error.  
13 Another interesting finding was the behavior of the standard errors of the coefficients. When the  
14 correlation between the covariates was high ( $r = 0.9$ ) and both covariates had no error, the  
15 standard errors for both coefficients were inflated by a factor of two; however, this phenomenon  
16 disappeared when the confounder had error. The effect of multicollinearity is generally even  
17 more complex when analyzing real data. For further discussion, see the 2004 PM AQCD  
18 (Sections 8.4.3 and 8.4.5).

19 Uncertainty remains as to the use of multipollutant regression models to assess the  
20 independent health effects of pollutants that are correlated. Particularly in the case of O<sub>3</sub>,  
21 concern remains as to whether multipollutant regression models for year-round data can adjust  
22 for potential confounding adequately due to the changing relationship between O<sub>3</sub> and other  
23 pollutants. Despite these limitations, multipollutant models are still the prevailing approach in  
24 most, if not all, studies of O<sub>3</sub> health effects and serve as an important tool in addressing the issue  
25 of confounding by copollutants, especially in season-stratified analyses.

### 27 **7.1.3.6 Model Uncertainty from Multiple Hypothesis Testing**

28 Epidemiologic studies that investigated the association between various measures of O<sub>3</sub>  
29 (multiple lags, different metrics, etc.) and various health outcomes often found significant  
30 effects. A major question is: Are these significant associations an artifact of model selection  
31 due to multiple testing and does this lead to overestimation of the effect estimates?

1 Multiple testing occurs when multiple health outcomes are examined, several lags are  
2 tested, different metrics of O<sub>3</sub> exposure are used, and many sub-populations are tested.  
3 Statistically testing a null hypothesis (i.e., there is no effect of O<sub>3</sub>) requires one to calculate the  
4 value of a test statistic (i.e., t-value). If the observed test statistic exceeds a critical value  
5 (oftentimes the 95th percentile) or is outside a range of values, the null hypothesis is rejected.  
6 However, when multiple testing is done using a critical value determined for a single test, the  
7 chance that at least one of the hypotheses is significant may be greater than the expected 5%  
8 error rate. This uncertainty clouds the interpretation and weakens the evidence against any of  
9 the null hypotheses. Still, multiple hypotheses testing may be of great value. For example,  
10 developing a few hypotheses a priori allows researchers to explore more thoroughly potential  
11 associations for an O<sub>3</sub>-related health effect. Sensitivity analyses, which are critical for model  
12 validation, also involve multiple testing. There are two types of sensitivity testing. One tests for  
13 the consistency of the effect when different adjustments are made for seasonal effects, or other  
14 covariates. Another tests for sensitive subpopulations and other specific conditions. In the  
15 former case, one should guard against a multiple testing error by restricting the inferences to  
16 consistency of the effect and not treat the hypotheses generated for sensitivity analyses as being  
17 confirmatory.

18 Recent attention has focused on Bayesian model averaging as a method to address model  
19 uncertainty from multiple hypothesis testing. In Bayesian model averaging, predictions and  
20 inferences are based on a set of models rather than a single model, and each model contributes  
21 proportionally to the support it receives from the observed data (Clyde, 1999). In addition to the  
22 uncertainty of effect estimation, Bayesian model averaging can incorporate uncertainty regarding  
23 the choice of confounding variables, pollutants, and lags. Koop and Tole (2004) used Bayesian  
24 model averaging to analyze the effect of various air pollutants, including O<sub>3</sub>, SO<sub>2</sub>, CO, NO, NO<sub>2</sub>,  
25 PM<sub>10-2.5</sub>, and PM<sub>2.5</sub>, on mortality in Toronto, Canada. The 50+ explanatory variables required the  
26 fitting of an enormous number of potential models. Clyde et al. (2000) and Clyde (2000) also  
27 used Bayesian model averaging to analyze the relationship between PM and mortality. Clyde  
28 (2000) noted that Bayesian model averaging did not take into consideration factors that might  
29 bias the estimated effect toward the null. For example, measurement error in the exposure  
30 variables was not considered. In addition, the Poisson model (similar to many other regression  
31 models) assumed that all individuals in a population had equal risks, including potentially

1 susceptible populations such as those with respiratory illnesses and outdoor workers. While  
2 Bayesian model averaging can theoretically be used to take into account uncertainty, claims of  
3 causality based on observational studies may be highly sensitive to the choice of prior  
4 distributions and class of models under consideration (Clyde et al., 2000). Another limitation of  
5 Bayesian model averaging is that the estimated posterior effects may be diluted (i.e., result in  
6 smaller coefficients) when variables are highly correlated, as may be the case for air pollution  
7 studies (George, 1999 in comments to Hoeting et al., 1999).

8 Additional methods to control for model uncertainty resulting from multiple hypothesis  
9 testing are by a priori deciding hypotheses that are confirmatory and exploratory, and limiting  
10 the number of confirmatory tests. For example, Dominici et al. (2003) used a minimum number  
11 of tests in the U.S. 90 cities study, which reduced the uncertainty associated with multiple  
12 testing. In addition, they performed sensitivity analyses to examine the consistency and  
13 robustness of the effects. Another approach is to partition the data into two sets, one for model  
14 identification and a second for model confirmation.

15 The summary of health effects in this chapter is vulnerable to the errors of publication bias  
16 and multiple testing. Recent studies (Bell et al., 2005; Ito et al., 2005; Lumley and Sheppard,  
17 2000) have found indications of what the magnitudes of these errors might be in some instances.  
18 Some researchers have used methods to protect their estimates against these errors. Efforts have  
19 been made to reduce the impact of multiple testing errors on the conclusions in this document.  
20 To address multiple hypothesis testing in this chapter, emphasis will be on a priori hypotheses.  
21 As identifying a priori hypotheses is difficult in the majority of the studies, the most common  
22 hypotheses will be considered. For example, although many studies examined multiple single-  
23 day lag models, priority would be given to the effects observed at 0- or 1-day lags rather than at  
24 longer lags. Both single- and multiple-pollutant models that include O<sub>3</sub> will be considered and  
25 examined for robustness of results. Analyses of multiple model specifications for adjustment of  
26 temporal or meteorological trends will be considered sensitivity analyses. Sensitivity analyses  
27 shall not be granted the same inferential weight as the original hypothesis-driven analysis;  
28 however, these analyses will be discussed in this chapter as appropriate given their valuable  
29 insights that may lead scientific knowledge in new directions.

### 7.1.3.7 Impact of GAM Convergence Issue on Ozone Risk Estimates

Generalized Additive Models (GAM) have been widely utilized for epidemiologic analysis of the health effects attributable to air pollution. The impact of the GAM convergence issue was thoroughly discussed in Section 8.4.2 of the 2004 PM AQCD. Reports have indicated that using the default convergence criteria in the Splus software package for the GAM function can lead to biased regression estimates for PM and an underestimation of the standard error of the effect estimate (Dominici et al., 2002; Ramsay et al., 2003). GAM default convergence criteria has a convergence precision of  $10^{-3}$  and a maximum number of 10 iterations. The more stringent convergence criteria refers to increased stringency of both the convergence precision and number of iterations. The default convergence criteria was found to be a problem when the estimated relative risks were small and two or more nonparametric smoothing curves were included in the GAM (Dominici et al., 2002). The magnitude and direction of the bias depend in part on the concurvity of the independent variables in the GAM and the magnitude of the risk estimate. Recent focus has been on the influence of the GAM function on effect estimates for PM. However, because O<sub>3</sub> covaries more strongly with both weather and time factors than does PM, the issue of GAM convergence criteria for O<sub>3</sub> also needs to be considered.

A meta-analysis by Stieb et al. (2003) found some difference in O<sub>3</sub>-mortality risk estimates between the GAM studies and non-GAM studies. GAM studies were defined as studies that analyzed effect estimates using nonparametric smoothing functions of time or weather. Non-GAM studies were all other studies, including those using Generalized Linear Models (GLM) and Generalized Estimating Equations (GEE) in their analysis. In the single-pollutant models, the O<sub>3</sub>-mortality risk estimates for the non-GAM studies (10 estimates) and GAM studies (15 estimates) were 1.8% (95% CI: 0.5, 3.1) and 2.2% (95% CI: 1.4, 2.8), respectively, per 40 ppb daily 1-h max O<sub>3</sub>. In the multipollutant models, the pooled risk estimate was 1.0% (95% CI: -0.5, 2.6) for non-GAM studies (7 estimates) and 0.5% (95% CI: -1.0, 1.9) for GAM studies (4 estimates).

Results from recent meta-analyses of O<sub>3</sub>-mortality effects suggest that there are no substantial differences between GAM-affected estimates and non-GAM-affected estimates (Bell et al., 2005; Ito et al., 2005; Levy et al., 2005). GAM-affected studies included those that used default convergence criteria. Non-GAM-affected studies included GAM studies that used stringent convergence criteria or those that used other modeling techniques. Ito et al. (2005)

1 found that the single-pollutant combined estimate for the GAM-affected studies (15 estimates)  
2 and non-GAM-affected studies (28 estimates) were 1.92% (95% CI: 1.02, 2.81) and 1.40%  
3 (95% CI: 0.78, 2.02), respectively, per 20 ppb increase in 24-h avg O<sub>3</sub>. In the analysis by Levy  
4 et al. (2005), the single-pollutant combined estimate for the GAM-affected studies (29 estimates)  
5 and non-GAM-affected studies (17 estimates) were 1.56% (95% CI: 1.01, 2.11) and 1.80%  
6 (95% CI: 1.17, 2.43), respectively, per 40 ppb increase in 1-h max O<sub>3</sub>. Bell et al. (2005) also  
7 reported that the pooled estimate was larger for the studies that were not GAM-affected.

8 A few GAM studies reanalyzed O<sub>3</sub> risk estimates using more stringent convergence criteria  
9 or GLM. Reanalysis of an asthma hospital admissions study in Seattle, WA (Sheppard et al.,  
10 1999; reanalysis Sheppard, 2003) indicated that there were only slight changes in the risk  
11 estimates when using more stringent convergence precision (10<sup>-8</sup>) in GAM. The original GAM  
12 analysis indicated an excess risk of 9% (95% CI: 3, 17) whereas the stringent GAM analysis  
13 found an excess risk of 11% (95% CI: 3, 19) per 30 ppb increase in 8-h max O<sub>3</sub> at a 2-day lag.  
14 Similar results were found using GLM with natural splines, 11% (95% CI: 2, 20). In the  
15 reanalysis of Santa Clara County, CA data, Fairley (1999; reanalysis Fairley, 2003) used the  
16 same methods as the original analysis except the convergence precision ( $\epsilon$ ) was increased from  
17 10<sup>-4</sup> to 10<sup>-12</sup> and the maximum number of iterations (M) were increased from 10 to 10<sup>7</sup>. The  
18 O<sub>3</sub>-mortality risk estimate slightly increased from 2.8% (95% CI not provided) using default  
19 GAM parameters to 2.9% (95% CI: -0.3, 6.0) using stringent GAM parameters per 30 ppb  
20 increase in 8-h max O<sub>3</sub> at a 0-day lag. The O<sub>3</sub>-mortality risk estimates further increased to 3.0%  
21 (95% CI: -0.3, 6.3) using GLM with natural cubic splines. In the reanalysis of the Netherlands  
22 data by Hoek et al. (2000; reanalysis Hoek, 2003), the O<sub>3</sub> nonaccidental mortality risk estimates  
23 increased from 1.3% (95% CI: 0.8, 1.9) using default GAM to 1.5% (95% CI: 1.0, 2.1) using  
24 stringent GAM ( $\epsilon = 10^{-8}$ , M = 10<sup>3</sup>) and 1.6% (95% CI: 0.9, 2.4) using GLM with natural splines  
25 per 30 ppb increase in 8-h avg O<sub>3</sub> (12 p.m. - 8 p.m.) at a 1-day lag.

26 In the limited number of studies that have reanalyzed O<sub>3</sub> risk estimates, there is little  
27 evidence that default GAM analyses resulted in positively biased estimates as observed for PM.  
28 Generally it appears that the use of default convergence criteria in GAM tends to bias risk  
29 estimates towards the null, in addition to underestimating the standard errors. However, one  
30 study by Cifuentes et al. (2000) in Santiago, Chile observed a large difference in the O<sub>3</sub>-  
31 mortality excess risks calculated using default GAM (0.9% [95%CI: 0.2, 1.6] per 40 ppb

1 increase in 1-h max O<sub>3</sub>) and GLM (0.1% [95% CI: -0.6, 0.8]). The GAM convergence problem  
2 appears to vary depending on data sets, and likely depends upon the intercorrelation among  
3 covariates and the magnitude of the risk estimate; thus, its impact on the results of individual  
4 studies cannot be known without a reanalysis. In uniformity with the approach used in the 2004  
5 PM AQCD, the results from studies that analyzed data using GAM with default convergence  
6 criteria and at least two nonparametric smoothing terms are generally not considered in this  
7 chapter, with some exceptions as noted.

#### 9 **7.1.4 Approach to Presenting Ozone Epidemiologic Evidence**

10 To produce a thorough appraisal of the evidence, key information (including study design,  
11 analysis, mean O<sub>3</sub> concentrations, and health outcome results) from important new studies is  
12 presented in summary tables in Chapter 7 of the Annex. Each section of the chapter starts by  
13 concisely highlighting important points derived from the 1996 O<sub>3</sub> AQCD assessment. In the  
14 main body of the chapter, particular emphasis is focused on studies and analyses that provide  
15 pertinent information for the critical assessment of health risks from O<sub>3</sub> exposure. Not all studies  
16 should be accorded equal weight in the overall interpretive assessment of evidence regarding  
17 O<sub>3</sub>-associated health effects. Among well-conducted studies with adequate control for  
18 confounding, increasing scientific weight should be accorded in proportion to the precision of  
19 their effect estimates. Small-scale studies without a wide range of exposures generally produce  
20 less precise estimates compared to larger studies with a broad exposure gradient. The size of the  
21 study, as indicated by the length of the study period and total number of events, and the  
22 variability of O<sub>3</sub> exposures are important components of the precision of the health effect  
23 estimates. More weight should be accorded to estimates from studies with narrow confidence  
24 bands.

25 Emphasis is placed on text discussion of (1) new multicity studies that employ  
26 standardized methodological analyses for evaluating O<sub>3</sub> effects across several or numerous cities  
27 and often provide overall effect estimates based on combined analyses of information pooled  
28 across multiple cities; (2) studies that consider O<sub>3</sub> as a component of a complex mixture of air  
29 pollutants including PM and other gaseous criteria pollutants (CO, NO<sub>2</sub>, SO<sub>2</sub>); and (3) North  
30 American studies conducted in the U.S. or Canada. Multicity studies are of particular interest  
31 and value due to their evaluation of a wider range of O<sub>3</sub> exposures and large numbers of



1 observations. They generally provide more precise effect estimates than most smaller scale  
2 studies of single cities. Compared to meta-analyses of multiple “independent” studies, a  
3 potential advantage of multicity studies is consistency in data handling and model specifications  
4 which eliminates variation due to analysis approach. Also, unlike meta-analyses, they do not  
5 suffer from potential omission of nonsignificant results due to “publication bias.” Furthermore,  
6 geographic patterns of air pollution effects have the potential to provide especially valuable  
7 evidence regarding relative homogeneity and/or heterogeneity of O<sub>3</sub> health effects relationships  
8 across geographic locations. Due to the potential for confounding by copollutants, preference is  
9 given to studies with effect estimates from multipollutant models, i.e., models with both O<sub>3</sub> and  
10 PM rather than O<sub>3</sub>-only models. The potential impacts of different health care systems and the  
11 underlying health status of populations also need to be accounted for in the assessment  
12 (Hubbell et al., 2005; Levy et al., 2001); thus, U.S. studies are emphasized over non-U.S.  
13 studies. In accordance to the emphasis placed on the O<sub>3</sub> epidemiologic studies in this chapter,  
14 the tables in the Chapter 7 Annex were organized by region with multicity studies in each region  
15 presented first.

16 In the coming sections, field/panel studies and studies of emergency department visits and  
17 hospital admissions, which contributed to the establishment of the revised 1997 NAAQS for O<sub>3</sub>,  
18 are presented first. This is followed by a discussion of O<sub>3</sub>-related mortality and effects of  
19 chronic exposures to O<sub>3</sub>. The chapter ends with an integrative discussion providing a summary  
20 and conclusions.

## 23 **7.2 FIELD STUDIES ADDRESSING ACUTE EFFECTS OF OZONE**

### 24 **7.2.1 Summary of Key Findings on Field Studies of Acute Ozone Effects** 25 **From the 1996 O<sub>3</sub> AQCD**

26 In the 1996 O<sub>3</sub> AQCD, individual-level camp and exercise studies provided useful  
27 quantitative information on the concentration-response relationships linking human lung  
28 function declines with ambient O<sub>3</sub> concentrations. The available body of evidence supported a  
29 dominant role of O<sub>3</sub> in the observed lung function decrements. Extensive epidemiologic  
30 evidence of pulmonary function responses to ambient O<sub>3</sub> came from camp studies. Six studies  
31 from three separate research groups provided a combined database on individual

1 concentration-response relationships for 616 children (mostly healthy, nonasthmatic) ranging in  
2 age from 7 to 17 years, each with at least six sequential measurements of FEV<sub>1</sub> (forced  
3 expiratory volume in 1 second) while attending summer camps (Avol et al., 1990; Higgins et al.,  
4 1990; Raizenne et al., 1987, 1989; Spektor et al., 1988a, 1991). In the combined reanalysis by  
5 Kinney et al. (1996a) using consistent analytical methods, these data yielded an average  
6 relationship between afternoon FEV<sub>1</sub> and concurrent-hour O<sub>3</sub> concentration of -0.50 mL/ppb  
7 (95% CI: -0.63, -0.36), with study-specific slopes ranging from -1.29 to -0.19 mL/ppb.  
8 Exposure in camp studies usually extended for multiple hours. Although the regression results  
9 noted above were based on one-hour O<sub>3</sub> levels, single- and multiple-hour averages were  
10 observed to be highly correlated; thus, these results might represent, to some extent, the  
11 influence of multihour exposures. In addition to the camp study results, two studies involving  
12 lung function measurements before and after well-defined exercise events in adults yielded  
13 concentration-response slopes of -0.4 mL/ppb (95% CI: -0.7, -0.1) (Selwyn et al., 1985) and  
14 -1.35 mL/ppb (95% CI: -2.04, -0.66) (Spektor et al., 1988b). Ozone concentrations during  
15 exercise events of approximately ½-hour duration ranged from 4 to 135 ppb in these studies.

16 Results from other field panel studies also supported a consistent relationship between  
17 ambient O<sub>3</sub> exposure and acute respiratory morbidity in the population. Respiratory symptoms  
18 (or exacerbation of asthma) and decrements in peak expiratory flow (PEF) occurred with  
19 increased ambient O<sub>3</sub> concentrations, especially in asthmatic children (Lebowitz et al., 1991;  
20 Krzyzanowski et al., 1992). The results showed greater responses in asthmatic individuals than  
21 in nonasthmatics (Lebowitz et al., 1991; Krzyzanowski et al., 1992), indicating that asthmatics  
22 might constitute a sensitive group in epidemiologic studies of oxidant air pollution. Since the  
23 1996 O<sub>3</sub> AQCD, new research has examined a broad scope of field studies which are presented  
24 next.

## 26 **7.2.2 Introduction to Recent Field Studies of Acute Ozone Effects**

27 Numerous field studies carried out over the past decade have tested for and, in many cases,  
28 observed acute associations between measures of respiratory ill-health and O<sub>3</sub> concentrations in  
29 groups of subjects (Table AX7-1 in Chapter 7 Annex). Acute field studies are distinguished  
30 from time-series study designs in that they recruit and collect data from individual human  
31 subjects instead of utilizing administrative data on aggregate health outcomes such as daily

1 mortality, hospital admissions, or emergency department visits. Although individual-level health  
2 outcome data are collected in field studies, ambient O<sub>3</sub> concentrations from centrally located  
3 monitoring stations are generally used to assess exposure. Because of the logistical burden  
4 associated with direct data collection from individual subjects, field/panel studies tend to be  
5 small in both numbers of subjects and in duration of follow-up. While this may limit the  
6 statistical power of field studies as compared with time-series studies, the ability to determine  
7 individual-level information on health outcomes and potentially confounding factors adds  
8 scientific value.

9 The most common outcomes measured in acute field studies on the effects of air pollution  
10 exposure are lung function and various respiratory symptoms. Other respiratory outcomes  
11 examined on a limited basis include inflammation and generation of hydroxyl radicals in the  
12 upper airways, and school absences. Several studies examined cardiovascular outcomes  
13 including heart rate variability (HRV) and risk of myocardial infarctions (MI). The first group  
14 of studies provided varying degrees of evidence supporting the conclusion that elevated O<sub>3</sub> levels  
15 could have negative impacts on lung function and symptoms, confirming and adding to the body  
16 of knowledge that was presented in the 1996 O<sub>3</sub> AQCD. Some emphasis has been placed in  
17 examining the independent role of O<sub>3</sub> in the presence of PM and other pollutants. The other new  
18 studies contribute information on cardiopulmonary outcomes which have not been as well  
19 documented previously.

### 21 **7.2.3 Acute Ozone Exposure and Lung Function**

22 As discussed in the 1996 O<sub>3</sub> AQCD and in the earlier chapter of this document on  
23 controlled human exposure studies (Chapter 6), a large body of literature from clinical and field  
24 studies has clearly and consistently demonstrated reversible decrements in pulmonary function  
25 following acute O<sub>3</sub> exposure. Significant O<sub>3</sub>-induced spirometric and symptom responses have  
26 been observed in clinical studies of exercising healthy young adults (see Section 6.2) and in  
27 some potentially susceptible subpopulations, namely asthmatics and children (see Sections 6.3.2  
28 and 6.5.1). Field studies of acute O<sub>3</sub> exposure that examine pulmonary function fall into two  
29 distinct groupings, those that conduct spirometry (measuring FEV<sub>1</sub>, FVC [forced vital capacity],  
30 and other spirometric indices) and those that measure PEF using peak flow meters. Results from  
31 the previous O<sub>3</sub> AQCD and Chapter 6 of this document support the conclusion that the

1 spirometric parameter FEV<sub>1</sub> is a strong and consistent measure of lung function and may be used  
2 in the assessment of asthma (Fuhlbrigge et al., 2001). PEF is a closely related but different  
3 metric of lung function. PEF measurements have been shown to be more variable than FEV<sub>1</sub> in  
4 some studies (Vaughan et al., 1989; Cross and Nelson, 1991), and can have an element of  
5 uncertain reliability when self-administered by study subjects. However, Lippmann and Spektor  
6 (1998) state that PEF measurements from small, inexpensive flow meters, which are more  
7 feasible to use in field studies, have been shown to produce similar results to PEF measured  
8 spirometrically.

9 Studies of FEV<sub>1</sub> will be presented first, followed by a discussion of PEF studies. Other  
10 dividing aspects within these two major types of lung function studies include health status of  
11 subjects (e.g., healthy, mildly asthmatic, severely asthmatic), age group, time spent outdoors,  
12 and exertion levels. Several studies brought these factors together to produce informative data.  
13 Some FEV<sub>1</sub> studies involved both increased outdoor O<sub>3</sub> exposure and higher exertion levels.  
14 The results from this group of subjects may be comparable to those from exercising subjects in  
15 the clinical studies discussed in Chapter 6.

### 17 **7.2.3.1 Acute Ozone Studies with Spirometry (FEV<sub>1</sub>)**

18 Studies published over the past decade have provided some new insights on the acute  
19 effects of O<sub>3</sub> on FEV<sub>1</sub>. The results of all studies that investigated quantitative O<sub>3</sub>-related effects  
20 on FEV<sub>1</sub> are summarized in the following tables. Tables 7-1a,b,c present changes in FEV<sub>1</sub>  
21 associated with O<sub>3</sub> exposure in adults while Tables 7-2a,b,c present effects in children. Tables  
22 7-1b and 7-2b present the effect of O<sub>3</sub> on FEV<sub>1</sub> measured either in the morning or afternoon;  
23 Tables 7-1c and 7-2c present O<sub>3</sub> effects on changes in FEV<sub>1</sub> across the day (afternoon  
24 FEV<sub>1</sub> – morning FEV<sub>1</sub>). Studies that did not provide quantitative O<sub>3</sub> data were not included in  
25 the tables (Cuijpers et al., 1994; Delfino et al., 2004; Frischer et al., 1997). The data presented in  
26 Höppe et al. (1995a) were further analyzed in a subsequent paper (Höppe et al., 2003); results  
27 from the latter paper are included in the tables. In general, the O<sub>3</sub> effect estimates showed  
28 decrements for FEV<sub>1</sub> across studies, especially in children. The studies presented in the tables  
29 are discussed in further detail, starting with the O<sub>3</sub> effect on individuals with elevated exertion  
30 levels and increased exposure due to time spent outdoors, followed by its effect on other  
31 potential risk groups.

**Table 7-1a. Field Studies that Investigated the Association Between Acute Ambient O<sub>3</sub> Exposure and Changes in FEV<sub>1</sub> in Adults**

Reference	Study Location	Study Period	Mean O <sub>3</sub> (SD) Level, ppb	O <sub>3</sub> Index
Korrick et al. (1998)	Mount Washington, NH	Summers 1991, 1992	40 (12)	8-h avg
Brauer et al. (1996)	Fraser Valley, British Columbia, Canada	Jun-Aug 1993	40.3 (15.2)	1-h max
Schindler et al. (2001)	Eight communities in Switzerland	May-Sep 1991	46.6 (1.5-127.6) <sup>a</sup>	8-h avg
Höppe et al. (2003)	Munich, Germany	Apr-Sep 1992-1995	65.9 - 70.4 <sup>b</sup>	½-h max
Romieu et al. (1998)	Mexico City	Mar-May 1996; Jun-Aug 1996	123 (40)	1-h max

<sup>a</sup> Range of 8-h avg concentrations is presented by Schindler et al. (2001).

<sup>b</sup> Range of mean ½-h max O<sub>3</sub> concentrations on high O<sub>3</sub> days is presented for Höppe et al. (2003).

1 *Exercise and outdoor worker panels*

2 The current 8-hour NAAQS for O<sub>3</sub> was originally based on results from controlled human  
3 exposure studies, as discussed in Chapter 6. These field studies with subjects at elevated  
4 exertion levels are of particular interest due to their similarities to the human chamber studies.  
5 The majority of human chamber studies have examined the effects of O<sub>3</sub> exposure in subjects  
6 performing continuous or intermittent exercise for variable periods of time (see Chapter 6 of  
7 this O<sub>3</sub> AQCD).

8 A study by Brauer and colleagues (1996) reported unusually large O<sub>3</sub> effects on lung  
9 function among outdoor workers. This study presented O<sub>3</sub> effects during an extended outdoor  
10 exposure period combined with elevated levels of exertion. The investigators repeatedly  
11 measured spirometric lung function before and after outdoor summer work shifts over 59 days  
12 on a group of 58 berry pickers in Fraser Valley, British Columbia, Canada. The subjects, both  
13 male and female native Punjabi-speakers, ranged in age from 10 to 69 years old, with a mean age  
14 of 44 years. Outdoor work shifts averaged 11 hours in duration. The mean 1-h max O<sub>3</sub>  
15 concentration was 40.3 ppb (SD 15.2). Exertion levels were estimated using portable heart rate  
16 monitors carried over a period of four or more hours by a representative subset of subjects  
17 during 16 work shifts. Heart rates over the work shift averaged 36% higher than resting levels.

**Table 7-1b. Percent Changes in FEV<sub>1</sub> (95% CI) Associated with Acute Ambient O<sub>3</sub> Exposures in Adults, Ordered by Size of the Estimate <sup>a</sup>**

Reference	Study Population/Analysis	N	% Change in FEV <sub>1</sub>
1 Brauer et al. (1996) <sup>b</sup>	Berry pickers, next morning	58	-6.36 (-8.02, -4.70)
2 Brauer et al. (1996) <sup>b</sup>	Berry pickers, afternoon	58	-5.40 (-6.51, -4.28)
3 Romieu et al. (1998) <sup>c</sup>	Street workers on placebo, 1st phase (lag 0-1)	19	-3.55 (-6.28, -0.82)
4 Schindler et al. (2001)	Adults who never smoked (lag 0)	3912	-2.96 (-5.11, -0.76)
5 Romieu et al. (1998) <sup>c</sup>	Street workers on placebo, 1st phase (lag 0)	19	-2.17 (-3.45, -0.89)
6 Höppe et al. (2003) <sup>b</sup>	Athletes, afternoon (lag 0)	43	-1.26 (-2.63, 0.10)
7 Romieu et al. (1998) <sup>c</sup>	Street workers on supplement, 1st phase (lag 0-1)	22	-1.25 (-4.36, 1.86)
8 Romieu et al. (1998) <sup>c</sup>	Street workers on supplement, 1st phase (lag 0)	22	-0.53 (-2.08, 1.01)
9 Romieu et al. (1998) <sup>c</sup>	Street workers on placebo, 2nd phase (lag 0)	23	-0.40 (-1.94, 1.14)
10 Romieu et al. (1998) <sup>c</sup>	Street workers on placebo, 2nd phase (lag 0-1)	23	-0.36 (-2.93, 2.20)
11 Höppe et al. (2003)	Elderly, morning (lag 2)	41	-0.22 (-3.86, 3.42)
12 Romieu et al. (1998) <sup>c</sup>	Street workers on supplement, 2nd phase (lag 0)	19	0.18 (-0.72, 1.08)
13 Höppe et al. (2003) <sup>b</sup>	Athletes, afternoon (lag 2)	43	0.24 (-0.64, 1.12)
14 Höppe et al. (2003) <sup>b</sup>	Athletes, afternoon (lag 1)	43	0.48 (-0.97, 1.94)
15 Höppe et al. (2003) <sup>b</sup>	Athletes, morning (lag 2)	43	0.62 (-0.45, 1.68)
16 Höppe et al. (2003) <sup>b</sup>	Athletes, morning (lag 1)	43	0.71 (-0.65, 2.07)
17 Höppe et al. (2003)	Elderly, afternoon (lag 0)	41	0.75 (-2.08, 3.58)
18 Romieu et al. (1998) <sup>c</sup>	Street workers on supplement, 2nd phase (lag 0-1)	19	0.82 (-0.77, 2.42)
19 Höppe et al. (2003)	Elderly, afternoon (lag 1)	41	1.16 (-1.26, 3.58)
20 Höppe et al. (2003)	Elderly, morning (lag 1)	41	1.82 (-2.19, 5.84)
21 Höppe et al. (2003)	Elderly, afternoon (lag 2)	41	2.88 (-0.24, 6.00)

<sup>a</sup>Change in FEV<sub>1</sub> is per standard unit ppb O<sub>3</sub> (40 ppb for ½-h max O<sub>3</sub> and 1-h max O<sub>3</sub>, 30 ppb for 8-h max O<sub>3</sub>, and 20 ppb for 24-hr avg O<sub>3</sub>).

<sup>b</sup>Brauer et al. (1996) and Höppe et al. (2003) studies also included children. The study population for Brauer et-al. ranged in age from 10 to 69 years (mean age 44 years). For Höppe et al. (2003), the athletes ranged in age from 13 to 38 years (mean age 18 years).

<sup>c</sup>Romieu et al. (1998) present change in FEV<sub>1</sub> (mL). The data from Romieu et al. (1998) were transformed to percent change by dividing the estimates by 3,300 mL (average FEV<sub>1</sub> for 40 year old Mexican-American males by Hankinson et al., 1999).

**Table 7-1c. Cross-day Percent Changes in FEV<sub>1</sub> (95% CI) Associated with Acute Ambient O<sub>3</sub> Exposures in Adults, Ordered by Size of the Estimate <sup>a</sup>**

Reference	Study Population/Analysis	N	Cross-day % Change in FEV <sub>1</sub>	
1	Korrick et al. (1998)	Hikers with wheeze or asthma (post-pre-hike)	40	-4.47 (-7.65, -1.29)
2	Korrick et al. (1998)	Hikers who hiked 8-12 hours (post-pre-hike)	265	-2.07 (-3.78, -0.36)
3	Korrick et al. (1998)	Hikers age 28-37 years (post-pre-hike)	185	-2.01 (-3.42, -0.60)
4	Korrick et al. (1998)	Hikers who never smoked (post-pre-hike)	405	-1.77 (-3.24, -0.30)
5	Korrick et al. (1998)	Hikers male (post-pre-hike)	375	-1.65 (-3.12, -0.18)
6	Korrick et al. (1998)	Hikers age 38-47 years (post-pre-hike)	142	-1.59 (-3.12, -0.06)
7	Korrick et al. (1998)	All hikers (post-pre-hike)	530	-1.53 (-2.82, -0.24)
8	Korrick et al. (1998)	All hikers, with PM <sub>2.5</sub> and acidity in model (post-pre-hike)	530	-1.44 (-3.32, 0.44)
9	Korrick et al. (1998)	Hikers age 18-27 years (post-pre-hike)	135	-1.29 (-2.88, 0.30)
10	Korrick et al. (1998)	Hikers female (post-pre-hike)	155	-1.17 (-3.46, 1.12)
11	Korrick et al. (1998)	Hikers age 48-64 years (post-pre-hike)	68	-1.14 (-3.08, 0.80)
12	Korrick et al. (1998)	Hikers without wheeze or asthma (post-pre-hike)	490	-1.08 (-2.49, 0.33)
13	Korrick et al. (1998)	Hikers who hiked 2-8 hours (post-pre-hike)	265	-0.99 (-2.70, 0.72)
14	Korrick et al. (1998)	Hikers who formerly smoked (post-pre-hike)	125	-0.72 (-3.07, 1.63)
15	Brauer et al. (1996) <sup>b</sup>	Berry pickers (post-pre-work shift)	58	0.00 (-1.66, 1.66)

<sup>a</sup>Change in FEV<sub>1</sub> is per standard unit ppb O<sub>3</sub> (40 ppb for ½-h max O<sub>3</sub> and 1-h max O<sub>3</sub>, 30 ppb for 8-h max O<sub>3</sub>, and 20 ppb for 24-h avg O<sub>3</sub>).

<sup>b</sup>Brauer et al. (1996) study also included children. The study population ranged in age from 10 to 69 years (mean age 44 years).

1 Post-shift FEV<sub>1</sub> and FVC decreased as a function of O<sub>3</sub> concentration and the effects of O<sub>3</sub>  
2 remained significant after adjusting for PM<sub>2.5</sub> in the analysis. Declines in lung function also  
3 were observed on the morning following high O<sub>3</sub> exposure. The effects seen in this study are  
4 larger than have been reported previously in studies with briefer exposure durations. For  
5 example, afternoon FEV<sub>1</sub> was 3.8 mL (95% CI: -4.6, -3.0) lower per 1 ppb increase in O<sub>3</sub>  
6 concentrations, compared to the decline of 0.4 mL/ppb and 1.35 mL/ppb observed in the earlier  
7 adult exercise studies (Spektor et al., 1988b; Selwyn et al., 1985). These results are consistent

**Table 7-2a. Field Studies that Investigated the Association Between Acute Ambient O<sub>3</sub> Exposure and Changes in FEV<sub>1</sub> in Children**

Reference	Study Location	Study Period	Mean O <sub>3</sub> (SD) Level, ppb	O <sub>3</sub> Index
Linn et al. (1996)	Rubidoux, Upland, and Torrance, CA	Fall-spring 1992-1993, 1993-1994	23 (12)	24-h avg
Scarlett et al. (1996)	Surrey, England	Jun-Jul 1994	50.7 (24.48)	8-h max
Höppe et al. (2003)	Munich, Germany	Apr-Sep 1992-1995	65.9 - 70.4 <sup>a</sup>	½-h max
Ulmer et al. (1997)	Freudenstadt and Villingen, Germany	Mar-Oct 1994	Freudenstadt: 50.6 (22.5-89.7) <sup>b</sup> Villingen: 32.1 (0.5-70.1) <sup>b</sup>	½-h max ½-h max
Castillejos et al. (1995)	SW Mexico City	Aug 1990-Oct 1991	112.3 (0-365) <sup>c</sup>	1-h max
Romieu et al. (2002)	Mexico City	Oct 1998-Apr 2000	102 (47)	1-h max
Chen et al. (1999)	Sanchun, Taihsi, and Linyuan, Taiwan	May 1995-Jan 1996	19.7 - 110.3 <sup>c</sup>	1-h max

<sup>a</sup> Range of mean ½-h max O<sub>3</sub> concentrations on high O<sub>3</sub> days is presented for Höppe et al. (2003).

<sup>b</sup> Median and 90th percentile interval are presented for Ulmer et al. (1997).

<sup>c</sup> Range of 1-h max O<sub>3</sub> concentrations are presented by Castillejos et al. (1995) and Chen et al. (1999).

1 with the interpretation that extended exposures to O<sub>3</sub> produce more marked effects on lung  
 2 function. Further, when data were restricted to days with 1-h max O<sub>3</sub> concentrations under  
 3 40 ppb, the O<sub>3</sub> effects on afternoon FEV<sub>1</sub> did not change in magnitude and remained significant.  
 4 However, a possible role of copollutants cannot be completely excluded.

5 In a Mexico City study of 47 outdoor street workers (Romieu et al., 1998), spirometry was  
 6 performed repeatedly at the end of the work shift over a two-month period. Subjects were  
 7 exposed to outdoor ambient O<sub>3</sub> levels for a mean of 7.4 hours during the workday. Among those  
 8 who had never taken an antioxidant supplement (subjects who received a placebo during the  
 9 first phase of the study), same day O<sub>3</sub> concentrations were associated with decreases in FEV<sub>1</sub>.  
 10 A mean change of -71.6 mL (95% CI: -113.9, -29.3) (approximately a 4% decline) was  
 11 observed per 40 ppb increase in 1-h max O<sub>3</sub>. The results from this study, in addition to those  
 12 from the Canadian study of berry pickers (Brauer et al., 1996), indicate that outdoor workers are  
 13 a potentially vulnerable population that may need protection from O<sub>3</sub> exposures.



**Table 7-2b. Percent Changes in FEV<sub>1</sub> (95% CI) Associated with Acute Ambient O<sub>3</sub> Exposures in Children, Ordered by Size of the Estimate <sup>a</sup>**

Reference	Study Population/Analysis	N	% Change in FEV <sub>1</sub>	
1	Ulmer et al. (1997) <sup>b</sup>	School children in Freudenstadt (lag 1)	57	-4.60 (-7.54, -1.67)
2	Ulmer et al. (1997) <sup>b</sup>	School boys in Freudenstadt and Villingen (lag 1)	67	-3.23 (-6.47, 0.00)
3	Ulmer et al. (1997) <sup>b</sup>	School children in Freudenstadt and Villingen (lag 1)	135	-2.98 (-5.33, -0.63)
4	Ulmer et al. (1997) <sup>b</sup>	School girls in Freudenstadt and Villingen (lag 1)	68	-2.32 (-5.53, 0.88)
5	Höppe et al. (2003) <sup>c</sup>	Asthmatics, afternoon (lag 2)	43	-2.08 (-6.24, 2.08)
6	Chen et al. (1999)	Children, with NO <sub>2</sub> in model (lag 1)	895	-1.97 (-3.51, -0.43)
7	Chen et al. (1999)	Children (lag 1)	895	-1.48 (-2.84, -0.12)
8	Romieu et al. (2002) <sup>b</sup>	Moderate to severe asthmatic children on placebo (lag 1)	35	-0.99 (-1.80, -0.18)
9	Romieu et al. (2002) <sup>b</sup>	Moderate to severe asthmatic children on placebo, with NO <sub>2</sub> and PM <sub>10</sub> in model (lag 1)	35	-0.97 (-1.87, -0.07)
10	Chen et al. (1999)	Children (lag 2)	895	-0.93 (-2.56, 0.71)
11	Ulmer et al. (1997) <sup>b</sup>	School children in Villingen (lag 1)	78	-0.79 (-3.93, 2.34)
12	Chen et al. (1999)	Children (lag 7)	895	-0.72 (-1.81, 0.37)
13	Höppe et al. (2003) <sup>c</sup>	Asthmatics, afternoon (lag 1)	43	-0.56 (-4.61, 3.50)
14	Linn et al. (1996) <sup>b</sup>	School children, next morning	269	-0.27 (-0.79, 0.24)
15	Linn et al. (1996) <sup>b</sup>	School children, afternoon	269	-0.19 (-0.73, 0.35)
16	Romieu et al. (2002) <sup>b</sup>	All asthmatic children on placebo (lag 1)	78	-0.19 (-0.71, 0.33)
17	Höppe et al. (2003)	Children, afternoon (lag 0)	44	-0.14 (-2.71, 2.42)
18	Höppe et al. (2003) <sup>c</sup>	Asthmatics, afternoon (lag 0)	43	-0.10 (-6.59, 6.39)
19	Romieu et al. (2002) <sup>b</sup>	Moderate to severe asthmatic on supplement (lag 1)	47	-0.04 (-0.80, 0.72)
20	Romieu et al. (2002) <sup>b</sup>	Moderate to severe asthmatic on supplement, with NO <sub>2</sub> and PM <sub>10</sub> in model (lag 1)	47	-0.01 (-0.82, 0.80)
21	Scarlett et al. (1996) <sup>d</sup>	School children (lag 1)	154	0.01 (-0.20, 0.22)
22	Romieu et al. (2002) <sup>b</sup>	All asthmatic children on supplement (lag 1)	80	0.04 (-0.52, 0.60)
23	Höppe et al. (2003) <sup>c</sup>	Asthmatics, morning (lag 1)	43	0.30 (-3.93, 4.53)
24	Höppe et al. (2003)	Children, morning (lag 1)	44	0.83 (-0.53, 2.20)

**Table 7-2b (cont'd). Percent Changes in FEV<sub>1</sub> (95% CI) Associated with Acute Ambient O<sub>3</sub> Exposures in Children, Ordered by Size of the Estimate <sup>a</sup>**

Reference	Study Population/Analysis	N	% Change in FEV <sub>1</sub>
25 Höppe et al. (2003)	Children, afternoon (lag 1)	44	0.93 (-0.80, 2.66)
26 Höppe et al. (2003)	Children, morning (lag 2)	44	1.17 (-0.36, 2.70)
27 Höppe et al. (2003)	Children, afternoon (lag 2)	44	1.20 (-0.12, 2.52)
28 Höppe et al. (2003) <sup>c</sup>	Asthmatics, morning (lag 2)	43	1.40 (-3.69, 6.49)

<sup>a</sup>Change in FEV<sub>1</sub> is per standard unit ppb O<sub>3</sub> (40 ppb for ½-h max O<sub>3</sub> and 1-h max O<sub>3</sub>, 30 ppb for 8-h max O<sub>3</sub>, and 20 ppb for 24-h avg O<sub>3</sub>).

<sup>b</sup>Linn et al. (1996), Romieu et al. (2002), and Ulmer et al. (1997) present change in FEV<sub>1</sub> (mL). The data were transformed to percent change by dividing the estimates by 1,900 mL (average FEV<sub>1</sub> among 8 to 10 year olds by Hankinson et al., 1999).

<sup>c</sup>Höppe et al. (2003) study also included young adults. The study population age for the asthmatics ranged from 12 to 23 years (mean age 15 years).

<sup>d</sup>FEV<sub>0.75</sub> results are presented in Scarlett et al. (1996).

**Table 7-2c. Cross-day Percent Changes in FEV<sub>1</sub> (95% CI) Associated with Acute Ambient O<sub>3</sub> Exposures in Children, Ordered by Size of the Estimate <sup>a</sup>**

Reference	Study Population/Analysis	N	Cross-day % Change in FEV <sub>1</sub>
1 Linn et al. (1996) <sup>b</sup>	School children (p.m. – a.m.)	269	-0.61 (-1.09, -0.14)
2 Castillejos et al. (1995)	Private primary school (post-pre-exercise)	40	-0.48 (-0.72, -0.24)

<sup>a</sup>Change in FEV<sub>1</sub> is per standard unit ppb O<sub>3</sub> (40 ppb for ½-h max O<sub>3</sub> and 1-h max O<sub>3</sub>, 30 ppb for 8-h max O<sub>3</sub>, and 20 ppb for 24-h avg O<sub>3</sub>).

<sup>b</sup>Linn et al. (1996) present change in FEV<sub>1</sub> (mL). The data were transformed to percent change by dividing the estimates by 1,900 mL (average FEV<sub>1</sub> among 8 to 10 year olds by Hankinson et al., 1999).

1 Höppe et al. (1995a) examined forestry workers (n = 41) for changes in pulmonary  
2 function attributable to O<sub>3</sub> exposure in Munich, Germany. In addition, athletes (n = 43) were  
3 monitored in the afternoon following a two-hour outdoor training period. Pulmonary function  
4 tests were conducted on days of both “high” (mean ½-h max O<sub>3</sub> of 64 to 74 ppb) and “low”  
5 (mean ½-h max O<sub>3</sub> of 32 to 34 ppb) ambient O<sub>3</sub> concentrations. From the average activity levels,  
6 ventilation rates were estimated. Athletes, who had a fairly high ventilation rate of 80 L/min,  
7 experienced a significant decrease of 60.8 mL (95% CI: 6.4, 115.2) in FEV<sub>1</sub> per 40 ppb increase

1 in ½-h max O<sub>3</sub>. Among the forestry workers, a similar O<sub>3</sub>-related decline in FEV<sub>1</sub> also was  
2 observed (-56.0 mL [95% CI: -118.4, 6.4]). In a subsequent study, Höpfe et al. (2003)  
3 reanalyzed the results of the athletes after stratifying the spirometric data by time of day  
4 (morning versus afternoon) and at different lag periods (lags of 0 to 2 days). The reanalysis  
5 indicated that O<sub>3</sub>-related decrements were observed only with the afternoon FEV<sub>1</sub> at a 0-day lag,  
6 -1.26% (95% CI: -2.63, 0.10) change in FEV<sub>1</sub> per 35 ppb increase in 3-h avg O<sub>3</sub>.

7 One FEV<sub>1</sub> study clearly demonstrated small but measurable effects of multihour O<sub>3</sub>  
8 exposures on adults exercising outdoors. In Korrnick et al. (1998), adult hikers (n = 530) of  
9 Mount Washington, NH performed spirometry before and after hiking for a mean of 8 hours  
10 (range 2–12). The mean hourly O<sub>3</sub> concentration ranged from 21 to 74 ppb. After the hike, all  
11 subjects combined experienced a small mean decline of 1.5% (95% CI: 0.2, 2.8) in FEV<sub>1</sub> and  
12 1.3% (95% CI: 0.5, 2.1) in FVC per 30 ppb increase in the mean of the hourly O<sub>3</sub> concentration  
13 during the hike. In addition, Korrnick et al. (1998) compared hikers who hiked 8 to 12 hours to  
14 those who hiked 2 to 8 hours. Among those who hiked longer, the percent change in FEV<sub>1</sub> was  
15 more than twofold greater per ppb exposure compared to those who hiked only for 2 to 8 hours.  
16 Each hour hiked, which may reflect dose, was associated with a decline of 0.3% (p = 0.05) in  
17 FEV<sub>1</sub>, after adjusting for O<sub>3</sub>.

18 In a Mexico City study, the O<sub>3</sub> effect attributable to exercise was determined using a group  
19 of school children (n = 40) chronically exposed to moderate to high levels of O<sub>3</sub> (Castillejos  
20 et al., 1995). Children were tested up to 8 times between August 1990 and October 1991.  
21 Spirometry was performed by the children before and after a one-hour intermittent exercise  
22 session outdoors. Outdoor O<sub>3</sub> levels ranged up to 365 ppb, with a mean of 112.3 ppb. Linear  
23 trend analyses indicated a relationship between quintiles of O<sub>3</sub> and percent change in lung  
24 function. However, stratified analyses indicated that significant changes were observed only  
25 with higher quintiles of O<sub>3</sub> exposure (72-125 ppb and 183-365 ppb). Therefore, children  
26 exercising at higher O<sub>3</sub> levels experienced declines in pulmonary function despite the repeated  
27 daily exposure to moderate and high levels of O<sub>3</sub> in Mexico City.

28 Collectively, the above studies confirm and extend clinical observations that prolonged  
29 exposure periods, combined with elevated levels of exertion or exercise, may magnify the effect  
30 of O<sub>3</sub> on lung function. The most representative data come from the Korrnick et al. (1998) hiker  
31 study. This U.S. study provided outcome measures stratified by several factors (e.g., gender,

1 age, smoking status, presence of asthma) within a population capable of more than normal  
2 exertion.

### 3 4 *Panel studies of children, elderly, and asthmatics*

5 Höppe et al. (1995a,b) examined several potentially susceptible populations for changes in  
6 pulmonary function attributable to O<sub>3</sub> exposure in Munich, Germany. The forestry workers and  
7 athletes were discussed in the previous section. Senior citizens (n = 41) and juvenile asthmatics  
8 (n = 43) were also monitored on “low” O<sub>3</sub> and “high” O<sub>3</sub> days. Subjects were requested to stay  
9 outdoors for at least 2 hours just before the afternoon pulmonary function test. Clerks (n = 40)  
10 were considered the nonrisk control group. Although clerks spent the majority of their time  
11 indoors, their outdoor exposures on high O<sub>3</sub> days were similar to that of the four other risk  
12 groups. The results showed no significant O<sub>3</sub> effects on the senior citizens. Clinical studies also  
13 have consistently shown that seniors are less responsive to O<sub>3</sub> (Bedi et al., 1989;  
14 Drechsler-Parks, 1995). Asthmatics and clerks experienced slight reductions in FEV<sub>1</sub> on high O<sub>3</sub>  
15 days. Among all risk groups, juvenile asthmatics experienced the largest O<sub>3</sub>-related decline in  
16 FEV<sub>1</sub>, -84.0 mL (95% CI: -196.4, 28.4) per 40 ppb increase in ½-h max O<sub>3</sub>. To further  
17 examine their hypotheses on characteristics of O<sub>3</sub> risk groups, Höppe et al. (2003) conducted a  
18 different analysis on a more expanded data base than utilized in the earlier study. Children were  
19 examined as an additional risk group. Höppe et al. (2003) presented both group mean values  
20 and analyses on an individual basis. For the group mean values, consistent O<sub>3</sub> effects were not  
21 detectable. On an individual basis, a potential pattern of O<sub>3</sub> sensitivity was observed (see  
22 Table AX7-1 in the Annex for details). About 20% of the children and asthmatics were regarded  
23 as O<sub>3</sub> responders (i.e., individuals with greater than 10% change in FEV<sub>1</sub>) compared to only 5%  
24 of the elderly and athletes. These results indicated that while the majority of the population did  
25 not react to O<sub>3</sub> exposure, a small group of susceptible individuals experienced health effects  
26 from O<sub>3</sub>. The sample size limits quantitative extrapolation to larger populations, but may allow  
27 cautious first estimates.

28 Several other panel studies performed spirometry in children, another potentially  
29 susceptible group (Avol et al., 1998; Chen et al., 1999; Cuijpers et al., 1994; Frischer et al.,  
30 1997; Linn et al., 1996; Romieu et al., 2002; Scarlett et al., 1996; Ulmer et al., 1997).  
31 All studies, with the exception of Avol et al. (1998) and Scarlett et al. (1996), observed a

1 decrease in FEV<sub>1</sub> associated with O<sub>3</sub> exposure. One large study measured spirometric lung  
2 function in 895 school children in three towns in Taiwan (Chen et al., 1999). Lung function was  
3 measured only once for each subject. The authors reported significant associations between  
4 diminished FEV<sub>1</sub> and FVC with a 1-day lag of O<sub>3</sub> concentrations. Effect sizes were typical of  
5 those observed in past studies, i.e., 0.5 to 1.0 mL decline in FEV<sub>1</sub> per ppb increase in O<sub>3</sub>  
6 concentration. Ozone was the only air pollutant associated with changes in lung function in  
7 multipollutant models including SO<sub>2</sub>, CO, PM<sub>10</sub>, and NO<sub>2</sub>.

8 Linn et al. (1996) repeatedly measured spirometric lung function among 269 school  
9 children in three southern California communities (Rubidoux, Upland, and Torrance). Lung  
10 function was measured over five consecutive days, once in each of three seasons over two school  
11 years. Between-week variability was controlled in the analysis by seasonal terms in the model.  
12 Statistical power was limited by the narrow range of exposures that were experienced within  
13 each week. In addition, the study was restricted to the school year, eliminating most of the  
14 “high” O<sub>3</sub> season from consideration. During the study period, 24-h avg O<sub>3</sub> levels at the central  
15 monitoring site ranged up to 53 ppb (mean 23 ppb) while personal measurements ranged up to  
16 16 ppb (mean 5 ppb). A mean change of -11.6 mL (95% CI: -20.6, -2.6) (approximately a 1%  
17 decline) in FEV<sub>1</sub> was observed from morning to afternoon per 20 ppb increase in 24-h avg O<sub>3</sub>.  
18 Other associations (involving individual morning or afternoon FVC and FEV<sub>1</sub> measurements)  
19 went in the plausible direction but the O<sub>3</sub> effect estimates were considerably smaller.

20 Ulmer et al. (1997) examined 135 children aged 8 to 11 years in two towns in Germany  
21 from March to October 1994 for O<sub>3</sub> effects on pulmonary function at four time periods. The  
22 cross-sectional results at each of the four time points showed limited FVC and no FEV<sub>1</sub>  
23 associations. However, the longitudinal analysis, which combined data from all four periods  
24 yielded a mean change of -87.5 mL (95% CI: -143.2, -31.7) (approximately a 5% decline)  
25 in FEV<sub>1</sub> per 40 ppb increase in ½-h max O<sub>3</sub> for the town with the higher O<sub>3</sub> levels (median ½-h  
26 max of 50.6 ppb versus 32.1 ppb). In the cross-sectional analysis, only between-person  
27 variability was analyzed. The longitudinal analysis, in which the subjects provided multiple  
28 days of measurements, provided information on both between- and within-subject responses.

29 There are a limited number of new epidemiologic studies examining the effects of O<sub>3</sub> on  
30 FEV<sub>1</sub>; however, results from these studies indicate that acute exposure to O<sub>3</sub> is associated with  
31 declines in FEV<sub>1</sub> in children. These results further support the negative effects of O<sub>3</sub> on lung

1 function observed in the meta-analysis on children attending summer camp (Kinney et al.,  
2 1996a) and in the clinical literature.

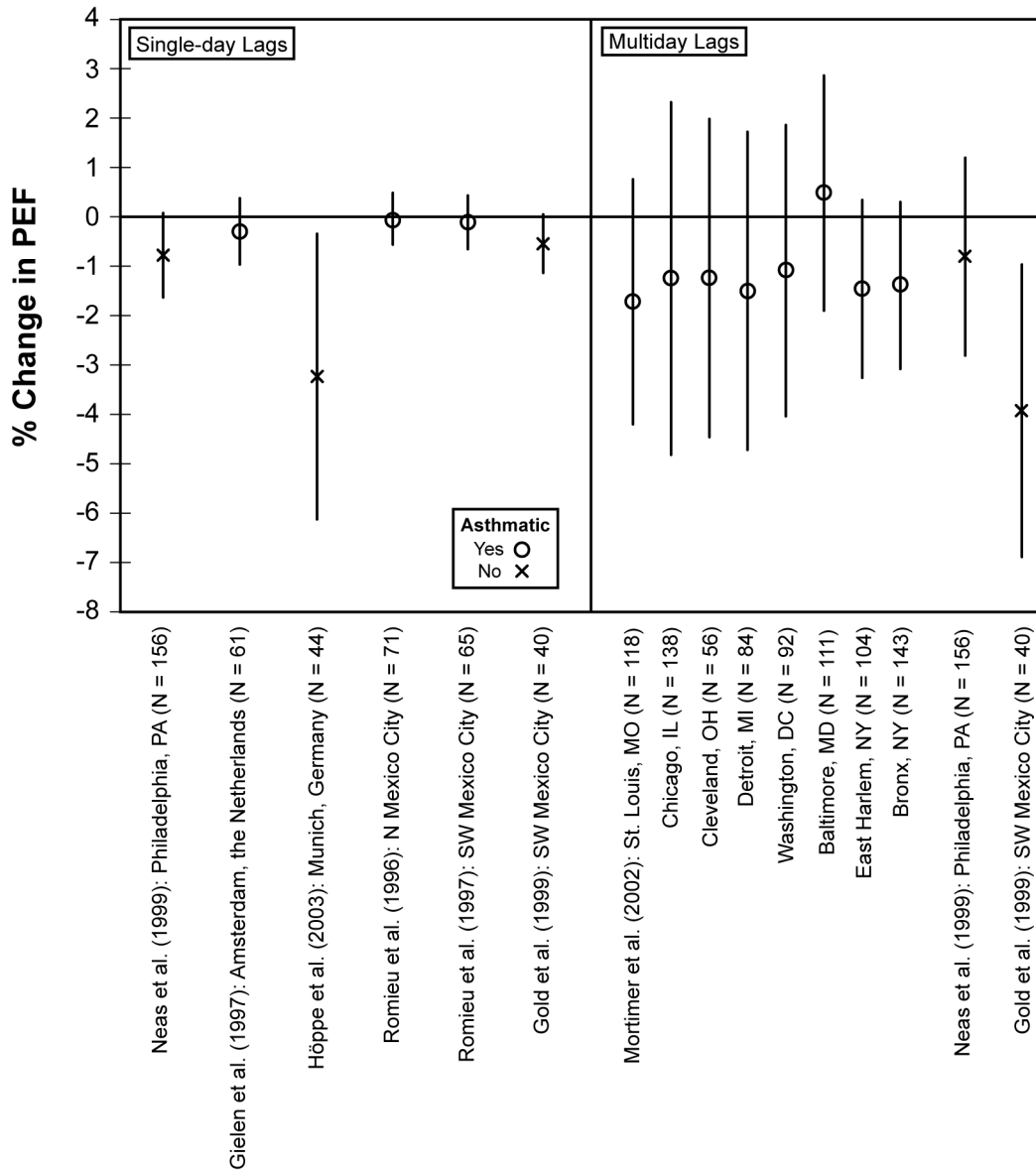
### 4 **7.2.3.2 Acute Ozone Studies of PEF**

5 Many studies of the acute effect of O<sub>3</sub> on PEF examined self-administered PEF levels  
6 daily, both in the morning and afternoon. PEF follows a circadian rhythm with the highest  
7 values found during the late afternoon and lowest values during the night and early morning.  
8 Due to the diurnal variation in PEF, most studies analyzed their data after stratifying by time of  
9 day. The peak flow studies examined both asthmatic panels and healthy individuals. The  
10 asthma panels are discussed first.

#### 12 *Asthma panels*

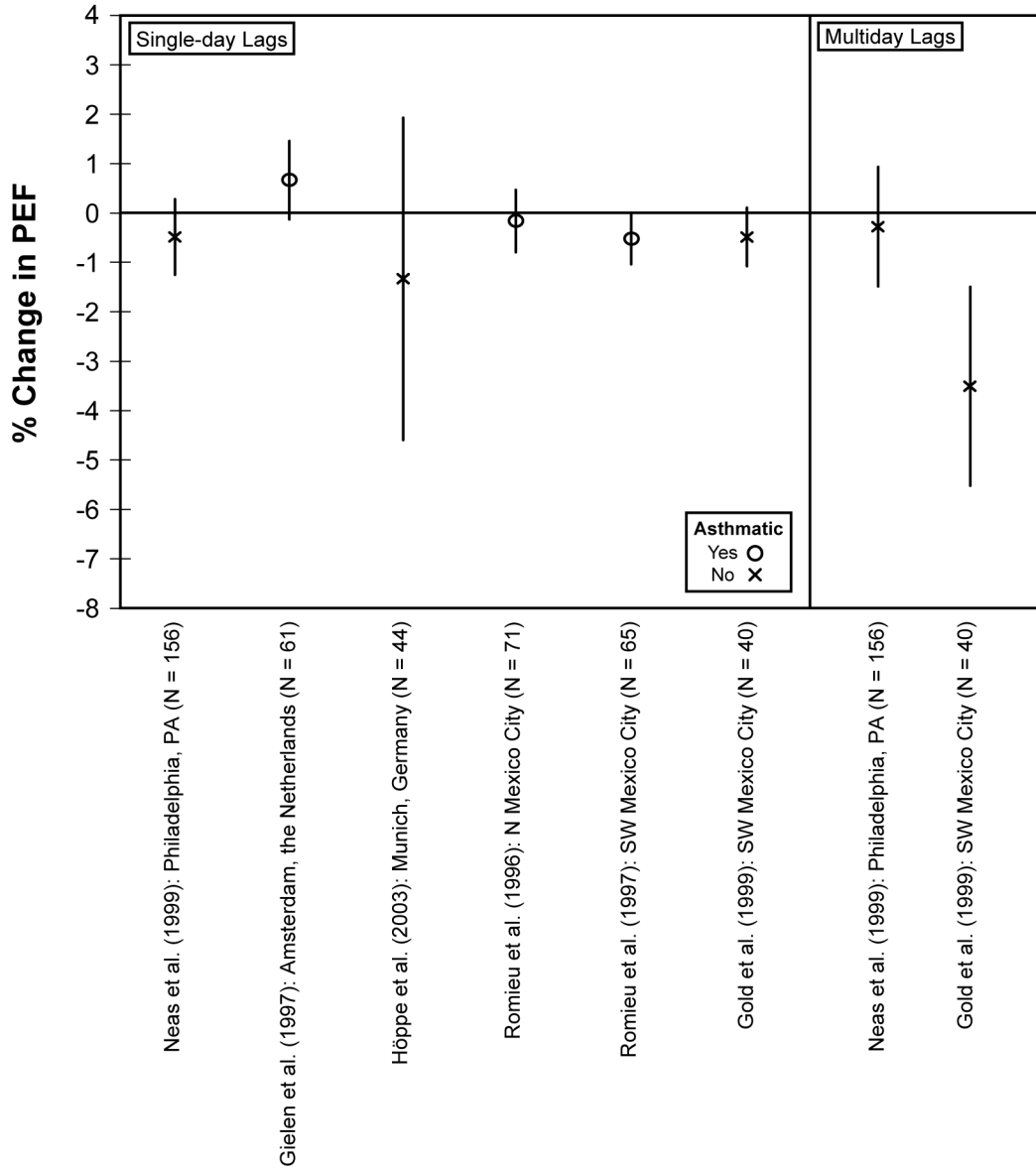
13 The effects of acute O<sub>3</sub> exposure on PEF in asthmatics were examined in several panel  
14 studies. Figures 7-1 and 7-2 present percent changes in morning and evening PEF outcomes  
15 from seven panel studies of children, mostly asthmatic, ranging in age from 5 to 13 years. The  
16 effect estimates from all single-day and multiday lag models are presented. Only single-city  
17 results with analyses stratified by morning and afternoon are included in the figure. Studies that  
18 examined cross-day changes and daily variability in PEF (e.g., Just et al., 2002; Thurston et al.,  
19 1997) are not included in the figure since such outcomes are not directly comparable.  
20 Collectively, nearly all of the studies indicated decrements of peak flow but most of the  
21 individual estimates were not statistically significant. The results from the individual studies are  
22 further discussed below.

23 In Mexico City, two studies of asthmatic school children were carried out simultaneously  
24 in the northern (Romieu et al., 1996) and southwestern sections of the city (Romieu et al., 1997).  
25 In the northern study, 71 mildly asthmatic school children aged 5 to 13 years old, were followed  
26 over time for daily morning (before breakfast) and afternoon (bedtime) PEF. In single-pollutant  
27 models, O<sub>3</sub> concentrations at 0-, 1-, and 2-day lags were associated with diminished morning and  
28 afternoon PEF, but only the 0-day lag morning effect was significant. The O<sub>3</sub> effect became  
29 nonsignificant when PM<sub>2.5</sub> was added to the model. In the southwestern study, 65 mildly  
30 asthmatic children aged 5 to 13 years old were followed during the summer and winter for daily  
31 morning and afternoon PEF. Ozone concentrations at a 0- and 1-day lag were associated with



**Figure 7-1. Percent change (95% CI) in morning PEF in children per standardized increment (see Section 7.1.3.2). For single-day lag models, previous day O<sub>3</sub> effects are shown. For multiday lag models, the cumulative effects of a 1- to 5-day lag are shown for Mortimer et al. (2002) and Neas et al. (1999), and the effect of a 1- to 10-day lag is shown for Gold et al. (1999).**

1 afternoon PEF, with larger effects at a 1-day lag. Associations involving O<sub>3</sub> were stronger than  
 2 those involving PM<sub>10</sub>. Several additional studies, both in the U.S. and in other countries,  
 3 reported significant associations between O<sub>3</sub> exposure and decrements in PEF among asthmatics



**Figure 7-2. Percent change (95% CI) in afternoon PEF in children per standardized increment (see Section 7.1.3.2). For single-day lag models, current day O<sub>3</sub> effects are shown. For multiday lag models, the cumulative effect of a 1- to 5-day lag is shown for Neas et al. (1999) and a 1- to 9-day lag is shown for Gold et al. (1999).**

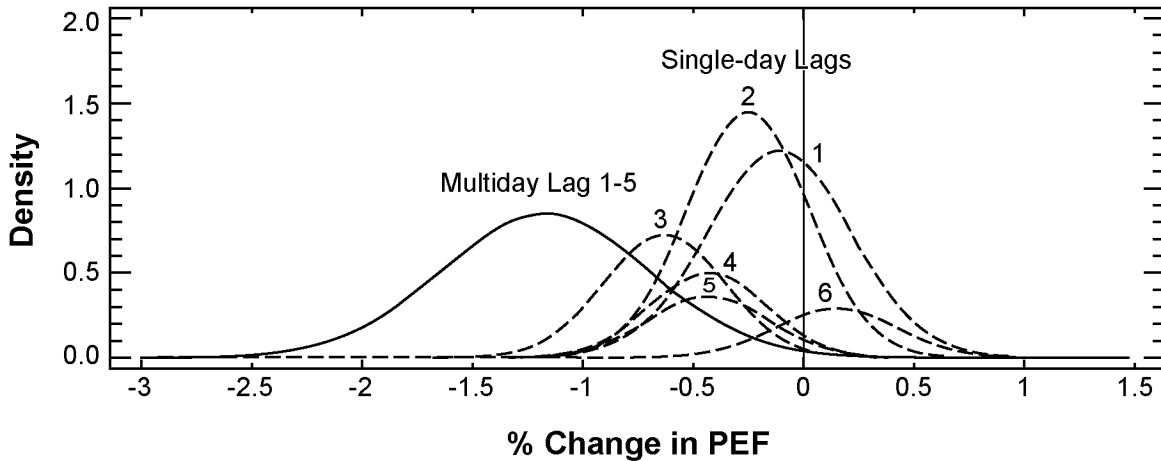
1 (Gielen et al., 1997; Jalaludin et al., 2000; Just et al., 2002; Ross et al., 2002; Thurston et al.,  
 2 1997).



1 Other epidemiologic studies did not find a significant O<sub>3</sub> effect on the lung function of  
2 asthmatics. Delfino et al. (1997a) examined morning and evening PEF among 22 asthmatics  
3 ranging in age from 9 to 46 years, living in Alpine, CA. Daily ambient 12-h avg O<sub>3</sub>  
4 (8 a.m.-8 p.m.) concentrations ranged from 34 to 103 ppb, with a mean value of 64 ppb.  
5 Unique to this study, personal O<sub>3</sub> exposures were measured using 12-h passive O<sub>3</sub> samplers that  
6 were worn by the subjects. The personal 12-h avg O<sub>3</sub> (8 a.m.-8 p.m.) concentrations, which had  
7 a mean value of 18 ppb, were much lower than the fixed-site ambient levels. Quantitative O<sub>3</sub>  
8 results were not reported but researchers stated that no O<sub>3</sub> effects were observed on morning and  
9 evening PEF. In Hiltermann et al. (1998), 60 nonsmoking adults aged 18 to 55 years in  
10 Bilthoven, the Netherlands, were followed between July and October 1995 with morning and  
11 afternoon PEF measurements. Although negative associations were observed between O<sub>3</sub> and  
12 cross-day changes in PEF, the results were not significant.

13 Mortimer et al. (2002) examined 846 asthmatic children from the National Cooperative  
14 Inner-City Asthma Study (NCICAS) for O<sub>3</sub>-related changes in PEF. Children from eight urban  
15 areas in the U.S. (St. Louis, MO; Chicago, IL; Detroit, MI; Cleveland, OH; Washington, DC;  
16 Baltimore, MD; East Harlem, NY; and Bronx NY) were monitored from June through August  
17 1993. This study provides representative data for the U.S. as children from multiple cities  
18 throughout the East and Midwest were examined. Asthmatic children from urban areas are an  
19 important subgroup of potentially at-risk populations. Study children either had physician-  
20 diagnosed asthma and symptoms in the past 12 months or respiratory symptoms consistent with  
21 asthma that lasted more than 6 weeks during the previous year.

22 Mortimer et al. (2002) examined O<sub>3</sub>-related changes in PEF for single-day lags from 1 to  
23 6 days and a multiday lag period of 5 days. Of all the pollutants examined, including O<sub>3</sub>, PM<sub>10</sub>,  
24 NO<sub>2</sub>, and SO<sub>2</sub>, none were associated with evening PEF. Only O<sub>3</sub> was found to be associated  
25 with morning PEF. The effect estimates of the association between O<sub>3</sub> and morning PEF for the  
26 single-day and multiday lags are depicted as error density curves in Figure 7-3 (for description of  
27 error density curves, see Annex Section AX7-2). Small morning effects were observed at 1- and  
28 2-day lags. The effect of O<sub>3</sub> on morning outcomes increased over several days. The strongest  
29 association between O<sub>3</sub> and PEF was found with a multiday lag period (cumulative lag of 1 to 5  
30 days). Unrestricted lag models suggested that the O<sub>3</sub> exposure from 3 to 5 days prior had a  
31 greater impact on morning % PEF than more immediate exposures. Mortimer et al. discussed

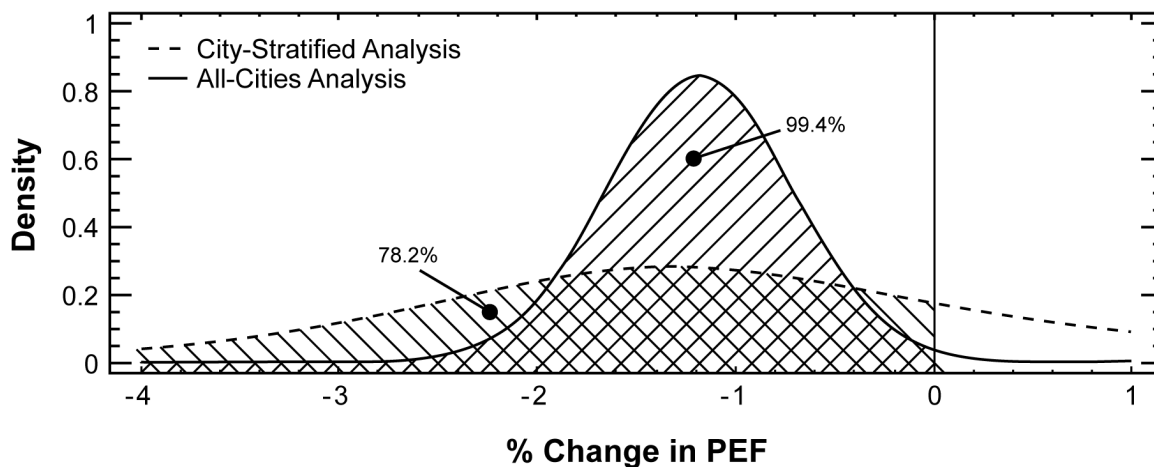


**Figure 7-3. Comparison of single-day lags (1-, 2-, 3-, 4-, 5-, and 6-day) to a cumulative multiday lag (1- to 5-day) for percent changes in PEF per 30 ppb increase in 8-h avg O<sub>3</sub> in urban children.**

Source: Derived from Mortimer et al. (2002).

1 biological mechanisms for delayed effects on pulmonary function, which included increased  
 2 bronchial reactivity secondary to airway inflammation associated with irritant exposure. Animal  
 3 toxicology and human chamber studies (see Chapters 5 and 6) provide further evidence that  
 4 exposure to O<sub>3</sub> may augment cellular infiltration and cellular activation, enhance release of  
 5 cytotoxic inflammatory mediators, and alter membrane permeability.

6 Figure 7-4 illustrates the probability density curves of the results from the city-stratified  
 7 analysis and that from the pooled analysis of all eight cities. The error density curve for the  
 8 all-cities analysis is a graphical presentation of the all-cities regression analysis presented by  
 9 Mortimer et al. (2002), a change in morning PEF of -1.18% (95% CI: -2.10, -0.26) per 30 ppb  
 10 increase in 8-h avg O<sub>3</sub> (10 a.m.-6 p.m.) with a cumulative lag of 1 to 5 days. The summary  
 11 density curve for the city-stratified analysis was calculated by summing together eight normal  
 12 distribution functions, one for each of the study cities, then taking the derivative of the summed  
 13 function (see Annex Section AX7-2 for further explanation of summary density curves). The  
 14 area under the density curve and to the left of a value on the x-axis is an estimate of the  
 15 probability that the effect estimate will be less than or equal to that value. For example, the area  
 16 under the density curve to the left of 0% change in PEF is 99% in the all-cities analysis. A wider



**Figure 7-4. Density curves of the percent change in PEF per 30 ppb increase in 8-h avg O<sub>3</sub> with a cumulative lag of 1 to 5 days for the individual eight NCICAS cities and the pooled average of all cities. Note that 99% and 78% of the areas under the curves are less than zero for the pooled cities analysis and individual cities analysis, respectively.**

Source: Derived from Mortimer et al. (2002).

1 distribution was observed in the city-stratified analysis, with only 78% of the area less than zero.  
 2 The all-cities analysis likely had a smaller standard error compared to the city-specific analysis  
 3 as it was based upon more subjects and considered differences between cities to  
 4 vary about the same mean effect. The regression analysis by Mortimer et al. (2002) suggested a  
 5 lack of heterogeneity by city, as indicated by the nonsignificant interaction term between O<sub>3</sub>  
 6 effect and city. As shown in Figure 7-4, the summary density curve of the city-stratified analysis  
 7 has a peak at about the same value as the curve of the all-cities analysis, suggesting a common  
 8 O<sub>3</sub> effect for all eight cities and small variation among them. The unimodal shape of the density  
 9 curve of the city-stratified analysis also indicates the absence of outlying cities.

10 Mortimer et al. (2002) further noted that small declines in morning PEF may be of  
 11 uncertain clinical significance, thus they calculated the incidence of  $\geq 10\%$  declines in PEF.  
 12 A 5 to 15% change in FEV<sub>1</sub> has been expressed as having clinical importance to asthma  
 13 morbidity (American Thoracic Society, 1991; Lebowitz et al., 1987; Lippmann, 1988).  
 14 Although greater variability is expected in PEF measurements, a  $\geq 10\%$  change in PEF also may  
 15 have clinical significance. In Mortimer et al. (2002), O<sub>3</sub> was associated with an increased

1 incidence of  $\geq 10\%$  declines in morning PEF (odds ratio of 1.30 [95% CI: 1.04, 1.61] per 30 ppb  
2 increase in 8-h avg  $O_3$  for a 5-day cumulative lag). This finding suggests that exposure to  $O_3$   
3 might be related to clinically important changes in PEF in asthmatic children. This study also  
4 observed that excluding days when 8-h avg  $O_3$  levels were greater than 80 ppb provided effect  
5 estimates which were similar to those when all days were included in the analysis, indicating that  
6 the negative effect of  $O_3$  on morning PEF persisted at levels below 80 ppb. There is some  
7 concern, however, regarding the lack of an association between  $O_3$  and afternoon PEF.

8 Results from the multicities study by Mortimer et al. (2002), as well as those from several  
9 regional studies provide evidence of a significant relationship between  $O_3$  concentrations and  
10 PEF among asthmatics. Collectively, these studies indicate that  $O_3$  may be associated with  
11 declines in lung function in this potentially susceptible population.

### 12 *Panels of healthy subjects*

13 The effect of  $O_3$  on PEF in healthy subjects also was investigated in several studies.  
14 A study of 162 children (9 years of age) in England examined the relationship between  $O_3$  and  
15 PEF in the winter and summer seasons (Ward et al., 2002). The  $O_3$  effect estimates were  
16 generally positive in the winter and negative in the summer. Single-day lags of 0- to 3-days  
17 were examined; however, the strongest association was found with a multiday lag period.  
18 During the summer, a decline of 11.10 L/min (95% CI: 0.18, 21.98) was observed in morning  
19 PEF per 20 ppb increase in 24-h avg  $O_3$  with a 7-day cumulative lag. Smaller  $O_3$  effects were  
20 observed on afternoon PEF.

21 During the summer of 1990, Neas et al. (1995) examined 83 children in Uniontown, PA  
22 and reported twice daily PEF measurements. Researchers found that evening PEF was  
23 associated with  $O_3$  levels weighted by hours spent outdoors. Using a similar repeated measures  
24 design, Neas et al. (1999) saw evidence for effects due to ambient  $O_3$  exposure among  
25 156 children attending two summer day camps in the Philadelphia, PA area. Associations were  
26 found between afternoon PEF (recorded before leaving camp) and same-day  $O_3$  concentrations,  
27 and between morning PEF (recorded upon arrival at camp) and previous-day  $O_3$  concentrations.  
28 However, as in the case of Ward et al. (2002), the relationship between PEF and  $O_3$  was  
29 significant only when a multiday lag period was considered. Naeher et al. (1999), in a sample of  
30

1 473 nonsmoking women (age 19 to 43 years) living in Vinton, VA, also showed the strongest  
2 association between O<sub>3</sub> and evening PEF with a 5-day cumulative lag exposure.

3 Another study in southwestern Mexico City analyzed morning and afternoon PEF data  
4 collected from 40 school children aged 8 to 11 years (Gold et al., 1999). Subjects provided  
5 measurements upon arriving and before departing from school each day. A negative effect of O<sub>3</sub>  
6 on PEF was observed, -1.60 mL/s (95% CI: -3.56, 0.36) and -1.80 mL/s (95% CI: -3.76,  
7 0.16) per 20 ppb increase in 24-h avg O<sub>3</sub> on the same day afternoon and next day morning PEF,  
8 respectively. A greater effect was observed for PEF regressed on O<sub>3</sub> concentrations with a  
9 cumulative 10-day lag period (-3.50 mL/s [95% CI: -5.52, -1.49] on same day afternoon).  
10 These results suggest a longer, cumulative effect of O<sub>3</sub> on PEF. Alternatively, the associations  
11 observed at the 10-day lag period may reflect confounding by other time-varying factors or be a  
12 chance finding from an exploratory analysis.

13 In a recent study of 43 mail carriers in Taichung City, Taiwan, PEF was monitored twice  
14 daily during a six-week period (Chan and Wu, 2005). The mean 8-h avg O<sub>3</sub> (9 a.m.-5 p.m.)  
15 concentration during their work shift was 35.6 ppb (SD 12.1). Associations were observed  
16 between evening PEF and O<sub>3</sub> concentrations at lags of 0, 1 and 2 days. The greatest effect was  
17 observed at a lag of 1 day, a 2.07% decline in PEF per 30 ppb increase in 8-h avg O<sub>3</sub>  
18 (quantitative results for 95% CI not provided). Similar O<sub>3</sub> effects on morning PEF were  
19 observed. The effect of O<sub>3</sub> on PEF was robust to adjustment for copollutants; no association  
20 with PEF was observed for PM<sub>10</sub> and NO<sub>2</sub> in multipollutant models.  
21

#### 22 **7.2.4 Respiratory Symptoms**

23 Studies published over the past decade represent an improved new body of data on the  
24 symptom effects of O<sub>3</sub>. Respiratory symptoms in acute air pollution field studies are usually  
25 measured using questionnaire forms or “daily diaries” that are filled out by study subjects,  
26 usually without the direct supervision of research staff. Questions address the daily experience  
27 of coughing, wheezing, shortness of breath (or difficulty breathing), production of phlegm, and  
28 others. While convenient and potentially useful in identifying acute episodes of morbidity,  
29 measurements of daily symptoms are prone to a variety of errors. These include  
30 misunderstanding of the meaning of symptoms, variability in individual interpretation of  
31 symptoms, inability to remember symptoms if not recorded soon after their occurrence, reporting

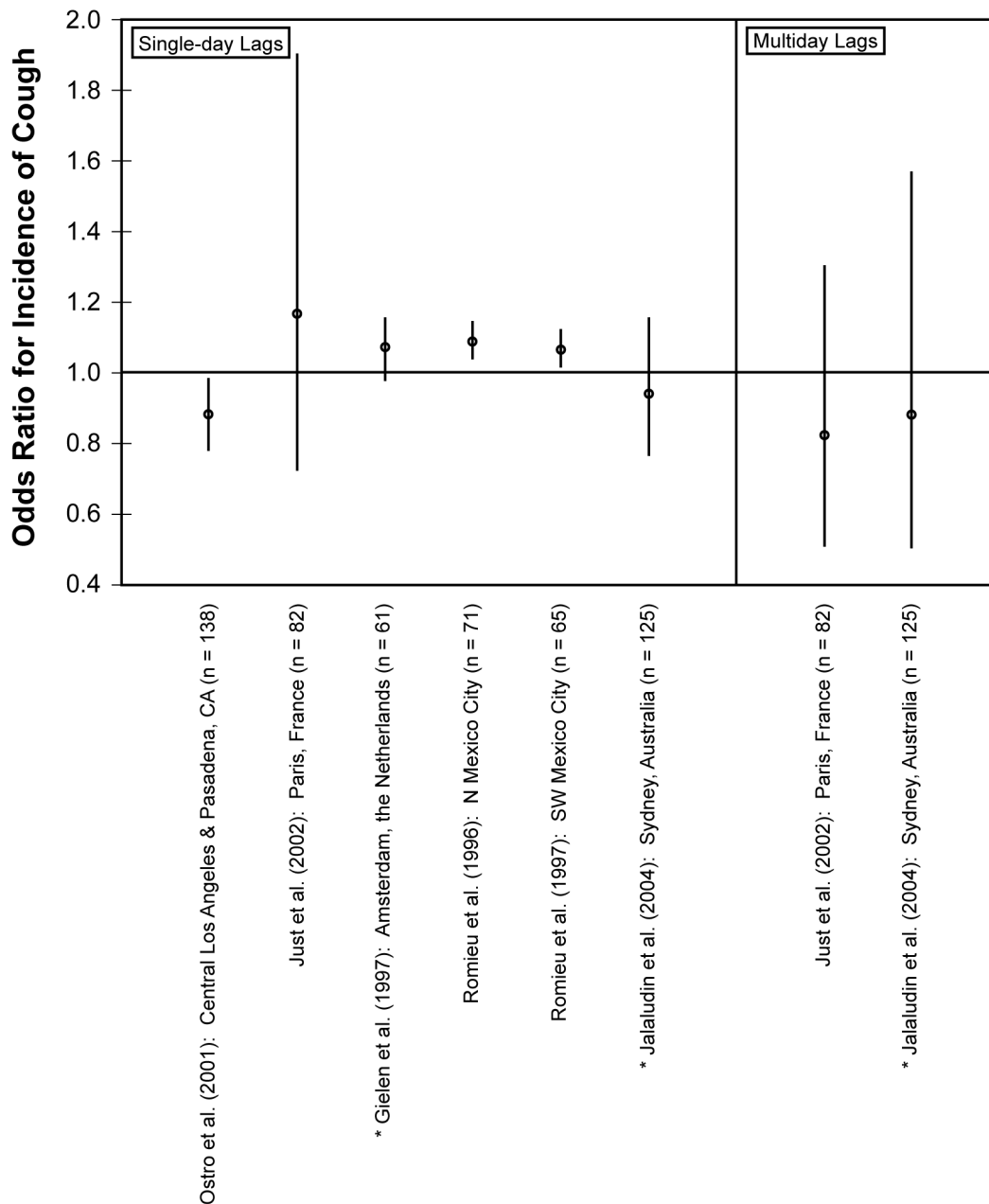
1 bias if days of high air pollution levels are identifiable by subjects, and the possibility of falsified  
2 data. In spite of these potential problems, the ease of data collection has made daily symptom  
3 assessment a common feature of field studies. Many of the studies reviewed above for lung  
4 function results also included measurements of daily symptoms. Pearce et al. (1998) reports that  
5 one advantage in the case of asthma panels is that the population is usually already familiar with  
6 symptom terms such as wheezing and cough. Delfino et al. (1998a) further states that the use of  
7 repeated daily symptom diaries has additional advantages of reducing recall bias given the  
8 proximity of events and allowing health effects to be modeled with each subject serving as their  
9 own control over time. Also, study design can blind the participants from the air pollution  
10 aspect of the study. Careful efforts by study staff can help ensure that the symptom diaries will  
11 provide information that is less affected by the potential problems noted.

12 Similar to studies of lung function, respiratory symptom studies can be divided into two  
13 groups, asthma panels or healthy subjects. Asthma panel studies are presented first.

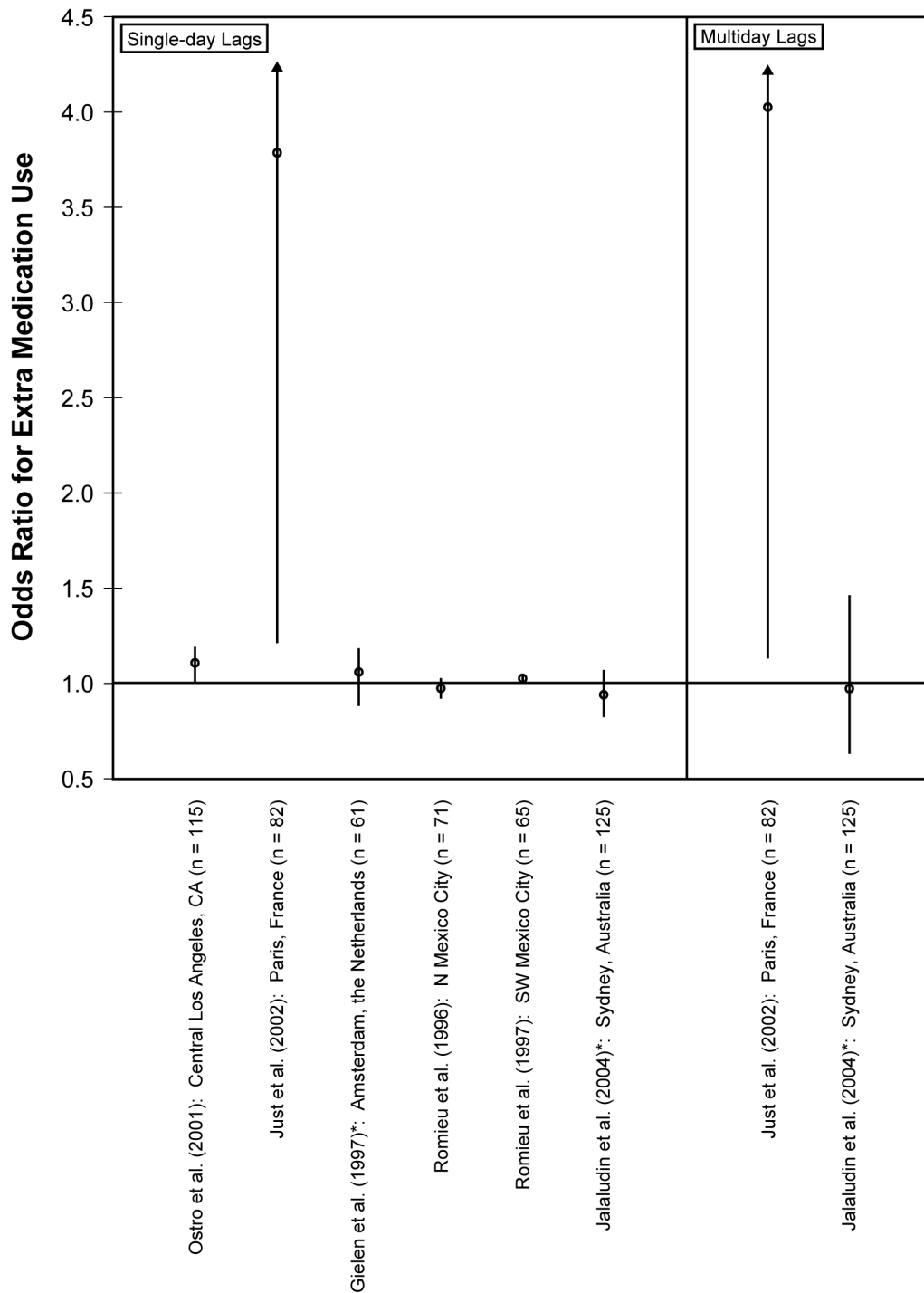
#### 14 *Asthma panels*

15 Most studies examining respiratory symptoms related to O<sub>3</sub> exposure focused on asthmatic  
16 children. Among the health outcomes, of particular interest were those associated with asthma,  
17 including cough, wheeze, shortness of breath, and increased medication use. Figures 7-5 and 7-6  
18 present the odds ratios for O<sub>3</sub>-related cough and medication use among asthmatic children from  
19 six studies (Gielen et al., 1997; Jalaludin et al., 2004; Just et al., 2002; Ostro et al., 2001; Romieu  
20 et al., 1996, 1997). Only single city/region studies that present odds ratios are included in the  
21 figure for consistency. Studies that present change in severity of symptoms, another informative  
22 health outcome, are not included in the figure since this symptom outcome differs from  
23 indicating simple presence of symptoms. The study by Gent et al. (2003) also is not included in  
24 this figure as odds ratios for cough and medication use were analyzed for quintiles of O<sub>3</sub>  
25 concentrations using the lowest quintile as the reference. These studies are discussed separately.

26 The various effect estimates for the association between O<sub>3</sub> concentrations and cough are  
27 depicted in Figure 7-5. Despite the variability in the individual effect estimates, there is some  
28 consistency in the O<sub>3</sub> effects. In general, the majority of the odds ratios appear to be greater than  
29 one among the single-day lag models, suggesting an association between acute exposure to O<sub>3</sub>  
30 and increased cough among asthmatic children. Figure 7-6 presents the odds ratios for  
31



**Figure 7-5. Odds ratios for the incidence of cough among asthmatic children per standardized increment (see Section 7.1.3.2). For single-day lag models, current day O<sub>3</sub> effects are shown with the exception of Ostro et al. (2001) which only presented results from a 3-day lag. For multiday lag models, the cumulative effects of a 0- to 4-day lag are shown. \*Note that Gielen et al. (1997) and Jalaludin et al. (2004) presented results for prevalence of cough.**



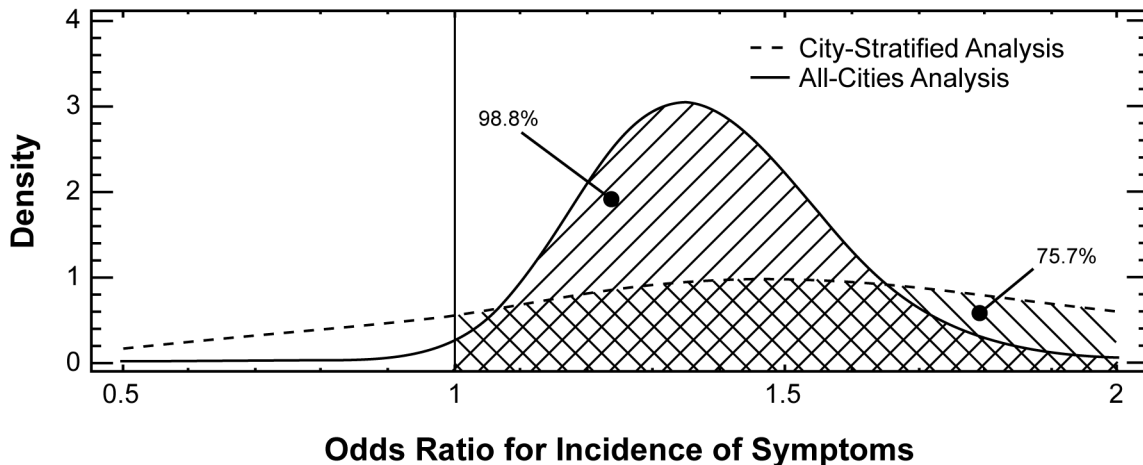
**Figure 7-6. Odds ratios for extra medication use among asthmatic children per standardized increment (see Section 7.1.3.2). For single-day lag models, current day O<sub>3</sub> effects are shown. For multiday lag models, the cumulative effects of a 0- to 4-day lag are shown.**



1 O<sub>3</sub>-associated bronchodilator use. The results from medication use are less consistent than those  
2 from cough; one study by Just et al. (2002) observed strong positive associations, but had wide  
3 confidence intervals.

4 Among the studies reporting results for daily symptoms and asthma medication use,  
5 several observed associations with O<sub>3</sub> concentrations that appeared fairly robust (Delfino et al.,  
6 2003; Desqueyroux et al., 2002a,b; Gent et al., 2003; Hiltermann et al., 1998; Just et al., 2002;  
7 Mortimer et al., 2000, 2002; Newhouse et al., 2004; Romieu et al., 1996, 1997; Ross et al., 2002;  
8 Thurston et al., 1997). Mortimer et al. (2002) reported morning symptoms in 846 asthmatic  
9 children from eight urban areas of the U.S. to be most strongly associated with a cumulative  
10 4-day lag of O<sub>3</sub> concentrations in the NCICAS. The NCICAS used standard protocols which  
11 included instructing caretakers of the subjects to record symptoms in the daily diary by  
12 observing or asking the child (Mitchell et al., 1997). Symptoms reported included cough, chest  
13 tightness, and wheeze. In the analysis pooling data from all eight cities, the odds ratio for the  
14 incidence of symptoms was 1.35 (95% CI: 1.04, 1.69) per 30 ppb increase in 8-h avg O<sub>3</sub>  
15 (10 a.m.-6 p.m.). Excluding days when 8-h avg O<sub>3</sub> was greater than 80 ppb, the odds ratio  
16 was 1.37 (95% CI: 1.02, 1.82) for incidence of morning symptoms. Figure 7-7 presents the  
17 probability density curves of the odds ratios for the incidence of symptoms from the city-  
18 stratified analysis and that from the all-cities analysis. This figure confirms the regression  
19 results that there is a significant increase in odds for incidence of symptoms, as the area under  
20 the density curve with an odds ratio greater than one is 99%. Mortimer et al. (2002) did not  
21 observe significant interactions among the eight cities, indicating that there was no heterogeneity  
22 among the city-specific estimates. The unimodal distribution of the city-stratified summary  
23 density curve also suggests a lack of significant heterogeneity in O<sub>3</sub> effects among the cities. It  
24 should be noted that other pollutants, including PM<sub>10</sub> (monitored in 3 cities), NO<sub>2</sub> (in 7 cities),  
25 and SO<sub>2</sub> (in all 8 cities), also were associated with increased incidence of morning symptoms.  
26 In multipollutant models, the O<sub>3</sub> effect was shown to be slightly diminished.

27 Another one of the larger studies was that of Gent and colleagues (2003), where  
28 271 asthmatic children under age 12 and living in southern New England were followed over  
29 6 months (April through September) for daily symptoms. The data were analyzed for two  
30 separate groups of subjects, 130 who used maintenance asthma medications during the follow-up  
31 period and 141 who did not. The need for regular medication was considered to be a proxy for



**Figure 7-7. Density curves of the odds ratios for the incidence of symptoms per 30 ppb increase in 8-h avg O<sub>3</sub> with a cumulative lag of 1 to 4 days for the individual eight cities and the pooled average of all cities. Note that 99% and 76% of the areas under the curves are greater than one for the pooled cities and individual cities analyses, respectively.**

Source: Derived from Mortimer et al. (2002).

1 more severe asthma. Not taking any medication on a regular basis and not needing to use a  
 2 bronchodilator would suggest the presence of very mild asthma. Effects of 1-day lag O<sub>3</sub> were  
 3 observed on a variety of respiratory symptoms only in the medication user group. Both daily  
 4 1-h max and 8-h max O<sub>3</sub> concentrations were similarly related to symptoms such as chest  
 5 tightness and shortness of breath. Effects of O<sub>3</sub>, but not PM<sub>2.5</sub>, remained significant and even  
 6 increased in magnitude in two-pollutant models. Some of the associations were noted at 1-h  
 7 max O<sub>3</sub> levels below 60 ppb. In contrast, no effects were observed among asthmatics not using  
 8 maintenance medication. In terms of person-days of follow-up, this is one of the larger studies  
 9 currently available that address symptom outcomes in relation to O<sub>3</sub>, and provides supportive  
 10 evidence for effects of O<sub>3</sub> independent of PM<sub>2.5</sub>. Study limitations include limited control for  
 11 meteorological factors and the post-hoc nature of the population stratification by medication use.

12 Some international studies have reported significant symptoms associations with O<sub>3</sub>.  
 13 The incidence of asthma attacks was associated with O<sub>3</sub> concentrations in a group of 60 severe  
 14 asthmatics (mean age 55 years) followed over a 13-month period in Paris (Desqueyroux et al.,

1 2002a). In a similar study, Desqueyroux et al. (2002b) observed O<sub>3</sub>-associated exacerbation of  
2 symptoms in 39 adult patients (mean age 67 years) with chronic obstructive pulmonary disease  
3 (COPD). Interestingly, in contrast to the controlled human studies (see Section 6.3.1, Subjects  
4 with COPD), the O<sub>3</sub> effect appeared larger among subjects who smoked and those with more  
5 severe COPD. However, the low O<sub>3</sub> concentrations experienced during this study (summer  
6 mean 8-h max O<sub>3</sub> of approximately 21 ppb [SD 9]) raise plausibility questions. In a study of  
7 60 nonsmoking asthmatic adults (aged 18 to 55 years) in Bilthoven, the Netherlands, Hilterman  
8 and colleagues (1998) reported associations between O<sub>3</sub> and daily symptoms of shortness of  
9 breath and pain upon deep inspiration. The O<sub>3</sub> associations were stronger than those of PM<sub>10</sub>,  
10 NO<sub>2</sub>, SO<sub>2</sub>, and black smoke (BS). No differences in response were evident between subgroups  
11 of subjects defined on the basis of steroid use or airway hyperresponsiveness. Daily use of  
12 bronchodilators or steroid inhalers was not found to be associated with O<sub>3</sub> in this study.

13 Other studies showed only limited or a lack of evidence for symptom increases associated  
14 with O<sub>3</sub> exposure (Avol et al., 1998; Chen et al., 1998; Delfino et al., 1996, 1997a, 1998a;  
15 Gielen et al., 1997; Jalaludin et al., 2004; Ostro et al., 2001; Taggart et al., 1996). Avol et al.  
16 (1998) studied symptoms in asthmatic, wheezy, and healthy children aged 10 to 12 years in  
17 southern California. Some symptom associations were noted but they were inconsistent. For  
18 example, children with wheeze were at increased risk of difficulty breathing and wheezing at  
19 low O<sub>3</sub> concentrations, but not at higher O<sub>3</sub> concentrations. Authors noted that O<sub>3</sub> concentrations  
20 were relatively low and that children studied did not spend substantial time outdoors engaged in  
21 physical activities. Ostro et al. (2001) reported no associations between daily symptoms and  
22 ambient O<sub>3</sub> concentrations in a cohort of 138 African-American children with asthma followed  
23 over 3 months (August to October) in Central Los Angeles and Pasadena, CA. However, the use  
24 of extra asthma medication was associated with 1-h max O<sub>3</sub> concentrations at a 1-day lag.  
25 Delfino and colleagues (1996) followed 12 asthmatic teens living in San Diego, CA for  
26 respiratory symptoms over a two-month period and saw no relationship with central site ambient  
27 O<sub>3</sub>. Personal O<sub>3</sub> exposures measured with passive diffusion monitors were associated with the  
28 composite symptom score and  $\beta_2$ -agonist inhaler use, but the relationship with symptom score  
29 disappeared when weekday/weekend differences were controlled in the statistical analysis.  
30 Study power was likely compromised by the small sample size. This observation of stronger  
31 associations with O<sub>3</sub> levels from personal monitors implies that gains in power may be achieved

1 if exposure misclassification is reduced through the use of personal exposure measurements  
2 rather than central site ambient O<sub>3</sub> concentrations. A similar study of 22 asthmatics in Alpine,  
3 CA observed no effects of O<sub>3</sub> on symptoms when personal O<sub>3</sub> exposure was used as the exposure  
4 metric (Delfino et al., 1997a). However, a later study in the same location involving 24 subjects  
5 (Delfino et al., 1998a) did find an association between respiratory symptoms and ambient O<sub>3</sub>  
6 exposure, with stronger O<sub>3</sub> effects experienced by asthmatics not on anti-inflammatory  
7 medication. In this study, a binary symptom score was used, whereas the earlier study used a  
8 linear symptom score of 0 through 6.

9 In conclusion, the various studies seem to indicate that O<sub>3</sub> concentrations are associated  
10 with respiratory symptoms and increased medication use in asthmatics. The multicities study by  
11 Mortimer et al. (2002) provides an asthmatic population most representative of the U.S., and  
12 several single-city studies also add to the knowledge base. However, there are a number of well-  
13 conducted, albeit relatively smaller studies that have not found these effects.

#### 14 *Panels of healthy subjects*

15 Fewer studies examined the effect of O<sub>3</sub> on respiratory symptoms in healthy individuals.  
16 Neas et al. (1995) reported that in school children, evening cough was associated with O<sub>3</sub> levels  
17 weighted by hours spent outdoors. The study by Linn and colleagues (1996) of 269 school  
18 children in southern California reported no associations between respiratory symptoms and O<sub>3</sub>,  
19 but subjects were exposed to fairly low O<sub>3</sub> concentrations as determined using personal  
20 monitors. Gold et al. (1999) examined symptoms in 40 healthy children in southwest Mexico  
21 City. Pollutant exposures were associated with increased production of phlegm in the morning,  
22 although the effects of the air pollutants (PM<sub>2.5</sub>, PM<sub>10</sub>, and O<sub>3</sub>) could not be separated in  
23 multipollutant models. Hoek and Brunekreef (1995) did not find a consistent association  
24 between ambient O<sub>3</sub> levels and the prevalence or incidence of respiratory symptoms in children  
25 living in two rural towns in the Netherlands. Collectively, these studies indicate that there is no  
26 consistent evidence of an association between O<sub>3</sub> and respiratory symptoms among healthy  
27 children.  
28  
29

## 7.2.5 Acute Airway Inflammation

Acute airway inflammation has been shown to occur among adults exposed to 80 ppb O<sub>3</sub> over 6.6 hours with exercise in controlled chamber studies (Devlin et al., 1991). Kopp and colleagues (1999) attempted to document inflammation of the upper airways in response to summer season O<sub>3</sub> exposures by following a group of 170 school children in two towns in the German Black Forest from March to October of 1994. To assess inflammation, the investigators collected nasal lavage samples at 11 time points spanning the follow-up period. The nasal lavage samples were analyzed for markers of inflammation, including eosinophil cationic protein, albumin, and leukocyte counts. Subjects who were sensitized to inhaled allergens were excluded. When analyzed across the entire follow-up period, no association was detected between upper airway inflammation and O<sub>3</sub> concentrations. More detailed analysis showed that the first significant O<sub>3</sub> episode of the summer was followed by a rise in eosinophil cationic protein levels, however, subsequent and even higher O<sub>3</sub> episodes had no effect. These findings suggest an adaptive response of inflammation in the nasal airways that is consistent with controlled human studies (see Section 6.9, Effects of Inflammation and Host Defense).

Frischer and colleagues (1993) collected nasal lavage samples from 44 school children in Umkirch, Germany the morning after “low” O<sub>3</sub> days (<140 µg/m<sup>3</sup> or approximately 72 ppb) and “high” O<sub>3</sub> days (>180 µg/m<sup>3</sup> or approximately 93 ppb) to measure levels of biochemical markers of inflammation. The researchers found that higher O<sub>3</sub> levels were associated with increased polymorphonuclear leukocyte counts in all children, and increases in myeloperoxidases and eosinophilic cation proteins among children without symptoms of rhinitis (n = 30). These results indicated that O<sub>3</sub> was associated with inflammation in the upper airways. Frischer et al. (1997) further investigated whether hydroxyl radical attacks played a role in mediating the O<sub>3</sub>-associated inflammatory response of the airways. *Ortho*- and *para*-tyrosine levels were measured in the nasal lavage samples and the *ortho/para* radical ratio was used to determine the generation of hydroxyl radicals. Significant increases in the *ortho/para* ratio were observed on days following high ambient O<sub>3</sub> levels. However, the *ortho/para* ratio was not related to polymorphonuclear leukocyte counts, suggesting that there was no detectable relationship between hydroxyl radical attacks and the inflammatory response seen in these children. Similar to the study by Kopp et al. (1999), the *ortho/para* ratio decreased at the end of the summer although O<sub>3</sub> concentrations were still high, providing additional evidence for a possible adaptive

1 response. These findings, however, do not preclude the possibility that other unmeasured  
2 effects, including cell damage or lower airway responses, may have occurred with ongoing  
3 summer season exposures. In fact, a study of joggers repeatedly exposed to O<sub>3</sub> while exercising  
4 over the summer in New York City suggested that cell damage may occur in the absence of  
5 ongoing inflammation (Kinney et al., 1996b).

6 In two Mexico City studies by Romieu et al. (1998, 2002), the effect of antioxidant  
7 supplements on the association between O<sub>3</sub> and lung function in outdoor workers and asthmatic  
8 children was investigated. Romieu and colleagues (1998) observed significant inverse  
9 associations between O<sub>3</sub> and lung function parameters, including FVC, FEV<sub>1</sub>, and FEF<sub>25-75</sub>  
10 (forced expiratory flow at 25 to 75% of FVC), among outdoor workers who were on the placebo,  
11 but not among those taking the antioxidant supplement during the first phase of testing.  
12 Likewise, O<sub>3</sub> concentrations were associated with declines in lung function among children with  
13 moderate-to-severe asthma who were on the placebo, but no associations were found among  
14 those who were taking the vitamin C and E supplement (Romieu et al., 2002). These results  
15 indicate that supplementation with antioxidants may modulate the impact of O<sub>3</sub> exposure on the  
16 small airways of two potentially at-risk populations, outdoor workers and children with  
17 moderate-to-severe asthma. In a further analysis, genetic factors were found to contribute to the  
18 variability between individuals in the effects of O<sub>3</sub> on lung function (Romieu et al., 2004).  
19 Individuals with polymorphism of the glutathione S-transferase gene (GSTM1 null genotype)  
20 lack glutathione transferase enzyme activity, which plays an important role in protecting cells  
21 against oxidative damage. Results from this analysis indicate that asthmatic children with  
22 GSTM1 null genotype were found to be more susceptible to the impact of O<sub>3</sub> exposure on small  
23 airways function. Romieu et al. (2004) noted that supplementation with the antioxidant vitamins  
24 C and E above the minimum daily requirement might compensate for the genetic susceptibility.  
25

## 26 **7.2.6 Acute Ozone Exposure and School Absences**

27 The association between school absenteeism and ambient air pollution was assessed in a  
28 few studies (Chen et al., 2000; Gilliland et al., 2001; Park et al., 2002). In the study by Chen and  
29 colleagues (2000), daily school absenteeism was examined in 27,793 students (kindergarten to  
30 sixth grade) from 57 elementary school students in Washoe County, NV over a two-year period.  
31 One major limitation of this study was that the percent of total daily absences was the outcome

1 of interest, not illness-related absences, as reasons for absences were not noted in all schools.  
2 In models adjusting for PM<sub>10</sub> and CO, ambient O<sub>3</sub> levels were associated with school  
3 absenteeism. With a distributed lag of 1 to 14 days, O<sub>3</sub> concentrations were associated with a  
4 10.41% (95% CI: 2.73, 18.09) excess rate of school absences per 40 ppb increase in 1-h max O<sub>3</sub>.  
5 PM<sub>10</sub> and CO concentrations also were associated with school absenteeism, however, the effect  
6 estimate for PM<sub>10</sub> was negative. The inverse relationship between O<sub>3</sub> and PM<sub>10</sub> may have  
7 partially attributed to the negative association observed between PM<sub>10</sub> and school absenteeism.

8 Ozone-related school absences also were examined in a study of 1,933 fourth grade  
9 students from 12 southern California communities participating in the Children's Health Study  
10 (Gilliland et al., 2001). Due to its comprehensive characterization of health outcomes, this study  
11 is valuable in assessing the effect of O<sub>3</sub> on illness-related school absenteeism in children. The  
12 study spanned a period, January through June 1996, that captured a wide range of exposures  
13 while staying mostly below the highest levels observed in the summer season. All school  
14 absences that occurred during this period were followed up with phone calls to determine  
15 whether they were illness-related. For illness-related absences, further questions assessed  
16 whether the illness was respiratory or gastrointestinal, with respiratory symptoms including  
17 runny nose/sneeze, sore throat, cough, earache, wheezing, or asthma attacks. Multiple pollutants  
18 were measured at a central site in each of the 12 communities. The statistical analysis controlled  
19 for temporal cycles, day of week, and temperature, and expressed exposure as a distributed lag  
20 out to 30 days. Associations were found between the 30-day distributed lag of 8-h avg O<sub>3</sub>  
21 (10 a.m.-6 p.m.) and all absence categories. Larger O<sub>3</sub> effects were seen for respiratory causes  
22 (147% [95% CI: 6, 478] increase in absences per 30 ppb increase in 8-h avg O<sub>3</sub>) than for  
23 nonrespiratory causes (61% [95% CI: 9, 138] increase). Among the respiratory absences, larger  
24 effects were seen for lower respiratory diseases than for upper respiratory diseases.  
25 Multipollutant analyses were not performed; however, in single-pollutant models neither PM<sub>10</sub> or  
26 NO<sub>2</sub> were associated with any respiratory or nonrespiratory illness-related absences. Some  
27 concern exists regarding the possibility of residual seasonal confounding given the six-month  
28 time span of the monitoring period and the long lag periods of exposure, which are likely to  
29 capture seasonally changing factors such as pollen episodes. Further, the biological relevance of  
30 O<sub>3</sub> concentrations lagged 30 days present an interpretive challenge.

1 Park et al. (2002) examined the association between air pollution and school absenteeism  
2 in 1,264 students, first to sixth grade, attending school in Seoul, Korea. The study period  
3 extended from March 1996 to December 1999, with 8-h avg O<sub>3</sub> concentrations ranging from  
4 3.13 ppb to 69.15 ppb (mean 22.86 ppb). Note that analysis was performed using Poisson GAM  
5 with default convergence criteria. Same day O<sub>3</sub> concentrations were positively associated with  
6 illness-related absences, but inversely associated with non-illness-related absences. PM<sub>10</sub>  
7 concentrations also were positively associated with illness-related absences. In two-pollutant  
8 models containing O<sub>3</sub> and PM<sub>10</sub>, both estimates were robust, with a slightly greater effect seen  
9 for O<sub>3</sub>.

10 Results from Chen et al. (2000), Gilliland et al. (2001), and Park et al. (2002) suggest that  
11 ambient O<sub>3</sub> concentrations, on the same day as well as accumulated over two to four weeks, may  
12 be associated with school absenteeism, particularly illness-related absences. Further replication  
13 is needed before firm conclusions can be reached regarding the effect of O<sub>3</sub> on school absences.  
14

## 15 **7.2.7 Cardiovascular Endpoints**

16 Several air pollution studies have examined various cardiovascular endpoints (Table  
17 AX7-2 in Chapter 7 Annex). The earlier studies focused on PM effects. For a more thorough  
18 discussion of these PM studies and their health endpoints, refer to the 2004 PM AQCD (Section  
19 8.3.1). More recently studies have examined associations of O<sub>3</sub> and other gaseous pollutants  
20 with various measures of heart beat rhythms in panels of elderly subjects, as discussed below.  
21 Other studies examined the increased risk of MI related to air pollutants exposures.  
22

### 23 **7.2.7.1 Cardiac Autonomic Control**

24 Alterations in heart rate and/or rhythm are thought to reflect pathophysiologic changes that  
25 may represent possible mechanisms by which ambient air pollutants such as O<sub>3</sub> may exert acute  
26 effects on human health. Decreased HRV has been identified as a predictor of increased  
27 cardiovascular morbidity and mortality. Brook et al. (2004) state that HRV, resting heart rate,  
28 and blood pressure are modulated by a balance between the two determinants of autonomic tone  
29 (the sympathetic and parasympathetic nervous systems). They note that decreased HRV predicts  
30 an increased risk of cardiovascular morbidity and mortality in the elderly and those with  
31 significant heart disease, which is generally determined by analyses of time (e.g., standard



1 deviation of normal R-R intervals) and frequency domains (e.g., low frequency/high frequency  
2 ratio by power spectral analysis, reflecting autonomic balance) measured during 24 hours of  
3 electrocardiography. Decreased parasympathetic input to the heart may provide an important  
4 mechanistic link between air pollution and cardiovascular mortality by promoting fatal  
5 tachyarrhythmias.

6 The potentially adverse effects of air pollutants on cardiac autonomic control were  
7 examined in a large population-based study, among the first in this field. Liao et al. (2004)  
8 investigated short-term associations between ambient pollutants and cardiac autonomic control  
9 from the fourth cohort examination (1996-1998) of the population-based Atherosclerosis Risk in  
10 Communities Study (ARIC). PM<sub>10</sub>, O<sub>3</sub>, and other gaseous air pollutants were examined in this  
11 study. PM<sub>10</sub> (24-h avg) and O<sub>3</sub> exposures (8-h avg, 10 a.m.-6 p.m.) one day prior to the  
12 randomly allocated examination date were used. They calculated 5-minute HRV indices  
13 between 8:30 a.m. and 12:30 p.m., and used logarithmically-transformed data on high-frequency  
14 (0.15 to 0.40 Hz) and low-frequency (0.04 to 0.15 Hz) power, standard deviation of normal R-R  
15 intervals, and mean heart rate. The effective sample sizes for O<sub>3</sub> and PM<sub>10</sub> were 5,431 and  
16 4,899, respectively, from three U.S. study centers in North Carolina, Minnesota, and Mississippi.  
17 PM<sub>10</sub> concentrations measured one day prior to the HRV measurements were inversely  
18 associated with both frequency- and time-domain HRV indices. Ambient O<sub>3</sub> concentrations  
19 were inversely associated with high-frequency power among whites. Consistently more  
20 pronounced associations were suggested between PM<sub>10</sub> and HRV among persons with a history  
21 of hypertension. Liao et al. note that these findings may represent potentially important  
22 arrhythmogenic mechanisms of ambient air pollution. The acute adverse effect of air pollution  
23 on cardiac autonomic control hypothesizes that increased air pollution levels may stimulate the  
24 autonomic nervous system and lead to an imbalance of cardiac autonomic control characterized  
25 by sympathetic activation unopposed by parasympathetic control. Such an imbalance of cardiac  
26 autonomic control may predispose susceptible people to greater risk of life-threatening  
27 arrhythmias and acute cardiac events. The findings from Liao et al. were cross-sectionally  
28 derived from a population-based sample and reflect the short-term effects of air pollution on  
29 HRV. When the regression coefficients for each individual pollutant model were compared, the  
30 effects for PM<sub>10</sub> were considerably larger than the effects for gaseous pollutants such as O<sub>3</sub>.  
31 Because of the population-based sample, this study does have better generalizability than other

1 smaller panel studies. The findings are suggestive of short-term effects of air pollutants,  
2 including O<sub>3</sub>, on HRV at the population level.

3 Another population-based study of air pollutants and HRV was conducted in Boston on  
4 497 men from the VA Normative Aging Study (NAS) (Park et al., 2005). Ozone showed several  
5 associations with HRV outcomes. Stronger associations were reported with PM<sub>2.5</sub>.

6 In two-pollutant models, the magnitude of the percent changes for both PM<sub>2.5</sub> and O<sub>3</sub> diminished  
7 slightly. In analyses by ischemic heart disease, hypertension, and diabetes status, stronger  
8 associations of HRV with O<sub>3</sub> and PM<sub>2.5</sub> were observed for individuals with ischemic heart  
9 disease and hypertension. These results are consistent with a Mexico City study (n = 34) by  
10 Holguín et al. (2003) which reported an HRV effect for O<sub>3</sub> in subjects with hypertension. The  
11 association of O<sub>3</sub> exposure with reduced low-frequency power in the full cohort seemed to be  
12 driven by subjects not taking calcium-channel blockers (Park et al., 2005). This suggests that  
13 this drug is blocking effects of O<sub>3</sub> on the sympathetic pathway. This study cohort consists of all  
14 males and almost all whites. This population-based study suggests that short-term exposures  
15 to O<sub>3</sub> are predictors of alteration in cardiac autonomic function as measured by HRV among  
16 older male adults.

17 Two related studies in Boston, MA, examined the association between air pollution and the  
18 incidence of ventricular arrhythmias (Dockery et al., 2005; Rich et al., 2005). A total of 203  
19 patients with implanted cardioverter defibrillators who lived within 25 miles of the ambient  
20 monitoring site at the Harvard School of Public Health were monitored. They had a total of  
21 635 person-years of follow-up or an average of 3.1 years per subject. In the analysis by Dockery  
22 et al. (2005), positive associations were observed between ventricular arrhythmias within three  
23 days of a prior event and a two-day mean of several air pollutants, including PM<sub>2.5</sub>, black carbon,  
24 NO<sub>2</sub>, CO, and SO<sub>2</sub>. No associations were observed with O<sub>3</sub>. There was, however, a suggestion  
25 of increasing risk with increasing quintiles of O<sub>3</sub> (p < 0.05). The analysis by Rich et al. (2005)  
26 observed stronger O<sub>3</sub> effects on ventricular arrhythmias using a case-crossover study design.  
27 Case periods were defined by the time each arrhythmic event began; for each case, three to four  
28 control periods were selected by matching on weekday and hour of the day within the same  
29 calendar month. For a 20 ppb increase in 24-hour moving average O<sub>3</sub>, a 27% (95% CI: 0, 60)  
30 increased risk of ventricular arrhythmias was estimated. Significant effects also were found  
31 for PM<sub>2.5</sub>, NO<sub>2</sub>, and SO<sub>2</sub>. In two-pollutant models, the O<sub>3</sub> effect was found to be generally

1 robust. Stratified analysis by the presence of a recent ventricular arrhythmia within the previous  
2 three days indicated that O<sub>3</sub> was associated with increased risk among subjects without a recent  
3 event (37% [95% CI: 6, 79]), but not among those with recent events (5% [95% CI: -27, 49]).  
4 Rich et al. explained that the use of the case-crossover study design and conditional analysis  
5 might have contributed to the stronger associations observed in their study compared to Dockery  
6 et al. In addition, the use of a 24-hour moving average instead of a calendar-day air pollution  
7 concentration might have reduced exposure misclassification, resulting in larger effect estimates.

8 Other studies do not provide evidence for an O<sub>3</sub> effect on HRV and cardiac arrhythmias  
9 (Peters et al., 2000a; Rich et al., 2004; Vedal et al., 2004). These studies, however, may have  
10 had limited power to examine subtle effects. Gold et al. (2000; reanalysis Gold et al., 2003)  
11 reported results that suggest that O<sub>3</sub> exposure may decrease vagal tone, leading to reduced HRV.  
12 Schwartz et al. (2005) reported a weak association of O<sub>3</sub> with the root mean squared differences  
13 between adjacent R-R intervals in a study of 28 elderly subjects and noted that lack of personal  
14 exposure measurements may render such studies less able to assess autonomic functions. This  
15 study reported the strongest effects for black carbon.

#### 16 17 **7.2.7.2 Acute Myocardial Infarction**

18 The effect of O<sub>3</sub> on the incidence of MI was examined in a limited number of studies.  
19 Acute MI was studied in relation to air pollution in Toulouse, France based on the existence of  
20 an acute MI registry (Monitoring Trend and Determinants in Cardiovascular Disease  
21 [MONICA]) and an air quality network covering the same population (Ruidavets et al., 2005).  
22 After adjustment for temperature, relative humidity, and influenza epidemics, the relative risk of  
23 acute MI occurrence was 1.76 (95% CI: 1.12, 2.45) for current day O<sub>3</sub> concentrations. The  
24 increased risk of MI was more evident in the oldest group, 55 to 64 years of age. Further, the  
25 oldest subjects without a personal history of ischemic heart disease were more susceptible to an  
26 acute event when O<sub>3</sub> levels increased. No PM data was reported in this study.

27 In a case-crossover study (n = 772) in Boston, MA, Peters et al. (2001) reported an odds  
28 ratio of 1.27 (95% CI: 0.87, 1.88) per 40 ppb increase in 2-h avg O<sub>3</sub> (1 hour before onset of  
29 event). Stronger effects on the incidence of MI were observed for PM<sub>2.5</sub> and PM<sub>10</sub>.

### 7.2.7.3 Cardiovascular Endpoints in Human Clinical Studies

In a controlled human exposure study discussed in Chapter 6, Sections 6.3.4 and 6.10, Gong et al. (1998a) studied 10 nonmedicated hypertensive and 6 healthy male adults exposed to 0.3 ppm O<sub>3</sub> with intermittent exercise in relation to various cardiovascular effects. The overall results did not indicate acute cardiovascular effects of O<sub>3</sub> in either the hypertensive or control subjects. The authors observed an increase in rate-pressure product and heart rate, a decrement for FEV<sub>1</sub>, and a >10 mm Hg increase in the alveolar/arterial pressure difference for O<sub>2</sub> following O<sub>3</sub> exposure. These findings suggest that O<sub>3</sub> can exert cardiovascular effects indirectly by impairing alveolar-arterial O<sub>2</sub> transfer and potentially reducing O<sub>2</sub> supply to the myocardium. Ozone exposure may increase myocardial work and impair pulmonary gas exchange to a degree that may be clinically important in persons with significant pre-existing cardiovascular impairment.

### 7.2.7.4 Summary of Field Studies with Cardiovascular Outcomes

A limited epidemiologic database examining cardiovascular outcomes in relation to O<sub>3</sub> exposures is available. Among these studies, three were population-based and involved cohorts such as the ARIC (Liao et al., 2004), MONICA (Ruidavets et al., 2005), and NAS (Park et al., 2005). Such studies may offer more informative results based on their large subject-pool and design. Results from these three studies were suggestive of an association between O<sub>3</sub> exposure and the cardiovascular endpoints studied. As in the case of respiratory disease outcomes, Brook et al. (2004) state that the increase in relative risk for cardiovascular disease due to air pollution is small compared with the impact of the established cardiovascular risk factors. However, because of the enormous number of people affected, even conservative risk estimates can translate into a substantial increase in mortality due to cardiovascular disease within the population. The impact of air pollution on cardiovascular disease therefore may represent a serious public health problem.

## 7.2.8 Summary of Field Studies Assessing Acute Ozone Effects

- Results from recent field/panel studies support the evidence from clinical studies that acute O<sub>3</sub> exposure is associated with a significant effect on lung function, as indicated by decrements in FEV<sub>1</sub>, FVC, and PEF. The declines in lung function were noted particularly in children and asthmatics.

- 1 • Limited evidence suggests that more time spent outdoors, higher levels of exertion,  
2 and the related increase in O<sub>3</sub> exposure may potentiate the risk of respiratory effects.  
3 In addition to children and asthmatics, adults who work or exercise outdoors may be  
4 particularly vulnerable to O<sub>3</sub>-associated health effects.  
5
- 6 • Many new studies have examined the association between O<sub>3</sub> concentrations and a  
7 wide variety of respiratory symptoms (e.g., cough, wheeze, production of phlegm,  
8 and shortness of breath). Collectively, the results suggest that acute exposure to O<sub>3</sub>  
9 is associated with increased respiratory symptoms and increased as-needed medication  
10 use in asthmatic children.  
11
- 12 • Additional panel studies investigated the effect of O<sub>3</sub> on other health outcomes,  
13 including school absences, and markers of inflammation and oxidative damage.  
14 Ozone exposure was associated with increases in respiratory-related school absences,  
15 as well as increased inflammation and generation of hydroxyl radicals in the upper  
16 airways. Use of antioxidant supplements was found to diminish the O<sub>3</sub> effect on lung  
17 function.  
18
- 19 • Some field studies have examined the association between O<sub>3</sub> and cardiac physiologic  
20 outcomes. The current evidence is rather limited but supportive of a potential effect on  
21 HRV, ventricular arrhythmias, and the incidence of MI. Additional studies need to be  
22 performed before any conclusions can be made regarding an O<sub>3</sub> effect on  
23 cardiovascular outcomes.  
24  
25

### 26 **7.3 ACUTE EFFECTS OF OZONE ON DAILY EMERGENCY** 27 **DEPARTMENT VISITS AND HOSPITAL ADMISSIONS**

#### 28 **7.3.1 Summary of Key Findings on Studies of Emergency Department Visits** 29 **and Hospital Admissions from the 1996 O<sub>3</sub> AQCD**

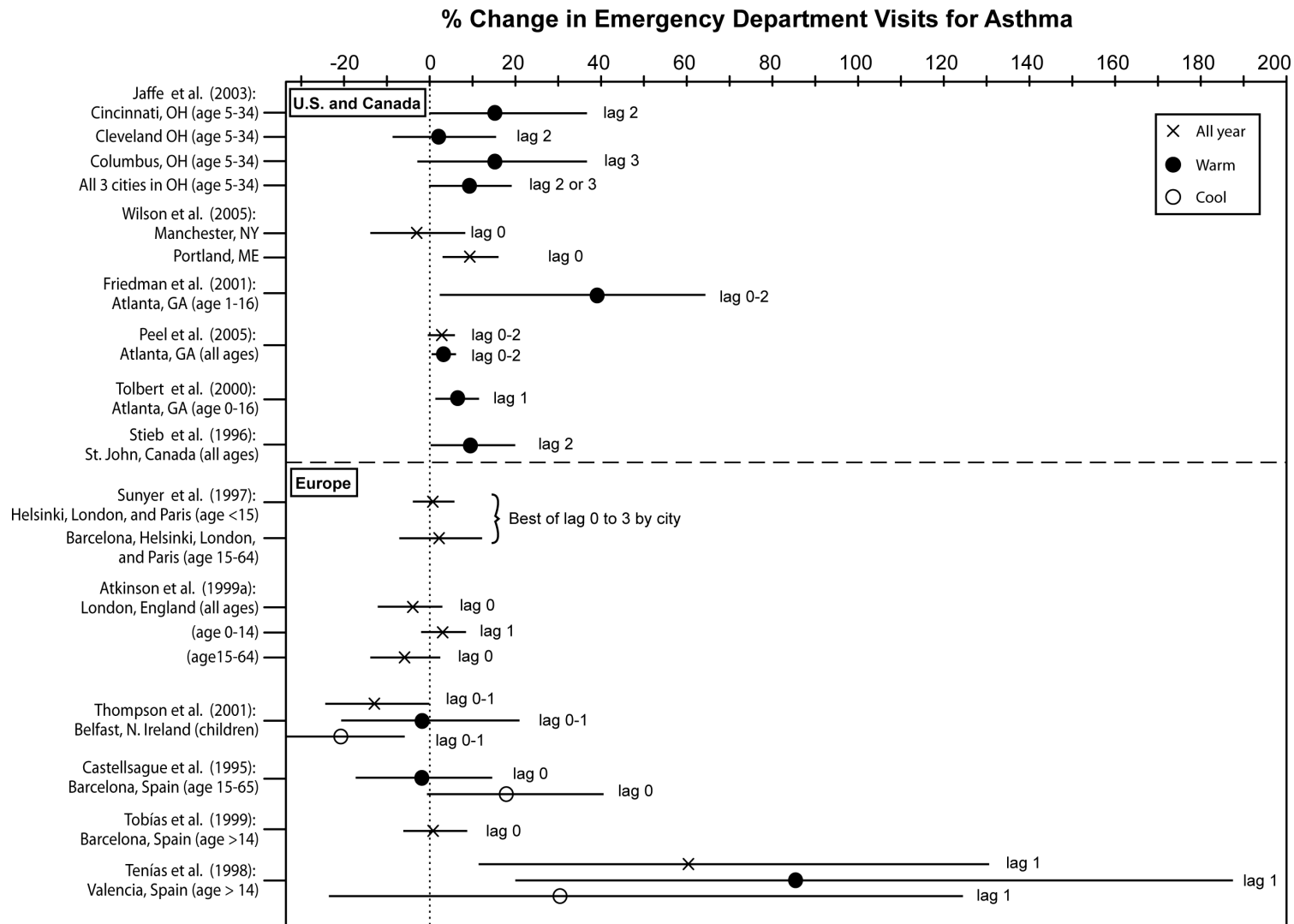
30 In the 1996 O<sub>3</sub> AQCD, aggregate population time-series studies of O<sub>3</sub>-related health effects  
31 provided relevant evidence of acute responses, even below a 1-h max O<sub>3</sub> of 0.12 ppm.  
32 Emergency room visits and hospital admissions were examined as possible outcomes following  
33 exposure to O<sub>3</sub>. In the case of emergency room visits, the evidence was limited (Bates et al.,  
34 1990; Cody et al., 1992; Weisel et al., 1995; White et al., 1994), but results generally indicated  
35 an O<sub>3</sub> effect on morbidity. The strongest and most consistent evidence of O<sub>3</sub> effects, at levels  
36 both above and below 1-h max O<sub>3</sub> levels of 0.12 ppm, was provided by the multiple studies that  
37 had been conducted on summertime daily hospital admissions for respiratory causes in various  
38 locales in eastern North America (Bates and Sizto, 1983, 1987, 1989; Burnett et al., 1994;

1 Lipfert and Hammerstrom, 1992; Thurston et al., 1992, 1994). These studies consistently  
2 demonstrated that O<sub>3</sub> air pollution was associated with increased hospital admissions, accounting  
3 for roughly one to three excess respiratory hospital admissions per million persons with each 100  
4 ppb increase in 1-h max O<sub>3</sub>. This association had been shown to remain even after statistically  
5 controlling for the possible confounding effects of temperature and copollutants (e.g., H<sup>+</sup>, SO<sub>4</sub><sup>-2</sup>,  
6 PM<sub>10</sub>), as well as when considering only days with 1-h max O<sub>3</sub> concentrations below 0.12 ppm.  
7 Overall, the aggregate population time-series studies considered in the 1996 O<sub>3</sub> AQCD provided  
8 strong evidence that ambient exposures to O<sub>3</sub> can cause significant exacerbations of preexisting  
9 respiratory disease in the general public.

### 11 **7.3.2 Review of Recent Studies of Emergency Department Visits for** 12 **Respiratory Diseases**

13 Emergency department visits represent an important acute outcome that may be affected by  
14 O<sub>3</sub> exposures. Morbidities that result in emergency department visits are closely related to, but  
15 are generally less severe than, those that result in unscheduled hospital admissions. In many  
16 cases, acute health problems are successfully treated in the emergency department; a subset of  
17 more severe cases that present initially to the emergency department may require admission to  
18 the hospital.

19 Several studies have been published in the past decade examining the temporal  
20 associations between O<sub>3</sub> exposures and emergency department visits for respiratory diseases  
21 (Table AX7-3 in Chapter 7 Annex). Total respiratory causes for emergency room visits may  
22 include asthma, pneumonia, bronchitis, emphysema, upper and lower respiratory infections such  
23 as influenza, and a few other minor categories. Asthma visits typically dominate the daily  
24 incidence counts. Chronic bronchitis and emphysema often are combined to define COPD,  
25 which is a prominent diagnosis among older adults with lung disease. Figure 7-8 presents  
26 percent changes in emergency department visits for asthma from single-pollutant models,  
27 with results expressed in standardized increments. The lags presented in the figure vary  
28 depending on reported results. Most studies reported effect estimates from a short lag period  
29 (0 to 2 days). Results from Weisel et al. (2002) are not included as comparable risks estimates  
30 for O<sub>3</sub> are not presented. Among the U.S. studies, there was one multicity study which examined  
31 three cities in Ohio (Jaffe et al., 2003). Several presented Atlanta, GA data. In general, O<sub>3</sub>



**Figure 7-8. Ozone-associated percent change (95% CI) in emergency department visits for asthma per standardized increment (see Section 7.1.3.2).**

1 effect estimates from warm season only analyses tended to be positive and larger compared to  
2 results from cool season or all year analyses.

3 Among studies with adequate controls for seasonal patterns, many reported at least one  
4 positive association with O<sub>3</sub>. These studies examined emergency department visits for total  
5 respiratory complaints (Delfino et al., 1997b, 1998b; Hernández-Gardûno et al., 1997;  
6 Ilabaca et al., 1999; Jones et al., 1995; Lin et al., 1999), asthma (Friedman et al., 2001; Jaffe  
7 et al., 2003; Stieb et al., 1996; Tenías et al., 1998; Tobias et al., 1999; Tolbert et al., 2000;  
8 Weisel et al., 2002), and COPD (Tenías et al., 2002).

9 One recent study examined emergency department visits for total and cause-specific  
10 respiratory diseases in Atlanta, GA over an 8-year period (Peel et al., 2005). A distributed lag of  
11 0 to 2 days was specified a priori. Ozone concentrations were associated with emergency  
12 department visits for total respiratory diseases and upper respiratory infections in all ages.  
13 A marginally significant association was observed with asthma visits (2.6% [95% CI: -0.5, 5.9]  
14 excess risk per 30 ppb increase in 8-h max O<sub>3</sub>), which became stronger when analysis was  
15 restricted to the warm months (3.1% [95% CI: 0.2, 6.2] excess risk). In multipollutant models  
16 adjusting for PM<sub>10</sub>, NO<sub>2</sub> and CO, O<sub>3</sub> was the only pollutant that remained significantly associated  
17 with upper respiratory infections. Another large asthma emergency department study was  
18 carried out during the months of May through September from 1984 to 1992 in St. John, New  
19 Brunswick, Canada (Stieb et al., 1996). Effects were examined separately among children aged  
20 less than 15 years and in persons aged 15 years and older. A significant effect of O<sub>3</sub> on  
21 emergency department visits was reported among persons 15 years and older. There was  
22 suggestion of a threshold somewhere in the range below a 1-h max O<sub>3</sub> of 75 ppb. A study in  
23 Valencia, Spain from 1994 to 1995 observed that emergency room visits for asthma among  
24 persons over 14 years old were robustly associated with relatively low O<sub>3</sub> levels (median 1-h  
25 max O<sub>3</sub> of 62.8 µg/m<sup>3</sup> or approximately 32.4 ppb) (Tenías et al., 1998). The excess risk of  
26 asthma emergency room visits was larger in the warm season (May to October), 85% (95%  
27 CI: 20, 188) excess risk per 40 ppb increase in 1-h max O<sub>3</sub>, compared to the cool season  
28 (November-April), 31% (95% CI: -24, 125) excess risk (Tenías et al., 1998).

29 Among the studies that observed a positive association between O<sub>3</sub> and emergency  
30 department visits for respiratory outcomes, O<sub>3</sub> effects were found to be robust to adjustment for  
31 PM<sub>10</sub>, NO<sub>2</sub>, SO<sub>2</sub>, and BS (Lin et al., 1999; Peel et al., 2005; Tenías et al., 1998). One study by



1 Tolbert and colleagues (2000) observed that the significant univariate effects of both O<sub>3</sub> and  
2 PM<sub>10</sub> on pediatric asthma emergency department visits in Atlanta, GA became nonsignificant in  
3 two-pollutant regressions, reflecting the high correlation between the two pollutants (r = 0.75).

4 For several other studies with total respiratory and asthma outcomes, inconsistencies  
5 confound an interpretation of likely causal effects. For example, in a Montreal, Canada study,  
6 O<sub>3</sub> effects on total respiratory emergency department visits were seen in a short data series from  
7 the summer of 1993 but not in a similar data series from the summer of 1992 (Delfino et al.,  
8 1997b). The significant 1993 results were seen only for persons older than 64 years. A very  
9 similar analysis of two additional summers (1989 and 1990) revealed an O<sub>3</sub> association only for  
10 1989 and again only in persons over 64 years old (Delfino et al., 1998b). An analysis of data on  
11 respiratory emergency department visits from June to August of 1990 in Baton Rouge, LA  
12 reported O<sub>3</sub> effects in adults, but not in children or among the elderly (Jones et al., 1995).

13 Tobías and colleagues (1999) showed that regression results for asthma emergency  
14 department visits could be quite sensitive to methods used to control for asthma epidemics.  
15 Ozone was associated with the outcome variable in only one of eight models tested. An Atlanta,  
16 GA study by Zhu et al. (2003) examined asthma emergency department visits in children during  
17 three summers using Bayesian hierarchical modeling to address model variability. Data were  
18 analyzed at the zip code level to account for spatially misaligned longitudinal data. Results  
19 indicated a positive, but nonsignificant relationship between O<sub>3</sub> and emergency room visits  
20 for asthma.

21 Other studies also reported no association between O<sub>3</sub> and emergency department visits for  
22 respiratory causes (Atkinson et al., 1999a; Castellsague et al., 1995; Chew et al., 1999; Hwang  
23 and Chan, 2002; Sunyer et al., 1997). Using Bayesian hierarchical modeling, Hwang and Chan  
24 (2002) examined the effect of air pollutants on daily clinic visits for lower respiratory illnesses  
25 across 50 cities in Taiwan. All pollutants except O<sub>3</sub> were associated with daily clinic visits. In a  
26 pooled analysis of emergency admissions for asthma in four European cities as part of the Air  
27 Pollution on Health: European Approach (APHEA) study, there was no overall effect of O<sub>3</sub>  
28 observed (Sunyer et al., 1997). Atkinson et al. (1999a) in London, England also did not find an  
29 association between O<sub>3</sub> and emergency department visits at a mean 8-h max O<sub>3</sub> concentration of  
30 17.5 ppb. One study by Thompson and colleagues (2001) in Belfast, Northern Ireland observed  
31 a decreased risk of childhood asthma admissions (-21% [95% CI: -33, -6] per 20 ppb increase

1 in 24-h avg O<sub>3</sub>) in the cold season (November-April). After adjusting for benzene levels, O<sub>3</sub> was  
2 no longer associated with asthma emergency department visits. The inverse relationship of O<sub>3</sub>  
3 with benzene concentrations ( $r = -0.65$ ), and perhaps with other pollutants, might have produced  
4 the apparent protective effect of O<sub>3</sub>. No significant O<sub>3</sub> effect was found in the warm season  
5 (May-October). The O<sub>3</sub> levels were low in both seasons, with a mean 24-h avg O<sub>3</sub> concentration  
6 of 18.7 ppb in the warm season and 17.1 ppb in the cold season. A study by Hajat et al. (1999,  
7 2002) of physician consultations for asthma, lower respiratory diseases, and upper respiratory  
8 diseases in London reported negative associations with O<sub>3</sub>, which was suggestive of residual  
9 confounding by copollutants or weather factors (note that data were analyzed using Poisson  
10 GAM with default convergence criteria). Several other emergency department studies looking at  
11 O<sub>3</sub> are more difficult to interpret due to inadequate control for seasonal patterns, very low O<sub>3</sub>  
12 levels, or because no quantitative results were shown for O<sub>3</sub> (Buchdahl et al., 1996, 2000; Garty  
13 et al., 1998; Holmén et al., 1997; Lierl and Hornung, 2003; Lipsett et al., 1997; Nutman et al.,  
14 1998).

15 Although several studies found a significant association between O<sub>3</sub> concentrations and  
16 emergency department visits for respiratory causes, some inconsistencies were observed. The  
17 inconsistencies may be attributable, at least partially, to differences in model specifications and  
18 analysis approach among the various studies. For example, ambient O<sub>3</sub> concentrations, length of  
19 the study period, and statistical methods used to control confounding by seasonal patterns and  
20 copollutants appear to affect the observed O<sub>3</sub> effect on emergency department visits. The body  
21 of evidence remains inconclusive regarding effects of O<sub>3</sub> on the risk of emergency department  
22 visits.

### 23 24 **7.3.3 Studies of Hospital Admissions for Respiratory Diseases**

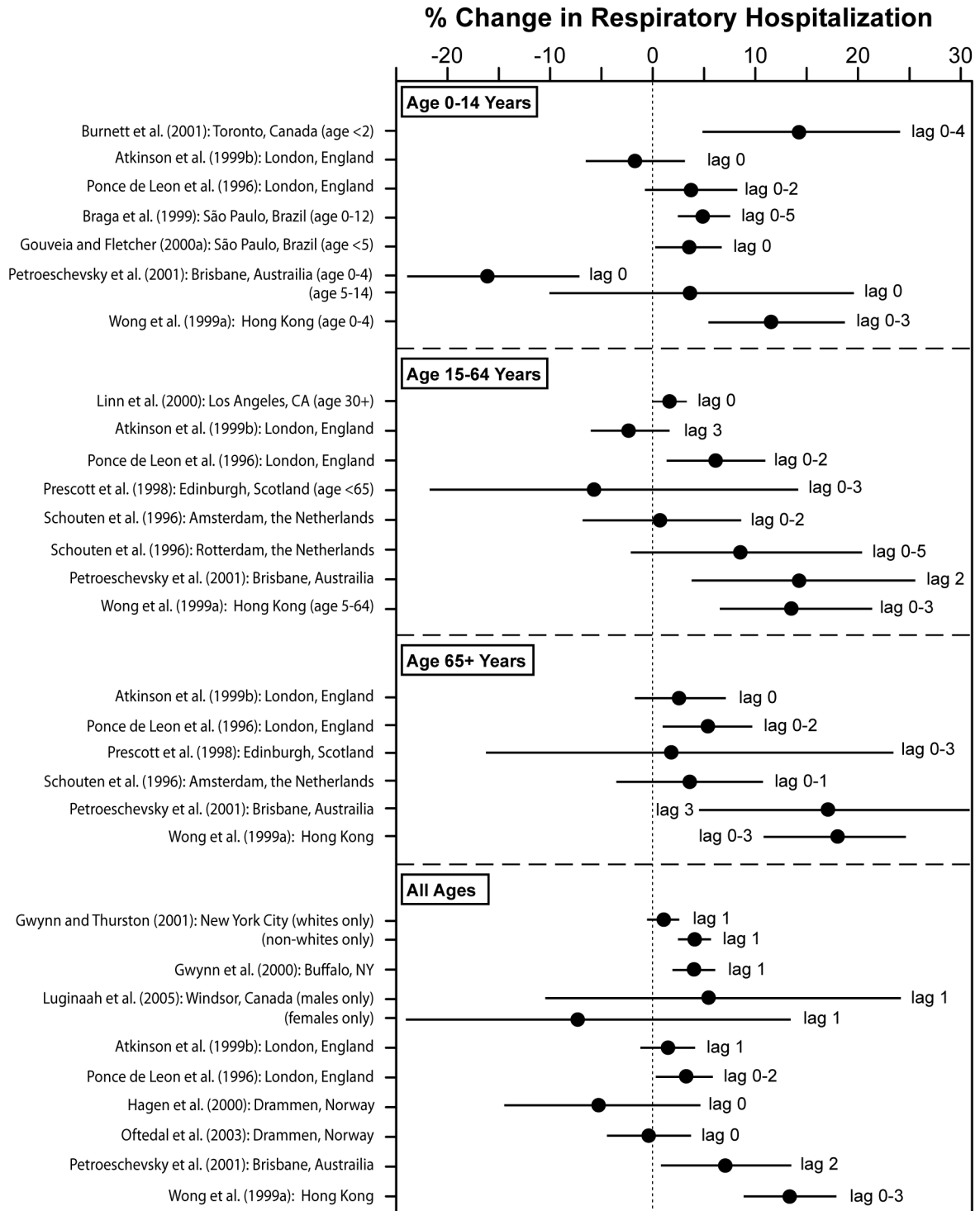
25 Hospital admissions represent a medical response to a serious degree of morbidity for a  
26 particular disease. Scheduled hospitalizations are planned in advance when a particular clinical  
27 treatment is needed. However, unscheduled admissions are ones that occur in response to  
28 unanticipated disease exacerbations and are more likely to be affected by environmental factors,  
29 such as air pollution. As such, the hospital admissions studies reviewed here focused  
30 specifically on unscheduled admissions. Study details and results from hospital admissions  
31 studies published over the past decade are summarized in Table AX7-4 (in the Chapter 7

1 Annex). As a group, these hospitalization studies tend to be larger in terms of geographic and  
2 temporal coverage, and indicate results that are generally more consistent than those reviewed  
3 above for emergency department visits. As in the case for all studies that examine changes in  
4 aggregate measures of acute disease outcomes over time, the following should be considered in  
5 comparing results: (1) difference in types of respiratory diseases for hospital admission; (2) age  
6 of study population; (3) mean level of O<sub>3</sub> during study; (4) single-city versus multicity studies;  
7 (5) length of study (e.g., <5 years versus >5 years); (6) analysis by season versus all year;  
8 (7) O<sub>3</sub>-only versus multipollutant models; (8) number of exposure lag days; and (9) type of study  
9 (e.g., case-crossover versus time-series). These factors are considered in the sections below with  
10 further discussion on potential confounding of the O<sub>3</sub> effect estimate by seasonal factors and  
11 copollutants.

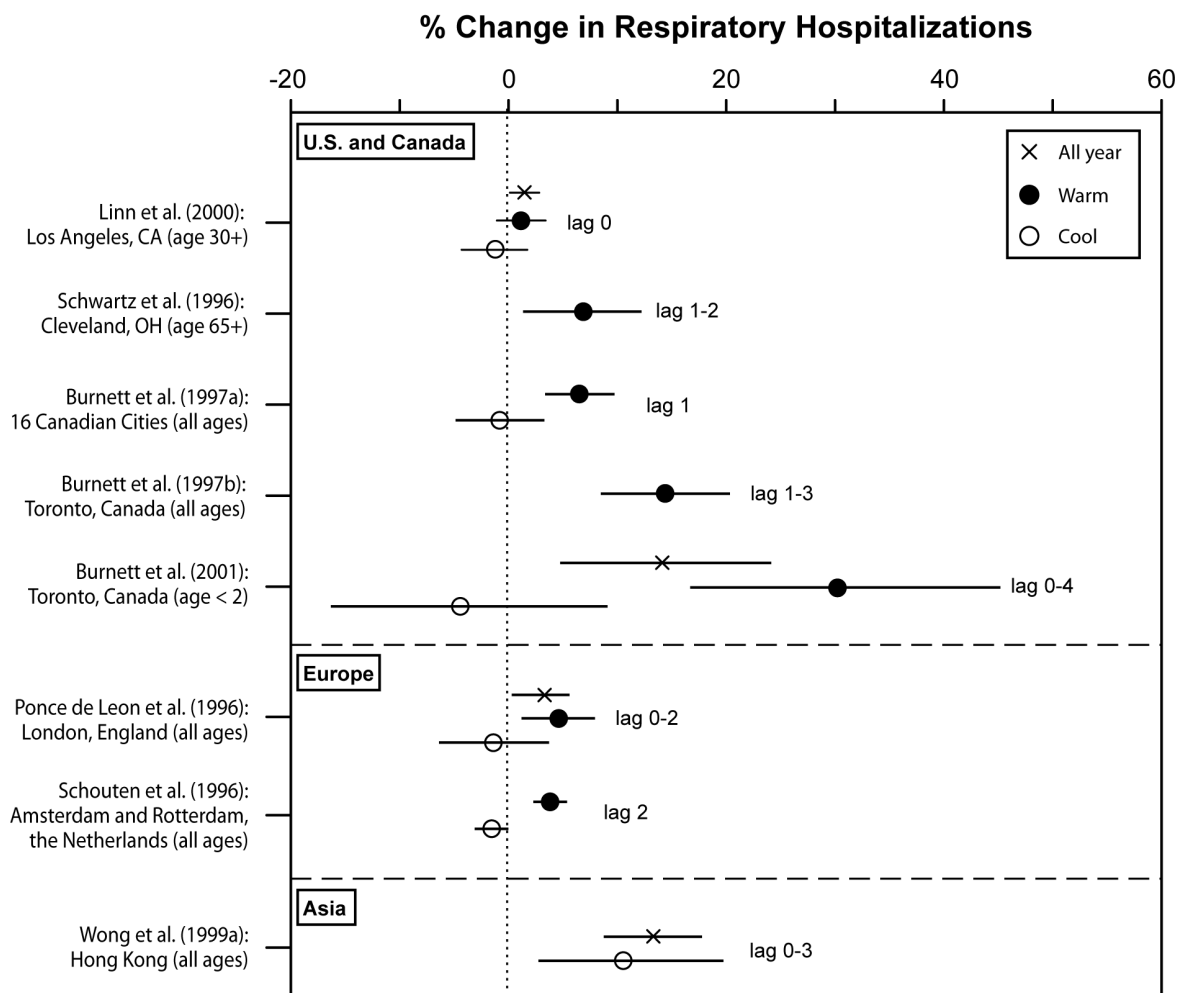
### 13 **7.3.3.1 All Year and Seasonal Effects of Ozone on Respiratory Hospitalizations**

14 The effect of O<sub>3</sub> on respiratory hospitalizations was examined in various studies conducted  
15 in the U.S. and abroad. Figures 7-9 and 7-10 present risk estimates from all total respiratory  
16 hospital admission studies. Burnett et al. (1995), which did not present quantitative results for  
17 O<sub>3</sub>, and Yang et al. (2003), which only presented odd ratios, were not included in the figures.  
18 In cases where multiple lags were presented, the multiday lag was selected to represent the  
19 cumulative effect from all days examined. If only single-day lags were analyzed, the effect  
20 estimate of the shortest lag time, usually a lag of 0 or 1 day, was presented. Figure 7-9 plots the  
21 effect estimates and 95% CIs from 15 studies that analyzed all year data. The risk estimates are  
22 arranged by age groups. The preponderance of positive risk estimates, with some that are  
23 statistically significant, is readily apparent. Figure 7-10 presents the season-stratified effect  
24 estimates by region. For studies that reported risk estimates from all four seasons, only the  
25 summer and winter estimates are presented. It appears that the warm season estimates,  
26 collectively, tend to be larger, positive values compared to all year and cool season estimates.  
27 All of the negative estimates were from analyses using cool season data only, which might  
28 reflect the inverse correlation between O<sub>3</sub> and copollutants, namely PM, during that season.  
29 These studies are discussed below in further detail.

30 Among the respiratory hospitalization studies, the most robust and informative results were  
31 observed when a broad geographic area was examined using a consistent analytical methodology



**Figure 7-9. Ozone-associated percent change (95% CI) in total respiratory hospitalizations for all year analyses per standardized increment (see Section 7.1.3.2). Effect estimates are arranged by age groups.**



**Figure 7-10. Ozone-associated percent change (95% CI) in total respiratory hospitalizations by season per standardized increment (see Section 7.1.3.2).**

1 (Anderson et al., 1997; Burnett et al., 1995, 1997a). These studies have all reported an O<sub>3</sub> effect  
 2 on respiratory hospital admissions. The largest such study to-date was carried out using data on  
 3 all-age respiratory hospital admissions from 16 Canadian cities with populations exceeding  
 4 100,000 during the period 1981 to 1991 (Burnett et al., 1997a). In addition to O<sub>3</sub>, the authors  
 5 evaluated health effects of SO<sub>2</sub>, NO<sub>2</sub>, CO, and coefficient of haze (a surrogate for black carbon  
 6 particle concentrations). Pooling the 16 cities, a positive association was observed between  
 7 respiratory hospital admissions and the 1-day lag O<sub>3</sub> concentration in the spring (5.6% [95%CI:

1 1.6, 9.9] excess risk per 40 ppb increase in 1-h max O<sub>3</sub>) and summer (6.7% [95%CI: 3.5, 10.0]).  
2 The results for fall were also positive, though of smaller magnitude (3.8% [95%CI: -0.2, 7.9]).  
3 There was no evidence for an O<sub>3</sub> effect in the winter season (-0.8% [95%CI: -4.8, 3.3]).

4 Control outcomes related to blood, nervous system, digestive system, and genitourinary system  
5 disorders were not associated with O<sub>3</sub>. In a previous study focused mainly on evaluating health  
6 impacts of sulfate particles, Burnett and colleagues (1995) reported results from a time-series  
7 analysis of all-age respiratory hospital admissions to 168 hospitals in Ontario, Canada over a  
8 6-year period (1983 to 1988). The outcome data were prefiltered to remove seasonal variations  
9 using a weighted 19-day moving average. The authors reported that O<sub>3</sub> was associated with  
10 respiratory hospital admissions; however, no quantitative results for O<sub>3</sub> were presented.

11 Results from an analysis of five European cities indicated strong and consistent O<sub>3</sub> effects  
12 on unscheduled hospital admissions for COPD (Anderson et al., 1997). The five cities  
13 examined — London, Paris, Amsterdam, Rotterdam, and Barcelona — were among those  
14 included in the multicity APHEA study. The number of years of available data varied from 5 to  
15 13 years among the cities. City-specific effect estimates were pooled across cities using  
16 weighted means. An association with O<sub>3</sub> was observed in full year analyses. Season-stratified  
17 analyses indicated that the O<sub>3</sub> effect was larger in the warm season (April-September), 4.7%  
18 (95% CI: 1.6, 7.9) excess risk per 40 ppb increase in 1-h max O<sub>3</sub>, compared to the cool season  
19 (October-March), 1.6% (95% CI: -3.1, 7.9) excess risk. There was no significant heterogeneity  
20 in O<sub>3</sub> effects among the cities.

21 Several additional studies carried out in one or two cities over a span of five or more years  
22 provided substantial additional evidence regarding O<sub>3</sub> effects on respiratory hospital admissions  
23 (Anderson et al., 1998; Burnett et al., 1999, 2001; Moolgavkar et al., 1997; Petroeschevsky et al.,  
24 2001; Ponce de Leon et al., 1996; Sheppard et al., 1999 [reanalysis Sheppard, 2003]; Yang et al.,  
25 2003). Moolgavkar and colleagues (1997) reported significant and robust O<sub>3</sub> effects on  
26 respiratory hospital admissions in adults 65 years and older in Minneapolis and St. Paul, MN,  
27 but not in Birmingham, AL. The absence of effects in the southern city may reflect less  
28 penetration of O<sub>3</sub> into the indoor environment due to greater use of air conditioning, and thus  
29 less correlation between central site O<sub>3</sub> monitoring and actual exposures of the urban populace.  
30 Ozone effects on all-age and age-stratified asthma and total respiratory hospital admissions were  
31 observed in Brisbane, Australia (Petroeschevsky et al., 2001). Effect sizes appeared consistent

1 in the warm and cool seasons (data not provided). Petroeschevsky et al. commented that the  
2 year-round effect of O<sub>3</sub> might reflect the relatively small degree of seasonal variation in O<sub>3</sub> levels  
3 observed in Brisbane. Although O<sub>3</sub> levels were quite low year-round, they did not notably  
4 decline during the winter period. The authors also noted that given the subtropical climate in  
5 Brisbane, characterized by warm, dry winters, perhaps the proportion of the population exposed  
6 to winter O<sub>3</sub> concentrations was higher than in cities where inclement winter weather might force  
7 populations indoors.

8 Another set of studies have examined associations between O<sub>3</sub> and respiratory  
9 hospitalizations in single cities over shorter (<5 years) time spans. Positive and significant O<sub>3</sub>  
10 effects were reported in Cleveland, OH (Schwartz et al., 1996); New York City (Gwynn and  
11 Thurston, 2001); Northern New Jersey (Weisel et al., 2002); Toronto, Canada (Burnett et al.,  
12 1997b); Helsinki, Finland (Pönkä and Virtanen, 1996); São Paulo, Brazil (Braga et al., 1999;  
13 Gouveia and Fletcher, 2000a); and Hong Kong (Wong et al., 1999a). The Helsinki study  
14 reported significant effects of O<sub>3</sub> on both asthma and on digestive disorders in a setting of very  
15 low O<sub>3</sub> concentrations (Pönkä and Virtanen, 1996), which raises questions of plausibility.

16 Less consistent effects of O<sub>3</sub> were seen in other respiratory hospitalization studies  
17 (Schouten et al., 1996; Lin et al., 2003; Lin et al., 2004; Morgan et al., 1998a; Oftedal et al.,  
18 2003). In a study conducted in Amsterdam and Rotterdam, the Netherlands, associations  
19 between O<sub>3</sub> and respiratory admissions were observed; however, results were difficult to  
20 interpret due to the large number of statistical tests performed (Schouten et al., 1996). In a  
21 California study by Neidell (2004), a negative association was observed between hospitalizations  
22 for asthma and naturally occurring seasonal variations in O<sub>3</sub> within zip codes in children aged  
23 0 to 18 years. However, the O<sub>3</sub> effect was found to be influenced by socioeconomic status.  
24 Among children of low socioeconomic status, O<sub>3</sub> generally was associated with increased  
25 hospitalizations, with statistical significance reached in certain age groups. Neidell further stated  
26 that avoidance behavior on high O<sub>3</sub> days may have attributed to the negative relationship  
27 observed in children of higher socioeconomic status.

28 No associations between respiratory hospital admissions and O<sub>3</sub> were seen in studies from  
29 Los Angeles, CA (Linn et al., 2000; Mann et al., 2002; Nauenberg and Basu, 1999); Vancouver,  
30 Canada (Lin et al., 2004); London, England (Atkinson et al., 1999b); Edinburgh, Scotland  
31 (Prescott et al., 1998); and Drammen, Norway (Hagen et al., 2000). Several of these studies

1 were carried out in locations with low O<sub>3</sub> levels, suggestive of a nonlinear concentration-  
2 response relationship (Lin et al., 2004; Prescott et al., 1998). The nonsignificant findings in the  
3 South Coast air basin, CA area are surprising given the elevated O<sub>3</sub> concentrations observed  
4 there (Mann et al., 2002). Inadequate control of seasonal confounding may underlie some of the  
5 nonsignificant and negative findings. An additional factor likely contributing to the variability  
6 of results is the relatively small sample sizes included in some of these studies.

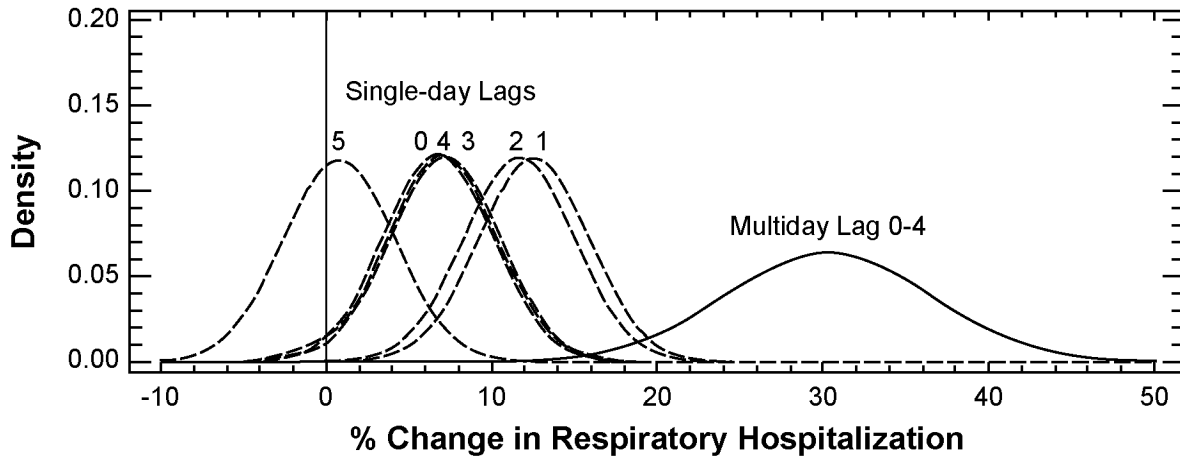
7 For respiratory hospitalization outcomes, the largest, most significant associations with O<sub>3</sub>  
8 concentrations were observed when using short lag periods, in particular a 0-day lag (exposure  
9 on same day) and a 1-day lag (exposure on previous day). In a study of 16 Canadian cities by  
10 Burnett et al. (1997a), the strongest association between O<sub>3</sub> and respiratory hospitalizations was  
11 found at a 1-day lag. A decline in the magnitude and significance of the effect was seen with  
12 increasing days lagged for O<sub>3</sub>. Anderson et al. (1997) investigated the association between O<sub>3</sub>  
13 and daily hospital admissions for COPD in five European cities. Lags up to 5 days were  
14 examined, and the largest risk estimates were found using 0- and 1-day lags. These results  
15 suggest that O<sub>3</sub> has a short-term effect on respiratory hospitalizations.

16 Burnett et al. (2001) investigated the association between respiratory hospitalizations and  
17 O<sub>3</sub> in children less than 2 years of age. Lags up to five days were examined after stratifying by  
18 season (Figure 7-11). In the summer season, significant associations between O<sub>3</sub> and daily  
19 admissions were found in several of the lags, with the largest risk estimate of 12.5% (95% CI:  
20 5.7, 19.7) excess risk per 40 ppb increase in 1-h max O<sub>3</sub> at a 1-day lag. In comparison, the  
21 O<sub>3</sub>-related risk estimate was 30.2% (95% CI: 18.0, 42.4) using a cumulative lag period of  
22 5 days. The large effect estimate for the cumulative lag period indicated that O<sub>3</sub> exposure likely  
23 had an immediate effect that persisted over several days.

24 Weisel et al. (2002) stated that a lag period of 1 to 3 days between exposure to O<sub>3</sub> and  
25 hospital admissions or emergency department visits for asthma was plausible because it might  
26 take time for the disease to progress to the most serious responses following exposure.  
27 In addition, taking medication could delay further the progression of the adverse effect. Thus,  
28 although strongest associations are found at lags of 0 and 1 days, examining longer single-day  
29 lag periods or multiday lag periods may further enhance understanding of the effect of O<sub>3</sub> on  
30 hospitalizations.

31





**Figure 7-11. Comparison of single-day lags (0-, 1-, 2-, 3-, 4-, and 5-day) to a cumulative multiday lag (0- to 4-day) for percent changes in total respiratory hospitalizations per 40 ppb increase in 1-h max O<sub>3</sub> in children less than two years of age.**

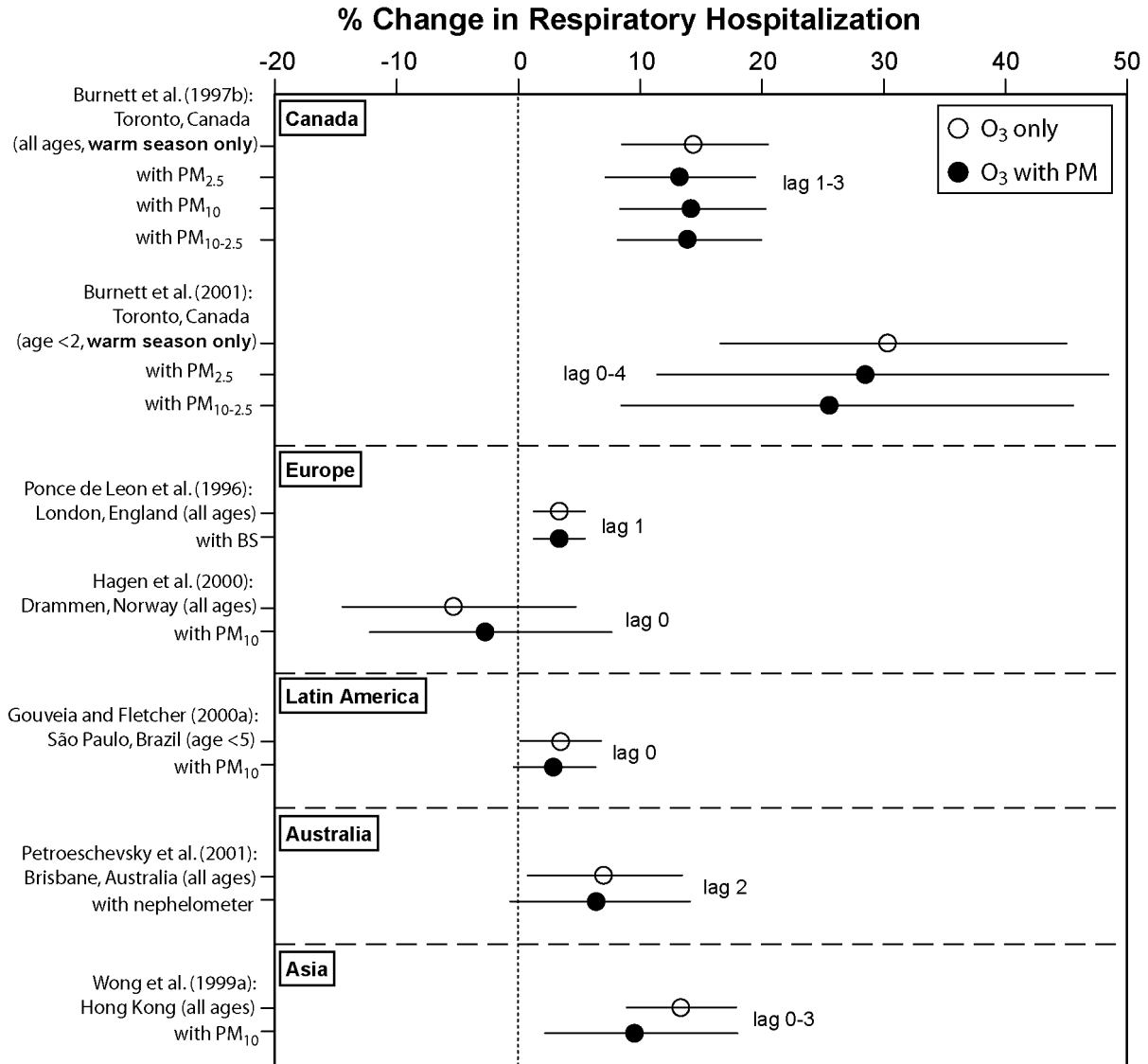
Source: Derived from Burnett et al. (2001).

1 In conclusion, while some inconsistencies are noted across studies, the evidence supports  
 2 the findings of significant and robust effects of O<sub>3</sub> on various respiratory disease hospitalization  
 3 outcomes. Large multicity studies, as well as many studies from individual cities have reported  
 4 significant O<sub>3</sub> associations with total respiratory hospitalizations, asthma, and COPD, especially  
 5 in studies analyzing the O<sub>3</sub> effect during the summer or warm season.

### 7.3.3.1 Potential Confounding of the Ozone Effect on Respiratory Hospitalizations by Copollutants

9 As in the case for most air pollution studies, potential confounding of the association  
 10 between O<sub>3</sub> and respiratory hospitalizations by copollutants generally was examined using  
 11 multipollutant regression models. Figure 7-12 compares the risk estimates from models with  
 12 and without adjustment for PM indices. This figure indicates that O<sub>3</sub> risk estimates are fairly  
 13 robust to PM adjustment in all year and warm season only data. None of the studies examined  
 14 PM-adjusted O<sub>3</sub> risk estimates in cool season only data.

15 Several analyses of a large data set from Toronto, Canada spanning the years 1980 to 1994  
 16 reported O<sub>3</sub> effects on respiratory hospitalizations for all ages (Burnett et al., 1997b, 1999) and



**Figure 7-12. Ozone-associated percent change (95% CI) in total respiratory hospitalizations with adjustment for PM indices per standardized increment (see Section 7.1.3.2). Analyses performed using all year data unless noted otherwise.**

1 for persons under the age of 2 years (Burnett et al., 2001). In the 1999 and 2001 studies,  
 2 analyses were performed using Poisson GAM (default convergence criteria) with a  
 3 nonparametric LOESS prefilter applied to the pollution and hospitalization data. All studies  
 4 demonstrated that O<sub>3</sub> effects were robust when adjusting for PM indices, whereas PM effects

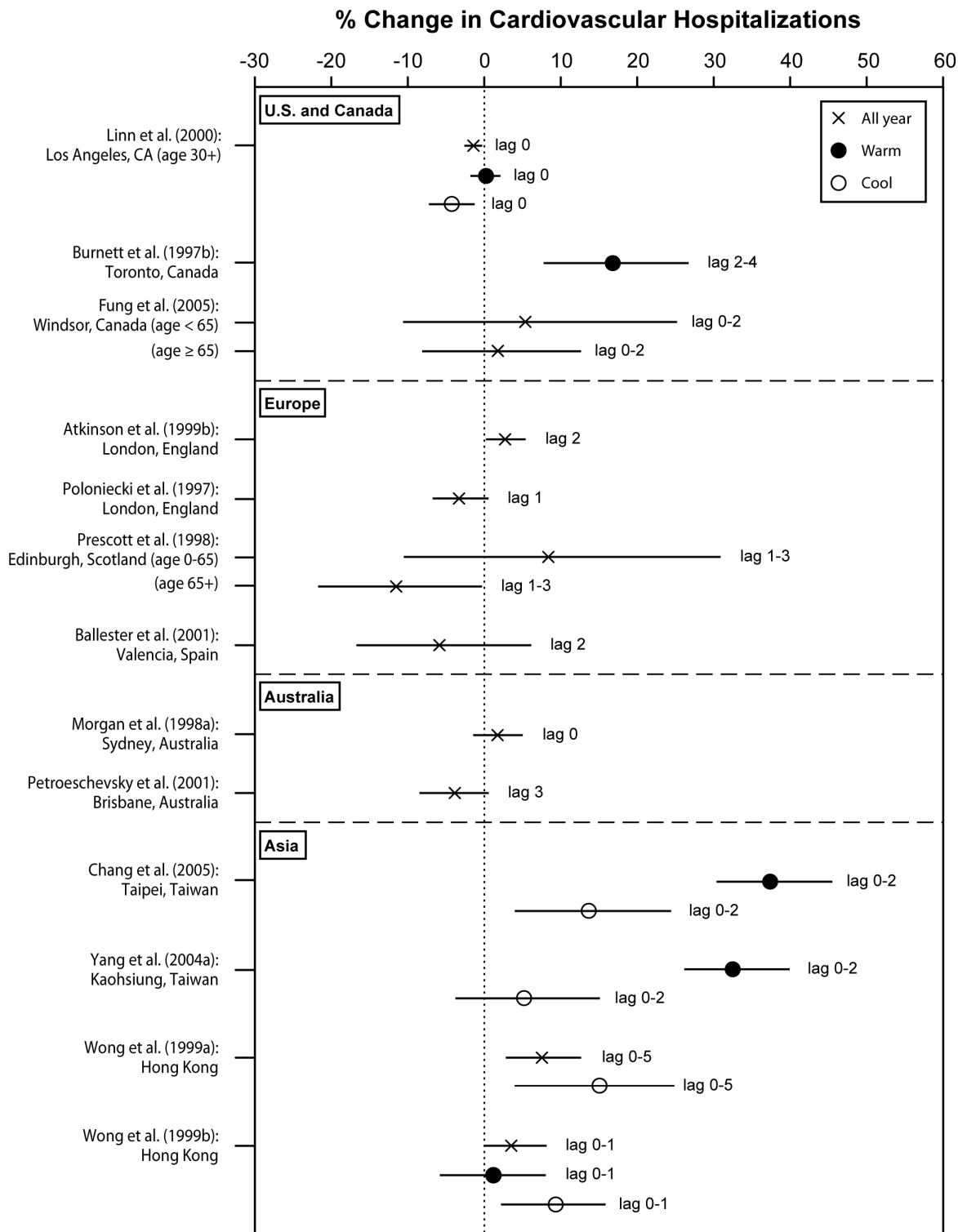
1 from single-pollutant models were markedly attenuated when O<sub>3</sub> was added to the regression.  
2 These results imply more robust associations with respiratory hospitalizations for O<sub>3</sub> than PM.

3 Results from the APHEA study indicated strong and consistent O<sub>3</sub> effects on unscheduled  
4 hospital admissions for COPD (Anderson et al., 1997). Significant effects also were seen for  
5 BS, TSP, and NO<sub>2</sub>. The authors reported that among all pollutants examined, the most consistent  
6 and significant findings were for O<sub>3</sub>. No two-pollutant model results were reported. Several  
7 additional studies also observed that there was no substantial difference in the O<sub>3</sub> effect after  
8 adjusting for PM in the regression model (Gouveia and Fletcher, 2000a; Petroeschovsky et al.,  
9 2001; Ponce de Leon et al., 1996).

10 Collectively, these results suggest that copollutants generally do not confound the  
11 association between O<sub>3</sub> and respiratory hospitalizations. Ozone risk estimates were robust to PM  
12 adjustment in all year and warm season only data.

#### 14 **7.3.4 Association of Ozone with Hospital Admissions for** 15 **Cardiovascular Disease**

16 A subset of hospital admissions studies have examined the association of O<sub>3</sub> with  
17 cardiovascular outcomes (see Figure 7-13). Several have found negative or inconsistent  
18 associations (Ballester et al., 2001; Burnett et al., 1999; Fung et al., 2005; Koken et al., 2003;  
19 Linn et al., 2000; Mann et al., 2002; Morgan et al., 1998a; Petroeschovsky et al., 2001;  
20 Poloniecki et al., 1997; Prescott et al., 1998). Other studies, especially those that examined  
21 the relationship when O<sub>3</sub> exposures were higher, have observed robust positive associations  
22 between O<sub>3</sub> and cardiovascular hospitalizations (Atkinson et al., 1999b; Burnett et al., 1997b;  
23 Chang et al., 2005; Tsai et al., 2003a; Wong et al., 1999a,b; Yang et al., 2004a). In Toronto,  
24 Canada, Burnett et al. (1997b) reported a positive association between O<sub>3</sub> and cardiovascular  
25 hospital admissions in a summer-only analysis. The results were robust to adjustment for  
26 various PM indices, while the PM effects diminished when adjusting for gaseous pollutants.  
27 Other studies stratified their analysis by temperature, warm days ( $\geq 20$  °C) versus cool days  
28 ( $< 20$  °C). The analysis using warm days consistently produced positive associations (Chang  
29 et al., 2005; Tsai et al., 2003a; Yang et al., 2004a). In two studies conducted in Hong Kong,  
30 total cardiovascular as well as circulatory, ischemic heart disease, and heart failure were all  
31 significantly associated with O<sub>3</sub> in the cool but not the warm season (Wong et al., 1999a,b).



**Figure 7-13. Ozone-associated percent change (95% CI) in total cardiovascular hospitalizations per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.**

1 In Wong et al. (1999b), O<sub>3</sub> concentrations were similar in both seasons, with warm season levels  
2 slightly lower, mean 8-h avg O<sub>3</sub> concentrations of 31.2 µg/m<sup>3</sup> (or 16.1 ppb), compared to the  
3 cool season, mean 34.8 µg/m<sup>3</sup> (or 18.0 ppb). The authors speculated that differing activity  
4 patterns and home ventilation factors may have contributed to the seasonal differences in O<sub>3</sub>  
5 effects. Weather in Hong Kong is mild throughout the year, but less humid and cloudy in the  
6 cool season. Thus, during the cool season people are more likely to open windows or stay  
7 outdoors, resulting in higher personal exposures even with similar ambient concentrations.

8 Among the growing group of hospitalization studies that examined the effect of O<sub>3</sub> on  
9 cardiovascular admissions, several have found inconsistent associations, especially for all year  
10 analyses. However, in studies that stratified analyses by seasonal or meteorological factors,  
11 evidence is suggestive of an association between O<sub>3</sub> and cardiovascular hospitalizations.

### 13 **7.3.5 Summary of Acute Ozone Effects on Daily Emergency Department** 14 **Visits and Hospital Admissions**

- 15 • The vast majority of emergency room visits and hospitalization studies conducted  
16 over the past decade have looked at effects of O<sub>3</sub> on either total respiratory diseases  
17 and/or asthma. Among the hospitalization studies, O<sub>3</sub> was found to be associated with  
18 both outcomes in many cases. Studies of emergency department visits for respiratory  
19 conditions also reported O<sub>3</sub> effects, but the results tended to be less consistent across  
20 studies.
- 21  
22 • Many of the daily emergency department visits and hospitalization studies analyzed  
23 O<sub>3</sub> risk estimates using year-round data. Given the strong seasonal variations in  
24 O<sub>3</sub> concentrations and the changing relationship between O<sub>3</sub> and other copollutants  
25 by seasons, inadequate adjustment for seasonal effects might have masked or  
26 underestimated the association between O<sub>3</sub> and the respiratory disease outcomes.  
27 Season-stratified analyses typically yielded more reliable O<sub>3</sub> effect estimates.
- 28  
29 • Several studies have examined the association between O<sub>3</sub> and respiratory  
30 hospitalizations while controlling for other pollutants in the analytical model.  
31 In most cases, O<sub>3</sub> effects have been reported to be robust to adjustment for copollutants,  
32 particularly PM. Therefore, the evidence is supportive of independent O<sub>3</sub> effects on  
33 respiratory hospital admissions.
- 34  
35 • A subset of hospital admission studies examined the effect of O<sub>3</sub> on cardiovascular  
36 outcomes. The evidence is inconclusive regarding the association between O<sub>3</sub> exposure  
37 and cardiovascular hospitalizations in year-round data analyses. However, in the limited  
38 number of studies that accounted for seasonal or meteorological factors, results suggested

1 that O<sub>3</sub> was associated with increased risk of cardiovascular hospital admissions during  
2 the warm season.  
3  
4

## 5 **7.4 ACUTE EFFECTS OF OZONE ON MORTALITY**

### 6 **7.4.1 Summary of Key Findings on Acute Effects of Ozone on Mortality** 7 **From the 1996 O<sub>3</sub> AQCD**

8 A limited number of studies examined O<sub>3</sub>-mortality associations at the time of the previous  
9 O<sub>3</sub> AQCD, most of which were from the 1950s and 1960s. The 1996 O<sub>3</sub> AQCD considered these  
10 historical studies to be flawed because of inadequate adjustment for seasonal trends or  
11 temperature and the use of questionable exposure indices. There were only a few time-series  
12 studies that examined O<sub>3</sub>-mortality associations between the 1980s and mid-1990s. These  
13 studies used more sophisticated approaches in addressing seasonal confounding and weather  
14 models. One of these studies (Shumway et al., 1988) focused on the associations with long-term  
15 fluctuations in Los Angeles, CA but did not examine short-term associations. A study that  
16 reanalyzed the Los Angeles, CA data with a focus on the short-term associations (Kinney and  
17 Özkaynak, 1991) did find that, of the PM and gaseous criteria pollutants, O<sub>3</sub> (reported as total  
18 oxidants) was most strongly associated with total nonaccidental mortality. Then two studies, one  
19 using Detroit, MI data (Schwartz, 1991) and the other using St. Louis, MO and Kingston-  
20 Harriman, TN data (Dockery et al., 1992), reported that PM but not O<sub>3</sub> was significantly  
21 associated with mortality. However, the 1996 O<sub>3</sub> AQCD discussed that without sufficient  
22 presentation of model specifications, it was difficult to evaluate whether the lack of O<sub>3</sub>-mortality  
23 associations was possibly due to mis-specification of the weather model. In summary, due to the  
24 insufficient number of studies that examined O<sub>3</sub>-mortality associations and the uncertainties  
25 regarding weather model specifications, the 1996 O<sub>3</sub> AQCD was unable to quantitatively assess  
26 O<sub>3</sub>-mortality excess risk estimates, or even provide qualitative assessment of the likelihood of  
27 O<sub>3</sub>-mortality associations.  
28

### 29 **7.4.2 Introduction to Assessment of Current Ozone-Mortality Studies**

30 Introductory discussions of the PM-mortality effects often cite historical air pollution  
31 incidents such as the 1952 London, England smog episode in which thousands of deaths were  
32 attributed to the air pollution from coal burning. There is no counterpart “historical episode” for

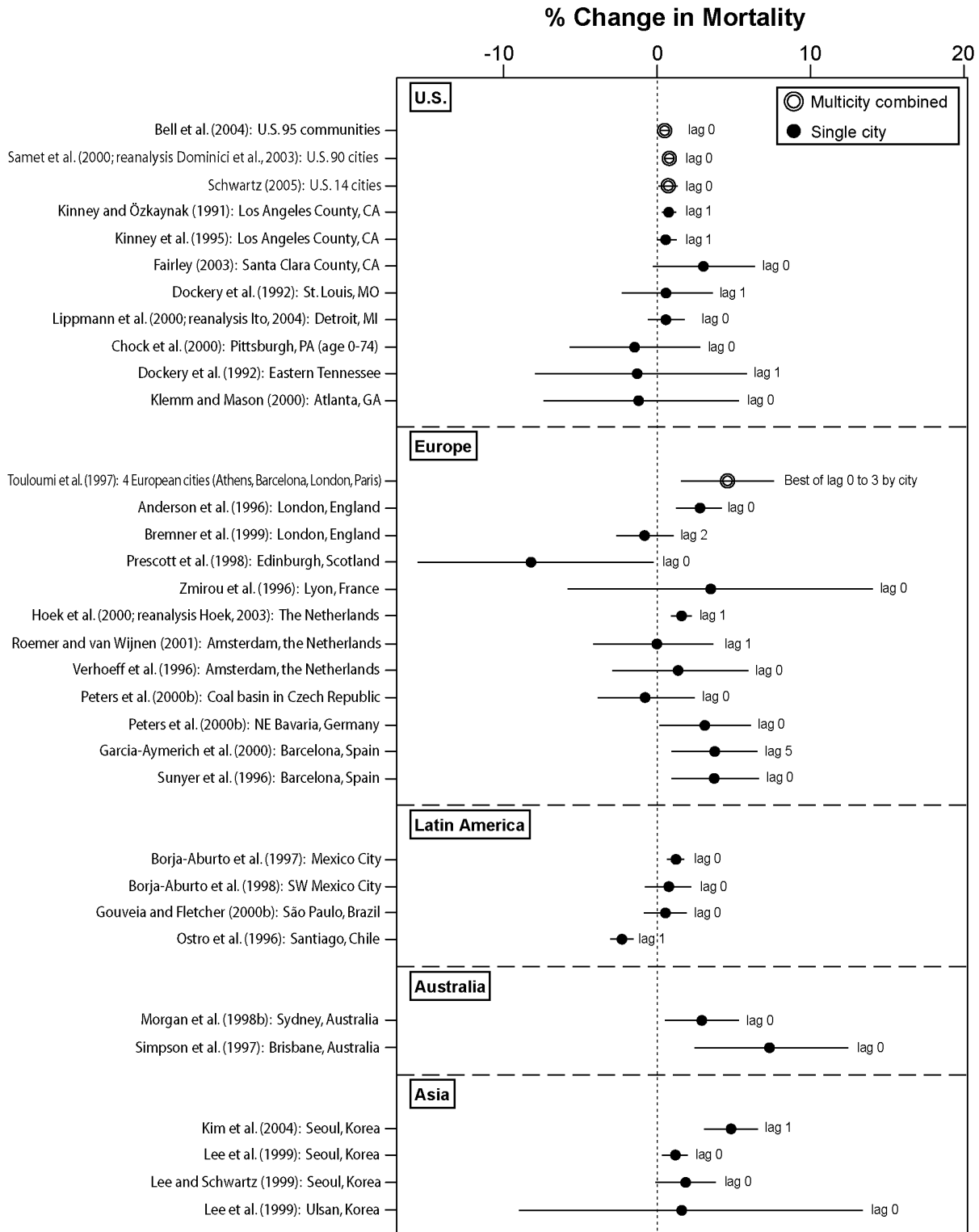
1 O<sub>3</sub>-mortality effects. Instead, the early recognition of the adverse health effects of summer  
2 oxidant air pollution, mainly from Los Angeles and other major cities with a high density of  
3 automobiles, were based on symptoms such as eye and throat irritations. Thus, the focus of PM  
4 epidemiology and that of O<sub>3</sub> epidemiology have been historically different.

5 As shown in Table AX7-5 in the Chapter 7 Annex, the number of short-term mortality  
6 studies that analyzed O<sub>3</sub> has increased markedly since the last publication of the O<sub>3</sub> AQCD in  
7 1996. The increased attention to PM-mortality associations in the early 1990s lead to the  
8 increase in studies that also examined O<sub>3</sub>, most often as a potential confounder for PM.

9 Although many of these PM studies also reported O<sub>3</sub> estimates, they often lacked specific  
10 hypotheses regarding mortality effects of O<sub>3</sub> as the focus of these studies was to examine the  
11 PM-mortality effect. This is in contrast to the O<sub>3</sub>-morbidity studies, most of which were  
12 specifically designed to examine effects of “summer haze” and O<sub>3</sub> (or oxidants) on respiratory  
13 and other symptoms, lung functions, and emergency department visits, etc. However, new  
14 studies with hypotheses developed specifically for O<sub>3</sub> effects on mortality have become  
15 available, such as the large U.S. 95 communities study by Bell et al. (2004), the U.S. 14 cities  
16 study by Schwartz (2005), and the 23 European cities study by Gryparis et al. (2004) discussed  
17 in the next section.

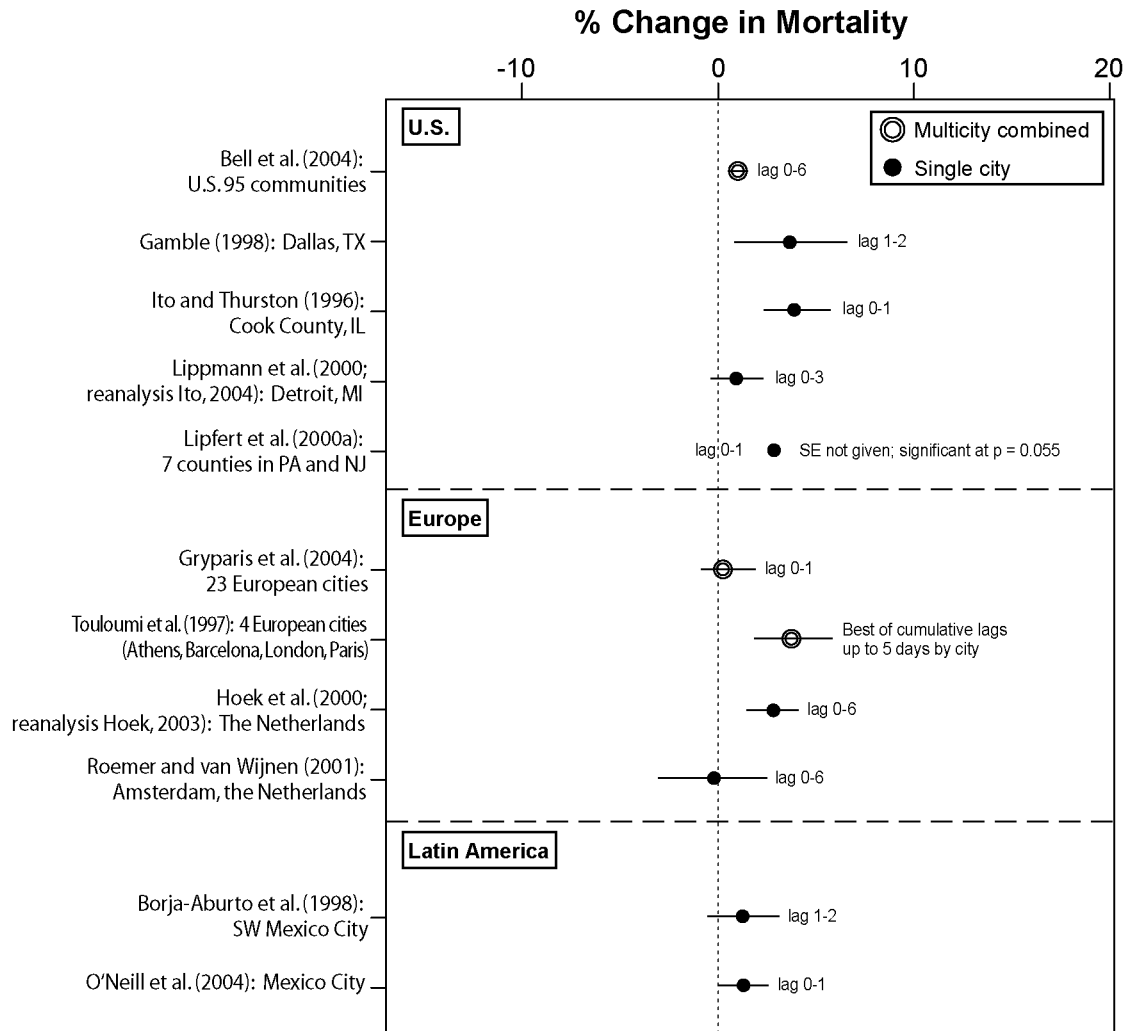
### 18 19 **7.4.3 Single-Pollutant Model Ozone-Mortality Risk Estimates**

20 To facilitate a quantitative overview of the O<sub>3</sub>-mortality effect estimates and their  
21 corresponding uncertainties, the percent excess risks of total nonaccidental mortality calculated  
22 using all year data are plotted in Figures 7-14 and 7-15. Studies that only conducted seasonal  
23 analyses will be presented in the next section. These figures do not include studies that only  
24 examined cause-specific mortality. Figure 7-14 only presents the results from single-day lag  
25 models. Results from multiday lag models are shown in Figure 7-15. All effect estimates are  
26 from single-pollutant models and include all age groups unless noted otherwise. The majority of  
27 the estimates are positive with a few exceptions. Five multicity studies, three from the U.S.  
28 (Bell et al., 2004; Samet et al., 2000 [reanalysis Dominici et al., 2003]; Schwartz, 2005) and two  
29 from Europe (Gryparis et al., 2004; Touloumi et al., 1997), also showed generally positive  
30 associations.



**Figure 7-14. All cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) for all year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted. Only results from single-day lag models are presented.**





**Figure 7-15. All cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) for all year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted. Only results from multiday lag models are presented.**

1 The initial primary objective of the original NMMAPS (Samet et al., 2000; reanalysis  
 2 Dominici et al., 2003) was to investigate the effects of PM, but the study also comprehensively  
 3 examined mortality risk estimates from gaseous pollutants in 90 U.S. cities over the period of  
 4 1987 to 1994. Among the 90 cities, 80 monitored O<sub>3</sub> either year-round or during the warm  
 5 season. The study illustrated that the mortality risk estimates for O<sub>3</sub> varied by season. The  
 6 estimate using all available data was about half of that for summer-only data at a lag of 1-day

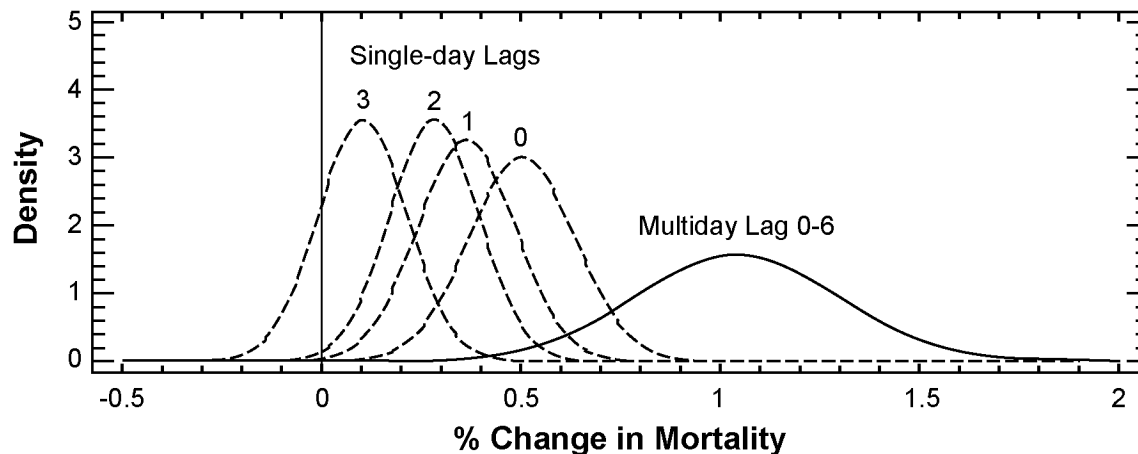
1 (see Section 7.6.3.2 for further discussion). Bell et al. (2004) extended the original NMMAPS  
2 by adding six more years (from 1987 to 2000) and 15 more communities (a total of  
3 95 communities), and examined the effects of O<sub>3</sub> on mortality. The results of this study are  
4 discussed in detail here because of the study's emphasis on U.S. data and the inclusion of 95  
5 large communities across the country, making this mortality study most representative of the  
6 U.S. population. In addition, this study is one of the few that have focused specifically on O<sub>3</sub>  
7 hypotheses testing and investigated several important issues. Among the 95 communities  
8 examined in this study, 55 monitored O<sub>3</sub> throughout the year and 32 only monitored during the  
9 warm season. Eight additional cities switched from warm season only to year-round monitoring  
10 or year-round to warm season only monitoring at some point during the study period. The mean  
11 24-h avg O<sub>3</sub> concentration was approximately 26 ppb for the 95 communities.  
12 Within-community results first were calculated using single-day lags of 0, 1, 2, and 3 days, and  
13 a 7-day distributed lag in O<sub>3</sub> exposure. A two-stage Bayesian hierarchical model was used to  
14 determine a national average effect estimate, taking into consideration city-to-city variation.  
15 Figure 7-16 presents the Bayesian community-specific and national average O<sub>3</sub> risk estimates for  
16 total mortality per 20 ppb increase in 24-h avg O<sub>3</sub> from a constrained 7-day distributed lag  
17 model. The Bayesian community-specific estimates were shrunk to the national average  
18 estimate by a factor that was inversely proportional to the heterogeneity of the community-  
19 specific relative rates. The heterogeneity of the effect estimates from the individual cities is  
20 partially attributable to differences in pollution characteristics, the use of air conditioning, time-  
21 activity patterns, and socioeconomic factors. Due to the random variation as well as the smaller  
22 sample sizes within each city, emphasis is given to the national average effect estimate.

23 In the U.S. 95 communities study, the largest risk estimate for O<sub>3</sub>-mortality was obtained  
24 with a 0-day lag, followed by diminishing risk estimates with 1-, 2-, and 3-day lags (Figure  
25 7-17). Ozone exposure at a 0-day lag was associated with a 0.50% (95% PI: 0.24, 0.78) excess  
26 risk in mortality per 20 ppb increase in 24-h avg O<sub>3</sub>. The 7-day distributed lag model, which  
27 examined the cumulative effect from the same day and six previous days, also is shown in  
28 Figure 7-17. A cumulative excess mortality risk of 1.04% (95% PI: 0.54, 1.55) per 20 ppb  
29 increase in 24-h avg O<sub>3</sub> during the previous week was observed. In a related U.S. study of the  
30 19 largest cities by Huang et al. (2005), the O<sub>3</sub> estimate for the summer season was 1.47% (95%  
31 PI: 0.54, 2.39) excess risk of cardiopulmonary mortality with current-day exposure. Smaller



**Figure 7-16. Community-specific Bayesian estimates and national average for the percent change (95% PI) in daily mortality per 20 ppb increase in 24-h avg O<sub>3</sub> in the previous week using a constrained distributed lag model for 95 U.S. communities (NMMAPS), arranged by size of the effect estimate. Results from all available data are presented (32 of the 95 communities only had warm season data).**

Source: Derived from Bell et al. (2004).



**Figure 7-17. Comparison of single-day lags (0-, 1-, 2-, and 3-day) to a cumulative multiday lag (0- to 6-day) for percent changes in all cause mortality per 20 ppb increase in 24-h avg O<sub>3</sub> in all ages.**

Source: Derived from Bell et al. (2004).

1 effects also were observed with 1- and 2-day lags of exposure. The effect estimate for the 7-day  
 2 distributed lag was 2.52% (95% PI: 0.94, 4.10) excess risk of cardiopulmonary mortality. These  
 3 findings suggest that the effect of O<sub>3</sub> on mortality is immediate, but also may persist over  
 4 multiple days.

5 The influence of higher O<sub>3</sub> levels on the risk estimate also was evaluated in the U.S.  
 6 95 communities study. When the data were restricted to days with 24-h avg O<sub>3</sub> levels less than  
 7 60 ppb for the 1-day lag analysis, the national estimate did not substantially change (0.30%  
 8 [95% PI: 0.06, 0.54] per 20 ppb increase for days with levels below 60 ppb versus 0.36% [95%  
 9 PI: 0.12, 0.61] for all days). These results suggest that the O<sub>3</sub>-mortality associations occur at  
 10 24-h avg O<sub>3</sub> levels below 60 ppb.

11 Schwartz (2005) examined O<sub>3</sub>-mortality associations using data from 14 U.S. cities.  
 12 A case-crossover study design was used to compare the influence of adjustment methods for  
 13 temperature (regression splines of temperature versus matching case and control periods by  
 14 temperature). The risk estimate obtained by matching (0.92% [95% CI: 0.06, 1.80] per 40 ppb  
 15 increase in 1-h max O<sub>3</sub>) was similar to that obtained with regression splines (0.76% [95% CI:

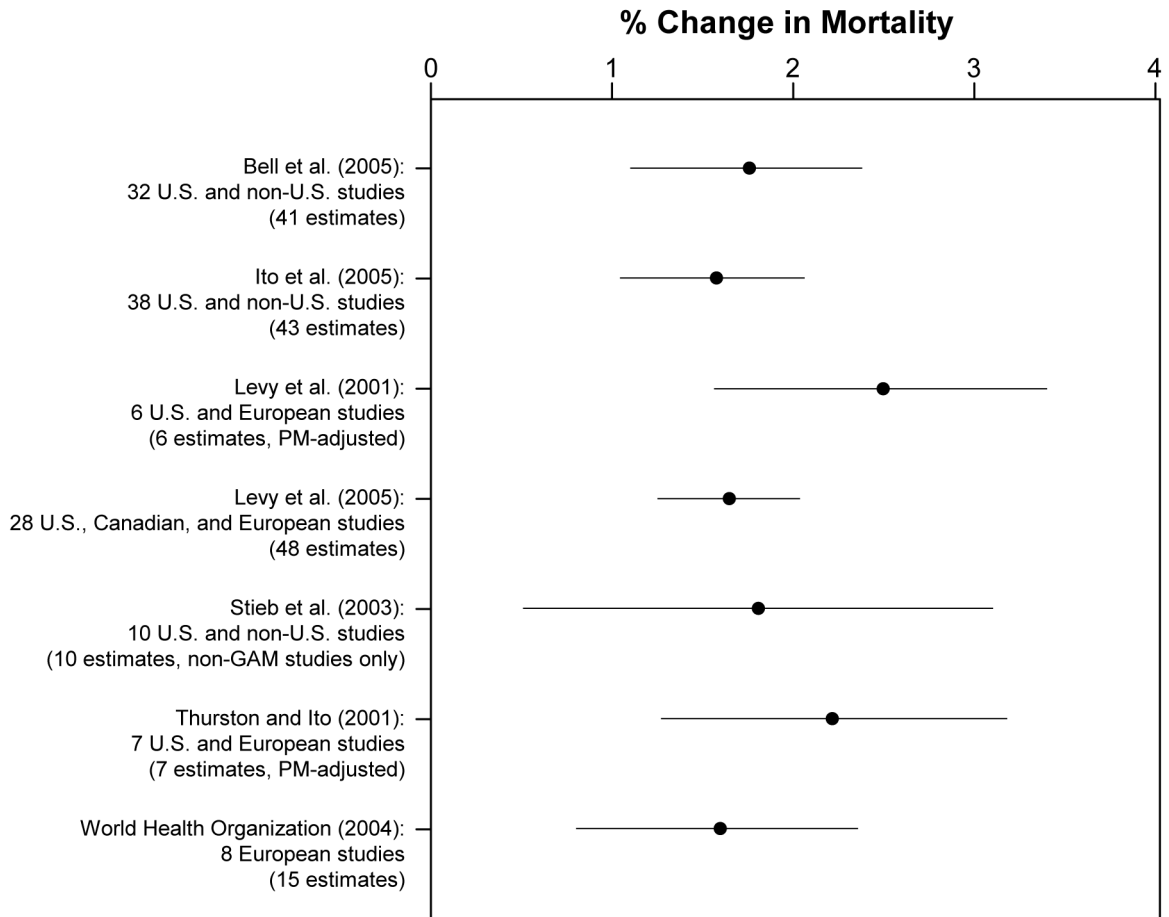
1 0.13, 1.40]), suggesting that the O<sub>3</sub>-mortality risk estimates were not sensitive to these  
2 adjustment methods for temperature.

3 The APHEA 1 project (Touloumi et al., 1997) reported a pooled random effects estimate of  
4 4.5% (95% CI: 1.6, 7.7) per 40 ppb increase in 1-h max O<sub>3</sub> using the best single-day lag model  
5 results from four European cities (London, Athens, Barcelona, and Paris). As an extension of  
6 the four European cities study, researchers of the APHEA 2 project investigated the effect of O<sub>3</sub>  
7 on total, cardiovascular, and respiratory mortality in 23 cities throughout Europe (Gryparis et al.,  
8 2004). Ozone data was available year-round in all 23 cities. A cumulative lag of 0 to 1 days was  
9 hypothesized a priori. A two-stage hierarchical model, which accounted for statistical variance  
10 and heterogeneity among cities, was used to estimate the pooled regression coefficients. Due to  
11 substantial heterogeneity among cities, random effects regression models were applied. The  
12 pooled effect estimate for the 23 European cities (0.23% [95% CI: -0.85, 1.95] per 40 ppb  
13 increase in 1-h max O<sub>3</sub> for all seasons) was positive but considerably smaller compared to that  
14 obtained in the APHEA 1 study. The researchers noted that there was a considerable seasonal  
15 difference in the O<sub>3</sub> effect on mortality, thus the small effect for the all year data might be  
16 attributable to inadequate adjustment for confounding by season. This seasonal effect will be  
17 discussed further in the next section.

18 Collectively, the single-pollutant model estimates from the single- and multiple-city  
19 studies shown in Figures 7-14 and 7-15 suggest an excess risk of total nonaccidental mortality  
20 associated with acute O<sub>3</sub> exposure. Despite the different analytical approaches and alternative  
21 model specifications used in the various studies, overall, the range of estimates were relatively  
22 narrow, with most of the positive estimates falling in the range from 0.5 to 5% excess risk in  
23 mortality per standardized increment.

#### 24 **7.4.4 Meta-analyses of O<sub>3</sub>-Mortality Risk Estimates**

25 Several studies in recent years conducted meta-analyses of O<sub>3</sub>-mortality associations (Levy  
26 et al., 2001; Stieb et al., 2002, 2003; Thurston and Ito, 2001; World Health Organization, 2004).  
27 Figure 7-18 presents the combined O<sub>3</sub> risk estimates from the various meta-analyses. Most of  
28 these studies included GAM studies using default convergence criteria except Stieb et al. (2003),  
29 which compared effect estimates from GAM-affected studies to non-GAM studies. All of these  
30 meta-analyses reported fairly consistent and positive combined estimates, approximately 2%  
31



**Figure 7-18. Combined all cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) from recent meta-analyses per standardized increment (see Section 7.1.3.2). Note that all meta-analyses, except Stieb et al. (2003), included studies which used Poisson GAM with default convergence criteria.**

1 excess total nonaccidental mortality per standardized increment (see Section 7.1.3.2). However,  
 2 most of these studies were not analytical in design in that they did not attempt to examine the  
 3 source of heterogeneity, although one suggested an influence of weather model specification  
 4 (Thurston and Ito, 2001) and another reported evidence of publication bias (World Health  
 5 Organization, 2004) in the past literature. None of these studies address the issue of season-  
 6 specific estimates, therefore, interpreting these combined estimates requires caution.

7 Most recently, three research groups conducted independent meta-analyses of O<sub>3</sub>-mortality  
 8 associations (Bell et al., 2005; Ito et al., 2005; Levy et al., 2005). These analyses attempted to  
 9 evaluate the source of heterogeneity using the most up-to-date literature database. These

1 analyses were also systematically compared and discussed (Bates, 2005; Goodman, 2005). The  
2 all-season combined point estimates per standardized increment from these three meta-analyses  
3 were remarkably consistent: 1.75% (95% PI: 1.10, 2.37), 1.6% (95% CI: 1.1, 2.0), and 1.64%  
4 (95% CI: 1.25, 2.03), for the Bell et al., Ito et al., and Levy et al. studies, respectively. All three  
5 studies also indicated that the estimates were higher in warm seasons. Each of these studies is  
6 briefly summarized below. Their findings related to specific issues are discussed later in the  
7 corresponding sections.

8 Bell et al. (2005) conducted a meta-analysis of 144 effect estimates from 39 U.S. and  
9 non-U.S. studies and estimated pooled effects by lags, age groups, specific causes, and exposure  
10 metrics. The results were also compared with their NMMAPS results (Bell et al., 2004).  
11 A two-stage Bayesian hierarchical model was used to estimate the combined estimate by taking  
12 into account the within-city variance (the statistical uncertainty) and between-study variance (the  
13 heterogeneity across cities). They concluded that the results provided strong evidence of a short-  
14 term association between O<sub>3</sub> and mortality that was not sensitive to adjustment for PM or model  
15 specifications (discussed in Section 7.4.6). However, they suggested that, based on comparisons  
16 between the meta-analysis results and NMMAPS results, there was evidence of publication bias  
17 (1.75% [95% CI: 1.10, 2.37] per 20 ppb increase in 24-h avg O<sub>3</sub> for meta-analysis versus 0.50%  
18 [95% CI: 0.24, 0.78] for NMMAPS 0-day lag results).

19 Ito et al. (2005) conducted a meta-analysis of 43 U.S. and non-U.S. studies but also  
20 analyzed data from 7 U.S. cities to further examine the issues identified in their meta-analysis.  
21 Adjusting for PM did not substantially influence the O<sub>3</sub>-mortality effect estimates in either the  
22 meta-analysis or 7 U.S. cities analysis. The multicity analysis further indicated that the  
23 difference in the weather adjustment model could result in a twofold difference in risk estimates  
24 (e.g., 1.96% versus 0.96% in multicity combined estimates across alternative weather models for  
25 the O<sub>3</sub>-only, all year case). In the meta-analysis, they found suggestive evidence of publication  
26 bias (a significant asymmetry in the funnel plot), but adjusting for the asymmetry reduced the  
27 combined estimate only slightly (from 1.6% [95% CI: 1.1, 2.0] to 1.4% [95% CI: 0.9, 1.9] per  
28 20 ppb increase in 24-h avg O<sub>3</sub>). The extent of potential bias implicated in this study differed  
29 compared to that in Bell et al. (2005). The source of this difference is not clear, but Ito et al.  
30 state that sensitivity analyses comparing estimates from commonly used weather model

1 specifications suggest that the stringent weather model used in NMMAPS may tend to yield  
2 smaller risk estimates than those used in other studies.

3 Levy et al. (2005) analyzed 48 estimates from 28 studies from the U.S., Canada, and  
4 Europe using an empiric Bayesian meta-regression with covariates including the relationship  
5 between O<sub>3</sub> and other pollutants, proxies for the relationship between personal exposure and  
6 ambient concentration such as air conditioning prevalence, and statistical methods used. They  
7 found that the air conditioning prevalence (a greater effect in cities with less air conditioning)  
8 and lag time (same-day effects larger than lagged effects) were the strongest predictors of  
9 between-study variability. The warm season estimates were larger than the cool season  
10 estimates. The influences of copollutants were inconsistent, but they found a potential influence  
11 of summertime PM<sub>2.5</sub>.

12 As stated earlier, the combined O<sub>3</sub> excess mortality risk estimates from the meta-analyses  
13 by Bell et al., Ito et al., and Levy et al. were very consistent. Although the analyses were  
14 conducted independently, there was considerable overlap among the estimates used in the three  
15 meta-analyses; thus, the agreement in the combined risk estimates was not unexpected. The  
16 common findings among these three meta-analyses, aside from the consistency in their combined  
17 estimates, include: (1) no difference in estimates between GAM studies using default versus  
18 stringent convergence criteria; (2) estimates were larger in warm seasons; and (3) no strong  
19 indication of PM confounding. Both Bell et al. and Levy et al. studies found that the estimates at  
20 lag 0-day were larger than longer lags. Both the Bell et al. and Ito et al. studies suggested  
21 evidence of publication bias. These three studies, along with the earlier meta-analyses, provide  
22 strong evidence that O<sub>3</sub> is associated with mortality. The combined effect estimates from the  
23 various meta-analyses ranged from 1.5 to 2.5% excess risk in all cause mortality.  
24

#### 25 **7.4.5 Seasonal Variation in Ozone-Mortality Risk Estimates**

26 Since the seasonal cycle of O<sub>3</sub> follows the seasonal cycle of temperature (which is  
27 inversely related to the mortality seasonal cycle), inadequate adjustment of temporal trends in  
28 the regression model may lead to negative O<sub>3</sub>-mortality risk estimates. In addition, as discussed  
29 in Section 7.1.3.5, in some cities low-level O<sub>3</sub> during the winter may be negatively correlated  
30 with PM and other primary pollutants, resulting in negative correlations between O<sub>3</sub> and



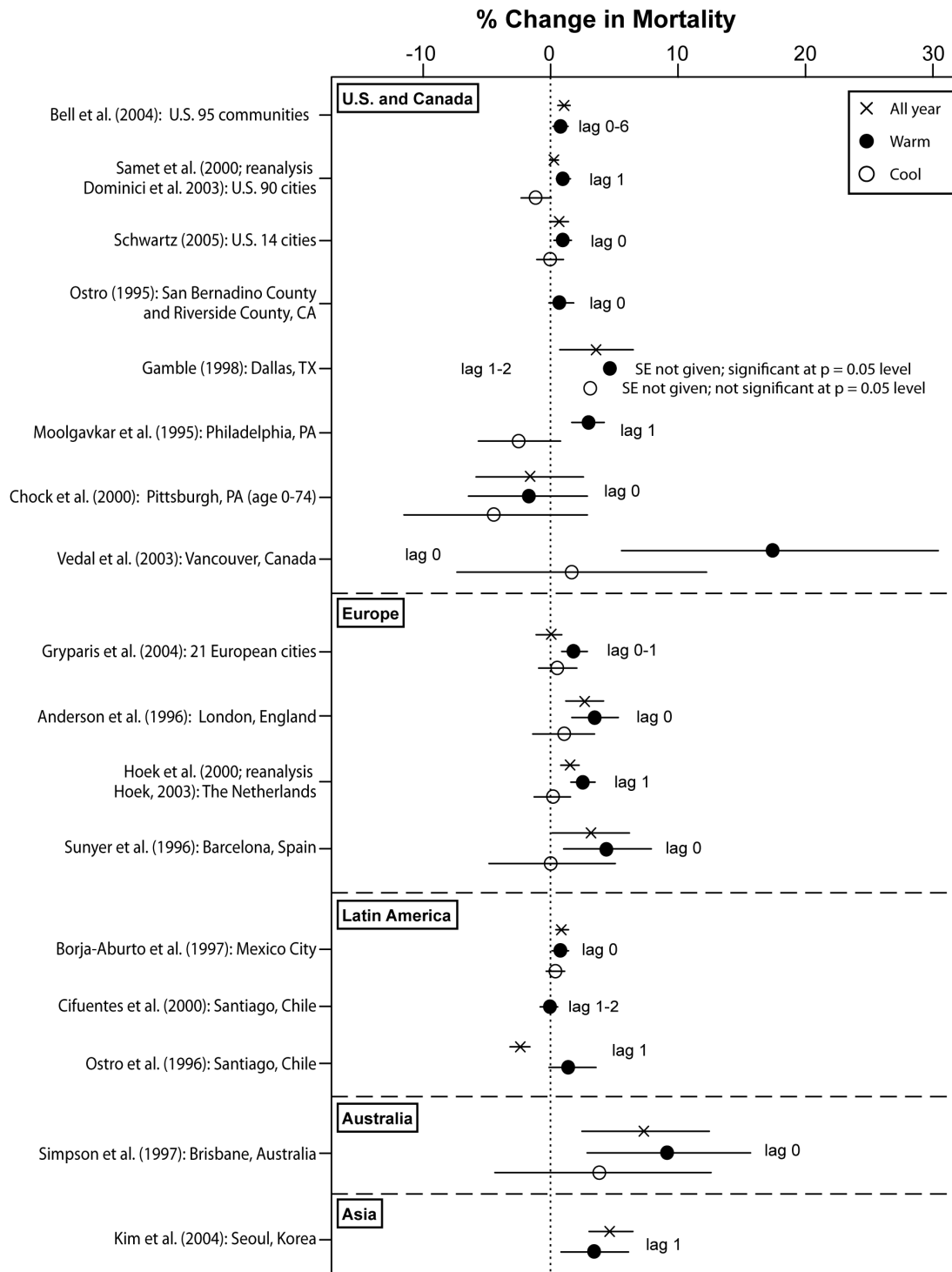
1 mortality even in short-term relationships. The confounding effect by season could be  
2 substantially reduced by conducting season-stratified analyses.

3 A fewer number of O<sub>3</sub>-mortality studies performed seasonal analyses. Figure 7-19 presents  
4 the studies that reported O<sub>3</sub> risk estimates for all cause mortality by season. For those studies  
5 that obtained O<sub>3</sub> risk estimates for each of the four seasons, only summer and winter results are  
6 shown. The estimates for year-round data analyses, when available, also are shown for  
7 comparisons. In all the studies, the O<sub>3</sub> risk estimates are larger during the warm season than the  
8 cool season, with the all year estimates generally in between the two seasonal estimates.

9 In three U.S. and European multicity studies (Gryparis et al., 2004; Samet et al., 2000  
10 [reanalysis Dominici et al., 2003]; Schwartz, 2005), season-stratified analyses indicated that the  
11 O<sub>3</sub>-mortality effect estimates were significant and positive in the warm season, with larger  
12 effects observed compared to the year-round analyses. The effect estimates from the cool season  
13 were notably smaller and less significant. In the case of the U.S. 90 cities study (of which  
14 80 cities had O<sub>3</sub> data available) the winter (December, January, and February) mortality estimate  
15 was negative, which was most likely attributable to the inverse relationship between O<sub>3</sub> and PM  
16 in the winter.

17 In the U.S. 95 communities study by Bell et al. (2004), no significant difference was  
18 observed between the estimates from all available data and warm season only data (April-  
19 October); cool season only analyses were not performed. The warm season effect estimate using  
20 the 7-day constrained distributed lag model was 0.78% (95% PI: 0.26, 1.30) excess risk per 20  
21 ppb increase in 24-h avg O<sub>3</sub>, compared to 1.04% (95% PI: 0.54, 1.55) calculated using all  
22 available data. In the 55 communities with year-round O<sub>3</sub> data, the all year effect estimate was  
23 0.96% (95% PI: 0.32, 1.57).

24 All three recent meta-analyses (Bell et al., 2005; Ito et al., 2005; Levy et al. 2005), found  
25 that the estimates for warm seasons were larger than all year estimates. In Bell et al., the warm  
26 season estimate was 3.02% [95% PI: 1.45, 4.63], compared to the all year estimate of 1.75%  
27 [95% PI: 1.10, 2.37]. In the subset of 10 cities examined in Ito et al., the warm season and all  
28 year estimates were 3.5% [95% CI: 2.1, 4.9] and 2.2% [95% CI: 0.8, 3.6], respectively.  
29 Likewise, Levy et al. observed a 3.38% [95% CI: 2.27, 4.42] excess risk in the warm season  
30 compared to a 1.64% [95% CI: 1.25, 2.03] excess risk using all year data. All results presented  
31 are percent excess risk in mortality per standardized increment.



**Figure 7-19. All cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) by season per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.**

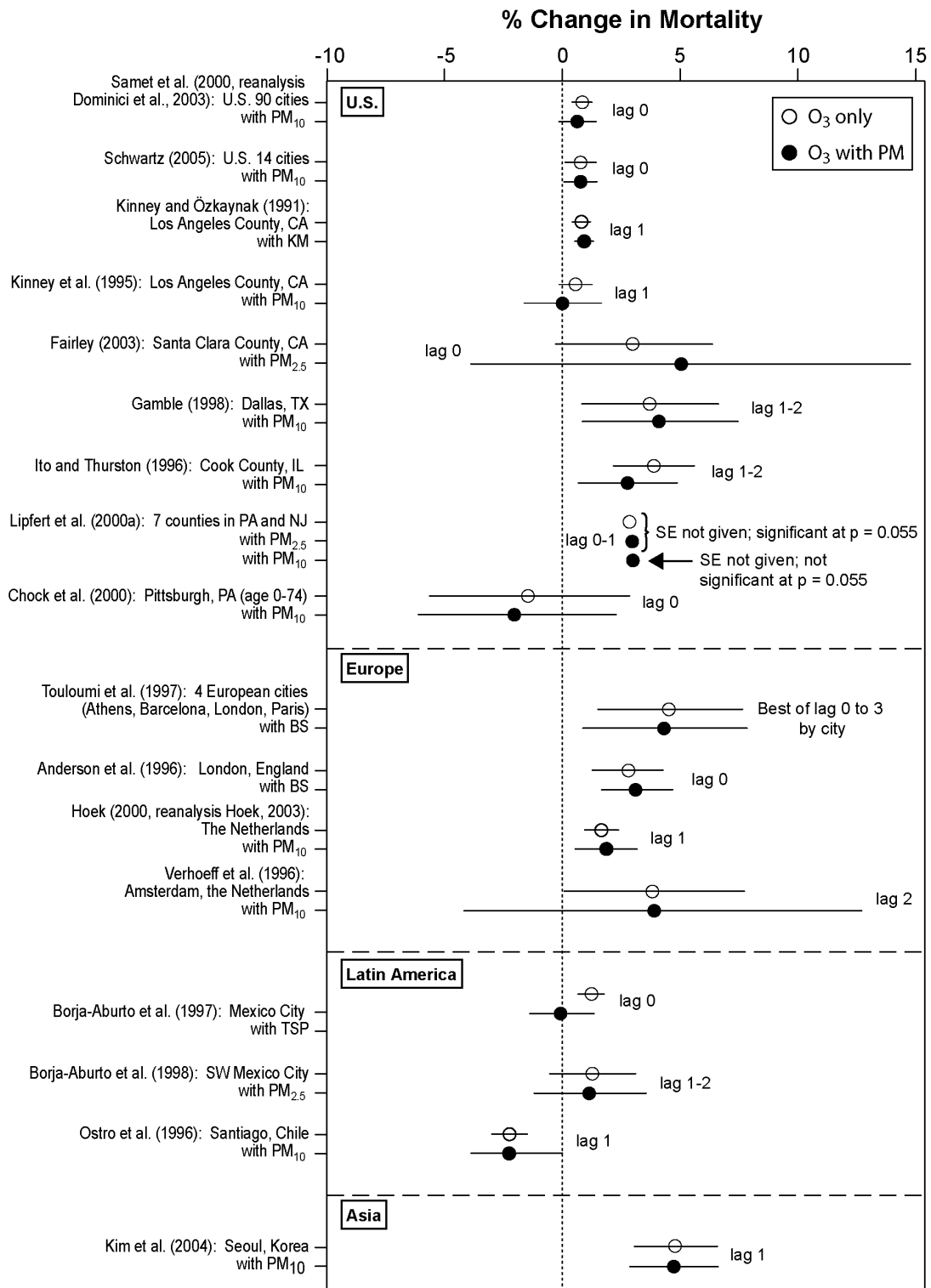
1 Studies that conducted analysis by season indicate that O<sub>3</sub>-mortality risk estimates are often  
2 larger in the warm season compared to the colder season. The seasonal dependence of  
3 O<sub>3</sub>-mortality effects complicates interpretation of O<sub>3</sub> risk estimates calculated from year-round  
4 data without adequate adjustment of temporal trends.  
5

#### 6 **7.4.6 Ozone-Mortality Risk Estimates Adjusting for PM Exposure**

7 The confounding between “winter type” pollution (e.g., CO, SO<sub>2</sub>, and NO<sub>2</sub>) and O<sub>3</sub> is not  
8 of great concern because the peaks of these pollutants do not strongly coincide. The main  
9 confounders of interest for O<sub>3</sub>, especially for the northeast U.S., are “summer haze” type  
10 pollutants such as acid aerosols and sulfates. Since very few studies had these chemical  
11 measurements, PM (especially PM<sub>2.5</sub>), may serve as surrogates. However, due to the expected  
12 high correlation among the constituents of the “summer haze mix,” multipollutant models  
13 including these pollutants may result in unstable coefficients, and therefore, an interpretation of  
14 such results requires some caution.

15 Figure 7-20 shows the O<sub>3</sub> risk estimates with and without adjustment for PM indices using  
16 all year data in studies that conducted two-pollutant analyses. Approximately half of the O<sub>3</sub> risk  
17 estimates slightly increased while the other half slightly decreased in value with the inclusion of  
18 PM in the models. In general, the O<sub>3</sub>-mortality risk estimates were robust to adjustment for PM  
19 in the models, with the exception of Los Angeles, CA data with PM<sub>10</sub> (Kinney et al., 1995) and  
20 Mexico City data with TSP (Borja-Aburto et al., 1997).

21 The U.S. 95 communities study by Bell et al. (2004) examined the sensitivity of acute  
22 O<sub>3</sub>-mortality effects to potential confounding by PM<sub>10</sub>. Restricting analysis to days when both  
23 O<sub>3</sub> and PM<sub>10</sub> data were available, the community-specific O<sub>3</sub>-mortality effect estimates as well as  
24 the national average results indicated that O<sub>3</sub> was robust to adjustment for PM<sub>10</sub> (Bell et al.,  
25 2004). There was insufficient data available to examine potential confounding by PM<sub>2.5</sub>.  
26 One study (Lipfert et al., 2000a) reported O<sub>3</sub> risk estimates with and without sulfate adjustment.  
27 Lipfert et al. (2000a) calculated O<sub>3</sub> risk estimates based on mean (45 ppb) less background (not  
28 stated) levels of 1-h max O<sub>3</sub> in seven counties in Pennsylvania and New Jersey. The O<sub>3</sub> risk  
29 estimate was not substantially affected by the addition of sulfate in the model (3.2% versus  
30 3.0% with sulfate) and remained statistically significant.



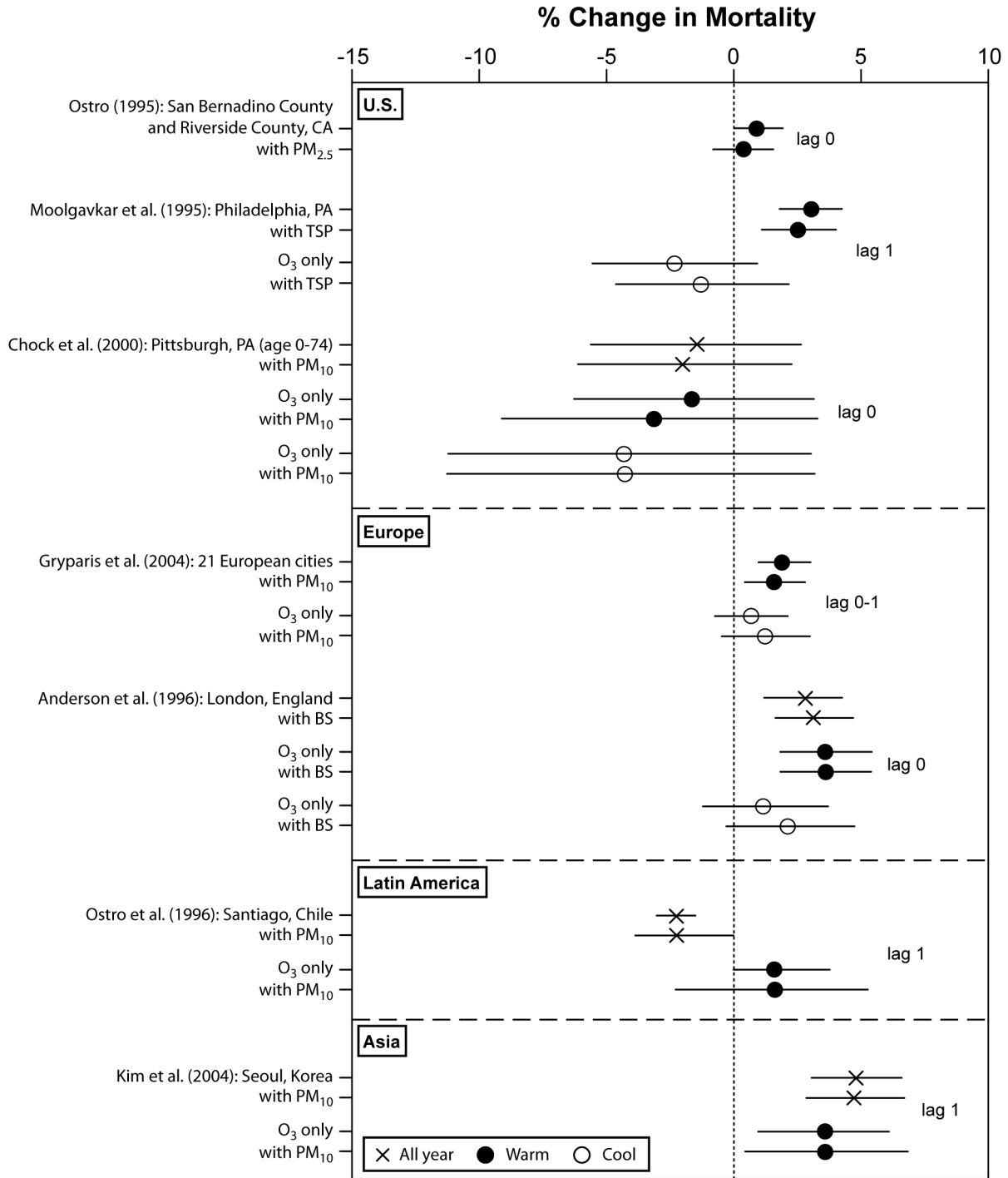
**Figure 7-20. All cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) with adjustment for PM indices for all year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.**

1 Several O<sub>3</sub>-mortality studies examined the effect of confounding by PM indices in different  
2 seasons (Figure 7-21). In analyses using all year data and warm season only data, O<sub>3</sub> risk  
3 estimates were once again fairly robust to adjustment for PM indices, with values showing both  
4 slight increases and decreases with the inclusion of PM in the model. In the analyses using cool  
5 season data only, the O<sub>3</sub> risk estimates all increased slightly with the adjustment of PM indices,  
6 although none reached statistical significance.

7 The three recent meta-analyses (Bell et al., 2005; Ito et al., 2005; Levy et al. 2005) all  
8 examined the influence of PM on O<sub>3</sub> risk estimates. No substantial influence was observed in  
9 any of these studies. In the analysis by Bell et al., the combined estimate without PM adjustment  
10 was 1.75% (95% PI: 1.10, 2.37) from 41 estimates, and the combined estimate with PM  
11 adjustment was 1.95% (95% PI: -0.06, 4.00) from 11 estimates per 20 ppb increase in 24-h avg  
12 O<sub>3</sub>. In the meta-analysis of 15 cities by Ito et al., the combined estimate was 1.6% (95% CI: 1.1,  
13 2.2) and 1.5% (95% CI: 0.8, 2.2) per 20 ppb in 24-h avg O<sub>3</sub> without and with PM adjustment,  
14 respectively. The additional time-series analysis of six cities by Ito et al. found that the  
15 influence of PM by season varied across alternative weather models but was never substantial.  
16 Levy et al. examined the regression relationships between O<sub>3</sub> and PM indices (PM<sub>10</sub> and PM<sub>2.5</sub>)  
17 with O<sub>3</sub>-mortality effect estimates for all year and by season. Positive slopes, which might  
18 indicate potential confounding, were observed for PM<sub>2.5</sub> on O<sub>3</sub> risk estimates in the summer and  
19 all year periods, but the relationships were weak. The effect of one causal variable (i.e., O<sub>3</sub>)  
20 is expected to be overestimated when a second causal variable (e.g., PM) is excluded from the  
21 analysis if the two variables are positively correlated and act in the same direction. However,  
22 the results from these meta-analyses as well as several single- and multiple-city studies indicate  
23 that copollutants generally do not appear to substantially confound the association between O<sub>3</sub>  
24 and mortality.

#### 25 26 **7.4.7 Ozone Risk Estimates for Specific Causes of Mortality**

27 In addition to all cause mortality, several studies examined broad underlying causes of  
28 mortality, such as cardiovascular and respiratory causes. The U.S. 95 communities study  
29 (Bell et al., 2004) analyzed O<sub>3</sub> effect estimates from cardiovascular and respiratory mortality.  
30 Significant effects were seen at 0- and 2-day lags with results similar to total mortality. The  
31 national average estimate from the constrained distributed lag model was slightly greater for

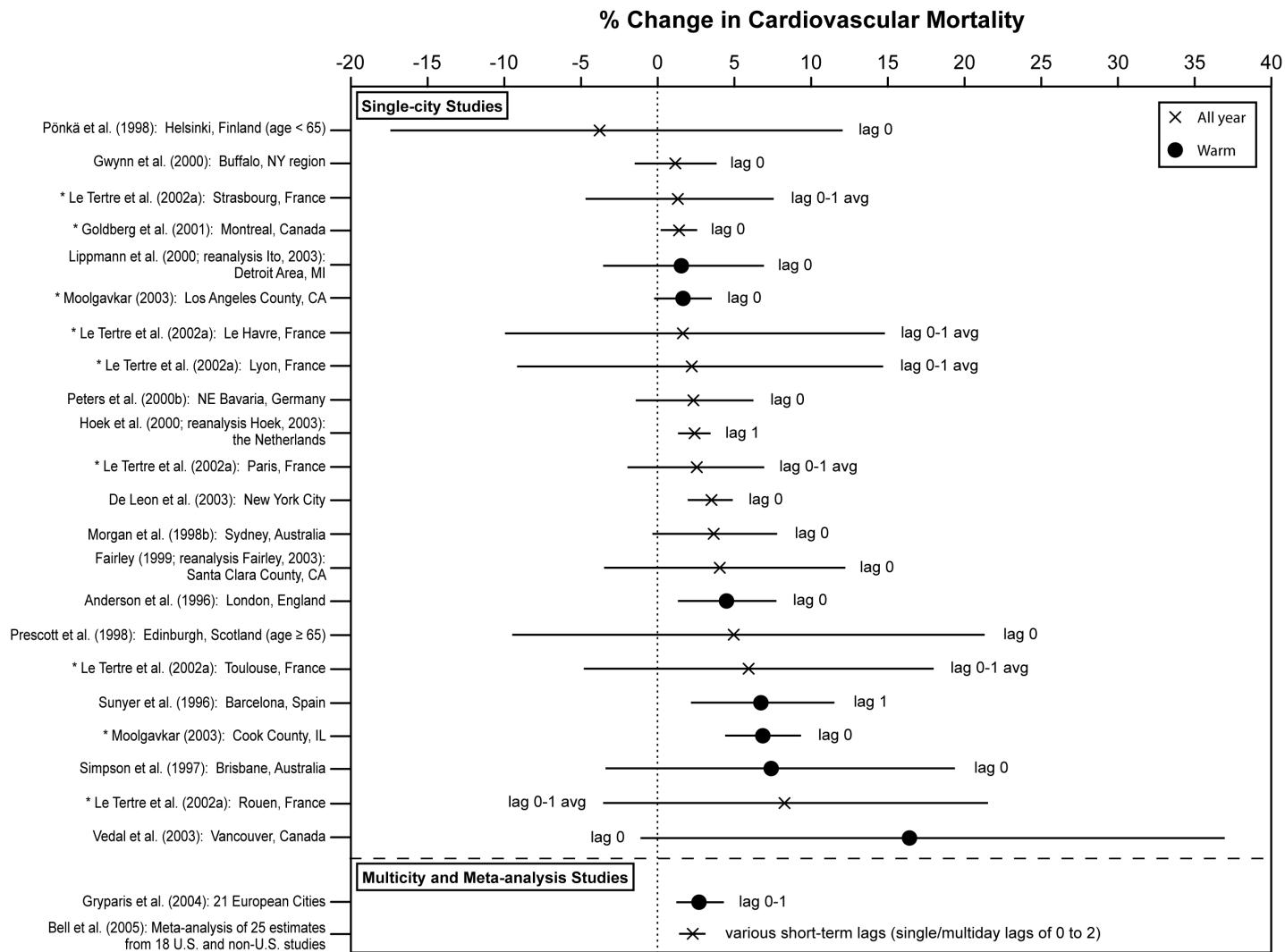


**Figure 7-21. All cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) with adjustment for PM indices by season per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.**

1 cardiopulmonary deaths than deaths from all causes, with an excess risk of 1.28% (95% PI:  
2 0.62, 1.97) compared to 1.04% (95% PI: 0.54, 1.55) per 20 ppb increase in 24-h avg O<sub>3</sub> in the  
3 preceding week. In a related study, Huang et al. (2005) examined O<sub>3</sub> effects on cardiopulmonary  
4 mortality during the summers (June to September) of 1987 to 1994 in 19 large U.S. cities from  
5 the NMMAPS database. In the 7-day distributed lag model, the O<sub>3</sub> effect estimate was  
6 2.52% (95% PI: 0.94, 4.10) excess risk in cardiopulmonary mortality per 20 ppb increase in  
7 24-h avg O<sub>3</sub>.

8 Figure 7-22 presents the effect estimates of the association between O<sub>3</sub> and cardiovascular  
9 mortality for all year and warm season analyses. All studies, with the exception of Pönkä et al.  
10 (1998), showed positive associations between O<sub>3</sub> and cardiovascular mortality. However, as  
11 with all cause mortality, there appears to be heterogeneity in the effect estimates across studies.  
12 The cardiovascular mortality estimate from the meta-analysis by Bell et al. (2005) appears to be  
13 close to the mode of the effect estimates from the various studies, as shown in Figure 7-22. This  
14 is expected as many of these studies are included in the meta-analysis. Bell et al. observed that  
15 the posterior mean estimate for cardiovascular causes (2.23% [95% PI: 1.36, 3.08] excess risk  
16 per 20 ppb increase in 24-h avg O<sub>3</sub> from 25 estimates) was slightly larger than that for total  
17 mortality (1.75% [95% PI: 1.10, 2.37] excess risk from 41 estimates). However, since  
18 cardiovascular deaths account for the largest fraction (over 40%) of total deaths, it is not  
19 surprising that the risk estimates for cardiovascular mortality are somewhat similar to those from  
20 all cause mortality. Overall, the cardiovascular mortality risk estimates in the current literature  
21 show consistently positive associations with some heterogeneity (most estimates fall within the  
22 range of 1 to 8% per 40 ppb increase in 1-h avg O<sub>3</sub>).

23 Several studies observed that the risk estimates for the respiratory category were larger  
24 than the cardiovascular and total nonaccidental categories (e.g., Anderson et al., 1996; Gouveia  
25 and Fletcher, 2000b; Gryparis et al., 2004; Zmirou et al., 1998). In the European 21 multicity  
26 study (Gryparis et al., 2004), the warm season effect estimate for respiratory mortality was  
27 6.75% (95% CI: 4.38, 9.10) excess risk per 30 ppb increase in 8-h max O<sub>3</sub>, compared to 2.70%  
28 (95% CI: 1.29, 4.32) for cardiovascular mortality and 1.82% (95% CI: 0.99, 3.06) for total  
29 mortality. In contrast, other studies have found that the risk estimates for the respiratory  
30 category were smaller or even negative while the risk estimates for total or cardiovascular  
31 categories were positive (e.g., Borja-Aburto et al., 1998; Bremner et al., 1999; Lipfert et al.,



**Figure 7-22. Ozone-associated cardiovascular mortality risk estimates (95% CI) per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted. \*Note that Goldberg et al. (2001), Le Tertre et al. (2002a), and Moolgavkar (2003) performed analyses using Poisson GAM with default convergence criteria.**



1 2000a; Morgan et al., 1998b). The apparent inconsistencies across studies may be due in part to  
2 the differences in model specifications, but they may also reflect the lower statistical power  
3 associated with the smaller daily counts of the respiratory category (usually accounting for less  
4 than 10% of total deaths) compared to the larger daily counts for the cardiovascular category  
5 (approximately 40 to 50% of total deaths). Thus, an examination of the differences in risk  
6 estimates across specific causes requires a large population and/or a long period of data  
7 collection. In the meta-analysis by Bell et al. (2005), which combined 23 estimates from  
8 17 studies for respiratory mortality, the effect estimate for respiratory causes was smaller (0.94%  
9 [95% PI: -1.02, 2.96] excess risk per 20 ppb increase in 24-h avg O<sub>3</sub>) compared to the estimates  
10 for total mortality (1.75% excess risk) and cardiovascular mortality (2.23% excess risk).

11 The analyses of a 9-year data set for the whole population of the Netherlands  
12 (population = 14.8 million) provided risk estimates for more specific causes of mortality,  
13 including COPD, pneumonia, and subcategories of cardiovascular causes (Hoek et al., 2000,  
14 2001; reanalysis Hoek, 2003). The effect estimate for total nonaccidental mortality was  
15 1.6% (95% CI: 0.9, 2.4) excess risk per 30 ppb increase in 8-h avg O<sub>3</sub>. In comparison, the  
16 excess risk estimates for pneumonia and COPD were 5.6% (95% CI: 1.8, 9.5) and 0.8% (95%  
17 CI: -2.4, 4.2), respectively. The effect estimates for some of the cardiovascular subcategories,  
18 including heart failure (3.8% [95% CI: 0.5, 7.3]) and thrombosis-related disease (6.0% [95% CI:  
19 1.1, 10.8]), showed greater risk estimates than that for total mortality. However, these  
20 associations were not specific to O<sub>3</sub>. For example, most of the pollutants examined, including  
21 PM<sub>10</sub>, BS, SO<sub>2</sub>, NO<sub>2</sub>, CO and NO<sub>3</sub><sup>-</sup>, were associated with pneumonia. Therefore, it is difficult to  
22 make a causal inference specific to O<sub>3</sub> based on these results.

23 De Leon et al. (2003) examined the role of contributing respiratory causes in the  
24 association between air pollution and nonrespiratory mortality (circulatory and cancer) in  
25 New York City during the period of 1985 to 1994. The main finding of this study was that the  
26 estimated excess mortality risks for PM<sub>10</sub> were higher for nonrespiratory deaths that had  
27 contributing respiratory causes compared to deaths without contributing respiratory causes in the  
28 older age (75+ years) group. This pattern was also seen for CO and SO<sub>2</sub>, but not for O<sub>3</sub>.  
29 Therefore, this study did not suggest a role of contributing respiratory causes in the association  
30 between O<sub>3</sub> and nonrespiratory causes of deaths.

1 In summary, several single-city studies observed positive associations between ambient O<sub>3</sub>  
2 concentrations and cardiovascular mortality. In addition, a meta-analysis that examined specific  
3 causes of mortality found that the cardiovascular mortality risk estimates were higher than those  
4 for total mortality. The findings regarding the effect size for respiratory mortality have been less  
5 consistent, possibly due to lower statistical power in this subcategory of mortality.  
6

#### 7 **7.4.8 Ozone-Mortality Risk Estimates for Specific Subpopulations**

8 Some studies examined O<sub>3</sub>-mortality risk estimates in potentially susceptible  
9 subpopulations, such as those with underlying cardiopulmonary disease. Sunyer et al. (2002)  
10 examined the association between air pollution and mortality in a cohort of patients (467 men  
11 and 611 women) with severe asthma in Barcelona, Spain during the period of 1986 to 1995.  
12 A case-crossover study design was used to estimate excess odds of mortality adjusting for  
13 weather and epidemics in three groups: (1) those who had only one asthma emergency  
14 department visit; (2) those who had more than one asthma emergency department visit; and  
15 (3) those who had more than one asthma and COPD emergency department visit. Those with  
16 more than one asthma emergency department visit showed the strongest associations with the  
17 examined air pollutants, with NO<sub>2</sub> being the most significant predictor, followed by O<sub>3</sub>. Sunyer  
18 et al. reported a significant association between O<sub>3</sub> and all cause deaths for this group during the  
19 warm season, with an odds ratio of 2.82 (95% CI: 1.15, 6.87) per 40 ppb increase in 1-h max O<sub>3</sub>,  
20 compared to an odds ratio of 1.03 (95% CI: 0.60, 1.78) for those with only one asthma  
21 emergency department visit and 1.08 (95% CI: 0.60, 1.92) for the group with a concomitant  
22 diagnosis of COPD. In another Barcelona study, Saez et al. (1999) examined asthma mortality  
23 death among persons aged 2 to 45 years. Once again, O<sub>3</sub> and NO<sub>2</sub> were the only air pollutants  
24 that were significantly associated with asthma mortality death. While the similarity of the  
25 patterns of associations between O<sub>3</sub> and NO<sub>2</sub> makes it difficult to speculate on the specific causal  
26 role of O<sub>3</sub>, the results of these studies suggest that individuals with severe asthma may make up a  
27 subpopulation that is sensitive to these pollutants.

28 Sunyer and Basagna (2001) also performed an analysis of emergency department visits by  
29 a cohort with COPD. The results from this study suggested that PM<sub>10</sub>, but not gases were  
30 associated with mortality risks for the COPD cohort. However, a Mexico City study by Téllez-

1 Rojo et al. (2000) observed a significant association between COPD mortality and O<sub>3</sub>, as well as  
2 PM<sub>10</sub>, among patients living outside a medical unit. For a cumulative 5-day lag, an excess risk of  
3 15.6% (95% CI: 4.0, 28.4) per 40 ppb increase in 1-h max O<sub>3</sub> was observed for COPD mortality.

4 Goldberg et al. (2003) investigated the association between air pollution and daily  
5 mortality with congestive heart failure as the underlying cause of death in patients aged 65 years  
6 or more in Montreal, Quebec, Canada during the period of 1984 to 1993. Analysis was stratified  
7 into two groups, those whose underlying cause of death was congestive heart failure and those  
8 with a diagnosis of congestive heart failure one year before their death. They found no  
9 association between daily mortality for congestive heart failure and any pollutants. However,  
10 they did find associations between daily mortality and coefficient of haze, SO<sub>2</sub>, and NO<sub>2</sub> among  
11 those who were classified as having congestive heart failure before death. In the case of O<sub>3</sub>,  
12 positive risk estimates were observed for year-round and warm season data; however, results  
13 were not significant. While the 10-year study period for this data was long, the daily mean death  
14 counts for the specific subcategory chosen was relatively small (0.7/day for mortality with  
15 congestive heart failure as underlying cause of death and 4.0/day for total mortality in patients  
16 previously diagnosed with congestive heart failure), limiting the power of the study.

17 In the meta-analysis by Bell et al. (2005), a combined estimate was obtained for the elderly  
18 population (age 64 years and older or 65 years and older) using 10 estimates from 9 studies. The  
19 posterior mean estimate for the elderly category (2.92% [95% PI: 1.34, 4.51] per 20 ppb  
20 increase in 24-h avg O<sub>3</sub>) was larger than that from all ages (1.75% [95% PI: 1.10, 2.37] from 41  
21 estimates). The results from this meta-analysis suggest that the elderly population may be  
22 particularly susceptible to O<sub>3</sub>-related mortality.

23 Few studies have examined O<sub>3</sub>-mortality effects for specific subpopulations. Among those  
24 that investigated the effect of air pollution in populations with underlying cardiopulmonary  
25 diseases, associations were not unique to O<sub>3</sub> but were shared with other pollutants. There is  
26 suggestive evidence that severe asthmatics may be susceptible to the mortality effects associated  
27 with NO<sub>2</sub> and O<sub>3</sub>. In addition, the meta-analysis by Bell et al. (2005) suggests that the elderly  
28 population may be more affected by O<sub>3</sub>.

## 7.4.9 Summary of Acute Ozone Effects on Mortality

- A substantial body of new data on acute mortality effects of O<sub>3</sub> has emerged since the previous O<sub>3</sub> AQCD. While uncertainties remain in some areas, it can be concluded that robust associations have been identified between various measures of daily O<sub>3</sub> concentrations and increased risk of mortality. Most of the single-pollutant model estimates from single-city studies fell in the range between 0.5 to 5% excess deaths per standardized increment. The corresponding summary estimates in large multicity studies and meta-analyses ranged between 0.5 to 2.5%, with some studies noting heterogeneity across cities and studies. These associations could not be readily explained by confounding due to time, weather, nor copollutants, but model specifications likely contributed to some of the observed heterogeneity in risk estimates across studies.
- The majority of the available O<sub>3</sub>-mortality risk estimates were computed using all year data. The results from the studies that conducted analysis by season suggest that the O<sub>3</sub> risk estimates were larger in the warm season. Some of the risk estimates in the cool season were negative, possibly reflecting the negative correlation between low-level O<sub>3</sub> and PM (and other primary pollutants) during that season. Thus, even with adjustment for temporal trends, the O<sub>3</sub> risk estimates obtained for year-round data may be misleading. In locations with considerable seasonal variation, season-specific analyses may better elucidate the effect of O<sub>3</sub> on mortality.
- The majority of the available O<sub>3</sub>-mortality risk estimates were computed for a single-day lag. Choosing the optimal lag out of several lags examined may bias the single-day risk estimate upward. However, recent findings from the largest U.S. 95 communities study indicated that a strong association between O<sub>3</sub> and mortality was observed with a 7-day distributed lag model. Thus, it is possible that the effect of acute O<sub>3</sub> exposure on mortality persists over several days. Further research is needed to understand the nature of cumulative effects.
- Some studies examined specific subcategories of mortality, but most of these studies had limited statistical power to detect associations due to the small daily mortality counts. A recent meta-analysis indicated that there was a slightly greater risk of cardiovascular mortality compared to total mortality.
- Few studies examined the effect of O<sub>3</sub> on mortality in subpopulations with underlying cardiopulmonary diseases. Similar to cause-specific mortality, these population-specific studies had limited statistical power to detect associations. The evidence suggests that individuals with severe asthma may be at increased risk of O<sub>3</sub>-related mortality; however, similar results were seen with other pollutants.

## 7.5 EFFECTS OF CHRONIC OZONE EXPOSURE

### 7.5.1 Summary of Key Findings on Studies of Health Effects and Chronic Ozone Exposure from the 1996 O<sub>3</sub> AQCD

The 1996 O<sub>3</sub> AQCD concluded that there was insufficient evidence from the limited number of studies to determine whether long-term ambient O<sub>3</sub> exposures resulted in chronic health effects. However, the aggregate evidence suggested that chronic O<sub>3</sub> exposure, along with other environmental factors, could be responsible for health effects in exposed populations.

### 7.5.2 Introduction to Morbidity Effects of Chronic Ozone Exposure

Several new longitudinal epidemiologic investigations have yielded information on health effects of long-term O<sub>3</sub> exposures. Epidemiologic interest in investigating long-term effects has been motivated by several considerations. Animal toxicology studies carried out from the late 1980s onward demonstrated that long-term exposures can result in permanent changes in the small airways of the lung, including remodeling of the airway architecture (specifically the distal airways and centriacinar region) and deposition of collagen, as discussed earlier in Chapter 5. These changes result from the damage and repair processes that occur with repeated exposure. Indices of fibrosis also were found to persist after exposure in some of the studies. Collectively, these findings provide a potential pathophysiologic basis for the changes in airway function observed in children in longitudinal studies. Seasonal ambient patterns of exposure may be of greater concern than continuous daily exposure. In the classical study by Tyler et al. (1988), young monkeys with seasonal exposure to O<sub>3</sub>, but not those with daily exposure, experienced increases in total lung collagen content, chest wall compliance, and inspiratory capacity, suggesting a delay in lung maturation in seasonally-exposed animals.

Controlled human exposure studies clearly demonstrated acute inflammation in the lung at ambient exposure levels. Epidemiologic studies could examine whether repeated exposures over multiple episode periods and/or multiple years would lead to persistent inflammation and result in damage to the human lung, especially in the small, terminal bronchiolar regions where vulnerability is greatest. However, the challenges to addressing these issues in epidemiologic studies are formidable, and as a result there exists relatively limited literature in this area. Long-term O<sub>3</sub> concentrations tend to be correlated with long-term concentrations of other pollutants, making specific attribution difficult. Subtle pulmonary effects require health outcome measures

1 that are sensitive, and must usually be directly collected from individual human subjects, rather  
2 than from administrative data bases. Although these factors make chronic studies difficult and  
3 expensive to conduct, efforts must be made to design studies with adequate power to examine  
4 the hypothesis being tested. Epidemiologic studies have the potential to provide important new  
5 insights on the links between chronic exposure to O<sub>3</sub> and the occurrence of human health effects.

6 This section reviews studies published from 1996 onward in which health effects were  
7 tested in relation to O<sub>3</sub> exposures extending from several weeks to many years (Table AX7-6 in  
8 the Chapter 7 Annex). The available literature falls into four general categories: (1) studies  
9 examining seasonal changes in lung function as related to O<sub>3</sub> exposures in peak season;  
10 (2) studies addressing smaller increases in lung function during childhood or decline of lung  
11 function beyond childhood in relation to long-term O<sub>3</sub> exposures; (3) studies addressing  
12 respiratory inflammation in high versus low exposure groups or time periods; and (4) studies  
13 addressing longitudinal and cross-sectional associations between long-term O<sub>3</sub> exposures and  
14 asthma development and prevalence.

### 16 **7.5.3 Seasonal Ozone Effects on Lung Function**

17 While it has been well-documented in both chamber and field studies that daily, multihour  
18 exposures to O<sub>3</sub> result in transient declines in lung function, much less is known about the effects  
19 of repeated exposures to O<sub>3</sub> over extended periods on lung function. Several new studies  
20 reported over the past decade have examined lung function changes over seasonal time periods  
21 with differing levels of O<sub>3</sub> exposures (Frischer et al., 1999; Horak et al., 2002a,b; Ihorst et al.,  
22 2004; Kinney and Lippmann, 2000; Kopp et al., 2000). The seasonal effects of O<sub>3</sub> are examined  
23 first in this section. In the next section is a discussion of effects over years, as opposed to over  
24 seasons, in addition to multiyear analyses of seasonal studies.

25 In a large Austrian study, Frischer et al. (1999) collected repeated lung function  
26 measurements in 1,150 school children (mean age 7.8 years) from nine towns that differed in  
27 mean O<sub>3</sub> levels. Lung function was measured in the spring and fall over a three-year period from  
28 1994 to 1996, yielding six measurements per child. Mean summertime O<sub>3</sub> exposure ranged from  
29 32.4 to 37.3 ppb during the three summers. Growth-related increases in lung function over the  
30 summer season were reduced in relation to seasonal mean O<sub>3</sub> levels. Ozone was associated with  
31 a change of -156.6 mL (95% CI: -209.5, -103.7) (central estimate: -0.029 mL/day/ppb × 90

1 days/year  $\times$  3 years  $\times$  20 ppb) in FEV<sub>1</sub> increase for each 20 ppb increase in mean 24-h avg O<sub>3</sub>  
2 concentrations over the three summers and -129.6 mL (95% CI: -193.1, -66.1) over the three  
3 winters. When analyses were restricted to children who had spent the whole summer period in  
4 their community, the changes were greater, with an O<sub>3</sub>-related -183.6 mL (95% CI: -278.9,  
5 -88.3) change in FEV<sub>1</sub> increase over three summers. Other pollutants (PM<sub>10</sub>, SO<sub>2</sub>, and NO<sub>2</sub>) had  
6 less consistent associations with changes in lung function. Horak et al. (2002a,b) extended the  
7 study of Frischer et al. (1999) with an additional year of data and stated that seasonal mean O<sub>3</sub>  
8 was associated with a negative effect on increases in lung function, confirming results from the  
9 previous three-year study. In an editorial, Tager (1999) stated that the Frischer et al. (1999) data  
10 provided the first prospective evidence of an association between exposure to ambient air  
11 pollution and alterations in lung function in children. Tager further noted that the prospective  
12 study design represented a substantial improvement over data derived from cross-sectional  
13 studies and should be emulated. However, Tager also cautioned that it was difficult to attribute  
14 the reported effects to O<sub>3</sub> alone independently of copollutants.

15 Kopp et al. (2000), in a cohort of 797 children in Austria and southwestern Germany,  
16 reported smaller increases in lung function in children exposed to high (44 to 52 ppb O<sub>3</sub>) levels  
17 of ambient O<sub>3</sub>. Children residing in low O<sub>3</sub> (24 to 33 ppb) areas experienced a 43 mL increase  
18 in FEV<sub>1</sub> whereas those in high O<sub>3</sub> areas only experienced a 16 mL increase during the summer of  
19 1994. Similar results were found in data from the summer of 1995. In another Austrian study,  
20 Ihorst et al. (2004) examined 2,153 children with a median age of 7.6 years and reported summer  
21 pulmonary function results revealing a significantly lower FVC and FEV<sub>1</sub> increase associated  
22 with higher O<sub>3</sub> exposures in the summer, but not in the winter.

23 In a pilot study (Kinney and Lippmann, 2000), 72 nonsmoking adults (mean age 20 years)  
24 from the second year class of students at the U.S. Military Academy at West Point, NY provided  
25 two lung function measurements, one before and one after a five-week long summer training  
26 program at four locations. There was a greater decline in FEV<sub>1</sub> among students at the Fort Dix  
27 location (78 mL) as compared to students at the other locations (31 mL). Ozone levels at Fort  
28 Dix averaged 71 ppb (mean of daily 1-h max O<sub>3</sub>) over the summer training period versus mean  
29 values of 55 to 62 ppb at the other three locations. In addition to the higher mean O<sub>3</sub> level, Fort  
30 Dix had greater peak O<sub>3</sub> values (23 hours >100 ppb) compared to the other locations (1 hour  
31 >100 ppb). Ambient levels of other pollutants, PM<sub>10</sub> and SO<sub>2</sub>, were relatively low during the

1 study and did not vary across the four sites. Though conclusions are limited by the small size of  
2 the study, results are consistent with a seasonal decline in lung function that may be due, in part,  
3 to O<sub>3</sub> exposures. An exploratory observation from this study was that there appeared to be a  
4 larger decline for those subjects who completed their post-summer lung function measurements  
5 in the first two weeks after returning from training compared to those measured three to four  
6 weeks after training, which is consistent with some degree of rebound of function following the  
7 summer exposure period.

8 Collectively, the above studies indicate that seasonal O<sub>3</sub> exposure is associated with  
9 smaller increases in lung function in children. The study by Kinney and Lippman (2000)  
10 provide limited evidence that seasonal O<sub>3</sub> also may affect lung function in adults, though the  
11 effect may be somewhat transient.

#### 13 **7.5.4 Chronic Ozone Exposure Effects on Lung Function and** 14 **Respiratory Symptoms**

15 Lung capacity grows during childhood and adolescence as body size increases, reaches  
16 a maximum during the 20s, and then begins to decline steadily and progressively with age.  
17 There has long been concern that long-term exposure to air pollution might lead to slower  
18 growth in lung capacity, diminished maximally attained capacity, and/or more rapid decline  
19 in capacity with age. The concern arises by analogy with cigarette smoking, where it is well-  
20 documented that lung function declines more rapidly with age in a dose-dependent manner  
21 among adults who smoke cigarettes. Adults who stop smoking return to a normal rate of decline  
22 in capacity, although there is no evidence that they regain the capacity previously lost due to  
23 smoking (Burchfiel et al., 1995). Because O<sub>3</sub> is a strong respiratory irritant and is associated  
24 with acute lung function declines as well as inflammation and re-structuring of the respiratory  
25 airways, it seems plausible that there might be a negative impact of long-term O<sub>3</sub> exposures on  
26 lung function. Exposures that negatively affect increases in lung function during childhood, in  
27 particular, might have greater long-term risks. Thus, studies of effects on the rates of increases  
28 in lung function in children are especially important.

29 Several studies published over the past decade have examined the relationship between  
30 lung function and long-term O<sub>3</sub> exposure. The most extensive and robust study of respiratory  
31 effects in relation to long-term air pollution exposures among children in the U.S. is the

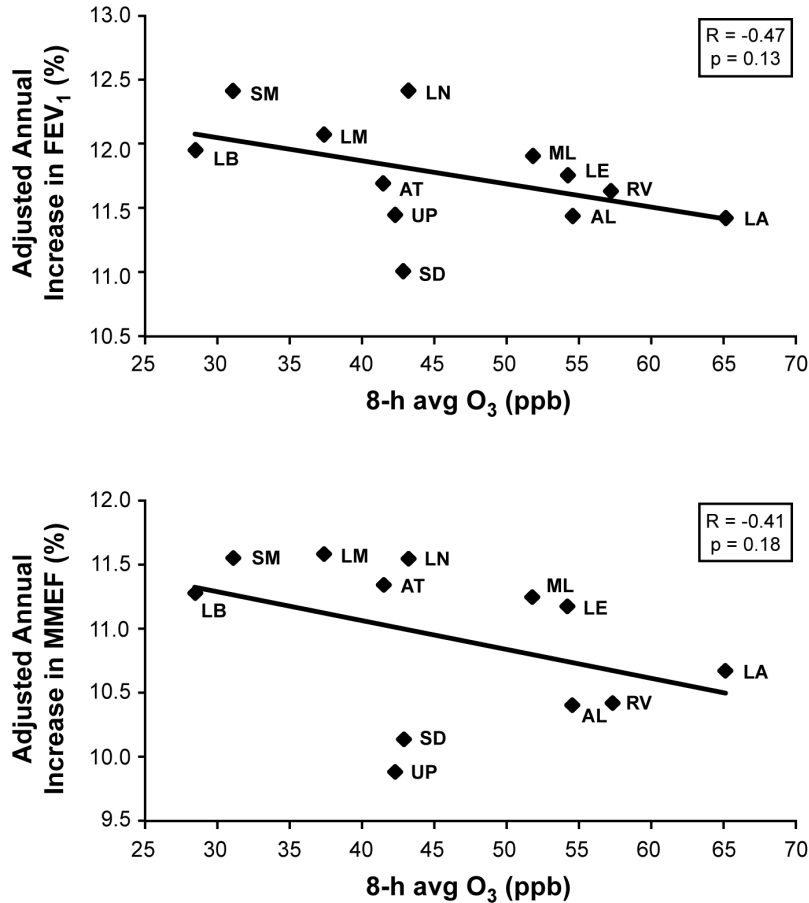


1 Children's Health Study carried out in 12 communities of southern California starting in 1993  
2 (Avol et al., 2001; Gauderman et al., 2000, 2002, 2004a,b; Peters et al., 1999a,b). The first  
3 cohort included children from the fourth, seventh, and tenth grades. A total of 3,676 students  
4 completed questionnaires regarding their lifetime residential histories, historic and current health  
5 status, residential characteristics, and physical activity. Among those students, 3,293 also  
6 performed pulmonary function tests at the time of enrollment. Peters et al. (1999a) examined the  
7 relationship between long-term (1986-1990) O<sub>3</sub> exposures and self-reports of respiratory  
8 symptoms and asthma in a cross-sectional analysis. For outcomes of current asthma, bronchitis,  
9 cough, and wheeze, the reported odds ratios were 0.95 (95% CI: 0.70,1.29), 1.14 (95% CI:  
10 0.84, 1.55), 0.98 (95% CI: 0.82, 1.17), and 1.08 (95% CI: 0.87, 1.35), respectively, per 40 ppb  
11 increase in 1-h max O<sub>3</sub>. In another cross-sectional analysis examining the relationship between  
12 lung function at baseline and levels of air pollution in the community, there was evidence that  
13 annual mean O<sub>3</sub> levels were associated with decreased FVC, FEV<sub>1</sub>, PEF, and FEF<sub>25-75</sub> (the latter  
14 two being statistically significant) among females but not males (Peters et al., 1999b).

15 Avol et al. (2001) examined 110 children from the first cohort who had moved from the  
16 participating communities in southern California to other states to determine whether changes in  
17 air quality caused by relocation were associated with changes in annual increases in lung  
18 function. With the exception of FEV<sub>1</sub>, the O<sub>3</sub> effect estimates for all other spirometric  
19 parameters were negative, but the associations were not as strong as those observed for PM<sub>10</sub>.

20 A second cohort of fourth graders (n = 1,678) were recruited in 1996 and followed over  
21 four years to examine the association between long-term exposure to air pollution and changes in  
22 lung function (Gauderman et al., 2002). In general, smaller increases in various lung function  
23 parameters were observed in communities with higher 4-year average O<sub>3</sub> levels (for examples,  
24 see Figure 7-23). The strongest effect of O<sub>3</sub> was on PEF — children from the least-polluted  
25 community had a 1.21% (95% CI: 0.36, 2.06) greater increase in PEF compared to those from  
26 the most-polluted communities. However, in the 4-year and 8-year longitudinal analysis of the  
27 first cohort, Gauderman et al. (2000, 2004) stated that the results provided little evidence that  
28 long-term exposure to ambient O<sub>3</sub> was associated with significant deficits in the growth rate of  
29 lung function in children.

30 In both cohorts of fourth graders, stratified analyses by time spent outdoors indicated a  
31 stronger association between long-term O<sub>3</sub> exposure and smaller increases in lung function in



**Figure 7-23. Adjusted average annual increases in FEV<sub>1</sub> and maximal midexpiratory flow (MMEF) versus the mean 8-h avg O<sub>3</sub> (10 a.m. to 6 p.m.) concentration over a 4-year period in the 12 southern California communities of the Children’s Health Study.**

AL = Alpine; AT = Atascadero; LA = Lake Arrowhead; LB = Long Beach; LE = Lake Elsinore; LM = Lompoc; LN = Lancaster; ML = Mira Loma; RV = Riverside; SD = San Dimas; SM = Santa Maria; UP = Upland

Source: Gauderman et al. (2002).

- 1 children who spent more time outdoors, as shown in Table 7-3 (Gauderman et al., 2002).
- 2 A study by Jedrychowski et al. (2001) found a link between repeated respiratory symptoms and
- 3 smaller lung function increases. Gauderman et al., therefore, suggested that the observation of
- 4 reduced increases in lung function with increasing annual average air pollution might be a
- 5 consequence of repeated acute respiratory events after short-term increases in pollutant levels.

**Table 7-3. Difference in Annual Percent Increases in Lung Function from the Least to the Most Polluted Community in the Children’s Health Study by Time Spent Outdoors<sup>a</sup>**

Lung Function Parameter	Cohort <sup>b</sup>	More Time Outdoors <sup>c</sup>	Less Time Outdoors <sup>c</sup>
		% Change (95% CI) <sup>d</sup>	% Change (95% CI) <sup>d</sup>
FVC	Cohort 1	-0.02% (-0.57, 0.54)	-0.04% (-0.45, 0.37)
	Cohort 2	-0.57% (-1.03, -0.09)	-0.06% (-0.76, 0.66)
FEV <sub>1</sub>	Cohort 1	-0.25% (-1.18, 0.68)	-0.05% (-0.58, 0.49)
	Cohort 2	-0.68% (-1.36, 0.00)	-0.29% (-1.02, 0.46)
MMEF	Cohort 1	-0.55% (-2.08, 1.01)	0.23% (0.89, 1.36)
	Cohort 2	-0.48% (-1.71, 0.78)	-0.80% (-2.07, 0.50)
PEF	Cohort 1	-0.77% (-2.03, 0.52)	0.25% (-0.65, 1.16)
	Cohort 2	-1.33% (-2.43, -0.24)	-0.71% (-1.71, 0.30)

<sup>a</sup> Results are derived from Gauderman et al. (2002).

<sup>b</sup> Cohort 1 includes children enrolled in 1993 as 4th graders and followed through 1997 (n = 1,457). Cohort 2 includes children enrolled in 1996 as 4th graders and followed through 2000 (n = 1,678).

<sup>c</sup> More or less time outdoors is based on reported time spent outdoors during weekday afternoons. Subjects were split into the two groups on the basis of the median reported time outdoors within each cohort.

<sup>d</sup> Percent change in lung function is per 30 ppb increase in 8-h avg O<sub>3</sub> (10 a.m.-6 p.m.).

1 The findings of larger deficits in children who spend more time outdoors in the afternoon adds  
2 some support to the possibility. Results from this study also indicate the importance of reducing  
3 exposure misclassification. Most long-term epidemiologic studies, including the Children’s  
4 Health Study, estimated O<sub>3</sub> exposure using centrally-located ambient monitors. Gonzales et al.  
5 (2003) and Künzli et al. (1997) evaluated the use of retrospective questionnaires to reconstruct  
6 past time-activity and location pattern information. Both studies found that questionnaires or  
7 activity diaries might improve the assessment of chronic exposure in epidemiologic studies.

8 In a study conducted in Austria and Germany, Ihorst et al. (2004) found that there were no  
9 associations between increases in lung function and mean summer O<sub>3</sub> levels for FVC and FEV<sub>1</sub>  
10 over a 3.5-year period, in contrast to the significant seasonal effects discussed in the earlier

1 section. Unlike the reduced increases in lung function parameters over the first two summers  
2 among children in high O<sub>3</sub> areas, a greater increase was observed during the third summer and no  
3 difference was observed during the fourth summer. The authors then concluded that medium-  
4 term effects on school children lung function were possibly present but were not detected over a  
5 3- to 5-year period due to partial reversibility. The study by Frischer et al. (1999) showed results  
6 similar to the Ihorst et al. (2004) study. Although O<sub>3</sub> was related to smaller increases in lung  
7 function when three years of data were analyzed collectively, the magnitude and direction of the  
8 effect changed throughout the years. Ozone was associated with a change of -34.0 mL (95% CI:  
9 -58.7, -9.3) in FEV<sub>1</sub> increase in the first year, compared to +7.3 mL (95% CI: -20.8, 35.6) in  
10 the third year for each 20 ppb increase in mean 24-h avg O<sub>3</sub> (Frischer et al., 1999).

11 Calderón-Garcidueñas et al. (2003) examined chest X-rays and lung function in  
12 174 children from Mexico City and 27 control children from low pollution areas (Tuxpam and  
13 Tlaxcala, Mexico). The Mexico City children exhibited lung hyperinflation (67%), interstitial  
14 markings (49%), and a mild restrictive pattern by spirometry (10%). In children with increased  
15 interstitial markings, FEF<sub>75</sub> values were significantly declined ( $r = 0.42$ ,  $p < 0.003$ ).  
16 No significant abnormalities were observed in the control children. In another study of similar  
17 design, the prevalence of respiratory symptoms was higher in Mexico City children compared to  
18 children from the clean coastal town of Manzanillo, Mexico (Calderón-Garcidueñas et al., 1999).

19 A study by Gong et al. (1998b) examined lung function changes in 164 nonsmoking adults  
20 (mean age 45 years) from a high O<sub>3</sub> community in southern California, recruited from a cohort  
21 of 208 who had been tested on two previous occasions. In the earlier analysis by Detels et al.  
22 (1987), a significant decline in lung function was observed from 1977/1978 to 1982/1983. In  
23 contrast, Gong et al. observed a slight increase in FVC and FEV<sub>1</sub> from 1982/1983 to 1986/1987.  
24 A consistent decline in FEV<sub>1</sub>/FVC ratio was observed at all three time points ( $p < 0.0001$ ).  
25 Among the 45 subjects who further participated in the controlled exposure study (0.40 ppm O<sub>3</sub>  
26 over 2 hours with intermittent exercise), acute changes in lung function were not associated with  
27 long-term changes in lung function over a decade.

28 Evidence for a relationship between long-term O<sub>3</sub> exposures and decrements in maximally  
29 attained lung function was observed in a nationwide cohort of 520 first year students at Yale  
30 College in New Haven, CT (Galizia and Kinney 1999; Kinney et al., 1998). Each student  
31 performed one lung function test in the spring of their first year at college. Ozone exposures

1 were estimated by linking 10-year mean summer season 1-h max O<sub>3</sub> levels at the nearest  
2 monitoring station to the residential locations reported each year from birth to the time of  
3 measurement. Students who had lived four or more years in areas with long-term mean O<sub>3</sub> levels  
4 above 80 ppb had significantly lower FEV<sub>1</sub> (-3.07% [95% CI: -0.22, -5.92]) and FEF<sub>25-75</sub>  
5 (-8.11% [95% CI: -2.32, -13.90]) compared to their classmates with lower long-term O<sub>3</sub>  
6 exposures. Stratification by gender indicated that males had much larger effect estimates than  
7 females, which might reflect higher outdoor activity levels and corresponding higher O<sub>3</sub>  
8 exposure during childhood.

9 A similar study of 130 first year college freshmen at the University of California at  
10 Berkeley also reported significant effects of O<sub>3</sub> on lung function (Künzli et al., 1997; Tager  
11 et al., 1998). Enrollment was limited to students from either the San Francisco or Los Angeles,  
12 CA metropolitan areas. After controlling for city of origin, long-term O<sub>3</sub> exposures were  
13 associated with declines in FEF<sub>25-75</sub> and FEF<sub>75</sub> (forced expiratory flow after 75% of FVC has  
14 been exhaled). No effects were seen for PM<sub>10</sub> and NO<sub>2</sub>. Künzli and colleagues noted that  
15 significant changes in these mid- and end-expiratory flow measures could be considered early  
16 indicators for pathologic changes that might ultimately progress to COPD, as evidenced by  
17 animal studies that showed that the primary site of O<sub>3</sub> injury in the lung was the centriacinar  
18 region (Chapter 5).

19 Sherwin et al. (2000) examined lungs from autopsies of young residents in Miami, FL and  
20 Los Angeles, CA for centriacinar region inflammatory diseases. A trend towards greater degrees  
21 of centriacinar region alterations was observed in the lungs of Los Angeles residents compared  
22 to Miami residents, independent of a smoking effect. The results suggest that the greater extent  
23 and severity of centriacinar region alterations might be related to the higher O<sub>3</sub> levels in Los  
24 Angeles. Beyond the challenge of differentiating the lifetime of exposure for subjects in the two  
25 cities, various confounding factors also can impact this study. The pathogenesis of centriacinar  
26 region alteration is undoubtedly multifactorial with respiratory infection and adverse  
27 environmental influences being two major considerations. In addition, Sherwin et al. (2000)  
28 noted that the study was limited due to the relatively small number of cases available.  
29 Nonetheless, as observed by Tager (1993), the use of human postmortem specimens is of interest  
30 in future epidemiologic studies.

1 The results of the southern California Children’s Health Study, as well as those from the  
2 European studies, provide little evidence for impacts of long-term O<sub>3</sub> exposures on lung function  
3 in children. However, further study is needed to better address this difficult question. There is  
4 limited evidence that young adults who grew up in high O<sub>3</sub> communities may have reduced lung  
5 function compared to those from low O<sub>3</sub> communities.  
6

### 7 **7.5.5 Chronic Ozone Exposure and Respiratory Inflammation**

8 As noted in Chapter 6, human chamber studies have demonstrated that brief (2 to  
9 6.6 hours) exposures to O<sub>3</sub> while exercising result in inflammation in the lung, including the  
10 alveolar region where gas exchange takes place. This acute exposure effect is potentially  
11 important for effects of chronic exposure because repeated inflammation can result in the release  
12 of substances from inflammatory cells that can damage the sensitive cells lining the lung. Over  
13 extended periods, repeated insults of this kind can lead to permanent damage to and restructuring  
14 of the small airways and alveoli. In addition, since inflammation is a fundamental feature of  
15 asthma, there is concern that O<sub>3</sub>-induced inflammation can exacerbate existing asthma or  
16 perhaps promote the development of asthma among genetically pre-disposed individuals.  
17 Several studies are discussed next, examining different outcomes related to inflammation.

18 In a study by Kinney et al. (1996b), bronchoalveolar lavage fluids were collected in the  
19 summer and winter from a group of 19 adult joggers living and working on an island in  
20 New York harbor. The mean 1-h max O<sub>3</sub> concentrations for a 3-month period were 58 ppb  
21 (maximum 110) in the summer and 32 ppb (maximum 64) in the winter. PM<sub>10</sub> and NO<sub>2</sub>  
22 concentrations were similar across the two seasons. There was little evidence for acute  
23 inflammation in bronchoalveolar lavage fluids collected during the summer as compared to that  
24 collected from the same subjects in the winter. However, there was evidence of enhanced cell  
25 damage, as measured by lactate dehydrogenase, in the summer lavage fluids. These results  
26 indicate that acute inflammatory responses may diminish with repeated exposures over the  
27 course of a summer (which have been demonstrated in multiday chamber exposures, Chapter 6,  
28 Section 6.9) but cell damage may be ongoing.

29 Pollution effects in the nose can be viewed as a potential surrogate measure for effects that  
30 may occur in the lungs, though doses to nasal tissues are usually higher for a given pollutant  
31 concentration. In Chapter 5, morphological effects of O<sub>3</sub> on the upper respiratory tract indicated

1 quantitative changes in the nasal transitional respiratory epithelium. The persistent nature of the  
2 O<sub>3</sub>-induced mucous cell metaplasia in rats, as discussed in Chapter 5, suggests that O<sub>3</sub> exposure  
3 may have the potential to induce similar long-lasting alterations in the airways of humans.  
4 A series of interesting studies in Mexico City have demonstrated inflammation and genetic  
5 damage to cells in the nasal passages of children chronically exposed to O<sub>3</sub> and other air  
6 pollutants (Calderón-Garcidueñas et al., 1995, 1997, 1999, 2001, 2003). Nasal lavage samples  
7 and nasal biopsies from children living in Mexico City were compared to those from children  
8 living in a clean coastal town with no detectable air pollutants. In the first study, urban children  
9 (n = 38) from Mexico City were found to have significantly higher polymorphonuclear leukocyte  
10 counts and abnormal nasal cytologies compared to nonurban children (n = 28) (Calderón-  
11 Garcidueñas et al., 1995). A more recent study of similar design examined nasal abnormalities  
12 and serum cytokines in both urban and nonurban children (Calderón-Garcidueñas et al., 2003).  
13 Twenty-two percent of the 112 Mexico City children showed a grossly abnormal nasal mucosa.  
14 No significant abnormalities were observed in the control children. In addition, the Mexico City  
15 children had more serum interleukin-10 and interleukin-6, and less serum interleukin-8 than  
16 controls. Twenty-five children with whitish-gray nasal lesions showed a significant association  
17 between tumor necrosis factor  $\alpha$  and interleukin-8 ( $r = 0.89$ ,  $p < 0.0001$ ), which suggested the  
18 potential importance of the nose in the production of proinflammatory cytokines.

19 Calderón-Garcidueñas et al. (1997) also observed that cells collected from the lining of the  
20 nose had significantly higher amounts of DNA damage in the urban children in Mexico City  
21 (n = 129) versus nonurban children (n = 19). Among exposed children, the extent of  
22 DNA damage was greater in older children, who had spent more time outdoors and were more  
23 engaged in physical activities compared to the younger children. Another study of 86 urban and  
24 12 nonurban children reported similar findings, and also noted increased levels of specific DNA  
25 mutations (Calderón-Garcidueñas et al., 1999). Fortoul et al. (2003) examined DNA strand  
26 breaks in nasal epithelial cells from asthmatic and nonasthmatic medical students in Mexico City  
27 and noted greater genotoxic damage in asthmatics. These results indicate that asthmatics may  
28 have a greater susceptibility for DNA damage, or a decreased ability to repair it, compared to  
29 nonasthmatic subjects. However, because of the complex mixture of pollutants present in  
30 Mexico City, it is not possible to uniquely attribute these observed changes to O<sub>3</sub> concentrations.

31

1 Another outcome of inflammation was examined in a study by Frischer et al. (2001).  
2 In this cross-sectional study, urinary eosinophil protein was analyzed as a marker of eosinophil  
3 activation in 877 school children living in nine Austrian communities with varying O<sub>3</sub> exposure.  
4 The results indicated that O<sub>3</sub> exposure was significantly associated with eosinophil  
5 inflammation.

6 In the Mexico City studies, specific attribution of these adverse respiratory and genotoxic  
7 effects to O<sub>3</sub> is difficult given the complex pollutant mixture present in the ambient air.  
8 In particular, the DNA effects seem more plausibly related to other components of urban air,  
9 such as semi-volatile organic compounds. However, the inflammatory changes such as  
10 increased eosinophil levels observed in the Austrian study would be consistent with known  
11 effects of O<sub>3</sub>.

### 13 **7.5.6 Risk of Asthma Development**

14 Recent longitudinal cohort studies have reported associations between the onset of asthma  
15 and long-term O<sub>3</sub> exposures (McConnell et al., 2002; McDonnell et al., 1999). Significant  
16 associations between new cases of asthma among adult males and long-term O<sub>3</sub> exposure were  
17 observed in a cohort of nonsmoking adults in California (Greer et al., 1993; McDonnell et al.,  
18 1999). The Adventist Health and Smog (AHSMOG) study cohort of 3,914 (age 27-87 years,  
19 36% male) was drawn from nonsmoking, non-Hispanic white California Seventh Day  
20 Adventists. Subjects were surveyed in 1977, 1987, and 1992. To be eligible, subjects had to  
21 have lived 10 or more years within 5 miles of their current residence in 1977. Residences from  
22 1977 onward were followed and linked in time and space to interpolate concentrations of O<sub>3</sub>,  
23 PM<sub>10</sub>, SO<sub>2</sub>, and NO<sub>2</sub>. New asthma cases were defined as self-reported doctor-diagnosed asthma  
24 at either the 1987 or 1992 follow-up questionnaire among those who had not reported having  
25 asthma upon enrollment in 1977. During the 10-year follow-up (1977-1987), the incidence of  
26 new asthma was 2.1% for males and 2.2% for females (Greer et al., 1993). A relative risk of  
27 3.12 (95% CI: 1.16, 5.85) per 10 ppb increase in annual mean O<sub>3</sub> (exposure metric not stated)  
28 was observed in males, compared to a relative risk of 0.94 (95% CI: 0.65, 1.34) in females.  
29 In the 15-year follow-up study (1977-1992), 3.2% of the eligible males and a slightly greater  
30 4.3% of the eligible females developed adult asthma (McDonnell et al., 1999). For males, the  
31 relative risk of developing asthma was 2.27 (95% CI: 1.03, 4.87) per 30 ppb increase in 8-h avg



1 O<sub>3</sub> (9 a.m.-5 p.m.). Once again, there was no evidence of an association between O<sub>3</sub> and new-  
2 onset asthma in females (relative risk of 0.85 [95% CI: 0.55, 1.29]). The lack of an association  
3 does not necessarily indicate no effect of O<sub>3</sub> on the development of asthma among females.  
4 For example, differences in time-activity patterns in females and males may influence relative  
5 exposures to O<sub>3</sub>, leading to greater misclassification of exposure in females. The consistency of  
6 the results in the two studies with different follow-up times and indices of O<sub>3</sub> exposure provides  
7 supportive evidence that long-term O<sub>3</sub> exposure may be associated with asthma incidence in  
8 adult males. However, as the AHSMOG cohort was drawn from a narrow subject definition, the  
9 representativeness of this cohort to the general U.S. population may be limited.

10 A similar study of incident asthma cases in relation to O<sub>3</sub> among children was carried out  
11 in the Children's Health Study (McConnell et al., 2002). 3,535 initially nonasthmatic children  
12 (ages 9 to 16 years at enrollment) were followed for up to 5 years to identify new-onset asthma  
13 cases. Communities were stratified by pollution levels, with six high-O<sub>3</sub> communities (mean 1-h  
14 max O<sub>3</sub> of 75.4 ppb [SD 6.8] over four years) and six low-O<sub>3</sub> communities (mean 50.1 ppb  
15 [SD 11.0]). A total of 265 children reported a new diagnosis of asthma during the follow-up  
16 period. Asthma risk was not higher for residents of the six high-O<sub>3</sub> communities versus residents  
17 of the six low-O<sub>3</sub> communities. However, within the high-O<sub>3</sub> communities, asthma risk was  
18 3.3 (95% CI: 1.9, 5.8) times greater for children who played three or more sports as compared  
19 with children who played no sports. This association was absent in the low-O<sub>3</sub> communities  
20 (relative risk of 0.8 [95% CI: 0.4, 1.6]). No associations with asthma were seen for PM<sub>10</sub>, PM<sub>2.5</sub>,  
21 NO<sub>2</sub>, or inorganic acid vapors. These results suggest effect modification of the impacts of O<sub>3</sub> on  
22 asthma risk by physical activity. Playing sports may indicate outdoor activity when O<sub>3</sub> levels are  
23 higher and an increased ventilation rate, which may lead to increased O<sub>3</sub> exposure. It should be  
24 noted, however, that these findings were based on a small number of new asthma cases (n = 29  
25 among children who played three or more sports) and it is not clear to what extent the key  
26 findings were based on a priori hypotheses. Replication of these findings in other cohorts would  
27 lend greater weight to a causal interpretation.

28 Recent cross-sectional surveys have detected no associations between long-term O<sub>3</sub>  
29 exposures and asthma prevalence, asthma-related symptoms, or allergy to common aeroallergens  
30 in children after controlling for covariates (Charpin et al., 1999; Kuo et al., 2002; Ramadour  
31 et al., 2000). It should be noted that O<sub>3</sub> levels were quite low in all cases, with a range of 16 to

1 27 ppb for 8-h max O<sub>3</sub>. The longitudinal study design, which observes new onset of asthma  
2 prospectively, provides stronger evidence on the question of asthma development.

### 3 4 **7.5.7 Respiratory Effects of Chronic Ozone Exposure on Susceptible** 5 **Populations**

6 Studies on the effect of long-term O<sub>3</sub> exposure on respiratory health has focused mostly on  
7 children, a potentially susceptible population. Ozone exposure was associated with smaller  
8 increases in lung function and respiratory inflammation in children. Other studies have  
9 investigated additional groups of potentially susceptible individuals. McConnell et al. (1999)  
10 examined the association between O<sub>3</sub> levels and the prevalence of chronic lower respiratory tract  
11 symptoms in southern California children with asthma (n = 3,676). In this cross-sectional study,  
12 bronchitis, phlegm, and cough were not associated with annual mean O<sub>3</sub> concentrations in  
13 children with asthma or wheeze. All other pollutants examined, PM<sub>10</sub>, PM<sub>2.5</sub>, NO<sub>2</sub>, and gaseous  
14 acid, was associated with an increase in phlegm, but not cough.

15 In another analysis from the Children's Health Study, McConnell et al. (2003) investigated  
16 the relationship between air pollutants and bronchitic symptoms among 475 children with  
17 asthma. For a 1 ppb increase in 8-h avg O<sub>3</sub> concentrations averaged over 4 years, the between-  
18 community odds ratio was 0.99 (95% CI: 0.98, 1.01) compared to the within-community odds  
19 ratio of 1.06 (95% CI: 1.00, 1.12). The authors commented that if the larger within-community  
20 effect estimates were correct, then other cross-sectional (between-community) studies might  
21 have underestimated the true effect of air pollution on bronchitic symptoms in children. These  
22 differences might be attributable to confounding by poorly measured or unmeasured risk factors  
23 that vary between communities. In two-pollutant models, the within-community effect estimates  
24 for O<sub>3</sub> were markedly reduced and in some cases no longer significant (odds ratios not provided).  
25 However, given the high correlation between O<sub>3</sub> and the other pollutants, a causal role for O<sub>3</sub>  
26 should not be excluded.

27 One recent study examined a susceptible group not examined before. Goss et al. (2004)  
28 investigated the effect of O<sub>3</sub> on pulmonary exacerbations and lung function in individuals with  
29 cystic fibrosis over the age of 6 years (n = 11,484). The study included patients enrolled in the  
30 Cystic Fibrosis Foundation National Patient Registry. The registry contained demographic and  
31 clinical data collected annually at accredited centers for cystic fibrosis. In 1999 and 2000, the

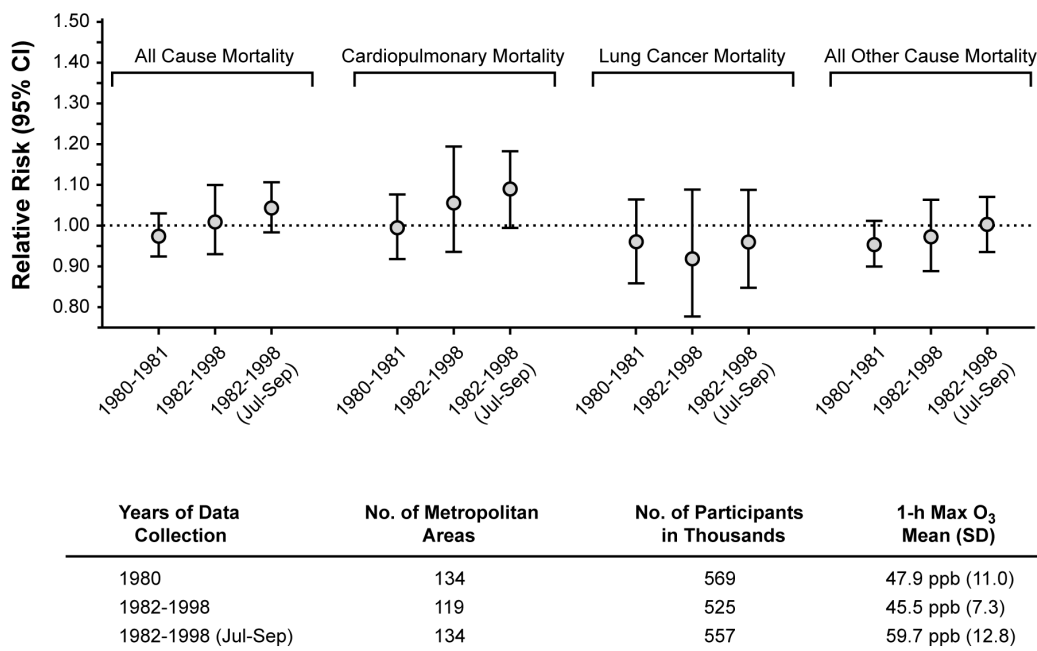
1 annual mean O<sub>3</sub> concentration from 616 monitors in the U.S. EPA Aerometric Information  
2 Retrieval System (AIRS) was 51.0 ppb (SD 7.3). Exposure was assessed by linking air pollution  
3 values from AIRS with the patient's home zip code. No clear association was found between  
4 annual mean O<sub>3</sub> and lung function parameters. However, a 40 ppb increase in annual mean 1-h  
5 max O<sub>3</sub> was associated with a 46% (95% CI: 13, 87) increase in the odds of two or more  
6 pulmonary exacerbations. Significant excess odds of pulmonary exacerbations also were  
7 observed with increased annual mean PM<sub>10</sub> and PM<sub>2.5</sub> concentrations.

8 In summary, some studies have identified and investigated potentially susceptible  
9 populations. Although effects are not specific to O<sub>3</sub> exposure, the results suggest that O<sub>3</sub> may  
10 contribute to the adverse respiratory health responses observed in individuals with asthma and  
11 cystic fibrosis.

### 13 **7.5.8 Effects of Chronic Ozone Exposure on Mortality and Cancer Incidence**

14 There is inconsistent and inconclusive evidence for a relationship between long-term O<sub>3</sub>  
15 exposure and increased mortality and cancer risk (see Table AX7-7 in the Chapter 7 Annex).  
16 In a large prospective cohort study of approximately 500,000 U.S. adults, Pope et al. (2002)  
17 examined the effects of long-term exposure to air pollutants on mortality. All cause,  
18 cardiopulmonary, lung cancer, and all other cause mortality risk estimates for long-term O<sub>3</sub>  
19 exposure are shown in Figure 7-24. Consistent positive associations were not observed between  
20 O<sub>3</sub> and mortality. The mortality risk estimates were larger when analyses were restricted to the  
21 summer months (July to September) when O<sub>3</sub> levels were generally higher. The O<sub>3</sub>-mortality  
22 risk estimates were positive for all cause and cardiopulmonary mortality, with a marginally  
23 significant estimate for cardiopulmonary mortality in the summer months. A negative,  
24 nonsignificant O<sub>3</sub> risk estimate was observed for lung cancer mortality. Consistent positive and  
25 significant effects of PM<sub>2.5</sub> were observed for both lung cancer and cardiopulmonary mortality.

26 Lipfert et al. (2000b, 2003) reported positive effects on all cause mortality for peak O<sub>3</sub>  
27 exposures (95th percentile levels) in the U.S. Veterans Cohort study of approximately 50,000  
28 male middle-aged men recruited with a diagnosis of hypertension. The actual analysis involved  
29 smaller subcohorts based on exposure and mortality follow-up periods. Four separate exposure  
30 periods were associated with three mortality follow-up periods. In a preliminary screening of  
31 regression results, Lipfert et al. (2000b) observed a negative association for mean O<sub>3</sub> and a



**Figure 7-24. Adjusted O<sub>3</sub>-mortality relative risk estimates (95% CI) by cause of mortality and time period of analysis per subject-weighted mean O<sub>3</sub> concentration in the Cancer Prevention Study II by the American Cancer Society.**

Source: Derived from Pope et al. (2002).

1 positive relationship for peak O<sub>3</sub>; thus, peak O<sub>3</sub> was used in subsequent analyses. The mean of  
 2 the peak values ranged from 85 to 140 ppb over the four exposure periods. For concurrent  
 3 exposure periods, peak O<sub>3</sub> was positively associated with all cause mortality, with a 9.4% (95%  
 4 CI: 0.4, 18.4) excess risk per mean 95th percentile O<sub>3</sub> less estimated background level (not  
 5 stated). When exposure periods preceding death were considered, no association between O<sub>3</sub>  
 6 and mortality was observed (-0.2% [95% CI: -12.5, 12.1]). In a further analysis, Lipfert et al.  
 7 (2003) reported the strongest positive association for concurrent exposure to peak O<sub>3</sub> for the  
 8 subset with low diastolic blood pressure during the period of 1982-1988. Once again, the O<sub>3</sub>  
 9 effect was diminished when exposure (1982-1988) preceded mortality (1989-1996).

10 A long-term prospective cohort study (AHSMOG) of 6,338 nonsmoking, non-Hispanic  
 11 white individuals living in California examined the association between air pollutants and lung  
 12 cancer incidence (Beeson et al., 1998). Over the follow-up period of 1977 to 1992, 20 females

1 (35% smokers, n = 7) and 16 males (37.5% smokers, n = 6) developed lung cancer.  
2 An association was observed between long-term O<sub>3</sub> exposure and increased incidence of lung  
3 cancer in males only. The relative risk for incident lung cancer among males was 3.56 (95% CI:  
4 1.35, 9.42) for an interquartile range increase in hours per year (556 hours/year) when O<sub>3</sub> levels  
5 exceeded 100 ppb (Beeson et al., 1998). A stronger association was observed in males who  
6 never smoked (4.48 [95% CI: 1.25, 16.04]) compared to those who smoked in the past (2.15  
7 [95% CI: 0.42, 10.89]) (Beeson et al., 1998).

8 A related study by Abbey et al. (1999) examined the effects of long-term O<sub>3</sub> exposure on  
9 all cause (n = 1,575), cardiopulmonary (n = 1,029), nonmalignant respiratory (n = 410), and lung  
10 cancer (n = 30) mortality in the same AHSMOG study population. A particular strength of this  
11 study was the extensive effort devoted to assessing long-term air pollution exposures, including  
12 interpolation to residential and work locations from monitoring sites over time and space.  
13 No associations with long-term O<sub>3</sub> exposure were observed for all cause, cardiopulmonary, and  
14 nonmalignant respiratory mortality. However, effects of O<sub>3</sub> on lung cancer mortality confirmed  
15 the results of the previous study by Beeson and colleagues. An association between lung cancer  
16 mortality and chronic O<sub>3</sub> exposure was observed in males only, with a relative risk of 4.19 (95%  
17 CI: 1.81, 9.69) (Abbey et al., 1999). The gender-specific O<sub>3</sub> effects may be partially attributable  
18 to the differences in activity and time spent outdoors. The questionnaires indicated that males  
19 spent approximately twice as much time outdoors and performed more vigorous exercises  
20 outdoors, especially during the summer, compared to the females. However, the very small  
21 numbers of lung cancer deaths (n = 12 for males) raise concerns with regard to the precision of  
22 the effect estimate, as evidenced by the wide confidence intervals. The lack of an association of  
23 chronic O<sub>3</sub> exposure with other mortality outcomes, which had much larger samples sizes, also is  
24 of concern. A study by Pereira et al. (2005) provides supportive evidence of an association  
25 between O<sub>3</sub> and increase risk of cancer. The correlation between average air pollution data from  
26 1981 to 1990 and cases of larynx and lung cancer in 1997 were assessed in communities of  
27 São Paulo, Brazil. Of all the pollutants examined (PM<sub>10</sub>, NO<sub>2</sub>, NO<sub>x</sub>, SO<sub>2</sub>, CO, and O<sub>3</sub>), O<sub>3</sub>  
28 was best correlated with cases of larynx (r = 0.9929, p = 0.007) and lung cancer (r = 0.7234,  
29 p = 0.277).

30 Few studies have examined the effect of chronic O<sub>3</sub> exposure on mortality outcomes and  
31 incidence of cancer. Consistent associations with long-term O<sub>3</sub> exposure were not observed for

1 all cause and cardiopulmonary mortality. There is limited evidence supportive of an association  
2 between O<sub>3</sub> exposure and lung cancer incidence and mortality; however, the small number of  
3 lung cancer cases, differential effects by gender, and the lack of O<sub>3</sub> effects on other mortality  
4 outcomes raise concerns regarding plausibility.  
5

### 6 **7.5.9 Effects of Ozone on Birth-Related Health Outcomes**

7 In recent years, air pollution epidemiologic studies have examined impacts on birth-related  
8 endpoints including intrauterine, perinatal, postneonatal, and infant deaths; premature births;  
9 intrauterine growth retardation; very low birth weight (weight <1500 grams) and low birth  
10 weight (weight <2500 grams); and birth defects. However, the majority of these studies did not  
11 examine the effect of O<sub>3</sub>. In the limited studies that investigated O<sub>3</sub>, no associations were  
12 observed between O<sub>3</sub> and birth outcomes, with the exception of birth defects. The following is a  
13 synopsis of the literature on this topic.

14 Pereira et al. (1998) investigated impacts of air pollution on intrauterine mortality in São  
15 Paulo, Brazil during 1991 and 1992. NO<sub>2</sub>, SO<sub>2</sub>, CO, O<sub>3</sub>, and PM<sub>10</sub> were examined. Intrauterine  
16 mortality was most significantly associated with NO<sub>2</sub>, and less for SO<sub>2</sub> and CO. No association  
17 was found for O<sub>3</sub> or PM<sub>10</sub>. Pereira et al. also sampled blood from the umbilical cord of healthy  
18 non-smoking pregnant women soon after delivery in 1995 and analyzed for levels of  
19 carboxyhemoglobin. They found an association between carboxyhemoglobin and ambient CO  
20 after adjusting for passive smoking and weight, suggesting an impact of CO on the fetus.  
21 Loomis et al. (1999) examined the association between air pollutants and infant mortality in  
22 Mexico City in the years 1993 to 1995. NO<sub>2</sub>, SO<sub>2</sub>, CO, O<sub>3</sub>, and PM<sub>2.5</sub> were examined. They  
23 reported that the strongest association was found for PM<sub>2.5</sub> with a 3- to 5-day cumulative lag.  
24 They noted that infant mortality also was associated with NO<sub>2</sub> and O<sub>3</sub> at a 3- to 5-day lag, but not  
25 as consistently as PM<sub>2.5</sub>. There have been air pollution studies that examined postneonatal  
26 mortality (Bobak and Leon, 1992; Bobak and Leon, 1999; Kaiser et al., 2004; Woodruff et al.,  
27 1997), but these studies did not examine O<sub>3</sub>.

28 Ritz and Yu (1999) investigated the effects of ambient CO on low birth weight among  
29 children born in southern California between 1989 and 1993. They focused on CO because  
30 “a biologic mechanism for fetal effects has been proposed for CO, but not for other air

1 pollutants.” They found that exposure to higher levels of ambient CO during the last trimester  
2 was associated with an increased risk for low birth weight. Using available data, they also  
3 estimated the last-trimester exposures for NO<sub>2</sub>, O<sub>3</sub>, and PM<sub>10</sub>. NO<sub>2</sub> and PM<sub>10</sub> were positively  
4 associated with CO, but O<sub>3</sub> was negatively associated with CO (r = -0.65). Ha et al. (2001;  
5 a GAM analysis using default convergence parameters) examined CO, NO<sub>2</sub>, SO<sub>2</sub>, O<sub>3</sub>, and TSP  
6 for their associations with low birth weight in Seoul, Korea during the years 1996 and 1997.  
7 They estimated first and third trimester exposures by averaging daily air pollution levels during  
8 the corresponding days for the registered births. Ha et al. found that first trimester exposures of  
9 CO, NO<sub>2</sub>, SO<sub>2</sub>, and TSP were associated with increased risk of low birth weight, whereas O<sub>3</sub> was  
10 associated with a decreased risk. The opposite pattern was observed for third trimester  
11 exposures, with an increased risk of low birth weight found only for O<sub>3</sub>. When exposures from  
12 both trimesters were examined simultaneously, the associations of first trimester exposures of  
13 CO, NO<sub>2</sub>, SO<sub>2</sub>, and TSP with increased risk of low birth weight remained; however, the  
14 association between third trimester O<sub>3</sub> exposure and low birth weight was diminished. Based on  
15 these results, Ha et al. concluded that exposures to CO, NO<sub>2</sub>, SO<sub>2</sub>, and TSP in the first trimester  
16 were risk factors for low birth weight. Note that neither of these studies examined the air  
17 pollution effect by season. Other studies that examined the associations between air pollution  
18 and low birth weight (Bobak, 2000; Bobak and Leon, 1999; Lin et al., 2001; Maisonet et al.,  
19 2001; Wang et al., 1997) did not examine O<sub>3</sub> data, and found associations between low birth  
20 weight and either one or more of CO, SO<sub>2</sub>, NO<sub>2</sub> and PM indices. Collectively, these results do  
21 not indicate strong evidence of the role of O<sub>3</sub> in low birth weight.

22 Two studies by Dejmek et al. (1999; 2000) examined the relationship between ambient air  
23 pollution and risk of intrauterine growth retardation in a highly polluted area of Northern  
24 Bohemia (Teplice District). Both studies, however, focused on PM indices and did not analyze  
25 gaseous pollutants.

26 A few studies have examined the association between air pollution and premature births  
27 (Bobak, 2000; Ritz et al., 2000; Xu et al., 1995), but only Ritz et al. (2000) included O<sub>3</sub> in their  
28 analysis. Ritz et al. evaluated the effect of air pollution exposure during pregnancy on the  
29 occurrence of preterm birth in a cohort of 97,518 neonates born in southern California. CO, NO<sub>2</sub>  
30 SO<sub>2</sub>, O<sub>3</sub>, and PM<sub>10</sub> data measured at 17 air quality monitoring stations were used to estimate the

1 average exposures for the first month and the last 6 weeks of pregnancy. They found  
2 associations between PM<sub>10</sub> levels averaged for the last 6 weeks of pregnancy as well as PM<sub>10</sub>  
3 levels averaged over the first month of pregnancy. Similar but weaker associations were found  
4 for CO. No association was found for O<sub>3</sub>. The reported correlation matrix indicated that O<sub>3</sub> was  
5 negatively correlated with CO ( $r = -0.45$ ) and only weakly correlated with PM<sub>10</sub> ( $r = 0.2$ ). The  
6 results from Beijing, China (Xu et al., 1995) and the Czech Republic (Bobak, 2000) suggested  
7 that SO<sub>2</sub> and TSP were associated with preterm births. Considering that O<sub>3</sub> tends to be  
8 negatively correlated with winter-type pollutants, O<sub>3</sub> is unlikely to be an important risk factor for  
9 preterm births.

10 Ritz et al. (2002) evaluated the effect of air pollution on the occurrence of birth defects in  
11 neonates and fetuses delivered in southern California from 1987 to 1993 as ascertained by the  
12 California Birth Defects Monitoring Program. They averaged air pollution (CO, O<sub>3</sub>, PM<sub>10</sub>, and  
13 NO<sub>2</sub>) levels measured at the assigned ambient station over the first, second, and third month of  
14 gestation. Conventional, polytomous, and hierarchical logistic regressions were used to estimate  
15 odds ratios for subgroups of cardiac and orofacial defects. Concentration-response relationships  
16 were observed for second month CO exposure on ventricular septal defects, and second month  
17 O<sub>3</sub> exposure on aortic artery and valve defects, pulmonary artery and valve anomalies, and  
18 conotruncal defects. The odds ratios observed for these outcomes were similar and quite large  
19 (e.g., the odds ratios comparing the highest [monthly 24-h avg mean 34.9 ppb] to lowest  
20 [monthly mean 6.4 ppb] O<sub>3</sub> quartiles ranged from 2.0 to 2.7), and were not sensitive in  
21 multipollutant models. Ritz et al. reported that they did not observe consistently increased risks  
22 and concentration-response patterns for NO<sub>2</sub> and PM<sub>10</sub> after controlling for the effects of CO and  
23 O<sub>3</sub>. Results from this study were in contrast to those with other birth-related outcomes in that  
24 both CO and O<sub>3</sub>, presumably negatively correlated pollutants, were associated with birth defects.  
25 Further, O<sub>3</sub> showed associations with more birth defect outcomes compared to CO. It should be  
26 noted, however, that the concentration-response relationships were quite specific to exposures  
27 during the second month. Associations with third month exposures were often negative (though  
28 not significantly). Since both CO and O<sub>3</sub> show strong seasonal peaks, it is possible that seasonal  
29 confounding could have played some role in these associations. This is the only study to date  
30 that examined the relationship between air pollution and birth defects.



1 In summary, O<sub>3</sub> was not an important predictor of birth-related outcomes including  
2 intrauterine and infant mortality, premature births, and low birth weight. Birth-related outcomes  
3 generally appear to be associated with air pollutants that tend to peak in the winter and are  
4 possibly traffic-related, most notably CO. The strong results for CO are consistent with its  
5 ability to cross the placental barrier and the high affinity that hemoglobin in fetal blood has for  
6 binding with it. However, most of these studies did not analyze the data by season, and therefore  
7 seasonal confounding may have influenced the reported associations. One study reported  
8 associations between exposures to O<sub>3</sub> in the second month of pregnancy and birth defects. Since  
9 the O<sub>3</sub> effect estimates were relatively large (odds ratios  $\geq 2.0$  at the highest O<sub>3</sub> quartile), the  
10 potential role of O<sub>3</sub> on birth defects should be further investigated.

#### 11

#### 12 **7.5.10 Summary of Chronic Ozone Exposure Effects on Morbidity**

#### 13 **and Mortality**

- 14 • In the past decade, important new longitudinal studies have examined the effect of chronic  
15 O<sub>3</sub> exposure on respiratory health outcomes, including seasonal declines in lung function,  
16 increases in inflammation, and development of asthma in children and adults. Seasonal  
17 O<sub>3</sub> effects on lung function have been reported in several studies; however, it remains  
18 uncertain to what extent these changes are transient. There is suggestive evidence that  
19 chronic exposure to O<sub>3</sub> also may be associated with airway inflammation. In contrast to  
20 the supportive evidence from chronic animal studies, epidemiologic studies of new asthma  
21 development and longer-term lung function declines remain inconclusive at present.  
22
- 23 • Few studies have investigated the effect of long-term O<sub>3</sub> exposure on mortality and cancer  
24 incidence. Uncertainties regarding the exposure period of relevance, differential effects  
25 by gender, and inconsistencies across outcomes raise concerns regarding plausibility.  
26 There is currently little evidence for a relationship between chronic O<sub>3</sub> exposure and  
27 increased risk of mortality.  
28
- 29 • A limited number of studies have examined the relationship between air pollution and  
30 birth-related health outcomes, including mortality, premature births, low birth weights,  
31 and birth defects. The most consistent associations with various birth outcomes were  
32 observed for CO. One study reported a large effect of O<sub>3</sub> on cardiac defects. The  
33 potential role of O<sub>3</sub> on birth defects needs to be further examined.  
34

## 7.6 INTERPRETIVE ASSESSMENT OF THE EVIDENCE IN EPIDEMIOLOGIC STUDIES OF OZONE HEALTH EFFECTS

### 7.6.1 Introduction

In the 1996 O<sub>3</sub> AQCD, the epidemiologic section focused primarily on individual-level camp and exercise studies, and studies of hospital admissions and emergency room visits. The field studies indicated concentration-response relationships of O<sub>3</sub> exposure from the ambient air with declines in pulmonary function, increases in respiratory symptoms, and exacerbation of asthma, especially in children. Numerous new studies provide additional evidence for evaluating associations between O<sub>3</sub> exposure and the above respiratory health outcomes. The 1996 O<sub>3</sub> AQCD review of aggregate population time-series studies indicated an association between ambient O<sub>3</sub> concentrations and increased hospitalizations. Limited studies examined the O<sub>3</sub>-mortality relationship. The current O<sub>3</sub> AQCD further presents results from time-series studies that have addressed previously unresolved issues regarding potential linkages between ambient O<sub>3</sub> concentrations and health outcomes, particularly mortality. Daily time-series studies minimize confounding by population characteristics (e.g., cigarette smoking, diet, occupation) by following the same population from day to day. However, confounders operating over shorter time scales can affect O<sub>3</sub> risk estimates in these studies.

In this section, the issues and attendant uncertainties that affect the interpretation of O<sub>3</sub> health effects will be discussed. The use of various indices to represent O<sub>3</sub> exposure in epidemiologic studies is discussed first. Also, of interest is the issue of confounding by temporal factors, meteorological factors, and copollutants. The shape of the concentration-response function and heterogeneity of O<sub>3</sub> effects also will be discussed briefly. All of these topics are of much importance for characterizing and interpreting ambient O<sub>3</sub>-health effects associations.

### 7.6.2 Ozone Exposure Indices

Three O<sub>3</sub> indices were used most often to indicate daily O<sub>3</sub> exposure: maximum 1-h average (1-h max); maximum 8-h average (8-h max); and 24-h average (24-h avg). The 8-h max O<sub>3</sub> is a frequently used index in newer epidemiologic studies, as it best reflects the current U.S. EPA NAAQS. The O<sub>3</sub> exposure indices are highly correlated as indicated in several studies. In the 21 European multicities acute mortality study (Gryparis et al., 2004), 1-h max O<sub>3</sub> was found to be highly correlated with 8-h max O<sub>3</sub>, with a median correlation coefficient of 0.98

1 (range 0.91–0.99). Among single-city studies, the 1-h max O<sub>3</sub> and 8-h max O<sub>3</sub> also were found  
2 to have correlation coefficients ranging from 0.91 to 0.99 in various cities such as Atlanta, GA  
3 (Tolbert et al., 2000; White et al., 1994); southern New England (Gent et al., 2003); Ontario,  
4 Canada (Burnett et al., 1994); and Mexico City (Loomis et al., 1996; Romieu et al., 1995).  
5 In addition, 1-h max O<sub>3</sub> was highly correlated with 24-h avg O<sub>3</sub>, as observed in the Mexico City  
6 study by Loomis et al. (1996) (r = 0.77) and in the Ontario, Canada study by Burnett et al. (1994)  
7 (r = 0.87).

8 All studies discussed in Sections 7.2 to 7.5 were examined for presentation of the three O<sub>3</sub>  
9 exposure indices. Several presented the concentration data and correlations among 1-h max, 8-h  
10 max, and 24-h avg O<sub>3</sub> ambient measures. Some presented the associated risk estimates of  
11 comparable analyses for the three exposure indices. No papers provided a statistical analysis  
12 comparing results from the different indices. Summary of the available data is provided below  
13 starting with two multicity mortality studies.

14 In the large U.S. 95 communities study by Bell et al. (2004), increases in O<sub>3</sub>-associated  
15 daily mortality were estimated using all three O<sub>3</sub> indices. The increments used in this document  
16 to standardize expressions of excess risks are 40 ppb for 1-h max O<sub>3</sub>, 30 ppb for 8-h max O<sub>3</sub>, and  
17 20 ppb for 24-h avg O<sub>3</sub>, as discussed in Section 7.1.3.2. For these increments, the effect  
18 estimates calculated by Bell et al. (2004) using all available data were 1.34% (95% PI: 0.84,  
19 1.85), 1.28% (95% PI: 0.88, 1.73), and 1.04% (95% PI: 0.54, 1.55) excess risk in mortality for  
20 1-h max O<sub>3</sub>, 8-h max O<sub>3</sub>, and 24-avg O<sub>3</sub>, respectively. A statistical test examining differences  
21 among these risk estimates indicated that there were no significant differences by exposure  
22 index. In the European study of 21 cities (of the 23 cities, two did not have 8-h max O<sub>3</sub> data),  
23 the O<sub>3</sub>-mortality effect estimate for the summer season was slightly smaller for 8-h max O<sub>3</sub>,  
24 1.82% (95% CI: 0.99, 3.06) excess risk, compared to 1-h max O<sub>3</sub>, 2.59% (95% CI: 1.32, 4.10)  
25 excess risk; however, the two risk estimates were not significantly different (Gryparis et al.,  
26 2004).

27 Several single-city mortality studies examined multiple O<sub>3</sub> exposure indices (Anderson  
28 et al., 1996; Dab et al., 1996; Sunyer et al., 2002; Zmirou et al., 1996; Borja-Aburto et al., 1997).  
29 These studies did not differentiate risk estimates by exposure index as the results were  
30 considered similar. Hospital admission studies also provided limited data for O<sub>3</sub> index  
31 comparisons. Schouten et al. (1996) found similar O<sub>3</sub> effects on total respiratory hospitalizations

1 from 8-h max O<sub>3</sub> and 1-h max O<sub>3</sub> in the summer. Both indices resulted in a 4.0% excess risk per  
2 standardized increment. For emergency department visits, the examples of Delfino et al.  
3 (1998b) and Weisel et al. (2002) indicated no difference in effect estimate when using various O<sub>3</sub>  
4 indices. Tolbert et al. (2000) noted an increase in emergency room visits of 4.0% per standard  
5 deviation increase (approximately 20 ppb) for both 1-h max O<sub>3</sub> and 8-h max O<sub>3</sub> as being  
6 expected since the correlation between the indices was 0.99.

7 Peak flow asthma panel studies generally used only one index; thus, there were limited  
8 data available for comparison. One respiratory symptom study (Gent et al., 2003) examined  
9 both 1-h max O<sub>3</sub> and 8-h max O<sub>3</sub> but noted no differences in the results. Only one FEV<sub>1</sub> panel  
10 study examined more than one O<sub>3</sub> exposure index. Chen et al. (1999) examined 1-h max O<sub>3</sub> and  
11 24-h avg O<sub>3</sub> and reported a decrement in FEV<sub>1</sub> of -25.6 mL (95% CI: -49.1, -2.1) for 1-h max  
12 O<sub>3</sub> and -13.6 mL (95% CI: -33.2, 6.0) for 24-h avg O<sub>3</sub> in children at a 1-day lag. For 2- and  
13 7-day lags, smaller differences were observed between the two indices. Despite the apparent  
14 differences, the effect estimates calculated using 1-h max O<sub>3</sub> and 24-h avg O<sub>3</sub> concentrations  
15 were not found to be significantly different for any of the lags examined.

16 Limited information is available to reach conclusions for comparison of the three indices  
17 1-h max O<sub>3</sub>, 8-h max O<sub>3</sub>, and 24-h avg O<sub>3</sub>. Studies conducted in various cities have observed  
18 very high correlations among the daily O<sub>3</sub> indices. For the same distributional increment, the  
19 excess health risk estimates and significance of associations were generally comparable for the  
20 three O<sub>3</sub> indices across all outcomes. The high correlation among the indices presents a  
21 challenge in distinguishing the most appropriate measure for epidemiologic studies. Exploratory  
22 analyses using various O<sub>3</sub> exposure indices are valuable in understanding relationships.  
23 However, to address the issue of multiple hypothesis testing, hypotheses that are confirmatory  
24 and exploratory should be decided a priori and reported accordingly.

### 25 26 **7.6.3 Confounding by Temporal Trends and Meteorologic Effects in** 27 **Time-Series Studies**

28 The challenge of analyzing acute O<sub>3</sub> effects in time-series studies is to avoid bias due to  
29 confounding by daily to seasonal temporal factors. On a seasonal scale, the analysis must  
30 remove the influence of the strong seasonal cycles that usually exist in both health outcomes and  
31 O<sub>3</sub>. On a daily scale, weather factors and other air pollutants also may confound the association

1 of interest. This section discusses the interpretation of effect estimates after adjusting for  
2 temporal trends and meteorologic effects.

### 3 4 **7.6.3.1 Assessment of Ozone Effects after Adjusting for Temporal Trends and** 5 **Meteorologic Effects**

6 The relationship between O<sub>3</sub> and health outcomes are significantly affected by temporal  
7 trends and meteorological factors, namely temperature. Analyses of the association between  
8 health outcomes and O<sub>3</sub> concentrations using raw data, therefore, can be misleading. In Díaz  
9 et al. (1999), a U-shaped relationship was observed between mortality and O<sub>3</sub> concentrations, in  
10 which the negative portion of the slope was likely due to the opposing seasonal cycles in  
11 mortality (high in winter) and temperature (low in winter). Goldberg and Burnett (2003) report a  
12 positive slope for the temperature-mortality relationship being fitted most tightly in the mild  
13 temperature range where mortality effects of temperature are not expected. It is possible that  
14 temperature has mortality effects in the mild temperature range, however because daily  
15 fluctuations of air pollution, especially O<sub>3</sub>, are strongly influenced by weather conditions,  
16 ascribing the association between temperature and mortality entirely to effects of temperature  
17 may underestimate the effects of air pollution.

18 Sensitivity analyses specifically for O<sub>3</sub> effects were performed in the U.S. 95 communities  
19 data by Bell et al. (2004). They found that varying the degrees of freedom from 7 to 21 per year  
20 did not significantly affect the O<sub>3</sub>-mortality estimates, with effect estimates ranging from 0.82 to  
21 1.08% excess risk per 20 ppb increase in 24-h avg O<sub>3</sub> during the previous week. Using more  
22 degrees of freedom in temporal trend fitting (i.e., controlling shorter temporal fluctuations)  
23 means ascribing more details of daily health outcomes to unmeasured potential confounders and  
24 possibly taking away real weather and air pollution effects. However, results from this large  
25 multicity study indicated that O<sub>3</sub> effects were robust to aggressive smoothing of temporal trends.  
26 In a related analysis of 19 U.S. cities by Huang et al. (2005), sensitivity of summertime O<sub>3</sub> risk  
27 estimates to varying degrees of freedom (4 to 16 per year) for temporal trend adjustment was  
28 examined. The extent of change in the risk estimates, while varied from city to city (graphically  
29 presented), was not substantial. Huang et al. concluded that the risk estimates were robust to the  
30 adjustment for long-term trends.

31 Ito et al. (2005) examined sensitivity of O<sub>3</sub>-mortality risk estimates to the extent of  
32 temporal trend adjustment and to alternative weather model specifications using data from seven

1 U.S. cities (Cook County, IL; Detroit, MI; Houston, TX; Minneapolis, MN; New York City;  
2 Philadelphia, PA; and St. Louis, MO). They found that varying the degrees of freedom from 4 to  
3 26 per year did not substantially or systematically affect the O<sub>3</sub>-mortality estimates, except for  
4 Cook County where the percent excess O<sub>3</sub>-mortality risk estimates were considerably reduced  
5 when the temporal adjustment term with 26 degrees of freedom was applied. Ito et al. noted that  
6 the O<sub>3</sub> risk estimates were generally more sensitive to alternative weather models than to the  
7 degrees of freedom for temporal adjustment.

8 Schwartz (2005) examined the sensitivity of the O<sub>3</sub>-mortality relationship to methods used  
9 to control for temperature. Initially, temperature lagged 0 and 1 day was controlled using  
10 nonlinear regression splines with 3 degrees of freedom each. In a comparison analysis, control  
11 days were restricted to a subset that was matched on temperature. The effect estimates for all  
12 year data using nonlinear regression splines (0.8% [95% CI: 0.1, 1.4] excess risk per 40 ppb  
13 increase in 1-h max O<sub>3</sub>) and temperature matched controls (0.9% [95% CI: 0.04, 1.8] excess  
14 risk) were not significantly different. Results were similar when restricting analysis to warm  
15 season only data.

16 Temporal cycles in daily hospital admissions or emergency department visits are often  
17 considerably more episodic and variable than is usually the case for daily mortality. As a result,  
18 smoothing functions that have been developed and tuned for analyses of daily mortality data  
19 may not work as well at removing cyclic patterns from morbidity counts. Two methods are  
20 commonly used to adjust for temporal trends. The pre-adjustment method involves applying the  
21 adjustment to both outcome and air pollution variables prior to the regression analysis. The  
22 co-adjustment method involves applying the adjustment as part of the regression analysis, by  
23 fitting a function of time while simultaneously fitting the regression effect of air pollution and  
24 weather factors. As shown in a hospital admissions study by Burnett et al. (2001; used Poisson  
25 GAM with default convergence criteria), the co-adjustment approach may lead to biased air  
26 pollution effect estimates in cases where both outcome and pollution variables exhibit strong  
27 seasonal cycles. Using year-round data, pre-adjustment followed by regression analysis yielded  
28 a 14% (95% CI: 5, 24) increase in admissions per 40 ppb increase in 1-h max O<sub>3</sub> with a  
29 multiday lag of 0 to 4 days. The co-adjustment method resulted in a 7% (95% CI: 3, 11)  
30 decrease in admissions. When the authors limited the analysis to the warm season (May-  
31 August), both methods yielded similar results (32% [95% CI: 21, 44] versus 30% [95% CI: 17,

1 45] increase for co-adjustment and pre-adjustment, respectively) implying that stratification by  
2 season can remove a significant amount of the confounding seasonality (which also may include  
3 seasonally-varying population behavior and ventilation conditions). This finding may be  
4 important to consider in reviewing the acute O<sub>3</sub> mortality and morbidity literature since the vast  
5 majority of studies published over the past decade have used the co-adjustment method.  
6 However, the use of pre-adjustment versus co-adjustment in time-series studies is an unresolved  
7 issue. More empirical research in different locales is needed to evaluate the merits of these two  
8 methods as far as O<sub>3</sub> is concerned, and to determine what endpoints may be affected.

9 More sensitivity analysis of O<sub>3</sub> effect estimates to the extent of adjustment for temporal  
10 trends and meteorological factors is needed, but perhaps it is equally as important to evaluate the  
11 epidemiologic adequacy of a given adjustment. For example, do the fitted mortality series  
12 sufficiently depict influenza epidemics? Or, when larger degrees of freedom (e.g., 12 degrees of  
13 freedom per year) are used, what “unmeasured” confounders, other than weather and pollution,  
14 are the investigators trying to adjust? Even in PM studies that conducted sensitivity analyses,  
15 investigators rarely stated assumptions clearly and not enough discussions were provided as to  
16 potential reasons for the sensitivity of results.

17 Given their relationship to health outcomes and O<sub>3</sub> exposure, adjusting for temporal trends  
18 and meteorologic factors is critical to obtain meaningful O<sub>3</sub> effect estimates. While the  
19 prevailing analytical approaches fit the data flexibly, the estimated effects of meteorologic  
20 variables and their impact on the adjusted O<sub>3</sub> effects are not adequately discussed. More work is  
21 needed in this area to reduce the uncertainty involved in the epidemiologic interpretation of O<sub>3</sub>  
22 effect estimates.

### 24 **7.6.3.2 Importance of Season-Specific Estimates of Ozone Health Effects**

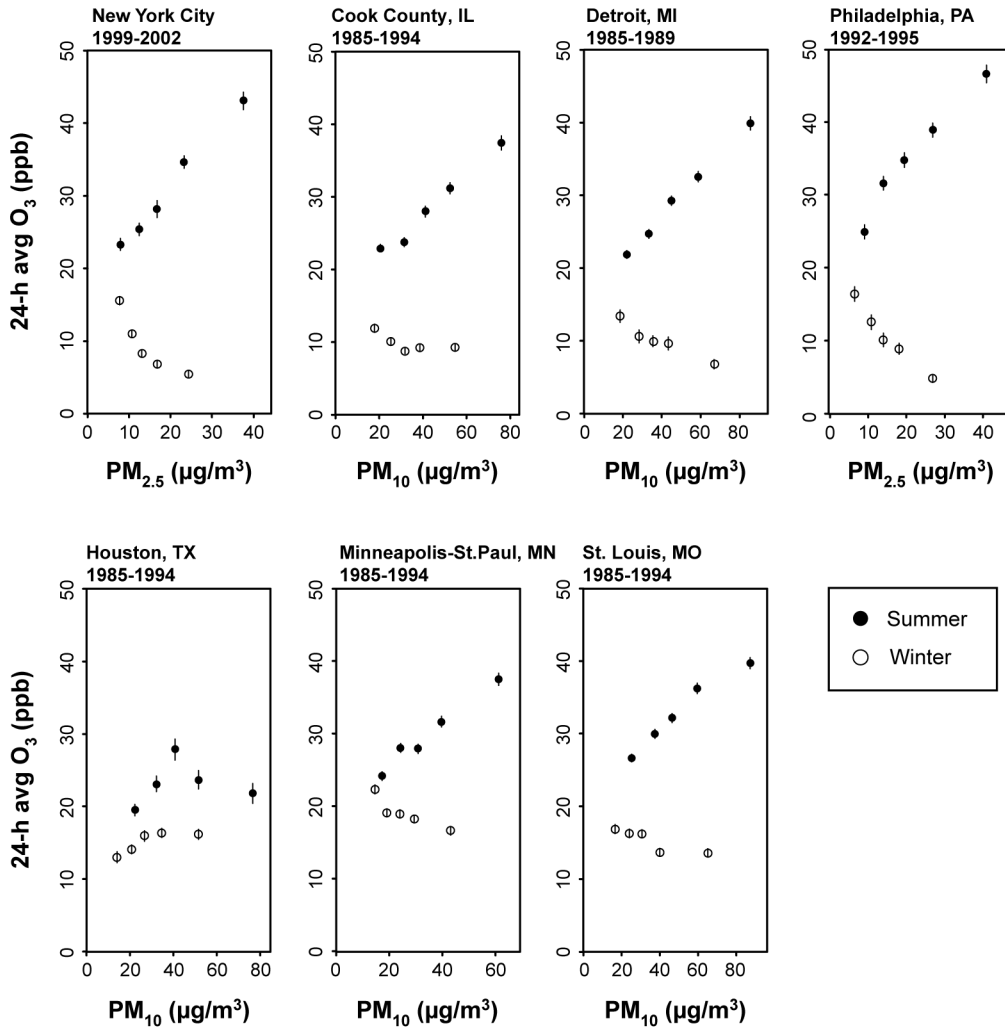
25 Analysis of O<sub>3</sub> health effects is further complicated as relationships of O<sub>3</sub> with other  
26 pollutants and with temperature appear to change across seasons. Moolgavkar et al. (1995)  
27 examined the relationship between daily mortality and air pollution by season in Philadelphia,  
28 PA for the period of 1973 to 1988. During the summer, there was a positive relationship  
29 between O<sub>3</sub> and TSP, as well as O<sub>3</sub> and SO<sub>2</sub>. In contrast, the relationship of O<sub>3</sub> with TSP and  
30 SO<sub>2</sub> inversed during the winter. Ozone showed positive associations only in the summer when  
31 the mean O<sub>3</sub> concentration was the highest. The effect of O<sub>3</sub> on mortality was negative (though

1 not significantly) in the winter when the mean O<sub>3</sub> concentration was low. In the summer  
2 multipollutant model, O<sub>3</sub> was the only pollutant that remained significant. Similar results were  
3 found in another Philadelphia study by Moolgavkar and Luebeck (1996). Both studies did not  
4 analyze year-round data, therefore the relationship between the excess risk estimates for all year  
5 and each season could not be compared. The results from these studies, however, suggest that  
6 year-round analyses may mask the positive (or negative) associations that may exist in particular  
7 seasons.

8 Ito et al. (2005) examined O<sub>3</sub>-mortality associations in seven U.S. cities, but also described  
9 the relationship between O<sub>3</sub> and PM for summer months (June-August) and winter months  
10 (December-February) in these cities (see Figure 7-25). The O<sub>3</sub>-PM relationships were positive in  
11 the summer and negative in the winter in all of these cities, except in Houston, where the O<sub>3</sub>-PM  
12 association was not clearly positive in the warmer months but positive in colder months.  
13 Ito et al. found that O<sub>3</sub>-mortality associations were mostly weaker, null, or even negative in the  
14 winter compared to the summer in most of these cities. Once again, the exception was Houston  
15 where the cold season O<sub>3</sub>-mortality association was positive and larger than those for year-round  
16 or warmer months. Findings from this study suggest the influence of seasonal O<sub>3</sub>-PM  
17 relationships on O<sub>3</sub>-mortality associations.

18 In the analyses of the U.S. 90 cities data (of which 80 cities had O<sub>3</sub> data available) by  
19 Samet et al. (2000; reanalysis Dominici et al., 2003), the focus of the study was PM<sub>10</sub>, but O<sub>3</sub> and  
20 other gaseous pollutants also were analyzed in single- and multiple-pollutant models. In the  
21 reanalysis (Dominici et al., 2003), O<sub>3</sub> was associated with an excess risk of mortality in analyses  
22 of all available data (0.4% [95% PI: 0.1, 0.7]) and summer only data (1.0% [95% PI: 0.5, 1.6]);  
23 however, a negative association was observed for the winter only analysis (-1.1% [95% PI:  
24 -2.2, 0.1]). A twofold greater effect was estimated using summer data compared to all available  
25 data. It should be noted that the analyses by Samet et al. and Dominici et al. used a weather  
26 model specification that is more detailed than other studies in that it had multiple terms for  
27 temperature and dewpoint (these two variables are generally highly correlated). Thus, it is  
28 possible that the high concurrency of O<sub>3</sub> with these weather covariates may have produced these  
29 conflicting results. Another possibility is that the apparent negative relationship between O<sub>3</sub> and  
30 mortality in the winter may have been due to confounding by PM. In the larger U.S. 95  
31 communities study by Bell et al. (2004), the all available data and summer only analyses also





**Figure 7-25. The relationship between PM and O<sub>3</sub> in the summer (June through August) and the winter (December through February) as sorted and averaged by quintiles of PM.**

Source: Derived from Ito et al. (2005).

- 1 indicated positive risk estimates (1.04% [95% PI: 0.54, 1.55] and 0.78% [95% PI: 0.26, 1.30]
- 2 excess risk per 20 ppb increase in 24-h avg O<sub>3</sub>, respectively, using a constrained distributed
- 3 7-day lag model), but the two estimates were similar in magnitude. Winter only analyses were
- 4 not performed. Note that 32 of the 95 communities only had warm season data available.
- 5 Results from the U.S. 95 communities study appear to conflict with the strong seasonal variation

1 observed in the U.S. 90 cities study. However, there are several differences between the two  
2 studies that might account for these results. First, the U.S. 95 communities study nearly doubled  
3 the study period by extending the analysis by six additional years (1987 to 2000 versus 1987 to  
4 1994) and included 15 additional cities to the original 80. Also note that the warm seasons are  
5 characterized differently in the two studies. The U.S. 90 cities study defined summer as a three-  
6 month period of June through August, while the 95 communities study defined warm season as a  
7 seven-month period of April through October. In addition, the results presented in the U.S. 90  
8 cities study were from a single-day lag model (lag 1-day) while the estimates from the 95  
9 communities study were calculated using a constrained distributed 7-day lag model. The  
10 difference in seasonal O<sub>3</sub> effects observed in the two related studies might be attributable to  
11 some of these factors.

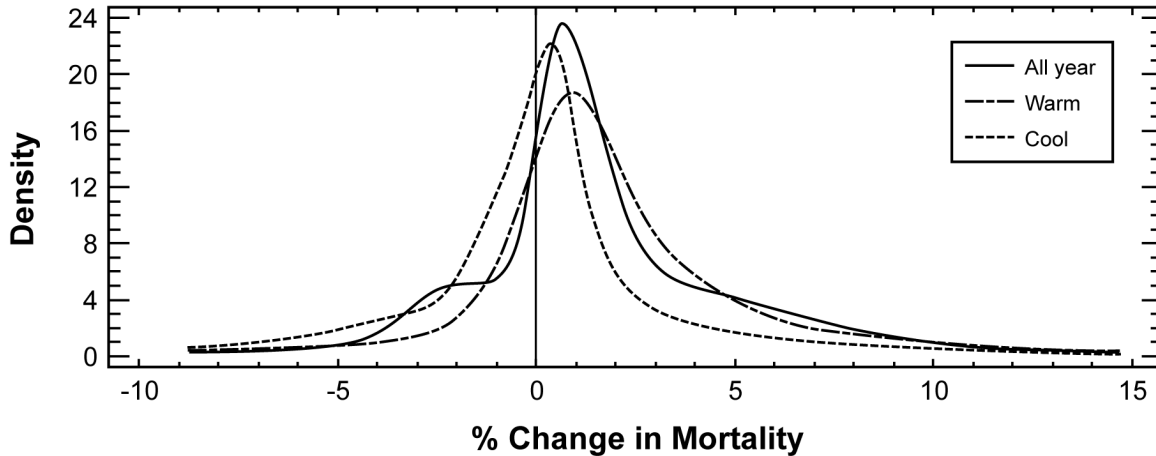
12 Many studies reported larger excess mortality risks in the warm (or summer) season than in  
13 the cool (or winter) season (see Figure 7-19 in Section 7.4.5). These studies showed cool season  
14 risk estimates that were either smaller compared to warm season estimates or slightly negative.  
15 Of the studies that analyzed data by season, only one study in Pittsburgh, PA (Chock et al.,  
16 2000) showed negative risk estimates in the summer. The studies that observed larger, positive  
17 associations between O<sub>3</sub> and mortality in warm seasons are consistent with the expectation that  
18 O<sub>3</sub>, if harmful, should have a stronger association with health outcomes in the summer when  
19 exposure to O<sub>3</sub> is higher. However, the negative O<sub>3</sub>-mortality associations seen in the winter  
20 suggest that further examination of this issue is required. Specifically, if the O<sub>3</sub> level in the  
21 winter is shown to be negatively associated with factors (e.g., PM) that are positively associated  
22 with mortality, then these potentially spurious negative O<sub>3</sub>-mortality associations can be  
23 explained. Several examples of this phenomenon also exist in morbidity studies investigating  
24 the effect of O<sub>3</sub> on daily hospital admissions and emergency department visits (Anderson et al.,  
25 1998; Burnett et al., 2001; Prescott et al., 1998; Thompson et al., 2001).

26 Unlike the time-series studies examining outcomes of mortality, hospital admissions, and  
27 emergency department visits, most acute field studies did not perform year-round analyses.  
28 These acute field studies that examined the relationship between O<sub>3</sub> and lung function,  
29 respiratory symptoms, and inflammation focused primarily on the O<sub>3</sub> effect during the warm  
30 season when O<sub>3</sub> levels were expected to be high and subjects spent more time outdoors and were  
31 physically active.

1           There are seasonal (e.g., air conditioning use) or seasonally-modified (e.g., time spent  
2 outdoors, air exchange rates) factors that affect the relationship between ambient concentrations  
3 and personal exposures to O<sub>3</sub>, as discussed in Section 3.9. The influence of combinations of  
4 these factors across seasons on air pollution health effects can become quite complex. For  
5 example, longer time spent outdoors in the summer may increase personal exposure to O<sub>3</sub> for  
6 some segment of the population, but the increased use of air conditioners may reduce exposures  
7 to ambient O<sub>3</sub> for those who spend much of their time indoors. In the meta-analysis by Levy  
8 et al. (2005), the combined risk estimate from the warm season was greater (3.38% [95% CI:  
9 2.27, 4.42] per 40 ppb increase in 1-h max O<sub>3</sub>) compared to the estimate from all year data  
10 (1.64% [95% CI: 1.25, 2.03]). However, further analysis suggested that the O<sub>3</sub>-mortality risk  
11 estimates were smaller in cities with high air conditioning prevalence. These seasonal factors  
12 that influence the relationship between ambient concentrations and personal exposures make the  
13 interpretation of the concentration-response relationships obtained from analyses of year-round  
14 data less straightforward.

15           In some cities, O<sub>3</sub> is only monitored during the warm season. For example, 34% of the  
16 communities in the U.S. 95 communities study only collected O<sub>3</sub> data during the warm season  
17 (Bell et al., 2004). The cities with larger populations and/or higher O<sub>3</sub> concentrations generally  
18 collected year-round data. There is some concern that differential data availability may  
19 contribute to the seasonal differences in O<sub>3</sub> health effects observed.

20           The potential influence of season on O<sub>3</sub> effect estimates was examined using summary  
21 density curves. The O<sub>3</sub> effect observed in all year data was compared to effects from warm  
22 season and cool season only data (Figures 7-26 and 7-27). Summary probability density curves  
23 were calculated to review the effect estimates from the various studies (see Annex Section  
24 AX7-2 for further explanation of summary density curves). The summary density curves shown  
25 in Figures 7-26 and 7-27 were smoothed by multiplying a constant to the standard error of each  
26 effect estimate in the calculation of the individual distribution functions. Since the normal  
27 distribution is unimodal, this constant will oversmooth when the density is multimodal.  
28 In Figure 7-26, the summary density curves of O<sub>3</sub>-associated all cause (nonaccidental) mortality  
29 are presented (see Figure 7-19 in Section 7.4.5 for the effect estimates). The summary density  
30 curves are calculated using results from 14 studies that reported at least two of the three  
31 estimates. This figure indicates that 75% of the area under the density curve has a value greater

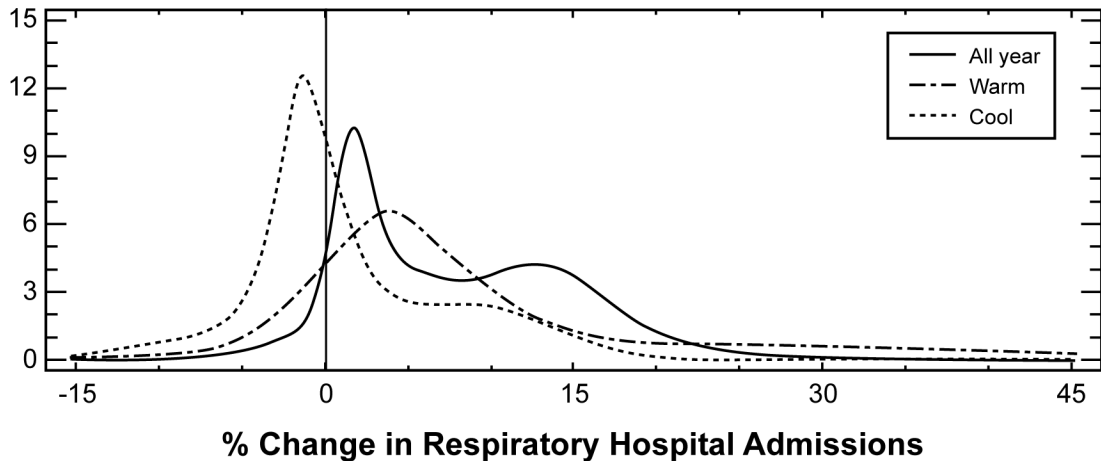


	All year	Warm season	Cool season
% area under the density curve and >0	75%	78%	62%
Mean (SD) effect estimates	1.2% (2.8%)	1.3% (2.6%)	0.1% (3.1%)
Mode effect estimates	0.7%	0.9%	0.4%

**Figure 7-26. Summary density curves of the percent change in all cause mortality for all year data and by season per standardized increment (see Section 7.1.3.2). Effect estimates from 14 studies have been included in the summary density curves (see Figure 7-19 in Section 7.4.5 for the effect estimates).**

1 than zero for all year data compared to 78% for warm season data and 62% for cool season data.  
 2 Therefore, both all year and warm season data generally indicate a positive O<sub>3</sub> effect on  
 3 mortality. The mean effect estimates for all year data and warm season only data are  
 4 1.2%(SD 2.8) and 1.3% (SD 2.6) excess risk in mortality per 40 ppb increase in 1-h max O<sub>3</sub>,  
 5 respectively. A slightly larger mode of effects is observed for warm season data (0.9% excess  
 6 risk) compared to all year data (0.7%). The cool season only data indicate that there is no excess  
 7 risk (mean 0.1% [SD 3.1]) associated with O<sub>3</sub> concentrations.

8 Similar observations are made when examining the O<sub>3</sub> effect on total respiratory hospital  
 9 admissions (Figure 7-27). Six studies provided season-specific estimates as well as all year  
 10 results (see Figure 7-10 in Section 7.3.3 for the effect estimates). Once again, a large percent of  
 11 the area under the summary density curve is greater than zero when using all year and warm  
 12 season data, 92% and 84%, respectively, compared to cool season data, 49%. The mean O<sub>3</sub>



	All year	Warm season	Cool season
% area under the density curve and >0	92%	84%	49%
Mean (SD) effect estimates	6.5% (6.4%)	6.3% (9.1%)	0.8% (6.1%)
Mode effect estimates	1.8%	4.0%	-1.3%

**Figure 7-27. Summary density curves of the percent change in total respiratory hospital admissions for all year data and by season per standardized increment (see Section 7.1.3.2). Effect estimates from six studies have been included in the summary density curves (see Figure 7-10 in Section 7.3.3 for the effect estimates).**

1 effect estimates for warm season data only, 6.3% (SD 9.1) excess risk per 40 ppb increase in 1-h  
 2 max O<sub>3</sub>, and all year analyses, 6.5% (SD 6.4) excess risk, are similar. A larger mode of effects is  
 3 observed for warm season data (3.9% excess risk) compared to all year data (1.8% excess risk).  
 4 A small O<sub>3</sub> effect (0.8% [SD 6.1] excess risk) is observed when using cool season data only.

5 Integrating seasonal influences across the various health outcomes supports the view that  
 6 O<sub>3</sub> effects are different in the cool and warm seasons, with greater effects observed during the  
 7 warm season. As this relates to potentially higher O<sub>3</sub> exposures during the warm season, the  
 8 larger effects are consistent with causal inference. Therefore, these results indicate that the focus  
 9 should be on warm season data to derive quantitative relationships for the effect of O<sub>3</sub> on health  
 10 outcomes. This conclusion is supported by epidemiologic researchers who mainly examine  
 11 warm season as an a priori hypothesis. However, studying summer data only when all year data

1 are available weakens the power of the study since less data are analyzed. In addition, increased  
2 adverse health outcomes are observed in the winter, some of which may be attributable to O<sub>3</sub>.  
3 The O<sub>3</sub> effect in the wintertime may be masked by the effects of PM due to the negative  
4 correlation between these variables (see Section 7.6.4.2 for further discussion). Therefore,  
5 analysis of all year data may be improved by adjusting for PM indices in addition to adequate  
6 adjustment of meteorological factors and temporal trends.

7 Seasonality influences the relationship between O<sub>3</sub> and health outcomes as it may serve as  
8 an indicator for time-varying factors, including temperature, copollutant concentrations,  
9 infiltration, and human activity patterns. Given the potentially significant effect of season, O<sub>3</sub>  
10 effect estimates computed for year-round data need to be interpreted with caution. Small or no  
11 effects may simply reflect the cancellation of positive associations in the summer and negative  
12 associations in the winter, or the presence of confounding due to the strong seasonal character of  
13 O<sub>3</sub> concentrations.

#### 15 **7.6.4 Assessment of Confounding by Copollutants**

16 Potential confounding by daily variations in copollutants is another analytical issue to be  
17 considered. With respect to copollutants, daily variations in O<sub>3</sub> tend to not correlate highly with  
18 most other criteria pollutants (e.g., CO, NO<sub>2</sub>, SO<sub>2</sub>, PM<sub>10</sub>), but may be more correlated with  
19 secondary fine PM (e.g., PM<sub>2.5</sub>, sulfates) measured during the summer months. Assessing the  
20 independent health effects of two pollutants that are correlated over time is not straightforward.  
21 If high correlations between O<sub>3</sub> and PM or other gaseous pollutants exist in a given area, then  
22 disentangling their relative individual contributions to observed health effects associations  
23 becomes very difficult. The changing relationship between O<sub>3</sub> and other copollutants also is of  
24 issue. In some urban locations, the correlation between PM indices and O<sub>3</sub> is positive in the  
25 summer and negative in the winter. This section will further discuss the correlation between O<sub>3</sub>  
26 and copollutants and confounding of the O<sub>3</sub> effect by copollutants.

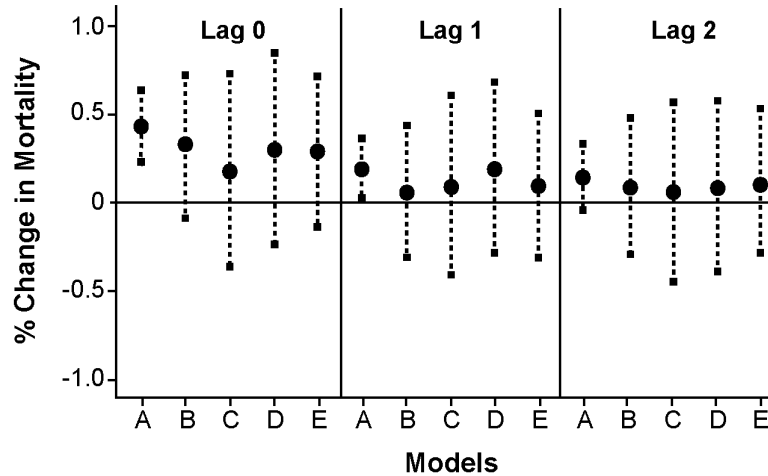
##### 28 **7.6.4.1 Relationship between Personal Exposure to Ozone and Copollutants**

29 Ambient levels of PM, NO<sub>2</sub>, SO<sub>2</sub>, and CO, measured at central monitoring sites, have been  
30 found to be highly correlated to ambient O<sub>3</sub> concentrations. A very limited number of studies  
31 have examined the association between personal O<sub>3</sub> concentrations and personal exposures to

1 other copollutants. An issue of particular interest is the correlation between personal exposure to  
2 O<sub>3</sub> and personal exposure to the ambient component of PM<sub>2.5</sub>. Only one study examined  
3 personal exposure to PM of ambient origin. In a Baltimore, MD study of susceptible populations  
4 (older adults, individuals with COPD, and children), Sarnat et al. (2001) found that ambient 24-h  
5 avg O<sub>3</sub> concentrations and ambient 24-h avg PM<sub>2.5</sub> levels were positively associated ( $\beta = 0.84$ ,  
6  $r = 0.67$ ) in the summer and negatively associated ( $\beta = -0.67$ ,  $r = -0.67$ ) in the winter.  
7 A significant association also was observed between ambient O<sub>3</sub> concentrations and personal  
8 PM<sub>2.5</sub> of ambient origin, with a mixed regression effect estimate of  $\beta = 0.37$  (95% CI: 0.25,  
9 0.49) in the summer and  $\beta = -0.36$  (95% CI: -0.31, -0.41) in the winter. However, no  
10 relationship was found between 24-h avg personal O<sub>3</sub> exposure and personal exposure to PM<sub>2.5</sub>  
11 of ambient origin. While the results from this study provide limited evidence for a lack of an  
12 association between personal O<sub>3</sub> levels and personal exposure to PM<sub>2.5</sub> of ambient origin,  
13 additional research is necessary to address this issue.

#### 14 15 **7.6.4.2 Assessment of Confounding Using Multipollutant Regression Models**

16 Multipollutant regression models are generally used to determine whether the pollutant-  
17 specific effect is robust. However, due to the multicollinearity among O<sub>3</sub> and pollutants, and the  
18 changing correlations by seasons, multipollutant models may not adjust for potential  
19 confounding adequately, especially when using year-round data. These limitations need to be  
20 considered when evaluating results from multipollutant models. Results from the U.S. 90 cities  
21 study, which included 80 cities with O<sub>3</sub> data, indicated that while the addition of PM<sub>10</sub> in the  
22 model did not substantially change the O<sub>3</sub>-mortality risk estimates, slight declines in the O<sub>3</sub>-  
23 effect were observed, as shown in Figure 7-28 (Samet et al., 2000; reanalysis Dominici et al.,  
24 2003). In the extended U.S. 95 communities study (Bell et al., 2004), the city-specific O<sub>3</sub>-  
25 mortality effects were robust to adjustment for PM<sub>10</sub>, as indicated by the nearly 1:1 ratio between  
26 estimates with and without PM<sub>10</sub> adjustment shown in Figure 7-29. This finding suggested that  
27 PM<sub>10</sub> generally did not confound the association between O<sub>3</sub> and mortality. Limited data were  
28 available to examine the potential confounding effect of PM<sub>2.5</sub> on the O<sub>3</sub>-mortality relationship.  
29 A weighted second-stage linear regression indicated that there was no association between long-  
30 term PM<sub>2.5</sub> average and the community-specific O<sub>3</sub>-mortality effect estimate. Several other  
31 mortality and morbidity studies have investigated confounding of O<sub>3</sub> risk estimates using



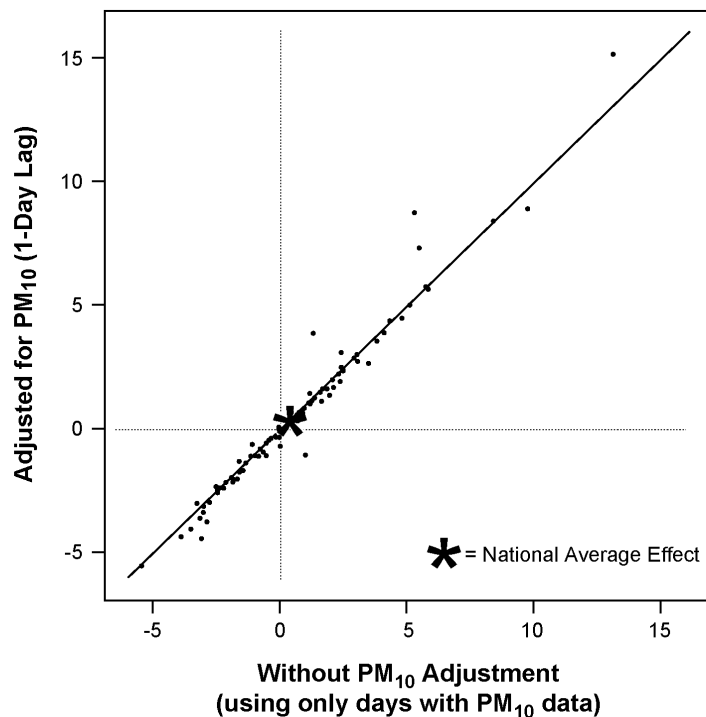
**Figure 7-28. Posterior means and 95% PIs of the national average estimate of O<sub>3</sub> effects on total mortality from non-external causes per 10 ppb increase in 24-h avg O<sub>3</sub> at 0-, 1-, and 2-day lags within sets of 80 U.S. cities with pollutant data available. Models A = O<sub>3</sub> only; B = O<sub>3</sub> + PM<sub>10</sub>; C = O<sub>3</sub> + PM<sub>10</sub> + NO<sub>2</sub>; D = O<sub>3</sub> + PM<sub>10</sub> + SO<sub>2</sub>; E = O<sub>3</sub> + PM<sub>10</sub> + CO.**

Source: Derived from Dominici et al. (2003).

1 multipollutant models with year-round data, and most have reported that O<sub>3</sub> effects were robust  
 2 to adjustment for copollutants (see Figures 7-11 and 7-20 in Sections 7.3.3 and 7.4.6,  
 3 respectively).

4 The pollutant most correlated with O<sub>3</sub> in the summer is sulfate (which is in the fine particle  
 5 size range), especially in the eastern U.S. Therefore, the main potential confounders of interest  
 6 for O<sub>3</sub> are PM<sub>2.5</sub> and sulfate in the summer. Once again, the results from two-pollutant  
 7 regression models with O<sub>3</sub> and sulfate (or PM<sub>2.5</sub>) should be interpreted with caution because both  
 8 of these pollutants are formed under the same atmospheric condition and are both part of the  
 9 “summer haze” pollution mix. A simple two-pollutant regression model does not address their  
 10 possible synergistic effects, and the high correlation between the two pollutants may lead to  
 11 unstable and possibly misleading results. In any case, most studies that analyzed O<sub>3</sub> with PM  
 12 indices did not have PM<sub>2.5</sub> data and very few examined sulfate data. The studies that did have  
 13 PM<sub>2.5</sub> data, including Santa Clara County, CA (Fairley, 1999; reanalysis Fairley, 2003),  
 14 Philadelphia, PA (Lipfert et al., 2000a), and Detroit, MI (Lippmann et al., 2000; reanalysis Ito,





**Figure 7-29. Maximum likelihood estimates of O<sub>3</sub>-mortality for 95 U.S. communities, determined using a constrained distributed lag model for lags 0 through 6 days. Same data set was used for O<sub>3</sub> estimates with and without adjustment for PM<sub>10</sub>.**

Source: Derived from Bell et al. (2004).

1 2003), examined copollutant models for year-round data only, but O<sub>3</sub>-mortality risk estimates  
 2 were not substantially affected by the addition of PM<sub>2.5</sub>. The updated analysis of Philadelphia  
 3 and Detroit data by season suggested that O<sub>3</sub>-mortality risk estimates were not sensitive to  
 4 adjustment for PM<sub>2.5</sub> in all year or seasonal analyses (Ito et al., 2005). A mortality study by  
 5 Lipfert et al. (2000a) also found that all year O<sub>3</sub> risk estimates were not affected by the addition  
 6 of sulfate.

7 Other studies have estimated O<sub>3</sub> health risks with copollutants in the model by season.  
 8 Respiratory hospitalization studies conducted during the warm season in Canada observed  
 9 consistent O<sub>3</sub> risk estimates with the inclusion of PM<sub>2.5</sub> in the model (Burnett et al., 1997b,  
 10 2001). In one of these studies (Burnett et al., 1997b), the effect of O<sub>3</sub> also was adjusted for  
 11 sulfate. With the addition of sulfate in the model, the risk estimate for O<sub>3</sub> on respiratory

1 hospitalizations remained relatively stable, from an 14.4% (95% CI: 8.7, 20.5) excess risk to a  
2 11.7% (95% CI: 5.6, 18.0) excess risk per 25 ppb increase in 12-h avg O<sub>3</sub> at a 1-day lag.  
3 In contrast, the effects for sulfate were reduced in half after adjusting for O<sub>3</sub>. Amongst the  
4 mortality studies (see Figure 7-21 in Section 7.4.6), adjusting for copollutants, in particular PM  
5 indices, did not substantially change the warm season O<sub>3</sub>-mortality effect estimates, with both  
6 slight reductions and increases observed in the adjusted estimates. In the analysis using cool  
7 season data only, the O<sub>3</sub> effect estimates were generally negative, but none were statistically  
8 significant. The O<sub>3</sub> risk estimates all increased slightly with the adjustment of PM indices.  
9 The inverse relationship between O<sub>3</sub> and PM during the cool season most likely influenced the  
10 O<sub>3</sub>-mortality effect estimates in the single-pollutant models.

11 In field studies, power to assess independent O<sub>3</sub> effects may be limited by small sample  
12 sizes and short follow-up times. Yet, the O<sub>3</sub> effect also was robust to the addition of copollutants  
13 in multipollutant models, with a few exceptions. For example, the effect of O<sub>3</sub> on PEF was not  
14 robust to adjustments for PM<sub>2.5</sub> and sulfate, in studies by Romieu et al. (1996) and Neas et al.  
15 (1999). In general, however, O<sub>3</sub> effects on respiratory symptoms (Romieu et al., 1996), lung  
16 function parameters (Brauer et al., 1996, Gold et al., 1999), and asthma medication use (Gent  
17 et al., 2003) were robust to inclusion of PM<sub>2.5</sub>. Further, the effects for O<sub>3</sub> were observed to be  
18 stronger than those for PM.

19 Multipollutant regression analyses indicated that O<sub>3</sub> risk estimates, in general, were not  
20 sensitive to the inclusion of copollutants, including PM<sub>2.5</sub> and sulfate. These results suggest that  
21 the effect of O<sub>3</sub> on respiratory health outcomes appears to be robust and independent of the  
22 effects of other copollutants.

## 24 **7.6.5 Concentration-Response Function and Threshold**

25 An important consideration in characterizing the public health impacts associated with O<sub>3</sub>  
26 exposure is whether the concentration-response relationship is linear across the full  
27 concentration range or instead shows evidence of a population threshold. Of particular interest is  
28 the shape of the concentration-response curve at and below the level of the current 8-h standard  
29 of 80 ppb. The slope of the O<sub>3</sub> concentration-response relationship has been explored in several  
30 studies.

1 To examine the shape of the concentration-response relationship between O<sub>3</sub> and mortality,  
2 Gryparis et al. (2004) used meta-smoothing to combine smooth curves across the 23 European  
3 cities in a hierarchical model. For the summer period, while the estimated concentration-  
4 response curve did not appear to deviate significantly from linearity, there were indications of  
5 decreasing effectiveness at lower exposures.

6 In the U.S. 95 communities study (Bell et al., 2004), effect estimates calculated using only  
7 days with 24-h avg O<sub>3</sub> levels less than 60 ppb were compared to those using all data. At a lag of  
8 1 day, O<sub>3</sub> was associated with an excess risk of 0.36% (95% PI: 0.12, 0.60) per 20 ppb increase  
9 in 24-h avg O<sub>3</sub> using data from all days and only a slightly smaller risk of 0.30% (95% PI: 0.08,  
10 0.54) when data were limited to days less than 60 ppb. These results suggest that if there is a  
11 threshold, it must be notably lower than a 24-h avg O<sub>3</sub> of 60 ppb.

12 Fairley (2003) reanalyzed the Santa Clara County mortality data using GAM with stringent  
13 convergence criteria and examined a new exposure index for O<sub>3</sub>. He noted O<sub>3</sub> concentrations  
14 exceeding 60 ppb each hour and calculated a daily sum of these exceedances. Fairley's index  
15 incorporates measures of concentration and exposure duration; this index represents a linear  
16 time-integrated concentration, also known as dosage. The O<sub>3</sub> index with the 60 ppb "threshold  
17 level" was found to be significantly associated with mortality in single-pollutant models as well  
18 as in multi-pollutant models. Two other "threshold levels" were examined, 40 ppb and 80 ppb.  
19 Both produced statistically significant results in single-pollutant models. These results suggest  
20 that the threshold for O<sub>3</sub>-mortality effects, if it exists, is likely less than 40 ppb. The implication  
21 for thresholds in terms of the three standard indices (i.e., 1-h max, 8-h max, and 24-h avg) is  
22 unclear, but there may be an empirical relationship.

23 Vedal et al. (2003) observed that the annual mean 1-h daily max O<sub>3</sub> concentration of  
24 27.3 ppb in Vancouver, Canada, was lower than that in any of the 80 NMMAPS cities (Samet  
25 et al., 2000); thus, a study in this city might focus better on the shape of the concentration-  
26 response curve at lower levels. In this study, an O<sub>3</sub> effect was observed on total mortality at a  
27 0-day lag during the summer. Ozone effects on respiratory mortality at a 2-day lag and  
28 cardiovascular mortality at a 0-day lag also were observed in the summer. The effect of O<sub>3</sub> on  
29 mortality was robust in two-pollutant models. Vedal et al. questioned if O<sub>3</sub>, other gaseous  
30 pollutants, and PM were acting as surrogate markers of pollutant sources that contain more toxic  
31 compounds, since the low measured concentrations were unlikely, in their opinion, to cause the

1 observed effects. They further stated that measurement error and interference by meteorological  
2 factors might have contributed to the inability to detect a threshold. Vedal et al. (2003)  
3 concluded that O<sub>3</sub> concentrations were associated with adverse effects on mortality even at low  
4 levels. Although this study supports the argument that there are no threshold concentrations  
5 below which adverse effects cannot be detected, the results must be interpreted with caution as  
6 concerns remain.

7 Kim et al. (2004) investigated the presence of a threshold in O<sub>3</sub>-mortality effects in Seoul,  
8 Korea by analyzing data using a log linear GAM (linear model), a cubic natural spline model  
9 (nonlinear model), and a B-mode splined model (threshold model). Models were stratified by  
10 season and adjusted for PM<sub>10</sub>, long-term time trend, and meteorological variables. An estimated  
11 threshold value of 47 ppb was observed for 1-h daily max O<sub>3</sub>. None of the other pollutants  
12 examined, including PM<sub>10</sub>, SO<sub>2</sub>, NO<sub>2</sub>, and CO, had a nonlinear association with mortality. Using  
13 summer data only, the B-spline model resulted in an excess mortality risk of 7.1% (95% CI: 3.1,  
14 11.2) per 40 ppb increase in 1-h max O<sub>3</sub>, compared to an excess risk of 3.6% (95% CI, 0.5, 6.8)  
15 calculated using the log linear model. If a threshold truly exists, results from the Kim et al. study  
16 suggest that the use of log-linear models may underestimate the O<sub>3</sub> effect on mortality at levels  
17 above the threshold.

18 Other studies examining the effect of O<sub>3</sub> on mortality also have found suggestive evidence  
19 for a possible threshold level. In a London, England study (Anderson et al., 1996), an adjusted  
20 O<sub>3</sub>-mortality bubble plot suggested that a threshold might exist around 50 ppb for 8-h avg O<sub>3</sub>.  
21 A study by Simpson et al. (1997) in Brisbane, Australia observed a significant excess risk in  
22 mortality only in the highest quintile of O<sub>3</sub> exposure, which had a mean concentration of 42 ppb  
23 for 1-h max O<sub>3</sub>.

24 Among several studies with morbidity outcomes, examination of the shape of the  
25 concentration-response function indicated evidence of an effect threshold. In a study of all-age  
26 respiratory hospital admissions in Toronto, Canada, effects of O<sub>3</sub> appeared to become apparent  
27 only above approximately 30 ppb daily 1-h max O<sub>3</sub> (Burnett et al., 1997b). In London, England,  
28 Ponce de Leon et al. (1996) observed an indication of a threshold in the O<sub>3</sub> effect on  
29 hospitalizations at 40 to 50 ppb for 8-h max O<sub>3</sub> and 50 to 60 ppb for 1-h max O<sub>3</sub>. In a study of  
30 emergency department visits for asthma in St. John, Canada, effects observed in the over  
31 15 years age group were apparent only when data above the 95th percentile (75 ppb daily 1-h

1 max O<sub>3</sub>) were included (Stieb et al., 1996). However, other morbidity studies observed a  
2 monotonic increase in the concentration-response function, suggesting that there was no  
3 threshold in O<sub>3</sub> effects on hospitalizations and emergency department visits (Burnett et al.,  
4 1997a; Jaffe et al., 2003; Petroeschevsky et al., 2001; Tenías et al., 1998).

5 In a field study by Mortimer et al. (2002), the associations of ambient O<sub>3</sub> levels with PEF  
6 and asthma symptoms were investigated in eight urban cities in the U.S. The mean 8-h avg O<sub>3</sub>  
7 was 48 ppb, with less than 5% of days exceeding 80 ppb. Analysis performed using all data  
8 indicated that a 15 ppb change in 8-h avg O<sub>3</sub> was associated with decrements in PEF (-0.59%  
9 [95% CI: -1.05, -0.13]) and increased incidence of respiratory symptoms (odds ratio of 1.16  
10 [95% CI: 1.02, 1.30]) over multiday lag periods. When data were restricted to days when  
11 ambient O<sub>3</sub> concentrations were less than 80 ppb, the O<sub>3</sub> effects persisted, with a significant PEF  
12 decline (-0.70% [95% CI: -1.29, -0.12]) and incidence of morning symptoms (odds ratio of  
13 1.17 [95% CI: 1.01, 1.35]). A study by Chen et al. (1999) also found that there was no clear  
14 threshold in the O<sub>3</sub> effect on FEV<sub>1</sub> and FVC in Taiwanese school children.

15 The studies of both Brauer et al. (1996) and Korrick et al. (1998) demonstrate that  
16 exposure duration and exercise level, in addition to O<sub>3</sub> concentration, must be considered when  
17 evaluating thresholds. In the study by Brauer et al., the mean O<sub>3</sub> concentration during the  
18 11-hour work shift was 26.0 ppb (SD 11.8). Workers experienced a change of -180.0 mL (95%  
19 CI: -227.0, -133.0) in FEV<sub>1</sub> levels the next morning per 40 ppb increase in 1-h max O<sub>3</sub>. The  
20 hikers in the study by Korrick et al. (1998) were exposed to mean O<sub>3</sub> levels of 40 ppb (SD 12)  
21 over the duration of their hike (mean 8 hours). Korrick et al. observed a mean change of -62.5  
22 mL (95% CI: -115.3, -9.7) in pre-hike to post-hike FEV<sub>1</sub> per 30 ppb increase in 8-h avg O<sub>3</sub>  
23 when all hikers were included in the analysis; however, when analysis was restricted to hikers  
24 with wheeze or asthma, a larger change of -182.5 mL (95% CI: -312.2, -52.9) was observed.  
25 In both studies, large reductions in lung function were observed in subjects exposed to relatively  
26 low levels of O<sub>3</sub> over multiple hours while active outdoors.

27 Note that adjusting for seasonal cycles does not address the issue of the changing  
28 relationship between O<sub>3</sub> concentrations and personal exposure across seasons. The ambient O<sub>3</sub>  
29 levels are lower in the cold season, but people are likely to be exposed to even lower levels of  
30 O<sub>3</sub> during this season due to the shorter time spent outdoors and the longer time spent indoors  
31 with closed windows. This is in contrast to what occurs with fine particles, which can

1 effectively penetrate the indoors. Thus, a more “accurate” concentration-response relationship  
2 may need to be examined in a summer-only data set. Even for summer data, however, an  
3 interpretation of the relationship is not straightforward because of the possible influence of the  
4 use of air conditioning (an effective remover of O<sub>3</sub>). Greater use of air conditioning is expected  
5 on hot days when the O<sub>3</sub> level is higher, but the use of air conditioning may also vary from city  
6 to city and across social class within a city. Using PM<sub>2.5</sub> and sulfate as an example, Brauer et al.  
7 (2002) observed that surrogate measures of exposure (i.e., those from centrally-located ambient  
8 monitors) that were not highly correlated with personal exposures obscured the presence of  
9 thresholds in epidemiologic studies of larger populations. Likewise, exposure measurement  
10 error may reduce the ability to detect a threshold in O<sub>3</sub> population studies that used ambient O<sub>3</sub>  
11 concentrations as an indicator of personal exposure.

12 Limited studies have examined the issue of thresholds in O<sub>3</sub> health effects studies. Some  
13 studies have found a low level threshold while others have found no threshold in O<sub>3</sub> effects.  
14 Levy et al. (2001) states that the molecular effects of O<sub>3</sub> are mediated by antioxidants in the lung  
15 lining fluid, which raises the possibility that there may be a threshold levels below which O<sub>3</sub>  
16 would have few or no adverse effects. However, due to the variability in individual sensitivities  
17 and antioxidant levels, this threshold may not be seen at the population level.

18 From 1990 to 2004, the 10th percentile values (which represent the lower concentration  
19 range) of the warm season (May to September) 8-h max O<sub>3</sub> concentrations averaged for all  
20 available monitors throughout the U.S. were approximately 40 ppb (see discussion in Section  
21 3.2). While no conclusion can be made regarding the threshold issue, the limited evidence  
22 suggests that if there is a threshold level in O<sub>3</sub> health effects, it is likely near the lower limit  
23 of ambient O<sub>3</sub> concentrations in the U.S.

## 24 25 **7.6.6 Heterogeneity of Ozone Health Effects**

26 As described in Chapter 3 of this AQCD, O<sub>3</sub> concentrations tend to be more spatially  
27 variable than PM<sub>2.5</sub> concentrations in urban areas. In addition, relative personal exposures to O<sub>3</sub>  
28 likely vary by region. The geographic variability in O<sub>3</sub> concentrations and personal exposures  
29 may contribute to the heterogeneity in observed O<sub>3</sub> health effects. The degree of influence of the  
30 geographic variability on heterogeneity in effects will vary by study as study design affects  
31 different aspects of exposure (e.g., time period and duration of exposure).

1 More than 80% of the O<sub>3</sub>-mortality estimates from the various studies conducted in North  
2 America, South America, Europe, and Australia were between 0.5 and 5% excess risk per 40 ppb  
3 increase in 1-h max O<sub>3</sub> using year-round data. In general, the O<sub>3</sub>-mortality estimates were  
4 greater when using summer only data compared to year-round data. Though not all statistically  
5 significant, most of the O<sub>3</sub>-mortality estimates were greater than zero, indicating a positive  
6 relationship between O<sub>3</sub> exposure and mortality. The O<sub>3</sub> risk estimates from the numerous  
7 hospitalization and emergency department visit studies were generally larger in magnitude and  
8 more variable from study to study compared to the mortality studies. These differences in the  
9 O<sub>3</sub> effect estimates may be attributable to the greater variability in the outcome measure in  
10 hospitalization studies compared to mortality studies, such as more subcategories of outcome  
11 and varying degrees of severity.

12 Three recent meta-analyses that included both U.S. and non-U.S. studies found consistent  
13 all-year combined point estimates: 1.75% (95% PI: 1.10, 2.40), 1.6% (95% CI: 1.1, 2.0), and  
14 1.64% (95% CI: 1.25, 2.03) per 20 ppb increase in 24-h average O<sub>3</sub>, for Bell et al. (2005),  
15 Ito et al. (2005), and Levy et al. (2005), respectively. Bell et al. further observed that the pooled  
16 estimate for U.S. studies (11 estimates), 1.69% (95% PI: 0.94, 2.78), was similar to the pooled  
17 estimate for the non-U.S. studies (30 estimates), 1.85% (95% PI: 0.94, 2.78). Levy et al.  
18 compared North American studies to European studies and also found nearly identical effect  
19 estimates.

20 As differences in study design, population, and data analysis may affect risk estimates,  
21 studies that were conducted in multiple cities using standardized methods were further examined  
22 to investigate the geographic heterogeneity of O<sub>3</sub> effects. Bell et al. (2004) conducted a time-  
23 series analysis of O<sub>3</sub> and mortality in 95 U.S. communities from 1987 to 2000. A 20 ppb  
24 increase in 24-h avg O<sub>3</sub> levels in the previous week was associated with an increase of 1.04%  
25 (95% PI: 0.54, 1.55) excess risk of mortality in the pooled analysis of 95 communities using all  
26 available data. Intercity heterogeneity was observed among the 95 communities, which the  
27 authors noted as plausible given the city-specific differences in pollution characteristics, the use  
28 of air conditioning, time-activity patterns, and socioeconomic factors. Although some  
29 heterogeneity was observed among the communities (see Figure 7-16 of Section 7.4.3), the  
30 range of the community-specific Bayesian estimates was fairly narrow. Note that the  
31 community-specific Bayesian estimates are shrunken estimates of the percent changes in daily

1 mortality. The larger the heterogeneity (across-community variance relative to within-  
2 community variance), the less the Bayesian estimates shrink toward the national average. Of the  
3 95 U.S. communities, 93 had positive O<sub>3</sub>-mortality risk estimates. Only 5 had risk estimates  
4 greater than 2.0% per 20 ppb increase in 24-h avg O<sub>3</sub> during the previous week, with all  
5 communities indicating an excess mortality risk less than 3.5%.

6 Greater heterogeneity was observed in the European study of 23 cities in 14 countries  
7 (Gryparis et al., 2004). In the year-round analyses, only 8 of the 23 cities had positive  
8 O<sub>3</sub>-mortality effect estimates. However, in the analyses using summer data only, the risk  
9 estimates were positive in 19 of the 23 cities, with a range of 0.8 to 8% excess risk per 40 ppb  
10 increase in 1-h max O<sub>3</sub>. The heterogeneity may be attributable to the considerable variability  
11 among countries in factors that may influence the relationship between ambient O<sub>3</sub>  
12 concentrations and personal exposure to O<sub>3</sub>, such as climate, use of air conditioning, personal  
13 activity patterns, and socioeconomic factors. In addition, the variability in the concentration and  
14 composition of copollutants by cities or countries may contribute to the heterogeneity in the  
15 O<sub>3</sub>-mortality effects. For example, concentrations of NO<sub>2</sub> may vary widely by region, depending  
16 on the differences in traffic density.

17 Among the hospitalization studies, Burnett et al. (1997a) conducted the largest study of  
18 16 Canadian cities. The mean daily 1-h max O<sub>3</sub> was 31 ppb in the 16 cities. The pooled O<sub>3</sub>  
19 estimate was 5.6% (95% CI: 3.4, 7.9) excess risk in respiratory hospitalization per 40 ppb  
20 increase in 1-h max O<sub>3</sub> using warm season data (April to December). The risk estimates were  
21 fairly homogenous across the 16 Canadian cities, ranging from 3.1% for Vancouver to 7.7% for  
22 Quebec City.

23 Anderson et al. (1997) investigated the association between O<sub>3</sub> and hospital admissions for  
24 COPD in five European cities — London, Paris, Amsterdam, Rotterdam, and Barcelona. The  
25 pooled effect estimate was 5.0% (95% CI: 2.6, 7.6) excess risk per 30 ppb increase in 8-h max  
26 O<sub>3</sub> for year-round data. Results from the APHEA study showed similar variability to that from  
27 the Burnett et al. (1997a) study. The year-round effects estimates were lower in the two Dutch  
28 cities (2.5% excess risk) compared to that in Paris (7.7% excess risk); however, analyses  
29 indicated that there was no significant heterogeneity in effects by city. The authors further noted  
30 that among the pollutants examined (O<sub>3</sub>, BS, TSP, SO<sub>2</sub>, and NO<sub>2</sub>), O<sub>3</sub> had the most consistent  
31 and significant findings.



1           Among the field studies, various respiratory health outcomes were examined, including  
2 PEF, spirometric parameters, respiratory symptoms, and medication use. Only one field study  
3 investigated the O<sub>3</sub> effect in several locations (Mortimer et al., 2002). Mortimer et al. (2002)  
4 investigated the association of ambient O<sub>3</sub> concentrations with PEF and asthma symptoms in  
5 asthmatic children living in eight urban cities in the U.S. — St. Louis, MO; Chicago, IL; Detroit,  
6 MI; Cleveland, OH; Washington, DC; Baltimore, MD; East Harlem, NY; and Bronx, NY. In the  
7 analysis pooling data from all eight cities, a 30 ppb increase in 8-h avg O<sub>3</sub> was associated with a  
8 decrement of -1.18% (95%CI: -2.10, -0.26) in morning PEF for a 5-day cumulative lag period.  
9 The percent changes in PEF were negative in all cities except for Baltimore, 0.49%. Among the  
10 other seven cities, the percent changes in PEF were quite homogenous, with values ranging from  
11 -1.08% for Washington, DC to -1.71% for St. Louis. A 30 ppb increase in 8-h avg O<sub>3</sub> also was  
12 associated with an increased incidence of morning symptoms in the pooled analysis (odds ratio  
13 of 1.35 [95% CI: 1.04, 1.69] for a 4-day cumulative lag period). In all cities except for  
14 St. Louis, there was an increase in the incidence of morning symptoms. In these cities, the odds  
15 ratios for incidence of morning symptoms varied more compared to the PEF measurements,  
16 ranging from 1.19 for Chicago to 2.96 for Detroit. The greater variance may indicate the lack of  
17 standardization in the use of symptoms as a health outcome measure.

18           Most of the multicity and meta-analyses studies consistently found positive associations  
19 between O<sub>3</sub> and mortality. Consistent O<sub>3</sub> effects on hospitalizations and various respiratory  
20 health outcomes also were found. The observed heterogeneity of O<sub>3</sub> effects may be partially  
21 attributable to the use of centrally-located ambient monitors to assess exposure. There may be  
22 differences in relative personal exposures to O<sub>3</sub> due to varying factors, such as use of air  
23 conditioning and activity patterns, that affect the relationship between personal exposure and  
24 ambient concentrations. For example, Levy et al. (2005) found suggestive evidence that air  
25 conditioning prevalence was a predictor of heterogeneity in O<sub>3</sub> risk estimates in their meta-  
26 analysis. The variability in the concentration and composition of other pollutants present also  
27 may contribute to the heterogeneity of the effect of O<sub>3</sub> on health outcomes as confounding by  
28 copollutants may vary by region.

## 7.6.7 Health Effects of Ozone in Susceptible Populations

In this section, the effects of O<sub>3</sub> on morbidity and mortality in potentially susceptible populations will be examined. In epidemiologic studies of O<sub>3</sub> health effects, the most widely studied subpopulation was asthmatics. Also of interest were the observed health effects of O<sub>3</sub> on different age groups, particularly children and the elderly. This section begins with a discussion of the O<sub>3</sub>-related health effects in asthmatics.

### 7.6.7.1 Health Effects Associated with Ambient Ozone Exposure in Asthmatics

Epidemiologic studies of health effects from acute O<sub>3</sub> exposure in asthmatics have examined a range of outcomes: pulmonary function, respiratory symptoms, inflammation, emergency room visits, hospital admissions, and mortality. Chronic O<sub>3</sub> exposure studies have investigated similar outcomes, with the exception of emergency room visits and hospitalizations. Both are discussed in the earlier text. This subsection draws together this information to examine whether the evidence indicates that O<sub>3</sub> exposure impacts asthmatics.

In Germany and Mexico City, O<sub>3</sub> exposure was associated with a decline in FEV<sub>1</sub> in asthmatic adults and children (Höppe et al., 1995a, 2003; Romieu et al., 2002). Change in FEV<sub>1</sub> also was examined in a group of asthmatic hikers in Mount Washington, NH (Korrick et al., 1998). Compared to the healthy subjects, the asthmatic subjects experienced a four-fold greater decline in FEV<sub>1</sub> with the same exposure to O<sub>3</sub> (mean change of -1.08% [95% CI: -2.49, 0.33] versus -4.47% [95% CI: -7.65, -1.29] per 30 ppb increase in 8-h avg O<sub>3</sub>). The results from the hiker study are consistent with those observed in controlled human exposure studies (discussed in Chapter 6), which also indicate greater decrements in FEV<sub>1</sub> among mild asthmatics versus nonasthmatic subjects with heavy intermittent exercise.

PEF was examined in panels of asthmatic children in several field studies (see Figures 7-1 and 7-2). Collectively, most of the studies indicated decrements of morning PEF, though only a few estimates were statistically significant. One multicity study of eight urban areas in the U.S. observed O<sub>3</sub>-related reductions in morning PEF that were not significant in each individual city (Mortimer et al., 2002); however, the analysis combining data from all eight cities indicated a significant decline in PEF with a cumulative lag of 1 to 5 days of O<sub>3</sub> exposure. The odds ratio for the incidence of ≥10% decline in morning PEF was greater than one, which was discussed by the author as an indication that O<sub>3</sub> exposure might be associated with clinically important

1 changes in PEF in asthmatic children. The study examined 846 asthmatic children, the largest  
2 asthma panel study reported.

3 Mortimer et al. (2000) observed that the subpopulation of asthmatic children with a history  
4 of low birth weight or premature birth had greater O<sub>3</sub>-associated declines in PEF (mean change  
5 of -3.66% [95% CI: -5.30, -2.02] per 30 ppb increase in 8-h avg O<sub>3</sub>) than normal birth weight  
6 children (-0.60% [95% CI: -1.58, 0.38]). Low birth weight and prematurity are associated with  
7 reduced lung function, higher levels of airway reactivity and increased susceptibility to lung  
8 damage (Barker et al., 1993; Rona et al., 1993), which may explain why these factors are found  
9 to increase susceptibility to respiratory insults of air pollution in children.

10 Lung function parameters have been evaluated for clinical significance. A reversible 5 to  
11 15% decline in FEV<sub>1</sub> in an individual may have clinical importance to asthma morbidity  
12 (American Thoracic Society, 1991; Lebowitz et al., 1987; Lippmann, 1988). The National  
13 Institutes of Health (1997) has stated that PEF below 80% of the personal best indicates a need  
14 for additional medication use in asthmatics. At a population level the mean changes in lung  
15 function attributable to O<sub>3</sub> exposure do not generally exceed 10% changes in FEV<sub>1</sub> or PEF per  
16 standardized increment of O<sub>3</sub>. At an individual level, a subpopulation of susceptible asthmatics  
17 are likely experiencing clinically significant declines in lung function. Höpfe et al. (2003)  
18 examined effects of O<sub>3</sub> on the lung function of potential risk groups, performing both group and  
19 individual analyses. For the group mean values, consistent O<sub>3</sub> effects were not detectable.  
20 On an individual basis, a potential pattern of O<sub>3</sub> sensitivity was observed. About 20% of the  
21 asthmatics and children were regarded as O<sub>3</sub> responders (i.e., individuals with >10% change in  
22 FEV<sub>1</sub>) compared to only 5% of the elderly and athletes. These results indicated that while the  
23 population as a whole was not reacting to O<sub>3</sub>, susceptible individuals were experiencing  
24 clinically significant declines in lung function in response to O<sub>3</sub> exposure.

25 Respiratory symptom increases in asthma panels were examined in several field studies,  
26 some of which also examined PEF as discussed above. The outcome definition of symptoms  
27 varied among these studies. Collectively, the results are suggestive of a potential O<sub>3</sub> effect on  
28 respiratory symptoms, but the evidence is not strong in the available studies. Two U.S. studies  
29 that examined larger panels might be better to draw inferences from as the large sample size  
30 provided greater power to examine the effect of O<sub>3</sub> on respiratory symptoms. The eight U.S.  
31 urban cities study reported that morning symptoms in the 846 asthmatic children were most

1 strongly associated with a 4-day cumulative lag of O<sub>3</sub> concentrations (Mortimer et al., 2002).  
2 A New England study examined 271 asthmatic children and observed an O<sub>3</sub> effect on a variety  
3 of respiratory symptoms at a lag of 1 day among the 130 subjects who used maintenance asthma  
4 medications (Gent et al., 2003).

5 Few epidemiologic studies have examined airway inflammation in asthmatics. A Mexico  
6 City study indicated that supplementation with antioxidants may modulate the impact of O<sub>3</sub>  
7 exposure on the small airways of children with moderate-to-severe asthma (Romieu et al., 2002).  
8 A related study indicated that asthmatic children with GSTM1 null genotype were found to be  
9 more susceptible to the impact of O<sub>3</sub> exposure on small airways (Romieu et al., 2004). A chronic  
10 exposure study in Mexico City examined DNA strand breaks in nasal epithelial cells in  
11 asthmatic and nonasthmatics medical students and noted greater genotoxic damage in asthmatics  
12 (Fortoul et al., 2003).

13 Emergency department visits for asthmatics have been examined in several studies and  
14 range from negative to positive results (see Figure 7-8 in Section 7.3.2). Warm season studies  
15 tended to yield positive outcomes, as expected based on earlier discussions. Two studies in  
16 Atlanta, GA (Tolbert et al., 2000) and Valencia, Spain (Tenías et al., 1998) indicated positive  
17 effects in warm season analyses. Further, a Canadian study, one of the larger studies conducted  
18 in the summertime, reported a large increase in asthma emergency department visits when the  
19 daily 1-h max O<sub>3</sub> concentration exceeded 75 ppb (Stieb et al., 1996). A three-city study in Ohio  
20 also indicated an increased risk of asthma visits during the summer (Jaffe et al., 2003). Other  
21 studies of mostly year-long data tended to produce inconsistent results, with some finding  
22 negative estimates (Atkinson et al., 1999a; Castellsague et al., 1995; Thompson et al., 2001;  
23 Tobías et al., 1999).

24 Hospital admission studies that specifically examined asthmatics were fewer in number  
25 than those that examined total respiratory diseases. Associations were noted in all age groups in  
26 studies conducted in Seattle, WA (Sheppard, 2003), New Jersey (Weisel et al., 2002), Toronto,  
27 Canada (Burnett et al., 1999), London, England (Anderson et al., 1998), Brisbane, Australia  
28 (Petroeschovsky et al., 2001), and Hong Kong (Wong et al., 1999a). However, several other  
29 studies, mostly examining the effect of O<sub>3</sub> on asthmatic children, did not observe a significant  
30 relationship (Gouveia and Fletcher, 2000a; Lin et al., 2003; Morgan et al., 1998; Nauenberg and  
31 Basu, 1999; Schouten et al., 1996).

1 Acute mortality related to asthma was examined in Barcelona, Spain (Saez et al., 1999;  
2 Sunyer et al., 2002). Severe asthmatics with more than one asthma emergency visit showed the  
3 strongest mortality associations with O<sub>3</sub> (Sunyer et al., 2002).

4 Recent reports from longitudinal cohort studies in California have reported associations  
5 between the onset of asthma and long-term O<sub>3</sub> exposures (Greer et al., 1993; McConnell et al.,  
6 2002; McDonnell et al., 1999). In adult studies, associations were seen in males but not females  
7 (Greer et al., 1993; McDonnell et al., 1999). Among children residing in high O<sub>3</sub> communities,  
8 McConnell et al. (2002) observed that asthma risk was elevated for those who played three or  
9 more sports as compared with those who did not play sports. Playing sports may indicate  
10 outdoor activity and an increased ventilation rate which may lead to increased dose of O<sub>3</sub>. These  
11 outcomes would benefit from replication in other cohorts in regards to indicating weight of a  
12 causal interpretation.

13 A few studies provide limited discussion of concentration-response functions and  
14 thresholds. In the eight U.S. urban cities study, the odds ratios for incidence of  $\geq 10\%$  decline in  
15 morning PEF and incidence of morning symptoms when excluding days with 8-h avg O<sub>3</sub> greater  
16 than 80 ppb were nearly identical to those including data from all days (Mortimer et al., 2002).  
17 In the New England asthma panel study (Gent et al., 2003), some of the associations for  
18 symptoms occurred at 1-h max O<sub>3</sub> levels below 60 ppb. In the St. John, Canada study (Stieb  
19 et al., 2003), an effect of O<sub>3</sub> on emergency department visits was reported with evidence of a  
20 threshold somewhere in the range below a 1-h max O<sub>3</sub> of 75 ppb in the 15 years and over age  
21 group.

22 Overall, asthma subjects have been examined across most health endpoints of interest. The  
23 results reported in these studies vary with some indicating a positive excess risk associated with  
24 O<sub>3</sub>. While no endpoint in itself seems to indicate an unquestionable demonstration of an  
25 association, studies with adequate sample size and power consistently provide positive results,  
26 especially during the summer months when higher O<sub>3</sub> levels occur. This view is strengthened as  
27 positive results are obtained cohesively across the varied outcomes. Therefore, based on the  
28 evidence, it seems prudent to consider asthmatics as a susceptible group that requires protection  
29 from O<sub>3</sub> exposures.  
30

### 7.6.7.2 Age-Related Differences in Ozone Effects

Several mortality studies have investigated age-related differences in O<sub>3</sub> effects. Among the studies that observed positive associations between O<sub>3</sub> and mortality, a comparison of all age or younger age ( $\leq 65$  years of age) O<sub>3</sub>-mortality risk estimates to that of the elderly population ( $>65$  years) indicates that, in general, the elderly population is more susceptible to O<sub>3</sub> effects (Borja-Aburto et al. 1997; Bremner et al., 1999; Gouveia and Fletcher 2000b; O'Neill et al., 2004; Simpson et al., 1997; Sartor et al., 1995; Sunyer et al., 2002). For example, a study by Gouveia and Fletcher (2000b) examined the O<sub>3</sub>-mortality effect by age in São Paulo, Brazil. There were 151,756 deaths for all non-violent causes over the period of 1991 to 1993, of which 49% occurred in the elderly. Among all ages, O<sub>3</sub> was associated with a 0.6% (95% CI: -0.8, 2.0) excess risk in all cause mortality per 40 ppb increase in 1-h max O<sub>3</sub>. In comparison, in the elderly population, the O<sub>3</sub>-mortality risk estimate was nearly threefold greater, 1.7% (95% CI: 0.0, 3.3). Similarly, a Mexico City study found that O<sub>3</sub>-mortality risk estimates were 1.3% (95% CI: 0.04, 2.6) and 2.8% (95% CI: 1.0, 4.6) per 20 ppb increase in 24-h avg O<sub>3</sub> concentration in all ages and the elderly, respectively (O'Neill et al., 2004).

The meta-analysis by Bell et al. (2005) found a larger effect estimate for the elderly (2.92% [95% PI: 1.34, 4.51] per 20 ppb increase in 24-h avg O<sub>3</sub>) than for all ages (1.75% [95% PI: 1.10, 2.37]). In the large U.S. 95 communities study (Bell et al., 2004), effect estimates were slightly higher for those aged 65 to 74 years, 1.40% (95% PI: 0.56, 2.25) excess risk per 20 ppb increase in 24-h avg O<sub>3</sub>, compared to individuals less than 65 years and 75 years or greater, 1.00% (95% PI: 0.20, 1.85) and 1.04% (95% PI: 0.36, 1.75), respectively, using a constrained distributed 7-day lag model. Bell et al. (2004) notes that despite somewhat similar effect estimates, the absolute effect of O<sub>3</sub> is substantially greater in the elderly population due to the higher underlying mortality rates, which leads to a larger number of extra deaths for the elderly compared to the general population.

Few mortality studies examined another potentially susceptible age group, young children under the age of 5 years. The results were mixed, with one Mexico City study showing a lower risk of O<sub>3</sub>-related all cause mortality in young children compared to all ages and the elderly (Borja-Aburto et al., 1997) and one São Paulo, Brazil study showing a greater risk in respiratory mortality in young children compared to the elderly (Gouveia and Fletcher, 2000b). It should be

1 noted that approximately 10% of mortality occurred in young children, thus the statistical power  
2 to study the O<sub>3</sub> effect in this age group was limited.

3 With respect to age-specificity of associations between O<sub>3</sub> and acute respiratory  
4 hospitalizations or emergency department visits, no clear pattern emerges from recent studies.  
5 Associations have been reported for all ages (Anderson et al., 1997; Burnett et al., 1995, 1997b,  
6 1999; Weisel et al., 2002), adults or elderly (Burnett et al., 1997a; Delfino et al., 1997b, 1998b;  
7 Moolgavkar et al., 1997; Schwartz et al., 1996; Yang et al., 2003), and children (Burnett et al.,  
8 2001; Gouveia and Fletcher, 2000a; Lin et al., 1999; Pönkä and Virtanen, 1996; Tolbert et al.,  
9 2000; Yang et al., 2003). Interestingly, studies that have examined effects in multiple age strata  
10 often have seen effects only in non-pediatric strata (Delfino et al., 1997b, 1998b; Stieb et al.,  
11 1996; Jones et al., 1995). Several studies that focused on children did not report significant O<sub>3</sub>  
12 effects, though in some cases these studies are limited by small size, inadequate control of  
13 seasonal patterns, or very low O<sub>3</sub> levels (Lierl and Hornung, 2003; Lin et al., 2003; Thompson  
14 et al., 2001). If O<sub>3</sub> is causally related to exacerbations of respiratory diseases leading to hospital  
15 usage, one would expect to see effects most prominently among children, for whom asthma is  
16 more prevalent and O<sub>3</sub> exposures may be greater. However, once again, children only comprised  
17 of about 20% of the total hospitalizations, which limits the power to examine age-specific O<sub>3</sub>  
18 effects.

19 A few field studies compared the effect of O<sub>3</sub> in different age groups. Korrick et al. (1998)  
20 examined changes in FEV<sub>1</sub> and FVC related to O<sub>3</sub> exposure in a group of hikers ranging in age  
21 from 18 to 64 years, and found that there was no association between O<sub>3</sub> responsiveness and age.  
22 Brauer et al. (1996), in a study of berry pickers aged 10 to 69 years, also observed that subject  
23 age was not significantly associated with O<sub>3</sub>-related changes in lung function. However, a study  
24 by Höpfe et al. (1995a, 2003) observed that children, but not seniors (69 to 95 years of age),  
25 experienced a decline in lung function associated with O<sub>3</sub> exposure. The results by Höpfe et al.  
26 are consistent with the diminishing responses to O<sub>3</sub> exposure with increasing age observed in  
27 clinical studies. The clinical studies by Drechsler-Parks (1995) and Bedi et al. (1989) found that  
28 subjects aged 56 to 89 years had markedly reduced responses to O<sub>3</sub> exposure compared to young  
29 adults.

30 Many field studies focused on the effect of O<sub>3</sub> on the respiratory health of school children.  
31 In general, children experienced decrements in pulmonary function parameters, including PEF,

1 FEV<sub>1</sub>, and FVC (Castillejos et al., 1995; Chen et al., 1999; Gielen et al., 1997; Gold et al., 1999;  
2 Jalaludin et al., 2000; Mortimer et al., 2002; Romieu et al., 1996; Thurston et al., 1997).  
3 Increases in respiratory symptoms (Delfino et al., 2003; Gold et al., 1999; Neas et al., 1995;  
4 Romieu et al., 1996, 1997; Thurston et al., 1997) and asthma medication use (Delfino et al.,  
5 1996; Just et al., 2002; Ostro et al., 2001; Thuston et al., 1997) also were observed in children.  
6 These respiratory health effects were observed in both healthy and asthmatic children. In one  
7 German study (Höppe et al., 2003), juvenile asthmatics and healthy children were found to be  
8 particularly susceptible to O<sub>3</sub> effects on lung function. Approximately 20% of the children and  
9 asthmatics experienced a greater than 10% change in FEV<sub>1</sub>, compared to only 5% of the elderly  
10 population and athletes.

11 The American Academy of Pediatrics (2004) notes that children and infants are among the  
12 most susceptible to many air pollutants, including O<sub>3</sub>. Eighty percent of alveolar are formed  
13 postnatally and changes in the lung continue through adolescence (Dietert et al., 2000). Children  
14 spend more time outdoors, which results in increased exposure to air pollutants (Wiley et al.,  
15 1991a,b). Further, children have a high minute ventilation and high levels of physical activity  
16 which increase their dose (Plunkett et al., 1992).

17 Collectively, there is supporting evidence of age-related differences in susceptibility to O<sub>3</sub>  
18 health effects. The elderly population (>65 years of age) appear to be at increased risk of  
19 O<sub>3</sub>-related mortality and hospitalizations, and children (<18 years of age) experience other  
20 potentially adverse respiratory health outcomes with increased O<sub>3</sub> exposure. One epidemiologic  
21 study also found that the lung function response to O<sub>3</sub> exposure may be diminished in elderly  
22 populations; this finding is further supported by evidence from clinical studies.

### 23 24 **7.6.8 Summary of Key Findings and Conclusions Derived From Ozone** 25 **Epidemiologic Studies**

26 In the previous 1996 O<sub>3</sub> AQCD, there was considerable evidence of O<sub>3</sub>-related respiratory  
27 health effects from individual-level camp and exercise studies, as well as some consistent  
28 evidence from time-series studies of emergency room visits and hospitalizations. Since the 1996  
29 document, more field studies have been conducted, with some emphasis on additional outcome  
30 markers such as respiratory symptoms and asthma medication use. Another significant addition  
31 to the current O<sub>3</sub> AQCD is the substantial number of short-term O<sub>3</sub> mortality studies. The recent



1 publication of an analysis examining the relationship between O<sub>3</sub> and mortality in 95 U.S.  
2 communities (Bell et al., 2004) and three meta-analysis on O<sub>3</sub>-mortality associations (Bell et al.,  
3 2005; Ito et al., 2005; Levy et al., 2005) also contribute significantly to the evidence base.  
4 Considering the wide variability in possible study designs and statistical model specification  
5 choices, the reported O<sub>3</sub> risk estimates for the various health outcomes are in reasonably good  
6 agreement. In the case of O<sub>3</sub>-mortality time-series studies, combinations of choices in model  
7 specifications (the number of weather terms and degrees of freedom for smoothing of mortality-  
8 temporal trends) alone may explain the extent of the difference in O<sub>3</sub> risk estimates across  
9 studies. As use of time-series studies to investigate air pollution effects has become more  
10 common, there has been a great effort to evaluate the issues surrounding these studies.

11 In this section, conclusions regarding O<sub>3</sub> health effects from the epidemiologic evidence  
12 and the issues that may affect the interpretation of the effect estimates are briefly summarized.  
13 A more integrative synthesis of all relevant information will be presented in Chapter 8 of this  
14 AQCD.

- 15  
16 (1) Field/panel studies of acute O<sub>3</sub> effects. Results from recent field/panel studies  
17 continue to confirm that short-term O<sub>3</sub> exposure is associated with acute decrements  
18 in lung function and increased respiratory symptoms, particularly in children and  
19 asthmatics. There is also suggestive evidence that O<sub>3</sub> is related to increased asthma  
20 medication use. Taken together with the evidence from controlled human exposure  
21 studies, O<sub>3</sub> is likely causally related to the various respiratory health outcomes. The  
22 current evidence is limited but supportive of a potential effect of O<sub>3</sub> on heart rate  
23 variability, ventricular arrhythmias, and the incidence of myocardial infarctions.  
24
- 25 (2) Acute O<sub>3</sub> effects on emergency department visits and hospitalizations. Large  
26 multicity studies, as well as many studies from individual cities have reported an  
27 association of O<sub>3</sub> concentrations with respiratory and cardiovascular hospital  
28 admissions. Studies using year-round data noted some inconsistencies in the O<sub>3</sub>  
29 effect on daily hospitalizations. However, studies with data restricted to the summer  
30 or warm season, in general, indicated positive and robust associations between  
31 ambient O<sub>3</sub> concentrations and cardiopulmonary hospital admissions. Results for  
32 emergency department visits are less consistent.  
33
- 34 (3) Acute O<sub>3</sub> effects on mortality. The majority of the studies suggest an elevated risk  
35 of all cause mortality associated with acute exposure to O<sub>3</sub>, especially in the summer  
36 or warm season when O<sub>3</sub> levels are typically high. Slightly greater O<sub>3</sub> effects were  
37 observed for cardiovascular mortality. Results from a recent, large U.S. multicity  
38 time-series study provide the strongest evidence to-date for O<sub>3</sub> effects on acute  
39 mortality. Recent meta-analyses also showed consistent risk estimates that are

1 unlikely to be confounded by PM; however, future work is needed to better  
2 understand the influence of model specifications on the risk coefficient.

- 3
- 4 (4) Chronic O<sub>3</sub> exposure effects on morbidity and mortality. Fewer studies have  
5 investigated the effect of chronic O<sub>3</sub> exposure on morbidity and mortality. The  
6 strongest evidence is for negative seasonal effects of O<sub>3</sub> on lung function in adults  
7 and children. Less conclusive are longer-term studies investigating the association  
8 of chronic O<sub>3</sub> exposure on yearly lung function, asthma incidence, and respiratory  
9 symptoms. Chronic O<sub>3</sub>-mortality studies observed inconsistencies across exposure  
10 periods, cause-specific mortality outcomes, and gender.
- 11
- 12 (5) Exposure assessment. Exposure misclassification may result from the use of  
13 stationary ambient monitors to determine exposure in population studies. Although  
14 central ambient monitors do not explain the variance of individual personal  
15 exposures, significant correlations are found between aggregate personal O<sub>3</sub>  
16 measurements and O<sub>3</sub> concentrations from ambient monitors. A simulation study  
17 indicated that the use of ambient monitor data will tend to underestimate the O<sub>3</sub>  
18 effect. A better understanding of the factors that affect the relationship between  
19 ambient concentrations and personal exposures will improve interpretation of the O<sub>3</sub>  
20 effect estimates.
- 21
- 22 (6) Ozone exposure indices. The three most commonly used daily O<sub>3</sub> exposure indices,  
23 1-h max O<sub>3</sub>, 8-max O<sub>3</sub>, and 24-h avg O<sub>3</sub>, were found to be highly correlated in  
24 studies conducted in various regions. In addition, the effect estimates and  
25 significance of associations across all health outcomes were comparable when using  
26 the standardized distributional increment of 40 ppb, 30 ppb, and 20 ppb for 1-h max  
27 O<sub>3</sub>, 8-h max O<sub>3</sub>, and 24-h avg O<sub>3</sub>, respectively.
- 28
- 29 (7) Lag structures for O<sub>3</sub> exposure and effect. The lag time between O<sub>3</sub> exposure and  
30 effect may differ depending on various factors such as the specific health outcome  
31 of interest, the mechanism of effect, and preexisting health conditions. The majority  
32 of the studies found an immediate O<sub>3</sub> effect, with the strongest associations  
33 observed between health outcomes and O<sub>3</sub> exposure on the same day and/or  
34 previous day. Some studies found large cumulative effects of O<sub>3</sub> over longer  
35 lag periods, indicating that multiday lags also may be relevant for some health  
36 outcomes, including mortality.
- 37
- 38 (8) Sensitivity to model specifications for temporal trends. Ozone effect estimates  
39 that were reported in studies whose main focus was PM often were calculated using  
40 the same model specifications as PM. While the sensitivity of the O<sub>3</sub> risk estimates  
41 to alternative model specifications has not been thoroughly investigated, limited  
42 evidence indicates that O<sub>3</sub> effects may be robust to various model specifications for  
43 temporal trend adjustment.
- 44
- 45 (9) Influence of seasonal factors. An evaluation of the confounding effects of  
46 meteorologic factors and copollutants on O<sub>3</sub> risk estimates is complicated by their  
47 changing relationships with O<sub>3</sub> across seasons. In addition, seasonal or seasonally-

1 modified factors (e.g., air conditioning use, time spent outdoors) complicate  
2 interpretation of all year effect estimates as they affect the relationship between  
3 ambient concentrations and personal exposures. Given the potentially significant  
4 influence of season, season-specific analyses are more informative in assessing O<sub>3</sub>  
5 health risks.  
6

- 7 (10) Confounding by copollutants. Multipollutant regression models often are used to  
8 adjust for confounding by copollutants. Although there is some concern regarding  
9 the use of multipollutant models given the varying concavity across pollutants,  
10 results generally suggest that the inclusion of copollutants into the models do not  
11 substantially affect O<sub>3</sub> risk estimates. These findings indicate that effects of O<sub>3</sub> on  
12 various health outcomes are robust and independent of the effects of other  
13 copollutants.  
14
- 15 (11) Concentration-response function. In the limited mortality and morbidity studies that  
16 have specifically examined the O<sub>3</sub> concentration-response relationship, the evidence  
17 is inconclusive regarding the presence of an effect threshold. Factors such as  
18 exposure measurement error may reduce the ability to detect a threshold in  
19 population studies.  
20
- 21 (12) Heterogeneity of O<sub>3</sub> health effects. Consistent O<sub>3</sub> effect estimates generally were  
22 observed for mortality, hospitalizations, and other respiratory health outcomes in  
23 multicity studies. Some of the observed geographic heterogeneity in effects may be  
24 attributable to the differences in relative personal exposure to O<sub>3</sub>, which is affected  
25 by factors such as air conditioning prevalence and activity patterns, and the varying  
26 concentrations and compositions of copollutants present by region.  
27
- 28 (13) Ozone health effects in asthmatics. The effects of O<sub>3</sub> on asthmatics have been  
29 examined widely in both time-series studies and panel studies. Associations of  
30 O<sub>3</sub> with various respiratory health outcomes, including lung function declines,  
31 increased respiratory symptoms, and emergency department visits, were observed.  
32 These findings, along with the pathophysiologic understanding of asthma as a  
33 chronic inflammatory disease, indicate that asthmatics may be a susceptible  
34 population that requires protection from O<sub>3</sub> exposures.  
35
- 36 (14) Age-related differences in O<sub>3</sub> health effects. Supporting evidence exists for  
37 heterogeneity in the effects of O<sub>3</sub> by age. The elderly population (>65 years of age)  
38 appear to be at greater risk of O<sub>3</sub>-related mortality and hospitalizations compared to  
39 all age or younger populations. In addition, potentially adverse respiratory health  
40 outcomes were associated with O<sub>3</sub> exposure in children (<18 years of age). One  
41 epidemiologic study provided limited evidence that lung function responses to O<sub>3</sub>  
42 exposure is diminished in the elderly population.  
43  
44

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# 8. INTEGRATIVE SYNTHESIS: OZONE EXPOSURE AND HEALTH EFFECTS

## 8.1 INTRODUCTION

This integrative synthesis is structured to provide a coherent framework for the assessment of health risks associated with human exposures to ambient surface-level (tropospheric) ozone (O<sub>3</sub>) in the United States. The main goal of the chapter is to integrate newly available scientific information with key findings and conclusions from the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996a), so as to address issues central to the EPA's assessment of evidence needed to support the current review of the primary O<sub>3</sub> NAAQS. The integrated assessment of key findings and conclusions provided here and elsewhere in this document with regard to O<sub>3</sub> exposure and health effects will be drawn upon and their policy implications considered in an Ozone Staff Paper prepared by EPA's Office of Air Quality Planning and Standards (OAQPS). The analyses provided in that Staff Paper aim to "bridge the gap" between scientific assessments in this criteria document and judgments required of the EPA administrator in evaluating whether to retain or, possibly, to revise the current primary O<sub>3</sub> NAAQS. Other types of scientific information concerning ambient O<sub>3</sub> welfare effects (i.e., tropospheric O<sub>3</sub> effects on vegetation and ecosystems, relationships to surface-level solar UV flux/climate changes, and effects on man-made materials) are assessed in ensuing Chapters 9, 10, and 11. That information will also be considered in the OAQPS staff paper in posing options regarding the secondary O<sub>3</sub> NAAQS.

As discussed in Chapter 2 of this document, O<sub>3</sub> found in the earth's troposphere generally originates from photochemical reactions that are predominantly catalyzed by the interaction of sunlight with precursor pollutants, especially nitrogen oxides (NO<sub>x</sub>) and hydrocarbons such as volatile organic compounds (VOCs), emitted by surface-level mobile and stationary sources. Other photochemical oxidants, such as peroxyacetyl nitrate (PAN) and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), are also generated along with O<sub>3</sub> by such atmospheric interactions. In addition to the tropospheric O<sub>3</sub> generated by these interactions, some O<sub>3</sub> is found near the earth's surface as the result of its downward transport from the stratosphere. However, in contrast to stratospheric O<sub>3</sub>, which plays an important role in maintaining the habitability of the planet by shielding the

1 surface from harmful solar ultraviolet (UV) radiation, tropospheric O<sub>3</sub> at the surface can exert  
2 adverse effects on humans, nonhuman animal species, and vegetation. As was the case for  
3 previous O<sub>3</sub>-related NAAQS criteria revisions, the present criteria document focuses mainly on  
4 the assessment of health and welfare effects resulting from exposures to surface-level  
5 concentrations of tropospheric O<sub>3</sub>, whereas less attention is accorded to the distinctly much more  
6 limited available information on other photochemical oxidants, e.g., PAN or H<sub>2</sub>O<sub>2</sub>.

7 Based on the criteria review completed in 1978, the original primary and secondary  
8 NAAQS set in 1971 for total photochemical oxidants were revised in 1979 to focus on O<sub>3</sub> as the  
9 indicator for new primary and secondary standards that were attained when the expected number  
10 of days per calendar year with maximum 1-h average O<sub>3</sub> concentrations >0.12 ppm did not  
11 exceed one. The NAAQS for ambient O<sub>3</sub> were revised in 1997 by replacing the 1-h standards  
12 with an 8-h primary standard that is met when the 3-year average of the annual fourth highest  
13 daily maximum 8-h average concentration is <0.08 ppm. The new 1997 primary standard was  
14 based on various scientific supportive data from experimental human exposure, animal  
15 toxicological and epidemiological studies, as assessed in the 1996 O<sub>3</sub> AQCD and in the 1996 O<sub>3</sub>  
16 Staff Paper (U.S. Environmental Protection Agency, 1996b).

### 17 18 **8.1.1 Chapter Organization**

19 In addition to providing the above brief background information regarding prior O<sub>3</sub>  
20 NAAQS reviews (including the 1997 EPA revision of the O<sub>3</sub> NAAQS), this first section  
21 (8.1 Introduction) of the integrative synthesis chapter aims to orient the reader to the  
22 organization and content of the chapter. The next section (Section 8.2) focuses on air quality  
23 trends and current ambient O<sub>3</sub> levels to help provide context for the ensuing discussions of O<sub>3</sub>  
24 exposures and associated health effects. The subsequent sections (8.3, 8.4, and 8.5)  
25 then integrate newly available key scientific information assessed in Chapters 4 through 7 of  
26 this document, including integration of information on O<sub>3</sub> dosimetry, toxicological information  
27 derived from controlled human exposure and laboratory animal studies, and epidemiologic  
28 evidence.

29 These sections collectively address the following topics: (1) ambient O<sub>3</sub> exposures,  
30 personal exposures, and dosimetric considerations; (2) experimental studies on toxicological  
31 responses to acute O<sub>3</sub> exposures in humans (clinical studies) and both acute and chronic effects

1 in animals; (3) epidemiological evidence for associations between O<sub>3</sub> exposure of human  
2 populations and health effects and the strength and robustness of these associations;  
3 (4) integration of the experimental and epidemiological evidence; (5) biological mechanisms and  
4 other evidence useful in judging the plausibility of adverse health effects being associated with  
5 human exposures to ambient O<sub>3</sub> levels encountered in the United States; and (6) identification of  
6 susceptible and vulnerable populations likely at increased risk for O<sub>3</sub>-related health effects and  
7 numbers of people potentially falling in such categories in the United States.

8 The present chapter mainly focuses on discussion of new scientific information that has  
9 become available since the 1996 O<sub>3</sub> criteria review that supported EPA's revision of the O<sub>3</sub>  
10 NAAQS in 1997. This includes assessment of information published or accepted for publication  
11 in peer-reviewed open literature mainly through December 2004, with a few particularly  
12 pertinent and important studies published beyond that point also being considered.

13 Important data gaps and uncertainties that still exist with regard to various important issues  
14 and research needs are also briefly noted for some key areas. Detailed discussion of such  
15 research needs is beyond the scope of this document; however, such discussion is typically  
16 undertaken later as part of EPA efforts focused on identification of O<sub>3</sub> research needs and  
17 development of associated research planning documents.

## 18 19 20 **8.2 AMBIENT OZONE AIR QUALITY IN UNITED STATES**

### 21 **8.2.1 Current Ozone Concentrations and Spatial Patterns**

22 Ambient air O<sub>3</sub> is monitored in the United States during 'ozone seasons', which vary in  
23 length depending on location. The ozone season extends all year in the Southwest. In most  
24 other areas of the country, O<sub>3</sub> is monitored typically from April to October. However, O<sub>3</sub> is  
25 monitored throughout the year in many urban areas, in as much as O<sub>3</sub> is present the year round  
26 not only in polluted areas but in clean areas as well. The median of the daily maximum 8-h  
27 average O<sub>3</sub> concentration in the United States, averaged over May to September from 2000 to  
28 2004 for all U.S. counties, was 0.049 ppm. In 95% of all counties, the median of the daily  
29 maximum 8-h average O<sub>3</sub> concentration was less than 0.057 ppm. However, it should be noted  
30 that most monitors are located in the East. The daily maximum 1-hour concentrations were



1 typically much higher in large urban areas or in areas downwind of them. For example, in  
2 Houston, TX they approached 0.20 ppm during this period. Daily 1-hour maximum ozone  
3 concentrations were lower in the rest of the country, but were still above 0.12 ppm in many  
4 locations. Eight hour daily maximum concentrations were not as high, but tend to be highly  
5 correlated with 1-hour daily maximums.

6 Within individual MSAs, O<sub>3</sub> concentrations tend to be well correlated across monitoring  
7 sites, although spatial variations in concentrations can be substantial. In many city centers, O<sub>3</sub>  
8 concentrations tend to be lower than in either upwind or downwind areas, largely due to reaction  
9 of O<sub>3</sub> with NO emitted by motor vehicles. For example, much lower O<sub>3</sub> concentrations overall  
10 are found in downtown Los Angeles (e.g., in Lynwood) than at sites located further downwind  
11 (e.g., in San Bernadino). The much higher downwind levels are formed from photochemical  
12 reactions involving the urban emissions, including products formed as the result of reactions  
13 titrating O<sub>3</sub> in the urban core. Thus, O<sub>3</sub> concentrations tend to be higher downwind of urban  
14 centers, and they decrease again in going to areas that are more remote from precursor sources.  
15 Likewise, surface-level O<sub>3</sub> can be depleted in rural areas close to NO sources, such as highways  
16 and powerplants.

## 17

### 18 **8.2.2 Diurnal and Seasonal Variations**

19 Ozone concentrations typically tend to peak in early to mid-afternoon in areas where there  
20 is strong photochemical activity and to peak later in the afternoon or during early evening in  
21 areas where transport is more important in determining the O<sub>3</sub> abundance. Summertime maxima  
22 in O<sub>3</sub> concentrations occur in those U.S. areas where substantial photochemical activity acts  
23 on O<sub>3</sub> precursors emitted as the result of human activities. Monthly maxima can occur anytime  
24 from June through August. However, springtime maxima are observed in some National Parks,  
25 mainly in the western United States, and at a number of other relatively unpolluted monitoring  
26 sites throughout the Northern Hemisphere. For example, the highest O<sub>3</sub> concentrations at  
27 Yellowstone National Park tend to occur during April and May. Typically, monthly minima  
28 tend to occur from November through February at polluted sites and during the fall at relatively  
29 remote sites.

### 8.2.3 Long-Term Trends

National attention started to be focused in the 1940s on O<sub>3</sub> and associated photochemical smog in the Los Angeles area. Prior to the adoption of stringent emissions controls, peak levels of O<sub>3</sub> were consistently higher in the Los Angeles area than are currently observed. For example, in 1958, peak O<sub>3</sub> concentrations measured in Los Angeles were about 0.6 ppm but have declined since then, although not at a steady rate. Peak O<sub>3</sub> levels of 0.2 to 0.5 ppm were still found at some locations in the Los Angeles basin during the 1970s. For example, on two days (October 13 and 14) during a 1978 episode, Tuazon et al. (1981) observed peak 1-h averaged values of O<sub>3</sub> of nearly 0.4 ppm and nearly 0.5 ppm. Currently, peak 1-h and 8-h average O<sub>3</sub> concentrations are about 0.17 and 0.15 ppm in the Los Angeles basin (cf. Figures 3-10 and 3-11). High O<sub>3</sub> levels were also earlier found throughout the rest of the United States as well, but peak O<sub>3</sub> levels have also gradually declined across the country during the 1980s. However, during one particularly hot summer (of 1988) in the East, peak 1-h O<sub>3</sub> concentrations of about 0.2 ppm were observed in many eastern U.S. cities (U.S. Environmental Protection Agency, 1990).

Historically high O<sub>3</sub> concentrations, as noted above, have not only been observed in the United States. For example, during an episode in Great Britain in 1976, peak O<sub>3</sub> levels exceeded 0.25 ppm and daily maximum 8-h O<sub>3</sub> concentrations were above 0.1 ppm for 18 consecutive days at one rural site (Wayne, 1991). Also, concentrations of O<sub>3</sub> in the range found in Los Angeles during the 1970s are still found in Mexico City.

Nationwide, 2nd highest 1-h ozone concentrations in the United States have decreased dramatically during the past several decades, i.e., by approximately 29 percent from 1980 to 2003 and 16 percent from 1990 to 2003. Also, 4th highest 8-h O<sub>3</sub> concentrations decreased by approximately 21 percent since 1980 and 9 percent since 1990 (U.S. Environmental Protection Agency, 2003). Trends in metrics for evaluating compliance with the O<sub>3</sub> NAAQS (i.e., changes in the 4th highest O<sub>3</sub> concentration) can be found in EPA's "National Air Quality and Emissions Trends Reports". These reports indicate that the 4th highest O<sub>3</sub> concentrations are still decreasing nationwide, but the rate of decrease has slowed since 1990. However, such trends have not been uniform across the United States. In general, reductions in the O<sub>3</sub> metrics given above have been largest in New England and in states along the West Coast and smallest in midwestern states. Downward trends in California O<sub>3</sub> concentrations have been driven mainly by notable decreases in Southern California, with reductions in other areas not being as large.

1 Trends in peak O<sub>3</sub> metrics do not necessarily reflect changes in O<sub>3</sub> values across the middle of  
2 the distribution of O<sub>3</sub> concentrations. Of note, O<sub>3</sub> concentrations towards the center of its  
3 nationwide distribution have not shown much change, and there are some indications that O<sub>3</sub>  
4 concentrations at the lower end of the distribution may even be increasing.  
5

#### 6 **8.2.4 Interrelationships Between Ozone and Other Ambient Pollutants**

7 Data on ambient concentrations of other oxidants (e.g., H<sub>2</sub>O<sub>2</sub>, PAN) and oxidation products  
8 (e.g., HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>) in the atmosphere are not nearly as abundant as they are for O<sub>3</sub>. Because  
9 data for such species are usually obtained only as part of specialized field studies, it is difficult to  
10 relate observed ambient O<sub>3</sub> concentrations to ambient levels of other oxidant species or oxidation  
11 products. In general, such secondary species are expected to be at least moderately positively  
12 correlated with O<sub>3</sub>. On the other hand, primary species are expected to be more highly correlated  
13 with each other than with secondary species, provided that the primary species originate from  
14 common sources in given areas. Measurements of gas phase oxidants conducted as part of the  
15 Southern Oxidants Study (SOS) indicated combined hydroperoxide (H<sub>2</sub>O<sub>2</sub>, CH<sub>3</sub>OOH, and  
16 HOCH<sub>2</sub>OOH) concentrations typically in the range of several ppb. Concentrations of PAN, PPN  
17 and MPAN also observed during the SOS likewise indicated combined concentrations in the  
18 range of several ppb. Oxidants are also present in airborne cloud droplets, rain drops, and  
19 particulate matter (PM). A few measurements of reactive oxygen species (expressed as  
20 equivalent H<sub>2</sub>O<sub>2</sub>) in ambient fine PM indicated levels of less than 1% of those for ambient O<sub>3</sub> on  
21 a molar basis. However, it should be noted that these measurements are potentially subject to  
22 both positive and negative artifacts.

23 Because PM is not a single distinct chemical species, but rather a mix of primary and  
24 secondary species, relationships between ambient O<sub>3</sub> and PM concentrations can be quite  
25 complex. As an example of this complexity, PM<sub>2.5</sub> concentrations positively correlated with O<sub>3</sub>  
26 during the summer, but negatively correlated with O<sub>3</sub> during the winter at Ft. Meade, MD. Also,  
27 Ito et al. (2005) examined relationships between PM<sub>10</sub> and O<sub>3</sub> on a seasonal basis in several  
28 urban areas (cf. Figure 7-24). Seasonal relationships with ambient O<sub>3</sub> similar to those at  
29 Ft. Meade were found, reflecting the dominant contribution of PM<sub>2.5</sub> to PM<sub>10</sub> in the urban areas  
30 studied (although PM<sub>10</sub> generally contains a higher fraction than does PM<sub>2.5</sub> of primary [mainly  
31 crustal] material). Possibly contributing to the higher correlations observed between fine PM

1 and O<sub>3</sub> in the summer is the fact that O<sub>3</sub> can contribute to formation of submicron particles via  
2 interactions with various other atmospheric constituents present, such as terpenes, and other  
3 biogenically derived hydrocarbons from trees, other vegetation, and wood products. Formation  
4 of ultrafine particles by this mechanism is most likely to occur during afternoons of summer  
5 days when temperatures and O<sub>3</sub> concentrations are sufficiently elevated to facilitate O<sub>3</sub> reactions  
6 with increased amounts of terpenes emitted from vegetation. Bursts of ultrafine particle  
7 formation have been observed repeatedly in both urban and rural air. Woo et al. (2001), for  
8 example, reported rapid formation of ultrafine particles in the ambient air of Atlanta typically  
9 around noon in both summer and winter. The mechanisms underlying such ultrafine particle  
10 formation events may also involve other atmospheric reactions that are related to O<sub>3</sub> formation,  
11 such as the nucleation of H<sub>2</sub>SO<sub>4</sub> (produced by oxidation of SO<sub>2</sub>) and, probably, NH<sub>3</sub>.

### 13 **8.2.5 Policy Relevant Background (PRB) Ozone Concentrations**

14 Background O<sub>3</sub> concentrations used for NAAQS-setting purposes are referred to as Policy  
15 Relevant Background (PRB) O<sub>3</sub> concentrations. Policy Relevant Background concentrations are  
16 those that would occur in the United States in the absence of anthropogenic emissions in  
17 continental North America (defined here as the United States, Canada, and Mexico). Such  
18 PRB O<sub>3</sub> concentrations include contributions from natural sources everywhere in the world and  
19 from anthropogenic sources outside these three countries. For the purpose of informing O<sub>3</sub>  
20 NAAQS decisions, EPA focuses on assessing risks to human health and environmental effects  
21 from O<sub>3</sub> levels in excess of PRB concentrations. Issues concerning the methodology for  
22 estimating PRB O<sub>3</sub> concentrations are discussed in detail in Section AX3.9 of Annex AX3.

23 Contributions to PRB O<sub>3</sub> include photochemical reactions involving natural emissions of  
24 VOCs, NO<sub>x</sub>, and CO, as well as the long-range transport of O<sub>3</sub> and its precursors from outside  
25 North America and the stratospheric-tropospheric exchange (STE) of O<sub>3</sub>. Processes involved in  
26 STE are described in detail in Section AX2.3 of Annex AX2. Natural sources of O<sub>3</sub> precursors  
27 include biogenic emissions, wildfires, and lightning. Biogenic emissions from agricultural  
28 activities are not considered in the formation of PRB O<sub>3</sub>.

29 Currently, estimates of PRB O<sub>3</sub> concentrations are based on predictions generated by the  
30 global scale, three dimensional, chemical transport model GEOS-CHEM (Fiore et al., 2003).  
31 Estimates of PRB O<sub>3</sub> concentrations cannot be derived solely from measurements of O<sub>3</sub> at

1 relatively unpolluted sites because of long-range transport from anthropogenic source regions  
2 within North America. It is impossible to determine sources of O<sub>3</sub> at a particular location  
3 without ancillary data that could be used as tracers of sources or to calculate photochemical  
4 production and loss rates for O<sub>3</sub>. Policy relevant background O<sub>3</sub> concentrations vary as a  
5 function of season, altitude, and total surface O<sub>3</sub> concentration, with PRB O<sub>3</sub> concentrations at  
6 the surface generally falling in the range of 0.015 to 0.035 ppm from 1300 to 1700 local time  
7 and tending to decline under conditions conducive to O<sub>3</sub> episodes. The PRB concentrations are  
8 highest during spring and decline into summer; and higher values tend also to occur at higher  
9 elevations during the spring due to contributions from hemispheric pollution and stratospheric  
10 intrusions. The contribution to surface O<sub>3</sub> by stratospheric intrusions is typically well below  
11 0.020 ppm. Stohl (2001) and Sprenger et al. (2003) found that the maximum probability of  
12 stratospheric intrusions reaching the 800 hPa level (~1800 m) was less than 1% and that higher  
13 probabilities (1 to 2%, and 10%) applied for stratospheric intrusions penetrating to the 600 hPa  
14 level (~4100 m) and 500 hPa level (~5400 m), respectively. Thus, stratospheric intrusions only  
15 rarely contribute to elevated surface-level O<sub>3</sub> concentrations at low altitude sites but have a  
16 higher (albeit still low) probability of elevating them at high-altitude sites.

### 19 **8.3 FACTORS AFFECTING HUMAN EXPOSURE TO AMBIENT OZONE**

20 Exposure to O<sub>3</sub> and related photochemical oxidants varies over time due to changes in their  
21 ambient concentrations and because people move between locations having notably different  
22 concentrations. The amount of O<sub>3</sub> delivered to the lung is not only influenced by the ambient  
23 concentration but also by the individual's breathing route and rate. Thus, activity level is an  
24 important consideration in determining the potential O<sub>3</sub> exposure and dose received.

25 The use of data from ambient air monitoring stations is still the most common surrogate for  
26 assigning exposure estimates in epidemiologic studies. Since the primary source of O<sub>3</sub> exposure  
27 is the ambient air, O<sub>3</sub> concentration data from outdoor community monitoring sites should  
28 provide a relative assignment of exposure with time, if: concentrations are relatively uniform  
29 across the region; time-activities pattern are roughly the same across the study population; and  
30 housing characteristics (such as ventilation rates and O<sub>3</sub> sinks contributing to indoor O<sub>3</sub> decay  
31 rates) are relatively constant for the study area. However, because these types of factors often do

1 vary across populations and locations, some error tends to be associated not only with estimates  
2 of the magnitude of O<sub>3</sub> exposure but, also, potentially with relative exposure assignments based  
3 solely on ambient monitoring data. Nevertheless, ambient O<sub>3</sub> monitoring data appear to provide  
4 the most useful index of human O<sub>3</sub> exposure currently available to help characterize health  
5 outcomes associated with O<sub>3</sub> exposures of large population groups.  
6

### 7 **8.3.1 Personal Exposure**

8 Personal O<sub>3</sub> concentrations have been measured for children, outdoor workers, and  
9 individuals with COPD, all being populations potentially susceptible to O<sub>3</sub> or other respiratory  
10 irritants. Outdoor workers can be expected to have somewhat higher O<sub>3</sub> exposures than other  
11 individuals, because they typically spend more time outdoors and often engage in prolonged  
12 moderate and heavy exertion activities. Children also tend to be more active outside and,  
13 therefore, often manifest a higher breathing rate than most adults. However, available exposure  
14 measurement studies are not sufficient to allow for highly confident broad quantitative  
15 generalization about the “typical” magnitudes of observed differences in exposure between the  
16 general population and such potentially susceptible subpopulations.  
17

### 18 **8.3.2 Indoor Concentrations**

19 Apart from only a few specific indoor sources such as photo-copying machines, O<sub>3</sub> indoors  
20 is derived from the infiltration of ambient air from outdoors. Generally, O<sub>3</sub> enters indoor  
21 environments through infiltration from outdoors and through building components, such as  
22 windows, doors, and ventilation systems. Ozone concentrations in indoor environments depend  
23 primarily on the outdoor O<sub>3</sub> concentration, outdoor/indoor infiltration and the air exchange rate  
24 (AER). Once indoors, O<sub>3</sub> reacts on various surfaces and with airborne components of either  
25 indoor or outdoor origin.

26 Indoor O<sub>3</sub> concentrations tend to reflect outdoor concentrations and, hence, are higher  
27 when outdoor O<sub>3</sub> is higher. However, because O<sub>3</sub> reacts indoors with surfaces and other  
28 contaminants, O<sub>3</sub> concentrations are typically lower indoors than outdoors. Gas phase reactions  
29 occurring outdoors also produce other oxidants analogous to the production of photochemical  
30 smog. The extent and rate of production of these other species indoors is a function of indoor O<sub>3</sub>

1 concentrations and the presence of other necessary precursors (i.e., VOCs), along with an  
2 optimal AER.

3 Several studies have measured O<sub>3</sub> concentrations in residences, schools, office buildings  
4 and museums; and typical concentrations varied across all such locations. However, indoor  
5 concentrations generally varied in relationship to the AER in the indoor environment (increasing  
6 with higher AER) and generally tended to be notably lower than outdoor ambient O<sub>3</sub> levels.  
7 For example, one study examining the relationship between O<sub>3</sub> concentrations indoors and  
8 outside of a school in New England reported average O<sub>3</sub> concentrations of 20 ppb (0.020 ppm)  
9 indoors and 40 ppb (0.040 ppm) outdoors. With regard to mobile source microenvironments, as  
10 is the case for other enclosed environments, O<sub>3</sub> exposures depend on the extent of mixing of  
11 outdoor air into the vehicle cabin. Thus, if windows are kept open, O<sub>3</sub> concentrations inside the  
12 vehicle may be expected to approach outdoor values; but, if windows are kept closed and there is  
13 air conditioning, then interior values can be much lower than those outside, especially if  
14 recirculated air is used. For example, in one N.C. study involving police cars with air  
15 conditioning and recirculated air, O<sub>3</sub> concentrations in the vehicle cabin (11.7 ppb average) were  
16 less than half those outside (28.3 ppb average at outdoor monitoring sites in the area).

17 Although concentrations of O<sub>3</sub> may be reduced to lower levels once ambient O<sub>3</sub> enters  
18 indoor environments, it should be kept in mind that the indoor O<sub>3</sub> may interact with other  
19 airborne substances of indoor or outdoor origin that may be present indoors. For example,  
20 Wainman et al. (2000) showed that O<sub>3</sub> reacts with d-limonene, a common component of air  
21 fresheners to produce submicron particles. These particles are found mainly in the size range  
22 from 0.1 to 0.3 μm. Wainman et al. noted that terpenes such as limonene are emitted by wood  
23 products; that they are used as solvents, as odorants in cleaning products, and as air fresheners;  
24 and, because of their widespread uses, their concentrations are often higher indoors than they are  
25 outdoors. In addition to particle formation, Weschler (2004) points out that gas phase products,  
26 such as aldehydes and hydroperoxides, produced by reactions of O<sub>3</sub> with terpenes and other  
27 unsaturated carbon compounds may also be of concern. During the formation of these products,  
28 OH radicals are also produced which can react with compounds that do not react with O<sub>3</sub>. To the  
29 extent that building ventilation rates etc. remain constant between days characterized by high  
30 and low O<sub>3</sub>, the concentrations of these other secondary pollutants formed indoors will tend to be  
31 correlated with ambient O<sub>3</sub>. Thus, ambient O<sub>3</sub> concentrations measured outdoors at community

1 monitoring sites and/or personal O<sub>3</sub> exposure monitor measurements may serve not only as  
2 indices of direct human exposure to O<sub>3</sub> per se, but also as surrogate indices of exposures to  
3 broader O<sub>3</sub>-containing mixtures of ambient or indoor air contaminants.

#### 6 **8.4 SYNTHESIS OF AVAILABLE INFORMATION ON OZONE-** 7 **RELATED HEALTH EFFECTS**

8 The integrated synthesis of the latest available information on O<sub>3</sub>-related health effects  
9 poses large challenges, especially in view of the emergence of certain important new information  
10 since the 1996 O<sub>3</sub> AQCD, which adds greatly to the complexity of an integrative assessment.  
11 Such information includes new findings from:

- 12 • Dosimetry studies that clarify further factors potentially affecting regional distribution  
of O<sub>3</sub> in the respiratory tract of humans and laboratory animals and providing improved  
bases by which to attempt animal-to-human extrapolations of experimentally-observed  
O<sub>3</sub>-induced health effects.
- 13 • Experimental toxicological studies using controlled human exposures and laboratory  
animals aimed at delineating exposure-response relationships and understanding potential  
biochemical mechanisms underlying toxic effects, pathology, and susceptibility;
- 14 • Epidemiological studies, reflecting progress in addressing many research needs identified  
during the last review, as well as raising new issues and reevaluating previously addressed  
issues that remain important in interpreting the body of epidemiological evidence and  
characterization of its strengths and limitations.

15  
16 Previous criteria assessments, including the 1996 O<sub>3</sub> AQCD, found that experimental  
17 studies of controlled human and laboratory animal exposures to O<sub>3</sub> provided the most clear cut  
18 and compelling evidence with regard to characterizing O<sub>3</sub>-related health effects. This section  
19 first summarizes key dosimetry and health related findings derived from the 1996 O<sub>3</sub> AQCD and  
20 then integrates those findings with new information obtained since 1996 from human and animal  
21 experimental studies. Ozone-induced physiological, pathological, cellular and biochemical  
22 alterations are evaluated in order to assess human health effects due to ambient O<sub>3</sub> exposures.  
23 Also, the influence of O<sub>3</sub>-induced changes at cellular and molecular levels are integrated to  
24 elucidate scientific bases for the observed physiological and pathological alterations. These  
25 research results are evaluated in order both to help assess the biological plausibility of health



1 outcome associations observed in epidemiologic studies and to assess the coherence of the  
2 overall body of evidence relevant to O<sub>3</sub>-related health outcome conclusions.

### 3 4 **8.4.1 Key Health-Related Findings and Conclusions from the 1996** 5 **Ozone Air Quality Criteria Document**

6 Based on extensive dosimetric and experimental data as well as growing epidemiologic  
7 evidence available at the time, the 1996 O<sub>3</sub> AQCD arrived at a set of findings and conclusions  
8 stated in relation to answering five key questions regarding potential health effects of ambient O<sub>3</sub>  
9 exposure. In general, the existing evidence was such to warrant a high degree of confidence in  
10 those conclusions derived from experimental (controlled exposure) studies. Considerable  
11 confidence could also be placed in the emerging field/panel studies providing observational  
12 study results substantiating and extending the controlled exposure study findings. Other  
13 epidemiologic studies provided highly suggestive, although less conclusive, indications of  
14 increased morbidity (e.g., as indexed by emergency department visits, hospital admissions, etc.)  
15 and, possibly, mortality being associated with exposure of human populations to ambient O<sub>3</sub>.  
16 The main findings and conclusions derived from the 1996 ozone criteria review are recapitulated  
17 (largely verbatim) below in relation to the five key questions addressed in the summary and  
18 conclusions of the Integrative Synthesis in the 1996 O<sub>3</sub> AQCD.

#### 19 20 ***1. What are the effects of short-term (<8-h) exposures to ozone?***

21 *Short-term O<sub>3</sub> exposure of laboratory animals and humans causes changes in pulmonary*  
22 *function, including tachypnea (rapid, shallow breathing), decreased lung volumes and flows,*  
23 *and increased airway responsiveness to nonspecific stimuli. Increased airway resistance occurs*  
24 *in both humans and laboratory animals, but typically at higher exposure levels than other*  
25 *functional endpoints. In addition, adult human subjects experience O<sub>3</sub> induced symptoms of*  
26 *airway irritation such as cough or pain on deep inspiration. The changes in pulmonary function*  
27 *and respiratory symptoms occur as a function of exposure concentration, duration, and level of*  
28 *exercise. Adult human subjects with mild asthma have qualitatively similar responses in lung*  
29 *volume and airway responsiveness to bronchoconstrictor drugs as nonasthmatics. Respiratory*  
30 *symptoms are also similar, but wheezing is a prevalent symptom in O<sub>3</sub>-exposed asthmatics in*  
31 *addition to the other demonstrated symptoms of airway irritation. Airway resistance, however,*

1 increases relatively more in asthmatics from an already higher baseline. Recovery from the  
2 effects of O<sub>3</sub> on pulmonary function and symptoms is usually complete within 24 h of the end of  
3 exposure, although other responses may persist somewhat longer.

4 • Increased O<sub>3</sub> levels are associated with increased hospital admissions and emergency  
department visits for respiratory causes. Analyses from data in the Northeastern  
United States suggest that O<sub>3</sub> air pollution is associated with a substantial portion (on the  
order of 10 to 20%) of all summertime respiratory hospital visits and admissions.

5 • Pulmonary function in children at summer camps in southern Ontario, Canada, in the  
northeastern United States, and in Southern California is associated with O<sub>3</sub>  
concentration. Meta-analysis indicates that a 0.50-mL decrease in FEV<sub>1</sub> is associated  
with a 1 ppb increase in O<sub>3</sub> concentration. For preadolescent children exposed to  
120 ppb (0.12 ppm) ambient O<sub>3</sub>, this amounts to an average decrement of 2.4 to 3.0% in  
FEV<sub>1</sub>. Similar responses are reported for children and adolescents exposed to O<sub>3</sub> in  
ambient air or O<sub>3</sub> in purified air for 1 to 2 h while exercising.

6 • Pulmonary function decrements are generally observed in healthy subjects (8 to 45 years  
of age) after 1 to 3 h of exposure as a function of the level of exercise performed and the  
O<sub>3</sub> concentration inhaled during the exposure. Group mean data from numerous  
controlled human exposure and field studies indicate that, in general, statistically  
significant pulmonary function decrements beyond the range of normal measurement  
variability (e.g., 3 to 5% for FEV<sub>1</sub>) occur

- 7 – at >0.50 ppm O<sub>3</sub> when at rest,  
8 – at >0.37 ppm O<sub>3</sub> with light exercise (slow walking),  
9 – at >0.30 ppm O<sub>3</sub> with moderate exercise (brisk walking),  
10 – at >0.18 ppm O<sub>3</sub> with heavy exercise (easy jogging), and  
11 – at >0.16 ppm O<sub>3</sub> with very heavy exercise (running).

12 • Smaller group mean changes (e.g., <5%) in FEV<sub>1</sub> have been observed at lower  
O<sub>3</sub> concentrations than those listed above. For example, FEV<sub>1</sub> decrements have been  
shown to occur with very heavy exercise in healthy adults at 0.15 to 0.16 ppm O<sub>3</sub>, and  
such effects may occur in healthy young adults at levels as low as 0.12 ppm. Also,  
pulmonary function decrements have been observed in children and adolescents at  
concentrations of 0.12 and 0.14 ppm O<sub>3</sub> with heavy exercise. Some individuals within a  
study may experience FEV<sub>1</sub> decrements in excess of 15% under these exposure  
conditions, even when the group mean decrement is less than 5%.

13 • For exposures of healthy subjects performing moderate exercise during longer duration  
exposures (6 to 8 h), 5% group mean decrements in FEV<sub>1</sub> were observed at

- 14 – 0.08 ppm O<sub>3</sub> after 5.6 h,  
15 – 0.10 ppm O<sub>3</sub> after 4.6 h, and



1 *experimental studies in laboratory animals and humans. An association between daily mortality*  
2 *and O<sub>3</sub> concentration for areas with high O<sub>3</sub> levels (e.g., Los Angeles) has been suggested,*  
3 *although the magnitude of such an effect is unclear.*

## 4 5 **2. What are the effects of repeated, short-term exposures to ozone?**

6 *During repeated short-term exposures, some of the O<sub>3</sub>-induced responses are partially or*  
7 *completely attenuated. Over a 5-day exposure, pulmonary function changes are typically*  
8 *greatest on the second day, but return to control levels by the fifth day of exposure. Most of the*  
9 *inflammatory markers (e.g., PMN influx) also attenuate by the fifth day of exposure, but markers*  
10 *of cell damage (e.g., lactate dehydrogenase enzyme activity) do not attenuate and continue to*  
11 *increase. Attenuation of lung function decrements is reversed following 7 to 10 days without O<sub>3</sub>.*  
12 *Some inflammatory markers are also reversed during this time period, but others still show*  
13 *attenuation even after 20 days without O<sub>3</sub>. The mechanisms and impacts involved in attenuation*  
14 *are not known, although animal studies show that the underlying cell damage continues*  
15 *throughout the attenuation process. In addition, attenuation may alter the normal distribution*  
16 *of O<sub>3</sub> within the lung, allowing more O<sub>3</sub> to reach sensitive regions, possibly affecting normal*  
17 *lung defenses (e.g., PMN influx in response to inhaled microorganisms).*

## 18 19 **3. What are the effects of long-term exposures to ozone?**

20 *Available data indicate that exposure to O<sub>3</sub> for months and years causes structural changes*  
21 *in several regions of the respiratory tract, but effects may be of the greatest importance in the*  
22 *centriacinar regions (where the alveoli and conducting airways meet); this region typically is*  
23 *affected in most chronic airway diseases of the human lung. This information on O<sub>3</sub> effects in*  
24 *the distal lung is extrapolated from animal toxicological studies because, to date, comparable*  
25 *data are not available from humans. The apparent lack of reversal of effects during periods of*  
26 *clean air exposure raises concern that seasonal exposures may have a cumulative impact over*  
27 *many years. The role of adaptive processes in this response is unknown but may be critically*  
28 *dependent on the temporal frequency or profile of exposure. Furthermore, the interspecies*  
29 *diversity in apparent sensitivity to the chronic effects of O<sub>3</sub> is notable, with the rat representing*  
30 *the lower limit of response, and the monkey the upper limit. Epidemiological studies attempting*

1 to associate chronic health effects in humans with long-term O<sub>3</sub> exposure provide only  
2 suggestive evidence that such a linkage exists.

3 Long-term exposure in the females of one strain of mice to high O<sub>3</sub> levels (1 ppm) caused a  
4 small, but statistically significant increase in lung tumors. There was no concentration-response  
5 relationship, and rats were not affected. Genotoxicity data are either negative or weak. Given  
6 the nature of the database, potential carcinogenicity in animals is uncertain. Ozone did not  
7 show tumor-promoting activity in a chronic rat study (at 0.5 ppm O<sub>3</sub>).

#### 8 9 **4. What are the effects of binary pollutant mixtures containing ozone?**

10 Combined data from laboratory animal and controlled human exposure studies of  
11 O<sub>3</sub> support the hypothesis that coexposure to pollutants, each at low-effect levels, may result in  
12 effects of significance. The data from human studies of O<sub>3</sub> in combination with NO<sub>2</sub>, SO<sub>2</sub>, H<sub>2</sub>SO<sub>4</sub>,  
13 HNO<sub>3</sub>, or CO show no more than an additive response for lung spirometry or respiratory  
14 symptoms. The larger number of laboratory animal studies with O<sub>3</sub> in mixture with NO<sub>2</sub> and  
15 H<sub>2</sub>SO<sub>4</sub> show that effects can be additive, synergistic, or even antagonistic, depending on the  
16 exposure regimen and the endpoint studied. This issue of exposure to copollutants remains  
17 poorly understood, especially with regard to potential chronic effects.

#### 18 19 **5. What population groups are at risk as a result of exposure to ozone?**

20 Identification of population groups that may show increased sensitivity to O<sub>3</sub> is based on  
21 their (1) biological responses to O<sub>3</sub>, (2) preexisting lung disease (e.g., asthma), (3) activity  
22 patterns, (4) personal exposure history, and (5) personal factors (e.g., age, nutritional status).

23 The predominant information on the health effects of O<sub>3</sub> noted above comes from clinical  
24 and field studies on healthy, nonsmoking, exercising subjects, 8 to 45 years of age. These studies  
25 demonstrate that, among this group, there is a large variation in sensitivity and responsiveness  
26 to O<sub>3</sub>, with at least a 10-fold difference between the most and least responsive individuals.  
27 Individual sensitivity to O<sub>3</sub> also may vary throughout the year, related to seasonal variations in  
28 ambient O<sub>3</sub> exposure. The specific factors that contribute to this large intersubject variability,  
29 however, remain undefined. Although differences may be due to the dosimetry of O<sub>3</sub> in the  
30 respiratory tract, available data show little difference on O<sub>3</sub> deposition in the lungs for  
31 inhalation through the nose or mouth.

1            *Daily life studies reporting an exacerbation of asthma and decrease in peak expiratory*  
2 *flow rates, particularly in asthmatic children, appear to support the controlled studies; however,*  
3 *those studies may be confounded by temperature, particle or aeroallergen exposure, and asthma*  
4 *severity of the subjects or their medication use. In addition, field studies of summertime daily*  
5 *hospital admissions for respiratory causes show a consistent relationship between asthma and*  
6 *ambient levels of O<sub>3</sub> in various locations in the Northeastern United States, even after*  
7 *controlling for independent contributing factors. Controlled studies on mild asthmatics suggest*  
8 *that they have similar lung volume responses but greater airway resistance changes to O<sub>3</sub> than*  
9 *nonasthmatics. Furthermore, limited data from studies of moderate asthmatics suggest that this*  
10 *group may have greater lung volume responses than nonasthmatics.*

11            *Other population groups with preexisting limitations in pulmonary function and exercise*  
12 *capacity (e.g., chronic obstructive pulmonary disease, chronic bronchitis, ischemic heart*  
13 *disease) would be of primary concern in evaluating the health effects of O<sub>3</sub>. Unfortunately, not*  
14 *enough is known about the responses of these individuals to make definitive conclusions*  
15 *regarding their relative responsiveness to O<sub>3</sub>. Indeed, functional effects in these individuals with*  
16 *reduced lung function may have greater clinical significance than comparable changes in*  
17 *healthy individuals.*

18            *Currently available data on personal factors or personal exposure history known or*  
19 *suspected of influencing responses to O<sub>3</sub> follow.*

- 20            ● *Human studies have identified a decrease in pulmonary function responsiveness to O<sub>3</sub>*  
*with increasing age, although symptom rates remain similar. Toxicological studies*  
*are not easily interpreted but suggest that young animals are not more responsive*  
*than adults.*
- 21            ● *Available toxicological and human data have not conclusively demonstrated that males*  
*and females respond differently to O<sub>3</sub>. If gender differences exist for lung function*  
*responsiveness to O<sub>3</sub>, they are not based on differences in baseline pulmonary function.*
- 22            ● *Data are not adequate to determine whether any ethnic or racial group has a different*  
*distribution of responsiveness to O<sub>3</sub>. In particular, the responses of nonwhite asthmatics*  
*have not been investigated.*
- 23            ● *Information derived from O<sub>3</sub> exposure of smokers is limited. The general trend is that*  
*smokers are less responsive than nonsmokers. This reduced responsiveness may wane*  
*after smoking cessation.*

- 1
- *Although nutritional status (e.g., vitamin E deficiency) makes laboratory rats more susceptible to O<sub>3</sub>-induced effects, it is not clear if vitamin E supplementation has an effect in human populations. Such supplementation has no or minimal effects in animals. The role of such antioxidant vitamins in O<sub>3</sub> responsiveness, especially their deficiency, has not been well studied.*

2           *Based on information presented in this document, the population groups that have*  
3 *demonstrated increased responsiveness to ambient concentrations of O<sub>3</sub> consist of exercising,*  
4 *healthy and asthmatic individuals, including children, adolescents, and adults.*

5  
6           Since the 1996 O<sub>3</sub> AQCD evaluations, a distinctly more extensive database of air pollution  
7 epidemiologic studies has become available. A subset of these studies which examined O<sub>3</sub>  
8 health effects have reported a variety of O<sub>3</sub>-related health effects associations. Based on the  
9 physiological, biochemical and molecular changes observed in controlled human exposure  
10 studies and animal toxicological studies, new evidence is now available by which to evaluate the  
11 biological plausibility and extent of coherence for various health outcomes (such as respiratory  
12 and cardiovascular effects, fetal and infant development effects, and mortality) reported in the  
13 epidemiologic studies as discussed in ensuing sections. Biological observations pointing  
14 towards putative mechanisms of action in developing hypotheses to interpret associated  
15 pathological symptoms reported in epidemiologic studies are also critically evaluated in  
16 subsequent sections, as are in vitro and in vivo experimental studies using novel molecular  
17 technologies to address potential mechanisms of action.

## 18 19 **8.4.2 Assessment and Integration of New Experimental Evidence**

### 20 **8.4.2.1 Background on Cross-Cutting Issues**

21           Discussion of several cross-cutting issues that will facilitate a clearer understanding of the  
22 ensuing assessment is provided here to enhance an integrated and comprehensive understanding  
23 of the experimental and epidemiologic studies on O<sub>3</sub> health effects. An important issue to  
24 be considered is the extrapolation of observed effects in animals to humans, from the perspective  
25 of dosimetry and the strength and weaknesses of such extrapolation models. The most  
26 challenging issue is the integration of (a) epidemiologic (observational) findings that suggest a  
27 potential causative role of ambient O<sub>3</sub> (with adjustments for other copollutants) in producing

1 health effects with (b) physiological, biochemical and toxicological findings from experimental  
2 studies.

#### 4 **8.4.2.2 Approaches to Experimental Evaluation of Ozone Health Effects**

5 Three chapters in the current document provide detailed discussion of various experimental  
6 approaches utilized to evaluate O<sub>3</sub>-related health effects. Chapter 4 discusses dosimetry issues  
7 pertinent to both animal and human exposure scenarios. Chapter 5 discusses the experimental  
8 studies of physiological, biochemical (cellular and molecular changes) and pathological  
9 observations in laboratory animals (including nonhuman primates, dogs, and rodent species) and  
10 in vitro studies using cell culture systems (in certain cases, on humans cells recovered from  
11 BALF postexposure to O<sub>3</sub>). Chapter 6 evaluates studies on human volunteers exposed to O<sub>3</sub>  
12 which have investigated a variety of physiological and biochemical endpoints.

13 In interpreting the results from the experimental approaches, one must consider the  
14 following three issues: (1) exposure/dose considerations; (2) role of confounders; and  
15 (3) interpretation of results from high dose exposures and animal to human extrapolations.  
16 Earlier animal toxicology studies were carried out using relatively high O<sub>3</sub> exposure  
17 concentrations/doses that do not necessarily reflect “real-world” exposure scenarios. Those  
18 experiments were primarily aimed at understanding the pathophysiology associated with O<sub>3</sub>  
19 exposure in healthy animals, to help understand potential mechanisms(s) of action, and to help  
20 validate health outcomes reported in epidemiologic studies. Since the 1996 O<sub>3</sub> AQCD, the  
21 majority of human and animal studies have used ambient and/or near ambient doses. Earlier  
22 controlled chamber exposure studies on human volunteers mainly limited exposures to O<sub>3</sub> alone  
23 in comparison to sham (clean air) exposures, thus providing evidence concerning direct effects  
24 of O<sub>3</sub> per se versus more closely mimicking real-world atmospheric exposures to multipollutant  
25 mixes. Some newer air pollution clinical studies are utilizing various co-exposure regimens to  
26 simulate more closely ambient exposure to air pollution mixtures; and the results from these  
27 studies will be highly useful in developing better models to interpret the toxicological effects  
28 associated with O<sub>3</sub>-containing ambient air pollutants mixes.

29 Interpretations of experimental studies of air pollution, as in the case of environmental  
30 comparative toxicology studies, are affected by limitations associated with animal extrapolation  
31 models. The differences between humans and rodents with regard to O<sub>3</sub> inhalability, absorption



1 and distribution profiles based on breathing pattern, exposure dose and differences in lung  
2 structure and anatomy (see Chapter 4 and 5 for details) have to be taken into consideration.  
3 Also, in spite of a high degree of homology and the existence of a high percentage of  
4 orthologous genes across human and rodents, particularly mice, extrapolation of molecular  
5 alterations at the gene level suffers from the regulatory control of various signaling units as  
6 simple as cis and trans activating transcription factor units. Given these molecular differences,  
7 extrapolation of physiological parameters (which are under the control of various biochemical,  
8 endocrine and neuronal controls) observed between human and rodents represents a difficult  
9 task.

### 11 **8.4.2.3 Interspecies Comparison of Experimental Results: Dosimetric Considerations**

12 In this section, a brief overview of the experimental results obtained from studies on  
13 human and laboratory animals are comparatively analyzed and presented to provide background  
14 for assessing biological plausibility and coherence discussed in detail in the following section.  
15 Each subsection starts off with an introduction to what was known at the time of the publication  
16 of the previous O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996), followed by  
17 discussion of new information.

#### 19 **Dosimetry Considerations**

20 Dosimetric studies demonstrate fundamental relationships between ambient exposures and  
21 doses to target tissues. While experimental and theoretical dosimetry (modeling) studies of O<sub>3</sub>  
22 have proved to be valuable in the assessment of toxicity, they are most useful when conducted as  
23 part of an integrated approach to determining the distribution of inhaled O<sub>3</sub> along the upper and  
24 lower respiratory tract. Derivation of credible dosimetry estimates greatly facilitates the  
25 development of useful extrapolation models by which to compare doses and effects across  
26 species and subpopulations.

27 The state-of-the-art of O<sub>3</sub> dosimetry, as described in 1996 O<sub>3</sub> AQCD, indicated consistency  
28 across data and models derived from in vivo human and animal studies, thus increasing the level  
29 of confidence in the development of dosimetric extrapolation models. Earlier dosimetry models  
30 predicted that the tissue dose of inhaled O<sub>3</sub> was greatest at the bronchoalveolar junction, the  
31 region experimentally shown to be most impacted by O<sub>3</sub>. Ozone bolus inhalation studies in

1 humans have indicated that inspired O<sub>3</sub> reaches the distal airways and alveoli of resting humans;  
2 and, with increased inspiratory flow rates due to exercise, O<sub>3</sub> penetrates deeper and in greater  
3 quantity to the distal regions of the lung. These findings have been corroborated by observations  
4 of <sup>18</sup>O<sub>3</sub> (oxygen-18-labeled ozone) in the BALF of humans and rats (Hatch et al.,1994).

5 Some acute responses to O<sub>3</sub> have compared well across species when controlled for dose,  
6 indicating that animals and humans (a) respond to O<sub>3</sub> in a dose-dependent manner, i.e., they  
7 exhibit increasing breathing frequency with an accompanying decrease in tidal volume  
8 (tachypnea), and (b) show similar changes in alveolar permeability as measured by protein in the  
9 bronchoalveolar lavage fluid (BALF). These parallel changes in humans and animals were  
10 sufficiently homologous to suggest a common mode of action. It has also been recognized  
11 that O<sub>3</sub>-induced spirometric changes, the hallmark of response in humans, also occur in exposed  
12 rats when hyperventilated with CO<sub>2</sub> stimulation. However, the effect of anesthesia in the rodent  
13 model in contrast to the awake human remains uncertain; but, as will be discussed, activity level  
14 differences between species appear to strongly influence dose. Nevertheless, most lung function  
15 decrements subside with repeated exposures in both humans and animals, with analogous  
16 attenuation of certain (but not all) parameters measured in the BALF. The mechanisms  
17 associated with attenuation are unclear but may involve endogenous antioxidants. The  
18 significance of non-attenuated markers in BALF has been interpreted to relate to potential  
19 chronicity of O<sub>3</sub> effects. Studies on long-term exposure in monkeys and rats do show long term  
20 changes in the distal lung that appear to be represented by a near-linear dose-response pattern.  
21 More thorough analysis of this dose response is needed, however.

22 As discussed in the 1996 O<sub>3</sub> AQCD, Hatch et al. (1994) compared responses of exercising  
23 humans (15-min intervals of rest and exercise at 60 L/min for 2 h) to those of resting rats also  
24 exposed to 0.4 ppm <sup>18</sup>O<sub>3</sub> (oxygen-18-labeled ozone) for 2 h. They observed 4 or 5 times the <sup>18</sup>O<sub>3</sub>  
25 dose (as adduct) in BALF constituents of humans as compared to those of F344 male rats. This  
26 4- to 5-fold difference appeared to be due to the exercise-stimulated hyperventilation of the  
27 humans when compared to the rats. *Only when the resting rats were exposed to 2 ppm O<sub>3</sub> for 2 h*  
28 *at rest did the <sup>18</sup>O<sub>3</sub> labeling of BALF constituents and indices of effect (i.e., BAL cells and*  
29 *protein at 24 h) compare favorably with those of the exercising humans exposed to 0.4 ppm <sup>18</sup>O<sub>3</sub>*  
30 *for 2 h with intermittent exercise.* Thus, the rat and human appear to have similar sensitivities  
31 to O<sub>3</sub> when exercise is taken into account as a dose modifier. It was further concluded in the

1 1996 O<sub>3</sub> AQCD that attempts to compare resting animal data to exercising human data obtained  
2 at similar O<sub>3</sub> concentrations would likely underestimate the dose to the lung and, presumably, the  
3 resultant risk of effect.

4 In the past decade, no further reports have been published on O<sub>3</sub> uptake studies in animals,  
5 although several controlled human bolus and/or general O<sub>3</sub> uptake studies have provided refined  
6 data. The bolus uptake studies suggest that prior exposure to O<sub>3</sub> diminishes bolus uptake. In the  
7 earlier document, the effect of mode of breathing (oral or nasal) on O<sub>3</sub> uptake was thought to be  
8 minimal, with approximately equal uptake via the nose or mouth. Newer bolus dose studies  
9 have demonstrated that the uptake and regional respiratory tract distribution of O<sub>3</sub> is sensitive to  
10 mode of breathing (nasal uptake greater than oral) and to air flow rate (uptake decreases with  
11 increasing flow). Similarly, the change in breathing with exercise vs rest causes a shift in  
12 regional O<sub>3</sub> distribution, allowing deeper respiratory tract penetration, with resultant greater dose  
13 and damage to respiratory bronchiolar and alveolar tissues (as predicted by the models described  
14 in the 1996 O<sub>3</sub> AQCD).

15 The efficiency of O<sub>3</sub> uptake is chemically rate dependent. The resultant reaction products  
16 (hydrogen peroxide, aldehydes, and hydroxyhydroperoxides) created by ozonolysis of lipids in  
17 airway and epithelial lining fluid are thought to mediate O<sub>3</sub> toxicity. The dependence of O<sub>3</sub>  
18 absorption on chemical-reaction rates is consistent with the observation of Bush et al. (2001),  
19 that the rate of O<sub>3</sub> uptake is lower than for Cl<sub>2</sub> despite the similar gas-phase diffusion coefficients  
20 of these two gases. The slower uptake rate of O<sub>3</sub> relative to Cl<sub>2</sub> appears due to the limiting  
21 reaction rate of O<sub>3</sub> in the epithelial lining fluid. The work by Rigas et al. (1997) using the O<sub>3</sub>  
22 bolus technique in humans, showing uptake to be increased by continuous exposure to NO<sub>2</sub>  
23 and SO<sub>2</sub> and decreased by continuous O<sub>3</sub> exposure, suggests an important role for copollutant  
24 exposures. Thus, an inflammatory response may magnify the production of O<sub>3</sub>-reactive  
25 substrates in the epithelial lining fluid when other oxidants are present.

26 New uptake studies (Ultman et al., 2004) carried out in controlled human clinical studies  
27 have observed gender-specific differences in the uptake of O<sub>3</sub>, but these differences do not  
28 correlate well with spirometric responses. Rather, they appear to be related to breathing pattern  
29 and lung size, with females having smaller lungs than males. Other uptake studies carried out in  
30 humans using environmentally relevant O<sub>3</sub> concentrations have demonstrated the significance of  
31 incorporating inter-subject variability in dose-response relationship prediction and extrapolation.

1 Thus, a number of variables seem to have a degree of impact on O<sub>3</sub> uptake, notably including  
2 age, route of breathing, breathing pattern, gender and certain pre-exposure conditions. These  
3 differences are important in order to interrelate biological effect and risk assessment estimates.

4 The general consistency observed in O<sub>3</sub> uptake in animal and human experimental  
5 exposure studies provides increased confidence in the use of theoretical dosimetry modeling and  
6 the use of animal toxicological data (see Chapter 4 for detailed discussion). Models have taken  
7 into consideration various factors such as age, as well as anatomical, physiological, and  
8 biochemical alterations. Incorporation of novel information, such as (a) the identification of  
9 primary site of acute cell injury, (b) the site of O<sub>3</sub> reaction/diffusion in the epithelial lining fluid,  
10 (c) the roles of intermediate reactive oxygen species (ROS) and lipid ozonation products in  
11 oxidative injury, and (d) the roles of metabolic enzyme profiles in developing lung tissue, can be  
12 expected to lead to refined novel models and better extrapolation.

#### 14 **8.4.2.4 Critical Analysis of Toxicological Effects of O<sub>3</sub> Exposure**

15 In the following subsections, research results generated from experimental studies on  
16 humans and animals during the past decade are assessed (keeping in view the interspecies  
17 differences discussed in the preceding section) in evaluating experimental evidence for  
18 biological plausibility and coherence for O<sub>3</sub> health effects discussed in the later sections.

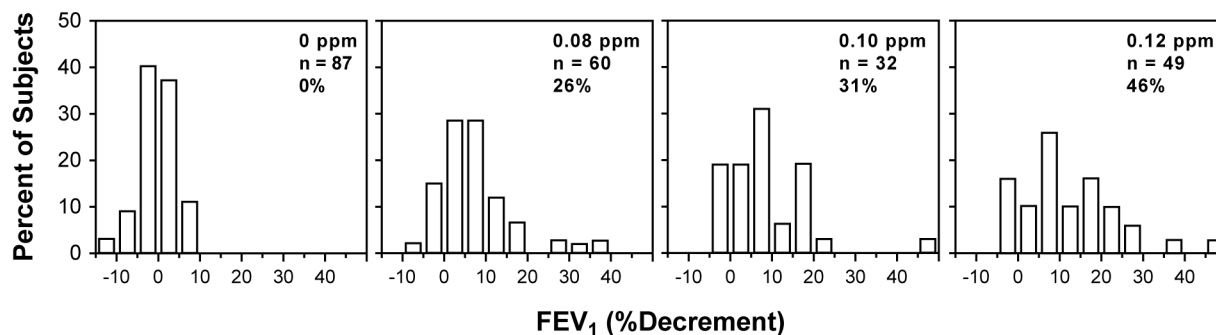
##### 20 **8.4.2.4.1 Pulmonary Function**

21 A number of controlled human exposure, animal, and epidemiological studies assessed in  
22 the 1996 O<sub>3</sub> AQCD demonstrated alterations in various measurements of pulmonary function.  
23 Inhalation of O<sub>3</sub> for several hours while physically active elicits both acute pathophysiologic  
24 changes and subjective respiratory tract symptoms. The pulmonary responses observed in  
25 healthy human subjects exposed to ambient O<sub>3</sub> concentrations include decreased inspiratory  
26 capacity; mild bronchoconstriction; rapid, shallow breathing pattern during exercise; and  
27 subjective symptoms of tracheobronchial airway irritation, including cough and pain during  
28 inspiration. Acute O<sub>3</sub> exposures also cause decreases in forced vital capacity (FVC), forced  
29 expiratory volume in 1 s (FEV<sub>1</sub>), and increased airways resistance (SR<sub>aw</sub>). The severity of  
30 symptoms and the magnitude of response depends on inhaled dose, individual O<sub>3</sub> sensitivity, and  
31 the extent of tolerance resulting from previous exposures.

1 A progressive decrease in tidal volume and a “compensatory” increase in frequency of  
2 breathing to maintain steady minute ventilation during exposure suggests a direct modulation of  
3 ventilatory control. These changes in humans parallel responses of many animal species  
4 exposed to O<sub>3</sub> and other lower airway irritants (Tepper et al., 1990). Pulmonary function  
5 evaluations carried out in several animal species on acute exposure to O<sub>3</sub> generally show  
6 responses similar to those observed in humans, such as increased breathing frequency, decreased  
7 tidal volume, increased resistance and decreased FVC. These effects are observed at relatively  
8 low O<sub>3</sub> concentrations (0.25 to 0.4 ppm) following several hours of exposure in many species.  
9 The alterations in breathing pattern return to normal within hours after exposure and the pattern  
10 of attenuation in responses following repeated exposures is similar to that observed in humans.  
11 When rats were exposed to concentrations  $\geq 1$  ppm even breathing mechanics were found to  
12 be affected.

13 The time course of spirometry responses to O<sub>3</sub> exposure depends on the specific exposure  
14 conditions. Early controlled human studies reviewed in the 1986 and 1996 O<sub>3</sub> AQCD typically  
15 reported statistically significant pulmonary responses in exercising (intermittent or continuous)  
16 subjects exposed for 2 h to a concentration in the range of 0.12 to 0.4 ppm O<sub>3</sub> (mimicking  
17 midday ambient O<sub>3</sub> peaks reported in Los Angeles, CA). Significant effects were not observed  
18 following 2 h exposures in sedentary subjects below 0.5 ppm O<sub>3</sub>. Some later human studies  
19 reviewed in the 1996 O<sub>3</sub> AQCD utilized 6-8 h exposures with exercise in order to better mimic  
20 longer exposures to ambient O<sub>3</sub> (recognizing the more prolonged elevated ambient O<sub>3</sub> levels  
21 often observed in some urban areas in the northeastern states) and provided some of the strongest  
22 and most quantifiable concentration-response data on the acute health effects of O<sub>3</sub> based on  
23 pulmonary function tests.

24 All evaluations have indicated that there exists considerable interindividual differences in  
25 the magnitude of responses to O<sub>3</sub>. However, an individual’s lung function and to a lesser extent,  
26 respiratory symptom responses to O<sub>3</sub> are reproducible over a period of time, indicating that some  
27 individuals are consistently more responsive than others to O<sub>3</sub>. Figure 8-1 illustrates the  
28 variability in FEV<sub>1</sub> responses in young healthy adults following a prolonged (6.6 h) exposure  
29 to O<sub>3</sub>, as summarized in the 1996 O<sub>3</sub> AQCD. Referring to this figure, the average FEV<sub>1</sub> response  
30 following exposure to 0.08 ppm O<sub>3</sub> is small (between a 5 and 10% decrement). However, ~18%  
31 of the exposed subjects had moderate FEV<sub>1</sub> decrements of 10 to 20% and ~8% experienced



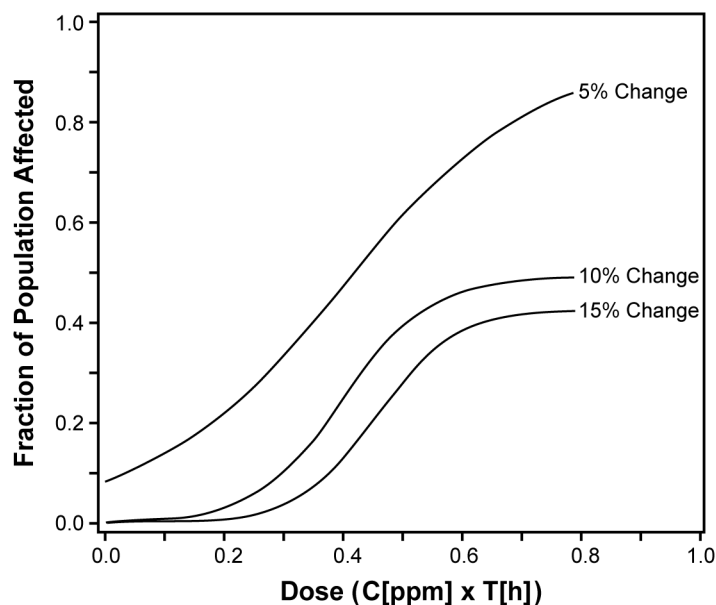
**Figure 8-1. Frequency distributions of FEV<sub>1</sub> decrements following 6.6-h exposures to O<sub>3</sub> or filtered air. During each hour of the exposures, subjects were engaged in moderate exercise for 50 minutes. With increasing O<sub>3</sub> concentration, the distribution of responses becomes asymmetric, with a few individuals exhibiting large FEV<sub>1</sub> decrements. The percentage in each panel indicates the portion of subjects having a FEV<sub>1</sub> decrement in excess of 10%.**

Source: McDonnell (1996).

1 large FEV<sub>1</sub> decrements of greater than 20%. This serves to emphasize that while average  
 2 responses may be small and seem physiologically insignificant, some individuals typically  
 3 experience distinctly more severe effects. As a further example of intersubject variability,  
 4 Figure 8-2 illustrates the portion of young healthy adult males (24 yr old) predicted to have FEV<sub>1</sub>  
 5 decrements of greater than 5, 10, and 15% when exposed to O<sub>3</sub> during moderate exercise, as also  
 6 presented in the 1996 O<sub>3</sub> AQCD.

7 New studies (assessed in Chapter 6 and Annex 6 of this document) which evaluated  
 8 responses in hundreds of subjects clearly indicate that FEV<sub>1</sub> decrements and symptom responses  
 9 decrease with age beyond young adulthood (18 to 20 years). Hazucha et al. (2003), for example,  
 10 examined gender and age differences in O<sub>3</sub> responsiveness and found that young females lose O<sub>3</sub>  
 11 sensitivity faster than young males, but the rate is about the same for both genders by middle age  
 12 (see Figure 8-3).

13 The development of effects is time-dependent during both exposure and recovery periods,  
 14 with considerable overlap of evolving and receding effects. In healthy human subjects exposed  
 15 to typical ambient concentrations (i.e., <0.2 ppm O<sub>3</sub>), spirometric responses largely resolve  
 16 within a few hours (4 to 6 h) postexposure; but cellular effects persist for longer periods (~24 h).  
 17 Persisting small residual lung function effects are almost completely resolved within 24 hours.

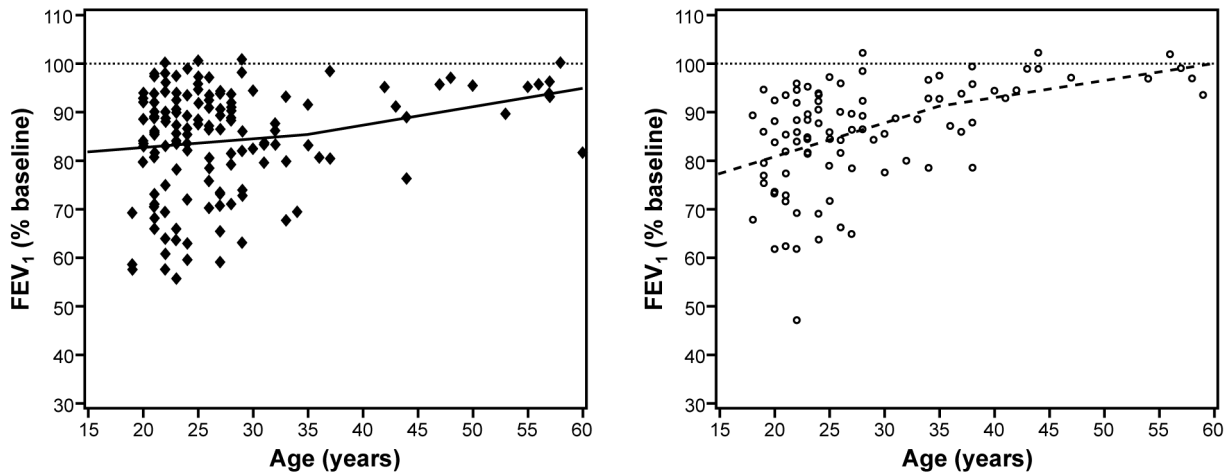


**Figure 8-2. Proportion of moderately exercising healthy adults (24 yrs old) predicted to have 5, 10, or 15% decrements in FEV<sub>1</sub> as a function of concentration (0 to 0.12 ppm O<sub>3</sub>) times exposure duration (1 to 6.6 h).**

Source: McDonnell et al. (1995).

1 In hyperresponsive individuals, the recovery takes longer (as much as 48 h) to return to baseline  
 2 values. The majority of these responses are attenuated after repeated exposure, but such  
 3 tolerance to O<sub>3</sub> is lost within a week postexposure. The biochemical indicators of lung injury  
 4 and associated morphological changes were not found to be attenuated in the majority of  
 5 laboratory animals. Unfortunately, no data are available on pulmonary function changes in  
 6 animals upon chronic exposure to O<sub>3</sub>. However, earlier work of repeated exposure of rats to an  
 7 episodic profile of O<sub>3</sub> demonstrated small but significant decrements in lung function that were  
 8 consistent with early indicators of focal fibrinogenesis in the proximal alveolar region.

9 In the 1996 O<sub>3</sub> AQCD, O<sub>3</sub>-induced decrease in inspiratory capacity was hypothesized to be  
 10 the result of neurogenic inhibition of maximal inspiration due to stimulation of C-fiber afferents  
 11 either directly or from O<sub>3</sub>-induced inflammatory mediators. Earlier human studies (Coleridge  
 12 et al., 1993; Hazucha and Sant’Ambrogio, 1993) reported a role for bronchial C-fibers and  
 13 rapidly adapting receptors as primary vagal afferents responsible for O<sub>3</sub>- induced changes in  
 14 ventilatory rate and depth. As discussed in Chapter 6, the newer results of Passannante et al.



**Figure 8-3. Effect of age on FEV<sub>1</sub> responses to O<sub>3</sub> exposure (0.42 ppm for 1.5 h with intermittent exercise). Left and right panels are data for males (n = 146; 19 to 60 yrs old) and females (n = 94; 18 to 59 yrs old), respectively. On average, FEV<sub>1</sub> responses to O<sub>3</sub> exposure decrease with increasing age. However, there is a large amount of intersubject variability in responses, e.g., responses of 20 to 25 year olds range from a small increase to greater than a 50% decrement in FEV<sub>1</sub> following O<sub>3</sub> exposure.**

Source: Adapted from Hazucha et al. (2003).

1 (1998) also support C-fiber stimulation as a primary mechanism of the O<sub>3</sub>-induced reduction in  
 2 inspiratory capacity and suggest a role for nociceptive mechanisms. This neurogenic mechanism  
 3 also likely has an effect on airway responsiveness and lung inflammation.

4 Lung function changes due to O<sub>3</sub> exposure have been evaluated in patients with preexisting  
 5 respiratory diseases under experimental controlled exposure regimens, with or without physical  
 6 exertion in the form of intermittent exercise. These new studies found minimal O<sub>3</sub>-induced  
 7 effects in COPD patients. For example, Gong et al. (1997a) exposed nine COPD patients  
 8 (0.24 ppm O<sub>3</sub> for 4 h with intermittent exercise) and observed a nonsignificant FEV<sub>1</sub> decrement  
 9 of -8% in COPD patients that was not statistically different from the -3% decrement seen in  
 10 healthy subjects. Augmenting observations discussed in the 1996 O<sub>3</sub> AQCD, newer studies of  
 11 asthmatics (see Chapter 6) continue to indicate that pulmonary function deficiencies detected by  
 12 spirometric analyses are somewhat increased relative to healthy controls. A tendency for  
 13 increased O<sub>3</sub>-induced pulmonary function responses were reported in asthmatics relative to



1 healthy subjects exposed to O<sub>3</sub> concentrations of ≤0.2 ppm for 4- 8 h duration (Scannell et al.,  
2 1996). Similarly, Alexis et al. (2000) observed statistically significant O<sub>3</sub>-induced decreases  
3 in FEV<sub>1</sub> in mild atopic asthmatics that tended to be greater than experienced by healthy control  
4 subjects. In a longer exposure duration (7.6 h) study, Horstman et al. (1995) reported that mild-  
5 to-moderate asthmatics exposed to 0.16 ppm O<sub>3</sub> had FEV<sub>1</sub> decrements that were significantly  
6 greater than in healthy subjects (19% versus 10% respectively). Moreover, Horstman et al.  
7 (1995) found that responses of asthmatics were more severe in patients with lower baseline lung  
8 function. Though most controlled human exposure studies may not provide the required  
9 statistical power (due to the limited number of subjects compared to panel or field studies),  
10 they do suggest that asthmatics are at least as sensitive, if not more, than healthy subjects.

11 In addition to effects of O<sub>3</sub> exposure on the large airways as indicated by spirometric  
12 responses, O<sub>3</sub> exposure also affects the function of the small airways and parenchymal lung.  
13 Studies reported by Foster et al. (1993, 1997) that examined the effect of O<sub>3</sub> on ventilation  
14 distribution in healthy adult males suggest a prolonged O<sub>3</sub> effect on the small airways and  
15 ventilation distribution in some individuals. Animal toxicology studies have shown the  
16 centriacinar region (CAR) of the lung (the segment between the last conducting airway and the  
17 gas exchange region) to be a region highly susceptible to O<sub>3</sub>-induced damage (epithelial cell  
18 necrosis and remodeling of respiratory bronchioles) and seem to be reasonably predictive of  
19 similar morphological changes as being likely to occur in humans. Unfortunately, common  
20 pulmonary function tests do not measure acute changes in the small airways of the CAR.  
21 Identification of acute effects of O<sub>3</sub> in small airways, if any, would lend additional support for  
22 concerns about long-term effects of repeated O<sub>3</sub> exposures.

#### 23 24 **8.4.2.4.2 Airway Responsiveness**

25 Increased airway responsiveness, also referred to as airway hyperresponsiveness (AHR) or  
26 bronchial hyperreactivity, is an indicator of enhanced reactivity of airways to  
27 bronchoconstriction induced by a variety of stimuli (exposure to cold air, allergens or exercise).  
28 AHR is assessed by airway function (either spirometry or plethysmography) after inhalation  
29 exposure to specific (antigen, allergen) or nonspecific (methacholine, histamine)  
30 bronchoconstrictor stimuli. It was recognized in the 1996 O<sub>3</sub> AQCD that exposure to O<sub>3</sub> to  
31 induce AHR in humans usually resolves in 18-24 h after exposure in a majority of subjects, but

1 may persist in some individuals for longer periods. Gong et al. (1997b) found that subjects with  
2 asthma developed tolerance to repeated O<sub>3</sub> exposures in a manner similar to normal subjects;  
3 however, there were more persistent effects of O<sub>3</sub> on airway responsiveness, which only partially  
4 attenuated when compared to filtered air controls. Such an occurrence and duration of increased  
5 nonspecific airway responsiveness following O<sub>3</sub> exposure could have clinical implications in  
6 asthmatics, possibly putting them at potential increased risk for more prolonged bouts of  
7 bronchoconstriction in response to various triggering stimuli (e.g., allergens, cold air, etc.).

8 Studies examining the effects of O<sub>3</sub> on exacerbations of antigen-induced asthma suggested  
9 that allergen-specific increased airway responsiveness indeed occurs in mild asthmatics upon  
10 exposure to O<sub>3</sub>. Jörres et al. (1996) confirmed that higher O<sub>3</sub> concentrations cause increased  
11 airway reactivity to specific antigens in subjects with mild allergic asthma and, to a lesser extent,  
12 in subjects with allergic rhinitis, after exposure to 0.25 ppm O<sub>3</sub> for 3 h. This enhancement of  
13 allergen responsiveness after O<sub>3</sub> exposure appears to be time dependent, suggesting that the  
14 timing of allergen challenge in O<sub>3</sub>-exposed subjects with allergic asthma is important.  
15 Significant, clinically relevant decreases in pulmonary function have been observed in the  
16 early phase allergen response in subjects with rhinitis after consecutive (4-day) exposure to  
17 0.125 ppm O<sub>3</sub> (Holz et al. 2002). Similar increased airway responsiveness to house dust mite  
18 antigen 16-18 h postexposure to a single dose of O<sub>3</sub> (0.16 ppm for 7.6 h ) was also observed in  
19 asthmatics. These observations suggest that O<sub>3</sub> exposure may be a clinically important factor  
20 that can exacerbate the response to ambient bronchoconstrictor substances in individuals with  
21 preexisting allergic asthma and that its influence may be both immediate and persist for  
22 relatively long periods of time.

23 An extensive laboratory animal study database (using rats, mice, guinea pigs, and rabbits),  
24 exploring the effects of acute, long-term, and repeated exposures to O<sub>3</sub>, indicates that induction  
25 of AHR occurs at relatively high O<sub>3</sub> concentrations. These studies provide clues to the roles of  
26 physiological and biochemical components involved in this process, but one has to exercise  
27 caution in the interpretation of these results, as different mechanisms may be involved in  
28 mediating high-dose and low-dose responses. Some of these studies indicated differences  
29 in O<sub>3</sub>-induced AHR between immature and adult rats and also between obese and lean mice  
30 strains. Some of the ex-vivo studies carried out in New Zealand white rabbits using  
31 environmentally relevant O<sub>3</sub> concentrations indicated O<sub>3</sub>-induced alterations in tracheal epithelial

1 functions and potential O<sub>3</sub>-induced direct vascular constriction. As observed in humans, the  
2 acute changes in AHR do not persist upon long-term exposure in animals exposed to near-  
3 ambient concentrations of O<sub>3</sub>; and attenuation has been observed. Both human and animal  
4 studies indicate that airway responses are not associated with inflammation, but they do suggest  
5 a likely role for neuronal involvement.

#### 7 **8.4.2.4.3 Morphological and Biochemical Abnormalities**

8 Most of the research results alluded to the ensuing discussion come from toxicology  
9 studies using various laboratory animal species that were usually exposed to relatively high,  
10 non-ambient concentrations of O<sub>3</sub>. However, these exploratory and mechanistic studies may  
11 provide important and useful hypotheses to consider in integrating various health outcomes  
12 observed or predicted by epidemiologic studies. A limited number of controlled human  
13 exposure studies evaluated cellular and biochemical parameters in the BALF. These studies  
14 have yielded limited evidence supporting the observations made in animal toxicology studies.  
15 Keeping in view the species- specific differences, the morphological and biochemical alterations  
16 in humans and animals are integrated in the following paragraphs to develop working hypotheses  
17 to interpret human health outcomes.

#### 18 *Lung Injury and Morphological Changes*

19 The 1996 O<sub>3</sub> AQCD stated that short-term O<sub>3</sub> exposure causes similar types of alterations  
20 in lung morphology in all laboratory animal species studied, including primates. The cells in the  
21 CAR have been recognized as a primary target, possibly because it receives the greatest dose of  
22 O<sub>3</sub> delivered to the lower respiratory tract. The ciliated cells in the nasal cavity and airways and  
23 Type I epithelial cells in the gas-exchange region are also identified as targets. Differences in  
24 the distribution of antioxidants in the CAR of the lung were responsible for the differences in  
25 injury and morphological changes observed between nonhuman primates and rodents. Though  
26 acute O<sub>3</sub> exposure induces structural changes such as fibrosis in the CAR, these structural  
27 alterations appear to be partially transient, with recovery shortly postexposure; but the time for  
28 recovery is dependent on species and the dose of O<sub>3</sub>.

29 New studies reviewed in the 1996 O<sub>3</sub> AQCD of lung morphological changes or damage  
30 due to long-term or prolonged exposure to O<sub>3</sub> found chronic lesions similar to early lesions of  
31

1 respiratory bronchiolitis, which have the potential to progress to fibrotic lung disease. Some of  
2 the morphological changes associated with long-term exposures, such as increases in  
3 hyperplastic epithelial cells, appear to reverse following cessation of O<sub>3</sub> exposure. However,  
4 in the underlying interstitium of the CAR, proliferation of fibroblasts creates excess noncellular  
5 matrices. These processes are only partially reversible and may progress following cessation of  
6 exposure. This suggests initiation of focal interstitial fibrosis, which can progress to chronic  
7 degenerative lung disease. Another important observation reported in the 1996 O<sub>3</sub> AQCD was  
8 that of greater injury observed in the monkey's lung upon intermittent exposure (simulated  
9 ambient) compared to continuous exposure, suggesting a role for loss of tolerance in this  
10 process.

11 Reports of morphological changes following chronic O<sub>3</sub> exposures in animal studies  
12 (rodents and primates) published since the 1996 AQCD allude to the earlier findings assessed in  
13 that document. In rats, the effects of chronic ~0.5 ppm O<sub>3</sub> exposure included mucous cell  
14 metaplasia, hyperplasia of the nasal epithelium, increased mucosubstances, and increased Bcl-2  
15 protein levels. In mice, lifetime exposures of 0.5 ppm O<sub>3</sub> were linked to similar outcomes.  
16 Taken together, the rodent studies suggest that O<sub>3</sub> exposure may have the potential to induce  
17 similar long-lasting alterations in human airways. A series of new studies that utilized infant  
18 rhesus monkeys and simulated seasonal ambient exposure (0.5 ppm 8 h/day for 5 days, every  
19 14 days for 11 episodes) reported remodeling in the distal airways; abnormalities in tracheal  
20 basement membrane; eosinophil accumulation in conducting airways; and decrements in airway  
21 innervation, again confirming the potential greater injury due to seasonal exposure compared to  
22 continuous exposure alluded to in the 1996 O<sub>3</sub> AQCD.

23 One epidemiologic report by Sherwin et al. (2000) compared results for autopsy of the  
24 lungs of Los Angeles and Miami residents and observed a significantly greater extent and  
25 severity of centriacinar region alterations in the lungs of Los Angeles residents independent of a  
26 smoking effect. These results suggest that the severity of CAR alterations may be related to the  
27 higher O<sub>3</sub> levels in Los Angeles. Similar observations of CAR thickening and deposition of  
28 collagen seen with chronic O<sub>3</sub> exposure in rat also suggest progressive structural lung injury that  
29 can evolve into a more chronic form, such as fibrosis. Again, however, one must be cautious in  
30 extrapolating these laboratory animal observations to humans, given the exposure regimens and  
31 doses used.

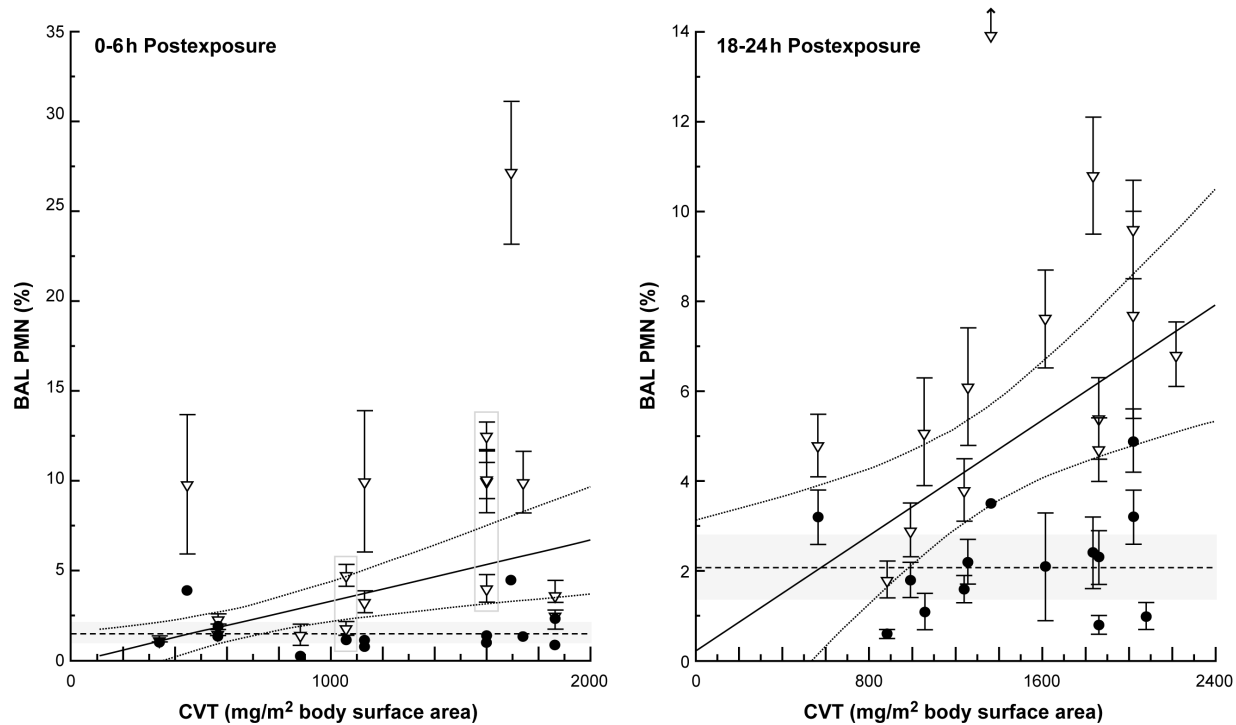
## ***Lung Inflammation and Permeability***

The 1996 O<sub>3</sub> AQCD recognized respiratory tract inflammation and increased cellular permeability as two important biological markers of ozone exposure in both animals and humans. These distinct, independent biological events have been observed in all species studied in response to acute exposure to O<sub>3</sub>. Increased epithelial permeability and inflammation in the lower respiratory tract are measured by increases in bronchoalveolar lavage fluid (BALF) protein and/or albumin and neutrophils (PMNs), respectively. Nasal lavage (NL) fluid and cells from O<sub>3</sub>-exposed humans were used to assess the inflammatory and permeability changes in the upper respiratory tract. Structural changes in nasal mucosa have been demonstrated after O<sub>3</sub> exposure in animals and humans. The presence of PMNs in the lung has long been accepted as a hallmark of inflammation and as an important indicator that O<sub>3</sub> causes inflammation in the lungs. Importantly, respiratory tract inflammation may lead to significant health effects, including impaired host defenses and irreversible structural alterations (as discussed earlier). Ozone-induced mucous membrane cell metaplasia observed in rodents appears to be mediated by inflammation.

Laboratory animals exhibit varying degrees of sensitivity to O<sub>3</sub> exposure (see Chapter 5 for detailed discussion); and this is evident even for the induction of pulmonary inflammation and permeability. Newer animal toxicology studies on O<sub>3</sub>-induced inflammation reviewed in Chapter 5 indicate that the lowest ozone concentration that had an effect on mouse lung inflammation was also 0.11 ppm for 24 hours. Shorter durations (8 h) required greater concentrations of ozone (0.26 ppm) for effects on epithelial permeability but had no effect on inflammation. The lowest concentration of ozone that had an effect on epithelial permeability or inflammation in the rat was 0.5 ppm for 3 hours. Subchronic exposures in animals suggest that permeability changes are transient (and species-dependent) and return to control levels even with continuing exposure. Chronic animal O<sub>3</sub> exposure studies suggest a role for persistent inflammation in O<sub>3</sub>-induced alterations in lung structure and function. Significant remodeling of epithelium and underlying connective tissues in distal airways have been reported in rat exposed to 0.25 ppm O<sub>3</sub> (12h/day for 6 wk) and in monkeys exposed to 0.2 ppm O<sub>3</sub> (8h/day for 90d). Various factors such as viral infection, chemotactants and oxidized matrix fragments are also implicated in the establishment and persistence of O<sub>3</sub>-induced inflammation.

1 A number of controlled human exposure studies reviewed in the 1996 O<sub>3</sub> AQCD clearly  
2 indicated that a single acute exposure (1-4 h) of humans to moderate O<sub>3</sub> concentrations  
3 (0.2-0.6 ppm) while exercising at moderate to heavy levels results in a number of cellular  
4 and biochemical changes suggesting pulmonary inflammation and increased lung permeability.  
5 Both the inflammatory response and increased lung permeability have been observed as early as  
6 1 h and persisted for at least 18 h. Devlin et al. (1991) reported these changes (increased  
7 neutrophils, inflammatory mediators such as PGE<sub>2</sub> and IL-6) to occur in humans exposed to 0.08  
8 to 0.12 ppm O<sub>3</sub> with moderate exercise for 6.6 h. The newer studies reviewed in this document  
9 (see Chapter 6 for details) have provided additional information on three different aspects of O<sub>3</sub>-  
10 induced inflammatory responses, such as (1) intersubject variability; (2) differential attenuation  
11 profile for various inflammatory markers; and (3) effects of repeated exposures.

12 Mean changes in inflammatory markers seen with exposure to ambient levels of O<sub>3</sub> (Devlin  
13 et al., 1991) exhibited interindividual differences; and, in some individuals, the changes were  
14 comparable to those observed in subjects exposed to 0.4 ppm (as reported by Koren et al., 1989),  
15 suggesting that some individuals in the population may be quite sensitive at ambient levels of O<sub>3</sub>.  
16 Mudway and Kelly (2004) examined O<sub>3</sub>-induced inflammatory responses (PMN influx) and  
17 altered epithelial permeability (protein leakage) via a meta-analysis of 21 controlled human  
18 exposure studies. Their analysis of PMN responses is illustrated in Figure 8-4. Tentatively,  
19 Mudway and Kelly (2004) suggested that the O<sub>3</sub> dose predicted to produce an average PMN  
20 influx exceeding the 95% confidence interval for PMN levels following filtered air (FA)  
21 exposures may, in essence, represent a threshold dose. For a 1 h exposure to 0.12 ppm O<sub>3</sub>, the  
22 threshold dose for early phase PMN responses would not be exceeded unless an individual was  
23 engaged in very heavy exercise ( $V_E = 90$  L/min). However, a longer 8 h exposure to 0.08 ppm  
24 O<sub>3</sub> could reach the early phase PMN dose threshold during relatively light sustained activity  
25 ( $V_E = 17$  L/min). For these same 1- and 8-h exposure scenarios, BALF protein levels would be  
26 predicted to increase by about 1.1-fold. Regarding late phase PMN responses, their threshold  
27 dose was 26% greater than the early phase responses. Mudway and Kelly (2004) did note that  
28 their “threshold” doses were for average PMN responses in healthy adults and that some  
29 individuals would respond at lower doses. Indeed, Krishna et al. (1998) and Stenfors et al.  
30 (2002) observed significant early and late phase PMN responses, respectively, at doses below the  
31 levels tentatively referred to as threshold doses by Mudway and Kelly (2004). Additionally,



**Figure 8-4. Neutrophilia response in the distal airways postexposure (PE) to O<sub>3</sub> or filtered air. Early (0-6 h PE) and delayed (18-24 h PE) responses illustrated in the left and right panel, respectively. Inhaled dose (CVT) is the product of O<sub>3</sub> concentration, minute ventilation per body surface area, and exposure duration. Data are mean (bars are standard errors) of neutrophils (% in BAL) after ozone (▽) or air (●) from 21 studies where subjects (18-40 yrs of age) were exposed to between 0.08 and 0.6 ppm O<sub>3</sub> for 1 to 6.6 h. The dashed lines (- -) are average percent neutrophils following air exposures and the shaded area is the 95% confidence interval (CI). Solid lines (—) illustrate the linear relationship between neutrophil response and O<sub>3</sub> dose with a 95% CI illustrated by dotted lines (····).**

Source: Adapted from Mudway and Kelly (2004).

1 significant inflammatory responses to O<sub>3</sub> exposures that did not elicit significant spirometric  
 2 responses have been reported (Holz et al., 2005; McBride et al., 1994).

3 Soluble mediators of inflammation (e.g., the cytokines IL-6 and IL-8) as well as  
 4 arachidonic acid metabolites (e.g., PGE<sub>2</sub>, PGF<sub>2α</sub>, thromboxane, and leukotrienes [LTs] such  
 5 as LTB<sub>4</sub>) have been measured in the BAL fluid of humans exposed to O<sub>3</sub>. In addition to their  
 6 role in inflammation, many of these compounds have bronchoconstrictive properties and may be

1 involved in increased airway responsiveness following O<sub>3</sub> exposure. The time course for the  
2 inflammatory responses (including recruitment of neutrophils and other soluble mediators) is not  
3 clearly established, but a differential attenuation profile for many of these parameters is evident  
4 from the meta-analysis of 21 controlled human exposure studies reviewed by Mudway and  
5 Kelly (2004).

6 Repeated exposures in humans also indicate ongoing cellular damage irrespective of  
7 attenuation of the inflammatory responses and lung function (Devlin et al. 1997; Jörres et al.  
8 2000). Devlin et al. (1997) examined the inflammatory responses of humans repeatedly exposed  
9 to 0.4 ppm O<sub>3</sub> for 5 consecutive days. Several indicators of inflammation (e.g., PMN influx,  
10 IL-6, PGE<sub>2</sub>, BAL protein, fibronectin) were attenuated after 5 days of exposure (i.e., values were  
11 not different from FA). Several other markers (LDH, IL-8, total protein, epithelial cells) did not  
12 show attenuation, indicating that tissue damage probably continues to occur during repeated  
13 exposure. The recovery of the inflammatory response occurred for some markers after 10 days,  
14 but some responses did not return to normal even after 20 days. When re-exposed 2 weeks later,  
15 changes in BALF indicated that epithelial cells appeared to be fully repaired (Devlin et al.,  
16 1997). Kopp et al. (1999) observed inflammatory responses only after the first O<sub>3</sub> peak in  
17 summer; and its absence late in summer (even after exposure to higher levels of O<sub>3</sub>) may be  
18 due to attenuation of the inflammatory response in the subjects.

19 Numerous studies reported acute O<sub>3</sub>-induced changes in lung epithelial permeability  
20 assessed by indirect assay (increased levels of albumin and protein in BALF). Few other studies  
21 demonstrated O<sub>3</sub>-induced epithelial cell permeability through direct assessment of clearance  
22 of <sup>99m</sup>Tc-DTPA (technetium-99m labeled diethylene triamine pentaacetic acid). For example,  
23 Kehrl et al. (1987) showed increased <sup>99m</sup>Tc-DTPA clearance in healthy young adults at  
24 75 minutes postexposure to 0.4 ppm O<sub>3</sub> for 2 h. More recently, Foster and Stetkiewicz (1996)  
25 have shown that increased <sup>99m</sup>Tc-DTPA clearance persists for at least 18-20 h post-O<sub>3</sub> exposure  
26 (130 min to average O<sub>3</sub> concentration of 0.24 ppm), and the effect is greater at the lung apices  
27 than at the base.

28 Interaction of O<sub>3</sub> with the constituents of the extracellular lining fluid and the induction of  
29 oxidative stress is implicated in injury and inflammation. Animal toxicology and human in vitro  
30 studies that evaluated biochemical mediators implicated in injury and inflammation found  
31 alterations in the expression of cytokines, chemokines, and adhesion molecules, indicative of an



1 ongoing active stress response as well as injury repair and regeneration processes. Both animal  
2 and human studies indicate cellular and biochemical changes associated with inflammation and  
3 increased permeability, but the relationship between these changes and their role in lung function  
4 and airway responses is not known.

### 6 ***Host Defense***

7 Evidence for O<sub>3</sub>-induced dysfunction of host defense components and for subsequent  
8 enhanced susceptibility to bacterial lung infection stems from studies carried out in laboratory  
9 animals. Acute exposures of 0.08 ppm (3 h) O<sub>3</sub> have been implicated in the mortality of mice  
10 due to Streptococcal bacterial infection. Changes in antibacterial defenses are dependent on  
11 exposure regimens, species and strain of test animal, species of bacteria, and age of animal (with  
12 young mice being more susceptible to the effects of O<sub>3</sub>, for example). Animal toxicology studies  
13 indicated that acute O<sub>3</sub>-induced suppression of alveolar phagocytosis and immune functions  
14 observed in animals appeared to be transient and were attenuated with continuous or repeated  
15 exposures. A single study reviewed in the 1996 O<sub>3</sub> AQCD reported decrements in the ability of  
16 alveolar macrophages (AMs) to phagocytose microorganisms upon exposure to 0.08-0.1 ppm O<sub>3</sub>  
17 for 6.6 h (Devin et al., 1991). It has also been reported that O<sub>3</sub> exposures can interfere with  
18 AM-mediated clearance in the respiratory region of the lung and with mucociliary clearance of  
19 the tracheobronchial airways. Ozone-induced perturbations in the clearance process have been  
20 found to be dose-dependent, with low dose exposures accelerating clearance and high doses  
21 slowing the clearance process. Some respiratory tract regional- and species-specific differences  
22 have also been observed.

23 In vitro cultures of epithelial cells obtained from nonatopic and mild atopic asthmatics,  
24 when exposed to O<sub>3</sub> (0.01-0.1 ppm), exhibited significantly increased permeability compared to  
25 cells from normal persons, thus indicating a potential inherent susceptibility of cells from  
26 asthmatics for O<sub>3</sub>-induced permeability. New animal toxicology studies reported O<sub>3</sub>-induced  
27 modulation of cell-mediated immune responses affecting the onset and persistence of infection  
28 in rats (Cohen et al. 2001, 2002). However, there is no compelling evidence from animal  
29 toxicological, human clinical or epidemiologic studies that O<sub>3</sub> enhances the incidence of  
30 respiratory viral infection in humans.

1           The available data at this time indicate that acute O<sub>3</sub> exposure has a potential to impair the  
2 host defense capability, primarily by interfering with the functions of alveolar macrophages.  
3 Any impairment in macrophage function may lead to decreased clearance of microorganisms or  
4 nonviable particles. Compromised alveolar macrophage functions in asthmatics may increase  
5 their susceptibility to other O<sub>3</sub> effects or to the effects of particles.

### 7 ***Biochemical Alterations***

8           An extensive experimental database, including research assessed in 1996 O<sub>3</sub> AQCD,  
9 suggests that potential biochemical alterations in various metabolic pathways (including  
10 xenobiotic metabolism) are involved in lung injury, inflammation, and functional alterations.  
11 Interaction of O<sub>3</sub> with the lipid constituents of pulmonary surfactant has been proposed as one of  
12 the key mechanisms by which O<sub>3</sub> exerts its toxic effects. Experimental evidence clearly  
13 indicates a role for the initial interaction of O<sub>3</sub> with lipid constituents of the ELF and generation  
14 of lipid ozonation products and secondary redox mediators in the initiation of site-specific cell-  
15 injury response cascades. One such lipid ozonation product, 4-hydroxynonenal, has been found  
16 to bind to proteins and increased protein adducts in human alveolar macrophages, suggesting a  
17 role for 4-hydroxynonenal in acute cell toxicity. Cholesterol, the most abundant neutral lipid in  
18 pulmonary surfactant is susceptible to attack by O<sub>3</sub> resulting in multiple oxidized cholesterol  
19 products, including the formation of cholesterol epoxide. A 20-fold increase in cholesterol  
20 epoxide in the BALF from mice exposed to 0.5 ppm O<sub>3</sub> for 3 h suggests a potential role for this  
21 oxidation product in O<sub>3</sub> toxicity (Pulfer et al., 2005). Species- and region-specific increases in  
22 lung xenobiotic metabolism have been observed in response to both short- and long-term O<sub>3</sub>  
23 exposure. It has been well recognized that antioxidants in the ELF confer protection against O<sub>3</sub>  
24 toxicity. But the observation of O<sub>3</sub> reactivity even with environmentally relevant exposures,  
25 questions their ability to quench O<sub>3</sub> reactivity. Species-specific and age-dependent changes in  
26 the antioxidant metabolism add another dimension to their role in this process. Carefully  
27 controlled studies of dietary antioxidant supplementation (Samet et al., 2001; Trenga et al.,  
28 2001) reported some protective effects of α-tocopherol and ascorbate for O<sub>3</sub>-induced spirometric  
29 lung function decrements but not for the intensity of subjective symptoms and inflammatory  
30 responses (including cell recruitment, activation, and a release of mediators). Dietary  
31 antioxidants have also afforded partial protection to asthmatics by attenuating postexposure

1 bronchial hyperresponsiveness (Trenga et al., 2001). In addition, two epidemiologic studies of  
2 street workers and asthmatic children in Mexico City found that subjects taking antioxidant  
3 supplements containing vitamins E and C were protected from O<sub>3</sub>-induced changes in lung  
4 function (Romieu et al., 1998, 2002).

5 Based on the above discussion, it is evident that a very extensive experimental database  
6 accumulated from animal toxicology studies (including nonhuman primate studies) and limited  
7 controlled human exposure studies, has provided important insights into various biochemical,  
8 cellular, and molecular alterations in lung tissue exposed to O<sub>3</sub>. The majority of these studies,  
9 although using acute exposure regimens and relatively high concentrations at times, do provide  
10 credible hypotheses regarding potential molecular mechanisms implicated in O<sub>3</sub> toxicity.  
11 Utilizing this information in relevant rodent-to-human extrapolation models with appropriate  
12 species-specific adjustments may well provide useful information on initial biochemical  
13 alterations that may aid in the development of suitable biomarkers for O<sub>3</sub> exposures/effects.

#### 14 15 ***Cardiovascular Effects***

16 Ozone-induced lung injury and permeability changes, as well as O<sub>3</sub>-induced alterations in  
17 the hemodynamics may lead to O<sub>3</sub> effects on the cardiovascular system. Also, the interaction of  
18 O<sub>3</sub> with ELF, lipids, and surfactants, and the lipid ozonation products and ROS generated in this  
19 process have the potential to penetrate the epithelial barrier and to initiate toxic effects on the  
20 cardiovascular system. An increasing body of animal toxicology evidence suggests that  
21 hematological and thermoregulatory alterations (in heart rate variability and/or core body  
22 temperature) may mediate acute cardiovascular effects. Studies carried out using isolated  
23 perfused rat lung model (Delaunois et al., 1998) indicate inhibition of pulmonary mechanical  
24 reactivity to bronchoconstrictors and persistent vasoreactivity of the vascular bed upon exposure  
25 to O<sub>3</sub> (0.4 ppm for 4 h). Earlier studies in rats indicate a potential role for platelet activating  
26 factor (PAF) in O<sub>3</sub>-inflammatory response. Recent observations of O<sub>3</sub>-induced generation of  
27 oxysterols and β-epoxides from cholesterol in surfactant suggest that these lipid ozonation  
28 products like lysophospholipids may initiate PAF-like activity and initiate clotting and  
29 thrombolytic effects in the cardiovascular system.

30 A few human experimental studies have examined the potential effects of O<sub>3</sub> exposure on  
31 cardiovascular functions. For example, Gong et al (1998) evaluated various cardiac function and

1 hemodynamic variables in healthy and hypertensive adult males and observed impairment of  
2 alveolar-arterial oxygen transfer, leading them to suggest that such an impairment could lead to  
3 decreased oxygen supply to the myocardium. Also, Foster et al (1993; 1997) have reported  
4 O<sub>3</sub>-induced ventilation-perfusion mismatch. Such an altered ventilation distribution profile  
5 observed even in relatively young healthy adults could contribute to the alveolar-arterial oxygen  
6 transfer impairment reported by Gong et al. (1998). Taken together, these observations support  
7 the regional differences in ventilation and perfusion in severe COPD patients reported by King  
8 and Briscoe (1968) and Kronenberg et al. (1973). Such preexisting compromised gas exchange  
9 abnormalities would likely make lungs of individuals with COPD more vulnerable to O<sub>3</sub>-induced  
10 gas exchange inhibition and reduced oxygen saturation.

### 11 **8.4.3 Assessment of Epidemiological Evidence**

12  
13 Based on the O<sub>3</sub> epidemiologic evidence available at the time, the 1996 O<sub>3</sub> AQCD arrived  
14 at the following conclusions:

15  
16 An association between daily mortality and O<sub>3</sub> concentration for areas with high  
17 O<sub>3</sub> levels (e.g., Los Angeles) has been suggested, although the magnitude of such  
18 an effect is unclear. Increased O<sub>3</sub> levels are associated with increased hospital  
19 admissions and emergency department visits for respiratory causes. Analyses  
20 from data in the northeastern United States suggest that O<sub>3</sub> air pollution is  
21 associated with a substantial portion (on the order of 10 to 20%) of all summertime  
22 respiratory hospital visits and admissions. Pulmonary function in children at  
23 summer camps in southern Ontario, Canada, in the northeastern United States,  
24 and in Southern California is associated with O<sub>3</sub> concentration.” (U.S. EPA, 1996,  
25 p1-29).

26  
27 The 1996 O<sub>3</sub> AQCD further stated that only suggestive epidemiologic evidence existed for health  
28 effects of chronic ambient O<sub>3</sub> exposure in the population, and this was partly due to an inability  
29 to isolate potential effects related to O<sub>3</sub> from those of other pollutants, especially PM (U.S.  
30 Environmental Protection Agency, 1996).

31 The discussion in this section of scientific strength and limitations of the growing body of  
32 epidemiologic evidence for associations between ambient exposure to O<sub>3</sub> and various health  
33 effects discussed is based primarily on Chapter 7 evaluations. The following criteria were  
34 considered in assessing the relative scientific quality of the epidemiologic studies: (1) the  
35 *strength* of reported associations, in terms of magnitude, statistical significance and statistical  
36 power of effects estimates; (2) *robustness* of reported associations (based on defined health  
37 endpoint criteria), potential confounding by copollutants; (3) *consistency and coherency* of the

1 effect associations (4) *temporality*, in terms of lag periods between exposure and observed  
2 effects; and (5) *biological plausibility* of the observed O<sub>3</sub>-related health effects assessed in terms  
3 of their coherence in relation to findings derived from controlled human exposure studies which,  
4 overall, provide insights into the plausibility of reported O<sub>3</sub> human health effects reflecting  
5 causal relationships.

6 Many newly available epidemiologic studies have provided additional evidence for  
7 O<sub>3</sub>-related health effects beyond that previously known. Significant statistical associations have  
8 been reported by various investigators between acute O<sub>3</sub> exposure and several respiratory and  
9 cardiovascular health endpoints, including: mortality; hospital admissions; emergency  
10 department visits; respiratory illness and symptoms; and changes in pulmonary function.  
11 Similarly, associations have been reported between long-term exposure to O<sub>3</sub> and increased  
12 morbidity; development of respiratory disease; and declines in lung function and lung function  
13 growth. The numerous new epidemiological studies that have been conducted in areas across the  
14 United States and Canada, as well as in Europe, Latin America, Australia and Asia, are  
15 summarized in the Annex to Chapter 7. Based on evidence extracted from the full body of  
16 epidemiologic studies carried out and reviewed since the 1996 O<sub>3</sub> AQCD, it has been well  
17 demonstrated that deleterious human health outcomes are positively associated with acute  
18 ambient O<sub>3</sub> concentrations currently encountered in the United States.

#### 19 20 **8.4.3.1 Strength and Consistency of Epidemiological Associations**

21 As quoted above, assessments in the 1996 O<sub>3</sub> AQCD supported a consistent relationship  
22 between O<sub>3</sub> concentration and respiratory illness, hospital visits and reduced lung function.  
23 However, due to insufficient evidence examining O<sub>3</sub>-mortality associations and uncertainties  
24 regarding weather model specification, the 1996 O<sub>3</sub> AQCD was limited to only a very qualitative  
25 assessment of O<sub>3</sub>-mortality associations. Since then, Generalized Additive Models (GAMs) have  
26 become widely utilized for epidemiologic time-series analysis of health effects attributable to air  
27 pollution, increasing our confidence in quantitative estimation of O<sub>3</sub>-mortality risks. Some  
28 concerns have been raised regarding the use of default convergence criteria in applications of  
29 commercially available software employed for GAM analyses to estimate air pollution-related  
30 health effects, as discussed in the 2004 PM AQCD (U.S. Environmental Protection Agency,  
31 2004a). However, reanalyses of a number of studies, comparing results using default GAM

1 convergence criteria to results from analyses using stringent GAM convergence criteria and/or  
2 from GLM analyses, found little difference among the O<sub>3</sub> effect estimates obtained (as discussed  
3 in Chapter 7 of this document). Overall, the magnitude of the effect-size estimates observed  
4 for O<sub>3</sub>-mortality relationships tend to be relatively consistent across the newly available studies.  
5 The effect estimates for O<sub>3</sub>-morbidity endpoints have greater variability, but consistent positive  
6 associations between ambient O<sub>3</sub> and various health outcomes have also been observed.

#### 8 **8.4.3.1.1 Acute Exposure Studies**

9 Numerous epidemiological studies carried out over the past decade have added evidence to  
10 the knowledge base assessed in the 1996 O<sub>3</sub> AQCD, which included both (a) individual-level  
11 camp and exercise studies that established a relationship between ambient O<sub>3</sub> exposure and  
12 human lung function decline and (b) aggregate time-series studies that suggested positive  
13 relationships for O<sub>3</sub>-related respiratory morbidity. The new studies reviewed in Chapter 7 in this  
14 document include numerous field/panel studies and time-series studies from various regions.  
15 In field/panel studies on the effects of air pollution exposure, the most common health outcomes  
16 measured were lung function and respiratory symptoms. The time-series studies examined daily  
17 emergency department visits, hospital admissions, and mortality data.

#### 19 **Field/Panel Studies of Acute Exposure Effects**

##### 20 ***Pulmonary Function And Respiratory Symptoms***

##### 21 *Healthy Individuals*

22 Many of the new field/panel studies reviewed in Chapter 7 and the controlled human  
23 exposure studies reviewed in Chapter 6 of this document provide additional data supporting two  
24 major findings reported in the 1996 O<sub>3</sub> AQCD. First, acute O<sub>3</sub> exposure is associated with a  
25 significant decline in lung function parameters. Ozone-related lung function decrements were  
26 most notable in children and asthmatics. In addition, adults who work or exercise outdoors also  
27 were found to be vulnerable to O<sub>3</sub>-associated declines in lung function due to their increased  
28 exposure to O<sub>3</sub>. Second, acute exposure to O<sub>3</sub> is associated with increased respiratory symptoms,  
29 particularly cough, and increased as-needed medication use in asthmatic children. Immediate  
30 effects of O<sub>3</sub> were observed on both lung function and respiratory symptoms, with the strongest

1 associations often observed at a lag of 0- or 1-day. The two health outcomes are further  
2 discussed below.

3 Pulmonary function was determined by either spirometry (forced expiratory volume in  
4 1 s [FEV<sub>1</sub>] and forced vital capacity [FVC]) or by peak expiratory flow (PEF) meters. The  
5 spirometric parameter, FEV<sub>1</sub> is a strong and consistent measure of lung function. PEF is a  
6 closely related but different metric of lung function, and PEF is more feasibly performed in field  
7 studies, using inexpensive peak flow meters that produce similar results to PEF measured  
8 spirometrically.

9 In a number of newly available field/panel studies, FEV<sub>1</sub> was measured in panels of  
10 exercising children, outdoor workers, and adult hikers exposed to ambient O<sub>3</sub> while experiencing  
11 elevated exertion levels. Collectively, the results of the new studies (discussed in Section  
12 7.2.3.1) confirm and extend those from analogous field/panel studies assessed in the 1996 O<sub>3</sub>  
13 AQCD and findings from experimental controlled human exposure studies indicating that  
14 acute O<sub>3</sub> exposures prolonged over several hours and combined with elevated levels of exertion  
15 or exercise magnify O<sub>3</sub> effects on lung function, as evaluated in terms of FEV<sub>1</sub>.

16 For example, six field studies by three different research groups of 7- to 17-year-old,  
17 healthy (nonasthmatic) children exposed for several hours to ambient O<sub>3</sub> during increased  
18 physical exertion in summer camp activities were assessed in the 1996 O<sub>3</sub> AQCD. When  
19 analyzed together by consistent statistical methods, the data from those studies showed an  
20 average relationship between afternoon FEV<sub>1</sub> and concurrent 1-h O<sub>3</sub> concentrations of  
21 -0.50 mL/ppb (95% CI: -0.63, -0.36), with individual slopes ranging from -0.19 to  
22 -1.29 mL/ppb (Kinney et al., 1996). Four new field/panel studies (assessed in Section 7.2.3.1 of  
23 this document) that evaluated pulmonary function in healthy school-aged children exposed to  
24 mean 1-h max O<sub>3</sub> concentrations ranging from 20 to 112 ppb found exposure-response functions  
25 of approximately -0.18 to -1.42 mL/ppb. Also, two other studies assessed in the 1996 O<sub>3</sub>  
26 AQCD that measured lung function before and after well-defined exercise events (1/2-h long) in  
27 adults during exposures to ambient O<sub>3</sub> across 4 to 135 ppb found exposure-response slopes of  
28 -0.4 mL/ppb (95% CI: -0.7, -0.1) (Selwyn et al., 1985) and -1.35 mL/ppb (95% CI: -2.04,  
29 -0.66) (Spektor et al., 1988). In comparison, new studies of healthy adult workers (street  
30 workers, berry pickers) and hikers engaged in prolonged (≥6 h) strenuous physical exertion at  
31 mean exposure levels ranging from 40 to 123 ppb 1-h max O<sub>3</sub> reported exposure-response slopes

1 of -1.40 to -3.8 mL/ppb (as assessed in Section 7.2.3.1 in Chapter 7 of this document). The  
2 most representative data are those of Korrick et al. (1998) from a U.S. study of adult hikers that  
3 provided outcome measures stratified by gender, age, smoking-status, and presence of asthma  
4 within a population capable of above-normal exertion.

5 Pulmonary function changes or declines measured in susceptible populations suggest that  
6 juvenile asthmatics are at greater risk with the largest O<sub>3</sub>-related decline of FEV<sub>1</sub> of -2.08%  
7 (95% CI: -6.24, 2.08) per 40 ppb increase in 1/2- h max O<sub>3</sub> at a 2-day lag (Höppe et al., 2003).  
8 An extended analysis by Höppe et al. (2003) examined individual susceptibility to O<sub>3</sub> effects.  
9 Ozone responders were regarded as those with a greater than 10% change in FEV<sub>1</sub>. Compared to  
10 athletes and the elderly, a greater percentage of children and asthmatics (20% versus 5%) were  
11 found to be sensitive to O<sub>3</sub> effects on lung function. The small sample sizes in these studies limit  
12 extrapolation to larger populations; however, the studies indicate a trend that should be evaluated  
13 in larger epidemiologic studies.

#### 14 *Asthma Panels*

15 Several studies assessed in the 1996 O<sub>3</sub> AQCD that evaluated elevated respiratory  
16 symptoms and/or pulmonary function decrements in asthmatic children showed greater  
17 responses in asthmatic than in nonasthmatic subjects, suggesting that asthmatic individuals  
18 might constitute a sensitive population group in O<sub>3</sub> epidemiologic studies.

19 Additional panel studies carried out over the past decade to understand the effect of acute  
20 exposure to O<sub>3</sub> in asthmatics evaluated lung function by PEF and/or respiratory symptoms (i.e.,  
21 cough, wheeze, shortness of breath and medication use) ascertained by questionnaire. Several  
22 additional studies (see Figures 7-1 and 7-2), both in the U.S. and in other countries, reported  
23 decrements in PEF to be associated with O<sub>3</sub> exposures among asthmatics. One large U.S.  
24 multicity study (Mortimer et al., 2002) associated O<sub>3</sub> concentrations with the incidence of ≥10%  
25 declines in morning PEF (odds ratio of 1.30 [95% CI: 1.04, 1.61] per 30 ppb increase in 8-h  
26 avg O<sub>3</sub> for a 5-day cumulative lag). In a group of adult hikers in Mount Washington, NH  
27 (Korrick et al., 1998), asthmatic subjects experienced a four-fold greater decline in FEV<sub>1</sub>  
28 compared to healthy individuals with the same exposure to O<sub>3</sub>. Asthmatic hikers experienced a  
29 mean change of -4.47% (95% CI: -7.65, -1.29) per 30 ppb increase in 8-h avg O<sub>3</sub> while other  
30 hikers had a mean change of -1.08% (95% CI: -2.49, 0.33).  
31



1 Several single-city studies did not observe statistically significant declines in lung function  
2 parameters (e.g., Delfino et al., 1997; Hilterman et al., 1998), which might be partially  
3 attributable to small sample sizes and/or low levels of O<sub>3</sub>. Although results were not always  
4 statistically significant in these single-city studies, effect estimates were consistently negative,  
5 providing suggestive evidence that O<sub>3</sub> was associated with lung function declines among  
6 asthmatics. Collectively, results from the large multicities study by Mortimer et al. (2002), as  
7 well as those from the smaller single-city studies suggest that exposure to O<sub>3</sub> may be associated  
8 with declines in lung function in this potentially susceptible population.

9 The majority of studies that evaluated respiratory symptoms (i.e., cough, shortness of  
10 breath, and wheeze) and the increased use of asthma medication related to O<sub>3</sub> exposure are also  
11 focused on asthmatic children. Two large U.S. studies (Mortimer et al., 2002; Gent et al., 2003)  
12 and some international studies (Hilterman et al., 1998; Desqueyroux et al., 2002a,b) suggest  
13 positive associations between O<sub>3</sub> ambient concentrations and increased symptoms or asthma  
14 medication use. In the multicities study by Mortimer et al. (2002), the odds ratio for the  
15 incidence of symptoms (including cough, chest tightness, and wheeze) was 1.35 (95% CI: 1.04,  
16 1.69) per 30 ppb increase in 8-h avg O<sub>3</sub> for a 4-day cumulative lag. As in the case of studies  
17 examining the O<sub>3</sub> effect on lung function, not all studies on respiratory symptoms observed  
18 consistent positive associations with O<sub>3</sub>. For example, Avol et al. (1998) studied symptoms in  
19 asthmatic, wheezy, and healthy children aged 10 to 12 years in southern California. Some  
20 symptom associations were noted but they were inconsistent, possibly due to relatively low O<sub>3</sub>  
21 concentrations during the study period. The authors also noted that the study children did not  
22 spend substantial time outdoors engaged in physical activities. Once again, the strong evidence  
23 from the large multicities study by Mortimer et al. (2002), along with less consistent but  
24 generally supportive evidence from several single-city studies suggest that O<sub>3</sub> exposure may be  
25 associated with increased respiratory symptoms and medication use in asthmatic children.

### 26 *School Absenteeism*

27 Two large U.S. studies (Chen et al., 2000; Gilliland, 2001) and one study from Seoul,  
28 Korea (Park et al., 2002) investigated the relationship between ambient O<sub>3</sub> concentrations and  
29 school absenteeism. Results from all these studies suggested a positive association between O<sub>3</sub>  
30 and absences from school, with each one arriving at these associations using different lag  
31

1 periods. Chen et al. (2000) reported a 10.4% (95% CI: 2.7, 18.1) excess rate of total daily  
2 school absences for 40 ppb increase in daily 1-h max O<sub>3</sub> with a distributed lag of 1 to 14 days.  
3 The 12 southern California communities study by Gilliland et al. (2001) also reported larger  
4 O<sub>3</sub>-related school absences due to respiratory causes, 147% (95% CI: 6, 478) per 30 ppb  
5 increase in 8-h -avg O<sub>3</sub> with a 30-day lag, compared to nonrespiratory causes, 61% (95% CI:  
6 9, 137). The studies reported by Park et al. (2002) were analyzed using GAM with default  
7 convergence criteria and indicated a positive association with same day O<sub>3</sub> (16% [95% CI: 12,  
8 22] per 30 ppb increase in 8-h -avg O<sub>3</sub>). Results from the three studies listed above suggest that  
9 ambient O<sub>3</sub> concentrations may be associated with school absenteeism, particularly illness-  
10 related absences. However, the associations observed during the long lag period of two to four  
11 weeks may reflect confounding by other time-varying factors or be a chance finding from an  
12 exploratory analysis. Additional studies and analysis using similar lag periods are needed to  
13 more clearly delineate quantitative relationships between ambient O<sub>3</sub> and school absences.  
14

#### 15 *Field Studies on Cardiovascular Effects*

16 A limited number of studies evaluated potential short term effects of air pollution on  
17 cardiovascular functions. Several of these studies evaluated the effects of PM, O<sub>3</sub> and other  
18 gaseous pollutants. Two major U.S. population-based studies (Liao et al., 2004; Park et al.,  
19 2005) suggested an association between short-term O<sub>3</sub> exposure and decreased heart rate  
20 variability (HRV). Park et al., (2005) reported stronger associations of HRV with PM<sub>2.5</sub> and O<sub>3</sub>  
21 in people with ischemic heart disease and hypertension. These results are consistent with a  
22 Mexico City study that observed an O<sub>3</sub>-induced HRV effect in individuals with hypertension  
23 (Holguin et al., 2003). Several other studies, on the other hand, did not find any such  
24 relationship, but these studies might have had limited power (e.g., low O<sub>3</sub> concentration ranges,  
25 small sample sizes) to examine the subtle effects. Studies that evaluated the relationship  
26 between air pollutants and the onset of myocardial infarction (Ruidavets et al., 2005; Peters  
27 et al., 2001) suggested a positive association with O<sub>3</sub>. However, due to lack of information on  
28 potential confounding by PM and the limited number of studies available, additional research is  
29 needed to confirm these observations.

30 Only a limited number of epidemiologic studies examined cardiovascular outcomes in  
31 relation to O<sub>3</sub> exposures. Among them, the larger population-based studies (Liao et al., 2004;

1 Park et al., 2005; Ruidavets et al., 2005) observed suggestive evidence of an association of O<sub>3</sub>  
2 exposure with decreased HRV and increased incidence of myocardial infarctions.

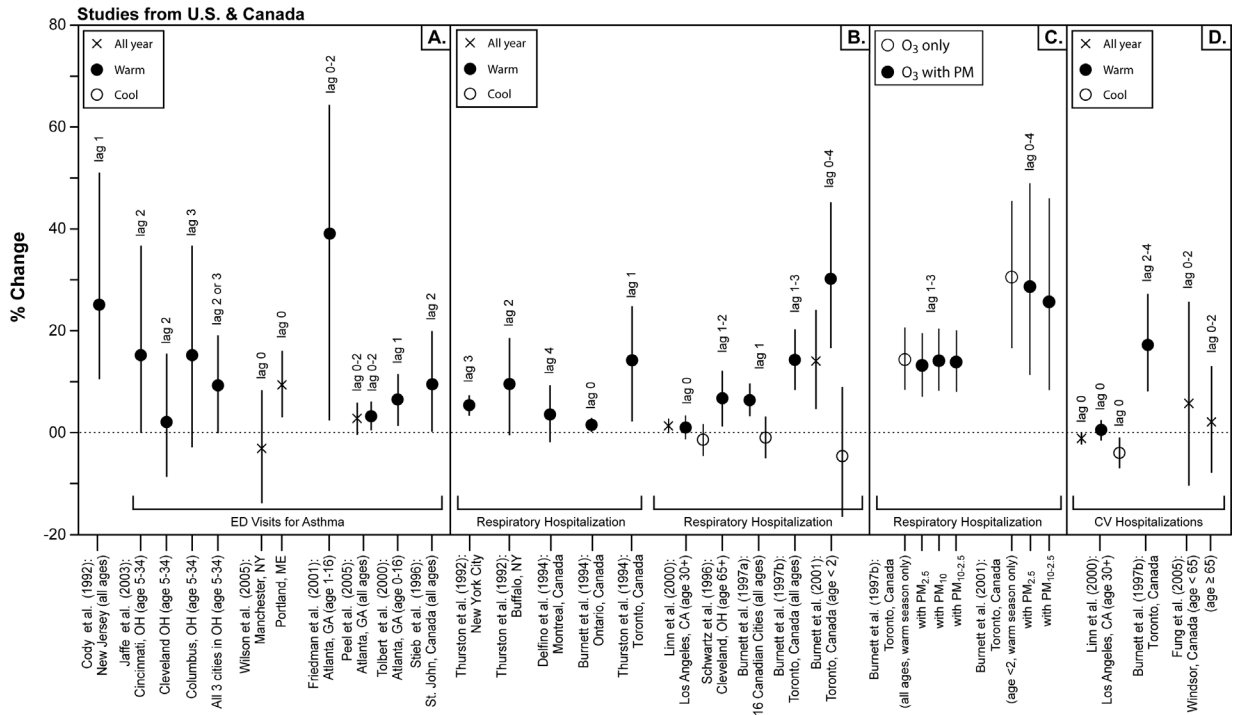
### 4 **Time-Series Analyses of Acute Exposure Effects**

#### 5 ***Emergency Department Visits and Hospital Admissions***

6 Many time-series studies reviewed in the 1996 O<sub>3</sub> AQCD indicated positive associations  
7 between O<sub>3</sub> air pollution and increased hospital admissions. Strong evidence establishing a  
8 correlation between O<sub>3</sub> exposure and increased exacerbations of preexisting respiratory disease  
9 in the general public were reported at 1 h-maximum O<sub>3</sub> concentrations <0.12 ppm.

10 Several studies published during the past decade examined temporal associations between  
11 O<sub>3</sub> exposures and emergency department visits or hospital admissions for respiratory diseases  
12 (see Table AX7-2 in Chapter 7 Annex). One of these studies by Peel et al. (2005) reported  
13 stronger and more positive associations between O<sub>3</sub> and emergency department visits due to  
14 respiratory-related diseases in the warm season. A 3.1% (95% CI: 0.2, 6.2) excess risk in  
15 asthma visits was associated with a standardized increment of 30 ppb in 8-h max O<sub>3</sub> (see  
16 Section 7.1.3.2). This risk was significantly associated with respiratory infections, when  
17 adjusted for PM<sub>10</sub>, NO<sub>2</sub>, and CO in multipollutant models. Several other U.S. and Canadian  
18 studies reported positive associations between O<sub>3</sub> concentrations and emergency department  
19 visits due to respiratory causes (Figure 8-5). However, several of the European studies observed  
20 no association between O<sub>3</sub> concentrations and emergency department visits for respiratory  
21 diseases. These inconsistent results might be partially attributable to differences in model  
22 specifications and statistical methods used to evaluate seasonal patterns and potential  
23 confounding by copollutants. Overall, then, the current body of evidence remains inconclusive  
24 regarding ambient O<sub>3</sub> effects on risk of emergency departments visits. Additional studies are  
25 needed to establish stronger and more convincing associations between increased concentrations  
26 of ambient O<sub>3</sub> and increased risk of emergency department visits.

27 Studies of acute O<sub>3</sub> exposure effects on respiratory disease-related hospital admissions  
28 (summarized in Section 7.3.3) have considered various factors in their analyses. The hospital  
29 admission data were assessed based on the type of respiratory disease (such as asthma, COPD),  
30 seasonal effects (summer vs. winter, O<sub>3</sub> concentration and temperature), age of the study  
31 population, studies carried out in single or multiple cities, effect of confounders and lag days



**Figure 8-5. Ozone-associated percent change (95% CI) in emergency department visits for asthma (A), total respiratory hospitalization by season (B), respiratory hospitalization with adjustment for PM indices (C) and (D) total cardiovascular hospitalization per standardized increment (see Section 7.1.3.2). Only results for U.S. and Canadian studies are presented.**

1 between exposure and hospital admissions (Figure 8-5). Two large studies (Burnett et al.,  
 2 1997a; Anderson et al. 1997) that used consistent analytical methodologies found significant  
 3 associations between O<sub>3</sub> concentrations and increased risk for unscheduled hospital admissions  
 4 for respiratory-related diseases, despite differences in geographic locations of the study  
 5 populations. In both studies, larger effects were observed in the warm season. The 16 cities  
 6 Canadian study by Burnett et al. (1997a) analyzed data for a population of 100,000 over a period  
 7 of 10 years. During the summer, an excess risk of 6.7% (95% CI: 3.5, 10.0) per 40 ppb standard  
 8 increment in 1-h max O<sub>3</sub> was observed. The five cities APHEA (Air Pollution on Health:  
 9 European Approach) study by Anderson et al. (1997) also reported significant associations  
 10 between O<sub>3</sub> and hospitalizations for COPD. An excess risk of 4.7% (95% CI: 1.6, 7.9) per 40  
 11 ppb increase in 1-h max O<sub>3</sub> was observed in the warm season. Several other studies (single city

1 or two city studies for five or more years) also suggested a positive association between increase  
2 in O<sub>3</sub> and increased risk of hospital admissions.

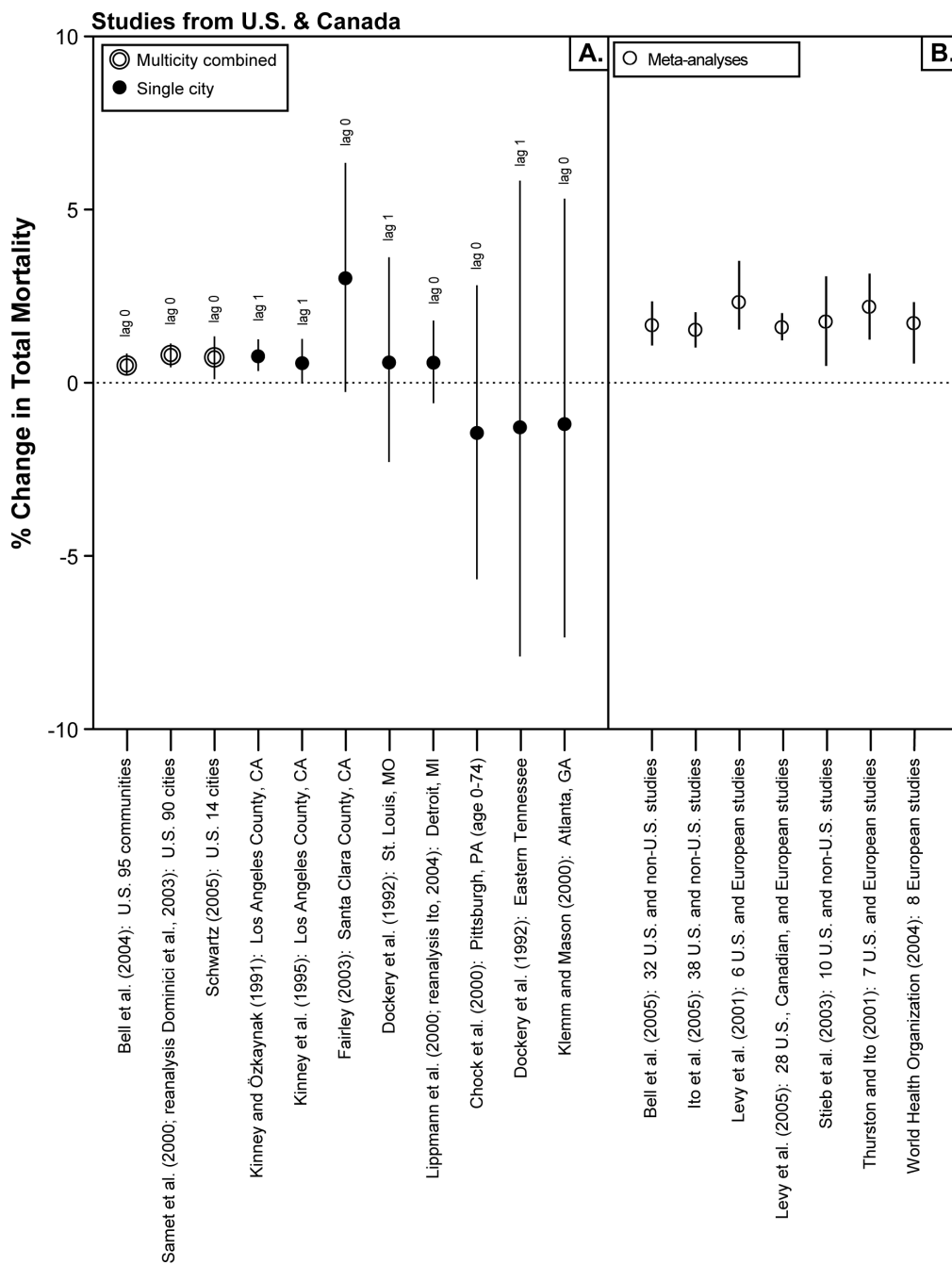
3 Many other studies reported less consistent or no associations between increases in O<sub>3</sub>  
4 concentrations and hospital admissions. A few other studies raise questions and concern about  
5 other factors in this relationship. Despite these inconsistencies noted across the studies, the  
6 collective evidence supports the findings of significant and robust effects of O<sub>3</sub> on respiratory  
7 hospitalization outcomes. Large multicity studies as well as several individual city studies have  
8 reported positive O<sub>3</sub> associations with total respiratory, asthma, and COPD hospitalizations,  
9 especially in those that analyzed the O<sub>3</sub> effect during the summer or warm season.

10 A subset of hospital admissions studies examined the effect of O<sub>3</sub> on cardiovascular  
11 outcomes (see Figure 8-5). The evidence is inconclusive on the association between O<sub>3</sub> exposure  
12 and cardiovascular hospitalizations with regard to year-round data. However, in studies that  
13 adjusted for seasonal or meteorological factors, there was suggestive evidence that O<sub>3</sub> was  
14 associated with increased risk of cardiovascular hospital admissions in the warm season.

#### 15 16 *Mortality-Related to Short-term O<sub>3</sub> Exposures*

17 Due to the limited number of studies and uncertainties regarding weather model  
18 specifications, no meaningful quantitative assessment of O<sub>3</sub>-mortality associations was possible  
19 in the 1996 O<sub>3</sub> AQCD. However, newly available large multicity studies designed specifically to  
20 examine the effect of O<sub>3</sub> on mortality have provided much more robust and credible information.

21 Two large multicity studies from the U.S. (Bell et al., 2004; Schwartz et al., 2005) and one  
22 from Europe (Gryparis et al., 2004) specifically evaluated O<sub>3</sub> effects on mortality and indicated  
23 positive associations between increased O<sub>3</sub> levels and mortality. Among the positive studies,  
24 risk estimates for (U.S. and Canadian) single-city studies carried out using a single-pollutant  
25 model are in the range of 0.8 to 3% excess deaths per 40 ppb increase in 1-h max O<sub>3</sub> (Figure  
26 8-6), while multicity studies and meta-analyses reported a risk estimate in the range of 0.5 to 2%  
27 excess risk with identified heterogeneity (due to model specifications) across cities and studies  
28 (Figure 8-6). Models examining different lag times, observed that there was an immediate effect  
29 of O<sub>3</sub> on mortality which persisted over several days, resulting in risk effect estimates at a  
30 cumulative lag of 0 to 7 days.



**Figure 8-6. A. All cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) for all year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted. Only results from single-day lag models are presented.**

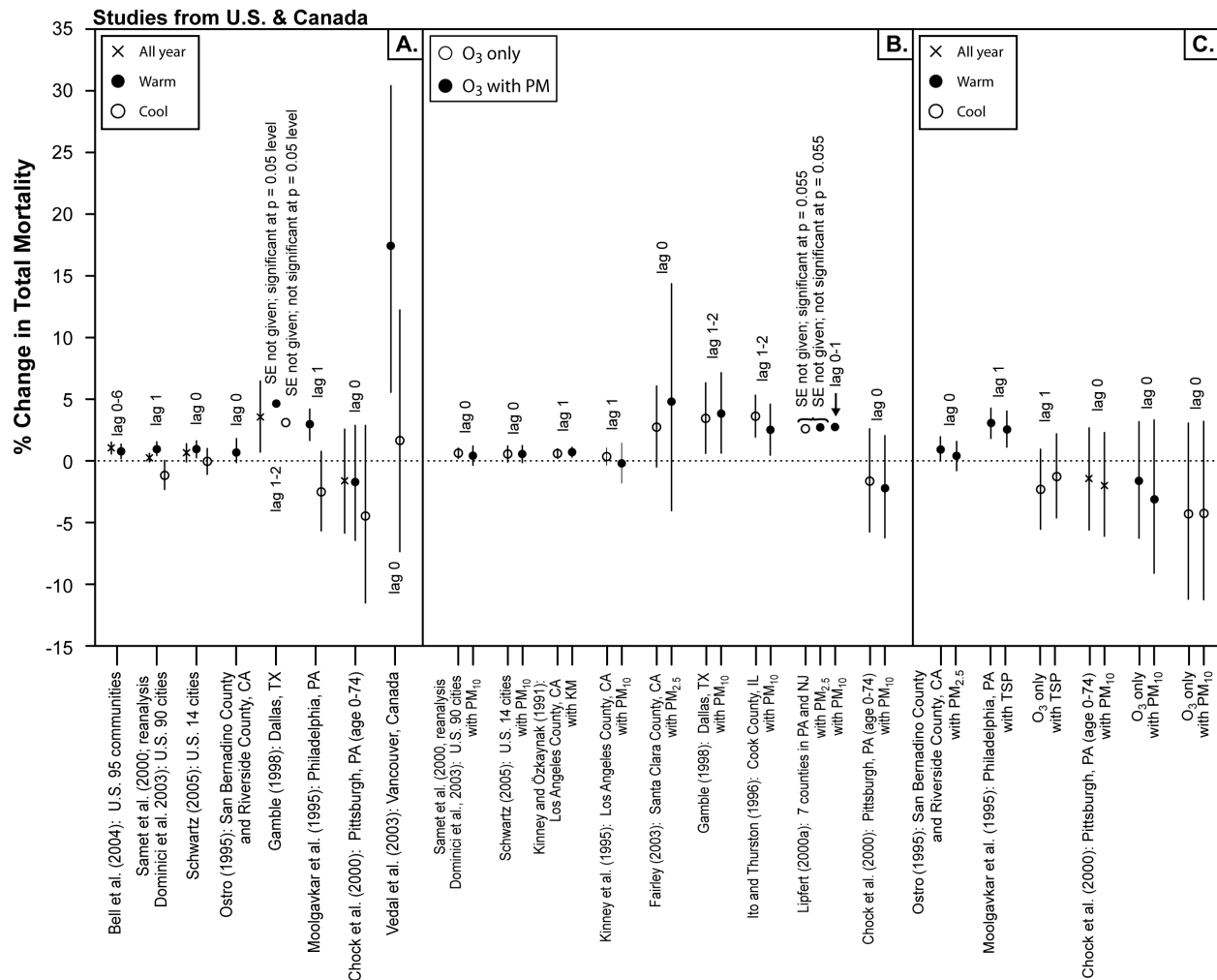
**B. Combined all cause (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) from recent meta-analyses per standard increment of 40 ppb in 1-h max O<sub>3</sub> or equivalent. Note that all meta-analyses, except Stieb et al. (2003), included studies which used Poisson GAM with default convergence criteria.**

1           Some of the epidemiological studies and, particularly, meta-analyses examined the  
2 influence of season on O<sub>3</sub> mortality associations and indicated larger O<sub>3</sub>-mortality risk estimates  
3 in the warm season compared to the colder season. These estimates appear to be consistent with  
4 a causal association when O<sub>3</sub> levels are high in the warm season. This seasonal dependence  
5 of O<sub>3</sub>-mortality effects complicates the evaluation and interpretation of risk estimates from year-  
6 round data without adjustment for temporal trends (Figure 8-7). The confounding effects of  
7 copollutants were examined by three recent meta-analyses (Bell et al., 2005; Ito et al., 2005;  
8 Levy et al., 2005). These, as well as other multicity and single-city studies, indicated that  
9 copollutants do not appear to substantially confound O<sub>3</sub>-mortality associations. Meta-analyses  
10 on the association between cause-specific mortality and O<sub>3</sub> levels observed larger positive  
11 associations with cardiovascular mortality compared to total mortality (Figure 8-8).  
12 Additional analyses carried out to examine specific population groups potentially susceptible  
13 to O<sub>3</sub>-mortality effects did not clearly identify any specific group, but they did suggest that  
14 severe asthmatics and elderly populations might be relatively more susceptible to O<sub>3</sub>.

#### 16 **8.4.3.1.2 Chronic Ozone Exposure Studies**

17           There were a limited number of studies reported in the 1996 O<sub>3</sub> AQCD that addressed  
18 potential health effects of long-term ambient O<sub>3</sub> exposures. Several longitudinal epidemiological  
19 studies carried out in the past decade evaluated the potential effects of chronic (several weeks to  
20 many years) O<sub>3</sub> exposure on lung function, respiratory symptoms, lung inflammation, asthma  
21 prevalence, cancer incidence, and mortality. Based on the available data at this time, no clear  
22 conclusions can be drawn now regarding the relationship between chronic O<sub>3</sub> exposure and such  
23 health outcomes. A limited number of studies also examined the potential effects of ambient O<sub>3</sub>  
24 exposure on birth defects, and these also suggest the need for additional studies and a larger  
25 database before drawing any conclusions regarding possible associations.

26           Very few studies have investigated the effects of long-term O<sub>3</sub> exposure on incidence of  
27 cancer and mortality. Uncertainties regarding the exposure period of relevance and  
28 inconsistencies across mortality outcomes and gender raise concerns regarding plausibility.  
29 The largest and most representative U.S. study, by Pope et al. (2002), observed positive but  
30 nonsignificant associations between O<sub>3</sub> exposure and all cause, cardiopulmonary, and lung



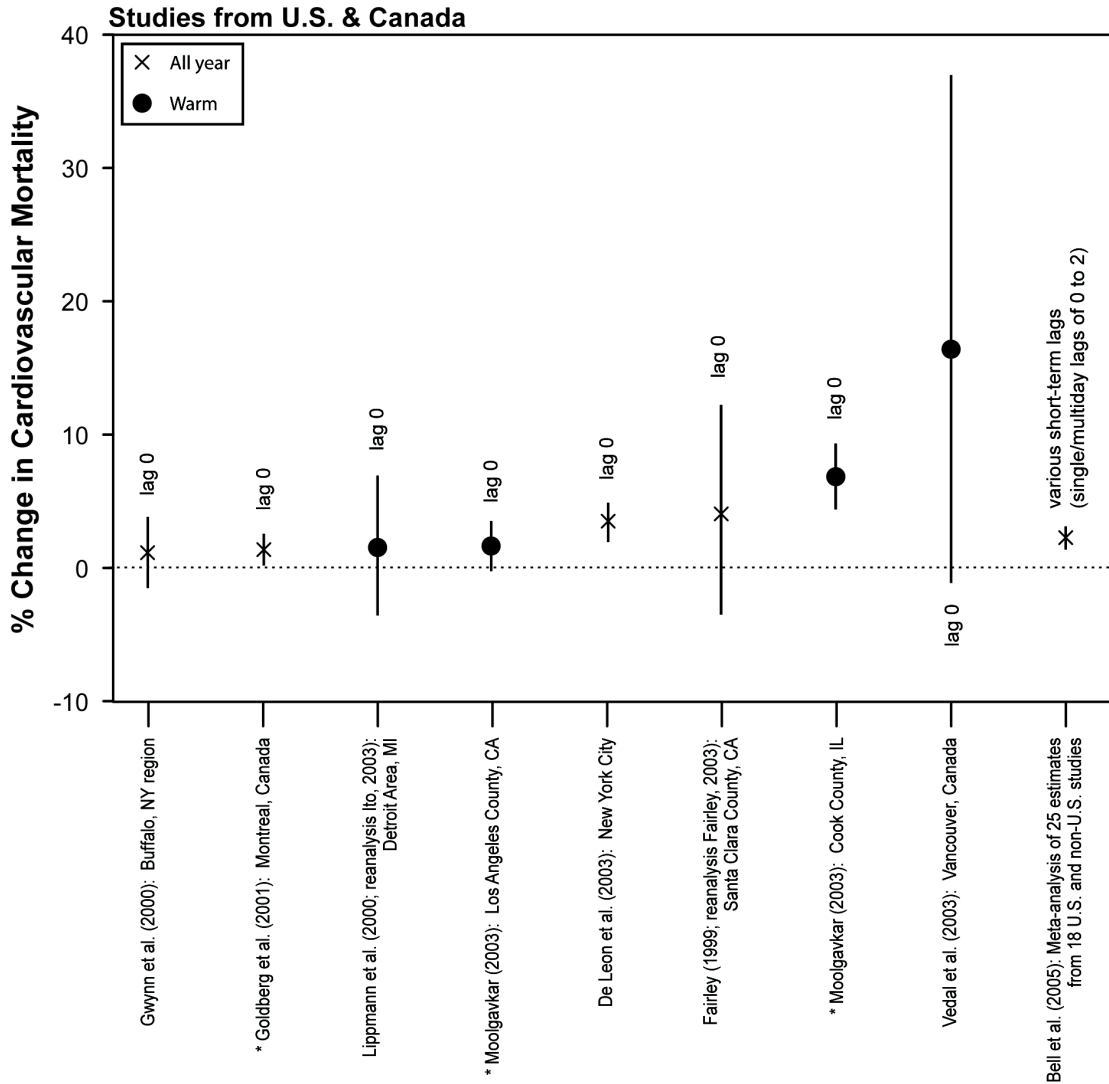
**Figure 8-7. All causes (nonaccidental) O<sub>3</sub> excess mortality risk estimates (95% CI) per standardized increment (see Section 7.1.3.2). (A) by season; (B) with adjustment for PM indices in all year analysis and (C) with adjustment for PM indices by season. Note: Only U.S. and Canadian studies presented.**

1 cancer mortality. Thus, the current evidence is inconclusive regarding a potential relationship  
 2 between chronic O<sub>3</sub> exposure and increased mortality risk.

### 4 8.4.3.2 Robustness of Epidemiological Associations

5 In evaluating the strength of the epidemiological evidence, the magnitude of observed O<sub>3</sub>  
 6 effect estimates and their statistical significance is important; however, consideration must be  
 7 given to the precision of the effect estimates and the robustness of the effects associations.





**Figure 8-8. Ozone-associated percent change (95% CI) in cardiovascular risk estimates per standardized increment (see Section 7.1.3.2). Analyses are for all ages. Only studies for U.S. and Canada presented. \*Analyses using GAM default convergency criteria.**

1 Examining the robustness of the associations includes evaluating the impact of alternative  
 2 models, model specifications for temporal trends and meteorological factors, and potential  
 3 confounding by copollutants. Also of interest are issues related to exposure assessment and  
 4 measurement error. A detailed discussion on each of these topics can be found in Chapter 7  
 5 (Sections 7-1 and 7-6). The following sections focus on the extent to which the current  
 6 epidemiological findings can be considered robust.

#### 1 **8.4.3.2.1 Exposure Issues: Ambient versus Personal**

2 In many air pollution epidemiologic studies, especially time-series studies utilizing  
3 administrative data on mortality and hospitalization outcomes, data from central ambient  
4 monitoring sites are generally used as the estimate of exposure. Personal exposures of individual  
5 study participants generally are not directly observed in epidemiologic studies. The relationship  
6 between ambient O<sub>3</sub> concentrations and personal O<sub>3</sub> exposure levels varies, depending on factors  
7 such as time spent outdoors, ventilation conditions, personal factors, and air quality indices.  
8 There is suggestive evidence that ambient O<sub>3</sub> concentrations from central monitors may serve as  
9 valid surrogate measures for aggregate personal O<sub>3</sub> exposures in time-series studies. However,  
10 ambient concentrations generally overestimate true personal O<sub>3</sub> exposures. Thus, use of ambient  
11 concentrations in risk calculations will likely result in effect estimates that are biased towards the  
12 null, resulting in biased descriptions of underlying concentration-response relationships. These  
13 effect estimates, though conservative from a testing perspective, must be evaluated and used  
14 with caution, as they may lead to an underestimation of the overall impact of air pollution on  
15 health effects.

#### 16 **8.4.3.2.2 Confounding by Temporal Trends and Meteorologic Effects**

17 The effect of seasonal differences in the health outcomes and O<sub>3</sub> exposure levels was  
18 recognized in the 1996 O<sub>3</sub> AQCD. This issue is discussed in detail in Section 7.6.5 of this  
19 document. Two important factors, i.e., temporal trends and meteorological factors must be  
20 considered in evaluating O<sub>3</sub> health effects estimates. In the U.S. 95 communities study (Bell  
21 et al., 2004), sensitivity analyses indicated that the O<sub>3</sub> risk estimates were robust to tripling the  
22 degrees of freedom for smoothing terms used to control for temporal trends. In a case-crossover  
23 study by Schwartz (2004), the O<sub>3</sub>-mortality risk estimates from an analysis using nonlinear  
24 regression splines to control for temperature were similar to those from an analysis that matched  
25 on temperature, indicating that the effect estimates were not sensitive to methods used to control  
26 confounding by temperature.

27  
28 Analysis of O<sub>3</sub> health effects is further complicated in view of the fact that the relationship  
29 of O<sub>3</sub> with temperature and with other pollutants appears to change across seasons. As shown in  
30 Figures 8-5, 8-7 and 8-8, the O<sub>3</sub> effect estimates from warm season data were consistently larger  
31 compared to those calculated using all-year data and cool-season data. In a study of daily

1 hospital admissions (Burnett et al., 2001), season-stratified analyses appeared to effectively  
2 control confounding by season.

3 In summary, adjusting for temporal trends and meteorological factors is critical to  
4 obtaining meaningful O<sub>3</sub>-effect estimates. Analyses have found that confounding by seasonal  
5 variability is controlled effectively by stratifying the data by season. Mortality and morbidity  
6 effect estimates computed using year-round data need to be interpreted with caution.

#### 7 8 **8.4.3.2.3 Assessment of Confounding by Copollutants**

9 The presence and influence of PM and other gaseous copollutants have to be considered in  
10 assessing O<sub>3</sub>-health effects associations found by observational studies. The potential for  
11 copollutant confounding in the epidemiological time-series studies was assessed in some detail  
12 in Section 7.6.6. Multipollutant modeling is the most common method used to test for potential  
13 confounding in epidemiological studies; however, interpretation of the results is often  
14 complicated by the high degree of correlation among air pollutants. Across the various health  
15 outcomes, O<sub>3</sub> effects were generally not confounded by PM and other gaseous copollutants.  
16 The O<sub>3</sub> effects on lung function, respiratory symptoms, respiratory hospitalizations, and  
17 mortality were robust to PM-adjustment in all year and warm season only analyses.

18 The O<sub>3</sub> mortality risk estimates from two-pollutant models adjusted for PM are presented  
19 in Figure 8-7 (U.S. and Canadian studies only). In the two multicity studies analyzed here, the  
20 addition of PM<sub>10</sub> did not substantially change the risk estimates (Samet et al., 2000; Dominci  
21 et al., 2003; Schwartz, 2004). The O<sub>3</sub>-mortality effects in single-city studies also were robust  
22 after adjusting for PM<sub>10</sub> indices, both in all-year and season-stratified analyses data.

23 In summary, assessing the health effects attributable to O<sub>3</sub> is very challenging, given the  
24 high covariation among the copollutants and the limitations in the statistical methodology to  
25 assess independent effects of such correlated variables. Definitive partitioning out of the  
26 individual pollutant-specific health outcomes from among an ambient mixture of multiple  
27 components is very difficult due to the dynamic nature of their interactions over time. However,  
28 the new time-series studies that made an exhaustive survey using populations from multiple U.S.  
29 cities do provide substantial epidemiologic evidence indicating that associations for O<sub>3</sub> with  
30 mortality and morbidity are robust to confounding by copollutants.

### 8.4.3.3 Lag Period between Ozone Exposure and Health Response

The lag times between cause and effect depend on underlying biological mechanisms involved in the processes. Different lag periods are appropriate for assessing different health outcomes. As discussed in Section 7.6.4, examining longer lag periods may be needed to understand more fully the O<sub>3</sub>-related health outcomes. The most significant associations between O<sub>3</sub> concentrations and mortality and respiratory hospitalization were observed with 0-day and 1-day lags. In the 95 U.S. communities study (Bell et al., 2004) and the related U.S. study of the 19 largest cities (Huang et al., 2005), the risk estimated over multiple days (cumulative lag of 0 to 6 days) using distributed lag models indicated a strong effect of O<sub>3</sub> on mortality. It should be noted that when there is a pattern of effects across lag periods, selecting a single-day lag effect estimate may underestimate the overall effect size and not fully capture the risk distributed over adjacent days. Longer averaging periods may aid in characterizing cumulative O<sub>3</sub>-related effects over several days; however, interpreting these results may not be straightforward.

### 8.4.3.4 Concentration-Response Functions and Threshold

Ozone concentration-response relationships have been explored in several studies with various health outcomes, including mortality, hospitalizations, emergency department visits, lung function, and respiratory symptoms. While some studies found no threshold for O<sub>3</sub> health effects, others have found that a very low-level threshold may be present. A study by Kim et al. (2004) specifically examined the presence of a threshold in O<sub>3</sub>-mortality effects in Seoul, Korea by analyzing data using a log linear GAM (linear model), a cubic natural spline model (nonlinear model), and a B-mode splined model (threshold model). An estimated threshold value of 47 ppb was observed for 1-h daily max O<sub>3</sub>. This study further observed that if a threshold truly exists, the use of log-linear models may underestimate the O<sub>3</sub> effect on mortality at levels above the threshold.

It should be noted that exposure measurement error may reduce the ability to detect a threshold in O<sub>3</sub> population studies that used ambient O<sub>3</sub> concentrations as an indicator of personal ambient exposure. In addition, due to the variability in individual sensitivities, a threshold may not be seen at the population level. The limited evidence suggests that if there is a

1 threshold level in O<sub>3</sub> health effects, it is likely near the lower limit of ambient O<sub>3</sub> concentrations  
2 in the United States.

#### 3 4 **8.4.3.5 Consistency of Findings Across Epidemiologic Studies**

5 Most of the multicity and meta-analyses studies consistently found positive associations  
6 between O<sub>3</sub> and mortality. Generally consistent O<sub>3</sub> effects on hospitalizations and various  
7 respiratory health outcomes also were found. Ozone concentrations tend to be spatially variable  
8 in urban areas. The geographic variability in O<sub>3</sub> concentrations and personal exposures may  
9 contribute to the heterogeneity in observed O<sub>3</sub> health effects. The degree of influence of the  
10 geographic variability on heterogeneity in effects tend to vary by study, as study design affects  
11 different aspects of exposure (e.g., time period and duration of exposure). In addition, some of  
12 the observed heterogeneity of O<sub>3</sub> effects may be partially attributable to the use of centrally-  
13 located ambient monitors to assess exposure. There may be differences in relative personal  
14 exposures to O<sub>3</sub> by region due to varying factors, such as use of air conditioning and activity  
15 patterns, that affect the relationship between personal exposure and ambient concentrations.

16 Among the field studies, various respiratory health outcomes were examined, including  
17 PEF, other spirometric parameters, respiratory symptoms, and medication use. One field study  
18 investigated the O<sub>3</sub> effect in asthmatic children living in eight urban cities in the U.S. (Mortimer  
19 et al., 2002). In the analysis pooling data from all eight cities, O<sub>3</sub> was associated with a  
20 decrement in morning PEF for a 5-day cumulative lag period. The percent changes in PEF were  
21 quite homogenous, with values ranging from -1.08% for Washington, DC to -1.71% for  
22 St. Louis. Ozone also was associated with an increased incidence of morning symptoms in the  
23 pooled analysis (Mortimer et al., 2002).

24 More than 80% of the O<sub>3</sub>-mortality estimates from the various studies conducted in North  
25 America, South America, Europe, and Australia fell between 0.5 and 5% excess risk per 40 ppb  
26 increase in 1-h max O<sub>3</sub> using year-round data. In general, the O<sub>3</sub>-mortality estimates were  
27 greater when using summer only data compared to year-round data. Though not all statistically  
28 significant, most of the O<sub>3</sub>-mortality estimates were greater than zero, indicating a positive  
29 relationship between O<sub>3</sub> exposure and mortality. Three recent mortality meta-analyses that  
30 included both U.S. and non-U.S. studies found consistent all-year combined point estimates of  
31 1.6 to 1.8% excess risk per 40 ppb increase in 1-h max O<sub>3</sub>. The O<sub>3</sub> risk estimates from the

1 numerous hospitalization and emergency department visit studies were generally larger in  
2 magnitude and more variable from study to study compared to the mortality studies.

3 Because differences in study design, population, and data analysis may affect risk  
4 estimates, one study investigated the geographic heterogeneity of O<sub>3</sub> effects in multiple cities  
5 using standardized methods. In the pooled analysis of 95 U.S. communities using all available  
6 data, intercity heterogeneity was observed among the 95 communities, which the authors noted  
7 as plausible given the city-specific differences in pollution characteristics, the use of air  
8 conditioning, time-activity patterns, and socioeconomic factors (Bell et al., 2004).

9 Overall, the epidemiological studies indicate that there are associations between acute O<sub>3</sub>  
10 exposures and morbidity and mortality outcomes in numerous locations across the United States.  
11 In general, fairly consistent O<sub>3</sub> effect estimates were observed for the various health outcomes,  
12 including pulmonary function, symptoms, hospitalization, and mortality.

#### 13 14 **8.4.3.6 Summary and Conclusions for Epidemiology Findings**

15 Discussions presented in the previous sections evaluated the merits of the epidemiologic  
16 studies to derive judgments about potential causal relationships between O<sub>3</sub> exposures and health  
17 outcomes. These evaluations were carried out in the context of the criteria listed in Section  
18 8.2.2. Information with regard to one of the criteria, i.e., coherence and biological plausibility, is  
19 discussed in the next section, which undertakes to provide an integrated analysis of the  
20 biological evidence from human and animal toxicology studies with the epidemiologic evidence.

21 The results from the new field/panel studies evaluated in this document provide additional  
22 evidence for likely causal relationships being reflected by significant associations between  
23 acute O<sub>3</sub> exposure and decrements in lung function. Several new studies also associated acute O<sub>3</sub>  
24 exposure with increased respiratory symptoms and use of asthma medication in children and,  
25 in some cases, adults. New population based time-series studies also indicated a positive  
26 association between acute O<sub>3</sub> exposure and respiratory morbidity indexed by hospital admissions  
27 and emergency visits, especially in season-stratified data. The results from large multicity  
28 studies and several meta-analyses consistently suggest an elevated risk of mortality for acute  
29 exposure to O<sub>3</sub>. Additional analyses evaluating the potential susceptibility of individuals with  
30 preexisting cardiovascular disease is rather limited. Analysis of the data from chronic mortality  
31 and morbidity studies indicate possible associations between O<sub>3</sub> and seasonal changes in lung

1 function; but, overall, the strength of the evidence does not allow establishment of a likely causal  
2 relationship between chronic O<sub>3</sub> exposure and these health outcomes.

3 Issues regarding strengths of models used in air pollution epidemiology were carefully  
4 considered. There have been improvements in the modeling to adjust for potential confounding  
5 variables, including temporal trends, meteorological factors, and copollutants. However, more  
6 sensitivity analyses would still be useful to examine the extent of adequate adjustment for  
7 confounding by these factors. Results from multipollutant models indicate that copollutants,  
8 e.g., PM, generally do not confound the association between O<sub>3</sub> and acute health outcomes,  
9 suggesting an independent effect of O<sub>3</sub>. The limited evidence suggests that if there is a threshold  
10 level in O<sub>3</sub> health effects, it is likely near the lower limit of U.S. ambient O<sub>3</sub> concentrations.

11 In conclusion, the epidemiological evidence continues to support likely causal associations  
12 between acute ambient O<sub>3</sub> exposures and increased risk of acute respiratory morbidity and  
13 mortality, based on the assessment of strength, robustness, and consistency of results reported  
14 from numerous studies reviewed in Chapter 7. There is a lack, however, of sufficient evidence  
15 by which to convincingly establish a positive association between chronic O<sub>3</sub> exposure and  
16 increased respiratory morbidity and mortality. Additional investigations are needed to further  
17 understand the health effects resulting from long-term O<sub>3</sub> exposure.

## 18 19 20 **8.5 BIOLOGICAL PLAUSIBILITY AND COHERENCE OF EVIDENCE FOR** 21 **OZONE-RELATED HEALTH EFFECTS**

22 This section is organized to integrate epidemiologic studies with toxicologic and  
23 mechanistic information obtained from controlled human exposure studies and animal  
24 toxicology studies for the two major health endpoints, morbidity and mortality reported to be  
25 associated with either short- or long- term exposure to ambient O<sub>3</sub>. Morbidity associations have  
26 been subdivided into (a) school absenteeism, (b) emergency department visits for asthma, and  
27 (c) hospitalizations due to respiratory and cardiovascular illnesses. Mortality associations were  
28 also critically evaluated to understand risk estimates for total nonaccidental mortality and  
29 mortality in specific susceptible populations. The discussion in each subsection concisely  
30 summarizes pertinent key information and then presents the plausibility of effects being  
31 reasonably attributed to the conclusions derived for the endpoint assessed. To facilitate an easy  
32 discussion and to recapitulate various biological endpoints that have been investigated in human

1 and animal toxicology studies (see Chapters 4, 5, and 6), the first subsection addresses the  
2 plausible interpretative assessments from the salient observations from experimental studies.

3 Several criteria listed in Section 8.4.2 are used in evaluating the available scientific support  
4 for conclusions regarding potential causal relationships between O<sub>3</sub> exposure and specific types  
5 of health outcomes. In addition to those criteria addressed in the preceding discussion of  
6 epidemiological evidence, certain other critical evaluation measures must be considered to  
7 ensure that these observations are biologically relevant and consistent with experimentally  
8 demonstrated biological mechanisms of action. For this assessment, the ensuing discussion on  
9 biological plausibility and coherence considers (a) the extent to which available epidemiological  
10 evidence logically ties to a range of relevant health endpoints (from cardiopulmonary  
11 physiological changes to morbidity to mortality) and (b) whether available toxicological and  
12 biochemical evidence supports plausible causal relationships for the observed epidemiological  
13 associations.

14 The ensuing discussion on biological plausibility and coherence also considers the  
15 following criteria for integrated assessment: (a) adequateness of the statistical power of the  
16 epidemiological studies to establish evidence for associations, (b) the location (urban vs. rural),  
17 (c) seasonal pattern vs. all year O<sub>3</sub> levels, (d) socioeconomic status of the population, and (e) the  
18 lag days (used in the epidemiologic analysis), considered in relation to the time course of likely  
19 biological mechanisms potentially implicated in the process.

### 21 ***Animal-to-Human Extrapolation Issues***

22 The physiological and biochemical observations reported in Table 8-1 represent the  
23 knowledge base available from toxicological studies in humans and animals that underlie the  
24 biological alterations that govern acute O<sub>3</sub>-induced health effects. This table is generated from  
25 the experimental database (see Annexes for Chapters 5 and 6 for experimental details) that  
26 utilized exposure regimens of varied concentration and duration that are environmentally  
27 relevant. As noted in the earlier section, most of the observed acute O<sub>3</sub> effects are transient and  
28 attenuate over time. However, the time-line for resolution of many of these physiological and  
29 biological parameters in normal and human subjects with underlying cardiopulmonary diseases  
30 follow different profiles as presented in Figure 8-9. Alterations in the cellular and molecular  
31 profiles observed in human airway epithelium upon acute exposure to O<sub>3</sub> evolve over time



**Table 8-1. Acute O<sub>3</sub>-induced Physiological and Biochemical Changes in Human and Animals**

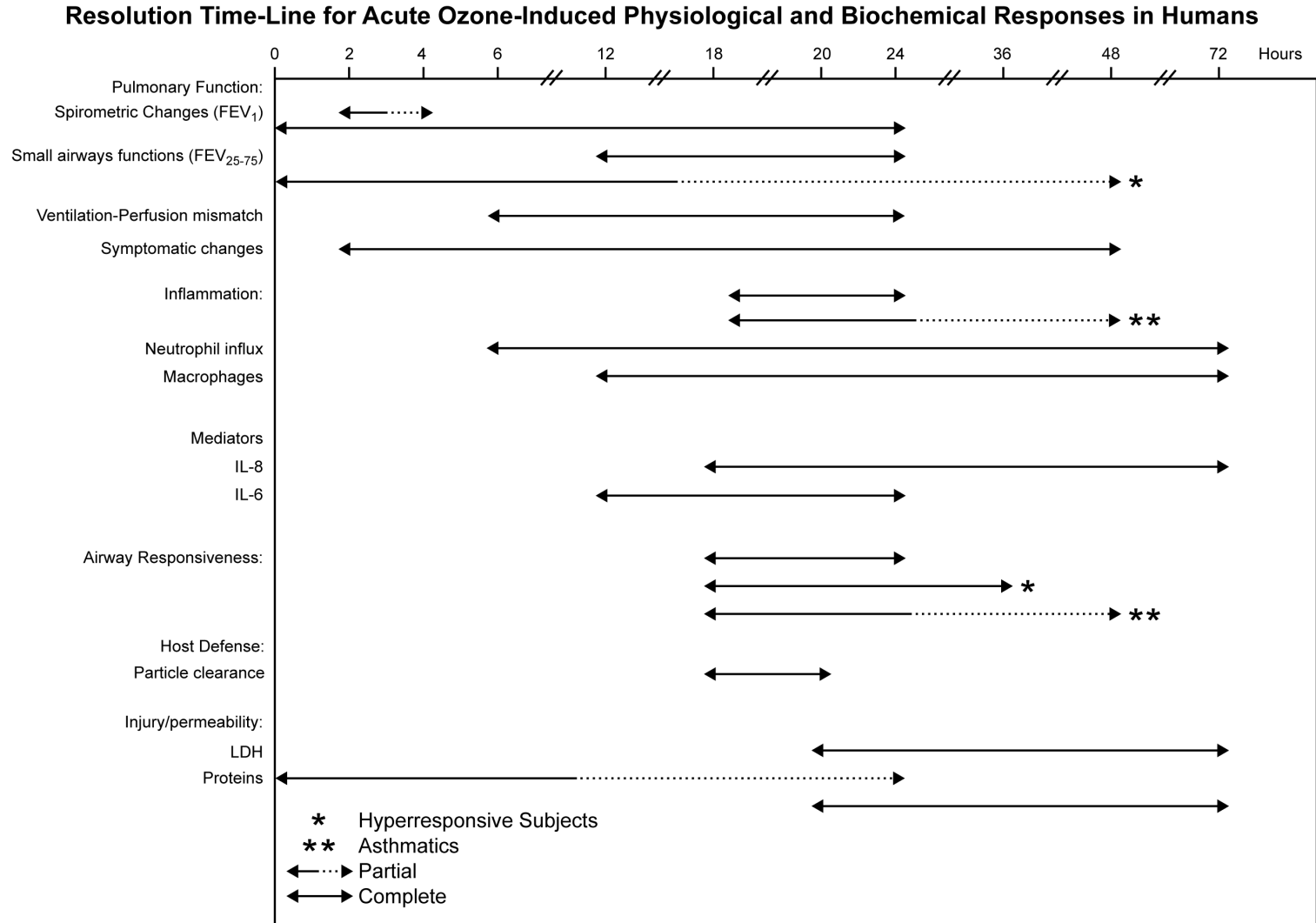
<b>Physiological/Biochemical Alterations</b>	<b>Human Exposure Studies<sup>1,2</sup></b>	<b>Animal Toxicology Studies<sup>3,4</sup></b>
Pulmonary Function:	↓ FEV <sub>1</sub> ↑ Frequency of breathing (rapid, shallow ) ↓ inspiratory capacity (cough, breathing discomfort, throat irritation, wheezing) Mild bronchoconstriction	↓ FEV <sub>1</sub> ↑ Frequency of breathing (rapid, shallow ) ↓ inspiratory capacity
Airway Responsiveness:	↑ (neuronal involvement) Change in lung resistance	↑ (vagal mediation) Change in lung resistance
Inflammation:	Yes ↑ inflammatory mediators	Yes ↑ inflammatory mediators
ROS	↑	↑
Host Defense:	↑ particle clearance ↑ permeability ↓ AM phagocytosis	↑ particle clearance ↑ permeability ↓ clearance of bacteria ↑ severity of infection ↑ mortality & morbidity
Lung injury: Morphology	Yes	Yes
Susceptibility:	Age, Inter individual variability Disease status Polymorphism in certain genes being recognized	Species specific differences Genetic basis for susceptibility indicated
Cardiovascular Changes:	Impairment in arterial O <sub>2</sub> transfer Ventilation-perfusion mismatch (suggesting potential arterial vasoconstriction)	Heart rate variability (HRV) ↓ core body temperature ↑ ANF Role for PAF indicated increased pulmonary vascular resistance

<sup>1</sup> Controlled chamber exposure studies in human volunteers were carried out for a duration of 1-6.6 h with O<sub>3</sub> concentration in the range of 0.08-0.4 ppm with intermittent exercise.

<sup>2</sup> Data on some of the biochemical parameters were obtained from in vitro studies using cells recovered from BALF.

<sup>3</sup> Responses were observed in animal toxicology studies with exposure for a duration of 2-72 h with O<sub>3</sub> concentration in the range of 0.1-2.0 ppm.

<sup>4</sup> Various species (mice, rat, guinea pigs and rabbit) and strains.

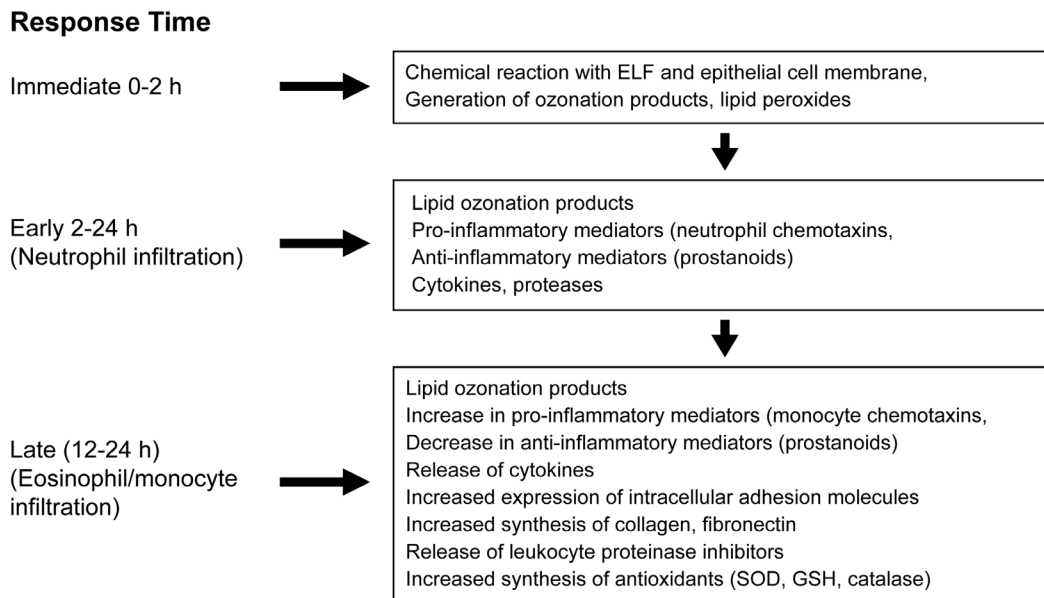


**Figure 8-9. Resolution time-line for the physiological and biochemical parameters are derived from studies reported in Chapter 6 and Chapter 6 Annex.**

1 (Figure 8-10), and the knowledge of this profile is valuable in assessing biological plausibility to  
2 integrate across evidence for various health endpoints.

3  
4

### Postulated Cellular and Molecular Changes in Human Airway Epithelial Cells on Acute Exposure to Ozone



**Figure 8-10. O<sub>3</sub>-induced cellular and molecular changes and their evolution depicted here is derived from the data reported in Leikauf et al. (1995) and Mudway and Kelly (2000).**

1 Basic similarities in physiological, biochemical, and pathological processes that exist  
2 between human and animal species are derived from the high degree of genome sequence  
3 homology that exists across species. This homology reinforces the significance of knowledge  
4 gained on the initiation, progression and treatment regimes for various disease processes across  
5 animal species. This homology is also apparent in acute O<sub>3</sub>-induced effects, especially on the  
6 respiratory tract of human and animal species as presented in Table 8-1 and Figures 8-9 and  
7 8-10. The commonality of phenomenon observed in humans and rats with regard to respiratory  
8 system effects (in terms of spirometry, ventilatory response, host defense and inflammation) and  
9 their attenuation adds strength to animal-human extrapolations. Such similarities observed at

1 higher levels of cellular organization (neutrophilic inflammation, macrophage phagocytosis  
2 processes) have increased the value and importance of animal studies in generating important  
3 data that is impossible to collect in human studies but which may corroborate both clinical and  
4 epidemiologic studies.

5 Quantitative extrapolation involves a combination of dosimetry, end point homology and  
6 species sensitivity, particularly in the case of exposure and health outcome analyses. However,  
7 extrapolation models have not been completely validated and, therefore, uncertainties do exist.  
8 Based on inflammatory markers in BALF, a 2 ppm O<sub>3</sub> exposure in nonexercising rats  
9 approximates to a 0.4 ppm exposure in exercising humans despite species-specific differences  
10 (Hatch et al., 1994). This observation lends support to the use of some of the animal toxicology  
11 data derived from relatively high O<sub>3</sub> concentration exposure regimens in understanding putative  
12 molecular changes associated with acute O<sub>3</sub> exposure in humans. Similarly, the presence of  
13 apparent O<sub>3</sub>-induced lesions in animals from chronic O<sub>3</sub> exposure studies (12 to 24 months)  
14 indicate functional defects that may potentially provide a means to facilitate more direct  
15 assessments of long-term health outcomes in humans.

### 16 17 **8.5.1 Acute Ozone Exposure-Induced Health Effects**

18 As noted in Section 8.4.2, several new epidemiologic (field/panel) studies show positive  
19 associations between short-term exposure to ambient O<sub>3</sub> and human health effects. These health  
20 effects as evaluated include school absenteeism, decline in lung function, increased use of  
21 asthma medication, and increased hospitalization, especially among individuals with asthma or  
22 certain known cardiopulmonary or cardiovascular diseases (see Chapter 7). The patterns of  
23 physiological and biochemical alterations reviewed earlier (see Figures 8-9, 8-10 and Table 8-2)  
24 tend to support certain hypotheses regarding underlying pathological mechanisms in the  
25 development of respiratory effects reported in the epidemiologic studies. Some of these  
26 mechanisms (see Table 8-2) include (a) decrements in lung function (capacities and volume), (b)  
27 bronchoconstriction, (c) increased airway responsiveness, (d) airway inflammation, (e) epithelial  
28 injury, (f) immune system activation, (g) host defense and (h) sensitivity of an individual such as  
29 age, genetic susceptibility and the extent of tolerance resulting from previous exposures. The  
30 time sequence, magnitude, and overlap of these complex events, both in terms of development

1 and recovery (see Figure 8-9 and 8-10), indicate the difficulties associated with the interpretation  
2 of biological plausibility associated with the cardiopulmonary health effects.

### 4 **Respiratory Health Effects**

5       Controlled human exposure studies have clearly demonstrated the following three types of  
6 respiratory responses to acute O<sub>3</sub> exposures: (1) irritative cough and substernal chest pain upon  
7 inspiration; (2) decrements in FVC and FEV<sub>1</sub> due to decreased inspiratory capacity rather than  
8 airways obstruction and (3) neutrophilic inflammation of the respiratory tract. Susceptibility or  
9 sensitivity to these effects were observed even among a carefully selected homogeneous study  
10 population. The sources of this heterogeneity are uncertain. As discussed in the earlier section,  
11 changes in baseline levels of various responses, the lag in the recovery phase and the role of  
12 residual defects in these mechanisms in hyperresponsive individuals suggest potential for  
13 increased health effects in cardiopulmonary compromised individuals such as people with  
14 asthma, COPD and cardiovascular diseases. Recent research has emphasized further  
15 characterization of the mechanisms and consequences of O<sub>3</sub>-induced pulmonary function and  
16 inflammatory responses. In addition, animal studies indicate morphological changes associated  
17 with acute O<sub>3</sub> exposures.

18       Ozone-induced altered breathing patterns (rapid shallow breathing) observed in controlled  
19 human exposure studies and animals occur without significantly affecting minute ventilation,  
20 suggesting compensatory changes in breathing pattern. Such a shift in breathing pattern  
21 diminishes deep lung penetration of O<sub>3</sub>. Breathing pattern is modulated by changes in peripheral  
22 mechanisms, such as direct or indirect stimulation of lung receptors and bronchial C-fibers. The  
23 activity of these afferents is integrated with input from sensory pathways and thus determines the  
24 depth and frequency of breathing. Stimulation of bronchial C-fibers along with inhibition of  
25 inspiration through local axon reflexes can induce neurogenic inflammation via tachykinins and  
26 other proinflammatory neuropeptides. Ozone-induced increases in the levels of neuropeptide  
27 substance P observed in the BALF of human subjects suggests potential neurogenic involvement  
28 in vascular permeability, plasma protein extravasation, bronchoconstriction and mucus secretion  
29 (Solway and Leff, 1991). Similar neurogenic involvement due to vagally mediated stimulation  
30 of C-fibers seen in animal toxicology studies support O<sub>3</sub>-induced bronchial hyperresponsiveness  
31 observed in humans.

1 An extensive database of animal, human, and in vitro studies supports the conclusion  
2 that O<sub>3</sub> interacts with airway epithelial cell membranes and lining fluid to form lipid ozonation  
3 products and ROS. These reactive products initiate a cascade of events leading to oxidative  
4 stress, injury, inflammation, airway epithelial damage and increased alveolar permeability to  
5 vascular fluids. Inflammation is the outcome of host response to injury and usually resolves  
6 completely. Continued irritant challenge may evolve into a chronic inflammatory state with  
7 simultaneous alterations in lung structure and function, leading to diseases such as fibrosis and  
8 emphysema. Continued inflammation can also alter the lung's ability to respond to infectious  
9 agents, allergens, and toxins. Acute inflammatory responses to O<sub>3</sub> exposure are well documented  
10 in humans and animals. As presented in Figure 8-10, the early inflammatory response  
11 to O<sub>3</sub>-induced lung injury is apparent in human subjects within 3 h postexposure. This initial  
12 neutrophilic inflammatory response phase is characterized by increases in PMNs in the BALF  
13 along with increased levels of inflammatory mediators such as interleukins, prostaglandins and  
14 complement component C3a. In vitro studies using human and animal lung cell culture systems  
15 have further examined the involvement of various inflammatory mediators and in some instances  
16 their downstream signaling pathways. The late inflammatory phase in the lung is characterized  
17 by increased levels of monocytes and eosinophils and respective mediators such as cytokines,  
18 leukotrienes, proteinases, and ROS.

19 Disruption of the lung's blood barrier by O<sub>3</sub> resulting in vascular permeability changes  
20 and plasma protein extravasation. BALF analysis on plasma influx markers such as albumin,  
21 proteins, immunoglobulins, and epithelial cell damage markers such as LDH indicate  
22 O<sub>3</sub>-induced lung epithelial injury. Ozone-induced lung injury and subsequent disruption of the  
23 airway epithelial barrier has been implicated in increased mucociliary clearance of particles  
24 observed in controlled human studies. Analogously, animal toxicology studies (see Chapter 5)  
25 have reported increased mortality to bacterial and viral infections subsequent to O<sub>3</sub> exposure and  
26 also increased clearance of particles.

27 Controlled O<sub>3</sub> exposure studies of healthy humans have indicated a large degree of  
28 intersubject variability. The spirometric and symptomatic responses are highly reproducible  
29 within the subject; but, within a group, pulmonary function measurements varied from -4% to  
30 56%. These are likely also to be genetic function, but as yet this factor remains of uncertain  
31 importance. Analysis of personal characteristics such as age, height, smoking history and

1 allergies indicated age as an important contributor. Based on FEV<sub>1</sub> measurements as criteria,  
2 young adults (18-25 yrs) were found to be more sensitive to O<sub>3</sub> than children or adults.  
3 Sensitivity to O<sub>3</sub> had been found to decline with older age (>60 yrs), but it should be noted that  
4 the baseline values for various measurable pulmonary functions are different in this group of  
5 population.

## 6 **Cardiovascular Health Effects**

8 There exist few experimental studies in animals and humans that have investigated  
9 potential cardiovascular effects with acute O<sub>3</sub> exposures. Ozone induces lung injury,  
10 inflammation, and impaired mucociliary clearance with a host of associated biochemical changes  
11 all leading to increased lung epithelial permeability. As discussed in the Section 5.4.2 the  
12 generation of lipid ozonation products and ROS in lung tissue can influence the pulmonary  
13 hemodynamics and ultimately cardiovascular system. Recent reports of interaction of O<sub>3</sub> with  
14 cholesterol in the lung surfactant and the generation of highly reactive products such as  
15 oxysterols and β-epoxide have indicated a role for cardiovascular effects and atherosclerosis  
16 (Pulfer and Murphy, 2004). Ozone-induced changes in heart rate variability, edema of heart  
17 tissue, and increased tissue and serum levels of ANF observed in animal toxicology studies lend  
18 support to potential cardiovascular effects of acute O<sub>3</sub> exposures. Such effects resulting from  
19 stimulation of airway irritant receptors, c-fiber activation, may result from either local or central  
20 nervous system involvement. The observation of O<sub>3</sub>-induced changes in the alveolar-arterial  
21 oxygen transfer in controlled human exposure studies on subjects with hypertension indicates  
22 potential complex ANF effects that need to be investigated further.

## 23 ***Coherence Between Epidemiologic and Experimental Evidence for Acute Respiratory 24 and Cardiovascular Effects***

26 Epidemiologic studies, indicate a positive association between exposure to ambient O<sub>3</sub> and  
27 declines in lung function in children and those with cardiopulmonary diseases such as asthma.  
28 Meta-analyses of children in summer camp studies (Kinney et al., 1996) and a multicity study by  
29 Mortimer et al (2002) supports the earlier observations that children and asthmatics are  
30 particularly susceptible to ambient O<sub>3</sub>. This association based on decrements in lung function  
31 and exacerbating pulmonary disease symptoms suggests that O<sub>3</sub> exposures may result in

1 increased use of medication in children and asthmatics. Studies that evaluated relationships  
2 between exposure to ambient O<sub>3</sub> and school absences in children provide corroborative positive  
3 associations. An increased incidence of disease exacerbations in people with cardiovascular  
4 diseases could not clearly be established.

5 Increased incidence of emergency department visits due to specific respiratory illness (e.g.,  
6 asthma) and hospitalization due to specific causes (e.g., respiratory or cardiovascular disease  
7 exacerbation) reported in various studies discussed in Chapter 7 (depicted in Figure 8-1 for  
8 studies from the United States and Canada) suggest a causal association supported by animal  
9 toxicology data. This association becomes more apparent when the data are analyzed for the  
10 influence of seasonal differences in ambient O<sub>3</sub> levels. Several controlled clinical studies  
11 reviewed in the 1996 O<sub>3</sub> AQCD on atopic and asthmatic subjects have not shown enhanced  
12 responsiveness to acute O<sub>3</sub> exposure compared to healthy subjects. The majority of the newer  
13 studies reviewed in Chapter 6 continue to suggest that asthmatics are as sensitive as, if not more  
14 sensitive than, normal subjects in manifesting O<sub>3</sub>-induced pulmonary function decrements.

15 Ozone-induced increases in neutrophils, protein, and IL-8 were found to be significantly  
16 higher in the BALF from asthmatics compared to healthy subjects. Similarly, subjects with  
17 allergic asthma exhibited increased airway responsiveness to inhaled allergens upon acute O<sub>3</sub>  
18 exposure. Consistent with these changes it is suggested that asthmatics will be more sensitive to  
19 small airway effects of ambient O<sub>3</sub>. Asthmatics present a differential response profile for the  
20 cellular, molecular, and biochemical parameters (Figure 8-10) in response to acute O<sub>3</sub> exposure.  
21 Increases in O<sub>3</sub>-induced nonspecific airway responsiveness incidence and duration could have  
22 important clinical implications for asthmatics.

23 Bronchial constriction following provocation with allergens presents a two-phase response.  
24 The early response is mediated by release of histamine and leukotrienes that leads to contraction  
25 of smooth muscle cells in the bronchi, narrowing the lumen and decreasing the airflow.  
26 In asthmatics, these mediators also attract accumulation eosinophils, followed by production  
27 of mucus and a late-phase bronchial constriction and reduced airflow. Holz et al (2002) reported  
28 an early phase response in subjects with rhinitis after a consecutive 4-day exposure to 0.125 ppm  
29 O<sub>3</sub> that resulted in a clinically relevant (>20%) decrease in FEV<sub>1</sub>. Allergen challenge in mild  
30 asthmatics 24 h postexposure to 0.27 ppm O<sub>3</sub> for 2 h had been found to significantly increase



1 eosinophil counts in BALF compared to healthy subjects (Vagaggini et al., 2002). Epithelial  
2 cells from the mucosal biopsies of allergic asthmatics indicated significant increase in the  
3 expression of IL-5, IL-8 and GM-CSF suggesting increased neutrophilic inflammation compared  
4 to healthy subjects (Bosson et al., 2003). In vitro exposure studies (0.1 ppm O<sub>3</sub>) of nasal  
5 epithelial cells from atopic asthma patients were found to release significantly greater amounts  
6 of neuropeptides, neurokinin A and Substance P, suggesting activation of neurogenic  
7 inflammation (Schierhorn et al., 1999). Collectively, these observations suggest that O<sub>3</sub>  
8 exposure may exacerbate pre-existing allergic asthma. People with allergic asthma may  
9 represent a segment of the population reported to have increased symptoms of respiratory illness  
10 exacerbations, emergency department visits, and hospital admissions in epidemiologic studies.

11 Recent population time-series studies (Figure 8-5) have also indicated a potential  
12 association between acute O<sub>3</sub> exposure and cardiovascular hospitalization. Additional well-  
13 designed time-series studies are needed to evaluate cardiovascular morbidity more specifically  
14 associated with acute O<sub>3</sub> exposure. The intimate hemodynamic and neurohumoral relationships,  
15 and potential cardiac consequences of pulmonary insults are well recognized. Two important  
16 observations in human clinical studies: (1) O<sub>3</sub>-induced impairment in alveolar-arterial oxygen  
17 transfer (Gong et al., 1998) and (2) O<sub>3</sub>-induced ventilation-perfusion mismatch (Foster et al.,  
18 1993, 1997) are consistent with potential cardiovascular impacts of O<sub>3</sub>. If such relationships are  
19 validated, they will aid in our understanding the role of O<sub>3</sub>-induced reductions in gas exchange  
20 and oxygen saturation in COPD patients with already compromised gas exchange process.  
21 Cardiovascular disease conditions and COPD are most common among old age groups.  
22 Thus, age-associated pulmonary function deficiencies in older people would add an additional  
23 burden with respect to O<sub>3</sub>-induced effects. The recent observations of air pollution-induced  
24 vasoconstriction in controlled human exposure studies by Brook et al. (2002) suggest a  
25 possible role for O<sub>3</sub>.

26 Animal toxicology studies indicate acute O<sub>3</sub>-induced microvascular leakage and  
27 subsequent edema of airways in guinea pigs (Inoue et al., 1997) and increased baseline values  
28 for total vascular resistance in rabbit pulmonary vessels in ex vivo studies (Delaunois et al.,  
29 1998). These observations support the possibility of potential cardiovascular effects.

## 8.5.2 Chronic O<sub>3</sub> Exposure-Induced Health Effects

The effects of chronic O<sub>3</sub> exposure in humans have been addressed primarily with cross-sectional epidemiologic studies. Due to lack of precise information on exposure, the possibility of selection bias and the difficulty of controlling for confounders, these findings are inconclusive. However, several new longitudinal epidemiological studies have evaluated the potential associations between chronic exposure to O<sub>3</sub> and morbidity and mortality (see Section 7.5). These studies suggest that long-term exposure may be related to changes in lung function, increased incidence of asthma, mortality, and, possibly, lung cancer. However, based on available evidence, no definitive relationship could be established between chronic O<sub>3</sub> exposure and these health outcomes. There are no data available from controlled human chamber studies that evaluated chronic exposure regimens.

The lack of adequate data from epidemiologic and clinical studies in human has directed attention to the results from chronic exposure studies in animals. Earlier chronic animal studies employed traditional exposure designs using chronic stable exposures. Later studies have attempted to incorporate design features that mimic diurnal and seasonal pattern of O<sub>3</sub> exposure and realistic exposure concentrations. Studies on monkeys that compared these two designs reported increased airway pathology with the latter design. Persistent and irreversible effects observed in chronic animal toxicology studies indicate the need for complementary human data from epidemiologic studies.

Animal toxicology data provide a clearer picture indicating that long-term O<sub>3</sub> exposure at levels found in the ambient air may have lasting effects. Chronic exposure studies in animals have reported biochemical and morphological changes suggestive of irreversible long-term O<sub>3</sub> impacts on the lung. Some of the studies in rats (0.5-1.0 ppm O<sub>3</sub> for 6 h/day) for 20 months and monkeys (0.61 ppm) for one year noted increased deposition of collagen and thickening of the CAR of the deep lung. Differences in this degree of lung damage have been observed with continuous exposure and seasonal pattern. A long term study of infant rhesus monkeys exposed to simulated seasonal O<sub>3</sub> (0.5 ppm 8 h/day for 5 days every 14 days for 11 episodes) resulted in remodeling in the distal airways, abnormalities in tracheal basement membrane, accumulation of eosinophils in conducting airways and decrements in airway innervation. Earlier studies in rats following seasonal episodic profiles also showed small, but significant, decrements in lung function that were consistent with focal fibrinogenesis in the proximal alveolar region. On the

1 other hand, chronic O<sub>3</sub> exposures in a range of 0.5 to 1.0 ppm induce epithelial hyperplasia that  
2 disappears in a few days, and the weight of evidence from new experimental animal studies  
3 (using non-lifetime exposures) does not support ambient O<sub>3</sub> as being a pulmonary carcinogen.

4 Collectively, the evidence from animal studies strongly suggest that O<sub>3</sub> is capable of  
5 damaging the distal airways and proximal alveoli, resulting in lung tissue remodeling leading to  
6 apparent irreversible changes. Compromised pulmonary function and structural changes due to  
7 persistent inflammation may exacerbate the progression and development of chronic lung  
8 disease.

### 10 **8.5.3 Mortality-Related Health Endpoints**

11 An extensive analysis of population time-series studies that evaluated the air pollution  
12 related mortality risk estimates presented in Section 7-4 utilized data from single and multicity  
13 studies from around the world. Mortality risk estimates derived from studies in U.S. and Canada  
14 coupled with meta-analyses (Figures 8-6 and 8-7) all indicate an elevated risk for mortality on  
15 acute O<sub>3</sub> exposure after adjustment for the influence of season and PM (Figure 8-7). Meta-  
16 analyses of large U.S. multicity studies also suggest a positive association. Mortality risk  
17 estimates derived from the studies that analyzed PM as a potential confounder suggest that the  
18 reported estimates are not attributable to confounding by PM. Several single-city studies that  
19 specifically evaluated the relationship between cardiovascular mortality and O<sub>3</sub> exposure also  
20 indicated a positive association.

21 The epidemiology results outlined above for mortality suggest a pattern of effects that may  
22 be biologically germane to interpretation of its causality, but our knowledge about potential  
23 underlying mechanisms remains very limited and suggests a need for further experimental  
24 support. The majority of the physiological and biochemical parameters evaluated both in human  
25 clinical and animal toxicology studies (Table 8-1; Figure 8-9) suggest a relatively transient  
26 nature for O<sub>3</sub>-induced biochemical perturbations. Most effects attenuate over time, depending on  
27 the preexisting pathophysiology. One can hypothesize a generic pathway of O<sub>3</sub>-induced lung  
28 damage, potentially involving oxidative lung damage with subsequent inflammation and/or  
29 decline in lung function leading to respiratory distress.

30 Recent analysis of third National Health and Nutrition Examination Followup study data  
31 indicated that about 20% of the adult population have reduced FEV<sub>1</sub> values indicative of

1 impaired lung function. The majority of these individuals have COPD, asthma or fibrotic lung  
2 disease (Mannino et al., 2003). These cardiopulmonary disease conditions are associated with  
3 persistent low-grade systemic inflammation. It has also been reported that patients with COPD  
4 are at increased risk for cardiovascular disease. Lung disease with underlying inflammation may  
5 also link to low-grade systemic inflammation associated with atherosclerosis. These effects in  
6 disease are independent of cigarette smoking (Sin et al., 2005). Lung function decrements in  
7 cardiopulmonary disease has also been associated with inflammatory markers such as C-reactive  
8 protein (CRP) in blood. In fact, at the population level, individuals with the lowest FEV<sub>1</sub> have  
9 the highest levels of CRP, while those with highest FEV<sub>1</sub> have the lowest values for CRP  
10 (Mannino et al., 2003; Sin and Man, 2003). The complex, physiological and biochemical  
11 perturbations that exist simultaneously (Figure 8-9 and 8-10) subsequent to acute exposure to O<sub>3</sub>  
12 may tilt the biological homeostasis mechanisms leading to adverse health effects in people with  
13 compromised cardiopulmonary systems. However, no experimental data are available at this  
14 time to support such a hypothesis as being operational in O<sub>3</sub>-induced cardiovascular mortality  
15 observed in the epidemiological studies. It is possible that reevaluation of some of the  
16 epidemiological panel studies that reported changes in CRP in the context of air pollution  
17 mortality evaluations focused on PM with correct adjustments for O<sub>3</sub> may shed some light on  
18 this potential relationship.

## 21 **8.6 SUSCEPTIBILITY FACTORS**

22 Many factors such as age, gender, disease, nutritional status, smoking, and genetic  
23 variability may contribute to the differential effects of environmental pollutants, including O<sub>3</sub>.  
24 Genetic factors, such as single nucleotide polymorphisms (SNPs) and developmental defects,  
25 can contribute to innate susceptibility, while acquired susceptibility may develop due to personal  
26 habits (smoking, diet, exercise) and other risk factors such as age, gender, pregnancy, and  
27 copollutants. However, the available information from animal toxicology and epidemiologic  
28 studies did not provide sufficiently clear scientific evidence by which to confidentially identify  
29 and/or associate any specific factor as contributing to adverse health effects of O<sub>3</sub> (U.S.  
30 Environmental Protection Agency, 1996a). However, advances in available research results

1 since then have improved our ability to delineate likely susceptible or vulnerable populations at  
2 increased risk for O<sub>3</sub>-induced health effects.

3 New animal toxicology studies using various strains of mice and rats have identified  
4 O<sub>3</sub>-sensitive and resistant strains and illustrated the importance of genetic background in  
5 determining O<sub>3</sub> susceptibility. Using subacute low exposure regimen (0.3 ppm O<sub>3</sub>, 48h) studies  
6 on inbred strains that have been designated as inflammation prone or resistant, Kleeberger et al.,  
7 (1997) identified the pro-inflammatory cytokine gene, *Tnf-α*, as a susceptibility gene. Further  
8 characterization of this model indicated a role for TNF receptors (TNFR1, TNFR2) in O<sub>3</sub>-  
9 induced pulmonary epithelial injury and inflammation (Cho et al., 2001). Studies on five inbred  
10 strains of mouse with differing response to O<sub>3</sub> exposure (acute high dose or low dose continuous  
11 exposure for 3 days), reported a protective role for clara cell secretory protein (CCSP) against  
12 O<sub>3</sub>-induced oxidative damage (Broeckeaert et al., 2003; Wattiez et al., 2003). The role for these  
13 genes and/or their orthologs in human susceptibility to O<sub>3</sub> exposure is yet to be examined.

14 Apart from age at the time of exposure, controlled human exposure studies have also  
15 indicated a high degree of interindividual variability in some of the pulmonary physiological  
16 parameters. Recent studies by David et al. (2003) and Romieu et al. (2004) reported a role for  
17 genetic polymorphism in antioxidant enzymes and genes involved in inflammation to modulate  
18 pulmonary function and inflammatory responses to O<sub>3</sub> exposure. Similar to mouse studies  
19 referred above, polymorphism in *Tnf-α* has been implicated in O<sub>3</sub>-induced lung function changes  
20 in healthy, mild asthmatics and individuals with rhinitis. These observations suggest a potential  
21 role for these markers in the innate susceptibility to O<sub>3</sub>, however, the validity of these markers  
22 and their relevance in the context of prediction to population studies need additional  
23 experimentation.

24 Biochemical and molecular parameters extensively evaluated in these experiments were  
25 used to identify specific loci on the chromosomes and, in some cases, to relate the differential  
26 expression of specific genes to biochemical and physiological differences observed among these  
27 species. Utilizing O<sub>3</sub>-sensitive and O<sub>3</sub>-resistant species, it has been possible to identify the  
28 involvement of AHR and inflammation processes in O<sub>3</sub> susceptibility. However, most of these  
29 studies were carried out using relatively high doses of O<sub>3</sub>, making the relevance of these studies  
30 questionable in human health effects assessment. No doubt, the molecular parameters identified  
31 in these studies may serve as useful biomarkers with the availability of suitable technologies and,

1 ultimately, can likely be integrated with epidemiological studies. Interindividual differences in  
2 O<sub>3</sub> responsiveness have been observed across a spectrum of symptoms and lung function  
3 responses do not yet allow identification of important underlying factors, except a significant  
4 role for age.

### 6 **8.6.1 Preexisting Disease as a Potential Risk Factor**

7 People with preexisting pulmonary disease may be at increased risk from O<sub>3</sub> exposure.  
8 Altered physiological, morphological and biochemical states typical of respiratory diseases like  
9 asthma, COPD and chronic bronchitis may render people sensitive to additional oxidative burden  
10 induced by O<sub>3</sub> exposure. Based on studies assessed in the 1996 O<sub>3</sub> AQCD (U.S. Environmental  
11 Protection Agency, 1996a), asthmatics appear to be at least as, or more, sensitive to the acute  
12 effects of O<sub>3</sub> as healthy nonasthmatic subjects. The new results reviewed in Chapters 6 and 7 of  
13 this document from controlled exposure and epidemiologic studies also suggest that asthmatics  
14 are a potentially sensitive subpopulation for O<sub>3</sub> health effects.

15 A number of time-series epidemiologic studies have reported increased risk in study  
16 subsets of individuals with preexisting lung diseases, among which tend to implicate asthmatics  
17 as potentially susceptible individuals. The epidemiologic studies of acute exposure to O<sub>3</sub>  
18 discussed in Section 8.4.2 indicate increased risk for exacerbation of disease symptoms during  
19 the warm season.

20 Newly available human exposure studies by Stenfors and coworkers have shown  
21 differences regarding PMN influx in BALF between asthmatics and healthy human subjects.  
22 In vitro studies (Stenfors et al., 2002) using nasal mucosal biopsies from atopic and nonatopic  
23 subjects exposed to 0.1 ppm O<sub>3</sub> found significant differences in the release of IL-4, IL-6, IL-8,  
24 and *TNF-α*. A subsequent study by the same group (Schierhorn et al., 2002) found a significant  
25 difference in the O<sub>3</sub>-induced release of the neuropeptides neurokinin A and substance P from  
26 allergic patients, compared to nonallergic controls, suggesting increased activation of sensory  
27 nerves by O<sub>3</sub> in the allergic tissues. Another report from Bayram et al. (2002) using in vitro  
28 culture of bronchial epithelial cells recovered from atopic and nonatopic asthmatics indicated the  
29 existence of a significant difference in permeability (by measuring the paracellular flux  
30 of <sup>14</sup>C-BSA). Additional controlled O<sub>3</sub> exposure studies in human subjects with intermittent  
31 asthma (Hiltermann et al., 1999), and asthmatics (Basha et al., 1994; Scannell et al., 1996)

1 reported increased secretion of IL-8 suggesting increased neutrophilic inflammation in those  
2 subjects. Two studies (Jörres et al. 1996; Holz et al. 2002) observed increased airway  
3 responsiveness to repeated daily O<sub>3</sub> exposure to bronchial allergen challenge in subjects with  
4 preexisting allergic airway disease.

5 Newly available reports from controlled human exposure studies (see Chapter 6) utilized  
6 subjects with preexisting cardiopulmonary diseases such as COPD, asthma, allergic rhinitis, and  
7 hypertension. The data generated from these studies that evaluated pulmonary function changes  
8 in spirometry did not clearly find differences between filtered air and O<sub>3</sub> exposure in COPD and  
9 asthmatic subjects. However, the data on airway responsiveness, inflammation and various  
10 molecular markers of inflammation and bronchoconstriction indicate that people with atopic  
11 asthma and allergic rhinitis are potentially susceptible groups for O<sub>3</sub>-induced adverse health  
12 effects. There was only one study with a limited number of subjects that evaluated the effects  
13 of O<sub>3</sub> exposure in hypertensive patients, and it did not find significant O<sub>3</sub>-induced changes in  
14 clinical parameters, such as heart rate, blood pressure and ECG.

15 The observation of increased pathology in long-term animal exposure studies in the  
16 absence of observable physiological changes also suggests that chronic exposure may increase  
17 susceptibility to adverse health effects, but this needs to be validated via long-term  
18 epidemiologic studies.

## 19 20 **8.6.2 Potential Public Health Impacts**

21 Exposure to ambient O<sub>3</sub> is associated with various health outcomes, including increased  
22 incidence of cough, reduction in lung function, increased inflammation, and increased hospital  
23 admissions and mortality. In protecting public health, a distinction must be made between health  
24 effects that are considered “adverse” and those that are not. What constitutes an adverse health  
25 effect varies for different population groups, with some changes in healthy individuals not being  
26 viewed as adverse but those of similar type and magnitude in other susceptible individuals being  
27 seen as adverse. Hence, the definition of adversity of health effects will be an important issue in  
28 ultimately considering and describing the rationale for decisions concerning review of the O<sub>3</sub>  
29 NAAQS.

1 **8.6.2.1 General Concepts Related to Defining of Adverse Health Effects**

2 The official statement of the American Thoracic Society (ATS) published on “What  
3 Constitutes an Adverse Health Effect of Air Pollution?” (ATS, 2000) updated guidance for  
4 defining adverse respiratory health effects published fifteen years earlier (ATS, 1985) to address  
5 new investigative approaches used to identify the effects of air pollution and to reflect the  
6 concern for the impacts of air pollution on specific susceptible groups.

7 In the 2000 update, there is an increased focus on quality of life measures as indicators of  
8 adversity and also a more specific consideration of population risk. Exposure to air pollution  
9 that increases the risk of an adverse effect to the entire population is adverse, even though it may  
10 not increase the risk of any identifiable individual to an unacceptable level. For example, a  
11 population of asthmatics could have a distribution of lung function such that no identifiable  
12 single individual has a level associated with significant impairment. Exposure to air pollution  
13 could shift the distribution to lower levels that still do not bring any identifiable individual to a  
14 level that is associated with clinically relevant effects. However, this would be considered to be  
15 adverse because individuals within the population would have diminished reserve function and,  
16 therefore, would be at increased risk if affected by another agent.

17 Reflecting new investigative approaches, the ATS statement also describes the potential  
18 usefulness of research into the genetic basis for disease, including responses to environmental  
19 agents, that will provide insights into the mechanistic basis for susceptibility, and provide  
20 markers of risk status. Likewise biomarkers, that are indicators of exposure, effect or  
21 susceptibility, may someday be useful in defining the point at which a response should be  
22 equated with an adverse effect.

23 The 1996 O<sub>3</sub> AQCD provided information useful in helping to define adverse health effects  
24 associated with ambient O<sub>3</sub> exposure by describing gradation of severity and adversity of  
25 respiratory-related effects, and those definitions are reproduced and presented here as Tables 8-2  
26 and 8-3. The severity of effects described in those tables and the approaches taken to define the  
27 adversity still appear to be valid and reasonable even in the context of the new ATS statement  
28 (ATS, 2000).



**Table 8-2. Gradation of Individual Responses to Short-Term Ozone Exposure in Healthy Persons \*\***

<b>Functional Response</b>	<b>None</b>	<b>Small</b>	<b>Moderate</b>	<b>Large</b>
FEV <sub>1</sub>	Within normal range ( $\pm 3\%$ )	Decrements of 3 to $\leq 10\%$	Decrements of $>10$ but $<20\%$	Decrements of $\geq 20\%$
Nonspecific bronchial responsiveness <sup>b</sup>	Within normal range	Increases of $<100\%$	Increases of $\leq 300\%$	Increases of $>300\%$
Duration of response	None	$<4$ hours	$>4$ hours but $\leq 24$ hours	$>24$ hours
<b>Symptomatic Response</b>	<b>Normal</b>	<b>Mild</b>	<b>Moderate</b>	<b>Severe</b>
Cough	Infrequent cough	Cough with deep breath	Frequent spontaneous cough	Persistent uncontrollable cough
Chest pain	None	Discomfort just noticeable on exercise or deep breath	Marked discomfort on exercise or deep breath	Severe discomfort on exercise or deep breath
Duration of response	None	$<4$ hours	$>4$ hours but $\leq 24$ hours	$>24$ hours
<b>Impact of Responses</b>	<b>Normal</b>	<b>Normal</b>	<b>Mild</b>	<b>Moderate</b>
Interference with normal activity	None	None	A few sensitive individuals choose to limit activity	Many sensitive individuals choose to limit activity

<sup>a</sup> See text for discussion; see Appendix A for abbreviations and acronyms.

<sup>b</sup> An increase in nonspecific bronchial responsiveness of 100% is equivalent to a 50% decrease in PD<sub>20</sub> or PD<sub>100</sub>.

\*This table is reproduced from the 1996 O<sub>3</sub> AQCD (Table 9-1, page 9-24) (U.S. Environmental Protection Agency, 1996a).

1 **8.6.2.2 Estimation of Potential Numbers of Persons in At-Risk Susceptible Population**  
 2 **Groups in the United States**

3 Although O<sub>3</sub>-related increases in individual health risks may appear to be small, they are  
 4 likely significant from an overall public health perspective due to the large number of  
 5 individuals in potential risk groups. Numerous subpopulations may be identified as having  
 6 increased susceptibility or vulnerability to adverse health effects from O<sub>3</sub>, including older adults,  
 7 children, individuals with preexisting cardiopulmonary disease, those of lower socioeconomic  
 8 status, and those with higher exposure levels. Clearly, the impact of O<sub>3</sub> on public health can be  
 9 very extensive.

**Table 8-3. Gradation of Individual Responses to Short-Term Ozone Exposure in Persons with Impaired Respiratory Systems<sup>a\*</sup>**

<b>Functional</b>	<b>None</b>	<b>Small</b>	<b>Moderate</b>	<b>Large</b>
FEV <sub>1</sub> change	Decrements of <3%	Decrements of 3 to ≤10%	Decrements of >10 but <20%	Decrements of ≥20%
Nonspecific bronchial responsiveness <sup>b</sup>	Within normal range	Increases of <100%	Increases of ≤300%	Increases of >300%
Airway resistance (SR <sub>aw</sub> )	Within normal range (±20%)	SR <sub>aw</sub> increased <100%	SR <sub>aw</sub> increased up to 200% or up to 15 cm H <sub>2</sub> O/s	SR <sub>aw</sub> increased >200% or more than 15 cm H <sub>2</sub> O/s
Duration of response	None	<4 hours	>4 hours but ≤24 hours	>24 hours

<b>Symptomatic</b>	<b>Normal</b>	<b>Mild</b>	<b>Moderate</b>	<b>Severe</b>
Wheeze	None	With otherwise normal breathing	With shortness of breath	Persistent with shortness of breath
Cough	Infrequent cough	Cough with deep breath	Frequent spontaneous cough	Persistent uncontrollable cough
Chest pain	None	Discomfort just noticeable on exercise or deep breath	Marked discomfort on exercise or deep breath	Severe discomfort on exercise or deep breath
Duration of response	None	< 4 hours	>4 hours, but ≤24 hours	>24 hours

<b>Impact of Responses</b>	<b>Normal</b>	<b>Mild</b>	<b>Moderate</b>	<b>Severe</b>
Interference with normal activity	None	Few individuals choose to limit activity	Many individuals choose to limit activity	Most individuals choose to limit activity
Medical treatment	No change	Normal medication as needed	Increased frequency of medication use or additional medication	Physician or emergency room visit

<sup>a</sup> See text for discussion; see Appendix A for abbreviations and acronyms.

<sup>b</sup> An increase in nonspecific bronchial responsiveness of 100% is equivalent to a 50% decrease in PD<sub>20</sub> or PD<sub>100</sub>.

\*This table is reproduced from the 1996 O<sub>3</sub> AQCD (Table 9-2, page 9-25) (U.S. Environmental Protection Agency, 1996a).

1 One consideration in the assessment of potential public health impacts is the size of various  
2 population groups that may be at increased risk for health effects associated with O<sub>3</sub>-related air  
3 pollution exposure. Table 8-4 summarizes information on the prevalence of chronic respiratory  
4 and circulatory conditions in the U.S. population in 2002 and 2003 (Dey and Bloom, 2005;  
5 Lethbridge-Çejku et al., 2004). Individuals with preexisting cardiopulmonary disease constitute  
6 a fairly large proportion of the population, with tens of millions of people included in each  
7 disease category. For respiratory conditions, approximately 11% of U.S. adults and 13% of  
8 children have been diagnosed with asthma, and 6% of adults have COPD (chronic bronchitis and  
9 emphysema). Table 8-5 provides further information on the number of various specific  
10 respiratory conditions per 100 persons by age among the U.S. population during the mid-1990s.  
11 Approximately 23 million people, or 11% of the U.S. adult population, have some type of heart  
12 disease, with 6% reporting diagnoses of coronary heart disease. Approximately 21% of the U.S.  
13 adult population has hypertension. Cardiovascular conditions are more common among older  
14 age groups, while asthma prevalence is higher in children.

15 In addition, subpopulations based on age group or socioeconomic status also comprise  
16 substantial segments of the population that may be potentially at risk for O<sub>3</sub>-related health  
17 impacts. Based on U.S. census data from 2003, about 26% of the U.S. population are under  
18 18 years of age and 12% are 65 years of age or older. Approximately 12% of the U.S.  
19 population (including 18% of children) are below the poverty level and 16% do not have health  
20 insurance coverage. Hence, large proportions of the U.S. population are included in groups that  
21 are considered likely to have increased susceptibility and vulnerability for health effects from  
22 ambient O<sub>3</sub> exposure.

23 The health statistics data illustrate what is known as the “pyramid” of effects. At the top of  
24 the pyramid, there are approximately 2.5 millions deaths from all causes per year in the U.S.  
25 population, with about 900,000 deaths due to circulatory diseases and 100,000 deaths from  
26 chronic lower respiratory diseases (Kochanek et al., 2004). For circulatory disease morbidity,  
27 there are approximately 6 million hospital discharges per year (DeFrances et al., 2005),  
28 4.5 million emergency department visits (McCaig and Burt, 2005), and 80 million ambulatory  
29 care visits (Woodwell and Cherry, 2004). For respiratory health diseases, there are nearly  
30 4 million hospital discharges per year (DeFrances et al., 2005), 14 million emergency  
31 department visits (McCaig and Burt, 2005), 112 million ambulatory care visits (Woodwell and

**Table 8-4. Prevalence of Selected Cardiorespiratory Disorders by Age Group and by Geographic Region in the United States (2002 [U.S. Adults] and 2003 [U.S. Children] National Health Interview Survey)**

Chronic Condition/Disease	Age (Years)						Region			
	Adults (18+ Years)		18-44	45-64	65-74	75+	Northeast	Midwest	South	West
	Cases ( $\times 10^6$ )	%	%	%	%	%	%	%	%	%
<b>Respiratory Conditions</b>										
Asthma	21.9	10.6	11.5	10.6	8.4	7.6	11	10.9	9.8	11.8
COPD										
Chronic Bronchitis	9.1	4.4	3.5	5.5	5.5	5.3	3.8	4	5.4	3.8
Emphysema	3.1	1.5	0.3	2	4.9	4.7	1.5	1.8	1.7	1.1
<b>Circulatory Conditions</b>										
All Heart Disease	22.7	11.2	4	12.7	26.3	36.6	10.5	11.7	11.6	10.7
Coronary Heart Disease	12.5	6.2	0.9	7.1	18.7	24.5	5.7	6.2	6.8	5.8
Hypertension	43.3	21.2	7.4	29	49.6	51.8	19.7	21.1	23.3	18.9
Stroke	4.8	2.4	0.4	2.5	6.4	11.1	2.4	2.3	2.4	2.7
Chronic Condition/Disease	Age (Years)					Region				
	Children ( $<18$ years)		0-4	5-11	12-17	Northeast	Midwest	South	West	
	Cases ( $\times 10^6$ )	%	%	%	%	%	%	%	%	
<b>Respiratory Conditions</b>										
Asthma	9.1	12.5	7.5	14	14.7	14	13.5	11.8	11.2	

Source: Lethbridge-Çejku et al. (2004) for data on adults (18+ years); Dey and Bloom (2005) for data on children (<18 years).

**Table 8-5. Acute Respiratory Conditions per 100 Persons/Year by Age Group in the United States (1996 National Health Interview Survey)**

Type of Acute Condition	All Ages	Under 5 Years	5-17 Years	18-24 Years	25-44 Years	45+ Years		
						Total	45-64 Years	65+ Years
Respiratory Conditions	78.9	129.4	101.5	86	76.9	53.3	55.9	49
Common Cold	23.6	48.6	33.8	23.8	18.7	16.1	16.4	15.7
Other Acute Upper Respiratory Infections	11.3	13.1	15	16.1	11.6	7	7.5	6.1
Influenza	36	53.7	44.3	40.5	38.1	23.3	26.1	18.6
Acute Bronchitis	4.6	7.2 <sup>a</sup>	4.3	3.9 <sup>a</sup>	5.1	3.8	3.5	4.4 <sup>a</sup>
Pneumonia	1.8	3.9 <sup>a</sup>	1.7 <sup>a</sup>	1.4 <sup>a</sup>	1.3 <sup>a</sup>	2.0 <sup>a</sup>	0.9 <sup>a</sup>	3.8 <sup>a</sup>
Other Respiratory Conditions	1.7	2.9 <sup>a</sup>	2.4 <sup>a</sup>	0.4 <sup>a</sup>	2.0 <sup>a</sup>	1.1 <sup>a</sup>	1.5 <sup>a</sup>	0.5 <sup>a</sup>

<sup>a</sup>Figure does not meet standard of reliability or precision.

Source: Adams et al. (1999).

1 Cherry, 2004), and an estimated 700 million restricted activity days per year due to respiratory  
 2 conditions (Adams et al., 1999). Combining small risk estimates with relatively large baseline  
 3 levels of health outcomes can result in quite large public health impacts. Thus, even a small  
 4 percentage reduction in O<sub>3</sub> health impacts on cardiopulmonary diseases would reflect a large  
 5 number of avoided cases.

6 Another key input for public health impact assessment is the range of concentration-  
 7 response functions for various health outcomes. Epidemiologic studies have reported  
 8 associations between short-term exposure to O<sub>3</sub> with mortality, hospitalizations for  
 9 cardiopulmonary diseases, reduced lung function, incidence of respiratory symptoms, and  
 10 changes in heartbeat rhythm and rate. Effect estimates for morbidity responses to short-term  
 11 changes in O<sub>3</sub> tend to be larger in magnitude than those for mortality.

12 A limited number of studies assessed the impact of reductions in air pollution levels on  
 13 health outcomes. A study by Neidell (2004) examined the relationship between air pollutants  
 14 and asthma hospitalizations in California. The most recent EPA O<sub>3</sub> report (U.S. Environmental  
 15 Protection Agency, 2004b) indicated that the fourth highest daily 8-h max O<sub>3</sub> levels in the pacific

1 southwest region had decreased by 16% from 1990 to 2003. This downward trend in O<sub>3</sub> levels  
2 was mostly influenced by the improvements in Los Angeles and other southern California  
3 metropolitan areas. Results from this study noted declines in levels of air pollutants since 1992  
4 and decreased asthma admissions in 1998 for children aged 1 to 18 years ranging from 5 to 14%,  
5 depending on the age group. The greatest decline (>10%) in air pollution-related asthma  
6 admissions was observed among 3- to 12-year old children. Although this benefit analysis was  
7 not specific to O<sub>3</sub>, it provides evidence of decreased morbidity resulting from reduced air  
8 pollutant concentrations, including O<sub>3</sub>.

9 An intervention study in Atlanta, GA, during the 1996 Summer Olympic Games examined  
10 the impact of a citywide decrease in automobile traffic on air quality and childhood asthma  
11 (Friedman et al., 2001). Citywide acute care visits and hospitalizations for asthma during the  
12 17 days of the Olympic Games were compared to a baseline period consisting of the four weeks  
13 before and after the Olympic Games. During the Olympic Games, levels for all pollutants  
14 generally declined, but the most dramatic change was observed for O<sub>3</sub>. The 1-h max O<sub>3</sub>  
15 concentration in Atlanta decreased 27.9% from a mean of 81.3 ppb during the baseline period to  
16 58.6 ppb during the Olympic Games. The number of asthma acute care events also decreased by  
17 41.6% in the Georgia Medicaid claims file, compared to a 3.1% decline in nonasthma acute care  
18 events. Combining data from the baseline and intervention periods, a 31% (95% CI: 0.8, 69.9)  
19 excess risk of asthma events was observed per 40 ppb increase in 1-h max O<sub>3</sub> at a cumulative lag  
20 of 0- to 2-days. Although a 16.1% decrease in PM<sub>10</sub> concentrations also occurred during the  
21 Olympic Games, there was no association between PM<sub>10</sub> and asthma acute care events.

22 Many studies have examined O<sub>3</sub>-related health effects, yet only a few have addressed the  
23 question as to the extent to which reductions in ambient O<sub>3</sub> actually lead to reductions in adverse  
24 health outcomes attributable to O<sub>3</sub>. While the findings from these studies suggest that decreases  
25 in ambient O<sub>3</sub> levels will likely lead to a reduction in asthma-related hospital admissions and  
26 emergency visits, more studies are needed to diminish uncertainty regarding this issue of  
27 accountability.

28 In addition to attribution of risks for the various health outcomes to O<sub>3</sub> and other  
29 copollutants, important considerations in assessing the impact of O<sub>3</sub> on public health include the  
30 size of the population at risk as well as the concentration-response relationship and the potential

1 identification of threshold levels. Taken together, it can be concluded that exposure to ambient  
2 O<sub>3</sub> likely has a significant impact on public health in the United States.

## 3 4 5 **8.7 SUMMARY AND CONCLUSIONS FOR OZONE HEALTH EFFECTS**

6 This section summarizes the main conclusions derived from this integrated synthesis of  
7 information regarding health effects associated with ambient O<sub>3</sub> exposures. The conclusions  
8 derived are based on an integrated analysis of animal, human clinical toxicological and  
9 epidemiological studies that have evaluated health effects associated with short-term, repeated,  
10 and long-term exposures to O<sub>3</sub> alone or in combination with other ambient pollutants.  
11 Experimental evidence from human and animal toxicological studies presented in Chapters 4,  
12 5, and 6 was utilized to provide biological plausibility for the health effects observed in  
13 epidemiologic studies. These empirical efforts are also aimed at identifying susceptible  
14 populations that are at potentially greater risk for effects of O<sub>3</sub> exposure.

### 15 16 ***1. Health effects of acute (short-term) exposures to Ozone***

17 Numerous field panel and time-series epidemiologic studies (using better weather models  
18 and adjustments to confounding copollutants than compared to those assessed in the 1996 O<sub>3</sub>  
19 AQCD ) have evaluated the effects of short-term exposure to O<sub>3</sub> on a wide range of health  
20 endpoints, from lung function decrements to mortality. Results from the majority of studies  
21 continue to support the conclusions reported in the 1996 O<sub>3</sub> AQCD.

- 22  
23 ● Panel studies typically have evaluated the effects of short-term O<sub>3</sub> exposure on both healthy  
24 individuals and people with cardiopulmonary diseases. These evaluations included  
25 measurement of lung function changes, respiratory symptoms and use of asthma medication.  
26
- 27 ● Clinical controlled exposure studies in humans indicate changes in lung function and  
28 respiratory symptoms that vary as a function of exposure concentration, duration and level  
29 of exercise.  
30

- 1 ● Newer meta-analyses confirmed the interindividual differences in lung function decrements  
2 reported in the 1996 O<sub>3</sub> AQCD. Age-specific differences in the lung function responses were  
3 also observed. Spirometric responses (due to decrements in lung function) in healthy adults  
4 exposed to near ambient O<sub>3</sub> levels typically resolve to near baseline values within 4-6 h.  
5
- 6 ● Meta-analyses of four controlled human exposure studies (two new and two reported in the  
7 1996 O<sub>3</sub> AQCD) reporting the effects of prolonged (6.6 h) exposures to 0.08 ppm O<sub>3</sub> during  
8 moderate exercise on pulmonary function in young healthy adults (M = 90, F = 30; mean  
9 age, 23 yrs) indicate an absolute FEV<sub>1</sub> decrease of 6%, whereas FEV<sub>1</sub> increased by 1%  
10 following free air (FA) exposures.  
11
- 12 ● Inflammatory responses (PMN, inflammation mediators such as cytokines and chemokines)  
13 and permeability changes (proteins, albumin), typically measured in BALF, also exhibit  
14 intersubject variability. Recent meta-analyses on numerous clinical studies indicate  
15 interindividual differences in response to short-term O<sub>3</sub> exposures.  
16
- 17 ● Inflammatory and permeability responses also resolve (in some instances complete recovery)  
18 and exhibit differential attenuation profiles between normal healthy subjects and people with  
19 preexisting respiratory diseases. Some lung inflammation markers may not resolve readily  
20 and mild persistent inflammation has been reported.  
21
- 22 ● Field/panel studies of healthy individuals and asthmatics have revealed a positive association  
23 between short-term exposure to O<sub>3</sub> and decrements in lung function.  
24
- 25 ● An association between on short-term O<sub>3</sub> exposures and school absenteeism (due to  
26 respiratory illness) has also been suggested.  
27
- 28 ● With regard to cardiac impacts, a limited number of field studies that examined the  
29 relationship between short-term O<sub>3</sub> exposures and cardiovascular effects (heart rate  
30 variability, myocardial infarction) suggest an association.  
31



- 1 ● A large multicity and several single-city studies have indicated a positive association  
2 between increased O<sub>3</sub> levels (especially during the warm season) and increased risk for  
3 hospital admissions. On the other hand epidemiologic data on emergency department visits  
4 do not suggest such an association with increase in ambient O<sub>3</sub> levels.  
5
- 6 ● Data from two large multicity studies from the U.S. and several single-city studies suggest a  
7 positive association between increase in O<sub>3</sub> levels and all cause (non-accidental) daily  
8 mortality. Meta-analyses on the influence of season suggest a causal association. Additional  
9 meta-analyses on cause-specific mortality are suggestive of a likely positive association  
10 between increases in ambient O<sub>3</sub> levels and cardiovascular mortality.  
11
- 12 ● Short-term O<sub>3</sub>-induced lung function decrements, respiratory symptoms, inflammation and  
13 permeability changes observed in animal toxicology studies are consistent with human  
14 studies.  
15

## 16 ***2. Health effects of repeated short-term exposures to Ozone***

17 The results of new controlled human exposure studies of repeated short-term O<sub>3</sub> exposures  
18 continue to support the health effects findings/conclusions reported in the 1996 O<sub>3</sub> AQCD.  
19

- 20 ● Repeated exposure studies at higher concentrations typically show that FEV<sub>1</sub> response to O<sub>3</sub>  
21 is enhanced on the second of several days of exposure. Such an enhanced response was not  
22 observed at lower O<sub>3</sub> concentrations. With repeated O<sub>3</sub> exposures over several days,  
23 spirometric and symptom responses become attenuated, but this tolerance is lost after about a  
24 week without exposure.  
25
- 26 ● In humans repeatedly exposed to 0.4 ppm O<sub>3</sub> for 5 consecutive days, several indicators of  
27 inflammation (e.g., PMN influx, IL-6, PGE<sub>2</sub>, BAL protein, fibronectin) were attenuated after  
28 5 days of exposure. Lung injury and permeability markers (LDH, IL-8, total protein,  
29 epithelial cells) did not show attenuation, indicating that tissue damage probably continues to  
30 occur during repeated exposure. The recovery of the inflammatory response occurred for  
31 some markers after 10 days, but some responses were not normalized even after 20 days.

1 The continued presence of cellular injury markers indicates a persistent effect that may not  
2 necessarily be recognized due to the attenuation of spirometric and symptom responses.

- 3
- 4 ● Repeated daily exposure to lower concentrations of O<sub>3</sub> (0.125 ppm for 4 days) causes an  
5 increased response to bronchial allergen challenge in subjects with preexisting allergic  
6 airway disease, with or without asthma. In these subjects, changes in airway responsiveness  
7 after O<sub>3</sub> exposure appear to be resolved more slowly than changes in FEV<sub>1</sub> or respiratory  
8 symptoms.

9

### 10 **3. Health effects of long-term exposures to Ozone**

11 Assessment of human health effects associated with long-term O<sub>3</sub> exposures is hampered  
12 by the lack of pertinent data from human clinical and epidemiologic studies. Chronic animal  
13 toxicology studies continue to support structural alterations in several regions of the respiratory  
14 tract and identify the CAR as the most affected region.

- 15
- 16 ● Animal toxicology studies that utilized exposure regimens to simulate seasonal exposure  
17 pattern also report increased lung injury compared to conventional chronic stable exposures.  
18 One long term study of infant rhesus monkeys exposed to simulated seasonal O<sub>3</sub> patterns  
19 (0.5 ppm 8h/day for 5 days, every 14 days for 11 episodes) demonstrated: (1) remodeling in  
20 the distal airways; (2) abnormalities in tracheal basement membrane; (3) eosinophil  
21 accumulation in conducting airways; (4) decrements in airway innervation. These findings  
22 advance earlier information regarding possible injury-repair processes occurring with  
23 seasonal O<sub>3</sub> exposures.

- 24
- 25 ● Effects of O<sub>3</sub> on the upper respiratory tract of F344 rats exposed to O<sub>3</sub> (0.12, 0.5, or 1.0 ppm  
26 for 20 months) included marked mucous cell metaplasia in the rats exposed to 0.5 and  
27 1.0 pm O<sub>3</sub>, but not at 0.12 ppm O<sub>3</sub>. The persistent nature of the O<sub>3</sub>-induced mucous cell  
28 metaplasia suggests that O<sub>3</sub> exposure may have the potential to induce similar long-lasting  
29 alterations in the airways of humans. Hyperplasia in the nasal epithelium of rats exposed to  
30 0.25 and 0.5 ppm, 8h/day, 7 days/week, for 13 weeks has been reported.

- Pathophysiological changes associated with chronic O<sub>3</sub> exposures observed in animal studies suggest possible similar alterations in humans. The pulmonary function changes observed in children in polluted metropolitan areas and lung structural alterations reported in autopsy study in Los Angeles suggest a role for long-term ambient O<sub>3</sub> exposure and need to be critically evaluated with proper study design.

#### ***4. Susceptibility factors associated with exposure to ozone***

Various factors such as age, gender, nutrition, socioeconomic, activity patterns, and disease status have been shown to influence the response to environmental air pollutants. Controlled human exposure studies clearly established differential biological response to O<sub>3</sub> based on physical activity (exertion) and age. These studies also demonstrated a large variation in sensitivity and responsiveness to O<sub>3</sub>. The specific factors that contribute to this intersubject variability are yet to be identified.

- Increased hospital admissions for asthma and COPD in summer (with increased levels of ambient O<sub>3</sub>) suggest that people with these respiratory diseases as potential sub-population for O<sub>3</sub>-induced health effects.
- Similarly, based on O<sub>3</sub>-induced differential responses in lung inflammation and in airway hyperresponsiveness, asthmatics (including children) appear to have potentially increased susceptibility to O<sub>3</sub>. However, there is no supportive data from controlled human studies suggesting individuals with COPD are more sensitive to O<sub>3</sub>-induced health effects.
- Animal toxicology studies provided supportive evidence to the observations of varied susceptibility. Various strains of mice and rats have demonstrated the importance of genetic background in O<sub>3</sub> susceptibility. Moreover, genetic and molecular characterization studies in laboratory animals identified genetic loci responsible for both sensitivity and resistance.
- Consistent with the 1996 O<sub>3</sub> AQCD, the scarcity of data prevents determination of the role of ethnic or racial background and nutrition status on O<sub>3</sub>-induced health effects. However, as presented in this document, exercising (moderate to high physical exertion) healthy,

1 adolescents, and asthmatics appear to demonstrate increased responsiveness to ambient  
2 concentrations of O<sub>3</sub> and may be susceptible for O<sub>3</sub>-induced health effects.

3  
4 **5. *Health effects of binary pollutant mixtures containing ozone***

5 A limited number of controlled human exposure studies and few animal toxicology studies  
6 with the binary mixtures containing O<sub>3</sub> suggest potential interactions depending on the exposure  
7 regimens and copollutant constituents.

- 8
- 9 ● Continuous exposure to SO<sub>2</sub> and NO<sub>2</sub> increased inhaled bolus O<sub>3</sub> absorption, while  
10 continuous exposure to O<sub>3</sub> decreased O<sub>3</sub> bolus absorption. Asthmatics exhibited enhanced  
11 airway reactivity to house dust mite following exposures to O<sub>3</sub>, NO<sub>2</sub>, and the combination of  
12 the two gases. Spirometric response, however, was impaired only by O<sub>3</sub> and O<sub>3</sub>+NO<sub>2</sub> at  
13 higher concentrations.
  - 14
  - 15 ● Animal toxicology studies with O<sub>3</sub> in mixture with NO<sub>2</sub>, formaldehyde, and PM  
16 demonstrated additive, synergistic or antagonistic effects depending on the exposure regimen  
17 and the endpoints evaluated.
  - 18
  - 19 ● One controlled exposure study of children, designed to approximate exposure conditions of  
20 an epidemiologic study by matching the population and exposure atmosphere (0.1 ppm O<sub>3</sub>,  
21 0.1 ppm SO<sub>2</sub> and 101 μg<sup>m<sup>3</sup></sup> H<sub>2</sub>SO<sub>4</sub>), failed to support the findings of the epidemiologic study.  
22 This study points out the difficulties in attempting to link the outcomes of epidemiologic and  
23 controlled studies with binary pollutant mixtures.
  - 24

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# 9. ENVIRONMENTAL EFFECTS: OZONE EFFECTS ON VEGETATION AND ECOSYSTEMS

## 9.1 INTRODUCTION

A number of ozone (O<sub>3</sub>) effects studies were published between 1996 and 2004, and they are reviewed in this document in the context of the previous O<sub>3</sub> air quality criteria documents (AQCDs) (U.S. Environmental Protection Agency, 1978, 1986, 1992, 1996). Data published since 1996 continue to support the conclusions of previous O<sub>3</sub> AQCDs that there is strong evidence that ambient O<sub>3</sub> concentrations cause foliar injury along with growth and yield damage to numerous common and economically valuable plant and tree species. Research to date has continued to be focused at the species level with very few studies at the ecosystem level. The lack of quantification of biotic and abiotic factors impinging on the individual to population organizational levels results in a limited ability to scale O<sub>3</sub> responses to the ecosystem level. Therefore, a high degree of uncertainty remains in our ability to assess ozone risk to ecological resources and the services they provide.

In general, there has been a shift away from chamber studies in favor of more field-based approaches, although chamber exposures still dominate the effects literature. Field-based approaches include surveys of visible injury, as well as physiological and growth studies using the non-chambered free-air CO<sub>2</sub> exposure (FACE) systems. The FACE systems have substantiated earlier growth and yield results for crop and tree species obtained in open-top chamber (OTC) systems. Increased emphasis has also been placed on quantifying aspects of ozone uptake to better link ambient exposure monitoring with plant/tree response. Much of the progress in quantifying uptake has occurred in Europe in the development of their ozone air quality management tool, the “critical level”. The research has developed exposure-response functions for several crops and tree seedlings using OTC studies, as well as in developing and testing models that simulate uptake. Evaluation of this new information has added to our knowledge and provides new research directions, but has not fundamentally altered the conclusions of 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996).

It is well known that O<sub>3</sub> is phytotoxic and that toxicity occurs only if O<sub>3</sub> or its reaction products reach the target tissues in the plant cell. Recent studies have provided an increased

1 understanding of how ozone interacts with the plant at the cellular level. This increased  
2 understanding of cellular-level O<sub>3</sub> effects have translated into better models, more detailed  
3 schema of how O<sub>3</sub> alters much of the basic metabolism of plants and how to construct an index  
4 that more aptly captures the species, climate, and site factors that alter uptake. These results  
5 have and will continue to lead to a better quantification of exposure and effect. However, the  
6 translation of these mechanisms into how O<sub>3</sub> is involved in altered cell metabolism and  
7 subsequent reductions in whole-plant productivity and ecosystem-level responses remain to be  
8 more been fully resolved.

9 The ensuing sections of this chapter (Section 9.2 to 9.8) are not intended to provide a  
10 complete review of the environmental effects of O<sub>3</sub>, but rather an assessment of key information  
11 published since the 1996 O<sub>3</sub> AQCD. More detailed discussion of the research since 1996 is  
12 provided in Chapter 9 Annex Sections AX9.1 to AX9.7 (in Volume 3 of this document). The  
13 framework for Chapter 9 follows the environmental effects chapter of the 1996 O<sub>3</sub> AQCD. First,  
14 an overview of various methodologies that have been, and continue to be, central to the  
15 quantification of O<sub>3</sub> effects on vegetation is provided in Section 9.2 below (see Section AX9.1  
16 for more detailed discussion). The adequacy of each methodology is discussed in the context of  
17 developing statistically robust data appropriate for assessing the risk of O<sub>3</sub> to vegetation  
18 resources. In Section 9.3, research is then reviewed from the molecular to the biochemical and  
19 physiological levels in plants that are impacted, which offers insight into the mode of action of  
20 O<sub>3</sub> (see also AX9.2). The manner in which plants respond to O<sub>3</sub>, as influenced by the numerous  
21 biotic and abiotic factors present in the environment, is next discussed (see also AX9.3).  
22 Quantifying these various modifiers is critical to being able to scale the response of individual  
23 plants to the community level and across varied landscapes and climates and is needed for  
24 regional to national assessments of risk. The development of indices of exposure or O<sub>3</sub> uptake is  
25 discussed in the context of their adequacy to realistically describe the ambient concentration-  
26 response relationships (see also AX9.4). The exposure-response relationships for a large number  
27 of crop species and cultivars, native vegetation, and tree species are reviewed, tabulated, and  
28 compared to form the basis for an assessment of the potential risk of current levels of O<sub>3</sub> on  
29 vegetation resources (see also AX9.5). Available research by which to assess the impact of O<sub>3</sub>  
30 on ecosystems is also reviewed, along with the potential data available for estimating the loss of

1 various ecosystem services (see also AX9.6). Finally, available research on the economic  
2 impact of ozone effects on vegetation resources is briefly discussed (see also AX9.7).

## 3 4 5 **9.2 METHODOLOGIES USED IN VEGETATION RESEARCH**

6 New methodological advancements since 1996 have not fundamentally altered our  
7 understanding of O<sub>3</sub> effects on plants or ecosystems. Most of the new information confirms  
8 earlier conclusions and provides additional support for OTC use in assessing sensitive species  
9 and developing exposure-response relationships. A more in-depth discussion of this topic can be  
10 found in Annex Section AX9.1.

11 The majority of ozone effects studies are fumigation studies conducted in controlled  
12 chambers, as noted in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996). That  
13 document noted that OTCs represented the best technology for determining statistically robust  
14 exposure response models of O<sub>3</sub> and crop yield and plant biomass at that time. While OTCs are  
15 still the best method for conducting controlled exposures of varying length and frequency for  
16 developing exposure-response relationships, several new approaches have been applied to O<sub>3</sub>  
17 effects research — most notably free-air exposure or “plume” systems. Free-air exposure  
18 systems (FACE) eliminate many of the concerns raised about closed or open-top chamber  
19 experiments including small plot size, altered microclimate within OTCs, and the effect of  
20 charcoal filtering on overall air quality within OTCs. FACE systems have increased our  
21 understanding in some areas; and results from FACE studies have, on the whole, confirmed what  
22 was already understood or hypothesized about how plants and plant assemblages respond to O<sub>3</sub>.  
23 Some shortcomings of using plume systems in O<sub>3</sub> research have also been identified, namely the  
24 relatively poor control of exposure levels, the presence of “hotspots” and the inability to  
25 decrease O<sub>3</sub> concentrations to below ambient levels when ambient concentrations are phytotoxic.  
26 Nonetheless, the application of FACE systems and other open-air systems to ozone exposure  
27 research have greatly helped our scaling efforts and are, perhaps, the best approach for studying  
28 the response of plant species mixtures to O<sub>3</sub> (Nussbaum and Fuhrer, 2000).

29 One of the advantages of the application of plume systems to O<sub>3</sub> research is the ability to  
30 compare response of plants in open-field systems with results from OTCs. In particular, studies  
31 with quaking aspen (*Populus tremuloides* L.) performed in OTCs, FACE, and also sites along an

1 ambient ozone gradient showed that ozone symptom expression was generally similar across  
2 these methodologies, supporting the previously observed level of variation among aspen clones  
3 in OTC studies (Isebrands et al., 2000, 2001; Karnosky et al., 1999). While this perhaps  
4 represents the first time direct comparisons were made, it supports the use of OTC data in the  
5 development of O<sub>3</sub> response functions for individual species. Concerns raised earlier about  
6 microclimate differences in chambers and the role of “chamber effects” (Fuhrer, 1994; Manning  
7 and Krupa, 1992) still persist; however, most evidence suggests that chamber effects result in  
8 altered O<sub>3</sub> uptake without altering the fundamental response of plants to ozone, thus reducing  
9 uncertainty in the use of data from OTCs. Extrapolation of the results from chamber studies  
10 depends on fully characterizing temperature, light, turbulence, and other chamber characteristics  
11 during exposures (Nussbaum and Fuhrer, 2000), but study design is equally important.  
12 Conducting studies with a large number of plant species across regions of the country where  
13 those species are indigenous is important in capturing regional climatic differences to reduce the  
14 uncertainty associated with extrapolating composited response functions across regions and to  
15 identify relative risk to vegetation in relation to given O<sub>3</sub> exposure values (U.S. Environmental  
16 Protection Agency, 1996).

17 The lack of rural monitors continues to be a major problem in the characterization of O<sub>3</sub>  
18 exposures in remote areas, as well as in linking effects to exposure in natural ecosystems. Since  
19 the 1996 O<sub>3</sub> AQCD, the use of passive samplers has expanded monitoring efforts to include  
20 remote areas that were previously uncharacterized. This has greatly enhanced our ability to link  
21 ozone symptomology with elevated O<sub>3</sub> exposure in such remote areas. However, passive  
22 samplers do not capture the temporal dynamics of exposure. Therefore, passive samplers cannot  
23 substitute for active monitors when attempting to link exposure dynamics to plant response or  
24 when developing exposure/dose-response relationships of much value as inputs for the standard  
25 setting process. To overcome this problem, Krupa et al. (2001, 2003) used models and data from  
26 a collocated O<sub>3</sub> monitor to estimate the underlying frequency distribution of hourly O<sub>3</sub>  
27 concentrations from passive samplers. Future development of passive monitor technology and  
28 data synthesis techniques holds promise, particularly since it is unlikely that extensive O<sub>3</sub>  
29 monitoring networks will be established in rural areas in the near future.

30 Exclusion methods that employ protective chemicals such as ethylenediurea (EDU) are the  
31 least disruptive of ambient culture conditions in the field, as noted in the 1996 O<sub>3</sub> AQCD.



1 However, the level of protection afforded by EDU is site- and species-specific and is subject to  
2 local meteorologic conditions. In addition, new evidence suggests that EDU does not always  
3 have greater effects at higher O<sub>3</sub> exposures and that the degree of protection by EDU largely  
4 depends on environmental conditions. Because of the variability observed in the level of  
5 protection provided, and the fact that mechanisms of protection afforded by EDU and other  
6 exclusion methods are unknown, caution is needed in applying this approach to the study of O<sub>3</sub>  
7 effects in the field.

8 Advancements in biomonitoring have been made since the 1996 O<sub>3</sub> AQCD, primarily in  
9 the area of identification and symptom verification of sensitive species (Flagler, 1998; Krupa  
10 et al., 1998; Innes et al., 2001; Smith et al., 2003). The U.S. Department of Agriculture (USDA)  
11 Forest Service continues its program to monitor ozone effects in forested ecosystems throughout  
12 the United States. Currently, 33 states participate in the program, which uses a grid system to  
13 identify the location of plants showing foliar injury. Although results cannot be used for  
14 developing exposure-response relationships or for quantifying responses to O<sub>3</sub>, they can provide  
15 an annual assessment and correlative information regarding the extent of O<sub>3</sub> injury occurring  
16 across many regions of the United States.

### 17 18 19 **9.3 SPECIES RESPONSE/MODE-OF-ACTION**

20 There are several steps in the process of O<sub>3</sub> uptake and toxicity that are now better  
21 understood than in 1996. These advancements are important in refining hypotheses on O<sub>3</sub> uptake  
22 and mode of action on plants and in developing a flux-based index for use in quantifying  
23 response and, ultimately, for potential use in developing a secondary national ambient air quality  
24 standard (SNAAQs). The new information available on the mode of action of O<sub>3</sub> is, in part, a  
25 result of improved molecular tools for following rapid changes that occur within the leaf (Pell  
26 et al., 1997; Sandermann, 2000; Ward et al., 1991). This new information is discussed in greater  
27 detail in Annex Section AX9.2.

28 Clearly, many changes occur within hours or possibly days following O<sub>3</sub> exposure  
29 (Sandermann, 1998). However, other O<sub>3</sub> effects take longer to occur and tend to be most  
30 obvious only under exposure to low O<sub>3</sub> concentrations for long periods (Andersen et al., 1997;  
31 Hogsett et al., 1989; Langebartels et al., 1997). These low-exposure chronic effects have been

1 linked to the senescence process or some physiological response very closely linked to  
2 senescence (e.g., translocation, reabsorption, allocation of nutrients and carbon).

3 Langebartels et al. (1997) discussed “memory” or “carry-over effects” within the plant to  
4 explain sensitivity to frost in the winter following summertime O<sub>3</sub> exposure. Others have argued  
5 that this sensitivity is due to the nutrient status of the tree during the over-wintering phase of its  
6 life and to chronic (on-going, less severe levels with fewer peaks at very high levels) exposure to  
7 ambient O<sub>3</sub> inducing (1) mineral nutrient deficiency; (2) alterations of normal metabolism,  
8 including translocation and allocation of carbohydrates and probably nitrogen; and/or (3)  
9 disturbance of normal transpiration and diurnal cycling, leading to water stress (Schmieden and  
10 Wild, 1995). While general nutrient concentrations within the foliage may not occur, localized  
11 deficiencies might. This is difficult to observe or prove without a great deal of work on all  
12 portions of a tree and without a general hypothesis of what is occurring.

13 It is important to note that the dramatic strides made over the last few years in  
14 understanding the genetic make-up of plants, gene control, and signal transduction/control will  
15 accelerate in the future and translate into better models of the hypotheses listed above as well as  
16 more detailed schemes of how O<sub>3</sub> alters basic plant metabolism. Thus, while our understanding  
17 of how O<sub>3</sub> interacts with the plant at the cellular level has dramatically improved (Assmann,  
18 2003; Assmann and Wang, 2001; Rao and Davis, 2001), the translation of those mechanisms  
19 into how O<sub>3</sub> is involved with altered cell metabolism and the subsequent reductions in whole  
20 plant productivity and other physiological facts has not yet been fully achieved. As the  
21 understanding of wounding responses in plants and more information on genome details and  
22 varied plant mutants becomes available, the cellular and physiological responses of plants to O<sub>3</sub>  
23 exposures are slowly becoming clearer. However, more studies on a larger variety of species are  
24 needed before this type of information can be incorporated into indices of response and for  
25 consideration in developing SNAAQs.

#### 26 27 28 **9.4 MODIFICATION OF FUNCTIONAL AND GROWTH RESPONSES**

29 It has been known for decades that several factors, both biotic and abiotic, alter plant  
30 response to ozone. However, only a few studies reported since the 1996 O<sub>3</sub> AQCD have  
31 improved our understanding of the role of these interactions in modifying plant O<sub>3</sub> response.

1 Quantifying how these interactions alter plant O<sub>3</sub> response is a critical first step to reducing the  
2 uncertainty in extrapolating individual plant responses to higher levels of biological  
3 organization, e.g., ecosystems. None of the recent studies have significantly improved our  
4 ability to quantify the degree to which these factors modify plant O<sub>3</sub> response; however, they  
5 have reinforced the conclusions of the 1996 O<sub>3</sub> AQCD with regard to factors known to alter  
6 plant response to O<sub>3</sub>. This new information is discussed in greater detail in Annex Section  
7 AX9.3.

8 In the area of biotic interactions, new evidence with regard to insect pests and diseases (see  
9 Docherty et al. (1997) and Flückiger et al. (2002) for recent reviews) has not reduced the  
10 uncertainties noted in the 1996 O<sub>3</sub> AQCD. Most interactions thought to affect crops, forest trees  
11 and other natural vegetation have yet to be studied. Recent studies have supported the earlier  
12 conclusion that O<sub>3</sub> often increases the likelihood and success of insect attacks, but only with  
13 respect to chewing insects (e.g., Percy et al., 2002; Kopper and Lindroth, 2003). With the  
14 economically important group of sucking insects (such as aphids), no clear trends have been  
15 revealed in the latest studies (see reviews by Docherty et al., 1997; Flückiger et al., 2002).  
16 Hence, although it seems likely that some insect problems could increase as a result of greater O<sub>3</sub>  
17 levels, we are still far from being able to predict the nature of any particular O<sub>3</sub>-plant-insect  
18 interaction, its likelihood, or its severity.

19 The situation is somewhat clearer with respect to interactions involving facultative  
20 necrotrophic plant pathogens, with O<sub>3</sub> exposure generally contributing to increased disease  
21 (Flückiger et al., 2002). With obligate biotrophic fungal, bacterial, and nematode diseases,  
22 however, twice as many reports indicate O<sub>3</sub>-induced inhibitions than enhancements. This pattern  
23 is supported by the concept put forth by Dowding (1988) that pathogens that benefit from  
24 damage to cells are enhanced by pollution stress of their hosts, whereas pathogens and pests that  
25 require healthy hosts are depressed by pollution stress. The frequent reports that infection by  
26 obligate biotrophs reduces the severity of O<sub>3</sub>-induced foliar injury (e.g., Schraudner et al., 1996)  
27 does not result in true “protection”, as the disease *per se* causes negative effects on the host  
28 plant. With obligate biotrophs, the nature of any interaction with O<sub>3</sub> is probably dictated by the  
29 unique, highly specific biochemical relationships between the pathogen and the host plant. At  
30 this time, therefore, although some diseases may become more widespread or severe as a result

1 of exposure to O<sub>3</sub>, it is still not possible to predict exactly which diseases are likely to present the  
2 greatest risks to crops and forests.

3 Recent studies have not greatly added to our understanding of the nature of interactions  
4 between O<sub>3</sub> and root symbionts, but have served to support conclusions put forth in the 1996 O<sub>3</sub>  
5 AQCD. Several studies have indicated that the functioning of tree root symbioses with  
6 mycorrhizae may be adversely affected by O<sub>3</sub> (e.g., Kytöviita et al., 2001), but there is also  
7 evidence that the presence of mycorrhizae may overcome O<sub>3</sub>-enhanced root diseases (Bonello  
8 et al., 1993). There is also evidence that O<sub>3</sub> may encourage the spread of mycorrhizae to the  
9 roots of uninfected trees. The role of O<sub>3</sub> in altering root symbionts, its interactions with soil  
10 organisms, and the subsequent feedback effects on plant growth represent one of the greatest  
11 areas of uncertainty in assessing the influence of O<sub>3</sub> on ecosystems (Andersen, 2003).

12 The few recent studies of the impact of O<sub>3</sub> on intraspecific plant competition confirmed  
13 that grasses frequently show greater resilience than other types of plants. In grass-legume  
14 pastures, the leguminous species tend to suffer greater growth inhibition (Johnson et al., 1996;  
15 Nussbaum et al., 2000). The suppression of Ponderosa pine (*Pinus ponderosa* Laws.) seedling  
16 growth by blue wild-rye grass (*Elymus glaucus* Buckl.) was markedly increased by O<sub>3</sub> (Andersen  
17 et al., 2001). However, we are far from being able to predict the outcome of the impact of O<sub>3</sub> on  
18 specific competitive situations, such as successional plant communities or crop-weed  
19 interactions.

20 Physical or abiotic factors play a large role in modifying plant response to O<sub>3</sub>, and new  
21 information is available that supports the conclusions of the 1996 O<sub>3</sub> AQCD. Although some  
22 recent field studies have indicated that O<sub>3</sub> impact significantly increases with increased ambient  
23 temperature (Ball et al., 2000; Mills et al., 2000), other studies have indicated that temperature  
24 has little effect (Balls et al., 1996; Fredericksen et al., 1996). Temperature affects the rates of all  
25 physiological processes based on enzyme-catalysis and diffusion; each process and overall  
26 growth (the integral of all processes) has a distinct optimal temperature range. It is important to  
27 note that a plant's response to changes in temperature will depend on whether it is growing near  
28 its optimum temperature for growth or near its maximum temperature (Rowland-Bamford,  
29 2000). But temperature is unquestionably an important variable affecting plant O<sub>3</sub> response in  
30 the presence of the elevated CO<sub>2</sub> levels contributing to global climate change. In contrast,  
31 evidence continues to accumulate to indicate that O<sub>3</sub> exposure sensitizes plants to low

1 temperature stress (Colls and Unsworth, 1992) and, also, that O<sub>3</sub> decreases below-ground  
2 carbohydrate reserves, which may lead to responses in perennial species ranging from rapid  
3 demise to impaired growth in subsequent seasons (i.e., carry-over effects) (Andersen et al.,  
4 1997).

5 Light, a component of the plant's physical environment, is an essential "resource" whose  
6 energy content drives photosynthesis and CO<sub>2</sub> assimilation. It has been suggested that increased  
7 light intensity may increase O<sub>3</sub> sensitivity of light-tolerant species while decreasing that of  
8 shade-tolerant species, but this appears to be an oversimplification with many exceptions. As  
9 pointed out by Chappelka and Samuelson (1998) and Topa et al. (2001), the interaction between  
10 O<sub>3</sub> sensitivity and light environment is complicated by developmental stage as well as the light  
11 environment of individual leaves in the canopy.

12 Although the relative humidity of the ambient air has generally been found to increase the  
13 adverse effects of O<sub>3</sub> by increasing stomatal conductance, and thereby increasing O<sub>3</sub> flux,  
14 abundant evidence also indicates that the ready availability of soil moisture results in greater O<sub>3</sub>  
15 sensitivity (Mills, 2002). The partial "protection" against the adverse effects of O<sub>3</sub> afforded by  
16 drought (as noted in previous O<sub>3</sub> AQCDs) has been observed in field experiments and modeled  
17 in computer simulations (Broadmeadow and Jackson, 2000). There is also compelling evidence  
18 that O<sub>3</sub> can predispose plants to drought stress (Maier-Maercker, 1998). Hence, the response  
19 will depend to some extent upon the sequence in which the stresses occur, but, even though the  
20 nature of the response is largely species-specific, successful applications of model simulations  
21 have led to larger-scale predictions of the consequences of O<sub>3</sub> × drought interactions. However,  
22 regardless of the interaction, the net result on short-term growth is negative; although in tree  
23 species, other responses such as increased water use efficiency could benefit long-term survival.

24 Somewhat analogous to temperature, it appears that any shift away from the nutritional  
25 optimum may lead to greater O<sub>3</sub> sensitivity; but the shift would have to be substantial before a  
26 significant effect on O<sub>3</sub> response was observed. Mineral nutrients in the soil, other gaseous air  
27 pollutants, and agricultural chemicals constitute chemical factors in the environment. The  
28 evidence regarding interactions with specific nutrients is still contradictory: some experimental  
29 evidence indicates that low general fertility increases sensitivity to O<sub>3</sub> (Whitfield et al., 1998;  
30 Landolt et al., 1997), although others have found less sensitivity with decreased fertility  
31 (Cardoso-Vilhena and Barnes, 2001). Simulation modeling of trees suggests that nutrient

1 deficiency and O<sub>3</sub> may act less than additively, but too many examples of contrary trends exist to  
2 permit any sweeping conclusions at this time.

3 Interactions of O<sub>3</sub> with other air pollutants have received relatively little recent attention  
4 since 1996 (see Barnes and Wellburn [1998] and Fangmeier et al. [2002] for recent reviews).  
5 The situation with SO<sub>2</sub> remains inconsistent, but SO<sub>2</sub> seems unlikely to pose any additional risk  
6 to those related to other individual pollutants. With NO and NO<sub>2</sub>, the situation is complicated by  
7 their nutritional value as a N source. Much more investigation is needed before we will be able  
8 to predict the outcomes of different O<sub>3</sub>-NO-NO<sub>2</sub> scenarios. The latest research into O<sub>3</sub> × acid  
9 rain interactions has confirmed that, at realistic acidities, significant interactions are unlikely  
10 (Momen et al., 1997; 1999; Laurence et al., 1997; Sayre and Fahey, 1999). A continuing lack of  
11 information precludes our offering any generalizations about interactive effects of O<sub>3</sub> with NH<sub>3</sub>,  
12 HF, or heavy metals. More evidence has been reported for protective effects against O<sub>3</sub> afforded  
13 by the application of fungicides (Wu and Tiedemann, 2002).

14 Considerable emphasis during the last decade has been placed on research evaluating  
15 potential O<sub>3</sub> interactions with the components of global climate change: increased atmospheric  
16 CO<sub>2</sub>, increased mean global temperatures, and increased surface level UV-B radiation.  
17 However, it must be noted that most of these studies have tended to regard increased CO<sub>2</sub> levels  
18 and increased mean temperatures as unrelated phenomena. Experiments into the effects of  
19 doubled CO<sub>2</sub> levels at today's mean ambient temperatures are of questionable value in trying to  
20 assess the impact of *climate change* on responses to O<sub>3</sub>. To date, the limited experimental  
21 evidence and that obtained by computer simulation suggest that even though an enriched CO<sub>2</sub>  
22 atmosphere (~600 ppm) would more than offset the impact of O<sub>3</sub> on responses as varied as wheat  
23 (*Triticum aestivum* L.) yield or young Ponderosa pine growth, the concurrent increase in  
24 temperature would reduce, but probably not eliminate, the net gain (Batts et al., 1997; Van Oijen  
25 and Ewart, 1999; Constable et al., 1996). There is also some recent evidence that O<sub>3</sub> and UV-B  
26 interact in their effects on plant injury and photosynthesis (Schnitzler et al., 1999), but additional  
27 research is needed to fully understand how O<sub>3</sub> interacts with multiple climate change factors.  
28

## 9.5 EFFECTS-BASED AIR QUALITY EXPOSURE INDICES

Since the 1996 O<sub>3</sub> AQCD, there has been no direct experimental testing of the adequacy of exposure indices proposed in 1996; therefore, there is no new information to alter the basic conclusions put forth in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996) with regard to exposure indices. A more detailed discussion of effects-based air quality indices can be found in Annex Section AX9.4.

Exposure indices are metrics that relate measured plant damage (i.e., reduced growth) to monitored ambient O<sub>3</sub> concentrations over time to provide a consistent metric for reviewing and comparing exposure-response effects obtained from various studies. The 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996) focused on the research used to develop various exposure indices to quantify growth and yield effects in crops, perennials, and trees (primarily seedlings), and not foliar injury. The proposed indices included various functional and statistical summaries of monitored hourly O<sub>3</sub> concentrations over designated time periods. The indices were developed through regression analyses of earlier exposure studies and was accomplished by ordering the measured responses of growth and/or yield of crops and tree (seedling) species in response to O<sub>3</sub>. Their development focused on consideration and inclusion of some, but not all, the factors that affect O<sub>3</sub> uptake and expression of effects (e.g., Lee et al. [1988]).

The few studies that have been published since the 1996 O<sub>3</sub> AQCD continue to support the earlier conclusions, including the importance of peak concentrations, and the duration and occurrence of O<sub>3</sub> exposures in altering plant growth and yield. In addition, a large body of new research, mostly out of Europe, addresses the need for an index related to the actual uptake of O<sub>3</sub> by the plant and the flux of O<sub>3</sub> from the atmosphere to the O<sub>3</sub> affected plant tissues. Despite additional research linking estimates of flux with plant response since 1996, information is still insufficient to identify a flux-based model that incorporates the necessary complexity across space and time to be non-site or non-species specific. Based on the current state of knowledge, exposure indices that cumulate and differentially weight the higher hourly average O<sub>3</sub> concentrations, but include the mid-level values, still represent the best approach for relating vegetation effects to O<sub>3</sub> exposure in the United States.

The new studies have also substantiated earlier conclusions on the role of exposure components including concentration, duration, and exposure patterns in determining plant growth response to O<sub>3</sub> (Oksanen and Holopainen, 2001; Yun and Laurence, 1999). Recent

1 studies using different exposure patterns have confirmed earlier studies on the role of higher  
2 concentrations and exposure duration (Nussbaum et al., 1995). A role for higher concentrations  
3 is inferred based on improved air quality in regions in the Western United States (Lefohn and  
4 Shadwick, 2000). For example, the O<sub>3</sub> reductions in the San Bernardino Mountain area since the  
5 late 1970s are associated with reductions in the higher hourly average O<sub>3</sub> concentrations, the  
6 number of hours of concentrations  $\geq 0.95$  ppm, and the cumulative concentration-weighted  
7 exposure index (Lee et al., 2003). The mid-range concentrations appeared to be relatively  
8 unchanged or even slightly increasing over the period of 1980 to 2000. General forest  
9 improvement has been reported following a decrease of O<sub>3</sub> along a decreasing gradient of  
10 exposure (Miller and Rechel, 1999; Arbaugh et al., 2003; Tingey et al., 2004). These studies  
11 suggest the focus should be on the higher O<sub>3</sub> concentrations, while including the lower levels,  
12 when estimating the effects of O<sub>3</sub> precursor emission reduction strategies on vegetation.

13 New studies have demonstrated the potential disconnection of peak events and maximal  
14 stomatal conductance at xeric to mesic sites in California (Panek et al., 2002; Grulke et al., 2002;  
15 Panek, 2004). In addition, a few studies have indicated that O<sub>3</sub> uptake during nighttime hours is  
16 greater than previously thought (Grulke et al., 2004; Massman, 2004); and a review of the  
17 literature suggests a large number of species exhibit some degree of conductance at night  
18 (Musselman and Minnick, 2000). These studies suggest a reconsideration of cumulating  
19 exposure 24 h/day and not just during daylight hours in exposure index determinations. This  
20 lack of coincidence in temporal patterns of conductance and peak ambient concentrations  
21 introduces uncertainty in assessing the impact of O<sub>3</sub>. The use of an exposure index that does not  
22 consider regionally unique climate and site factors that modify stomatal conductance may, as a  
23 result, under- or over- estimate growth effects. The shortcomings of an ambient exposure-based  
24 index is especially apparent when assessing the potential impact of O<sub>3</sub> across broad climatic  
25 regions of the United States or Europe. Various means to overcome this potential problem were  
26 addressed with several new studies; one solution would be to add other components to the  
27 present statistical summaries of exposure indices (e.g., meteorological) to develop flux-based  
28 indices. However, the increased biological and meteorological information in these indices may  
29 make them more regional in their applicability.

30 A number of studies have taken a flux-based approach to improve upon the  
31 concentration-based (i.e., exposure indices) approach as a means to address the issue of



1 assessing risk of O<sub>3</sub> across different climatic regions. The European acceptance and use of the  
2 flux-based critical values is, in part, a recognition of the landscape scaling problems associated  
3 with ambient exposure-based indices. A great deal of progress has occurred in developing and  
4 testing stomatal models that may be generally applicable across certain vegetation types  
5 (Danielsson et al., 2003; Emberson et al., 2000; Grünhage and Jäger, 2003; Matyssek et al.,  
6 2004; Pleijel et al., 2000). While a flux-based approach is preferred, a cautionary argument was  
7 advanced in a few publications based on the nonlinear relationship between O<sub>3</sub> uptake and foliar  
8 injury (growth was not assessed). The concern is that not all O<sub>3</sub> stomatal uptake results in a  
9 yield reduction, which depends to some degree on the amount of internal detoxification  
10 occurring with each particular species. Those species having high amounts of detoxification  
11 potential may, in fact, show little relationship between O<sub>3</sub> stomatal uptake and plant response  
12 (Musselman and Massman, 1999).

13 Given the current state of knowledge and the best available data, exposure indices that  
14 cumulate and differentially weight the higher hourly average concentrations, and also include the  
15 mid-level values, continue to offer the most defensible approach for use in developing response  
16 functions and comparing studies as well as future indices for vegetation protection. A large  
17 database exists that has been used for establishing exposure-response relationships; however, at  
18 this time, such a database does not exist for relating O<sub>3</sub> flux to growth response.

19 It is anticipated that as the overlapping relationships of conductance, concentration, and  
20 defense mechanisms are better defined, the flux-based indices may be able to predict vegetation  
21 injury and/or damage across varied landscapes and climates with more accuracy than the  
22 exposure-response models. However, it is unclear that such is the case at this time. The  
23 translation of these indices from research and assessment tools to air quality standards has the  
24 additional need to be simple, understandable, and adaptive to a manageable monitoring program.

## 25 26 27 **9.6 OZONE EXPOSURE-PLANT RESPONSE RELATIONSHIPS**

28 Data published since 1996 continue to support the conclusions of previous O<sub>3</sub> AQCDs that  
29 there is strong evidence that ambient concentrations of O<sub>3</sub> cause foliar injury and growth and  
30 yield damage to numerous common and economically valuable plant and tree species. For  
31 annual vegetation, the data summarized in Table AX9-16 (see Annex Section AX9.5) show a

1 range of growth and yield responses both within species and among species. Nearly all of these  
2 data were derived from studies in OTCs, with only two studies using open-air systems in the  
3 United Kingdom (Ollerenshaw et al., 1999; Ollerenshaw and Lyons, 1999). It continues to be  
4 difficult to compare studies that report O<sub>3</sub> exposure using different indices, such as AOT40,  
5 SUM06, W126, or 7-h or 12-h mean values.

6  
7 The AOT40, SUM06, and W126 indices are defined as follows:

8 **AOT60:** the seasonal sum of the difference between an hourly concentration above the  
9 threshold value of 60 ppb, minus the threshold value of 60 ppb;

10  
11 **SUM06:** the seasonal sum of hourly concentrations at or above the threshold value of 60 ppb;  
12 and

13  
14 **W126:** a sigmoid functional weighting of all hourly concentrations for the season.  
15

16 When such index comparisons can be made, the results of recent research confirm earlier  
17 results summarized in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996).  
18 A summary of earlier literature concluded that a 7-h, 3-month mean of 49 ppb O<sub>3</sub> corresponding  
19 to a SUM06 exposure of 26 ppm·h, would cause 10% loss in 50% of 49 experimental cases  
20 (Tingey et al., 1991). Recent data summarized in Table 9-16 support this conclusion and more  
21 generally indicate that ambient O<sub>3</sub> exposures can reduce the growth and yield of annual species.  
22 Some annual species such as soybean [*Glycine max* (L.) Merr.] are more sensitive, and greater  
23 losses in such species may be expected (Table 9-16). Thus, the recent scientific literature  
24 supports the conclusions of the 1996 O<sub>3</sub> AQCD that ambient O<sub>3</sub> concentrations are reducing the  
25 yield of major crops in the United States.

26 Much research in Europe has used the cutoff-concentration weighted, cumulative-exposure  
27 statistic AOT40; and substantial effort has gone into developing “Level-1” critical levels for  
28 vegetation using this index. Based on regression analysis of 15 OTC studies of spring wheat  
29 including one U.S. study and 14 studies from locations ranging from southern Sweden to  
30 Switzerland, an AOT40 value of 5.7 ppm·h was found to correspond to a 10% yield loss, and a  
31 value of 2.8 ppm·h corresponded to a 5% yield loss (Fuhrer et al., 1997). Because a 4 to 5%  
32 decrease could be detected with a confidence level of 99%, 3 ppm·h was selected by the  
33 European Union as the AOT40 critical level in 1996 (Kärenlampi and Skärby, 1996).

1 In addition to reductions in crop yield, O<sub>3</sub> may also reduce the quality or nutritive value of  
2 annual species. Many studies have found O<sub>3</sub> effects on various measures of plant organs that  
3 affect quality, with most of those studies focusing on characteristics important for food or  
4 fodder. These studies indicate that ambient O<sub>3</sub> may have economically important effects on the  
5 quality of crop and forage species. Previous O<sub>3</sub> AQCDs have concluded that visible symptoms  
6 on marketable portions of crops and ornamental plants can occur with seasonal 7-h mean O<sub>3</sub>  
7 exposures of 40 to 100 ppb (U.S. Environmental Protection Agency, 1978; 1986; 1992; 1996).  
8 The recent scientific literature does not refute this conclusion.

9 The use of OTCs may reverse the usual vertical gradient in O<sub>3</sub> that occurs within a few  
10 meters above the ground surface (see Annex Section AX9.1). This reversal suggests that OTC  
11 studies may, to some degree, overestimate the effects of an O<sub>3</sub> concentration as measured several  
12 meters above the ground. However, such considerations do not invalidate the conclusion of the  
13 1996 O<sub>3</sub> AQCD that ambient O<sub>3</sub> exposures are sufficient to reduce the yield of major crops in the  
14 United States.

15 As found for single-season agricultural crops, yields of multiple-year forage crops are  
16 reduced at ozone exposures that occur over large areas of the United States. This result is  
17 similar to that reported in the 1996 O<sub>3</sub> AQCD (U.S. Environmental Protection Agency, 1996).  
18 When species are grown in mixtures, O<sub>3</sub> exposure can lead to the increased growth of O<sub>3</sub>-tolerant  
19 species while exacerbating the growth decrease of O<sub>3</sub>-sensitive species. Because of this  
20 competitive interaction, the total growth of the mixed-species community may not be affected by  
21 O<sub>3</sub> exposure. However, in some cases, mixtures of grasses and clover species have shown  
22 significant decreases in total biomass growth in response to O<sub>3</sub> exposure in studies in the United  
23 States and in Sweden. In Europe, a provisional AOT40 critical level of 7 ppm·h over 6 months  
24 has been proposed by the European Union as a value to protect sensitive herbaceous perennial  
25 plant species from the adverse effects of O<sub>3</sub>.

26 For deciduous tree species, recent evidence from FACE and OTC studies supports results  
27 observed in previous OTC studies. For example, a series of O<sub>3</sub>-FACE studies was undertaken in  
28 Rhineland, WI (Isebrands et al., 2000, 2001). These studies showed that O<sub>3</sub> symptom  
29 expression was generally similar in OTCs, FACE, and also at sites along an ambient O<sub>3</sub> gradient,  
30 supporting the previously observed variation among aspen clones obtained using OTCs

1 (Karnosky et al., 1999). As has been observed in previous O<sub>3</sub> AQCDs, root growth is often  
2 found to be the most sensitive biomass response to O<sub>3</sub>.

3 Results since 1996 support the conclusion of the 1996 O<sub>3</sub> AQCD (U.S. Environmental  
4 Protection Agency, 1996) that deciduous trees are generally less sensitive to O<sub>3</sub> than are most  
5 annual plants, with the exception of a few very sensitive genera such as *Populus* and sensitive  
6 species such as black cherry (*Prunus serotina* Ehrh.). However, the data presented in  
7 Table AX9-18 (see Annex Section AX9.5) suggest that ambient exposures that occur in the  
8 United States can sometimes reduce the growth of seedlings of deciduous species. Results from  
9 multi-year studies sometimes show a pattern of increased effects in subsequent years. In some  
10 cases, however, growth decreases due to O<sub>3</sub> may become less significant or even disappear over  
11 time. While some mature trees show greater O<sub>3</sub> sensitivity than do seedlings in physiological  
12 parameters such as net photosynthetic rate, these effects may not translate into measurable  
13 reductions in biomass growth. However, because even multi-year experiments do not expose  
14 trees to O<sub>3</sub> for more than a small fraction of their life span and because competition may, in  
15 some cases, exacerbate the effects of O<sub>3</sub> on individual species, determining O<sub>3</sub> effects on mature  
16 trees remains a significant challenge.

17 In Europe, a Level I critical level has been set for forest trees based on OTC studies of  
18 European beech (*Fagus sylvatica* L.) seedlings: defined as an AOT40 value of 10 ppm·h for  
19 daylight hours for a 6-month growing season (Kärenlampi and Skärby, 1996). However, other  
20 studies show that other species, such as silver birch (*Betula pendula* Roth.), may be more  
21 sensitive to O<sub>3</sub> than beech (Pääkkönen et al., 1996).

22 As found for other tree species, various evergreen tree species and genotypes have widely  
23 varying O<sub>3</sub> sensitivities. Based on OTC studies with seedlings, major evergreen species in the  
24 United States are generally less sensitive than are most deciduous trees, and slower-growing  
25 evergreen species are less sensitive than are faster-growing species. There is evidence that  
26 interacting stress, such as competition, may increase the O<sub>3</sub> sensitivity of trees. As in studies of  
27 deciduous species, most experiments with evergreen species have only covered a very small  
28 portion of the life span of a tree and have been conducted with seedlings, so estimating effects  
29 on mature evergreens is difficult.

30 For all types of perennial vegetation, cumulative effects over more than one growing  
31 season may be important; and studies for only a single season may underestimate effects.

1 Mature trees may be more or less sensitive to O<sub>3</sub> than are seedlings, depending on the species,  
2 but information on physiological traits can be used to predict such differences in specific cases.  
3 In some cases, mature trees may be more sensitive to O<sub>3</sub> than seedlings due to differences in gas  
4 exchange rates, differences in growth rates, greater cumulative exposure, or the interaction of O<sub>3</sub>  
5 with other stressors.

## 8 **9.7 EFFECTS OF OZONE EXPOSURE ON NATURAL ECOSYSTEMS**

9 There is evidence that tropospheric O<sub>3</sub> is an important stressor of ecosystems, with  
10 documented impacts on the biotic condition, ecological processes, and chemical/physical nature  
11 of natural ecosystems (See Table AX9-22; Annex Section AX9.6). In turn, the effects of O<sub>3</sub> on  
12 individual plants and processes are scaled up through the ecosystem affecting processes such as  
13 energy and material flow, inter- and intraspecies competition, and net primary productivity  
14 (NPP). Thus, effects on individual keystone species and their associated microflora and fauna,  
15 which have been shown experimentally, may cascade through the ecosystem to the landscape  
16 level, although this has not yet been demonstrated. By affecting water balance, cold hardiness,  
17 tolerance to wind and by predisposing plants to insect and disease pests, O<sub>3</sub> may even impact the  
18 occurrence and impact of natural disturbance (e.g., fire, erosion). Despite the probable  
19 occurrence of such effects, there are essentially no instances where ecosystem level, highly  
20 integrated studies have conclusively shown that ozone is indeed altering ecosystem structure  
21 and/or function.

22 Systematic injury surveys demonstrate that foliar injury occurs to O<sub>3</sub> sensitive species in  
23 many regions of the United States (Smith et al., 2003; Coulston et al., 2003; Chappelka et al.,  
24 1997) (Campbell et al., 2000) and Europe (Braun et al., 1999). However, the frequent lack of  
25 correspondence between foliar symptoms and growth effects means that other methods must be  
26 used to estimate the regional effects of O<sub>3</sub> on tree growth rates (e.g., Rebbeck, 1996; Kouterick  
27 et al., 2000). Investigations of the radial growth of mature trees in combination with data from  
28 many controlled studies with seedlings and a few studies with mature trees suggest that ambient  
29 O<sub>3</sub> is reducing the growth of mature trees in some locations (Somers et al., 1998). Studies using  
30 models based on tree physiology and forest stand dynamics suggest that modest effects of O<sub>3</sub> on  
31 growth may accumulate over time and may interact with other stressors (Laurence et al., 2001)

1 (Laurence et al., 2003). For mixed-species stands, such models predict that overall stand growth  
2 rate is generally not likely to be affected. However, competitive interactions among species may  
3 change as a result of growth reductions of O<sub>3</sub>-sensitive species (Weinstein et al., 2001). These  
4 results suggest that O<sub>3</sub> exposure over decades may be altering the species composition of forests  
5 in some regions.

6 Despite increased understanding of possible ecosystem effects of ozone, the data base  
7 demonstrating and quantifying the degree to which O<sub>3</sub> is altering natural ecosystems is sparse.  
8 Much of the speculation of ozone impact on ecosystems must be inferred from a number of case  
9 studies of forest plot field-based data reporting on a number of different species. One means to  
10 discuss our current knowledge is by listing the areas in which more information is needed.

11 These include:

12 *Ecosystem processes.* Very little is known about the effects of O<sub>3</sub> on water, carbon, and  
13 nutrient cycling, particularly at the stand and community levels. Effects on below-ground  
14 ecosystem processes in response to O<sub>3</sub> exposure alone, and in combination with other stressors,  
15 are critical to projections at the watershed and landscape levels. Little is yet known about the  
16 effects of O<sub>3</sub> on structural or functional components of soil food webs or how these impacts  
17 could affect plant species diversity (Andersen, 2003).

18 *Biodiversity and genetic diversity.* The study of genetic aspects of O<sub>3</sub> impacts on natural  
19 ecosystems has been largely based on correlations, and it remains to be shown conclusively  
20 whether O<sub>3</sub> affects biodiversity or genetic diversity (Pitelka, 1988; Winner et al., 1991; Davison  
21 and Barnes, 1998). Studies of competitive interactions under elevated O<sub>3</sub> levels are needed  
22 (Laurence and Andersen, 2003). Reexaminations via new sampling of population studies to  
23 bring in a time component to previous studies showing spatial variability in population responses  
24 to O<sub>3</sub> are also needed. These studies could be strengthened by modern molecular methodologies  
25 to quantify impacts on diversity.

26 *Natural ecosystem interactions with the atmosphere.* Little is known about feedbacks  
27 between O<sub>3</sub> and climate change on production of volatile organic compounds, which, in turn,  
28 could affect O<sub>3</sub> production (Fuentes et al., 2001). At moderate to high O<sub>3</sub> exposure sites,  
29 aberrations in stomatal behavior could significantly affect individual tree water balance of O<sub>3</sub>-  
30 sensitive trees; and if the sensitive tree species is dominant, the hydrologic balance at the  
31 watershed and landscape levels could be affected. This has not been addressed in any model,

1 because O<sub>3</sub>-exposure effects, if included at all in the modeling effort, have assumed a linear  
2 relationship between assimilation and stomatal conductance. Interaction studies with other  
3 components of global change (i.e., warming, increasing atmospheric CO<sub>2</sub>, N deposition, etc.) or  
4 with various biotic stressors are needed to better predict complex interactions likely in the future  
5 (Laurence and Andersen, 2003). Whether O<sub>3</sub> will negate the positive effects of an elevated CO<sub>2</sub>  
6 environment on plant carbon and water balances is not yet known, nor is it known if these effects  
7 will scale up through the ecosystem. How O<sub>3</sub> affects the progress of pest epidemics and insect  
8 outbreaks as concentrations increase is unclear (Ball et al., 1998). Information concerning the  
9 impact of O<sub>3</sub> on plant pest and insect reproductive processes and reproductive development  
10 under realistic field or forest conditions are needed as well as examination of reproductive  
11 effects under interacting pollutants (Black et al., 2000).

12 *Scaling.* The vast majority of O<sub>3</sub> studies of trees have been conducted with young,  
13 immature trees and in trees that have not yet formed a closed canopy. Questions remain as to the  
14 comparability of O<sub>3</sub> effects on juvenile and mature trees and on trees grown in the open versus  
15 those in a closed forest canopy in a competitive environment (Chappelka and Samuelson, 1998;  
16 Kolb and Matyssek, 2001; Samuelson and Kelly, 2001). Merging the effects of O<sub>3</sub> across spatial  
17 scales is also difficult. Scaling responses of a single or a few plants to effects on communities  
18 and ecosystems are complicated matters that will require a combination of manipulative  
19 experiments with model ecosystems; community and ecosystem studies along natural O<sub>3</sub>  
20 gradients; and extensive modeling efforts to project landscape level, regional, national and  
21 international impacts of O<sub>3</sub>. Linking these various studies via impacts on common research  
22 quantification across various scales using measures of such factors as leaf area index or spectral  
23 reflective data, which could eventually be remotely sensed (Kraft et al., 1996; Panek et al.,  
24 2003), would provide powerful new tools for ecologists.

25 *Identifying endpoints.* In general, methodologies to determine the important values of  
26 services and benefits derived from natural ecosystems are lacking. Identifying and quantifying  
27 factors that could be used in comprehensive risk assessment for O<sub>3</sub> effects on natural ecosystems  
28 would increase societal awareness of the importance of protecting ecosystems (Heck et al.,  
29 1998).

## 9.8 ECONOMICS

Substantial progress has been made over the past two decades in our understanding of the effects of ozone and other oxidants on vegetation, particularly for agriculturally important plant species. See Annex Section AX9.7 for a more detailed discussion. The physical and economic effects on agriculture are well documented and provide useful information for the consideration of establishing air quality standards for crops (e.g., Spash, 1997).

Since the completion of the National Crop Loss Assessment Network (NCLAN) program in the late 1980s, the number of economic assessments of air pollution studies focusing on terrestrial ecosystems in general, and agriculture in particular, has declined. For example, for the period of 1980 to 1990, 33 economic studies of O<sub>3</sub> and other air pollutant effects on U.S. crops were published in peer-reviewed journal outlets (Spash, 1997). However, in preparing this section of the current O<sub>3</sub> AQCD, only four peer-reviewed economic assessments were found for the decade of 1991 to 2000 that addressed vegetation in the United States. In addition, one peer-reviewed article (Kuik et al., 2000) was found dealing with agriculture in the Netherlands. Recent interest in global climate change and the potential effects of global warming on O<sub>3</sub> and other photochemical oxidants, has renewed interest in the effects of air pollution on both managed and unmanaged terrestrial ecosystems (Adams et al., 1998). In addition, concern is growing for regarding the effects of air pollutants on natural ecosystems and on the services they provide (Daily, 1997). Unfortunately, this interest has not yet translated into additional peer-reviewed publications addressing O<sub>3</sub> or other air pollutants effects on ecosystems.

A study by Murphy et al. (1999) of the economic effects of tropospheric O<sub>3</sub> on U.S. agriculture is of note here, because it confirms the general magnitude of economic effects reported by the two key studies performed a decade earlier (Adams, 1986; 1985). Specifically, Murphy et al. (1999) evaluated benefits to eight major crops associated with several scenarios concerning the reduction or elimination of O<sub>3</sub> precursor emissions from motor vehicles in the United States. Their analysis reported a \$2.8 to 5.8 billion (1990 dollars) benefit from complete elimination of O<sub>3</sub> exposures from all sources, i.e., ambient O<sub>3</sub> reduced to a background level assumed to be 0.025 to 0.027 ppm. While the analytical framework is similar to Adams et al. (1986) in the use of NCLAN-based yield response functions and a mathematical programming-based economic optimization model, the study is novel in its focus on the role of motor vehicle emissions of VOCs/NO<sub>x</sub> in total anthropogenic O<sub>3</sub> levels. The study is also



1 notable in its careful attention to federal farm program effects, particularly the deficiency  
2 payment component.

3       There have been a number of recent studies of air pollutant effects on tree species in the  
4 literature. Some have reported changes in total biomass and focused on European species  
5 (Kurczynska et al., 1997). Other studies have assessed changes in composition of forest species  
6 (biodiversity) or forest health due to exposure to air pollutants (Bringmark and Bringmark, 1995;  
7 McLaughlin and Percy, 1999; Vacek et al., 1999). As noted previously, changes in forest  
8 biomass and composition are more difficult to value than marketable products. However,  
9 measures of forest composition or health have implications for an area of increasing policy  
10 concern, that being the effect of air pollutants and other environmental stressors on unmanaged  
11 (natural) ecosystems and the services they provide (Goulder and Kennedy, 1997; Pimentel et al.,  
12 1997). Considerable discussion has occurred among ecologists and economists as to the  
13 appropriate means for valuing these services (Anderson, 1990; Carpenter and Dixon, 1985;  
14 Common and Perrings, 1992). A number of conceptual articles have been published on this  
15 issue in both economic and ecological journals (Bergstrom, 1990; Castle, 1993; Pearce, 1993;  
16 Suter, II, 1990).

17       Effects on forests and natural ecosystems remain problematic, due to limitations in  
18 biological response data and economic methods. The problem is even more acute for valuing  
19 natural ecosystem service flows. The current limitations surrounding forests and natural  
20 ecosystems present a rich research agenda. Areas of greatest potential value in terms of regional  
21 policymaking need to be prioritized. Such priority setting can be assisted by sensitivity analyses  
22 with existing economic models. By measuring the changes in economic effects arising from  
23 changes in key parameters, it is possible to identify those research data gaps most likely to affect  
24 economic values.

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52

1                   **10. TROPOSPHERIC OZONE EFFECTS ON**  
2                   **UV-B FLUX, AND ITS ROLE IN CLIMATE CHANGE**

3  
4  
5                   **10.1 INTRODUCTION**

6                   In addition to exerting direct effects on human health and vegetation/ecosystems, as  
7                   discussed in earlier chapters, tropospheric ozone (O<sub>3</sub>) influences the ground-level flux of solar  
8                   ultraviolet (UV) radiation, as well as other processes that alter the Earth’s radiative balance and  
9                   contribute to climate change. This chapter discusses tropospheric O<sub>3</sub> and (1) its role in in  
10                  determining surface-level UV flux and, (2) its involvement in global climate change.

11  
12  
13                  **10.2 THE ROLE OF TROPOSPHERIC OZONE IN DETERMINING**  
14                  **GROUND-LEVEL UV-B FLUX**

15                  Atmospheric O<sub>3</sub> plays a crucial role in reducing the exposure of living organisms to solar  
16                  UV radiation. Approximately 90% of the total atmospheric O<sub>3</sub> burden is located in the  
17                  stratosphere; therefore, photochemical processes that alter the concentration of stratospheric O<sub>3</sub>  
18                  are of particular concern within the global community. The importance of stratospheric O<sub>3</sub>  
19                  depletion due to the release of long-lived anthropogenic chlorinated- and fluorinated  
20                  hydrocarbons was recognized over a period of several years during the 1970s and early 1980s  
21                  and led to the international treaty for the protection of stratospheric O<sub>3</sub>: the 1987 Montreal  
22                  Protocol on Substances that Deplete the Ozone Layer.

23                  While roughly representing only 10% of the total atmospheric O<sub>3</sub> burden, tropospheric O<sub>3</sub>,  
24                  like other tropospheric pollutants, can influence the flux of biologically-damaging UV radiation  
25                  at the Earth’s surface. This section summarizes the available information on the factors  
26                  governing UV flux at the Earth’s surface, first, followed by a discussion the status of scientific  
27                  understanding of the factors governing human UV exposure and, then, of the links between  
28                  UV exposure and human disease.



## 10.2.1 Factors Governing Ultraviolet Radiation Flux at the Earth's Surface

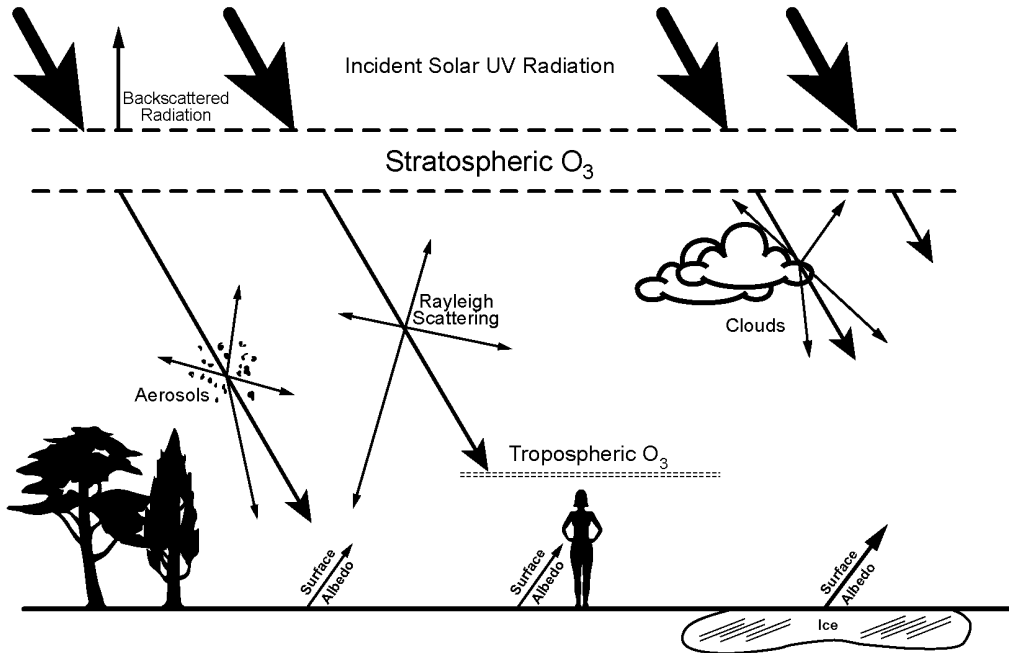
The Montreal Protocol requires routine review of the latest scientific information available on the status of the O<sub>3</sub> layer and of UV radiation levels at the Earth's surface. The World Meteorology Organization (WMO) and U.N. Environmental Program (UNEP) are responsible for assessing the state of the science regarding the O<sub>3</sub> layer and for reporting on trends in surface UV radiation levels. The latest WMO/UNEP assessment was published in 2002 (WMO/UNEP, 2002).

An outcome of the on-going atmospheric chemistry research effort devoted to tracking stratospheric O<sub>3</sub> depletion and its effects is a body of literature, though limited, that describes the effects of tropospheric pollutants, particulate matter (PM), and O<sub>3</sub>, on ground-level UV radiation flux. Drawing from the WMO/UNEP assessment and more recent literature, this section describes the current level of scientific understanding of the factors influencing ground-level UV radiation flux such as geophysical factors, tropospheric O<sub>3</sub>, PM, and cloud cover. Figure 10-1 visually summarizes some of the factors that influence the flux of UV-B at the Earth's surface.

### 10.2.1.1 UV Radiation:: Wavelengths, Energies and Depth of Atmospheric Penetration

Designations for portions of the electromagnetic spectrum have evolved over time and are usually associated with general functions or effects caused by photons within a given wavelength range. The energy possessed by a photon is inversely proportional to its wavelength. For example, gamma rays, having wavelengths  $\leq 0.1$  nm, are especially damaging high-energy photons emitted during radioactive decay and by stellar activity. Radiowaves, having wavelengths  $\geq 10^8$  nm, are very low in energy and function as carriers for broadcast communications.

The wavelengths ranging between 50 and 400 nm in length are denoted "ultraviolet." Solar radiation of wavelengths  $< 280$  nm, including UV-C (200 to 280 nm), is almost entirely blocked by the Earth's upper atmosphere due to photoionization and photodissociation processes. Figure 10-2 compares the solar flux above the atmosphere with ground-level flux. Solar UV-B radiation (280 - 320 nm) is absorbed or scattered in part within the atmosphere, while UV-A radiation (320 - 400 nm) can be scattered but not absorbed to any meaningful degree by atmospheric gases. Both UV-B and UV-A photons contain sufficient energy to break (photolyze) chemical bonds and are associated with human health- and ecosystem-damaging

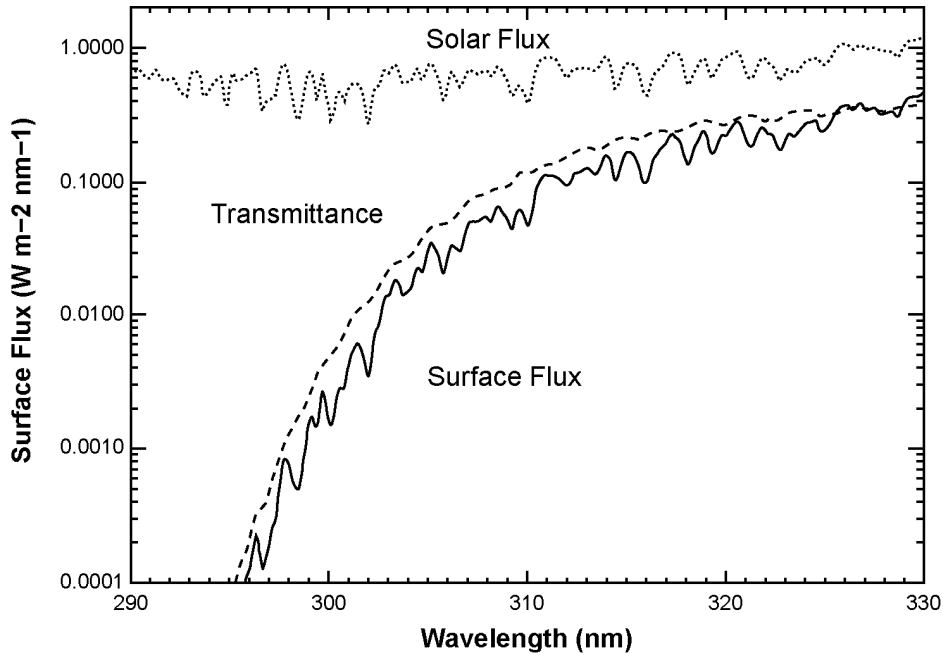


**Figure 10-1. Complexity of factors that determine human exposure to UV radiation. In addition to the geophysical/atmospheric factors (e.g., stratospheric and tropospheric O<sub>3</sub>, clouds, aerosols, and Rayleigh scattering) that affect the solar flux of UV radiation at surface level, there are human physical, behavioral and demographic factors that influence human exposure to UV radiation.**

1 effects. However, because UV-B is more energetic, it is potentially capable of producing  
 2 substantially more biological damage than UV-A.

#### 3 4 **10.2.1.2 Temporal Variations in Solar Flux**

5 The magnitude of the solar radiation flux entering the atmosphere depends upon long-term  
 6 solar activity, sunspot cycle (11 years), solar rotation (27 days) and the position of the Earth in  
 7 its orbit around the sun. A variety of changes in solar irradiance can be found in historical data,  
 8 from 1700 to the present. Solanki and Fligge (2000) concluded that solar irradiance changes on  
 9 time-scales of days to centuries can be attributed to variations in solar magnetic features. Since  
 10 the last Maunder minimum in 1700, solar irradiance has increased slightly, at ~3.0% for  
 11 wavelengths in the UV-C range and at ~1.3% for wavelengths in the UV-B and UV-A ranges.



**Figure 10-2. Comparison of solar flux above the atmosphere with flux at the Earth’s surface. The dotted line represents extraterrestrial solar flux measured by the satellite UARS SOLSTICE instrument (dotted line). The dashed line represents calculated atmospheric transmittance and the solid line is the calculated absolute flux of UV radiation for a solar zenith angle of 50deg, total column O<sub>3</sub> of 275 DU, and a surface reflectivity of 8%. The fine structure on the surface flux trace results from Fraunhofer lines (absorption specific wavelengths within the solar atmosphere).**

Source: Krotkov et al. (1998).

1 Including visible wavelengths, Solanki and Fligge (2000) estimated that the overall increase in  
 2 solar irradiance was ~0.3%. Rozema et al. (2001) pointed out that any increase in wavelengths  
 3 <300 nm (UV-C) would initiate additional O<sub>3</sub> formation in the stratosphere. This suggests that  
 4 any increase in UV-B and/or UV-A solar flux would be offset by a more absorptive stratosphere.

5 Solar rotation and sunspot activity have the greatest effects on radiation flux originating in  
 6 the highest levels of the solar atmosphere. The amplitude of the associated cyclical changes in  
 7 solar shortwave radiation flux follows an inverse relationship between photon wavelength and  
 8 the solar altitude at which it was emitted. The maximum level of radiation (solar-max) differs

1 from the minimum (solar-min) by as much as 10% for wavelengths near 160 nm. This peak-to-  
2 trough difference declines to around 1% for 300 nm (UV-B range) (Salby, 1996).

3 The combined effects of the Earth's obliquity (the angle of the Earth's axis of rotation with  
4 respect to the plane of its orbit around the sun) and its precession (the rotation of the Earth's axis  
5 with respect to a perpendicular line through the plane of its solar orbit) yield variations of up to  
6 30% in total summertime solar flux, depending on latitude (Hartmann, 1994).

### 7 8 *Zenith Angle: Latitude, Season, and Time of Day*

9 The sun's relative elevation is measured with respect to the vertical and is known as its  
10 "zenith angle." This angle varies hourly, seasonally, and with latitude. Daily and seasonal  
11 changes in solar zenith angle result in the largest changes in the magnitude of solar radiation  
12 flux, with higher zenith angles corresponding to lower solar flux. The largest natural fluxes  
13 occur in the tropical regions, where solar noon occurs at a zenith angle at or near 0°. Seasonal  
14 variation in solar flux ranges from small changes at the equator to very large changes at high  
15 latitudes. Daily variations in solar flux, from sunrise to sunset, show added wavelength  
16 dependence as a function of zenith angle, because transmission of some wavelengths are  
17 sensitive to atmospheric pathlength due to scattering and absorption processes. These processes  
18 will be discussed further below.

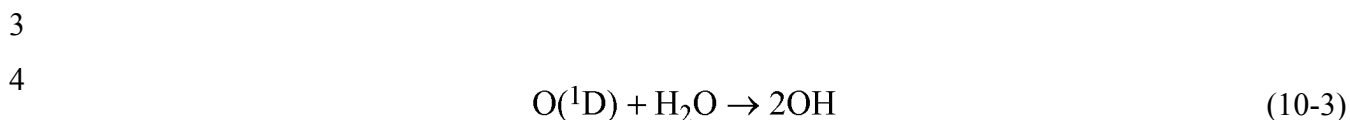
### 19 20 **10.2.1.3 Atmospheric Radiative Interactions with Solar Ultraviolet Radiation**

#### 21 *Radiative Interactions in the Stratosphere*

22 As noted, earlier, the stratosphere contains 90% or more of the total column density of O<sub>3</sub>,  
23 the principle gas phase absorber of UV-B. Ozone interacts with UV radiation by scattering the  
24 photon, or absorbing and transforming its energy. Upon absorbing a UV photon, O<sub>3</sub> may  
25 photodissociate, or become electronically and vibrationally excited.

26 Photoabsorption by O<sub>3</sub> occurs with very high efficiency. After electronically excited O<sub>3</sub>  
27 (O<sub>3</sub><sup>\*</sup>) is formed, it will either lose its excess electronic energy via a collision with another gas  
28 molecule (M) or dissociate into ground-state oxygen, O<sub>2</sub>, and an electronically excited oxygen  
29 radical, O(<sup>1</sup>D) (See Reactions 1 and 2). Intermolecular collisions degrade the excess electronic  
30 energy of the O<sub>3</sub><sup>\*</sup> molecule by transferring it to other molecules as vibrational, rotational, and/or  
31 translational energies, that warm the atmosphere. An O(<sup>1</sup>D) radical can react with H<sub>2</sub>O to form

1 two hydroxyl (OH) radicals (Reaction 3). See Chapter 2 for further discussion of odd oxygen  
2 and HO<sub>x</sub> photochemistry.



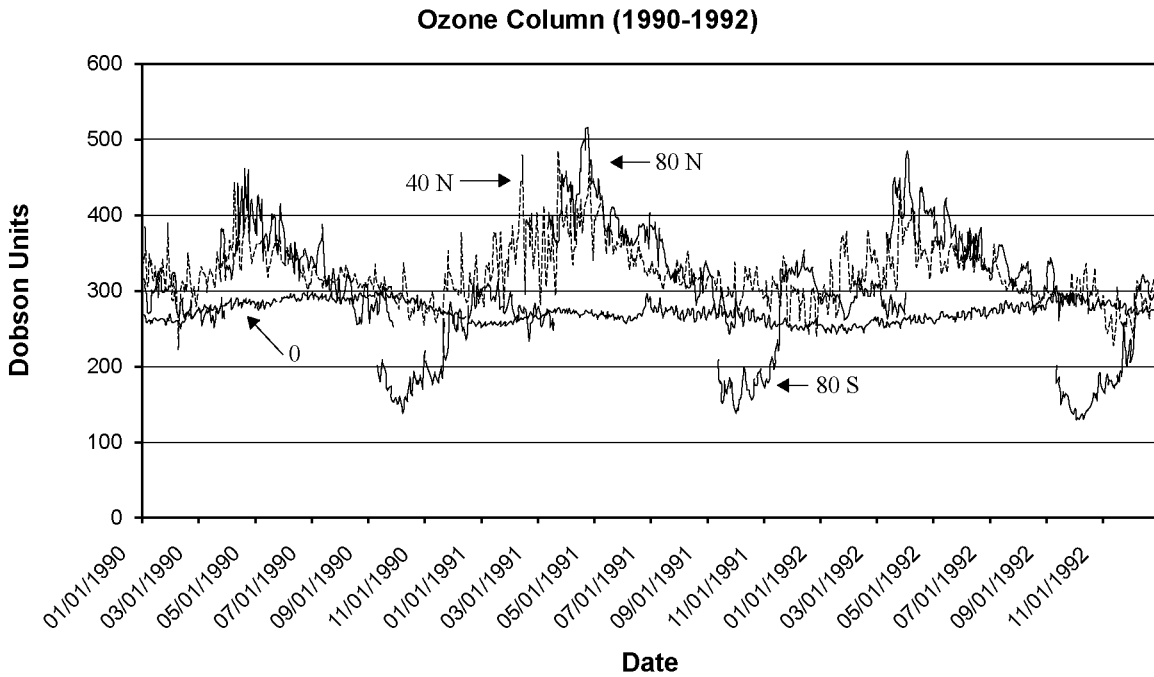
7 Either of these photochemical processes transforms the energy of the UV photon into heat, a  
8 form of energy that, in this context, lacks the potential for human health or ecosystems damage.

9 The WMO/UNEP (2002) scientific assessment reported that global average total  
10 column O<sub>3</sub> had declined by 3% from pre-1980 levels, due to the presence of anthropogenic O<sub>3</sub>-  
11 depleting substances in the atmosphere. Ozone depletion has a strong latitude and seasonal  
12 dependence. The seasonality of total O<sub>3</sub> changes differ between the Northern and Southern  
13 Hemispheres. In the northern midlatitudes, total column O<sub>3</sub> declined by ~4% during the  
14 winter/spring seasons and by approximately half that amount in the summer/fall of the  
15 1997-2001 time period, relative to pre-1980 total column O<sub>3</sub> levels. In southern midlatitudes,  
16 total column O<sub>3</sub> declined ~6% during all seasons.

17 The concentration of O<sub>3</sub> in a vertical column extending from the Earth's surface is  
18 expressed in Dobson Units (DU) corresponding to the column height in hundredths of a  
19 millimeter of O<sub>3</sub> at standard temperature and pressure (273 K and 1 atmosphere) (Wayne, 2000).  
20 One DU = 2.587 × 10<sup>15</sup> molecules of O<sub>3</sub>/cm<sup>2</sup>. The total O<sub>3</sub> column effectively prevents any  
21 UV-C from reaching the surface and reduces the penetration of UV-B to the surface, but it does  
22 little to attenuate the intensity of UV-A except at the shorter wavelengths close to the cutoff for  
23 UV-B. Cutchis (1974) calculated that with overhead sun, a 10% decrease in the O<sub>3</sub> column  
24 would lead to 20, 250, and 500% increases in flux at 305, 290, and 287 nm, respectively, values  
25 that have been supported by ground observations in Toronto, ON (49° N; Kerr and McElroy,  
26 1993). Rapid changes of this magnitude appear to happen naturally. As seen in data collected

1 by the Total Ozone Mapping Satellite (TOMS) (Figure 10-3), the total O<sub>3</sub> column undergoes  
2 wide natural variation on short timescales (Cockell, 2001).

3



**Figure 10-3. Ozone column abundances from the years 1990 to 1992 for 0, 40, and 80° N as well as 80° S. The data for 80° S are incomplete, but the graph shows the effects of the Antarctic O<sub>3</sub> hole on total column abundances at this latitude. The data for the Northern Hemisphere illustrate the natural variations in the O<sub>3</sub> column over time. The data are taken from the TOMS (Total Ozone Monitoring Satellite) data set (1979 to 1993).**

Source: Cockell (2001).

1 Nacreous and polar stratospheric clouds, and aerosols, such as those injected into the  
2 stratosphere by explosive volcanic eruptions, both absorb and scatter radiation. Relative to the  
3 troposphere, the stratosphere is low in atmospheric pressure. Stratospheric clouds and aerosols  
4 are also more dispersed than those in the troposphere. Consequently, UV radiation can traverse  
5 the stratosphere with a substantially lower probability of encountering a gas molecule or cloud or  
6 an aerosol particle than it would in the troposphere. In the radiative transfer literature, the

1 stratosphere is described as a “single scattering” regime for UV radiation, and UV that has  
2 penetrated the stratosphere is referred to as “direct beam UV.” The troposphere, due to its high  
3 gas and particle concentrations is referred to as a “multiple scattering” regime.  
4

### 5 ***Radiative Interactions in the Troposphere: Solar Irradiance Versus Actinic Flux***

6 The troposphere contains  $\leq 10\%$  of the total column  $O_3$  but  $\sim 78\%$  of the total atmospheric  
7 mass (including clouds, and gas- and particle-phase radiation scatterers and absorbers), making it  
8 a “multiple scattering” regime for UV radiation. These scattering processes increase the mean-  
9 free path a photon must travel before reaching the Earth’s surface, transforming the direct beam  
10 UV solar irradiance that has penetrated the stratosphere into diffuse, or actinic, UV irradiance  
11 (Brühl and Crutzen, 1988).

12 Rayleigh and Mie scattering of solar radiation are sensitive to molecular and particle size.  
13 In Rayleigh scattering, gas molecules that are smaller than the wavelength of the incident photon  
14 isotropically deflect incoming photons. Conversely, aerosol and cloud droplets scatter incoming  
15 radiation with distinctive forward- and backscattering tendencies, i.e., Mie scattering. Actinic  
16 flux, especially at the Earth’s surface, is directly proportional to surface albedo (Wendisch and  
17 Mayer, 2003). Surface albedo is very strongly wavelength dependent. For example, fresh and  
18 wet snow reflect 60 to 90% of incident violet light, while soil and grass surfaces reflect  $<5\%$   
19 (Xenopoulos and Schindler, 2001). In their in situ measurement and modeling study of the  
20 vertical distribution of solar irradiance, Wendisch and Mayer (2003) found that surface albedos  
21 must be measured in order to accurately simulate solar flux, due to the large variations in albedo  
22 that may occur within a given surface type. Snow cover, even many kilometers from  
23 measurement sites is known to increase detected UV irradiances. Complicated interactions  
24 result when radiation is scattered by snow (or other bright surfaces) and backscattered or  
25 absorbed by atmospheric particles and clouds in the same vicinity (WMO/UNEP, 2002).  
26

### 27 ***Variation in Solar Flux with Altitude***

28 Solar flux increases with altitude above sea level, due to the decreased presence of clouds  
29 and declining concentrations of scattering and absorbing atmospheric pollutants. Rayleigh  
30 scattering, also lessens with decreasing atmospheric pressure. A number of measurements of  
31 UV radiation have been taken at various altitudes and are reviewed by Xenopoulos and Schindler

1 (2001). Increases in flux as a function of altitude are given as percent irradiance enhancement  
2 per 1000 m relative to sea level. The effect can range from 9 to 24% /1000 m as function of the  
3 altitude at which the measurement was taken (Xenopoulos and Schindler, 2001). The effect  
4 corresponds to the relative pathlength traveled by the solar photon: flux is strongest when the  
5 photon is not impeded by atmospheric scattering or absorbing agents. Similarly, this effect is  
6 seen as a function of solar zenith angle, i.e., flux is at its maximum when the atmospheric depth  
7 through which the photon must pass is at its shallowest.

### 8 9 ***Clouds***

10 In principle, clouds have the largest influence on surface-level UV irradiance, but their  
11 effects are difficult to quantify. The depth and composition of a cloud determine, in part, the  
12 amount and wavelengths of radiation that it will scatter or absorb. Geometry is an especially  
13 important factor, as the reduction in irradiance may be small with scattered or broken clouds – or  
14 may be enhanced by scattering between clouds, increasing surface flux (WMO/UNEP, 2002).  
15 Quantifying the effect of clouds on surface UV flux, therefore, requires detailed information on  
16 cloud composition, geometry, altitude, and the position of the sun relative to the cloud and the  
17 underlying surface as a function of time. Provided that all of this information is available, a  
18 three-dimensional model is then required to calculate surface-level reductions or enhancements  
19 in UV flux.

### 20 21 ***Particulate Matter***

22 On a zonally averaged basis, PM does not contribute significantly to lower tropospheric  
23 absorption of UV radiation. However, in urban areas or other areas subject to high smog levels  
24 (areas of significant biomass combustion), PM may be the most important determinant of  
25 ground-level erythemal UV flux, second only to cloud cover (U.S. Environmental Protection  
26 Agency, 2004; WMO/UNEP, 2002). Model-to-measurement comparisons of ground-level flux  
27 for Greece and Toronto, Canada, have shown 20 and 5-10% reductions, respectively (McKenzie  
28 et al., 2003). Increases over the past 20 to 30 years in combustion-associated PM and black  
29 carbon may account for the inability to detect a surface trend in UV-B radiation caused by a  
30 known decrease in stratospheric O<sub>3</sub> over the Northern Hemisphere (Barnard et al., 2003).

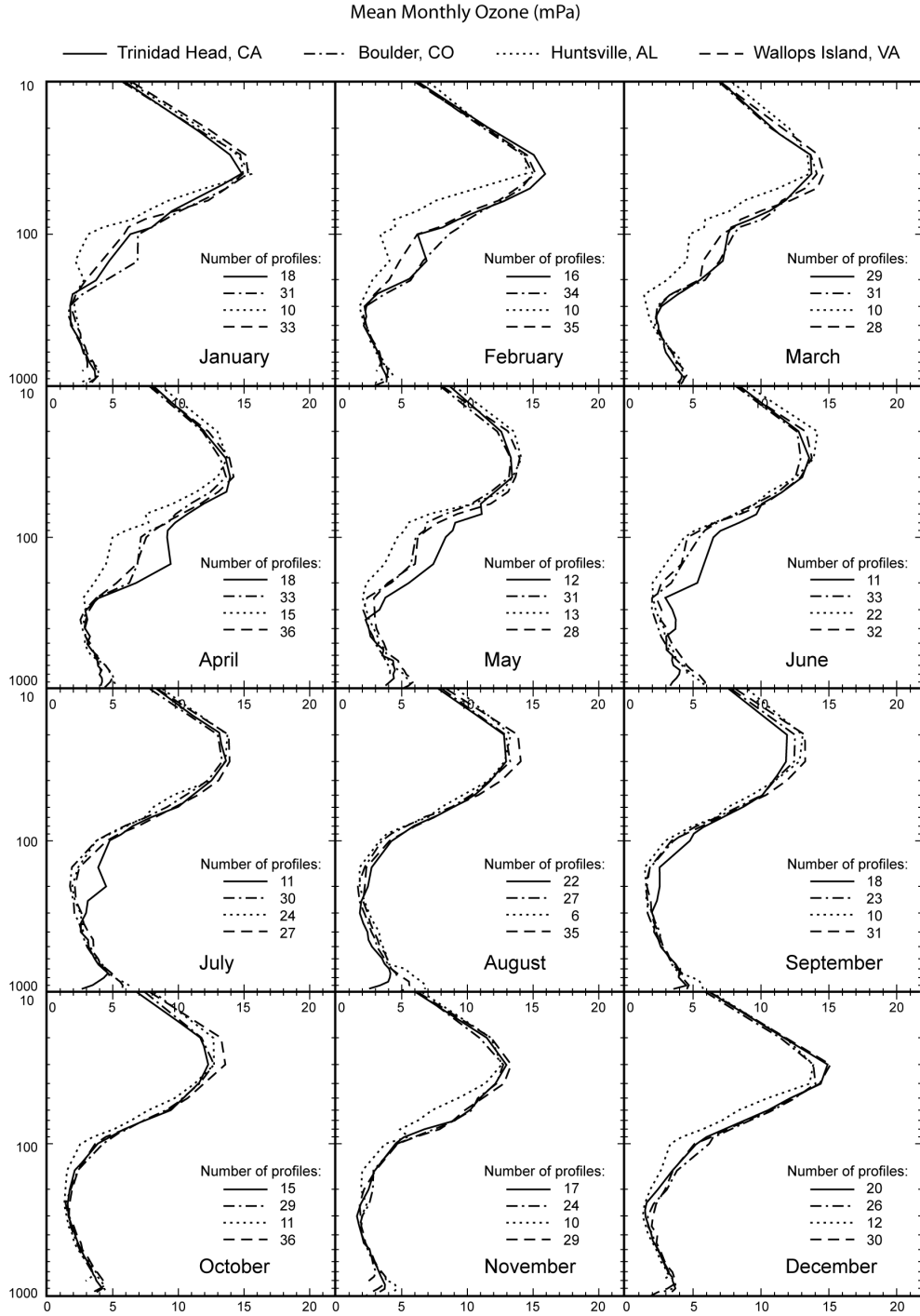


## 1 **Gases**

2 In the upper troposphere, the UV-absorbing gases O<sub>3</sub> and, of lesser importance,  
3 formaldehyde (CH<sub>2</sub>O) and SO<sub>2</sub> are vented or diffuse from the surface. Stratospheric intrusions  
4 force O<sub>3</sub>-rich air into the troposphere where it mixes, increasing regional background O<sub>3</sub> levels  
5 (see Chapter 3). Tropospheric O<sub>3</sub> data are typically expressed on a concentration basis, e.g.,  
6 parts per billion by volume (ppbv), where 1 ppbv tropospheric O<sub>3</sub> ≡ 0.65 DU (IPCC, 2001a).  
7 Ozone concentrations decrease with increasing altitude from the surface up to, roughly, the mid-  
8 troposphere, then increase up into the stratosphere. Figure 10-4 shows a series of O<sub>3</sub> vertical  
9 profiles for 4 sites within the continental U.S., i.e., plots of O<sub>3</sub> concentrations as a function of  
10 atmospheric pressure (correlating to altitude). The mean values of O<sub>3</sub> in the free troposphere  
11 reported in the literature range from ~50 to ~80 ppbv, with higher values occurring at the  
12 tropopause. For example, a series of ozonesonde soundings over France from 1976 to 1995  
13 showed an O<sub>3</sub> increase from 48.9 ppbv in the 2.5 to 3.5 km layer to 56.5 ppbv in the 6.5 to  
14 7.5 km layer, although the data revealed no statistically significant increasing trend over time  
15 (Ancellet and Beekmann, 1997).

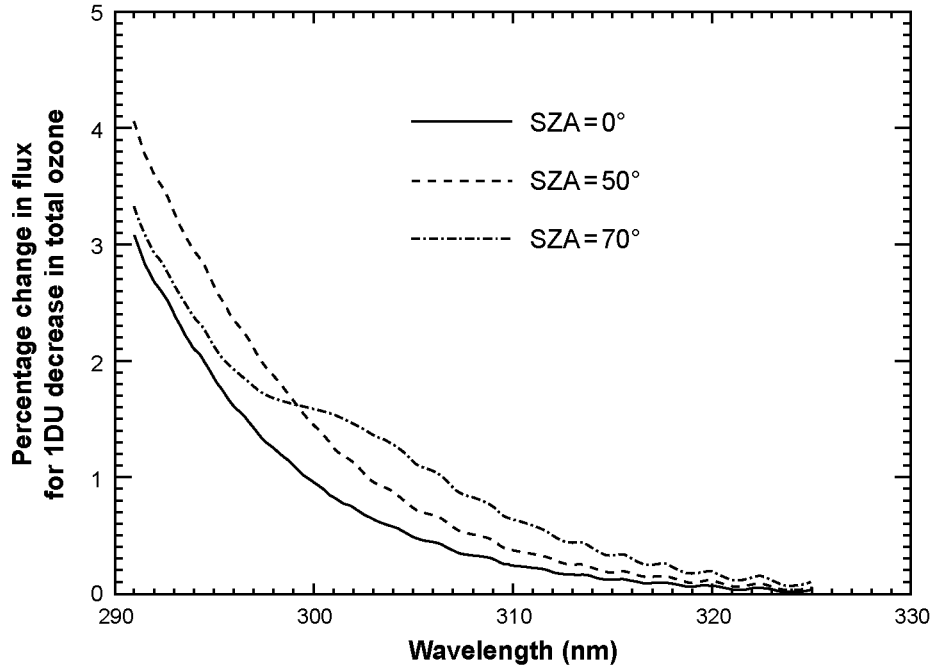
16 Photochemistry produces a diurnal rise and fall in O<sub>3</sub> and PM concentrations in polluted  
17 urban settings. Temperature inversions that often occur in these settings prevent the upward  
18 mixing and dilution of ground-level O<sub>3</sub>, also trapping primary and secondary PM within the  
19 boundary layer. A recent study of the concurrence of O<sub>3</sub> and PM is provided by  
20 Koloutsou-Vakakis et al. (2001). No measurement technique is currently available that can  
21 distinguish between absorption of incident UV radiation by O<sub>3</sub> versus absorption by PM.

22 Ultraviolet absorption by gases becomes significant under aerosol- and cloud-free  
23 conditions. Figure 10-5 shows a calculation by Krotkov et al. (1998) of the sensitivity, as a  
24 function of wavelength, of ground-level UV flux to a 1-DU decrease in total column O<sub>3</sub> under  
25 cloud- and aerosol-free conditions. A 1991 to 1992 study in Chicago in which ambient O<sub>3</sub>,  
26 broadband UV irradiance, and total sunlight were monitored (Frederick et al., 1993) found a  
27 significant negative correlation between the UV irradiance and ambient O<sub>3</sub> when the atmosphere  
28 was relatively free of clouds and haze. Although Frederick et al. (1993) estimated that a  
29 10-ppbv reduction in O<sub>3</sub> was associated with a 1.3% increase in erythemally-weighted UV-B,  
30 they cautioned that this value had a comparatively large uncertainty (±1.2%, or nearly 100% of  
31 the predicted increase).



**Figure 10-4. Monthly averaged vertical O<sub>3</sub> profiles (partial pressure in mPa) as a function of atmospheric pressure (in mBar) for Trinidad Head, CA (solid line); Boulder, CO (dot-dashed line); Huntsville, AL (dotted line); and Wallops Island, VA (dashed line). The number of launches at each site for each month are indicated on the charts.**

Source: Newchurch et al. (2003).



**Figure 10-5. The sensitivity of ground-level UV flux to a 1 DU change in total column O<sub>3</sub>, under clear sky conditions, as a function of solar zenith angle (SZA).**

Source: Krotkov et al. (1998).

1 Note that when attempting to apply the results of studies such as Krotkov et al. (1998) in  
 2 an analysis of the importance of surface O<sub>3</sub> to protecting humans from UVB-related diseases, is  
 3 that flux is ordinarily determined as a function of total column O<sub>3</sub> density, which includes the  
 4 stratospheric and upper tropospheric O<sub>3</sub>, not simply surface-level pollutant O<sub>3</sub>. Nearly all of the  
 5 routine data on tropospheric O<sub>3</sub> concentrations in the U.S. is from ground-level O<sub>3</sub> monitors,  
 6 such as those used to determine the attainment of the O<sub>3</sub> air quality standards. Such  
 7 measurements, alone, are not sufficient information for making reliable estimates of ambient O<sub>3</sub>  
 8 concentrations above the boundary layer.

9

10 **10.2.1.4 Data Requirements for a Surface UV-B Climatology**

11 A means of establishing the range of variability in UV-B at ground level would be in the  
 12 development of a map of flux levels under typical seasonal conditions based on historical

1 records. In the atmospheric sciences community, a map of this type is referred to as a  
2 “climatology.”

3 The WMO/UNEP (2002) stated that, in principle, if the spatial distribution of all UV  
4 absorbers and scatterers were fully known, the wavelength and angular distribution of the UV  
5 irradiance at the Earth’s surface could be determined with model calculations. However, the  
6 very limited information available on the distribution of the primary components (i.e., clouds,  
7 particles, O<sub>3</sub>, and surface albedo) makes detailed predictions impossible. In an earlier  
8 assessment of the environmental effects of stratospheric O<sub>3</sub> depletion (WMO/UNEP, 1999), the  
9 UNEP concluded that, in view of the high spatial and temporal variability of surface UV  
10 radiation and the difficulty in maintaining calibration within networks of UV monitoring  
11 instruments, satellite-based observations are necessary to develop a satisfactory UV climatology.  
12 Furthermore, satellite-derived estimates of surface UV are limited by the availability of  
13 instruments in orbit, with available datasets comprising interpolations based upon a single  
14 satellite overpass per day for a given region. Complete assessment of the uncertainties in  
15 predictions of UV surface flux would require comparisons between sparse available ground-level  
16 observations and satellite data over longer periods of time and for different geographical  
17 locations. No such assessment has been reported in the scientific literature.

### 18 19 **10.2.2 Factors Governing Human Exposure to Ultraviolet Radiation**

20 An assessment of public health benefits due to the attenuation of UV-B radiation by  
21 surface-level O<sub>3</sub> requires appropriate consideration of: (1) the multiple factors that alter the flux  
22 of UV-B radiation at ground-level, as described above; (2) the factors that influence the extent of  
23 human exposure to UV-B radiation, particularly behavioral decisions; and (3) the effects of UV-  
24 B radiation exposure on human health. Consideration must also be given to the public health  
25 benefits from exposure to UV-B radiation. The present section outlines the most recent  
26 information on the determinants of exposure to UV-B radiation in human populations.  
27 Quantitative evaluation of human exposure to UV-B radiation is scientifically necessary to  
28 perform health risk assessment and to define subpopulations at risk for UV-B-related health  
29 effects.

### 10.2.2.1 Outdoor Activities

Exposure to solar UV radiation is related to one predominating factor: time spent outdoors during daylight hours. A large U.S. study was conducted using the EPA National Human Activity Pattern Survey (NHAPS) to assess UV radiation dose in Americans (Godar, 2001; Godar et al., 2001, 2003). The EPA NHAPS recorded the activity profiles of 9,386 Americans (age 0 to 60+ years) over a 24-month period to assess their exposure to various environmental pollutants, including UV radiation. Available UV radiation was assessed using the EPA UV-monitoring program. Solar radiation in the UV-A and UV-B waveband regions were measured daily at a monitoring site in each quadrant of the U.S. There is considerable error associated with quantifying UV radiation dose from exposure surveys and four UV-monitoring sites across the country; however, the qualitative information regarding factors that increase human exposure to UV radiation is still of relevance. The EPA-UV monitoring network has since expanded to 21 sites, located in 14 U.S. national parks and 7 urban areas across the U.S.

(<http://www.epa.gov/uvnet/>). A UV-B monitoring network by the U.S. Department of Agriculture is also available for the quantitative assessment of UV radiation exposure (<http://uvb.nrel.colostate.edu/UVB/>). This monitoring network has 30+ monitoring sites across the U.S. and three additional sites in Canada and New Zealand.

Godar et al. (2001) observed a strong seasonal preference for outdoor activities, with people spending the most time outdoors during the summer followed by spring, fall, and, lastly, winter. Because the solar erythemal (i.e., skin reddening) UV radiation dose is also highest during the summer, the estimated UV radiation dose of Americans was more than 10-fold greater in the summer compared to the winter season (Godar et al., 2001).

Vacationing at the beach in the summer was associated with higher UV radiation exposures (Godar et al., 2001; Thieden et al., 2001). Even after accounting for sunscreen use at the beach, the erythemal UV radiation doses were more than 40% higher during a 3-week beach vacation compared to a 3-week stay at home (Godar et al., 2001). Danish children and adolescents were found to receive >50% of their annual UV radiation dose while vacationing at European beaches (Thieden et al., 2004a). Sunbathing also was associated with increased annual UV radiation dose in the Canadian National Survey on Sun Exposure and Protective Behaviours (Shoveller et al., 1998). Among the 3,449 adults (age 25+ years) who completed the telephone household survey, 21% stated that they spent time actively sunbathing. In a Danish study with 164

1 participants, all children (age 1 to 12 years) and teenagers (age 13 to 19 years) as well as 94% of  
2 adults (age 20 to 76 years) had days with risk behavior (Thieden et al., 2004b). Teenagers, who  
3 had the highest number of risk-behaviors days, were found to have the highest annual UV  
4 radiation doses. Among teenagers, 76% (95% CI: 41, 98) of their UV radiation dose during the  
5 measurement period was received on risk-behavior days, as determined using personal electronic  
6 UV dosimeters and exposure diaries (Thieden et al., 2004b).

7 An Australian study examining time profiles of daily UV radiation exposure among 8th  
8 grade students observed that up to 47% of the daily UV radiation dose fell within the time  
9 periods when students were outdoors during school hours, sitting under shaded structures during  
10 lunch breaks and participating in routine outdoors or sports activities (Moise et al., 1999). Other  
11 studies also have found that participation in outdoor sports (e.g., basketball, soccer, golfing,  
12 swimming, cycling) significantly increased UV radiation exposure (Moehrle, 2001; Moehrle  
13 et al., 2000; Thieden et al., 2004a,b).

#### 14 15 **10.2.2.2 Occupation**

16 Of the various factors that affect human exposure to UV radiation, occupation is also  
17 important. Approximately 5% of the American workforce work outdoors, as determined by the  
18 EPA NHAPS (Godar et al., 2001). On average, American indoor workers spend ~10% of their  
19 day outdoors. During their time outdoors, they are exposed to ~30% of the total ground-level  
20 UV flux, as measured by the EPA UV-monitoring program (Godar et al., 2001). Compared to  
21 indoor or in-home workers, outdoor workers are exposed to much higher levels of UV radiation  
22 (Kimlin et al., 1998a; Thieden et al., 2004a), frequently at levels that are above current exposure  
23 limits set by the International Commission on Non-Ionizing Radiation Protection (ICNIRP,  
24 2004). For example, Thieden et al. (2004a) observed that the annual UV radiation dose,  
25 estimated using personal electronic UV dosimeters and exposure diaries, was ~70% higher for  
26 gardeners than indoor workers. The gardeners received the majority (55%) of their UV radiation  
27 dose on working days (Thieden et al., 2004a). Another study found that outdoor workers  
28 received three to four times the annual UV radiation exposure of indoor workers (Diffey, 1990).  
29 At-risk working populations include farmers (Airey et al., 1997; Schenker et al., 2002),  
30 fishermen (Rosenthal et al., 1988), landscapers (Rosenthal et al., 1988), building and  
31 construction workers (Gies and Wright, 2003), physical education teachers (Vishvakarman et al.,

2001), mail delivery personnel (Vishvakarman et al., 2001), and various other workers who spend the majority of their day outdoors during peak UV radiation hours.

### 10.2.2.3 Age

Age may be a factor that influences human exposure to UV radiation. In a large U.S. study using the EPA NHAPS, the average UV radiation dose among American children (age <12 years) was estimated to be slightly higher (~20%) than that of adolescents (age 13 to 19 years) (Godar, 2001). A large Canadian survey found that 89% of children (age <12 years) had 30 minutes or more of daily UV exposure compared to 51% for both adults (age 25+ years) and youth (age 15 to 24 years) (Lovato et al., 1998a, 1998b; Shoveller et al., 1998). In an English study (Diffey et al., 1996), UV radiation exposure was estimated in 180 children (age 9 to 10 years) and adolescents (age 14 to 15 years) using personal film badges and exposure records. Once again, children were found to have received higher UV radiation exposure compared to adolescents (Diffey et al., 1996). However, as discussed earlier, a Danish study found that the annual UV radiation dose in teenagers (age 13 to 19 years) was 14-24% higher compared to children (age 1 to 12 years) and adults (Thieden et al., 2004b). This increase in UV radiation dose in the Danish teenagers was attributed to their increased risk-behavior days. Therefore, age may affect human exposure to UV radiation by influencing other factors of exposure, such as outdoor activity and risk behavior.

Two studies examined lifetime UV radiation exposure among persons in the U.S. (Godar et al., 2003) and Denmark (Thieden et al., 2004b). Both studies observed that while there are slight differences in UV radiation dose by age, generally people receive fairly consistent UV doses at different age intervals throughout their lives.

### 10.2.2.4 Gender

Studies have indicated that females generally spend less time outdoors and, consequently, have lower UV radiation exposure compared to males (Gies et al., 1998; Godar et al., 2001; Shoveller et al., 1998). The U.S. study by Godar et al. (2001) observed that while both males and females had relatively consistent erythemal UV radiation doses throughout their lives, males consistently received higher overall UV doses compared to females at all age groups. Among all Americans, the lowest exposure to UV radiation was received in females during their child-

1 raising years (age 22 to 40 years) (Godar et al., 2001). The highest exposure was observed in  
2 males aged 41 to 59 years in the U.S. study (Godar et al., 2001). A similar Canadian survey  
3 found that younger adult males had the greatest exposures to UV radiation (Shoveller et al.,  
4 1998).

#### 6 **10.2.2.5 Geography**

7 In the U.S. study by Godar et al. (2001), erythemal UV radiation doses were examined in  
8 persons living in northern and southern regions. Northerners and southerners were found to  
9 spend an equal amount of time outdoors; however, the higher solar flux at lower latitudes  
10 significantly increased the annual UV radiation dose for southerners (Godar et al., 2001). The  
11 annual UV radiation doses in southerners were 25 and 40% higher in females and males,  
12 respectively, compared to northerners (Godar et al., 2001). Other studies also have shown that  
13 altitude and latitude influence personal exposure to UV radiation (Kimlin et al., 1998b; Rigel  
14 et al., 1999).

#### 16 **10.2.2.6 Protective Behavior**

17 Protective behaviors such as using sunscreen (e.g., Nole and Johnson, 2004), wearing  
18 protective clothing (e.g., Rosenthal et al., 1988; Sarkar, 2004; Wong et al., 1996), and spending  
19 time in shaded areas (Moise et al., 1999; Parisi et al., 1999) have been shown to reduce exposure  
20 to UV radiation. In one study, the use of sunscreen was associated with extended intentional UV  
21 radiation exposure (Autier et al., 1999); however, a follow-up study indicated that sunscreen use  
22 increased duration of exposures to doses of UV radiation that were below the threshold level for  
23 erythema (Autier et al., 2000).

24 In a national study of U.S. youths aged 11 to 18 years, the most prevalent protective  
25 behavior was sunscreen use (39.2%) followed by use of a baseball hat (4.5%) (Davis et al.,  
26 2002). There were significant differences in the use of sunscreen by age group and gender, with  
27 the younger age group (age 11 to 13 years) and girls having greater likelihood (47.4 and 48.4%,  
28 respectively) of using sunscreen (Davis et al., 2002). The Canadian National Survey on Sun  
29 Exposure and Protective Behaviours observed that less than half of the adults (age 25+ years,  
30 n = 3,449) surveyed took adequate protective actions (Shoveller et al., 1998). Once again,  
31 children (age <12 years, n = 1,051) were most protected from exposure to UV radiation, with



1 76% using sunscreen and 36% avoiding the sun, as reported by their parents (Lovato et al.,  
2 1998a). However, the protection level was still not adequate, as indicated by the high 45% rate  
3 of erythema in children. Among Canadian youth (age 15 to 24 years, n = 574), protective  
4 actions from UV radiation exposure included wearing a hat (38%) and seeking shade and  
5 avoiding the sun between the peak hours of 11:00 a.m. to 4:00 p.m. (26%) (Lovato et al., 1998b).  
6 The lowest prevalence of protective behavior among the youth was likely responsible for the  
7 highest proportion of erythema (68%) experienced in this age group. A Danish study observed  
8 that both children and teenagers applied sunscreen on more days than adults, but teenagers had  
9 the most days with erythema, due to their increased risk behavior (Thieden et al., 2004b).  
10 A survey in Switzerland of 1,285 individuals, including children and parents, indicated that  
11 sunscreen use was the protective action most commonly used, but only at the beach and not in  
12 routine daily exposure (Berret et al., 2002). In general, protective clothing and avoiding the sun  
13 were not highly used among these individuals to protect against UV-related health effects.  
14

#### 15 **10.2.2.7 Summary of Factors that Affect Human Exposures to Ultraviolet Radiation**

16 The factors that potentially influence UV radiation doses were discussed in the previous  
17 sections and include outdoor activities, occupation, age, gender, geography, and protective  
18 behavior. Results from the various studies indicate that the following subpopulations may be at  
19 risk for higher exposures to UV radiation:

- 20 • Individuals who engage in high-risk behavior, viz., sunbathing;
- 21 • Individuals who participate in outdoor sports and activities;
- 22 • Individuals who work outdoors with inadequate shade, e.g., farmers, fishermen,  
gardeners, landscapers, building and construction workers; and
- 23 • Individuals living in geographic areas with higher solar flux (i.e., lower latitudes  
[e.g., Honolulu, HI] and higher altitudes [e.g., Denver, CO]).

#### 25 **10.2.3 Factors Governing Human Health Effects due to Ultraviolet Radiation**

26 Ultraviolet radiation occupies a specific region of the electromagnetic spectrum of  
27 wavelengths and can be further subdivided into three parts, UV-A (320 to 400 nm), UV-B  
28 (280 to 320 nm), and UV-C (200 to 280 nm). Most of the health risks associated with UV  
29 radiation exposure are wavelength dependent. Wavelengths <180 nm are of little practical

1 biological significance as they are almost completely absorbed by the stratosphere (ICNIRP,  
2 2004).

3 “Action spectra” of a given biological response to UV radiation across its spectral range  
4 are used to estimate exposure by weighting individual wavelength intensities according to the  
5 associated response. The overall effectiveness of the incident flux at inducing the biological  
6 response of interest is computed by means of the relationship:

$$7 \quad \text{effective irradiance} = \int_{\lambda} I_{\lambda} E_{\lambda} d\lambda \quad (10-4)$$

10 where  $I_{\lambda}$  and  $E_{\lambda}$  are, respectively, the irradiance and its relative effectiveness at wavelength  $\lambda$ .

11 Until 1980, it was generally thought that wavelengths <315 nm were responsible for the  
12 most significant adverse UV radiation health effects; however, recent studies have found that the  
13 longer wavelengths in the UV-A range also may produce adverse responses at substantially  
14 higher doses (ICNIRP, 2004). As UV-A radiation is not absorbed by O<sub>3</sub>, health effects solely  
15 induced by UV-A exposure are not relevant in an analysis of public health risks/benefits  
16 associated with O<sub>3</sub>-related UV attenuation. Therefore, this section focuses on the latest available  
17 information on the various adverse health effects associated with acute and chronic UV-B  
18 radiation exposure.  
19

### 20 **10.2.3.1 Erythema**

#### 21 *Association Between Ultraviolet Radiation Exposure and Erythema*

22 The most conspicuous and well-recognized acute response to UV radiation is erythema, or  
23 the reddening of the skin, which is likely caused by direct damage to DNA by UV-B and UV-A  
24 radiation (Matsumura and Ananthaswamy, 2004). Indirect oxidative damage also may occur at  
25 longer wavelengths (Matsumura and Ananthaswamy, 2004). Skin type appears to play a large  
26 role in the sensitivity to UV radiation-induced erythema. The Fitzpatrick classifications for skin  
27 types are: (1) skin type I – individuals with extremely sensitive skin that sunburns easily and  
28 severely, and is not likely to tan (e.g., very fair skin, blue eyes, freckles); (2) skin type  
29 II – individuals with very sensitive skin that usually sunburns easily and severely, and tans  
30 minimally (e.g., fair skin, red or blond hair, blue, hazel or brown eyes); (3) skin type  
31

1 III – individuals with sensitive skin that sunburns moderately and tans slowly (e.g., white skin,  
2 dark hair); (4) skin type IV – individuals with moderately sensitive skin that sunburns minimally  
3 and usually tans well (e.g., white or light brown skin, dark hair, dark eyes); (5) skin type  
4 V – individuals with minimally sensitive skin that rarely sunburns and tans deeply (e.g., brown  
5 skin); and (6) skin type VI – individuals with nonsensitive skin that never sunburns and tans  
6 profusely (e.g., dark skin). Harrison and Young (2002) found that the perceptible minimal  
7 erythemal dose was approximately twofold greater for individuals with skin type IV compared to  
8 skin type I, although there was considerable overlap in the minimal erythemal dose among the  
9 four skin types. Waterston et al. (2004) further observed that within an individual, erythemal  
10 response differed by body site (e.g., abdomen, chest, front upper arm, back of thigh). These  
11 differences were likely attributable to body site-specific variations in melanin pigmentation.

12 Kollias et al. (2001) investigated the change in erythemal response following a previous  
13 exposure to UV radiation. Body sites that received a second exposure to UV radiation always  
14 showed a reduced erythemal response compared to body sites with a single exposure, especially  
15 when the first exposure was at levels greater than the minimal erythemal dose. The suppression  
16 of erythema was more pronounced when the second exposure was given 48 hours after the first.  
17 These findings support the well established notion that repeated exposures to UV radiation  
18 results in adaptation (e.g., stimulation of melanogenesis). Kaidbey and Kligman (1981)  
19 examined individuals with skin types I, II, and III, and found that multiple exposures to  
20 subthreshold doses of UV radiation at 24-hour intervals resulted in cumulative injury to the skin,  
21 as indicated by a lowering of the minimal erythemal dose. These results suggest that a longer  
22 time period than 24 hours may be necessary to repair damage from a single exposure to UV  
23 radiation. Henriksen et al. (2004) also observed a lowering of the minimal erythemal dose with  
24 repeated exposure at 24-hour intervals in 49 healthy volunteers with skin types II, III, IV, and V.  
25 However, adaptation was reached after the 4th consecutive exposure. Henriksen et al. further  
26 found that the change in threshold depended on skin type. After 4 days of repeated UV  
27 radiation, there was little change (10 to 20%) in the erythemal threshold dose with repeated  
28 exposure to UV radiation in the fair-skinned individuals. Among the darker-skinned individuals,  
29 the minimal erythemal dose was lowered by 40 to 50%. However, both the initial UV dose and  
30 the dose to erythema after four days of exposure was still higher in the dark-skinned persons.

1 A reference erythema action spectrum was adopted by the Commission Internationale de  
2 l'Eclairage (International Commission on Illumination, CIE) in 1987 (McKinlay and Diffey,  
3 1987). The CIE erythema action spectrum indicates that UV-B radiation is orders of magnitude  
4 more effective per unit dose than UV-A radiation.  
5

### 6 ***Risk of Erythema from Changes in Tropospheric O<sub>3</sub> Levels***

7 There is no literature examining the risk of erythema associated with changes specifically  
8 in tropospheric or ground-level O<sub>3</sub> levels. The scientific studies, available to date, focus on the  
9 effects of a reduction in stratospheric ozone. One such study has assessed the effects of  
10 stratospheric O<sub>3</sub> depletion on the risk of erythema (Longstreth et al., 1998). The analysis by  
11 Longstreth et al. (1998) concluded that the risk of erythema would not appreciably increase with  
12 depletion of the stratospheric O<sub>3</sub> layer. This is due to the powerful adaptation of the skin to  
13 different levels of UV radiation, as evidenced by its ability to cope with changes in UV radiation  
14 by season (van der Leun and de Gruijl, 1993). Gradual exposure to increasing UV radiation from  
15 the winter to summer leads to decreased sensitivity of the skin. In midlatitudes, the UV-B  
16 radiation in the summer is 10-fold greater than in the winter. In contrast, the steady depletion of  
17 the O<sub>3</sub> layer has been estimated to result in an approximately 20% increase in UV-B over 10 years  
18 (Longstreth et al., 1998). The comparatively small increase in UV radiation throughout the years,  
19 therefore, would not significantly increase the risk of erythema. Tropospheric O<sub>3</sub> constitutes no  
20 more than 10% of total atmospheric O<sub>3</sub>. Given that stratospheric O<sub>3</sub> depletion was unlikely to  
21 increase the risk of erythema, one could reasonably conclude that small changes in ground-level  
22 O<sub>3</sub> that take place with attainment of the O<sub>3</sub> NAAQS would also not result in increased risk.  
23

### 24 **10.2.3.2 Skin Cancer**

25 According to the American Academy of Dermatology, one in five Americans develop skin  
26 cancer during their lifetime. The three main forms of skin cancer include basal cell carcinoma  
27 and squamous cell carcinoma, which are both nonmelanoma skin cancers, and malignant  
28 melanoma. Nonmelanoma skin cancers constitute more than one-third of all cancers in the U.S.  
29 and ~90% of all skin cancers, with basal cell carcinoma being approximately four times as  
30 common as squamous cell carcinoma (Diepgen and Mahler, 2002; ICNIRP, 2004). The  
31 incidence of malignant melanoma is much lower than nonmelanoma skin cancers. In 2004 more

1 than one million cases of basal and squamous cell skin cancer are expected to be newly  
2 diagnosed, compared to 40,780 cases of melanoma (Jemal et al., 2004). However, melanoma  
3 has great metastatic potential and accounts for the majority of skin cancer deaths.

4 Exposure to UV radiation is considered to be a major risk factor for all three forms of skin  
5 cancer (Gloster and Brodland,1996; Diepgen and Mahler, 2002; IARC, 1992). Ultraviolet  
6 radiation is especially effective in inducing genetic mutations and acts as both a tumor initiator  
7 and promoter. Keratinocytes have evolved DNA repair mechanisms to correct the damage  
8 induced by UV; however, mutations can occur, leading to skin cancers that are appearing with  
9 increasing frequency (Hildesheim and Fornace, 2004). The relationship between skin cancer and  
10 chronic exposure to UV radiation is further explored below, followed by discussion of the  
11 influence of O<sub>3</sub> on the incidence of skin cancer.

### 13 **10.2.3.3 Ultraviolet Radiation Exposure and the Incidence of Nonmelanoma Skin Cancers**

14 The incidence of all three types of cancers has been shown to rise with increasing UV  
15 radiation concentrations across the U.S. (de Gruijl, 1999); however, the most convincing  
16 evidence for a causal relationship exists between UV radiation and squamous cell carcinoma.  
17 Squamous cell carcinoma occurs almost exclusively on skin that is regularly exposed to the sun,  
18 such as the face, neck, arms, and hands. The incidence is higher among whites in areas of lower  
19 latitudes, where solar flux is greater (Krickler et al., 1994). The risk of squamous cell carcinoma  
20 was shown to increase with life-long accumulated exposure to UV radiation in one cross-  
21 sectional study (Vitasa et al., 1990); however, increased risk was found to be associated only  
22 with exposure 10 years prior to diagnosis in a case-control study (Gallagher et al., 1995a). One  
23 of the major concerns with both types of studies is the potential for recall bias in reporting past  
24 UV radiation exposure by individuals already aware of their disease status.

25 Ultraviolet radiation also has been linked to basal cell carcinoma. Basal cell carcinoma is  
26 common on the face and neck (80-90%) but rarely occurs on the back of the hands (de Gruijl,  
27 1999). While cumulative UV radiation exposure was not associated with an increased risk of  
28 basal cell carcinoma (Vitasa et al., 1990), increased risk was observed in individuals with greater  
29 recreational UV radiation exposure in adolescence and childhood (age <19 years) and  
30 individuals with a history of severe erythema in childhood (Gallagher et al., 1995b). Once again,  
31 consideration must be given to potential recall bias in assessing these results. Thus, there is

1 suggestive evidence that UV radiation also plays a role in the development of basal cell  
2 carcinoma, but the etiologic mechanisms for squamous cell carcinoma and basal cell carcinoma  
3 likely differ. In an Australian study conducted in a subtropical community, the factors of having  
4 fair skin, a history of repeated sunburns, and nonmalignant solar skin damage diagnosed by  
5 dermatologists were strongly associated with both types of nonmelanoma skin cancer (Green  
6 et al., 1996). The authors attributed the finding that outdoor occupation was not associated with  
7 nonmelanoma skin cancer to self-selection. Individuals with fair or medium complexions and a  
8 tendency to sunburn accounted for more than 80% of the community study sample; however,  
9 they were systematically underrepresented among outdoor workers (Green et al., 1996). Such  
10 self-selection bias might partly explain the lack of consistent quantitative evidence of a causal  
11 link between UV radiation and skin cancer in humans.

12 De Gruijl et al. (1993) assessed the action spectrum for nonmelanoma skin cancers using  
13 hairless albino mice. Human data are not available regarding wavelength dependence of the  
14 carcinogenicity of UV radiation. After adjusting for species differences, the Skin Cancer  
15 Utrecht-Philadelphia action spectrum indicated the highest effectiveness in the UV-B range with  
16 a maximum at 293 nm, which dropped to  $10^{-4}$  of this maximum at the UV-A range above  
17 340 nm (de Gruijl et al., 1993). The mutations commonly present in the *p53* tumor suppressor  
18 gene in individuals with squamous cell carcinoma and basal cell carcinoma are called the  
19 “signature” mutations of UV-B radiation (de Gruijl, 2002). UV-B radiation is highly mutagenic,  
20 because DNA is a chromophore for UV-B, but not for UV-A radiation (Ichihashi et al., 2003).  
21 Nevertheless, other studies have found that UV-A radiation, in addition to UV-B radiation, can  
22 induce DNA damage (Persson et al, 2002; Runger et al., 2000). DNA damage by UV-A is  
23 mediated by reactive oxygen species, making it indistinguishable from damage caused by other  
24 agents that generate reactive oxygen species (de Gruijl, 2002). Epidemiologic evidence of a  
25 carcinogenic effect of UV-A was found in a study of psoriasis patients receiving oral psoralen  
26 and UV-A radiation treatment (Stern et al., 1998). High-dose exposure to oral psoralen and  
27 UV-A radiation was associated with a persistent, dose-related increase in the risk of squamous  
28 cell cancer. Risk of basal cell cancer also was increased in those patients exposed to very high  
29 levels of UV-A radiation. Therefore, although UV-B radiation has long been considered the  
30 main culprit for nonmelanoma skin cancer, limited evidence suggest that UV-A radiation may  
31 also play a role.

1           Susceptible populations for nonmelanoma skin cancers include individuals with reduced  
2 capacity for nucleotide excision repair, the primary repair mechanism for UV radiation-induced  
3 DNA lesions (Ichihashi et al., 2003). At particular risk are individuals with xeroderma  
4 pigmentosum, as they have defective nucleotide excision repair in all tissues (Kraemer, 1997;  
5 Sarasin, 1999). Skin type also largely affects susceptibility to skin cancer. Of the six skin  
6 phenotypes, the most sensitive individuals are those with skin types I and II, who have a fair  
7 complexion, blue or green eyes, and red or blond hair (Diepgen and Mahler, 2002). These  
8 individuals tend to sunburn easily, tan poorly, and freckle with sun exposure. A history of  
9 repeated sunburns also appears to increase the risk of both cancers, while sunburns during  
10 childhood are more associated with increased basal cell carcinoma (Gallagher et al., 1995b;  
11 Green et al., 1996).

### 13 ***Ultraviolet Radiation and the Incidence of Cutaneous Malignant Melanoma***

14           From 1973 to 1994, the incidence rate of melanoma increased 120.5% along with an  
15 increased mortality rate of 38.9% among whites in the U.S. (Hall et al., 1999). The ICNIRP  
16 (2004) states that during the past 40 years or so, each decade has seen a twofold increase in the  
17 incidence of malignant melanoma in white populations, with increased incidence observed more  
18 prominently in individuals living in lower latitudes. Cutaneous malignant melanoma has a  
19 multifactorial etiology with environmental, genetic, and host factors (Lens and Dawes, 2004).  
20 The major environmental factor of malignant melanoma has been identified as UV radiation  
21 exposure (Diepgen and Mahler, 2002); therefore, the increased incidence of melanoma  
22 throughout the years might be partially attributable to changes in human activity patterns (e.g.,  
23 increased outdoor activity) that influence UV exposure or increased UV radiation at the ground  
24 level. The risk of melanoma appears to depend on the interaction between the nature of the  
25 exposure and skin type (Lens and Dawes, 2004).

26           Fears et al. (2002) examined the association between invasive cutaneous melanoma and  
27 UV radiation in non-Hispanic whites using a case-control study design. Lifetime residential  
28 history was coupled with mid-range UV-B radiation flux measurements to reduce exposure  
29 misclassification and recall bias. A 10% increase in the average annual UV-B flux was  
30 significantly associated with a 19% (95% CI: 5, 35) increase in individual odds for melanoma in  
31 men and a 16% (95% CI: 2, 32) increase in women. Whiteman et al. (2001) conducted a

1 systematic review of studies that examined the association between childhood UV radiation  
2 exposure and risk of melanoma. Researchers found that ecological studies assessing ambient  
3 sun exposure consistently reported higher risks of melanoma among people who resided in an  
4 environment with high UV radiation during their childhood (Whiteman et al., 2001). The lack of  
5 consistency among the case-control studies was likely due to the varying methods used to assess  
6 UV radiation dose.

7 While the evidence is generally suggestive of a causal relationship between UV radiation  
8 and malignant melanoma, possibly conflicting data also has been observed. For example, the  
9 highest occurrence of malignant melanoma is on men's backs and women's legs, areas that do  
10 not have prolonged exposure to the sun (Rivers, 2004). This indicates that, unlike nonmelanoma  
11 skin cancers, malignant melanoma tends to occur in sites of intermittent, intense sun exposure  
12 (trunk and legs), rather than in areas of cumulative sun damage (head, neck, and arms) (Swetter,  
13 2003). A study by Whiteman et al. (2003) observed that individuals with melanomas of the  
14 trunk had more melanocyte nevi and less solar keratoses compared to individuals with head and  
15 neck melanomas and, suggesting that cutaneous melanomas may arise through two pathways,  
16 one associated with melanocyte proliferation and the other with chronic exposure to sunlight.  
17 Green et al. (1999) also found that melanomas of the soles and palms resembled other cutaneous  
18 melanomas in their association with sun exposure, but were distinguished from them by their  
19 strong positive associations with nevi on the soles.

20 The available data conflict with regard to the relative importance of UV-A versus UV-B in  
21 inducing melanomas. UV-A has a much higher flux rate at the Earth's surface, as it is not  
22 absorbed by O<sub>3</sub> and it is able to penetrate more deeply into the skin surface due to its longer  
23 wavelength. However, UV-B, as mentioned earlier, is much more energetic and, therefore, more  
24 effective in photochemically altering DNA. The individual roles of UV-A and UV-B in the  
25 development of cutaneous malignant melanoma have been examined in several studies.

26 A case-control study of 571 patients and 913 matched controls found an elevated odds ratio of  
27 1.8 (95% CI: 1.2, 2.7), after adjusting for skin type, hair color, raised nevi, and number of  
28 sunburns, for developing malignant melanoma in individuals who regularly used tanning beds,  
29 which typically are UV-A sources (Westerdahl et al., 2000). In a study by Setlow et al. (1993),  
30 an action spectrum using the tropical fish *Xiphophorus* indicated that UV-A range wavelengths  
31 were especially important in malignant melanoma induction. However, an action spectrum



1 using the opossum *Monodelphis domestica* found that the potency of UV-A for melanoma  
2 induction was extremely low compared to that of UV-B (Robinson et al., 2000). A recent study  
3 by De Fabo et al. (2004) examined the differences in wavelength effectiveness using a  
4 hepatocyte growth factor/scatter factor-transgenic mouse model. The epidermal tissue of these  
5 transgenic mice behaves similar to the human epidermis in response to UV exposure. Given the  
6 absence of a mammalian melanoma action spectrum, the standardized CIE erythema action  
7 spectrum was used to deliver identical erythemally effective doses. Only UV-B radiation was  
8 found to initiate mammalian cutaneous malignant melanoma. UV-A radiation, even at doses  
9 considered physiologically relevant, were ineffective at inducing melanoma (De Fabo et al.,  
10 2004). Overall, current evidence suggests that UV-B, and not UV-A, is the primary risk factor  
11 for malignant melanoma (ICNIRP, 2004).

12 The populations susceptible for malignant melanoma are similar to those for nonmelanoma  
13 skin cancers. Once again, individuals with xeroderma pigmentosum or a reduced capacity for  
14 nucleotide excision repair are at increased risk (Tomescu et al., 2001; Wei et al., 2003).  
15 Individuals with skin types I and II, or the fair-skin phenotype (blue or green eyes; blond or red  
16 hair; skin that freckles, sunburns easily, and does not tan), have increased susceptibility to  
17 malignant melanoma (Evans et al., 1988; Swetter, 2003; Veierød et al., 2003). However, the  
18 incidence of melanoma was also positively associated with UV radiation in Hispanics and blacks  
19 (Hu et al., 2004). Although the incidence of melanoma is much lower in Hispanics and blacks  
20 compared to whites, melanomas in these populations are more likely to metastasize and have a  
21 poorer prognosis (Black et al., 1987; Bellows et al., 2001). Among children, malignant  
22 melanoma appears to have similar epidemiologic characteristics to the adult form of the disease  
23 (Whiteman et al., 1997). Individuals with intermittent, intense sun exposure, particularly during  
24 childhood, were found to have increased risk of melanoma (Whiteman et al., 2001), in contrast  
25 to the association between cumulative exposure and increased risk of squamous cell carcinoma.  
26 One study found that a personal history of nonmelanoma skin cancer or precancer, higher  
27 socioeconomic status, and increased numbers of nevocytic nevi also were associated with  
28 increased incidence of melanoma (Evans et al., 1988).

## *Effect of Changes in Tropospheric O<sub>3</sub> Levels on Skin Cancer Incidence*

The current evidence strongly suggests a causal link between exposure to UV radiation and the incidence of both nonmelanoma and melanoma skin cancer. Genetic factors, including skin phenotype and ability to repair DNA, affect an individual's susceptibility to skin cancer. Quantifying the relationship between UV radiation and skin cancer is complicated by the uncertainties involved in the selection of an action spectrum and appropriate characterization of dose (e.g., peak or cumulative levels of exposure, childhood or lifetime exposures). In addition, there are multiple complexities in attempting to quantify the effect of tropospheric O<sub>3</sub> levels on UV-radiation exposure, as described in Section 10.2. The absence of published studies that critically examine increased incidence of skin cancer attributable to decreased tropospheric O<sub>3</sub> levels reflects the significant challenges in determining ground-level O<sub>3</sub>-related changes in UV radiation exposure. An analysis by Lutter and Wolz (1997) attempted to examine the effects of a nationwide 10 ppb reduction in seasonal average tropospheric O<sub>3</sub> on the incidence of nonmelanoma and melanoma skin cancers and cataracts. Their estimate, however, depended upon several simplifying assumptions, ranging from an assumed generalized 10 ppb reduction in O<sub>3</sub> column density, national annual average incidence rates for the two types of skin cancer, and simple, linear biological amplification factors. Further, the methodologies used in this analysis inherently have ignored area-specific factors that are important in estimating the extent to which small, variable changes in ground-level O<sub>3</sub> mediate long-term exposures to UV-B radiation. More reasonable estimates of the human health impacts of enhanced UV-B penetration following reduced surface O<sub>3</sub> concentrations require both a solid understanding of the multiple factors that define the extent of human exposure to UV-B at present, and well-defined and quantifiable links between human disease and UV-B exposure. The reader is referred to the U.S. EPA 2002 Final Response to Court Remand (Federal Register, 2003) for detailed discussions of the data and scientific issues associated with the determination of public health benefits resulting from the attenuation of UV-B by surface-level O<sub>3</sub>.

In the absence of studies specifically addressing the reduction of tropospheric O<sub>3</sub> (by assuming that the key variable is total column O<sub>3</sub> density), inferences could be made concerning the effects of reduced tropospheric O<sub>3</sub>-related increases in UV-B exposure on the basis of studies focused on stratospheric O<sub>3</sub> depletion. Several studies have examined the potential effect of stratospheric O<sub>3</sub> depletion on the incidence of skin cancer (de Gruijl, 1995; Longstreth et al.,

1 1995; Madronich and de Gruijl, 1993; Slaper et al., 1996; Urbach, 1997). Note that several of  
2 the concerns expressed for Lutter and Wolz (1997) are relevant here as well. Stratospheric O<sub>3</sub>  
3 depletion is likely to increase the ground-level UV-B flux, as O<sub>3</sub> absorbs radiation in that  
4 wavelength range with high efficiency. Because UV-B radiation is primarily implicated in the  
5 induction of skin cancer, especially among persons with skin phenotypes I and II, there is  
6 concern that the depletion of the O<sub>3</sub> layer would result in significantly increased incidence of  
7 skin cancers.

8 Estimation of the increased risk in melanoma associated with stratospheric O<sub>3</sub> depletion  
9 cannot be done adequately due to the lack of a mammalian action spectrum for melanoma.  
10 Furthermore, the complexity of the UV-related induction mechanism of melanoma adds an  
11 additional layer of uncertainty to the calculations. The excess risk in nonmelanoma skin cancers  
12 associated with a decrease in stratospheric O<sub>3</sub> was estimated using the Skin Cancer Utrecht-  
13 Philadelphia action spectrum based on hairless albino mice (Longstreth et al., 1995).

14 Quantification of how much more UV radiation would reach ground level with each percentage  
15 decrease in O<sub>3</sub> required several assumptions: (1) annual doses were an appropriate measure; (2)  
16 personal doses were proportional to ambient doses; and, most notably, (3) each percentage  
17 decrease in O<sub>3</sub> was associated with a 1.2% increase in UV radiation. Next, the relationship  
18 between UV radiation and nonmelanoma skin cancer incidence was determined: each percent  
19 increase in annual UV radiation dose was estimated to cause a 2.5% increase in squamous cell  
20 carcinoma and 1.4% increase in basal cell carcinoma over a human lifetime. Incorporating all  
21 these factors, Longstreth et al. (1995) calculated that a sustained 10% decrease in stratospheric  
22 O<sub>3</sub> concentration would result in 250,000 additional nonmelanoma skin cancer cases per year.  
23 Madronich and de Gruijl (1993) noted that the largest percent of O<sub>3</sub>-induced nonmelanoma skin  
24 cancer increases would be at high latitudes, where baseline incidence of skin cancer is usually  
25 small. Assuming a phaseout of primary O<sub>3</sub>-depleting substances by 1996, as established by the  
26 Copenhagen Amendments in 1992, Slaper et al. (1996) estimated that the number of excess  
27 nonmelanoma skin cancers in the U.S. caused by O<sub>3</sub> depletion would exceed 33,000 per year (or  
28 approximately 7 per 100,000) around the year 2050.

29 However, estimating the increase in nonmelanoma skin cancer incidence attributable to the  
30 depletion of the stratospheric O<sub>3</sub> layer is marred by uncertainty. The following statement by

1 Madronich and de Gruijl (1994) describes the uncertainty of estimating the effect of  
2 stratospheric O<sub>3</sub> depletion on the incidence of skin cancer:

3  
4 Extrapolating trends and effects of UV into the future is very hypothetical due  
5 to uncertainties that arise from atmospheric chemistry, epidemiology, and related  
6 disciplines. The values that we calculated are one plausible measure of the  
7 magnitude of the O<sub>3</sub>-UV effects....The timescales for atmospheric change and  
8 skin-cancer development are still far from certain: O<sub>3</sub> reductions are expected  
9 to continue well into next century, and the time between UV exposure and  
10 development of skin cancer is essentially unknown.  
11

12 Therefore, much caution is necessary when assessing and interpreting the quantitative results of  
13 excess nonmelanoma skin cancer incidence due to stratospheric O<sub>3</sub> depletion. Although the  
14 effect of reductions in tropospheric or ground-level O<sub>3</sub> concentrations on skin cancer incidence  
15 has not been assessed, it would be expected to be much less compared to the effect from the  
16 depletion of the stratospheric O<sub>3</sub> layer, given that tropospheric O<sub>3</sub> makes up  $\leq 10\%$  of the total  
17 atmospheric O<sub>3</sub>.

#### 18 19 **10.2.3.4 Ocular Effects of Ultraviolet Radiation Exposure**

##### 20 *Ultraviolet Radiation Exposure and Risk of Ocular Damage*

21 Ocular damage from UV radiation exposures includes effects on the cornea, lens, iris, and  
22 associated epithelial and conjunctival tissues. Absorption of UV radiation differs by  
23 wavelength, with short wavelengths (<300 nm) being almost completely absorbed by the cornea,  
24 whereas longer wavelengths are transmitted through the cornea and absorbed by the lens  
25 (McCarty and Taylor, 2002). The most common acute ocular effect of environmental UV  
26 exposure is photokeratitis, also known as snowblindness, caused by absorption of short  
27 wavelength UV radiation by the cornea. The action spectrum indicated that maximum  
28 sensitivity of the human eye was found to occur at 270 nm (ICNIRP, 2004; Pitts, 1993). The  
29 threshold for photokeratitis in humans varied from 4 to 14 mJ/cm<sup>2</sup> for wavelengths 220 to  
30 310 nm.

31 Exposure to longer wavelengths has been shown to cause both transient and permanent  
32 opacities of the lens, or cataracts. Extensive toxicologic and epidemiologic evidence supports  
33 the causal association between UV radiation and cataracts (Hockwin et al., 1999; McCarty and  
34 Taylor, 2002). Ultraviolet radiation-induced cataracts are hypothesized to be caused by

1 oxidative stress leading to increased reactive species in the lens, which then causes damage to  
2 lens DNA and cross-linking of proteins. Exposure time to low-dose UV radiation was found to  
3 strongly influence cataract formation (Ayala et al., 2000). An action spectrum determined using  
4 young female rats indicated that the rat lens was most sensitive to 300 nm, correcting for corneal  
5 transmittance (Merriam et al., 2000). Oriowo et al. (2001) examined the action spectrum for  
6 cataract formation using whole cultured lens from pigs. As pigs lens are similar in shape and  
7 size to the human lens, some inferences may be made. Results indicated that the 270 to 315 nm  
8 waveband was most effective in producing UV-induced cataracts in vitro. However, the  
9 threshold values varied widely within that range, from 0.02 J/cm<sup>2</sup> for 285 nm to 0.74 J/cm<sup>2</sup> for  
10 315 nm (Oriowo et al., 2001). At wavelengths >325 nm, the threshold levels were orders of  
11 magnitude larger, with a minimum threshold value of 18.7 J/cm<sup>2</sup>.

12 An epidemiologic study examined the effects of UV radiation on cataract formation in  
13 watermen (e.g., commercial fishermen, boat workers) who worked on Chesapeake Bay, MD  
14 (Taylor et al., 1988). Among the 838 individuals surveyed in this study, 111 had cortical  
15 cataracts and 229 had nuclear cataracts. Results indicated that UV-B radiation was significantly  
16 associated with cortical, but not nuclear, cataract formation. For a given age, a doubling of  
17 cumulative UV-B exposure was associated with a 60% excess risk (95% CI: 1, 164) of cortical  
18 cataracts. No association was observed between cataracts and UV-A radiation in this outdoor-  
19 working population.

### 21 ***Risk of Ocular Damage from Changes in Tropospheric Ozone Levels***

22 Cataracts are the most common cause of blindness in the world. McCarty et al. (2000)  
23 calculated that ocular UV radiation exposure accounted for 10% of the cortical cataracts in an  
24 Australian cohort of 4,744 individuals from both urban and rural areas. A study by Javitt and  
25 Taylor (1994-1995) found that the probability of cataract surgery in the U.S. increased by 3% for  
26 each 1° decrease in latitude. These results suggest that depletion of the stratospheric O<sub>3</sub> layer  
27 may increase UV radiation-induced cataract formation. After assuming a certain wavelength  
28 dependency along with several additional assumptions, every 1% decrease in the stratospheric O<sub>3</sub>  
29 layer was estimated to be associated with a 0.3 to 0.6% increase in cataracts (Longstreth et al.,  
30 1995). Longstreth et al. (1995) noted that this estimate has a high degree of uncertainty due to  
31 inadequate information on the action spectrum and dose-response relationships. Quantitative

1 estimates have not been possible for photokeratitis, pterygium, or other UV-related ocular effects  
2 due to lack of epidemiologic and experimental data.

3 As is the case for all of the other UV-related health outcomes, there is no published  
4 information on the potential effects on cataract formation due to any changes in surface-level  
5 UV flux resulting from decreases in tropospheric O<sub>3</sub>.

### 7 **10.2.3.5 Ultraviolet Radiation and Immune System Suppression**

8 Experimental studies have suggested that exposure to UV radiation may suppress local and  
9 systemic immune responses to a variety of antigens (Clydesdale et al., 2001; Garssen and van  
10 Loveren, 2001; Selgrade et al., 1997). In rodent models, these effects have been shown to  
11 worsen the course and outcome of some infectious diseases and cancers (Granstein and Matsui,  
12 2004; Norval et al., 1999). Granstein and Matsui (2004) stated that exposure to UV-B radiation  
13 caused immunosuppression in mice ultimately by releasing cytokines that prevent antigen-  
14 presenting cells from performing their normal functions and causing direct damage to epidermal  
15 Langerhans cells. Noonan et al. (2003) investigated UV skin cancer induction in two strains of  
16 reciprocal F1 hybrid mice and found that genetically determined differences in susceptibility to  
17 UV-induced immunosuppression was a risk factor for skin cancer. At high UV radiation doses,  
18 mice with greater susceptibility to immune suppression had a larger proportion of skin tumors  
19 compared to those with lower susceptibility (Noonan et al., 2003). In a study by Yoshikawa  
20 et al. (1990), development of contact hypersensitivity to dinitrochlorobenzene on irradiated  
21 buttock skin was examined. Individuals who failed to develop contact hypersensitivity were  
22 considered to be susceptible to UV-B radiation. Virtually all skin cancer patients (92%) were  
23 susceptible to UV-B radiation-induced suppression of contact hypersensitivity, compared to  
24 approximately 40% of healthy volunteers. Others studies have observed increased skin cancer in  
25 immune suppressed organ transplant patients (Caforio et al., 2000; Lindelöf et al., 2000).  
26 Collectively, results from these studies suggest that immune suppression induced by UV  
27 radiation may be a risk factor for skin cancer induction (Ullrich, 2005).

28 There is also some evidence that UV radiation has indirect involvement in viral  
29 oncogenesis through the human papillomavirus (Pfister, 2003). Additional evidence of  
30 UV-related immunosuppression comes from an epidemiologic study of 919 patients with rare  
31 autoimmune muscle diseases from 15 cities on four continents with variable UV radiation

1 intensity (Okada et al., 2003). Ultraviolet radiation was strongly associated with the prevalence  
2 of dermatomyositis, an autoimmune disease distinguished by the presence of photosensitive  
3 pathognomonic rashes (Okada et al., 2003). In patients with the human immunodeficiency virus,  
4 UV-B radiation lead to activation of the virus in their skin through the release of cytoplasmic  
5 nuclear factor kappa B (Breuher-McHam et al., 2001). In a study by Selgrade et al. (2001),  
6 UV-induced immunosuppression was examined in 185 subjects with different skin  
7 pigmentations. To assess immune suppression, dinitrochlorobenzene was applied to irradiated  
8 buttock skin 72 hours after irradiation. Differences in sensitivity were unrelated to skin type  
9 based on the Fitzpatrick classification or minimal erythemal dose (Selgrade et al., 2001).  
10 However, erythemal reactivity, assessed by the steepness of the erythemal dose-response curve,  
11 was shown to be significantly associated with UV-induced immunosuppression. Only subjects  
12 with steep erythemal responses, which included individuals with skin types I through V, showed  
13 a dose-response relationship between UV exposure and immune suppression (Selgrade et al.,  
14 2001).

15 In other studies, UV radiation was associated with decreased autoimmune diseases.  
16 Several ecologic studies observed a decreased prevalence of multiple sclerosis, insulin-  
17 dependent diabetes mellitus, and rheumatoid arthritis in regions with lower latitude (i.e., higher  
18 UV radiation exposure) (Ponsonby et al., 2002). These results may be attributable to UV  
19 radiation-induced immunosuppression and UV-B-related production of vitamin D, which has  
20 immunomodulatory effects (Cantorna et al., 2000). The protective effects of UV radiation  
21 resulting from its active role in vitamin D production are further discussed in the next section.

22 Most action spectrum investigations have concluded that immunosuppression is caused  
23 most effectively by the UV-B waveband (Garssen and van Loveren, 2001). The effects of UV-A  
24 on local and systemic immunosuppression have been unclear and inconsistent. There is some  
25 evidence that high doses of UV-A is protective of immunosuppression induced by UV-B  
26 exposure (Halliday et al., 2004). Given the variety of outcomes of immune suppression and  
27 possible mechanisms of effect, little detailed information exists on UV radiation action  
28 spectrums and dose-response relationships. The available data are insufficient to conduct a  
29 critical risk assessment of UV radiation-induced immunosuppression in humans.  
30

### 10.2.3.6 Protective Effects of Ultraviolet Radiation – Production of Vitamin D

Any risk assessment that attempts to quantify the consequences of increased UV-B exposure on humans due to reduced ground-level O<sub>3</sub> must include consideration of both negative and positive effects. A potential health benefit of increased UV-B exposure relates to the production of vitamin D in humans. Most humans depend on sun exposure to satisfy their requirements for vitamin D (Holick, 2004). UV-B photons are absorbed by 7-dehydrocholesterol in the skin, leading to its transformation to previtamin D<sub>3</sub>, which is rapidly converted to vitamin D<sub>3</sub>. Vitamin D<sub>3</sub> is metabolized in the liver, then in the kidney to its biologically active form of 1.25-dihydroxyvitamin D<sub>3</sub>. One minimal erythemal dose produces vitamin D equivalent to an oral dose of 20,000 IU vitamin D, which is 100 times the recommended dietary allowance for adults under 50 years of age (Giovannucci, 2005; Holick, 2004).

Vitamin D deficiency can cause metabolic bone disease among children and adults, and also may increase the risk of many common chronic diseases, including type I diabetes mellitus and rheumatoid arthritis (Holick, 2004). Substantial in vitro and toxicologic evidence also support a role for vitamin D activity against the incidence or progression of various cancers (Giovannucci, 2005; Studzinski and Moore, 1995). Large geographical gradients in mortality rates for a number of cancers in the U.S. are not explained by dietary or other risk factors; therefore, it has been hypothesized that some carcinomas are due to insufficient UV-B radiation.

Published literature indicates that solar UV-B radiation, by increasing vitamin D production, is associated with a reduced risk of cancer. Most of these studies used an ecologic study design, in which latitude gradient was examined in relation to cancer rates. Kimlin and Schallhorn (2004) observed that latitude was a valid predictor of vitamin D-producing UV radiation. The strongest evidence exists for an association between UV radiation and reduced risk of colorectal cancer (Giovannucci, 2005; Grant and Garland, 2004; Freedman et al., 2002). Several other studies also have found an inverse relationship between UV radiation and various other cancers, including cancer of the breast (Freedman et al., 2002; Garland et al., 1990; Gorham et al., 1990; Grant, 2002a; John et al., 1999), ovary (Freedman et al., 2002; Lefkowitz and Garland, 1994), and prostate (Freedman et al., 2002; Hanchette and Schwartz, 1992), as well as non-Hodgkin lymphoma (Hughes et al., 2004; Hartge et al., 1996). Eight other cancers (i.e.,



1 bladder, esophageal, kidney, lung, pancreatic, rectal, stomach, and corpus uteri) have been found  
2 to exhibit an inverse correlation between mortality rates and UV-B radiation (Grant, 2002b).

3 Using UV-B data from July 1992 and U.S. cancer mortality rates from 1970 to 1994,  
4 premature cancer deaths attributable to insufficient UV-B exposure were analyzed in an ecologic  
5 study (Grant, 2002b). The minimum mortality rate, which was determined as the value  
6 corresponding to the maximum UV-B dose, was used to calculate the number of premature  
7 deaths. This analysis observed that the annual number of premature deaths from various cancers  
8 due to inadequate UV-B exposures was 21,700 (95% CI: 20,400, 23,400) for white Americans;  
9 1,400 (95% CI: 1,100, 1,600) for black Americans; and 500 (95% CI: 400, 600) for Asian  
10 Americans and other minorities. Uncertainty in the estimations of UV-B exposure limits the  
11 confidence for the estimates of excess cancer deaths attributable to insufficient exposure.  
12 Caution is required in interpreting results from ecologic data; however, no strong alternative  
13 explanation is indicated in the association observed between UV radiation and the decreased risk  
14 of cancer (Giovannucci, 2005). No study has assessed the decreased risk of cancer mortality  
15 resulting from increased UV radiation attributable to decreased tropospheric O<sub>3</sub> levels, but the  
16 change in risk is expected to be unappreciable.

17 In establishing guidelines on limits of exposure to UV radiation, the ICNIRP agreed that  
18 some low-level exposure to UV radiation has health benefits (ICNIRP, 2004). However, the  
19 adverse health effects of higher UV exposures necessitated the development of exposure limits  
20 for UV radiation. The ICNIRP recognized the challenge in establishing exposure limits that  
21 would achieve a realistic balance between beneficial and adverse health effects.

#### 22 23 **10.2.4 Summary and Conclusions for Ozone Effects on UV-B Flux**

24 Latitude and altitude are primary variables in defining UV-B flux at the Earth's surface,  
25 immediately followed in importance by clouds, surface albedo, particulate matter concentration  
26 and composition, and then by gas phase pollution. Of all of these, only latitude and altitude can  
27 be defined with small uncertainty in any effort to develop a UV climatology for use in a public  
28 health benefits analysis relevant to the areas not presently attaining the NAAQS for O<sub>3</sub>. Cloud  
29 cover, and its effect on surface UV flux, continues to be extremely difficult to define and predict.  
30 Particulate matter and gas-phase tropospheric pollutants are subject to similarly high degrees of  
31 uncertainty in predicting their relative concentration distributions. Land cover and,

1 consequently, surface albedos are highly variable at the geographic scales relevant to NAAQS  
2 attainment.

3         Within the uncertain context of presently available information on UV-B surface fluxes, a  
4 risk assessment of UV-B-related health effects would need to factor in human habits (e.g., daily  
5 activities, recreation, dress, and skin care) in order to adequately estimate UV-B exposure levels.  
6 Little is known about the impact of variability in these human factors on individual exposure to  
7 UV radiation. Furthermore, detailed information does not exist regarding the relevant type (e.g.,  
8 peak or cumulative) and time period (e.g., childhood, lifetime, or current) of exposure,  
9 wavelength dependency of biological responses, and interindividual variability in UV resistance.  
10 Recent reports of the necessity of UV-B in the production of vitamin D – a vitamin in which  
11 many individuals are deficient – suggests that increased risks of human disease due to a slight  
12 excess in UV-B radiation exposure may be offset by the benefits of enhanced vitamin D  
13 production. However, as with other impacts of UV-B on human health, this beneficial effect of  
14 UV-B has not been studied in sufficient detail to allow for a credible health benefits assessment.  
15 In conclusion, the effect of changes in surface-level O<sub>3</sub> concentrations on UV-induced health  
16 outcomes cannot yet be critically assessed within reasonable uncertainty.

### 19 **10.3 TROPOSPHERIC OZONE AND CLIMATE CHANGE**

20         Water vapor, CO<sub>2</sub>, O<sub>3</sub>, N<sub>2</sub>O, CH<sub>4</sub>, CFCs, and other polyatomic gases present in the Earth’s  
21 troposphere, trap infrared radiation emitted by the Earth’s surface, leading to surface warming.  
22 This phenomenon is widely known as the “Greenhouse Effect” (Arrhenius, 1896), and the gases  
23 involved are known as “greenhouse gases” (GHGs). The term used for the role a particular  
24 atmospheric component, or any other component of the greater climate system, plays in altering  
25 the Earth’s radiative balance is “forcing.” In the past decade, the global atmospheric sciences  
26 and climate communities have made significant progress in determining the specific role O<sub>3</sub>  
27 plays in forcing climate.

28         The Intergovernmental Panel on Climate Change (IPCC) was founded in 1988 by the  
29 World Meteorological Society (WMO) and the United Nations Environmental Program (UNEP)  
30 to support the work of the Conference of Parties (COP) to the United Nations Framework  
31 Convention on Climate Change (UNFCCC). Drawing from the global climate and atmospheric

1 sciences community for its authors and reviewers, the IPCC produces reports containing  
2 thorough assessments of the available peer-reviewed science regarding the physical climate  
3 system, past and present climate, and evidence of human-induced climate change. This section  
4 will summarize the reviews of the available information on the forcing properties of tropospheric  
5 O<sub>3</sub> as provided by the IPCC Third Assessment Report (IPCC, 2001a), and will also describe  
6 some of the more recent developments on the subject.

7 The projected effects of global climate change will be briefly explained to provide the  
8 context within which O<sub>3</sub> serves as a regional, and possibly global, anthropogenic pollutant.  
9 The concept of climate forcing is also explained, along with the factors that influence the extent  
10 of climate forcing by O<sub>3</sub>. The section concludes with a summary of the various estimates that  
11 have been placed on the amount of globally averaged forcing due to O<sub>3</sub>.

### 12 13 **10.3.1 The Projected Impacts of Global Climate Change**

14 The study of the atmospheric processes involved in global climate change, and its potential  
15 consequences for human health and global ecosystems, is an area of active research. The IPCC  
16 Third Assessment Report (TAR) is the most thorough evaluation available of the science  
17 concerning climate change. In addition to the first and second IPCC assessments in 1990 and  
18 1995, along with other IPCC reports, earlier assessments included those conducted by the UNEP  
19 (1986), the WMO (1988), the U.S. Environmental Protection Agency (1987), and others (e.g.,  
20 Patz et al., 2000a,b). The reader is referred to those documents for a complete discussion of  
21 climate change science. An abbreviated list of the IPCC conclusions to date and a short  
22 discussion of the potential impacts of climate change on human health and welfare is provided  
23 here to serve as the context for the discussion of the role of the increasing tropospheric O<sub>3</sub>  
24 concentration in climate change.

25 According to various historic and modern measurement records, atmospheric GHG  
26 concentrations have increased dramatically in the past century, and have been attributed to  
27 human activities. The IPCC TAR describes the scientific theory and evidence linking increases  
28 in GHGs to human activities (IPCC, 2001a).

29 An increasing body of geophysical observations shows that the Earth is warming and that  
30 other climate changes are underway. These observations include the global surface temperature  
31 record assembled since the year 1860, the satellite temperature record begun in 1979, recorded

1 changes in snow and ice cover since the 1950s, sea level measurements taken throughout the  
2 20th century, and sea surface temperature observations recorded since the 1950s.

3 Observations (Levitus et al., 2005) show that ~84% of the total heating of the Earth System  
4 (oceans, atmosphere, continents, and cryosphere) over the last 40 years has gone into warming  
5 the oceans. Barnett et al. (2005) have reported the emergence of a clear pattern of ocean surface  
6 warming associated with anthropogenic GHGs. The authors constructed a model-based  
7 fingerprint (i.e., a map of predicted changes in the vertical temperature profile of the Earth's six  
8 major oceans), and compared this map to the newly upgraded and expanded ocean temperature  
9 data set (Levitus et al., 2005). They concluded that the warming signal far exceeds what would  
10 be expected from natural variability, a finding that was in compelling agreement with GHG-  
11 forced model profiles. Other evidence of ocean warming includes a marked increase in the  
12 frequency, intensity, and persistence of the zonal atmospheric circulation shifts known as the El  
13 Niño-Southern Oscillation (ENSO) over the past 100 years. ENSO events occur when the  
14 tropical Pacific Ocean has accumulated a large, localized mass of warm water that interrupts  
15 cold surface currents along South America, altering precipitation and temperature patterns in the  
16 tropics, subtropics, and the midlatitudes.

17 IPCC (1998, 2001a) reports also describe the results of general circulation model (GCM)  
18 studies predicting that human activities will alter the climate system in a manner likely to lead to  
19 marked global and regional changes in temperature, precipitation, and other climate properties.  
20 These changes are expected to increase the global mean sea level as well as increase the number  
21 of extreme weather events such as floods and droughts, increased wind speeds and precipitation  
22 intensity of tropical cyclones, and changes in soil moisture. These predicted changes can be  
23 expected to directly impact human health, ecosystems, and global economic sectors (e.g.,  
24 hydrology and water resources, food and fiber production) (IPCC, 1998, 2001b). Table 10-1  
25 summarizes these projected impacts.

26 Wide variations in the course and net impacts of climate change in different geographic  
27 areas are expected. In general, the projected changes constitute additional stressors on natural  
28 ecosystems and human societal systems already impacted by increasing resource demands,  
29 unsustainable resource management practices, and pollution. Some of the predicted changes  
30 include alterations in ecological balances; in the availability of adequate food, water, clean air;

**Table 10-1. Examples of Impacts Resulting From Projected Changes in Extreme Climate Events**

<b>Projected changes during the 21st Century in Extreme Climate Phenomena and their Likelihood<sup>a</sup></b>	<b>Representative Examples of Projected Impacts<sup>b</sup> (all high confidence of occurrence in some areas<sup>c</sup>)</b>
<i>Simple Extremes</i>	
Higher maximum temperatures; more hot days and heat waves <sup>d</sup> over nearly all land areas ( <i>very likely</i> <sup>a</sup> )	<ul style="list-style-type: none"> <li>• Increased incidence of death and serious illness in older age groups and urban poor</li> <li>• Increased heat stress in livestock and wildlife</li> <li>• Shift in tourist destinations</li> <li>• Increased risk of damage to a number of crops</li> <li>• Increased electric cooling demand and reduced energy supply reliability</li> </ul>
Higher (increasing) minimum temperatures; fewer cold days, frost days, and cold waves <sup>d</sup> over nearly all land areas ( <i>very likely</i> <sup>a</sup> )	<ul style="list-style-type: none"> <li>• Decreased cold-related human morbidity and mortality</li> <li>• Decreased risk of damage to a number of crops, and increased risk to others</li> <li>• Extended range and activity of some pest and disease vectors</li> <li>• Reduced heating energy demand</li> </ul>
More intense precipitation events ( <i>very likely</i> <sup>a</sup> over many years)	<ul style="list-style-type: none"> <li>• Increased flood, landslide, avalanche, and mudslide damage</li> <li>• Increased soil erosion</li> <li>• Increased flood runoff could increase recharge of some floodplain aquifers</li> <li>• Increased pressure on government and private flood insurance systems and disaster relief</li> </ul>
<i>Complex Extremes</i>	
Increased summer drying over most midlatitude continental interiors and associated risk of drought ( <i>likely</i> <sup>a</sup> )	<ul style="list-style-type: none"> <li>• Decreased crop yields</li> <li>• Increased damage to building foundations caused by ground shrinkage</li> <li>• Decreased water resource quantity and quality</li> <li>• Increased risk of forest fire</li> </ul>
Increase in tropical cyclone peak wind intensities, mean and peak precipitation intensities ( <i>likely</i> <sup>a</sup> over some areas) <sup>e</sup>	<ul style="list-style-type: none"> <li>• Increased risk to human life, risk of infections, disease epidemics, and many other risks</li> <li>• Increased coastal erosion and damage to coastal buildings and infrastructure</li> <li>• Increased damage to coastal ecosystems such as coral reefs and mangroves</li> </ul>
Intensified droughts and floods associated with El Niño events in many different regions ( <i>likely</i> <sup>a</sup> ) (see also under droughts and intense precipitation events)	<ul style="list-style-type: none"> <li>• Decreased agricultural and rangeland productivity in drought- and flood-prone regions</li> <li>• Decreased hydropower potential in drought-prone regions</li> </ul>
Increased Asian summer monsoon precipitation variability ( <i>likely</i> <sup>a</sup> )	<ul style="list-style-type: none"> <li>• Increased flood and drought magnitude and damages in temperate and tropical Asia</li> </ul>

**Table 10-1 (cont'd). Examples of Impacts Resulting From Projected Changes in Extreme Climate Events**

Projected changes during the 21st Century in Extreme Climate Phenomena and their Likelihood <sup>a</sup>	Representative Examples of Projected Impacts <sup>b</sup> (all high confidence of occurrence in some areas <sup>c</sup> )
<i>Complex Extremes</i> (cont'd)	
Increased intensity of midlatitude storms (little agreement between current models) <sup>d</sup>	<ul style="list-style-type: none"> <li>• Increased risks to human life and health</li> <li>• Increased property and infrastructure losses</li> <li>• Increased damage to coastal ecosystems</li> </ul>

<sup>a</sup>Likelihood refers to judgmental estimates of confidence used by TAR WGI: *very likely* (90-99% chance); *likely* (66-90% chance). Unless otherwise stated, information on climate phenomena is taken from the Summary for Policymakers, TAR WGI. TAR WGI = Third Assessment Report of Working Group I (IPCC, 2001a).

<sup>b</sup>These impacts can be lessened by appropriate response measures.

<sup>c</sup>High confidence refers to probabilities between 67 and 95%.

<sup>d</sup>Information from TAR WGI, Technical Summary.

<sup>e</sup>Changes in regional distribution of tropical cyclones are possible but have not been established.

Source: IPCC (2001b).

1 and in human health and safety. Poorer nations can be expected to suffer the most, given their  
2 limited adaptive capabilities.

3 Although many regions are predicted to experience severe, possibly irreversible, adverse  
4 effects due to climate change, beneficial changes may also take place. For example, certain  
5 regions may benefit from warmer temperatures or increased CO<sub>2</sub> fertilization, e.g., U.S. West  
6 Coast coniferous forests, and some Western rangelands. Specific benefits may include reduced  
7 energy costs for heating, reduced road salting and snow-clearance costs, longer open-water  
8 seasons in northern channels and ports, and improved agricultural opportunities in the northern  
9 latitudes as well as in the Western interior and coastal areas. For further details about the  
10 projected effects of climate change on a U.S.-regional scale, the reader is also referred to several  
11 regionally-focused reports (MARAT, 2000; Yarnal et al., 2000; NERAG, 2001; GLRAG, 2000),  
12 as well as a report on potential impacts of climate change on human health (Bernard et al.,  
13 2001a,b). The IPCC report, “The Regional Impacts of Climate Change,” (IPCC, 1998) describes  
14 the projected effects of human-induced climate change on various regions of the globe including  
15 Africa, the Arctic and Antarctic, the Middle East and arid Asia, Australasia, Europe, Latin  
16 America, North America, the small island nations, temperate Asia, and tropical Asia.

1 While current climate models can successfully simulate the present-day annual mean  
2 global climate and the seasonal cycles on a continental scale, they have been less successful on a  
3 regional scale. Clouds and humidity, essential factors in defining local and regional (sub-grid  
4 scale) climate, are significantly uncertain (IPCC, 2001a). Due to modeling uncertainties, both in  
5 reproducing regional climate and in predicting the future economic activity that will govern  
6 future GHG emissions, the projected impacts discussed above are also uncertain.

7 Findings from the U.S. Global Change Research Program (USGCRP) (NAST, 2000) and  
8 related reports illustrate the considerable uncertainties and difficulties in projecting likely  
9 climate change impacts at the regional or local scale. The USGCRP findings also reflect the  
10 mixed nature of projected potential climate change impacts, i.e., combinations of deleterious and  
11 beneficial effects, for U.S. regions and the variation of projected impacts across different  
12 regions. Difficulties in projecting region-specific climate change impacts are complicated by the  
13 need to evaluate the potential effects of regional- or local-scale changes in key air pollutants not  
14 only on global-scale temperature trends, but also on regional- or local-scale temperature and  
15 precipitation patterns. The EPA is currently leading a research effort that uses regional-scale  
16 climate models with the goal of identifying changes to O<sub>3</sub> and PM concentrations that may occur  
17 in a warming climate. An assessment of the results of this effort will be available by the next  
18 review of the O<sub>3</sub> NAAQS. This focused effort to determine the impact of a warming climate on  
19 criteria air pollution requires regional-scale models with improved skill in reproducing climate  
20 history and predicting change. Among the innovations being employed in this effort is the  
21 downscaling of global circulation model outputs to provide boundary conditions for model  
22 calculations at the regional scale (Liang et al., 2005). While focusing on projecting the impact of  
23 a warming climate on regional O<sub>3</sub> concentrations, the effort applied to improving regional-scale  
24 modeling will also lead to improved estimates of current and projected future impacts of  
25 tropospheric O<sub>3</sub> on climate.

### 27 **10.3.2 Solar Energy Transformation and the Components of the Earth's** 28 **Climate System**

29 Mass, in any form, has the capacity to interact with solar and terrestrial radiation, but the  
30 manner in which it interacts with radiation is governed by its particular physical form and/or  
31 molecular properties. Water provides one of the most interesting examples of how physical form

1 affects radiative properties. In its gaseous form, water is the most important GHG present in the  
2 climate system, due to its ability to absorb long-wave terrestrial radiation. Conversely, in its  
3 frozen form as snow or sea ice, water plays a very important role in the climate system by  
4 scattering UV and visible solar radiation back to space, i.e., decreasing the Earth's net solar  
5 radiation receipts by increasing the Earth's reflective properties (albedo). In its liquid aerosol  
6 form as clouds, water also greatly increases the Earth's albedo. In its bulk liquid form as ocean  
7 water, it absorbs terrestrial radiation, and represents the Earth's most important reservoir of heat  
8 energy.

9 The atomic composition and molecular structure of a gas determines the wavelengths of  
10 light it can absorb and, therefore, its role in defining the heat capacity of the atmosphere. Ozone  
11 and O<sub>2</sub> provide examples of the relative importance of these molecular properties. While these  
12 molecules are both composed solely of oxygen atoms, their bond structures are distinct. Ozone  
13 has a three-atom, bent molecular structure, giving it the capacity to absorb terrestrial (infrared)  
14 wavelengths – making it a GHG. At any altitude, i.e., in the stratosphere or troposphere, O<sub>3</sub> has  
15 the capacity to absorb UV radiation of 320 nm and shorter, further increasing the energy-  
16 absorbing capacity of the troposphere. Conversely, O<sub>2</sub>, due to its diatomic, linear structure, is  
17 limited to absorbing very short-wave UV light – and does so at altitudes too high to influence the  
18 climate system significantly.

19 Each component of the climate system plays a role in absorbing, transforming, storing,  
20 dispersing, or scattering solar radiation. Weather is a tangible consequence of the transformation  
21 and dispersion of terrestrial radiation within the atmosphere. The term “weather” refers to the  
22 condition of the Earth's atmosphere at a specific time and place. It is defined by several specific  
23 variables: the air temperature, air pressure, humidity, clouds, precipitation, visibility, and wind  
24 speed. The “climate” for a given place on the Earth's surface is a long-term average of these  
25 variables accounting for daily and seasonal weather events. The frequency of extreme weather  
26 events is used to distinguish among climates that have similar averages (Ahrens, 1994).

27 Climate components, including GHGs, land, oceans, sea ice, land ice and snow,  
28 atmospheric particles, vegetation, clouds, etc., all contribute to the Earth's heat capacity, i.e., its  
29 ability to absorb and retain solar energy. Changes in the properties (or mass) of these  
30 components will “force” the climate system in one direction or the other, i.e., warmer versus  
31 cooler. The transformation of atmospheric O<sub>2</sub> into O<sub>3</sub> by way of air pollution chemistry,



1 enhances the heat capacity of the atmosphere. The principles behind the important concept of  
2 climate forcing are further described, below.

### 3 4 **10.3.3 The Composition of the Atmosphere and the Earth's** 5 **Radiative Equilibrium**

6 The Greenhouse Effect is the term given to the decreased rate of reemission of absorbed  
7 solar energy due to the heat-retaining properties of the Earth's atmosphere. According to simple  
8 radiative transfer theory, at thermal equilibrium, the Earth's temperature should be near  $-15\text{ }^{\circ}\text{C}$ .  
9 This is the temperature of a theoretical "black body" that is receiving and then reemitting  
10  $342.5\text{ Wm}^{-2}$ , i.e., the globally averaged amount of full-spectrum solar energy absorbed and then  
11 reemitted by the Earth as infrared terrestrial radiation per square meter. In fact, satellite  
12 observations well above the atmosphere indicate that the Earth's average *planetary* temperature  
13 is remarkably close to its theoretical black body value at  $-18\text{ }^{\circ}\text{C}$ , a temperature at which liquid  
14 water ordinarily does not exist.

15 At its *surface*, however, the Earth's average temperature is  $+15\text{ }^{\circ}\text{C}$ . The  $+33\text{ }^{\circ}\text{C}$   
16 temperature differential between the Earth's planetary and surface temperatures is due to the  
17 presence of infrared (IR) radiation-absorbing components in the atmosphere such as water  
18 vapor,  $\text{CO}_2$ ,  $\text{CH}_4$ , several other trace gases, and some types of particles and clouds.

19 The atmosphere, when cloud-free, is largely transparent in the solar wavelength range.  
20 A small fraction of this radiation is absorbed and reemitted as black body radiation by dark  
21 atmospheric particles (IPCC, 2001a). However, the majority of clouds and particles, in part,  
22 offset the greenhouse effect by increasing the Earth's albedo, thereby decreasing the overall  
23 amount of solar radiation absorbed by the Earth system.

24 Ozone,  $\text{SO}_2$  and  $\text{NO}_2$  also absorb ultraviolet and near ultraviolet wavelengths, in addition to  
25 infrared radiation. Once absorbed by a gas molecule, the energy introduced by a photon may  
26 induce a photochemical reaction with the residual energy thermally exciting (heating) the  
27 products of the reaction. Alternatively, the energy introduced into the molecule by the photon  
28 may be dispersed amongst neighboring molecules via intermolecular collisions, or reemitted in  
29 part as a lower energy (i.e., IR) photon.

30 Radiation from the sun or the Earth's surface that is absorbed by gases and particles is  
31 reemitted isotropically, i.e., it is equally likely to be emitted in all directions. Therefore, to a

1 first approximation, half of the radiation trapped by the Earth’s atmosphere is reflected back to  
2 its surface. A portion of this radiation is transformed into the heat energy that drives the  
3 atmospheric processes that form the basis of weather and climate. Radiation that is not absorbed  
4 by gases and aerosols reaches the Earth’s surface where it is scattered (reflected) or absorbed,  
5 depending on the surface albedo.

6 Successful modeling of the Earth’s climate and, therefore, the assessment of the extent of  
7 human-induced climate change and development of appropriate policy depend on the quality of  
8 available information on the relative efficiencies, amounts, and spatial and temporal distributions  
9 of the various radiatively active components of the atmosphere that absorb and/or reflect solar  
10 and terrestrial radiation, along with all the other nonatmospheric components of the Earth  
11 system.

### 13 **10.3.3.1 Forcing of the Earth’s Radiative Balance**

14 As mentioned earlier, the commonly used measure of the relative influence of a given  
15 component of the climate system on the Earth’s radiative balance is its radiative forcing (IPCC,  
16 2001a; Houghton et al., 1990). Radiative forcing, in  $Wm^{-2}$ , is a quantity that was developed by  
17 the climate modeling community as a first-order-only means of estimating relative effects of  
18 individual anthropogenic and natural processes on the energy balance within the climate system.

19 When the effect of a particular component of the climate system is to reduce the amount of  
20 solar energy absorbed, usually by increasing the Earth’s albedo, this component is said to  
21 provide a “negative” forcing, measured in  $Wm^{-2}$ . The convention assigns a positive value to the  
22 forcing induced by climate system components that enhance the Greenhouse Effect or otherwise  
23 act to increase the heat-absorbing capacity of the Earth system. Purely reflective atmospheric  
24 aerosol, clouds, white rooftops, snow-covered land surfaces, and dense sea ice provide a  
25 negative forcing, while highly absorbing dark-colored atmospheric aerosols, GHGs, and  
26 increases dark ocean surface area, due to the melting of sea ice sheets, positively force the  
27 climate system.

28 Global and regional climate are roughly defined by the balance between the large number  
29 of positive and negative forcings induced by the many different components of the Earth system  
30 and any changes in the properties of those components due to natural processes or anthropogenic  
31 activities. Following a perturbation or added forcing, such as an increase in GHG concentrations

1 or modification the Earth's albedo through changes in land use, this balance is re-established via  
2 a complex redistribution of energy within the Earth system. Feedback mechanisms that are  
3 theorized but difficult to resolve at the quantitative level further complicate the prediction of the  
4 sensitivity of variables, such as surface temperature, to changes in forcing.

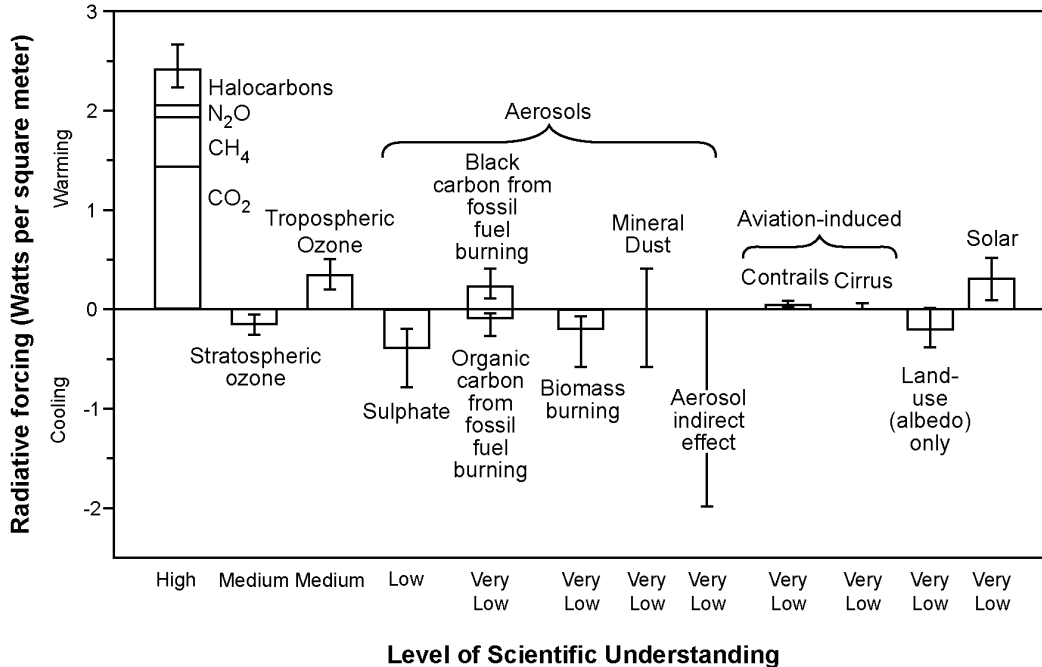
5 A simple example of a positive feedback would be melting sea ice. As sea ice melts with  
6 increasing ocean temperatures, the dark ocean surface that is revealed is more efficient at  
7 absorbing IR radiation, further increasing the rate of warming. A negative feedback would be  
8 the formation of clouds over a moist, warming surface. As clouds form, less radiation is  
9 available to warm the surface, leading to cooling. The role of feedbacks in determining the  
10 sensitivity of climate to changes in radiative forcing is described in detail in the IPCC TAR  
11 (IPCC, 2001a).

12 Discussions are presently underway within the climate community regarding a metric to  
13 replace forcing as the standard measure of climate impact – one that will account for more of the  
14 factors that determine the effectiveness of a specific change in altering climate. However,  
15 forcing remains the current standard (NRC, 2005).

16 The IPCC has reported estimated values for forcing by individual radiatively active gases,  
17 and by particle-phase components of the atmosphere that were derived primarily through expert  
18 judgment incorporating the results of peer-reviewed modeling studies. The forcing estimates,  
19 shown in Figure 10-6, are global averages attributed to known GHGs, including O<sub>3</sub>, particles,  
20 anthropogenic cirrus clouds, land-use change, and natural solar flux variations. Uncertainty  
21 ranges are assigned to reflect the range of modeled values reported in those studies. The current  
22 estimate of forcing due to long-lived, well-mixed, GHGs accumulated in the atmosphere from  
23 the preindustrial era (ca., 1750) through the year 2000 is  $+2.4 \text{ Wm}^{-2} \pm 10\%$  (IPCC, 2001a). An  
24 indication of the level of confidence in each of these estimates is given along the bottom of this  
25 figure, again reflecting the expert judgment of the IPCC.

26 The IPCC reported a global average forcing value of  $0.35 \pm 0.15 \text{ Wm}^{-2}$  for tropospheric O<sub>3</sub>,  
27 based on model calculations constrained by climatological observations. The considerations and  
28 studies used to estimate this value will be outlined below. Hansen and Sato (2001), accounting  
29 for uncertainties in pre-industrial emissions levels, more recently estimated a value of  $0.5 \pm 0.2$   
30  $\text{Wm}^{-2}$  for forcing by O<sub>3</sub>.

31



**Figure 10-6. Estimated global mean radiative forcing exerted by gas and various particle phase species for the year 2000, relative to 1750.**

Source: IPCC (2001a).

### 10.3.4 Factors Affecting the Magnitude of Climate Forcing by Ozone

The radiative properties of O<sub>3</sub> are distinct from those of other important GHGs in that it is capable of absorbing both UV and IR radiation. Furthermore, it is able to absorb long-wave radiation in a portion of the IR spectrum where water vapor does not absorb, i.e., the 9 to 10 μm wavelength range, meaning that its ability to trap heat and force climate are unchanged by variations in humidity. Given its relatively short atmospheric lifetime in comparison to other GHGs, the distribution of tropospheric O<sub>3</sub> is highly variable in geographic extent and time. These properties enhance the prospect of attributing a unique geographic and time-dependent pattern or fingerprint to forcing by O<sub>3</sub>.

Due to human activities, tropospheric O<sub>3</sub> is estimated to have provided the third largest increase in direct forcing since preindustrial times. It may also play a role in indirect forcing through its participation in the oxidative removal of other radiatively active trace gases, such as CH<sub>4</sub> and the HCFCs. The direct and indirect forcing that tropospheric O<sub>3</sub> imposes on the climate

1 system depends upon its geospatial and temporal distribution, but it also depends upon the  
2 albedo of the underlying surface and its vertical position (altitude) in the atmosphere.

#### 3 4 **10.3.4.1 The Global Burden of Tropospheric Ozone**

5 Little historical data exist that may be used to estimate the global O<sub>3</sub> burden prior to  
6 industrialization, although a few late 19th-century measurements suggest that O<sub>3</sub> has more than  
7 doubled in Europe during the 20th century. The insufficient data record on preindustrial  
8 tropospheric O<sub>3</sub> distributions introduces a major uncertainty in the estimation of the change  
9 in O<sub>3</sub>-induced forcing since that period (IPCC, 2001a; Mickley et al., 2004a; Mickley et al.,  
10 2001; Shindell and Faluvegi, 2002).

11 Ozone reacts photochemically at time scales that are generally shorter than those for large-  
12 scale mixing processes in the atmosphere. Concentrated O<sub>3</sub> plumes evolve downwind of strong  
13 sources of its precursor pollutants: reactive nitrogen, CO, and non-methane hydrocarbons  
14 (NMHCs). The most important of these sources are midlatitude industrialized areas and tropical  
15 biomass burning. When viewed from above the atmosphere by satellite-borne spectrometers, O<sub>3</sub>  
16 enhancements appear as relatively localized air masses or regional-scale plumes, usually  
17 originating from industrialized areas or areas in which active biomass burning is underway. The  
18 IPCC (2001a) describes the efforts of several research teams who have analyzed data supplied by  
19 the satellite-borne Total Ozone Mapping Spectrometer (TOMS) and other remote-sensing  
20 instruments to map the global distribution of tropospheric O<sub>3</sub> and to attempt to identify processes  
21 that influence the global tropospheric O<sub>3</sub> budget (IPCC, 2001a). More recently, coincident  
22 observations of total O<sub>3</sub> by TOMS and the Solar Backscattered UV (SBUV) instrument were  
23 used by Fishman et al. (2003) to construct well-resolved spatial and temporal maps of the  
24 regional distribution of tropospheric O<sub>3</sub>. Their results were consistent with those reported by  
25 others, but with higher regional-scale resolution. They reported large O<sub>3</sub> enhancements in the  
26 southern tropics in austral spring, and in the northern temperate latitudes in the summer. The  
27 regional nature of high O<sub>3</sub> concentrations was clearly visible in northeastern India, the eastern  
28 United States, eastern China, and west and southern Africa, each coincident with high population  
29 densities. Fishman et al. (2003) noted, as have the other groups cited above, significant  
30 interannual variability in the concentrations observed over these regions. *In situ* measurements  
31 of tropospheric O<sub>3</sub> concentrations range from 10 ppb over remote oceans to 100 ppb in both the

1 upper troposphere and in plumes downwind from polluted metropolitan regions (IPPC, 2001a).  
2 Ground-level concentrations in urban areas are often >100 ppb. In the Southern Hemisphere,  
3 one of the largest sources of O<sub>3</sub> precursors is biomass burning. Biomass burning elevates O<sub>3</sub> on  
4 large spatial scales, particularly in the tropical Atlantic west of the coast of Africa and in  
5 Indonesia.

6 In its third assessment report, the IPCC estimates placed the global burden of tropospheric  
7 O<sub>3</sub> at a highly uncertain 370 Tg, equivalent to an average column density of 34 Dobson Units (1  
8 DU = 2.687 × 10<sup>16</sup> molecules/cm<sup>-2</sup>) or a mean concentration of 50 ppb (IPCC, 2001a).

9 Accounting for differences in levels of industrialization between the hemispheres, the average  
10 column burden in the Northern Hemisphere is estimated to be 36 DU, with the Southern  
11 Hemisphere estimated to average 32 DU. Due to its rapid photochemistry, individual surface  
12 measurements of tropospheric O<sub>3</sub> cannot capture large-scale concentrations, nor will they  
13 represent the higher altitude concentrations. Dense surface and vertical measurements  
14 (ozonesondes) would be required to supplement available output from remote sensing  
15 instruments to provide the complete set of observations needed to derive a credible global O<sub>3</sub>  
16 budget. Such a measurement program appears, at present, to be impractical.

#### 18 **10.3.4.2 Background Concentrations versus Regionally-Oriented Ozone Enhancements**

19 Vingarzan (2004) surveyed the air quality literature and reported that annual average  
20 background O<sub>3</sub> concentrations at ground level in the Northern Hemisphere appear to range  
21 between 20 and 45 ppb, depending upon geographic location, elevation, and the influence of  
22 local sources. Fiore et al. (2003) modeled the U.S. continental O<sub>3</sub> concentrations and found that  
23 surface background levels overlap the lower end of the range reported by Vingarzan (2004), e.g.,  
24 15 to 35 ppb, with higher levels (40 to 50 ppb) arising at high-elevation sites due to the influence  
25 of the upper troposphere (See Chapter 3 and its associated annexes for a complete discussion of  
26 “policy relevant background [PRB]). Local- and regional-scale enhancements in O<sub>3</sub> may be  
27 thought of as roughly superimposed upon these background levels, with the exception of longer  
28 stagnation events in which preexisting background O<sub>3</sub> reacts away or is deposited as fresh O<sub>3</sub> is  
29 produced from local precursors.

30 Lin et al. (2001) analyzed the EPA AIRS database for the 1980-1998 period and noted  
31 that O<sub>3</sub> concentrations have declined at the high end of the probability distribution, consistent

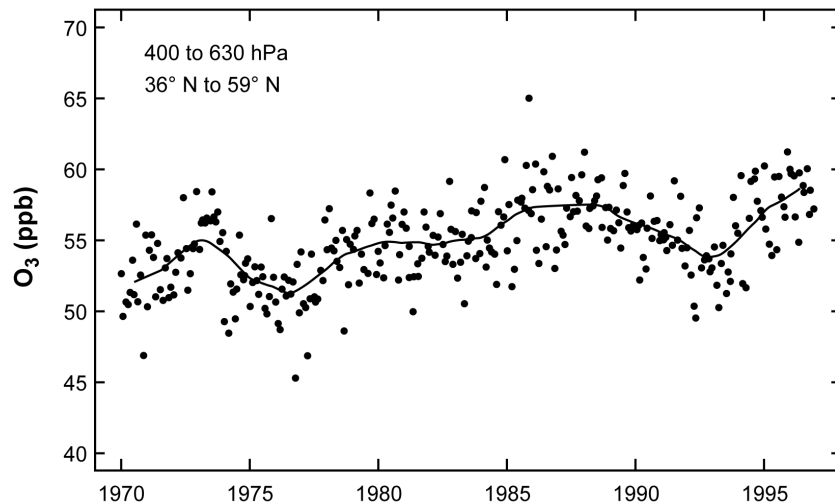
1 with the effects of emissions controls, but had increased at the low end of the distribution by 3-5  
2 ppb. They divided the monitoring data for the continental U.S. into 4 quadrants by geography,  
3 and noted a pattern of increase for the Western states that might be attributed to the long-range  
4 transport of O<sub>3</sub> precursors from Asia. They found, however, that the Northeastern quadrant had  
5 the highest increase in the low end of the concentration probability distribution, which could not  
6 be reasonably attributed to transboundary transport of O<sub>3</sub> precursors.

7 While not representing an ideal source of information for assessing the climatic effects  
8 of O<sub>3</sub> within the continental United States, data from the large air-quality-focused ground-based  
9 monitoring network may be used to identify boundary-layer geospatial and temporal patterns  
10 in O<sub>3</sub> concentrations for comparison to regional-scale chemistry/climate models. Extensive  
11 analysis of data available within the EPA AQS database can be found in Chapter 3 of this  
12 document, including an analysis showing the diurnal O<sub>3</sub> concentration patterns for several large  
13 metropolitan areas with peak values ranging up to 160 ppb (Los Angeles). Lehman et al. (2004)  
14 analyzed the AQS database of daily 8-h maximum O<sub>3</sub> concentrations collected for 1,090 stations  
15 in the eastern half of the United States for the 1993 to 2002 period. They applied a rotated  
16 principle component analysis to a reasonably complete, spatially representative, nonurban subset  
17 of the database in order to identify coherent, regionally oriented patterns in O<sub>3</sub> concentrations.  
18 Five spatially homogenous regions were identified: the U.S. Northeast, Great Lakes, Mid-  
19 Atlantic, Southwest (including Alabama, Louisiana, Texas, Oklahoma), and Florida. The  
20 Mid-Atlantic region displayed the highest mean concentration (52 ppb) of all of the regions  
21 analyzed, with the Great Lakes, Southwest, and Northeast regions following with around 47 ppb.  
22 The average concentration derived for Florida was 41 ppb. The authors found strong  
23 correlations in measured concentrations among stations within the same region, suggesting that  
24 the geospatial patterns of pollutant emissions and meteorological activity may also have a  
25 regional orientation. These results that these regions may define natural domains for regional  
26 scale modeling studies of the influence of O<sub>3</sub> (as well as PM) on climate.

#### 27 28 **10.3.4.3 Ozone Trends: Globally and in North America**

29 For the Northern Hemisphere, weekly continuous data are available from 1970 for only  
30 nine stations in the latitude range 36° N to 59° N (IPCC, 2001a). Available tropospheric O<sub>3</sub>  
31 measurements do not reveal a clear trend in concentration, while trends in the stratosphere are

1 more readily identified. Different trends are seen at different locations for different periods,  
2 consistent with regional changes in pollutant emission, especially  $\text{NO}_x$ . Logan et al. (1999)  
3 analyzed the composite record of mid-tropospheric  $\text{O}_3$  abundance from the nine-station network.  
4 A plot of data is shown in Figure 10-7. While no clear trend appeared for 1980 through 1996,  
5 the average level for second half of this record (about 57 ppb) is clearly greater than for the first  
6 half (about 53 ppb). The trend may be consistent with changes in regional  $\text{NO}_x$  emission rates  
7 occurring due to pollution reduction efforts in developed countries and increasing emissions in  
8 rapidly growing economies in Asia. The measurements shown in Figure 10-7 are for surface  
9 concentrations only. Fewer locations have measured changes in the concentrations of  $\text{O}_3$  as a  
10 function of altitude. Fewer still are locations that have collected and maintained data records  
11 prior to 1970. The absence of historical data on the vertical distribution of  $\text{O}_3$  adds to the  
12 difficulty in estimating historical atmospheric burdens and trends in  $\text{O}_3$ -related climate forcing.



**Figure 10-7. Mid-tropospheric  $\text{O}_3$  abundance (ppb) in northern midlatitudes ( $36^\circ\text{N}$ - $59^\circ\text{N}$ ) for the years 1970 to 1996. Observations between 630 and 400 hPa are averaged from nine ozonesonde stations (four in North America, three in Europe, two in Japan), following the data analysis of Logan et al. (1999). Values are derived from the residuals of the trend fit, with the trend added back to allow for discontinuities in the instruments. Monthly data (points) are shown with a smoothed 12-month mean (line).**

Source: IPCC (2001a).



1 The IPCC (2001a) surveyed the results of published chemistry transport model (CTM)  
 2 modelling studies (see Table 10-2) that estimated the global average increases in total column O<sub>3</sub>  
 3 since the preindustrial era. Model estimates ranged from +7 to +12 DU. On the basis of these  
 4 estimates, available measurements, and other analyses, the IPCC estimated that total column O<sub>3</sub>  
 5 has increased by 9 DU with a 67% confidence range of +6 to +13 DU. In some of the modelling  
 6 studies, emissions scenarios predicted a further increase in column O<sub>3</sub> due to growing emissions  
 7 of O<sub>3</sub> precursors. Fusco and Logan (2003) stated that, according to models, increased NO<sub>x</sub>  
 8 emissions from fossil fuel combustion have had the greatest effect on O<sub>3</sub> in the lower  
 9 troposphere since the 1970s. In addition, increases in background CH<sub>4</sub> have also contributed as  
 10 much as 20% to the increase in tropospheric O<sub>3</sub>, in the northern latitudes. Given its longer  
 11 atmospheric residence time, CH<sub>4</sub> can serve as an O<sub>3</sub> precursor at much longer distances from its  
 12 source than can other O<sub>3</sub> precursors, and, therefore, has a more uniform effect across the globe.  
 13  
 14

**Table 10-2. CTM Studies Assessed by the IPCC for its Estimate of the Change in Global and Total Column O<sub>3</sub> Since the Preindustrial Era**

<b>Estimated Change in Column O<sub>3</sub> in DU</b>	<b>Model Used</b>	<b>References</b>
7.9	<i>GFDL</i>	Haywood et al. (1998)
8.9	<i>MOZART-1</i>	Hauglustaine et al. (1998)
8.4	<i>NCAR/2D</i>	Kiehl et al. (1999)
9.5	<i>GFDL-scaled</i>	Levy et al. (1997)
12	<i>Harvard/GISS</i>	Mickley et al. (1999)
7.2	<i>ECHAM4</i>	Roelofs et al. (1997)
8.7	<i>UKMO</i>	Stevenson et al. (2000)
9.6	<i>UIO</i>	Berntsen et al. (1999)
8	<i>MOGUNTIA</i>	VanDorland et al. (1997)

Source: IPCC (2001a)

1 Fusco and Logan (2003) found a 10% increase in O<sub>3</sub> concentrations year-round over  
 2 Canada, Europe, and Japan and a 20% increase for Japan and Europe during spring and summer.  
 3 It was expected — but not found — that O<sub>3</sub> concentrations over Japan would increase in line

1 with emissions from China. The authors suggested that convective activity over Asia is stronger  
2 than that seen over other industrialized areas of the global. Such a meteorological characteristic  
3 would result in an injection of pollutants into the free troposphere, allowing long-range transport  
4 to North America. Their suggestion is supported by evidence of increasing background  
5 concentrations within the United States (Fiore et al., 1998).

6 NARSTO (2000), in its assessment of the available information on O<sub>3</sub> pollution in North  
7 America, stated that no single pattern for trends in O<sub>3</sub> over North America can be found in the  
8 available monitoring data. In the United States, the average 1-h concentration at surface  
9 monitoring sites decreased by 15% between 1986 and 1996, with most of the observed  
10 declines occurring in urban and urban-influenced locations. The largest declines occurred in  
11 Los Angeles, New York, and Chicago. Free tropospheric O<sub>3</sub> concentrations appeared to hold  
12 steady, or only declined slightly, from the 1980s, forward.

13 In preparation for the IPCC TAR (IPCC, 2001a), research groups engaged in modeling  
14 global-scale tropospheric chemistry were invited to participate in a model intercomparison  
15 focusing on potential changes in the oxidative capacity of the atmosphere (OxComp), which  
16 included O<sub>3</sub> concentrations, for the 2000 to 2100 period. Participating groups employed the  
17 IPCC A2p scenario, i.e., including the highest emissions levels, to calculate the geospatial  
18 distribution of O<sub>3</sub> up to 20 km. The predicted spatial distributions of O<sub>3</sub> were quite variable,  
19 but the predictions for total column O<sub>3</sub> density change fell within 9 DU of each other (11.4 to  
20 20.5 DU) in all cases and that was considered to be encouraging by the authors. Fusco and  
21 Logan (2003) pointed out that several unresolved issues may limit the ability of models in  
22 reproducing observed trends in tropospheric O<sub>3</sub>. Among these are the use of different  
23 meteorological inputs, photochemical reaction schemes, and predicted cloud cover — each  
24 contributing to different predictions in O<sub>3</sub> production and loss rates.

#### 25 26 **10.3.4.4 The Sensitivity of Ozone-Related Forcing Surface to Albedo**

27 The characteristics of the surface underlying an O<sub>3</sub> enhancement play a role in the O<sub>3</sub>  
28 forcing effect. Highly reflective surfaces, such as light-colored deserts, sea ice and snow, scatter  
29 solar short wave (UV and visible) radiation. UV and visible radiation can then be absorbed,  
30 transformed into long-wave radiation, and reemitted in part back to the surface by tropospheric  
31 O<sub>3</sub>. Studies by two groups, Hauglustaine et al. (1998) and Mickley et al. (1999), have shown

1 that industrial pollution that has been transported to the Arctic induces a high, regional O<sub>3</sub>-  
2 related forcing due to the highly reflective underlying ice and snow surface.

3 Liao et al. (2004) calculated that the maximum change in O<sub>3</sub>-related top-of-the-atmosphere  
4 forcing occurs over high albedo regions in high northern latitudes. Surface forcing was  
5 calculated to be greatest at high northern latitudes as well as at dust-source regions, which also  
6 tend to have high surface albedos.

#### 7 8 **10.3.4.5 The Altitude Dependence of Forcing by Tropospheric Ozone**

9 Altitude plays an important role in the forcing effect of tropospheric O<sub>3</sub> (IPCC, 1992;  
10 Gauss et al., 2003). The efficiency of IR absorption by O<sub>3</sub> depends upon its temperature – at  
11 atmospheric temperatures that are low, relative to the Earth’s surface, it has the capacity to  
12 absorb more IR radiation than O<sub>3</sub> at temperatures close to that of the surface. While this  
13 temperature effect applies to all GHGs, it introduces a complication for estimating forcing by O<sub>3</sub>,  
14 because O<sub>3</sub> is not homogeneously mixed within the troposphere. Ozone forcing estimates must  
15 account for these difficult-to-predict vertical inhomogeneities. However, as part of the OxComp  
16 modeling intercomparison, Gauss et al. (2003) found that the overall forcing by O<sub>3</sub> can be  
17 calculated within reasonable uncertainty simply on the basis of total column density.

#### 18 19 **10.3.4.6 Co-occurrence of Ozone with Particulate Matter**

20 Analysis of the 2001 data from the AQS database showed infrequent co-occurrence of  
21 high PM<sub>2.5</sub> and O<sub>3</sub> concentrations (Chapter 3 of this document). For those cases when O<sub>3</sub>  
22 production is high, in combination with high PM concentrations, there is a suggestion in the  
23 literature that heterogeneous chemistry on PM surfaces may lead to reduced gas-phase O<sub>3</sub>. Liao  
24 et al. (2004) modeled heterogeneous chemistry taking place on PM, and found a significant  
25 titration of O<sub>3</sub> and its NO<sub>x</sub> precursors. The importance of this titration effect remains an open  
26 question, given the difficulty in obtaining in situ measurements to validate model calculations.

27 Liao et al. (2004) also estimated that forcing by BC, mineral dust, and organic carbon  
28 aerosols substantially offsets forcing by tropospheric O<sub>3</sub>, yielding an overall negative globally  
29 averaged forcing at both the top of the atmosphere and at the Earth’s surface. However, such  
30 estimates neglect the regional aspects of forcing by these individual pollutants. Elevated  
31 concentrations of these very different types of pollutants often appear independently of the

1 others, such as with biomass burning plumes, Saharan dust, and organic aerosols associated  
2 with biogenic terpene emissions by forests. It is unlikely that a global average of the forcing  
3 effects of these individual pollutants will adequately capture their impacts on climate at the  
4 regional scale.

### 6 **10.3.5 Estimated Forcing by Tropospheric Ozone**

#### 7 **10.3.5.1 Direct Climate Forcing Due to Ozone**

8 The inhomogeneous distribution of O<sub>3</sub> within the troposphere, coupled with the large  
9 uncertainty in the global O<sub>3</sub> budget, significantly complicates the matter of estimating the global  
10 average direct forcing due to O<sub>3</sub>. The IPCC TAR (2001a) lists the results of several modeling  
11 studies that estimated the annual change in the relative forcing by O<sub>3</sub> from preindustrial times.  
12 It was noted that the differences among the estimates were most likely due to differences in  
13 predicted O<sub>3</sub> chemistry, including the emissions inventories used and the chemical process  
14 and transport mechanisms incorporated into the models, rather than by factors relating to  
15 radiative transfer. The IPCC intercomparison of the models and their results indicate that the  
16 uncertainties in estimated forcings due to O<sub>3</sub> have decreased since the IPCC Second Assessment  
17 Report (1996).

18 The O<sub>3</sub>-related forcings estimated by studies considered by the IPCC (2001a) are listed in  
19 Table 10-3. Ten of the listed estimates are based on global CTM calculations. One study was  
20 constrained by a climatology derived from observations. Given the differences in calculated  
21 total column O<sub>3</sub> among the models, a normalized forcing (Wm<sup>-2</sup> per Dobson Unit of  
22 tropospheric O<sub>3</sub> change) is listed in addition to the absolute forcing (Wm<sup>-2</sup>) estimated by each  
23 model. Both clear sky (cloud-free) and total sky (including clouds) forcing estimates are listed.

24 The largest O<sub>3</sub>-related forcings coincide with the strongest sources of tropospheric O<sub>3</sub>,  
25 which the models predict occur in the northern midlatitude regions (40° to 50° N) and reach as  
26 much as 1 Wm<sup>-2</sup> in the summer as well as in the tropics, and are related to biomass burning.  
27 In general, the estimates are comparable in magnitude and show similarity in geographic  
28 distribution. For total sky conditions, the range in globally and annually averaged  
29 tropospheric O<sub>3</sub> forcing from all of these models is from 0.28 to 0.43 Wm<sup>-2</sup>, while the  
30 normalized forcing is 0.033 to 0.056 Wm<sup>-2</sup> per DU. As expected, they are opposite in sign to  
31 the forcing estimated for sulfate aerosols, which scatter radiation. The range in normalized

**Table 10-3. Tropospheric O<sub>3</sub> Change (O<sub>3</sub>) in Dobson Units (DU) Since Preindustrial Times, and the Accompanying Net (SW plus LW) Radiative Forcings (Wm<sup>-2</sup>), After Accounting for Stratospheric Temperature Adjustment (using the Fixed Dynamical Heating Method). Estimates are Taken From the Published Literature. Normalized Forcings (norm.) Refer to Radiative Forcing per O<sub>3</sub> Change (Wm<sup>-2</sup> per DU)**

Reference	Clear sky conditions			Total sky conditions	
	ΔO <sub>3</sub>	Net	Net (norm.)	Net	Net (norm.)
Berntsen et al. (1997) – [Reading model]	7.600	0.310	0.041	0.280	0.037
Stevenson et al. (1998)	8.700	0.391	0.045	0.289	0.033
Berntsen et al. (1997) – [Oslo model]	7.600	0.390	0.051	0.310	0.041
Haywood et al. (1998)	7.900	0.380	0.048	0.310	0.039
Kiehl et al. (1999)	8.400	0.379	0.045	0.320	0.038
Berntsen et al. (2000)	9.600	0.428	0.045	0.342	0.036
Brasseur et al. (1998)	—	—	—	0.370	—
van Dorland et al. (1997)	8.070	0.443	0.055	0.380	0.047
Roelofs et al. (1997)	7.200	0.397	0.055	0.404	0.056
Lelieveld and Dentener (2000)	—	—	—	0.420	—
Hauglustaine et al. (1998)	8.940	0.511	0.057	0.426	0.048
Mean	8.224	0.403	0.049	0.343	0.042

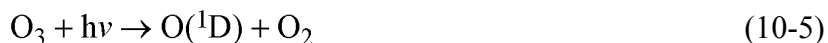
Source: IPCC (2001a).

1 forcings emphasizes the differences in assumptions used by the different models. The  
2 tropospheric O<sub>3</sub> forcing constrained by the observational climatology is 0.32 Wm<sup>-2</sup> for globally  
3 averaged, total sky conditions. As shown in Figure 10-6, the IPCC (2001a) concluded that  
4 0.35 ± 0.15 Wm<sup>-2</sup> represents the most likely value for annually and globally-averaged forcing  
5 by tropospheric O<sub>3</sub>. Not included here is the study by Hansen and Sato (2001), that evaluated  
6 forcing by O<sub>3</sub> with corrections made to the assumptions concerning pre-industrial O<sub>3</sub>  
7 concentrations and the effects of natural O<sub>3</sub> precursors, especially NO<sub>x</sub> generated by lightning.  
8 Hansen and Sato (2001) concluded that a more likely range for globally averaged forcing by O<sub>3</sub>  
9 is 0.4 to 0.8 Wm<sup>-2</sup>, with 0.5 Wm<sup>-2</sup> as their best estimate.

1 Since the publication of the IPCC TAR (2001a), new studies have been published that  
2 illuminate some of the regionally-relevant details associated with direct forcing by O<sub>3</sub>  
3 (Mickley et al., 2004a; Liao et al., 2004). Forcing by O<sub>3</sub>, due to its capacity for absorbing solar  
4 UV as well as solar and terrestrial IR radiation, can be divided into “shortwave” forcing and  
5 “long-wave” forcing. These forcings occur under different conditions. Shortwave forcing can  
6 only take place during daytime, while long-wave forcing can occur at all hours as a function of  
7 the diurnally varying concentration of atmospheric O<sub>3</sub>. As noted, earlier, unlike CO<sub>2</sub>, the  
8 absorption spectrum for O<sub>3</sub> is distinct from that of water vapor — meaning that O<sub>3</sub> will absorb  
9 and reemit long-wave radiation without interference by water under high humidity conditions.  
10 Mickley et al. (2004a) reported that, according to their modeling study, surface temperature  
11 response to the predicted O<sub>3</sub> enhancement since the preindustrial period differs greatly from that  
12 of the CO<sub>2</sub> response, and that this difference can only be explained by the geographical  
13 distribution and absorption properties of O<sub>3</sub>. Liao et al. (2004) estimated globally averaged  
14 top-of-the-atmosphere separate short- and long-wave forcings to O<sub>3</sub> to be 0.21 W/m<sup>2</sup> and  
15 0.32 W/m<sup>2</sup>, respectively.

#### 17 **10.3.5.2 Indirect Forcing Due to Ozone**

18 Ozone has an indirect climate forcing effect due to its role in the oxidative removal of  
19 other reactive GHGs, including CH<sub>4</sub>, hydrofluorocarbons (HFCs), and other reactive NMHCs.  
20 The primary actor in this effect is a second generation product of the photolysis of O<sub>3</sub>, the  
21 hydroxyl (OH) radical. Hydroxyl radicals are produced by way of a pair of reactions that start  
22 with the photodissociation of O<sub>3</sub> by solar UV.



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32 Reactions with OH are the primary removal mechanism for CH<sub>4</sub> and NMHCs as well as the  
33 pollutants NO<sub>x</sub> and CO. Methane and CO are in especially high abundance in the global  
34 atmosphere. OH is estimated to react with these two gases within 1 second of its formation.

1 In addition to CH<sub>4</sub>, NO<sub>x</sub>, CO, and the NMHCs, OH concentrations are controlled by local  
2 concentrations of H<sub>2</sub>O (i.e., humidity) and the intensity of solar UV. Different atmospheric  
3 concentrations of the required precursors suggest that preindustrial OH concentrations were  
4 likely to have been different from present-day concentrations, but there is no consensus on the  
5 magnitude of this difference. Observations of global atmospheric concentrations of  
6 methylchloroform (CH<sub>3</sub>CCl<sub>3</sub>), a well-mixed tropospheric species that also reacts with OH, have  
7 been used to estimate OH abundances. Independent studies have shown overlapping trends for  
8 the period 1978 to 1994, but none of the trends are outside the given uncertainty ranges (0.5 ±  
9 0.6%/year) (Prinn et al., 1995; Krol et al., 1998). The IPCC (2001a) reported a range of +5% to  
10 -20% for predicted changes in global OH abundances.

11 Given the difficulty in estimating global OH abundances in the past, present, and future,  
12 estimates of indirect forcing due to O<sub>3</sub> have been difficult to obtain and are highly uncertain.

### 14 **10.3.5.3 Predictions for Future Climate Forcing by Anthropogenic Ozone**

15 The rate of increase in surface O<sub>3</sub> in Europe and North America since 1980 appears to be  
16 slowing, likely due to control measures intended to improve urban air quality. Not surprisingly,  
17 CTM modeling attempts to predict future precursor emissions and resulting O<sub>3</sub> abundances  
18 indicate that the largest future O<sub>3</sub>-related forcings will be related to population growth and  
19 economic development in Asia (van Dorland et al., 1997; Brasseur et al., 1998). The results of  
20 these modeling studies predict that the globally averaged total radiative forcing due to O<sub>3</sub> from  
21 preindustrial times 0.66 Wm<sup>-2</sup> will rise to 0.63 Wm<sup>-2</sup> by 2050. Chalita et al. (1996) predicted a  
22 change in the globally averaged radiative forcing from preindustrial times to 2050 of 0.43 Wm<sup>-2</sup>.  
23 Stevenson et al. (1998) predicted an O<sub>3</sub>-related forcing of 0.48 Wm<sup>-2</sup> in 2100. Applying the  
24 SRES scenario projecting the highest emissions out to the year 2100 (IPCC, 2000), the OxComp  
25 model intercomparison study yielded a projected O<sub>3</sub>-induced forcing ranging from 0.40 to  
26 0.78 Wm<sup>-2</sup>. The authors concluded, given their prediction for forcing by well-mixed GHGs of  
27 5.6 Wm<sup>-2</sup>, that O<sub>3</sub> would remain an important contributor to overall anthropogenic forcing well  
28 into the future. However, all of these predictions must be viewed with caution given the  
29 considerable uncertainties associated with the long-term economic activity projections required  
30 for such estimates.

### 10.3.6 The Impact of a Warming Climate on Atmospheric Ozone Concentrations

Evaluation of the potential impact of climate warming on U.S. air quality is currently underway. Initial modeling results reported by Mickley et al. (2004b) suggest that reduced cyclone frequency in a warmer climate will lead to increases in the severity of summertime pollution episodes. Cyclonic weather patterns are known to play an important role in ventilating pollution away from the surface. They note that compelling evidence is accumulating that the frequency of these cyclones has decreased over the past few decades. An early study by Jacob et al. (1993) found a correlation between O<sub>3</sub> concentrations and temperature was due to the effect of O<sub>3</sub> on atmospheric chemistry, biogenic emissions, and stagnation.

### 10.3.7 Conclusion

The general consensus within the atmospheric sciences community, as represented by the United Nations Intergovernmental Panel on Climate Change (IPCC), is that human activities have a discernable effect on the Earth's climate. However, quantifying the extent of human-induced forcing on climate requires detailed information about human-induced change on the components of the Earth System that govern climate. Tropospheric O<sub>3</sub> is a well-known GHG, but information regarding its historical trends in concentration, its current and future atmospheric burden, and other critical details needed for estimating its direct and indirect forcing effects on the climate system are highly uncertain.

The IPCC has estimated that the globally averaged forcing due to O<sub>3</sub> is approximately  $0.35 \pm 0.15 \text{ Wm}^{-2}$ , with an updated value of  $0.5 \pm 0.2 \text{ Wm}^{-2}$  provided by Hansen and Sato (2001). However, the most important role of O<sub>3</sub> in climate is likely to be at the regional scale, adjacent to the sources of its chemical precursors. This expectation is consistent with satellite observations of high regional scale column O<sub>3</sub> densities near large urban areas and large-scale biomass burning activity. Modeling studies evaluated by the IPCC have estimated that regional-scale forcing due to O<sub>3</sub> can approach  $1 \text{ Wm}^{-2}$ , or as much as 40% of the globally averaged forcing due to the well-mixed GHGs. While more certain estimates of the overall importance of global-scale forcing due to tropospheric O<sub>3</sub> await further advances in monitoring and chemical transport modeling, the overall body of scientific evidence suggests that high concentrations of O<sub>3</sub> on the regional scale could have a discernable influence on climate, leading to surface



1 temperature and hydrological cycle changes. Confirming this effect requires improvement in  
2 regional-scale modeling — an activity that is currently underway.

3

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# 11. EFFECT OF OZONE ON MAN-MADE MATERIALS

Ozone (O<sub>3</sub>) and other photochemical oxidants react with many economically important man-made materials, decreasing their useful life and aesthetic appearance. Some materials known to be damaged by ozone include elastomers, fibers, dyes, and paints. This chapter provides a brief discussion of O<sub>3</sub> effects on man-made materials, including denoting of damage mechanisms and, where possible, concentration-response relationships. Much of what is known about ozone effects on man-made materials is derived from research conducted in the 1970's, 1980's, and early 1990's, with very little new research on the subject having been conducted since then. Since only very limited new information has been published on effects of ozone on materials, this chapter mainly summarizes key information assessed in the previous 1996 Air Quality Criteria Document for Ozone and other photochemical oxidants (1996 O<sub>3</sub> AQCD) (U.S. Environmental Protection Agency, 1996) and provides detailed discussion of the very limited new information that has become available since then. In the ensuing sections, discussion is focused on ozone effects on: elastomers (Sect 11.1); textiles and fabrics (11.2); dyes, pigments, and inks (11.3); artist's pigments (11.4); and surface coatings (11.5). Evaluation of the relevance and economic importance of O<sub>3</sub> materials damage information, as it affects productivity or cultural resources (such as museums), is beyond the scope of this chapter. The reader is referred to the previous criteria document (1996 O<sub>3</sub> AQCD) for a more detailed discussion of the earlier studies summarized below.

## 11.1 ELASTOMERS

The elastomeric compounds, natural rubber and synthetic polymers and copolymers of butadiene, isoprene, and styrene, are particularly susceptible to even low levels of ozone. Elastomeric compounds are long chain unsaturated organic molecules. Ozone damages these compounds by breaking the molecular chain at the carbon-carbon double bond; a chain of three oxygen atoms is added directly across the double bond, forming a five-membered ring structure (Mueller and Stickney, 1970). The change in structure promotes the characteristic cracking of stressed/stretched rubber called "weathering." A 5% tensile strain will produce cracks on the

1 surface of the rubber that increase in number with increased stress/stretching. The rate of crack  
2 growth is dependent on the degree of stress, the type of rubber compound, concentration, time of  
3 exposure, velocity, and temperature (Bradley and Haagen-Smit, 1951; Lake and Mente, 1992)  
4 (Gent and McGrath, 1965). Once cracking occurs, there is further penetration, additional  
5 cracking, and eventually mechanical weakening or stress relaxation (U.S. Environmental  
6 Protection Agency, 1996). Razumovskii et al. (1988) demonstrated the effect of ozone on stress  
7 relaxation of polyisoprene vulcanizates. A decrease in stress (stress relaxation) is caused by  
8 ozone-induced cracks in exposed elastomers resulting in irreversible changes in the elastomer  
9 dimensions and decreased tensile strength.

10 To counteract ozone effects on elastomers, antiozonants and wax are often added to the  
11 elastomeric formulations during processing. An antiozonant is an additive used to protect a  
12 polymer against the effects of ozone-induced degradation and, hence, is used mainly in diene  
13 rubbers. Antiozonant protection works either (a) by providing a physical barrier to ozone  
14 penetration via forming a thin surface film of an ozone-resisting wax or (b) by chemically  
15 reacting with ozone or polymer ozonolysis products, as do aromatic diamines such as  
16 p-phenylene diamine derivatives. The antiozonant diffuses to the surface of the elastomeric  
17 material, where it reacts with ozone faster than ozone reacts to break the molecular chain and the  
18 carbon-carbon double bond, or the antiozonant diffuses to the surface of the material but is not  
19 reactive with ozone and serves as a protective coating against ozone attack. The antiozonant  
20 may also serve to scavenge ozone while also providing protective film against ozone attack  
21 (Andries et al., 1979; Lattimer et al., 1984).

22 Most of the studies on ozone effects on elastomers were designed to evaluate the  
23 effectiveness of antiozonants in counteracting the rubber cracking produced by ozone exposure.  
24 Consequently, many of the studies were conducted using ozone concentrations higher than those  
25 typically found in the ambient air. Natural rubber strips exposed to high concentrations of ozone  
26 (20,000 ppm) under stressed conditions cracked almost instantaneously and were broken within  
27 1 sec. When the ozone concentration was lowered (0.02 to 0.46 ppm), the time to required to  
28 produce cracks in the exposed rubber material was increased (Bradley and Haagen-Smit, 1951).  
29 Lake and Mente (1992) studied the effect of temperature on ozone-induced elastomer cracking  
30 and antiozonant protection on natural rubber, epoxidised natural rubber, and two acrylonitrile-  
31 butadiene copolymers under constant strain. Temperatures ranged from -20 °C to +70 °C. The

1 elastomers were exposed to 0.05 to 1,000 ppm ozone for 16 h. Ozone cracking decreased at  
 2 lower ambient temperatures; however, diffusing of both chemical and wax antiozonants also  
 3 were slowed at the lower temperatures. Cracking was slightly increased at the higher  
 4 temperatures but the antiozonants offered more protection.

5 Serrano et al. (1993) evaluated the appropriateness of using ozone-induced elastomer  
 6 cracking to estimate ambient ozone concentrations. Two vulcanized natural rubber compounds  
 7 were exposed for 24 h to varying ozone concentrations under stressed conditions. Ozone  
 8 concentrations were 60, 80, 90, 100, and 120 ppb for durations of 2, 4, or 6 h. The 24 h average  
 9 ozone concentrations ranged from 31 to 57.5 ppb. There was a clear relationship between the  
 10 24-h average ozone concentration and the distribution of crack length frequencies on the rubber  
 11 surface. Table 11-1 gives the average 24-h ozone concentration and lengths for two vulcanized  
 12 natural rubber strips.

**Table 11-1. Average 24-h Ozone Concentrations Producing the Highest Frequency of Cracks of a Certain Length in the Middle and Central Zones of the Rubber Test Strips**

Crack Length (mm)	1% Antiozonant 4010NA #		0.5% Antiozonant 4010NA	
	Middle Zones	Central Zones	Middle Zones	Central Zones
0.05 - 0.10	37.5	37.5	40.0	42.5
0.10 - 0.15	45.0	48.0	48.0	53.0
0.15 - 0.20	48.0	≥57.5	≥57.5	≥57.5
0.20 - 0.40	≥57.5	≥57.5	≥57.5	≥57.5

Ozone concentrations given in ppb.

Adapted from Serrano et al. (1993).

## 1 11.2 TEXTILES AND FABRICS

2 Ozone can damage textiles and fabrics by methods similar to those associated with  
 3 elastomers. Generally, synthetic fibers are less affected by ozone than natural fibers; however,  
 4 ozone contribution to the degradation of textiles and fabrics is not considered significant (U.S.

1 Environmental Protection Agency, 1996). A study reported by Bogaty et al. (1952) showed that  
 2 ozone affects moistened cloth more than dry cloth. Scoured cotton duck cloth and commercially  
 3 bleached cotton print cloth were exposed to 20 to 60 ppb for 1,200 h (50 days). The rate of  
 4 deterioration was measured by the changes in cuprammonium fluidity values and the fabric  
 5 breaking strength. At the end of the 1,200-h exposure, there was a 20% loss in breaking  
 6 strength. Table 11-2 list the changes in cuprammonium fluidity values for both fabrics.

**Table 11-2. Cuprammonium Fluidity of Moist Cotton Cloth Exposed to 20 to 60 ppb Ozone**

	<b>Duration of Exposure (h)</b>	<b>Cuprammonium Fluidity (rhcs)</b>
Duck Cloth	0	2.6
	200	2.8
	680	4.0
	960	6.8
	1200	9.5
Bleached Print Cloth	0	8.2
	200	8.7
	510	9.4
	650	12.0
	865	12.7
	1500	16.5

Adapted from Bogaty et al. (1952).

### 1 **11.3 DYES, PIGMENTS, AND INKS**

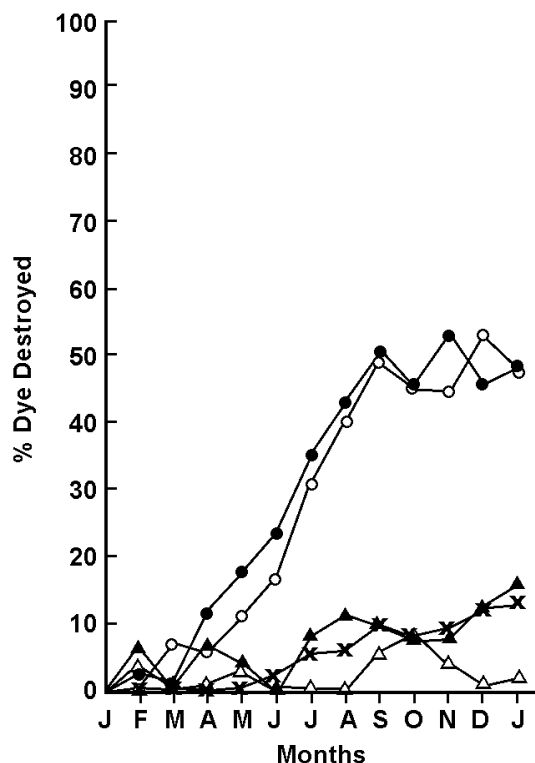
2 Ozone fading of textile dyes is diffusion-controlled; the rate of fading is controlled by the  
 3 diffusion of the dye to the fiber surface. Many textile dyes react with ozone; however, the rate  
 4 and severity of the ozone attack is influenced by the chemical nature of the textile fiber and the  
 5 manner in which the dye is applied. Ozone molecules break the aromatic ring portion of the dye  
 6 molecule, oxidizing the dye (U.S. Environmental Protection Agency, 1996). In case of aromatic  
 7 azo dyes, ozone attacks the aromatic rings and electron rich nitrogen atoms (Matsui et al., 1988).  
 8 Grosjean et al. (1987; 1988a,b) proposed a mechanism for reactions of ozone with indigo,  
 9 thioindigo, and dibromoindigo, alazarin, and curcumin dyes under dark conditions. Ozone

1 attaches to the dye molecule at the unsaturated carbon = carbon bond. An ozone adduct is  
2 formed (1,2,3-trioxolane), followed by scission of the carbon–carbon bond and the subsequent  
3 formation of the corresponding Criegee biradical. A similar mechanism was proposed for the  
4 reaction of ozone with triphenylmethane colorant Basic Violet 14. Ozone attacked Basic Violet  
5 14 at the carbon=carbon unsaturated bond and at the carbon–nitrogen unsaturated bond under  
6 dark conditions. Other members of the group of triphenylmethane colorants with unsaturated  
7 carbon–carbon bonds also are expected to be subject to ozone fading. Triphenylmethane  
8 colorants that are expected to be ozone-fugitive include the amino-substituted cationic dyes  
9 (Malachite Green, Brilliant Green, Crystal Violet, Pararosaniline Chloride, Methyl Green, and  
10 others) (Grosjean et al., 1989).

11 An indication that ozone caused textile dye fading was first reported by Salvin and Walker  
12 (1955). The researchers found that the fading was primarily the result of the destruction of the  
13 blue dye molecule. Drapes made of acetate, Arnel, and Dacron and dyed with anthraquinone  
14 blue dye exhibited a decrease in shade that was not accompanied by the characteristic reddening  
15 caused by NO<sub>x</sub>. Figures 11-1 and 11-2 demonstrate the effect of ozone exposure on nylon 6 yarn  
16 colored with several blue dyes. Nylon samples inside the home were located on a wall away  
17 from sunlight. Outside nylon samples were placed on a covered patio or under the eaves of the  
18 house to minimize exposure to sunlight and rain. Ozone concentrations ranged from 2 to 5 ppb  
19 outside and 0 to 2 ppb inside. The percent change in dye color was determined monthly by  
20 extraction and analysis of the remaining dye or by instrumental measurement of the color change  
21 (Haylock and Rush, 1978).

#### 22 23 24 **11.4 ARTISTS' PIGMENTS**

25 Several artists' pigments are sensitive to fading and oxidation by ozone when exposed to  
26 concentrations found in urban areas (Shaver et al., 1983; Drisko et al., 1985; Whitmore et al.,  
27 1987; Whitmore and Cass, 1988; Grosjean et al., 1993). The organic pigments that are ozone  
28 fugitive include alizarin red pigments containing lakes of the polycyclic aromatic compound  
29 1,2-dihydroxyanthraquinone, blue-violet pigments containing substituted triphenylmethane  
30 lakes, indigo, and yellow coloring agents containing polyfunctional, polyunsaturated compounds  
31 such as curcumin (Grosjean et al., 1987). Because of the potential of ozone to damage works of



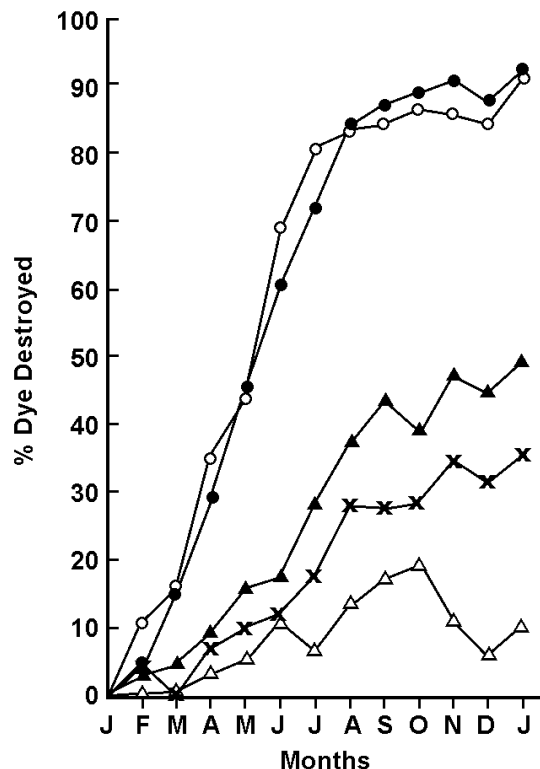
**Figure 11-1. In-service fading of nylon 6 yarn inside house. ● = C.I. Disperse Blue 3; ○ = C.I. Basic Blue 22; ▲ = C.I. Acid Blue 27; x = C.I. Disperse Blue 56; △ = C.I. Acid Blue 232.**

Source: Haylock and Rush (1978).

1 art, recommended limits on ozone concentrations in museums, libraries, and archives are  
 2 relatively low, ranging from 0.013 to 0.01 ppm.

3 Experimental studies demonstrate a concentration  $\times$  time ( $C \times T$ ) relationship between  
 4 ozone concentration, exposure time, and pigment fading. Cass et al. (1991) summarized some of  
 5 the earlier research on the effects of ozone on artists' pigments. In studies evaluating the effect  
 6 of ozone on organic and inorganic watercolors and traditional organic pigments, only the  
 7 traditional organic pigments showed measurable fading from ozone exposure. Of the inorganic  
 8 pigments tested, only the arsenic sulfides showed ozone-related changes. The pigments were  
 9 exposed to 0.3 to 0.4 ppm ozone for 3 mo in the absence of light, at 22 °C and 50% RH. The  
 10 authors equated this exposure to a  $C \times T$  of 6 to 8 years inside a Los Angeles museum with air  
 11 conditioning but without a pollutant removal system.





**Figure 11-2. In-service fading of nylon 6 yarn outside house. ● = C.I. Disperse Blue 3; ○ = C.I. Basic Blue 22; ▲ = C.I. Acid Blue 27; x = C.I. Disperse Blue 56; △ = C.I. Acid Blue 232.**

Source: Haylock and Rush (1978).

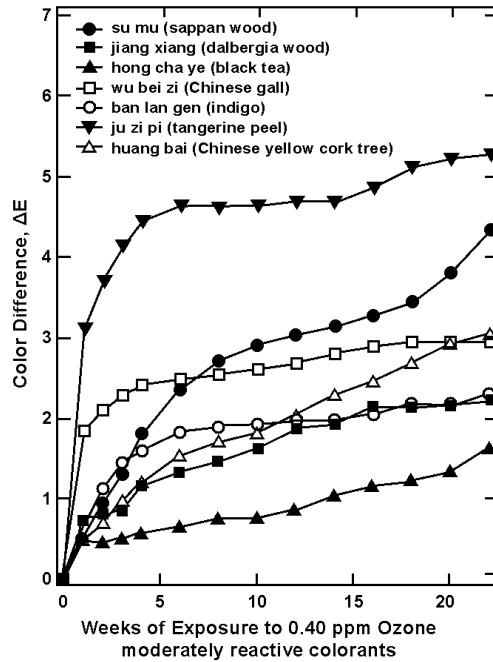
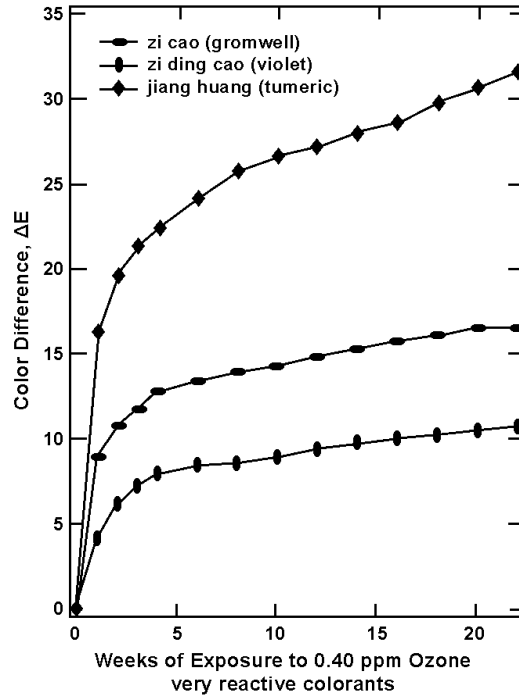
1 Whitmore and Cass (1988) studied the effect of ozone on traditional Japanese colorants.  
 2 Most of these compounds are insoluble metal salts that are stable in light and air. Suspensions or  
 3 solutions of the colorants were airbrushed on hot-pressed watercolor paper or silk cloths.  
 4 A sample of Japanese woodblock print also was included in the analysis. Samples were exposed  
 5 to 0.4 ppm ozone at 22 °C, 50% relative humidity, in the absence of light for 12 wk. Changes in  
 6 reflectance spectra were used to evaluate the effect of ozone exposure on colorant fading.  
 7 Among the colorants tested on paper, curmin, indigo, madder lake, and lac lake were the most  
 8 sensitive to ozone exposure. Gamboge was relatively insensitive to ozone. The blue and green  
 9 areas of the sample from the woodblock print was very reactive due to the indigo dye ozone  
 10 sensitivity. The other colorants, red, yellow, and purple, showed very little sensitivity to ozone.

1 The textiles dyes that reacted with ozone were indigo, alone or in combination with several  
2 yellow dyes.

3 Ye et al. (2000) reported the rate of ozone fading of traditional Chinese plant dyes. Twelve  
4 different colorants were applied to watercolor paper and silk and exposed to 0.4 ppm ozone at  
5 25 °C, at 50% RH, in the absence of light for 22 wks. Dye fading was greater when the colorant  
6 was applied to the watercolor paper compared to the silk cloth due to the darker initial depth of  
7 the shade, the greater saturation of the colorant throughout the cloth. Tumeric, gromwell, and  
8 violet on paper was particularly reactive. Tangerine peel was moderately reactive and sappan  
9 wood, dalbergia wood, Chinese gall, indigo, and Chinese yellow cork tree were slightly reactive  
10 to ozone. Black tea was not reactive to ozone. The colorants on silk samples showing color  
11 changes were gromwell, sappan wood, gardenia, tumeric, and violet. Figures 11-3 and 11-4  
12 demonstrate the color change of the various colorants on watercolor paper and silk.

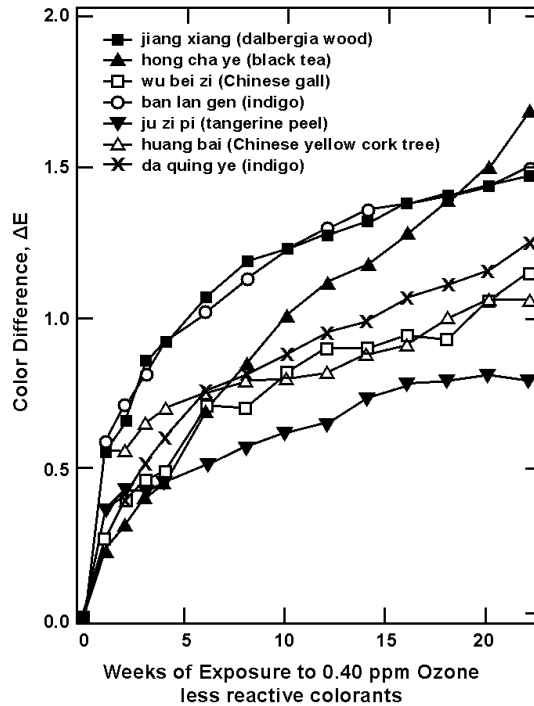
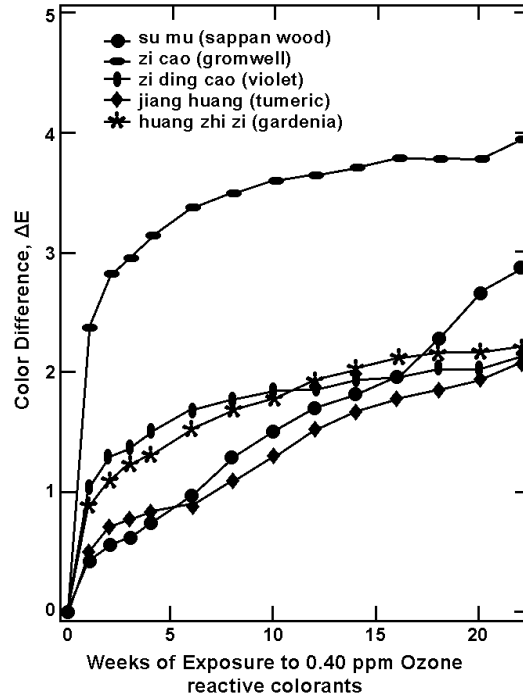
13 Artists' pigments also have exhibited fading when exposed to a mixture of photochemical  
14 oxidants. Grosjean et al. (1993) exposed 35 artists' pigments to a mixture of photochemical  
15 oxidants consisting of ozone, nitrogen dioxide (NO<sub>2</sub>), and peroxyacetyl nitrate (PAN) for  
16 12 wks. Weekly average photochemical concentrations were 200 ppb for ozone, 56 ± 12 to  
17 99 ± 24 for NO<sub>2</sub>, and 11 ± 3 to 18 ± 2 for PAN. All exposures were carried out at room  
18 temperature in the absence of light. To determine the effect of humidity on pigment fading, the  
19 relative humidity was increased from 46% after 8 weeks of exposure to 83% for a 2 week period  
20 and then returned to 46% for the remainder of the exposure.

21 Table 11-3 lists the artists' pigment and degree of fading. Eleven of the pigments  
22 exhibited negligible color change, 12 had small color changes, 3 had modest color changes, and  
23 9 exhibited substantial color changes. Fading of Disperse Blue 3 and Reactive Blue 2 were  
24 likely the result of NO<sub>2</sub> exposure, the fading of triphenylmethanes is consistent with exposure to  
25 nitric acid formed under high humidity conditions. Fading of the indigos was dominated by  
26 ozone exposure and curcumin was faded by all of the photochemicals studied. Increasing the  
27 relative humidity resulted in a substantial color change for all of the pigments, with the  
28 exception of curcumin and indigo.



**Figure 11-3. Observed color changes for natural colorant-on-paper systems during exposure to 0.40 ppm ozone at 25 °C ± 1 °C, 50% RH, in the absence of light.**

Source: Ye et al. (2000).



**Figure 11-4. Observed color changes for natural colorant-on-site during exposure to 0.40 ppm ozone at 25 °C ± 1 °C, 50% RH, in the absence of light.**

Source: Ye et al. (2000).

**Table 11-3. Color Change After 12 Weeks of Exposure to a Mixture of Photochemical Oxidants**

<b>Colorant*</b>	<b>Color Change (<math>\Delta E</math> units)<sup>†</sup></b>	<b>Chemical Functionality or Chemical Composition</b>
Acid Red 37 (17045) <sup>‡</sup>	11.7 ± 0.5	Aminophenyl-substituted azo dye, sulfonate salt
Acid Yellow 65 <sup>‡</sup>	1.8 ± 0.5	Nitro- and phenyl-substituted azo dye, sulfonate salt
Alizarin Carmine	1.8 ± 0.2	Alizarin lake
Alizarin Crimson (Pigment Red 83)	1.4 ± 0.2	Alizarin lake
Aurora Yellow (77199)	0.5 ± 0.1	Cadmium sulfide
Basic Fuschin (42510) <sup>‡</sup>	33.4 ± 3.0	Amino-substituted triphenylmethane
Brilliant Green (42040) <sup>‡</sup>	20.6 ± 2.1	Amino-substituted triphenylmethane
Brown Madder	1.7 ± 0.1	Alizarin lake
Cadmium Yellow (77199)	0.4 ± 0.1	Cadmium sulfide
Carmine	1.8 ± 0.2	Lake of cochineal (substituted anthraquinone)
Chrome Yellow (77600) <sup>‡</sup>	1.7 ± 1.2	Lead chromate
Copper phthalocyanine (Pigment Blue 15)	1.0 ± 0.1	Copper phthalocyanine
Crimson Lake	3.5 ± 0.3	Alizarin lake
Curcumin (Natural Yellow 3)	15.2 ± 2.6	1,7 bis (4-hydroxy-3-methoxyphenyl)-1,6-heptadiene-3,5-dione
Disperse Blue 3	10.8 ± 0.1	Amino-substituted anthraquinone
French Ultramarine Blue	0.8 ± 0.3	
Gamboge (Natural Yellow 24)	0.4 ± 0.1	Gambogic acid
Hooker's Green Light	1.5 ± 0.4	Chlorinated copper phthalocyanine plus ferrous beta naphthol derivative
Indigo (a formulation)	1.1 ± 0.1	Alizarin lake plus lampblack plus copper phthalocyanine
Indigo carmine <sup>‡</sup>	14.0 ± 1.9	5,5-indigo disulfonic acid, sodium salt
Indigo (73000) <sup>‡</sup>	64.1 ± 4.5	
Mauve	3.6 ± 0.5	Lake of triphenyl methane (basic fuschin) plus copper phthalocyanine
New Gamboge	0.9 ± 0.1	Arylamide yellow (CI 11680) plus toluidine red

**Table 11-3 (cont'd). Color Change After 12 Weeks of Exposure to a Mixture of Photochemical Oxidants**

<b>Colorant*</b>	<b>Color Change (<math>\Delta E</math> units)<sup>†</sup></b>	<b>Chemical Functionality or Chemical Composition</b>
Pararosaniline base (42500) <sup>‡</sup>	25.6 ± 4.7	Amino-substituted triphenylmethane
Payne's Grey	1.0 ± 0.1	Alizarin lake plus prussian blue plus lampblack plus ultramarine blue
Permanent Magenta	1.1 ± 0.1	Quinacridone
Permanent Rose	2.0 ± 0.1	Quinacridone
Prussian Blue	0.7 ± 0.2 1.6 ± 0.3	Ferric ferrocyanide
Prussian Green	0.9 ± 0.2	Arylamide yellow plus prussian blue
Purple Lake	2.3 ± 0.3	Alizarin lake
Reactive Blue 2 (61211) <sup>‡</sup>	14.4 ± 1.1	Amino-substituted anthraquinone, sulfonate salt
Rose Carthane (12467)	0.8 ± 0.2	Arylamide (Pigment Red 10) plus xanthene (Pigment Red 90)
Rose Doré	2.0 ± 0.2	Quinacridone plus Yellow 3
Thioindigo Violet (73312) <sup>‡</sup>	1.9 ± 1.2	Chlorinated thioindigo
Winsor Yellow (11680)	0.5 ± 0.2	Arylamide yellow

\* On watercolor paper unless otherwise indicated. Color Index (CI) names or CI numbers are given in parentheses.

<sup>†</sup> Mean ± one standard deviation for triplicate samples calculated from the parameters  $L^*$ ,  $a^*$ , and  $b^*$  measured with the color analyzer.

<sup>‡</sup> On Whatman 41 paper.

Source: Grosjean et al. (1993).

## 1 11.5 SURFACE COATINGS

2 Ozone will act to erode some surface coatings (paints, varnishes, and lacquers). However,  
3 many of the available studies on ozone degradation of surface coatings do not separate the  
4 effects of ozone from other pollutants or environmental factors such as weather, humidity, and  
5 temperature. Campbell et al. (1974) attempted to demonstrate an ozone related effect on oil  
6 house paint, acrylic latex coating, alkyd industrial maintenance coating, urea alkyd coil coating,  
7 and nitrocellulose/acrylic automotive paint. Painted test panels were exposed to 100 and

1 1,000 ppb ozone in a xenon arc accelerated weathering chamber for up to 1,000 h. Using weight  
2 loss as a measure of ozone-induced erosion the researchers concluded that all of the paints tested  
3 suffered degradation in the presence of ozone and that the automotive finish suffered the most  
4 ozone-induced degradation. When ozone degradation was measured using scanning electron  
5 microscopy, the oil house paint and latex coating samples showed erosion above that seen with  
6 clean air but only at the highest exposure level. No effects were noted for the automotive paint.  
7 The other painted surfaces were not evaluated.

8 Spence et al. (1975) studied the effect of air pollutants and relative humidity on oil based  
9 house paint, acrylic latex house paint, acrylic coil coating, and vinyl coil coating under  
10 laboratory conditions. Test panels were exposed in weathering chambers equipped with a xenon  
11 arc light for simulating sunlight to low and high levels of ozone (0.08 and 0.5 ppm), sulfur  
12 dioxide (0.03 and 0.5 ppm), and nitrogen dioxide (0.05 and 0.5 ppm) and relative humidity  
13 (50 and 90%). Samples were exposed for a total of 1000 h. The exposure cycle consisted of  
14 20 min of dew and 20 min of light. The effects of the exposure on the painted surfaces were  
15 measured by weight loss and loss in film thickness. The acrylic coil coating had the lowest  
16 erosion rate of the surface coatings tested. However, ozone was the only pollutant that had a  
17 significant effect on the surface erosion. Sulfur dioxide and relative humidity were significant  
18 factors in the erosion of oil base house paints and vinyl coil coating. The findings for acrylic  
19 latex house paint were not reported.

## 22 **11.6 CONCLUSIONS**

23 Ozone and other photochemical oxidants react with many economically important  
24 man-made materials, decreasing their useful life and aesthetic appearance. Some materials  
25 known to be damaged by ozone include elastomers, fibers and dyes, and paints. Most studies  
26 have been on single compounds rather than complex materials.

27 The elastomeric compounds, natural rubber and synthetic polymers and copolymers of  
28 butadiene, isoprene, and styrene, are particularly susceptible to even low concentrations of  
29 ozone. Ozone damages these compounds by breaking the molecular chain at the carbon-carbon  
30 double bond; a chain of three oxygen atoms is added directly across the double bond. The  
31 change in structure promotes the characteristic cracking of stressed/stretched rubber called

1 “weathering.” Tensile strain produces cracks on the surface of the rubber that increase in  
2 number with increased stress/stretching.

3 The rate of crack growth is dependent on the degree of stress, the type of rubber  
4 compound, ozone concentration, time of exposure, ozone velocity, and temperature. After initial  
5 cracking, there is further ozone penetration, resulting in additional cracking and, eventually,  
6 mechanical weakening or stress relaxation.

7 Ozone can damage textiles and fabrics by methods similar to those associated with  
8 elastomers. Generally, synthetic fibers are less affected by ozone than natural fibers; however,  
9 ozone contribution to the degradation of textiles and fabrics is not considered significant .

10 Ozone fading of textile dyes is a diffusion-controlled process; the rate of fading is  
11 controlled by the diffusion of the dye to the fiber surface. Many textile dyes react with ozone.  
12 The rate and severity of the ozone attack is influenced by the chemical nature of the textile fiber  
13 and the manner in which the dye is applied.

14 Several artists’ pigments are also sensitive to fading and oxidation by ozone when exposed  
15 to concentrations found in urban areas.

16



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